Electronic Supplementary Information (ESI)

Europium doped LaF_3 nanocrystals with organic 9-oxidophenalenone capping ligands that display visible light excitable steady-state blue and time delayed red emission

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Figure S2: TEM images of citrate capped $5\% \text{ Eu}^{3+}$ -doped LaF₃ NCs (above) and histogram showing TEM size distribution of the NCs.



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Figure S4: Electrospray ionisation mass spectrum in positive ion mode (ESI-MS⁺) of 9hydroxyphenalenone ligand recorded on Bruker Maxis Impact mass spectrometer.



Figure S5: The ¹H NMR spectrum of 9-hydroxyphenalenone in $CDCl_3$, (a) full spectrum and (b) expansion of 7.1-8.2 ppm region. The spectrum was recorded on 400 MHz Bruker Avance III spectrometer. The proton resonances are assignments according to the literature (*J. Chem. Edu.*, 2008, **85**, 413).



Figure S6: The ¹H NMR spectrum 9-oxidophenalenone coated \sim 5% Eu³⁺-doped LaF₃ NCs in DMSO-d₆. The spectrum was recorded on 400 MHz Bruker Avance III spectrometer.

The proton resonances observed in the spectrum are weak due to poor dispersibility of the NCs. However, the proton resonances at 8.4 and 8.3 ppm, both doublets and integrating for two protons each, can be easily related to the similar resonances observed in the ¹H NMR spectrum of free 9-hydroxyphenalenone ligand at 8.10 ppm and 8.03 ppm and assigned to H-3, H-7 and H-4, H-6 protons, respectively (Figure S5, ESI⁺). The triplet observed at 7.74 ppm, which integrates for one proton can be related to the triplet observed at 7.61 ppm and assigned to H-5 proton in the free 9-hydroxyphenalenone ligand. Finally a doublet observed at 7.25 ppm and integrating for two protons can be easily related to a doublet at 7.18 ppm in the free 9-hydroxyphenalenone and assigned to H-2 and H-8 protons. No resonance around 16 ppm due to OH protons is observed in the 9-oxidophenalenone capped NCs. The resonances observed in the free ligand are clearly very sharp as compared to the resonances observed in the NCs (Figure S5, ESI⁺). A sharp resonance at 7.6 ppm, integrating for less than one proton with respect to resonances identified for capping ligands, could most probably be due to some solvent impurity in the NMR tube and is visible only due to weak intensity (because of poor dispersibility of NCs in DMSO-d6) of the proton resonances of the NCs.



Figure S7: An overlap of the absorption spectra of 9-hydroxyphenalenone ligand (red) and 9oxidophenalenone coated \sim 5% Eu³⁺-doped LaF₃ NCs (black) in THF.



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