Neutral and methylated cationic Re(I) tetrazolato complexes: comparative photophysics, transient Raman spectroscopy and fabrication of OLED and LEEC devices

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Electronic Supplementary Information

X-ray Structural Determinations



Figure S1. Molecular structure of fac-[Re(**phen**)(CO)₃(**Tbdz**CH₃)]⁺ with key atoms labelled. Displacement ellipsoids are at 30 % probability level. H-atoms have been omitted for clarity.



Figure S2. Molecular structure of fac-[Re(**bipy**)(CO)₃(**Tbdz**CH₃)]⁺ with key atoms labelled. Displacement ellipsoids are at 30 % probability level. H-atoms have been omitted for clarity.



Figure S3. Molecular structure of fac-[Re(**phen**)(CO)₃(**Tmeb**CH₃)]⁺ with key atoms labelled. Displacement ellipsoids are at 30 % probability level. H-atoms have been omitted for clarity.



Figure S4. Molecular structure of fac-[Re(**bipy**)(CO)₃(**Tmeb**CH₃)]⁺ with key atoms labelled. Displacement ellipsoids are at 30 % probability level. H-atoms have been omitted for clarity.

Absorption Spectra



Figure S5. Absorption spectra of dichloromethane solutions (ca. 10^{-5} M) of the *fac*-[Re(**bipy**)(CO)₃(L)][PF₆] complexes, where L = **Tph**CH₃ (red trace), **Tbdz**CH₃ (blue trace) and **Tmeb**CH₃ (black trace).



Figure S6. Absorption spectra of dichloromethane solutions (ca. 10^{-5} M) of the *fac*-[Re(**phen**)(CO)₃(**L**)][PF₆] complexes, where **L** = **Tph**CH₃ (red trace), **Tbdz**CH₃ (blue trace) and **Tmeb**CH₃ (black trace).





Figure S7 Comparison of the emission spectra obtained from dichloromethane fluid solutions at RT (red trace) and 77 K (blue trace) containing fac-[Re(**phen**)(CO)₃(**Tph**CH₃)][PF₆] (top) and fac-[Re(**bipy**)(CO)₃(**Tph**CH₃)][PF₆], (bottom).



Figure S8. Comparison of the emission spectra obtained from dichloromethane fluid solutions at RT (red trace) and 77 K (blue trace) containing fac-[Re(**phen**)(CO)₃(**Tbdz**CH₃)][PF₆] (top) and fac-[Re(**bipy**)(CO)₃(**Tbdz**CH₃)][PF₆], (bottom).



Figure S9. Comparison of the emission spectra obtained from dichloromethane fluid solutions at RT (red trace) and 77 K (blue trace) containing fac-[Re(**phen**)(CO)₃(**Tmeb**CH₃)][PF₆] (top) and fac-[Re(**bipy**)(CO)₃(**Tmeb**CH₃)][PF₆], (bottom).

TD-DFT Calculations

Table S1. Summary of TD-DFT data for fac-[Re(phen)(CO)₃(TphCH₃)]⁺ and its neutral counterpart.

					Group Electron Transition Density (%)			
L=	Transition #	Wavelength (nm)	f	Orbital contributions	Re(CO)),	Phen	L
	1	435	0.0432	HOMO->LUMO (97%)	37 →	5	3 → 94	61 → 0
Tph	2	406	0.0071	H-1->LUMO (98%)	91 →	5	6 → 94	$3 \rightarrow 0$
	3	399	0.0091	HOMO->L+1 (97%)	$37 \rightarrow$	1	$3 \rightarrow 99$	$61 \rightarrow 0$
	4	383	0.0009	H-3->LUMO (48%), H-2->LUMO (51%) H-3->LUMO (12%), H-1->L+1 (79%), H-2->LUMO	71 →	5	6 → 94	$24 \rightarrow 0$
	5	369	0.0026	(8%) H-3->LUMO (39%), H-2->LUMO (40%), H-1->L+1	88 →	2	6 → 98	6 → 0
	6	357	0.0796	(19%)	75 →	4	$6 \rightarrow 95$	$19 \rightarrow 0$
	7	347	0.0009	H-3->L+1 (40%), H-2->L+1 (59%)	70 →	1	$6 \rightarrow 99$	$24 \rightarrow 0$
	8	334	0.0123	H-3->L+1 (58%), H-2->L+1 (38%)	73 →	1	$5 \rightarrow 99$	$22 \rightarrow 0$
	9	314	0.0051	H-5->LUMO (80%), H-4->LUMO (8%)	6 →	5	$75 \rightarrow 95$	$19 \rightarrow 0$
	10	310	0.0001	H-4->LUMO (92%)	1 →	5	6 → 94	93 → O
	1	393	0.0017	HOMO->LUMO (98%)	89 →	6	6 → 93	5 → 1
	2	373	0.0576	H-2->LUMO (18%), H-1->LUMO (78%)	82 \rightarrow	6	$16 \rightarrow 93$	$2 \rightarrow 1$
	3	363	0.0295	H-2->LUMO (76%), H-1->LUMO (11%), HOMO->L+1 (12%)	92 →	6	$7 \rightarrow 94$	1 -> 1
	4	349	0.0778	H-1->LUMO (11%), HOMO->L+1 (82%)	88 →	2	$7 \rightarrow 98$	$5 \rightarrow 0$
	5	338	0.0213	H-1->L+1 (91%)	77 →	1	21 → 98	$2 \rightarrow 0$
TphCH₃	6	327	0.0068	H-2->L+1 (94%)	94 →	1	$6 \rightarrow 99$	$1 \rightarrow 0$
(N1)	7	314	0.0008	H-3->LUMO (89%), H-6->L+1 (7%)	15 →	6	83 → 93	$2 \rightarrow 1$
	8	297	0.0003	H-8->LUMO (35%), H-3->L+1 (63%)	11 →	3	$88 \rightarrow 97$	$2 \rightarrow 0$
	9	291	0.0248	H-1->L+2 (12%), H-1->L+3 (20%), H-1->L+4 (13%), HOMO->L+2 (16%), HOMO->L+3 (18%), H-1->L+5 (8%)	83 →	24	14 → 22	4 → 54
	10	290	0.0271	H-1->L+2 (14%), H-1->L+3 (18%), HOMO->L+2 (17%), HOMO->L+3 (18%), HOMO->L+4 (11%), HOMO->L+5 (6%)	84 →	23	12 → 21	4 → 55
	1	389	0.0014	H-1->LUMO (98%)	92 →	6	7 → 93	1 → 0
	2	382	0.0808	HOMO->LUMO (87%), H-2->LUMO (9%)	$81 \rightarrow$	6	$13 \rightarrow 94$	$6 \rightarrow 0$
TphCH₃ (N2)	3	372	0.0215	H-2->LUMO (89%), HOMO->LUMO (8%)	96 \rightarrow	6	4 → 93	$1 \rightarrow 0$
	4	348	0.0255	HOMO->L+1 (85%), H-1->L+1 (9%)	$78 \rightarrow$	1	$16 \rightarrow 99$	$6 \rightarrow 0$
	5	346	0.0416	H-1->L+1 (84%), HOMO->L+1 (9%)	$90 \rightarrow$	1	$8 \rightarrow 99$	$2 \rightarrow 0$
	6	333	0.0023	H-2->L+1 (98%)	97 \rightarrow	1	$3 \rightarrow 99$	$0 \rightarrow 0$
	7	314	0.0055	H-4->LUMO (21%), H-3->LUMO (87%), H-8->L+1 (7%)	$7 \rightarrow$	6	54 → 94	$40 \rightarrow 0$
	8	300	0.0013	H-1->L+2 (71%), H-1->L+3 (21%)	92 \rightarrow	5	7 → 2	$1 \rightarrow 93$
	9	297	0.0003	H-8->LUMO (38%), H-4->L+1 (15%), H-3->L+1 (45%)	$4 \rightarrow$	3	$69 \rightarrow 97$	$27 \rightarrow 0$
	10	295	0.0334	HOMO->L+2 (82%), HOMO->L+5 (7%), H-4->LUMO (7%)	74 \rightarrow	15	15 → 9	$10 \rightarrow 76$

Table S2. Summary of TD-DFT data for fac-[Re(**phen**)(CO)₃(**Tmeb**CH₃)]⁺ and its neutral counterpart.

					Group Electron Transition Density (%)				
L=	Transition #	Wavelength (nm)	f	Orbital contributions	Re(CO)	Phen	L		
	1	424	0.0513	HOMO->LUMO (95%)	50 →	5 4 →	$94 45 \rightarrow 0$		
Tmah	2	405	0.0124	H-1->LUMO (97%)	90 →	$5 6 \rightarrow$	$94 3 \rightarrow 0$		
	3	389	0.0120	HOMO->L+1 (97%)	50 →	$1 4 \rightarrow$	$99 46 \rightarrow 0$		
	4	382	0.0001	H-2->LUMO (94%)	88 →	$5 2 \rightarrow$	$94 9 \rightarrow 0$		
	5	366	0.0150	H-1->L+1 (93%)	89 →	$1 6 \rightarrow$	$99 4 \rightarrow 0$		
Theb	6	346	0.0476	H-3->LUMO (72%), H-2->L+1 (18%)	52 → ·	4 8 →	$95 40 \rightarrow 0$		
	7	346	0.0114	H-3->LUMO (17%), H-2->L+1 (77%)	80 <i>-</i>	$2 4 \rightarrow$	98 17 \rightarrow 0		
	8	323	0.0108	H-3->L+1 (91%)	$39 \rightarrow$	1 11 →	$99 50 \rightarrow 0$		
	9	313	0.0336	H-5->LUMO (74%), H-4->LUMO (10%), H-6->L+1 (6%)	8 →	$5 67 \rightarrow$	$90 25 \rightarrow 5$		
	10	308	0.4649	HOMO->L+2 (90%)	48 →	$4 \qquad 7 \rightarrow$	$3 \qquad 45 \rightarrow 93$		
	1	392	0.0018	HOMO->I LIMO (98%)	89 →	$6 \rightarrow$	93 $5 \rightarrow 1$		
	2	372	0.0607	H-2-> UMO (20%) H-1-> UMO (74%)	82 →	$6 16 \rightarrow$	93 $2 \rightarrow 1$		
	3	363	0.0350	H-2->LUMO (75%) H-1->LUMO (13%) HOMO->L+1 (10%)	92 →	$5 7 \rightarrow$	93 $1 \rightarrow 1$		
TmebCH₃ (N1)	4	348	0.0785	H-1->LUMO (10%) HOMO->L+1 (83%)	88 →	2 8 →	98 $4 \rightarrow 0$		
	5	337	0.0218	H-1->I +1 (90%)	77 →	1 22 →	98 $2 \rightarrow 0$		
	6	327	0.0064	H-2->L+1 (94%)	94 →	1 5 →	$0 \rightarrow 0$		
(/	7	314	0.0006	H-3->LUMO (79%) HOMO->L+2 (11%) H-4->L+1 (6%)	24 →	6 74 →	$82 2 \rightarrow 12$		
	8	313	0.0188	H-3->LUMO (10%), HOMO->L+2 (86%)	81 →	2 15 →	$11 5 \rightarrow 87$		
	9	305	0.0079	H-1->L+2 (90%)	7 9 →	2 19 →	$1 2 \rightarrow 96$		
	10	297	0.0001	H-4->LUMO (35%), H-3->L+1 (60%)	14 →	3 85 →	93 $1 \rightarrow 4$		
	1	388	0.0017	H-1->I UMO (96%)	91 →	6 7 →	93 1 \rightarrow 0		
TmebCH₃ (N2)	2	381	0.0793	H-2->LUMO (10%) HOMO->LUMO (84%)	82 →	6 13 →	93 $5 \rightarrow 0$		
	3	371	0.0230	H-2->LUMO (89%) HOMO->LUMO (9%)	$95 \rightarrow$	$6 4 \rightarrow$	93 $1 \rightarrow 0$		
	4	347	0.0293	H-1->L+1 (16%), HOMO->L+1 (78%)	81 →	$1 \qquad 14 \rightarrow$	$99 5 \rightarrow 0$		
	5	345	0.0381	H-1->L+1 (77%), HOMO->L+1 (16%)	88 →	1 10 →	$99 2 \rightarrow 0$		
	6	333	0.0023	H-2->L+1 (98%)	$97 \rightarrow$	1 3 →	$0 \rightarrow 0$		
	7	318	0.0070	HOMO->L+2 (97%)	$79 \rightarrow$	1 15 →	$0 \qquad 6 \rightarrow 99$		
	8	316	0.0082	H-1->L+2 (89%), H-1->L+3 (7%)	$90 \rightarrow$	1 9 →	$1 \rightarrow 96$		
	9	314	0.0041	H-3->LUMO (84%), H-5->L+1 (6%)	1 3 →	6 82 →	$91 6 \rightarrow 3$		
	10	305	0.0002	H-2->L+2 (93%), H-2->L+3 (6%)	$97 \rightarrow$	$1 3 \rightarrow$	$0 0 \rightarrow 98$		

Raman Spectroscopy



Figure S10. FT-Raman and calculated Raman spectra for *fac*-[Re(phen)(CO)₃(TphCH₃)]⁺.



Figure S11. FT-Raman and calculated Raman spectra for *fac*-[Re(phen)(CO)₃(TmebCH₃)]⁺.

Resonance Raman Spectroscopy



Figure S12. Raman spectra of the neutral *fac*-[Re(**phen**)(CO)₃(**Tph**)]. Tetrazolate modes are labelled with dotted lines and **phen** modes with dashed lines.



Figure S13. Raman spectra of the cationic *fac*-[Re(**phen**)(CO)₃(**Tph**CH₃)]⁺. Methyltetrazole modes are labelled with dotted lines and **phen** modes with dashed lines.



Figure S13. Raman spectra of the neutral *fac*-[Re(**phen**)(CO)₃(**Tbdz**)]. Tetrazolate modes are labelled with dotted lines and **phen** modes with dashed lines.



Figure S14. Raman spectra of the cationic *fac*-[Re(**phen**)(CO)₃(**Tbdz**CH₃)]⁺. Methyltetrazole modes are labelled with dotted lines and **phen** modes with dashed lines.

LEEC Performance



Figure S15. Luminance as a function of time for the LEEC devices having *fac*- $[\text{Re}(\text{phen})(\text{CO})_3(\text{TphCH}_3)]^+$ as a single component in the active layer, operating under the bias of 6 (back), 7 (blue) and 8 V (red line).



Figure S16. Electroluminescence spectrum of LEEC device having the *fac*- $[Re(phen)(CO)_3(TphCH_3)]^+$ complex as an emitting layer.