Electronic Supplementary Information

Switching nuclearity and Co(II) content through stoichiometry adjustment: {Co^{II}₆Co^{III}₃} and {Co^{II}Co₄^{III}} mixed valent complexes and their magnetic properties study

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| | 1 | 2 |
|---|--|--|
| Empirical Formula | C ₆₈ H ₁₃₈ Co ₉ N ₃ O ₃₆ | C ₄₆ H _{90.8} Co ₅ N ₅ O ₂₆ |
| Formula weight | 2104.18 | 1424.68 |
| $T(\mathbf{K})$ | 293 (2) | 151 (1) |
| Crystal system | Triclinic | Monoclinic |
| Space Group | P-1 | P21/n |
| (9) | | |
| a (A) | 14.7903 (15) | 20.3316 (7) |
| <i>b</i> (A) | 15.2087 (12) | 10.9609 (5) |
| <i>c</i> (A) | 25.894 (2) | 28.8394 (12) |
| α (°) | 91.970 (6) | 90 |
| $\beta(^{\circ})$ | 104.308 (8) | 108.706 (4) |
| $\gamma(^{\circ})$ | 118.052 (9) | 90 |
| $V(Å^3)$ | 4903.1 (7) | 6087.44 |
| Z | 2 | 4 |
| $D_{\text{calc}} (\text{mg/m}^3)$ | 1.425 | 1.555 |
| μ (mm ⁻¹) | 1.558 | 1.418 |
| <i>F</i> (000) | 2196 | 2979 |
| Crystal size (mm) | 0.16 x 0.21 x 0.37 | 0.09 x 0.14 x 0.59 |
| Crystal color/shape | purple blocks | red blocks |
| Radiation, graphite | $M_0 K \alpha \lambda = 0.71073 \text{ Å}$ | $M_0 K \alpha \lambda = 0.71073 \text{ Å}$ |
| monochr. | | |
| θ Range | 357 - 2889 | 355 - 270 |
| data collection (°) | | |
| Index ranges | $-13 \le h \le 18$ | $-24 \le h \le 25$ |
| | $-19 \le k \le 19$ | $-13 \le k \le 13$ |
| | $-32 \le l \le 33$ | $-36 \le l \le 35$ |
| Reflections | $41016/20972(R_{int}=0.1092)$ | $41683/13141(R_{int}=0.0713)$ |
| collected/unique | (int) | , (int |
| Observed reflections | 9993 | 8733 |
| $\frac{[1>2\sigma(1)]}{C_{\text{constrained}}}$ | 00.1 | 07.0 |
| Completeness (%) | 99.1 | 97.9 |
| transmission | 1.000 / 0.548 | 1.000 / 0.819 |
| Pafinament mathad | full matrix losst squares on E^2 | full matrix losst squares on E^2 |
| Kermement method | $\frac{1}{1} = \frac{1}{2} \frac{1}{1} $ | $1/[-2(E^2) + (0.0287D)^2 + 11.8422D]$ |
| Weights, w | where $P = (F_0^2 + 2F_c^2)/3$ | where $P = (F_0^2 + 2F_c^2)/3$ |
| Data/restraints/parameters | 20972/105/1098 | 13141/4/748 |
| Goodness-of-fit (GOF) | 1.028 | 1.053 |
| on F^2 | 1.028 | 1.035 |
| Final <i>R</i> -index $[I \ge 2\sigma(I)]$ | 0 1010/ 0 1840 | 0.0624/.0.1067 |
| / all data | 0.1010/ 0.1049 | 0.0024/ 0.100/ |
| wR index $[I > 2\sigma(I)]$ | 0 2287/ 0 3027 | 0 1179/ 0 1387 |
| /all data | 0.22077 0.3027 | 0.1177/ 0.1507 |
| Largest peak and hole | 1 037 and -0 878 | 0.658 and -0.579 |
| $(e A^{-3})$ | 1.057 and -0.076 | 0.050 and -0.577 |

Table S1. Crystallographic data of 1 and 2.

| 1 | | |
|--------------|-----------|--|
| N17—Co4 | 1.965(5) | |
| N15—Co1 | 1.981(4) | |
| 011—Co1 | 1.912(3) | |
| O19—Co2 | 2.360(3) | |
| O19—Co3 | 1.896(4) | |
| 019—Co1 | 1.892(3) | |
| O25—Co2 | 2.041(3) | |
| O25—Co1 | 1.872(3) | |
| O110—Co5 | 1.884(4) | |
| O110—Co4 | 1.879(3) | |
| O28—Co5 | 1.867(4) | |
| O28—Co4 | 1.921(4) | |
| O27—Co4 | 1.887(3) | |
| O27—Co2 | 2.031(4) | |
| O23—Co4 | 1.915(3) | |
| O15—Co3 | 1.901(3) | |
| O15—Co1 | 1.870(4) | |
| O16—Co3 | 1.878(4) | |
| O26—Co2 | 2.036(4) | |
| O26—Co3 | 1.898(3) | |
| N16—Co3 | 1.964(4) | |
| O21—Co3 | 1.914(3) | |
| O22—Co1 | 1.916(4) | |
| O14—Co5 | 1.916(4) | |
| O18—Co5 | 1.869(3) | |
| O18—Co2 | 2.004(4) | |
| O13—Co5 | 1.933(3) | |
| O47—Co4 | 1.890(4) | |
| N18—Co5 | 1.978(6) | |
| Co5—Co4 | 2.772(1) | |
| Co3—Co1 | 2.7701(9) | |
| Co2—O19—Co3 | 94.3(1) | |
| Co2-019-Co1 | 94.4(1) | |
| Co3-019-Co1 | 94.0(1) | |
| Co2-025-Co1 | 106.5(2) | |
| Co5-0110-Co4 | 94.9(2) | |
| Co5—O28—Co4 | 94.0(2) | |

Co4—O27—Co2

Co3-015-Co1

Co2—O26—Co3 Co5—O18—Co2 112.5(2)

94.5(2) 105.7(2)

112.6(2)

Table S2. Main bond angles (°) and distances (Å) of 1 and 2.

| 2 | | |
|--|----------------------|--|
| Co1-032 | 1.95(1) | |
| Co1-02 | 1.978(8) | |
| Co1-010 | 1.959(6) | |
| Co1-030 | 1.98(1) | |
| $C_{02} - 013$ | 2 14(1) | |
| $C_{02} - 016$ | 2.03(1) | |
| $C_{02} - 014$ | 2.098(9) | |
| $C_{02} = 02$ | 2 133(8) | |
| $C_0^2 = 01$ | 2.045(5) | |
| $C_{02} = 0.033$ | 2.019(9) 2 289(8) | |
| $C_{03} - 01$ | 2.209(0) 2 050(8) | |
| $C_{03} - 05$ | 2.030(0) 2.002(5) | |
| $C_{03} - 018$ | 2.002(3) | |
| $C_{03} - 017$ | 2.07(1) | |
| C_{03} 017 | 2.020(8) | |
| Col_013 | 2.024(0) 1 014(0) | |
| $C_{04} = 020$ | 1.914(9) 1.842(7) | |
| $C_04 = 0.010$ | 1.043(7) | |
| $C_{04} = 019$ | 1.071(0) 1.052(6) | |
| $C_04 = 07$ | 1.932(0) | |
| C_04 N_2 | 1.901(9) | |
| $C_{04} = 00$ | 1.070(0) 1.002(7) | |
| $C_{05} = 0.0000000000000000000000000000000000$ | 1.092(7) | |
| $\frac{C05-022}{Co5-N2}$ | 1.891(0) | |
| C_{0} | 1.98(1) | |
| $C_{05} = 021$ | 1.941(9) | |
| $\frac{1000-07}{1000}$ | 1.894(3) | |
| $C_{00} = 00$ | 1.929(7) | |
| $C_{00} = 029$ | 2.049(8) | |
| $\begin{array}{c} C00 - 02 \\ \hline C_2 (- 01 \\ \hline \end{array}$ | 2.101(7) | |
| $\begin{array}{c} Cob-OI \\ \hline Coc \left(-O4 \right) \end{array}$ | 2.068(8) | |
| $\begin{array}{c} C0004 \\ \hline C_{2}(0) \end{array}$ | 2.081(5) | |
| $\begin{array}{c} C06-09 \\ \hline \end{array}$ | 2.256(8) | |
| $C_{00} = 03$ | 2.110(6) | |
| $C_0/-024$ | 2.100(8) | |
| $C_0/-08$ | 2.15/(9) | |
| $C_07 - 023$ | 2.022(7) | |
| <u>Co7</u> _034 | 2.054(8) | |
| <u>Co7</u> _04 | 2.063(6) | |
| <u>Co7</u> _03 | 2.097(6) | |
| 026 | 2.200(8) | |
| <u>Co8</u> <u>O2</u> / | 2.115(8) | |
| 09 | 2.134(6) | |
| <u>Co8</u> <u>O4</u> | 1.997(7) | |
| <u>Co8</u> 08 | 2.252(8) | |
| Co8—O28 | 2.039(9) | |
| <u>Co9</u> _010 | 1.890(7) | |
| Co9—O8 | 1.903(7) | |

| Со9—О9 | 1.894(7) |
|---------|----------|
| Co9—O3 | 1.899(8) |
| Co9—N1 | 1.91(1) |
| Co9—O25 | 1.910(8) |
| Co2—Co6 | 3.070(2) |
| Co4—Co5 | 2.837(2) |
| Co6—Co8 | 3.008(2) |
| Со7—Со9 | 2.893(3) |

| Co2—O1—Co3 | 110.2(3) |
|-------------|----------|
| Co2—O1—Co6 | 96.5(3) |
| Co3—O1—Co6 | 134.4(3) |
| Co1—O10—Co9 | 121.9(4) |
| Co1—O2—Co2 | 113.1(4) |
| Co1—O2—Co6 | 114.9(4) |
| Co2—O2—Co6 | 92.9(3) |
| Соб—ОЗ—Со7 | 99.8(3) |
| Соб—ОЗ—Со9 | 101.5(3) |
| Со7—О3—Со9 | 92.6(3) |
| Co5—O34—Co7 | 118.0(3) |
| Соб—О4—Со7 | 102.1(3) |
| Co6—O4—Co8 | 95.0(2) |
| Co7—O4—Co8 | 105.2(3) |
| Co3—O5—Co4 | 109.9(3) |
| Co4—O6—Co5 | 96.4(3) |
| Co4—O7—Co5 | 95.1(3) |
| Co7—O8—Co8 | 94.0(3) |
| Со7—О8—Со9 | 90.7(3) |
| Co8—O8—Co9 | 99.6(3) |
| Соб—О9—Со8 | 86.5(2) |
| Соб—О9—Со9 | 96.8(3) |
| Со8—О9—Со9 | 104.2(3) |



Figure ESI1. Intra-molecular H-bonding interactions in compound 1 structure. Hydrogen atoms and tert-butyl groups have been omitted for sake of clarity. Atoms colours code: violet: cobalt; red: oxygen; blue: nitrogen and gray: carbon.



Figure ESI2. Crystal packing views of compound **1**. Top: *a*-axis direction; Bottom: *b*-axis direction. Hydrogen atoms have been omitted for sake of clarity.



Figure ESI3. Inter-molecular H-bonding interactions in compound 1 structure. Hydrogen atoms and tert-butyl groups have been omitted for sake of clarity. Atoms colours code: violet: cobalt; red: oxygen; blue: nitrogen and gray: carbon. Symmetry codes, a: -x, -y, -z; b: x, -1+y, z; c: -x, 1-y, z.



Figure ESI4. Molecular representation of the tetrahedral environment of Co(II) ion in compound **2**, remarking the short contact Co2-O19 distance. Hydrogen atoms and tertbutyl groups have been omitted for sake of clarity. Atoms colours code: violet: cobalt; red: oxygen; blue: nitrogen and gray: carbon.



Figure ESI5. Crystal packing view of compound **2** along *b*-axis direction. Hydrogen atoms have been omitted for sake of clarity.



Figure ESI6. Low lying spin multiplet energy level plot arising from best fitting parameters of compound **1**. The dotted line remarks the level crossing field.



Figure ESI7. Powder X-band EPR spectra at 1.5 and 5 K of compound **2** as $S_{eff}=1/2$. Simulation linewidth: 260 Oe; *g*-strain: [0,0.3,0.3]; *H*-strain/ Oe: [0, 2300, 210].



Figure ESI8. Spin density (isosurface: 0.03 a.u.; white: positive; blue: negative) of the different single-determinant configurations calculated for the extraction of the exchange coupling constant values. First line: HS state; From top-bottom and left-right: BS1 to BS9 states respectively.



Figure ESI9. *D* vs *g* error surface contour plot from magnetization data simulations of compound **2**.



Figure ESI10. Compound 2 χ_m '' vs external applied H_{DC} , 0-1500 Hz (driving field 3 Oe) plot, at 2-K.



Figure ESI11. Compound 2 χ_m '' vs driving frequency (logarithmic scale), 0-1500 Hz (driving field 3 Oe) plot, in the range 2-12.5 K under a 1400 Oe DC applied field.



Figure ESI12. Cole-Cole plot at 2600 Oe DC applied field of compound **2**, in the 2-6 K range. Circles: experimental data; Lines: best fitting (see Text).



Figure ESI13. Arrhenius type plot for the *T* dependence of the characteristic relaxation time at 1400 Oe DC external field for compound **2**. Squares: experimental data; Full lines: best fitting curves (see Text).