

## ELECTRONIC SUPPLEMENTARY INFORMATION

### Gas sensing and electrochemical properties of tetra and octa 2H-chromen-2-one substituted iron(II) phthalocyanines

Selçuk Altun, Efe Baturhan Orman, Zafer Odabaş, Ahmet Altındal and Ali Rıza Özkaya

**Table S1.** Elemental analysis, spectral and MALDI-TOF data of coumarin substituted FePcs (**5-8**).

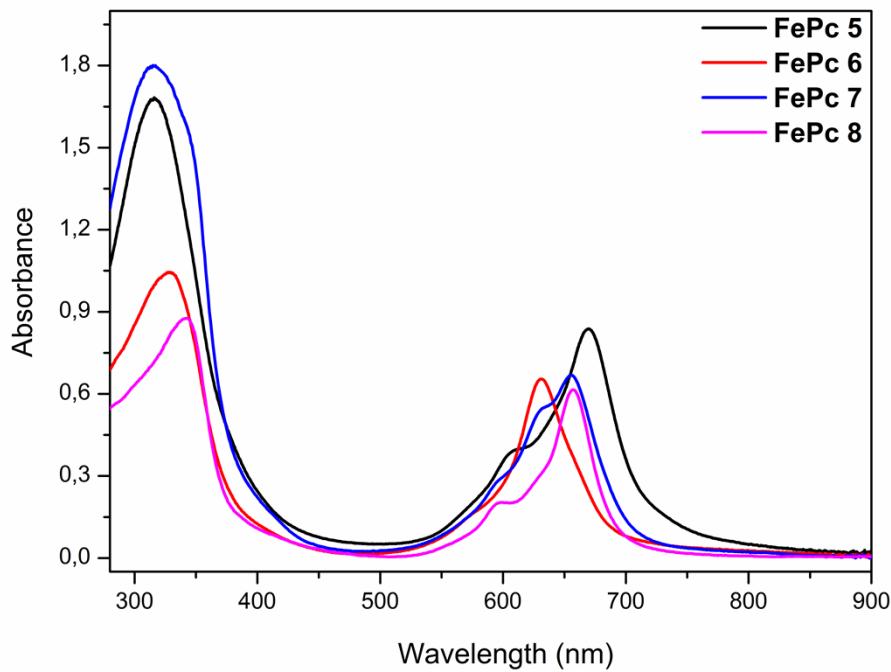
Pcs	Empirical Formula (MALDI-TOF (m/z))	Elemental Analysis (%) Found (Calc.)	UV-vis absorption in DMSO (λnm (log <sub>e</sub> ))	IR Spectral Data (cm <sup>-1</sup> )	Yield (%)
<b>5</b>	C <sub>100</sub> H <sub>64</sub> FeN <sub>8</sub> O <sub>16</sub> (1689.67)	C 71.09(69.91); H 3.82 (3.75); N 6.63(6.81)	315 (5.526), 605 (4.883), 669 (5.225)	742, 833, 956, 1083, 1175, 1244, 1366, 1421, 1480, 1511, 1582, 1603, 1723, 2837, 2927, 3000, 3069	7.62
<b>6</b>	C <sub>100</sub> H <sub>64</sub> FeN <sub>8</sub> O <sub>16</sub> (1689.63)	C 71.09(70.89); H 3.82 (4.01); N 6.63(6.74)	328 (5.321), 631(5.115)	747, 832, 955, 1080, 1174, 1248, 1366, 1469, 1511, 1595, 1728, 2837, 2929, 3065	9.16
<b>7</b>	C <sub>100</sub> H <sub>64</sub> FeN <sub>8</sub> O <sub>16</sub> (1827.26)	C 65.73(65.57); H 3.31 (3.43); N 6.13(5.95)	316 (5.557), 594 (4.739), 631(5.037), 656 (5.125)	744, 838, 897, 1045, 1085, 1178, 1250, 1369, 1410, 1456, 1513, 1604, 1731, 2854, 2926, 3007, 3071	19.66
<b>8</b>	C <sub>168</sub> H <sub>112</sub> FeN <sub>8</sub> O <sub>32</sub> (2810.74)	C 71.79(71.61); H,4.02 ( 3.88); N, 3.99(4.11)	344 (5.243), 595 (4.602), 657 (5.090)	747, 836, 888, 1087, 1178, 1252, 1369, 1398, 1438, 1513, 1599, 1732, 2905, 2956, 2999, 3071	9.60

**Table S2** The humidity dependence of the sensor parameters for the 60 nm **6** coated film kept at 298 K and exposed to various concentrations of CO<sub>2</sub> gas in air.

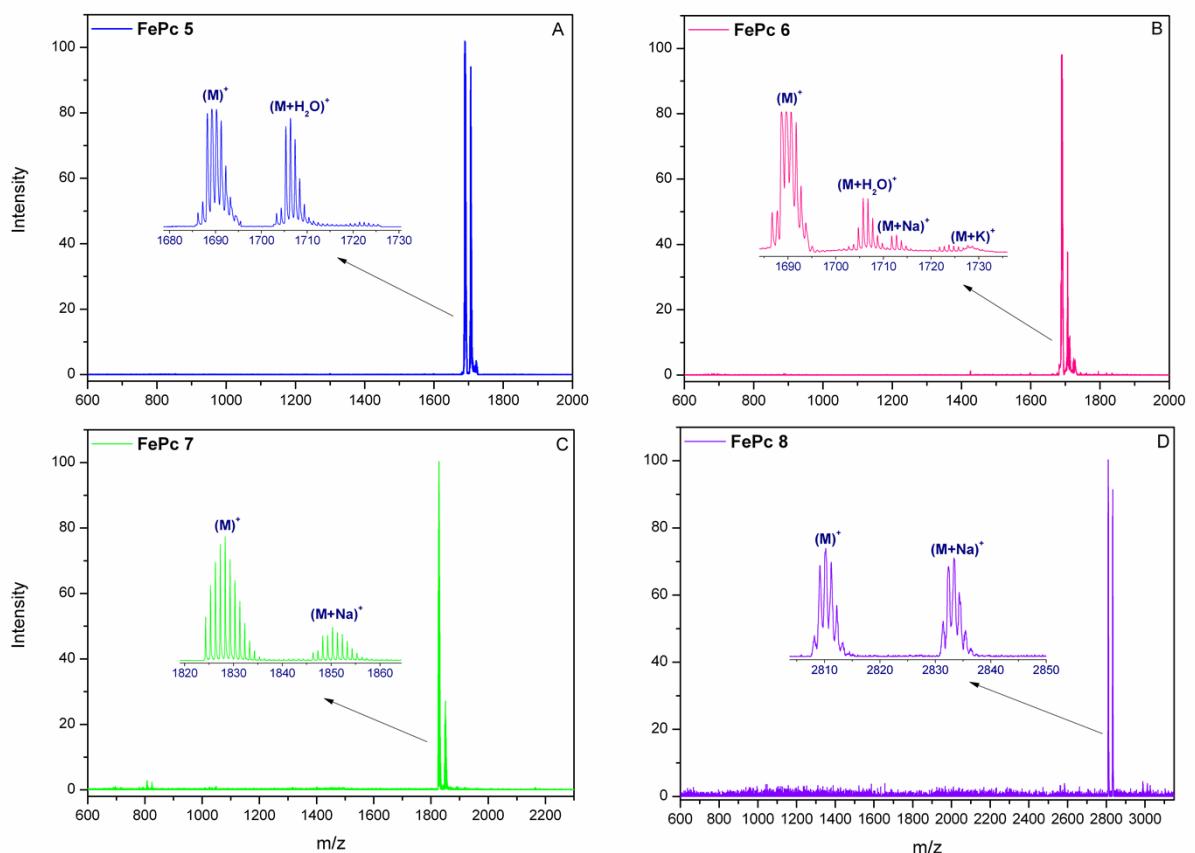
RH	200 ppm CO <sub>2</sub>			400 ppm CO <sub>2</sub>			600 ppm CO <sub>2</sub>			800 ppm CO <sub>2</sub>			1000 ppm CO <sub>2</sub>		
	S (ppm) <sup>-1</sup>	τ <sub>90</sub> (s)	τ <sub>10</sub> (s)	S (ppm) <sup>-1</sup>	τ <sub>90</sub> (s)	τ <sub>10</sub> (s)	S (ppm) <sup>-1</sup>	τ <sub>90</sub> (s)	τ <sub>10</sub> (s)	S (ppm) <sup>-1</sup>	τ <sub>90</sub> (s)	τ <sub>10</sub> (s)	S (ppm) <sup>-1</sup>	τ <sub>90</sub> (s)	τ <sub>10</sub> (s)
0%	7.36x10 <sup>-3</sup>	250	238	8.30x10 <sup>-3</sup>	240	215	9.38x10 <sup>-3</sup>	185	230	1.10x10 <sup>-2</sup>	135	200	1.12x10 <sup>-2</sup>	158	180
40%	6.41x10 <sup>-3</sup>	230	280	6.93x10 <sup>-3</sup>	240	200	8.38x10 <sup>-3</sup>	190	340	8.54x10 <sup>-3</sup>	140	205	1.00x10 <sup>-2</sup>	150	230
50%	5.76x10 <sup>-3</sup>	230	240	6.00x10 <sup>-3</sup>	220	185	7.10x10 <sup>-3</sup>	180	350	7.85x10 <sup>-3</sup>	130	180	8.85x10 <sup>-3</sup>	145	300
60%	5.10x10 <sup>-3</sup>	205	230	5.23x10 <sup>-3</sup>	210	210	6.46x10 <sup>-3</sup>	160	320	7.04x10 <sup>-3</sup>	120	160	7.30x10 <sup>-3</sup>	150	280
70%	4.28x10 <sup>-3</sup>	220	255	4.67x10 <sup>-3</sup>	225	200	5.42x10 <sup>-3</sup>	165	325	6.45x10 <sup>-3</sup>	125	175	7.05x10 <sup>-3</sup>	140	310
80%	3.85x10 <sup>-3</sup>	200	245	4.11x10 <sup>-3</sup>	215	190	4.52x10 <sup>-2</sup>	150	390	6.14x10 <sup>-3</sup>	110	140	6.55x10 <sup>-3</sup>	130	280

**Tabla S3.** The humidity dependence of the sensor parameters for the same sensor (60 nm **6** coated sensor) kept at 298 K and exposed to various concentrations of CO gas in N<sub>2</sub>.

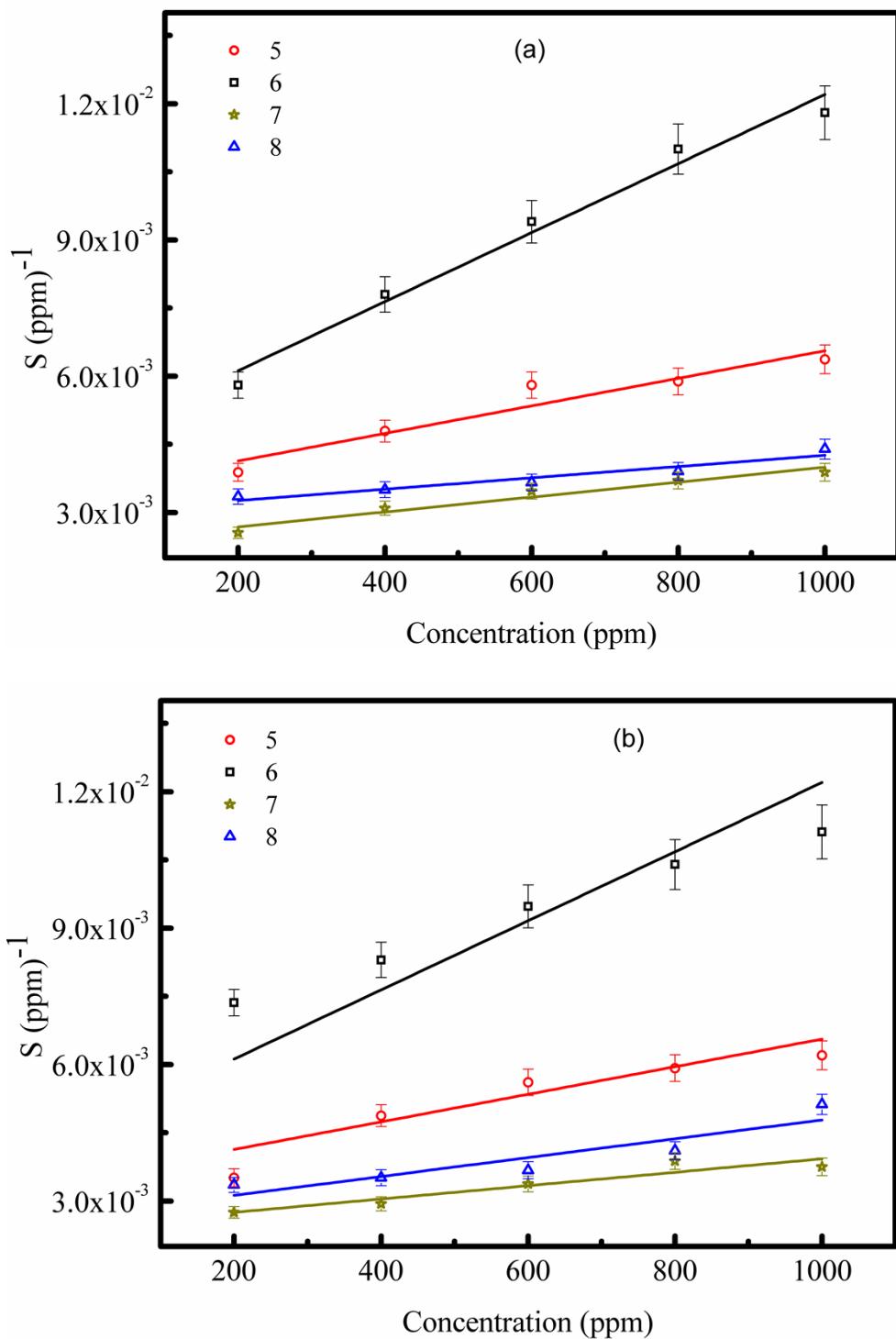
RH	200 ppm CO <sub>2</sub>			400 ppm CO <sub>2</sub>			600 ppm CO <sub>2</sub>			800 ppm CO <sub>2</sub>			1000 ppm CO <sub>2</sub>		
	S (ppm) <sup>-1</sup>	$\tau_{90}$ (s)	$\tau_{10}$ (s)												
0%	1.99x10 <sup>-3</sup>	80	150	5.19x10 <sup>-3</sup>	300	310	5.17x10 <sup>-3</sup>	256	400	5.72x10 <sup>-3</sup>	250	500	6.49x10 <sup>-3</sup>	186	510
40%	2.75x10 <sup>-3</sup>	120	200	5.49x10 <sup>-3</sup>	311	450	5.85x10 <sup>-3</sup>	310	350	6.86x10 <sup>-3</sup>	320	550	8.42x10 <sup>-3</sup>	340	420
50%	3.45x10 <sup>-3</sup>	220	170	6.12x10 <sup>-3</sup>	270	550	7.22x10 <sup>-3</sup>	300	370	8.58x10 <sup>-3</sup>	415	500	9.24x10 <sup>-3</sup>	390	430
60%	3.90x10 <sup>-3</sup>	280	320	7.42x10 <sup>-3</sup>	300	500	9.04x10 <sup>-3</sup>	320	510	1.04x10 <sup>-2</sup>	350	530	1.47x10 <sup>-2</sup>	340	435
70%	3.05x10 <sup>-3</sup>	150	240	4.04x10 <sup>-3</sup>	300	440	6.84x10 <sup>-3</sup>	330	515	1.05x10 <sup>-2</sup>	265	550	9.54x10 <sup>-3</sup>	325	565
80%	4.11x10 <sup>-3</sup>	200	220	9.02x10 <sup>-3</sup>	300	480	1.27x10 <sup>-2</sup>	310	515	1.42x10 <sup>-2</sup>	320	555	1.73x10 <sup>-2</sup>	330	560



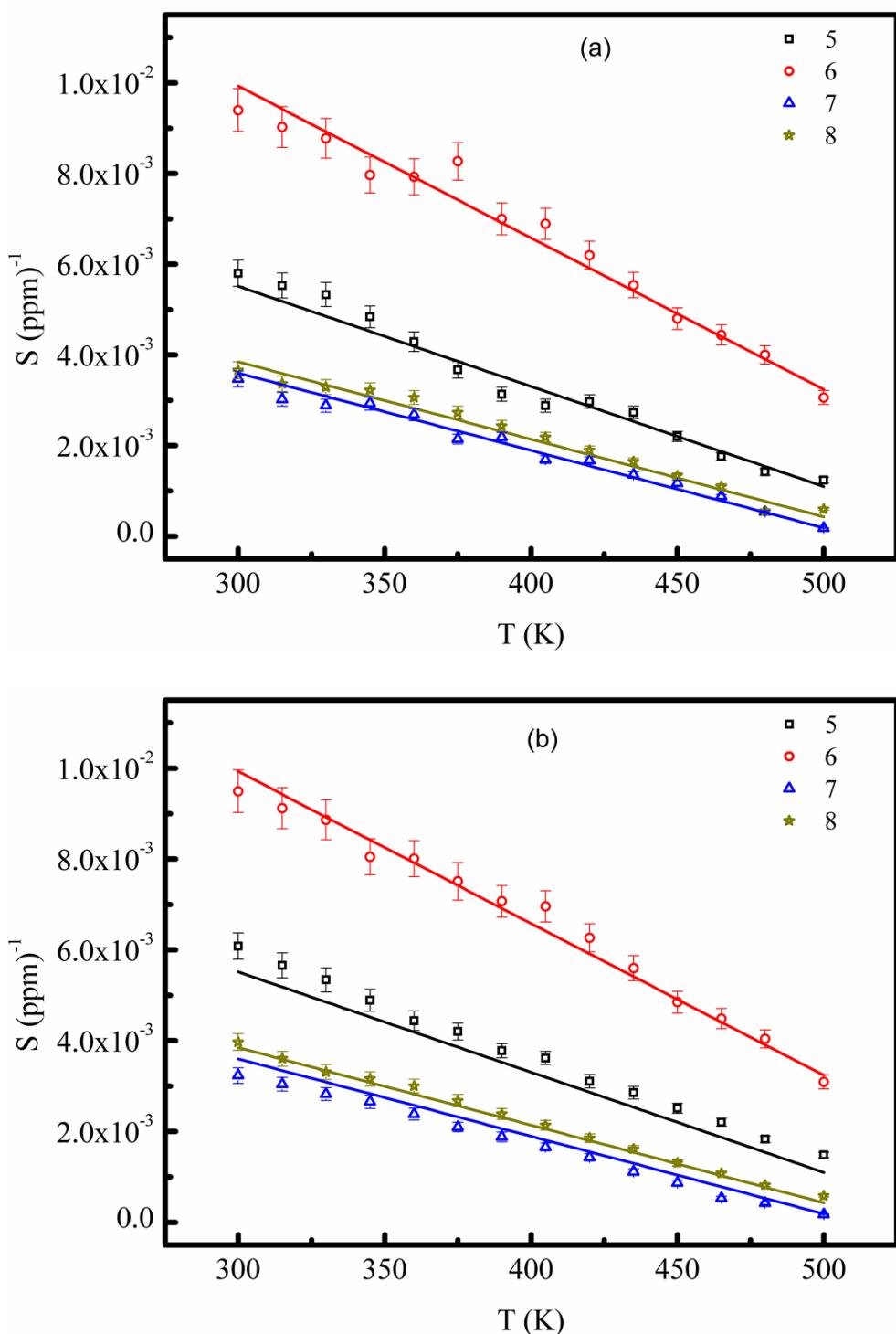
**Fig. S1** Electronic absorption spectra of  $5.0 \times 10^{-6}$  M **5**, **6**, **7** and **8** in DMSO



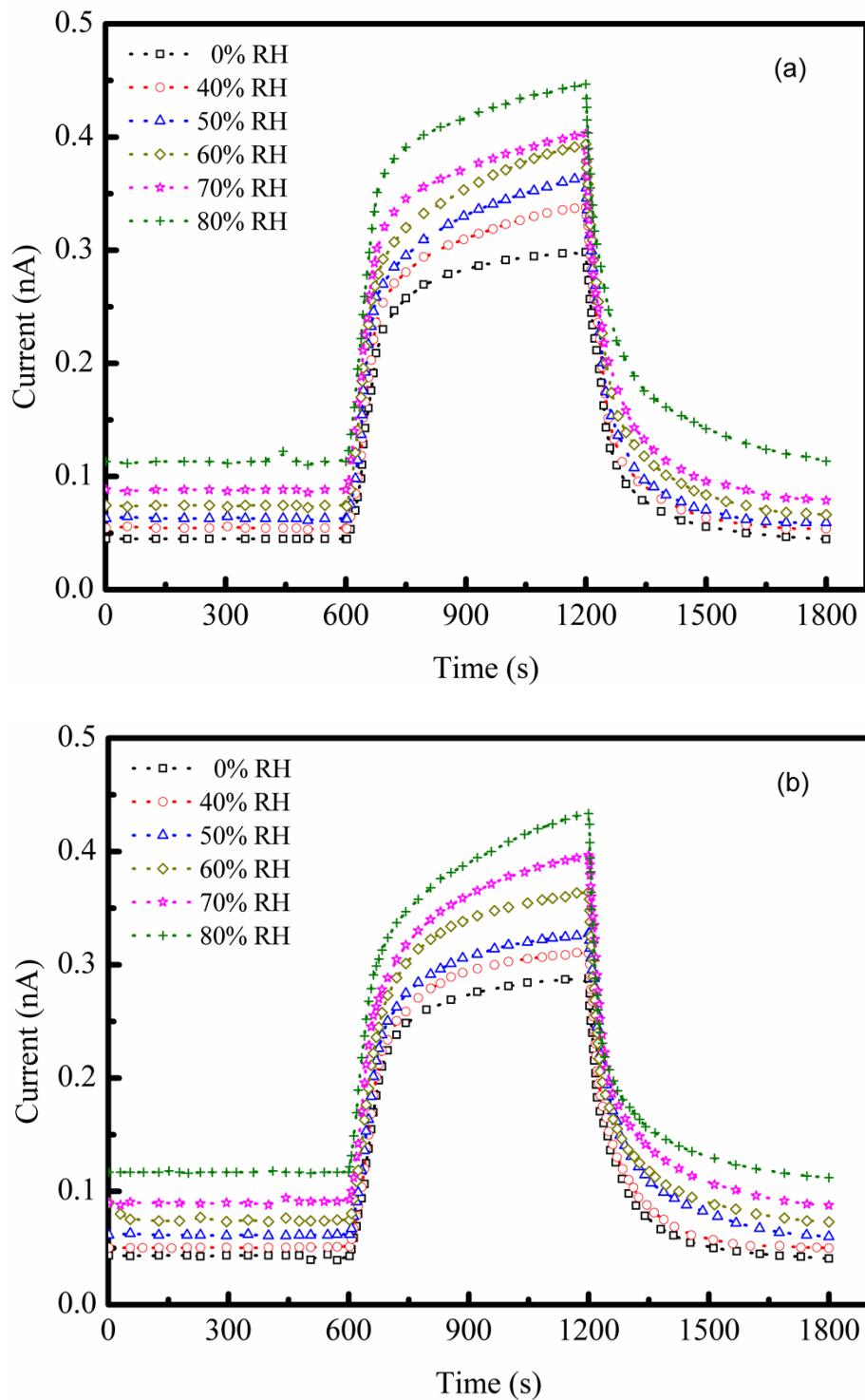
**Fig. S2** Positive ion and reflectron mode MALDI-TOF-MS spectrum of (A) **6**, (B) **7**, (C) **8** and (D) **9** in  $\alpha$ -cyano-4-hydroxycinnamic acid MALDI matrix using nitrogen laser (at 337 nm wavelength) accumulating 50 laser shots.



**Fig. S3** Concentration dependence of the sensitivity of the 60 nm **5-8** coated sensors at 298 K in (a)  $N_2$  and (b) air atmosphere.



**Fig. S4** The effect of the operating temperature on the sensitivity of the sensors for 600 ppm  $CO_2$  (a) in  $N_2$  (b) in air.



**Fig. S5** Variation of the sensor current vs. the time exposition 600 ppm  $\text{CO}_2$  gas and varying RH condition for **6** coated sensor at 298 K in (a)  $\text{N}_2$  (b) air.