

Understanding the relative binding ability of hydroxyfullerene to divalent and trivalent metals

*Jessica Heimann,^a Lauren Morrow,^a Robin E. Anderson^a and Andrew R. Barron^{*a,b,c}*

^aDepartment of Chemistry, Rice University, Houston, Texas 77005, USA

^bDepartment of Materials Science and Nanoengineering, Rice University, Houston, Texas 77005,
USA

^cEnergy Safety Research Institute, College of Engineering, Swansea University, Singleton Park,
Swansea, SA2 8PP, Wales, UK

Electronic Supplementary Information

Table S1. Observed reactivity between metal salts (500 mM) with fullerenol (4.6 mM).

Metal source	Observation
AgNO ₃	red-brown precipitate
Al(NO ₃) ₃	red-brown precipitate
B(OH) ₃	no precipitate
CaCl ₂	red-brown precipitate
Cd(NO ₃) ₂	red-brown precipitate
CoCl ₂	red-brown precipitate
CuCl ₂	red-brown precipitate
KCl	no precipitate
La(NO ₃) ₃	red-brown precipitate
KMnO ₄	dark solution
MnCl ₂	red-brown precipitate
NaOH	no precipitate
Nd(NO ₃) ₃	red-brown precipitate
Ni(NO ₃) ₂	red-brown precipitate
ZnCl ₂	red-brown precipitate

Table S2. Particle size as a function of reaction time as measured by DLS.

[Fe ⁿ⁺] (mM)	[fullerenol] (mM)	Reaction time (min.)	Particle size (μm)
5.0	0.46	12	243-359
0.5	0.046	1	0.25
		1.5	0.80
		2.5	12
0.05	0.0046	14	0.20
		16	2.7

Table S3. Selected bond lengths (Å) and angles (°) for K₃[Ga(catecholate)₃]·1.5 H₂O X-ray crystallographic data and *ab initio* calculations.

	Crystallographic data ^a	Gaussian model data ^b
Ga-O	1.969(2) - 2.005(2)	2.013
O1 - C	1.342(3) - 1.355(3)	1.335
O-Ga-O _{chelate}	83.61(7) - 83.89(7)	82.22 - 82.23
O-Ga-O _{cis}	89.01(8) - 98.09(8)	91.72 - 94.99
O-Ga-O _{trans}	170.41(8) - 170.97(8)	171.08 - 171.09
Ga - O1 - C11	109.9(2) - 111.4(2)	112.37 - 112.38
O6 - C22 - C21	117.4 (2)	116.507

^aB. A Borgias, S. J. Barclay and K. N. Raymond, *J. Coord. Chem.*, 1986, **15**, 109-123.^bB3LYP/3-21G*.

Table S4. Selected calculated bond lengths (Å) and angles (°).

	C-O	O-C-C(O) ^a	O-C-C(H) ^b
Catechol	1.401, 1.382	113.06, 119.24 ^c	126.14, 121.44 ^c
Catecholate	1.297	120.88	123.42

^aCarbon atom attached to the second oxygen. ^bCarbon atom attached to hydrogen. ^cHydrogen bonded OH, see Fig. S9.

Table S5. Selected *ab initio* calculated bond lengths (Å) and angles (°) for [M(catecholate)_n]ⁿ⁻.

	M-O	C-O	O-M-O _{chelate}	O-M-O _{cis} ^a	O-M-O _{trans}	O-C-C ^b	O-C-C ^c
[Zn(catecholate) ₂] ²⁻	1.889	1.343	90.12	119.93	n/a	116.91	124.70
[Cd(catecholate) ₂] ²⁻	2.190	1.343	79.29	126.36	n/a	119.67	122.43
[Al(catecholate) ₃] ³⁻	1.932	1.332	83.45	91.35	172.39	114.79	126.36
[Ga(catecholate) ₃] ³⁻	2.181	1.335	82.23	93.74	171.09	116.51	124.91
[In(catecholate) ₃] ³⁻	2.181	1.337	77.14	91.72	166.38	118.09	123.69

^aFor tetrahedral complexes this is the non-chelate O-M-O angle. ^bEndo the 5-membered cycle.

^cExo the 5-membered cycle.

Table S6. Selected *ab initio* calculated bond lengths (Å) and angles (°) for [M(L₃)_n]²⁻ and [M(L₂)_n]²⁻.^a

	M-O	C-O	O-M-O _{chelate}	O-M-O ^b
[Zn(L ₃) ₂] ²⁻	1.858, 1.888	1.412, 1.427	90.12	105.04, 105.05, 116.66, 129.01
[Cd(L ₃) ₂] ²⁻	2.144, 2.178	1.415, 1.441	79.29	101.86, 101.86, 122.30, 144.25
[Zn(L ₂) ₂] ²⁻	1.871, 1.902	1.410, 1.406	101.59	108.85, 108.85, 113.99, 122.47
[Cd(L ₂) ₂] ²⁻	2.157, 2.197	1.412, 1.415	95.19	108.49, 108.49, 112.90, 136.31

^aL₃ = *cis,cis*-1,3,5-trihydroxycyclohexane, L₂ = *cis*-1,3-dihydroxycyclohexane. ^bNon-chelate angles. ^cNon chelate.

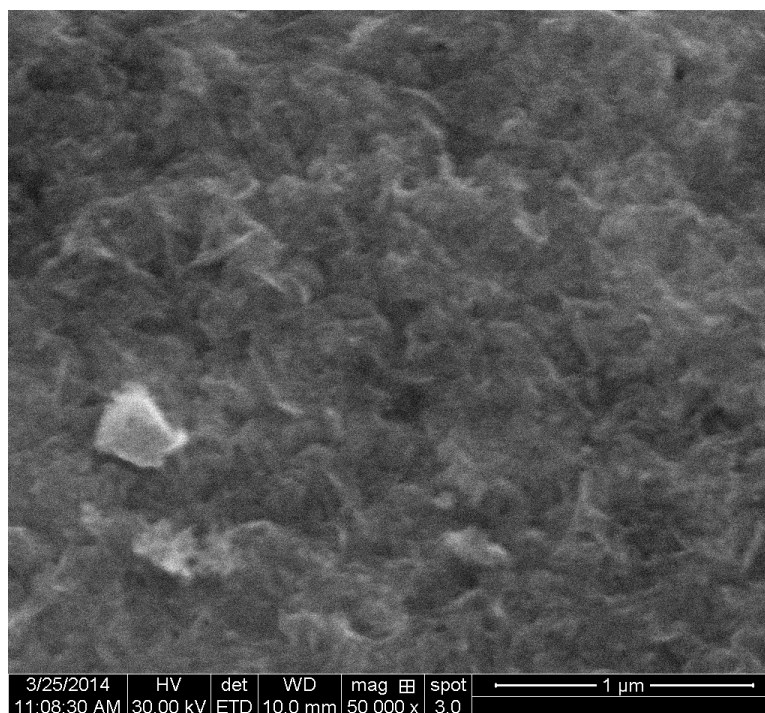


Fig. S1 SEM image of Fe³⁺/Cd²⁺ cross-linked fulleranol.

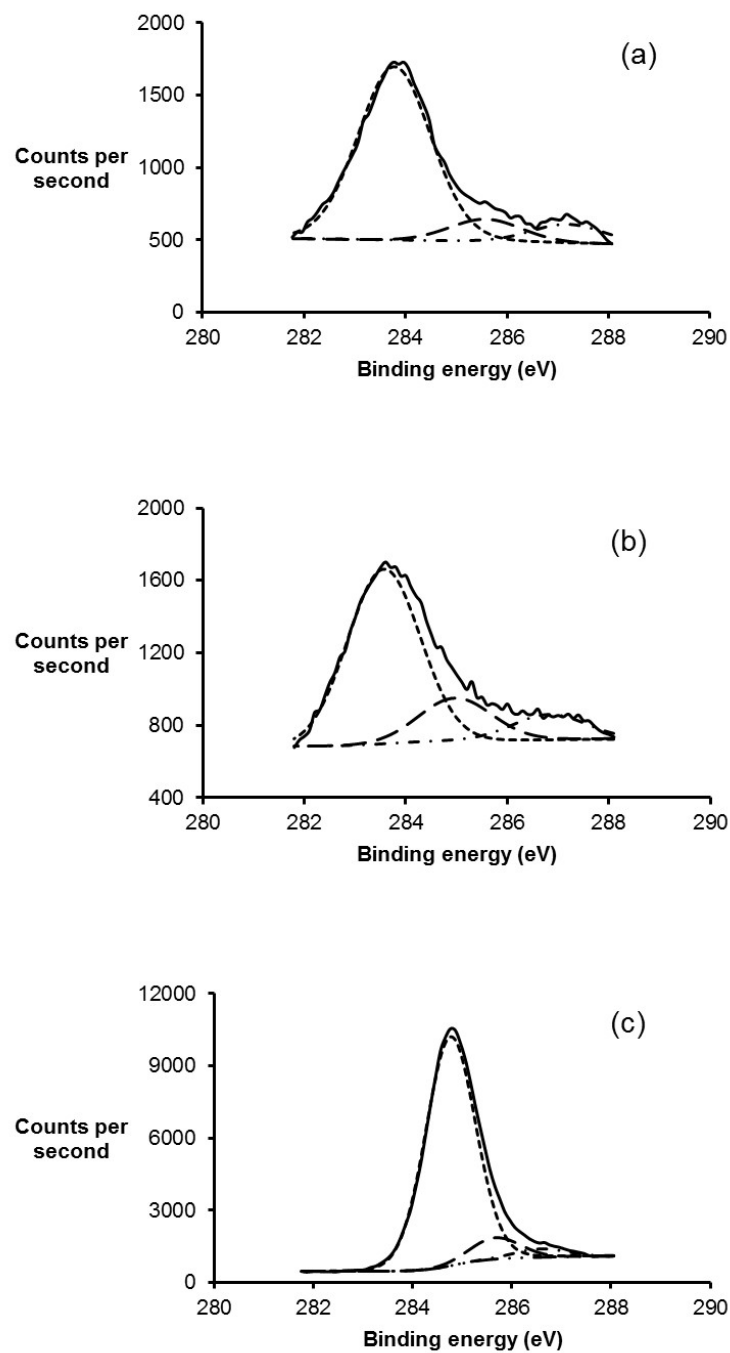


Fig. S2 High resolution C1s XPS data for (a) fullereneol, (b) Fe³⁺ cross-linked fullereneol, and (c) Fe³⁺/La³⁺ cross-linked fullereneol.

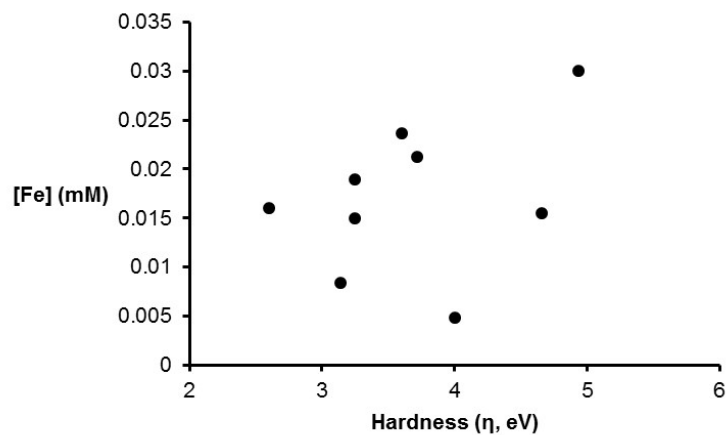


Fig. S3 Plot of $[\text{Fe}^{3+}]$ in competitive binding with M^{n+} versus Pearson's absolute hardness (η) of M^{n+} .

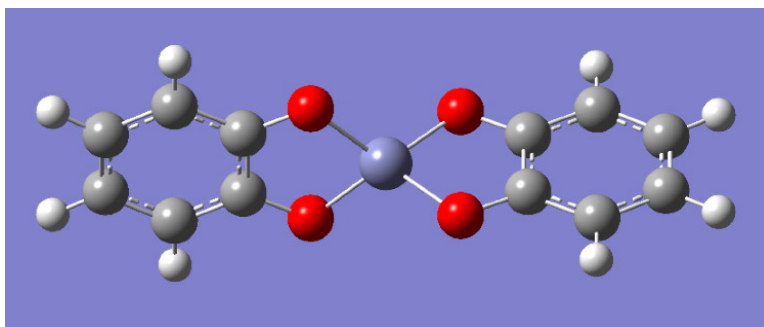


Fig. S4 Calculated structure of $[\text{Zn}(\text{catecholate})_2]^{2-}$.

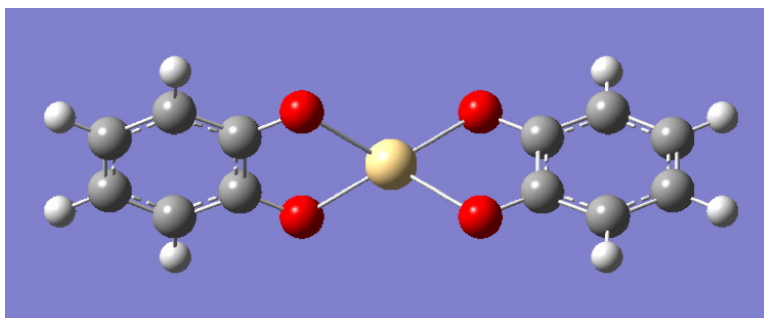


Fig. S5 Calculated structure of $[\text{Cd}(\text{catecholate})_2]^{2-}$.

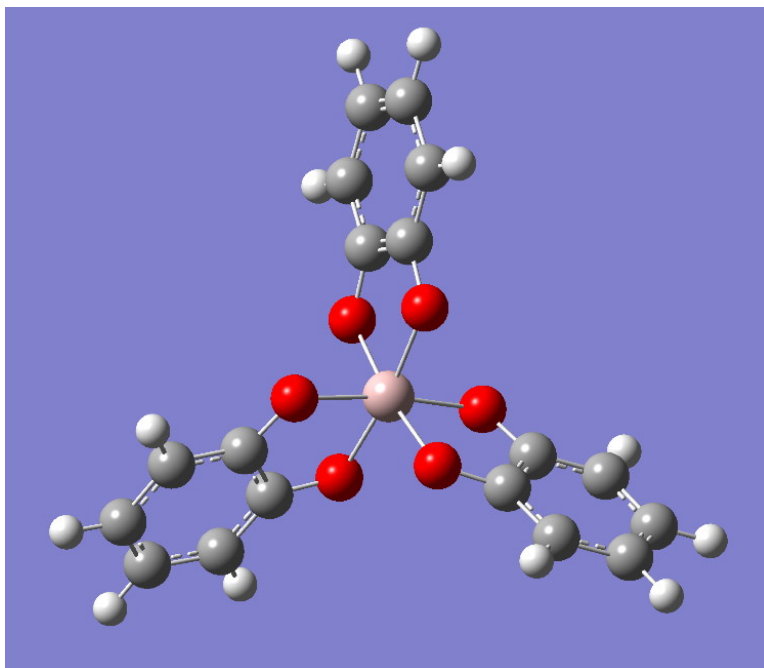


Fig. S6 Calculated structure of $[\text{Al}(\text{catecholate})_3]^{3-}$.

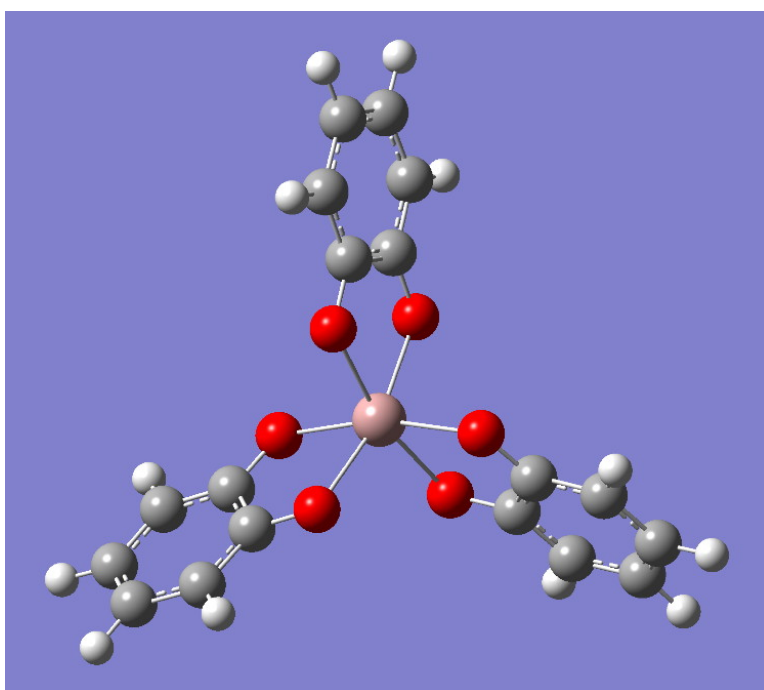


Fig. S7 Calculated structure of $[\text{Ga}(\text{catecholate})_3]^{3-}$.

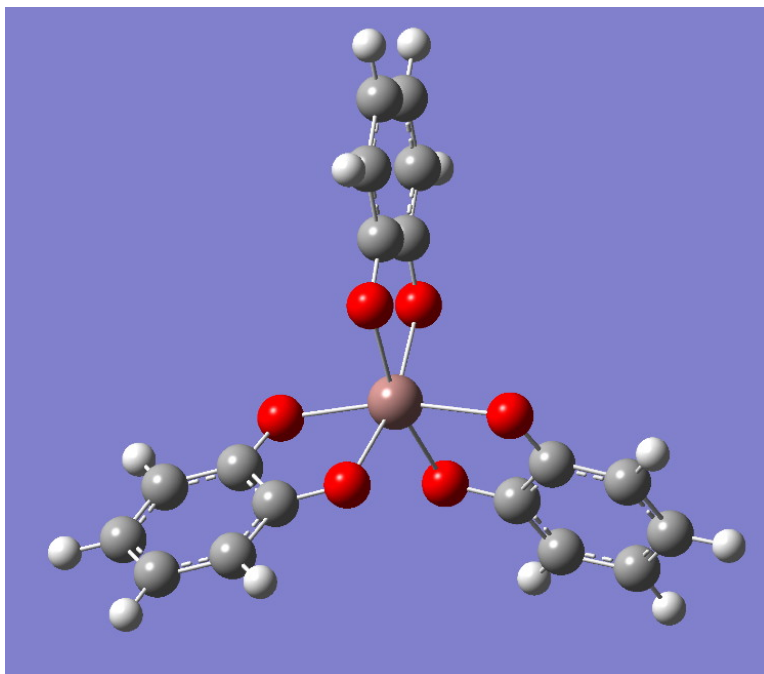


Fig. S8 Calculated structure of [In(catecholate)₃]³⁻.

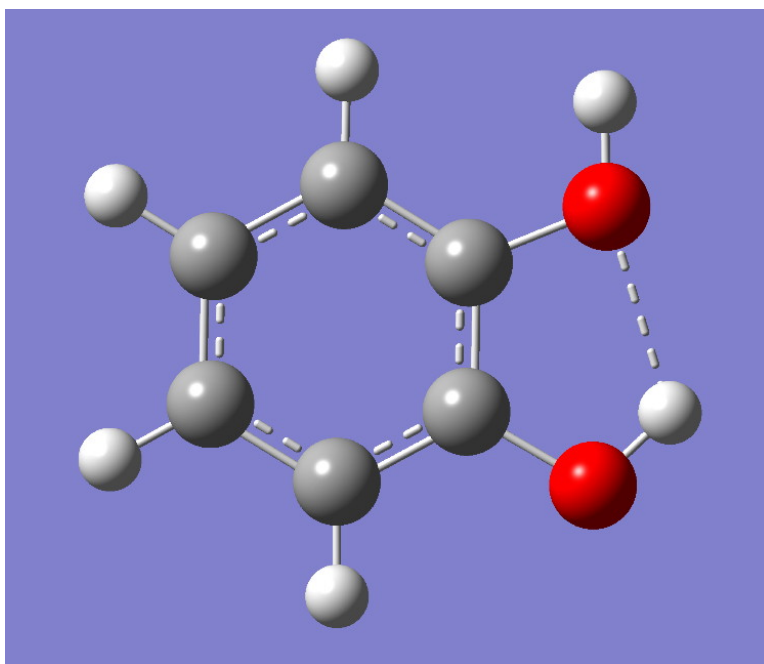


Fig. S9 Calculated structure of catechol.

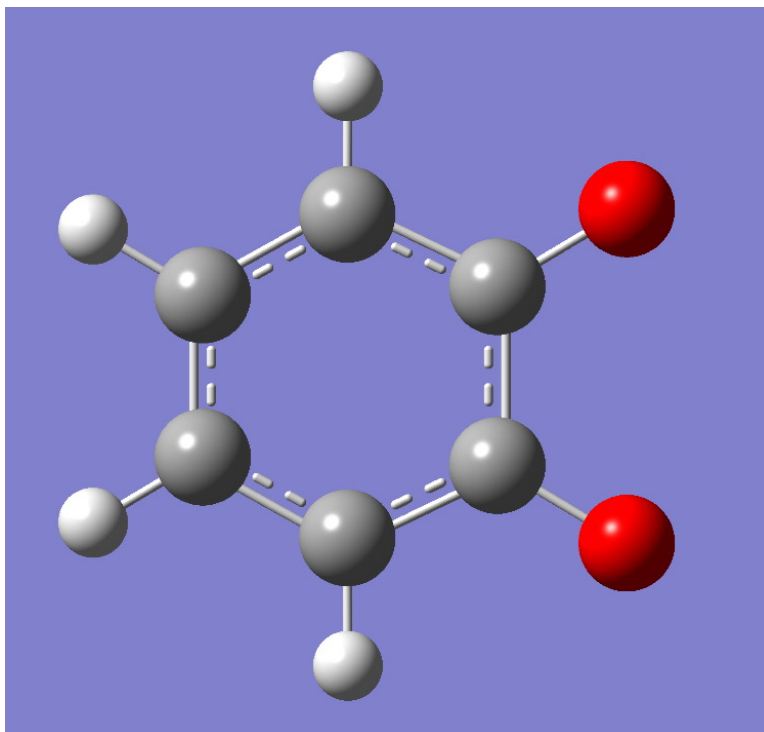


Fig. S10 Calculated structure of [catecholate]⁻.

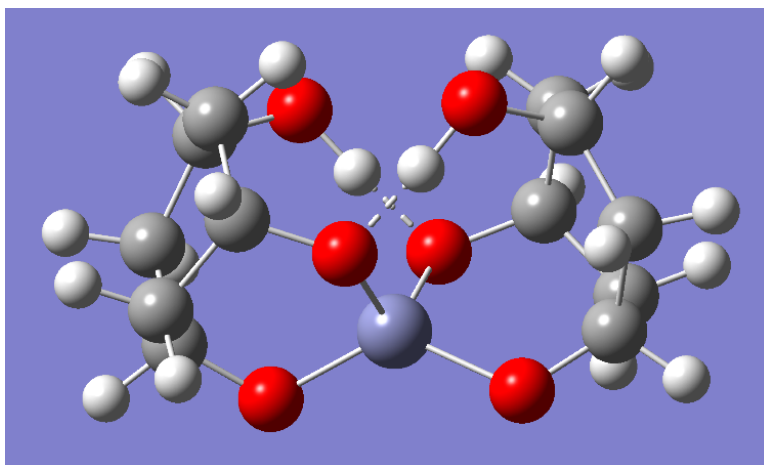


Fig. S11 Calculated structure of [Zn(L₃)₂]²⁻.

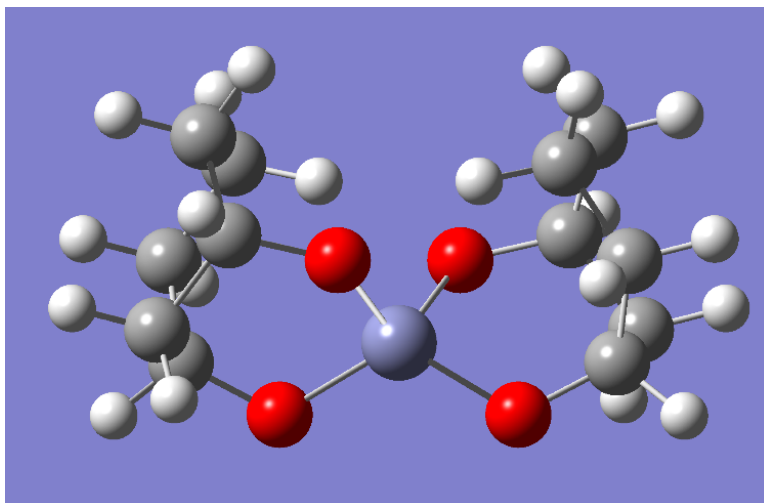


Fig. S12 Calculated structure of $[\text{Zn}(\text{L}_2)_2]^{2-}$.

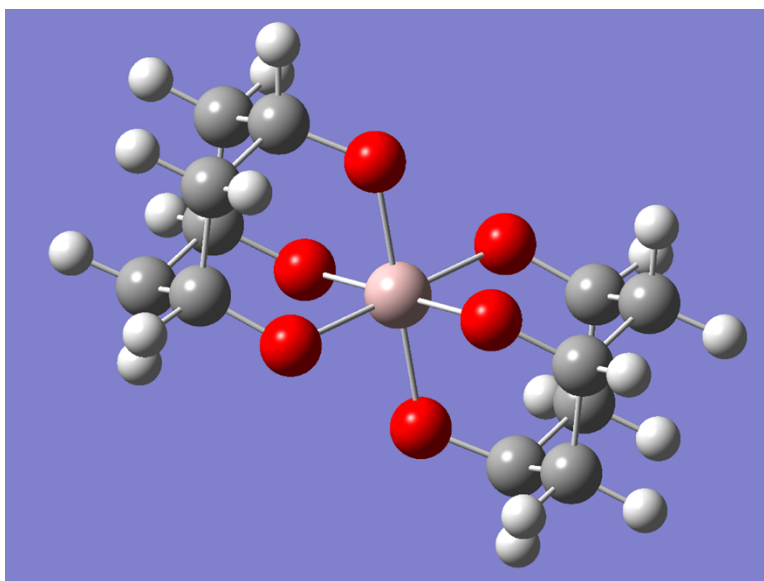


Fig. S13 Calculated structure of $[\text{Al}(\text{L}_3)_2]^{3-}$.

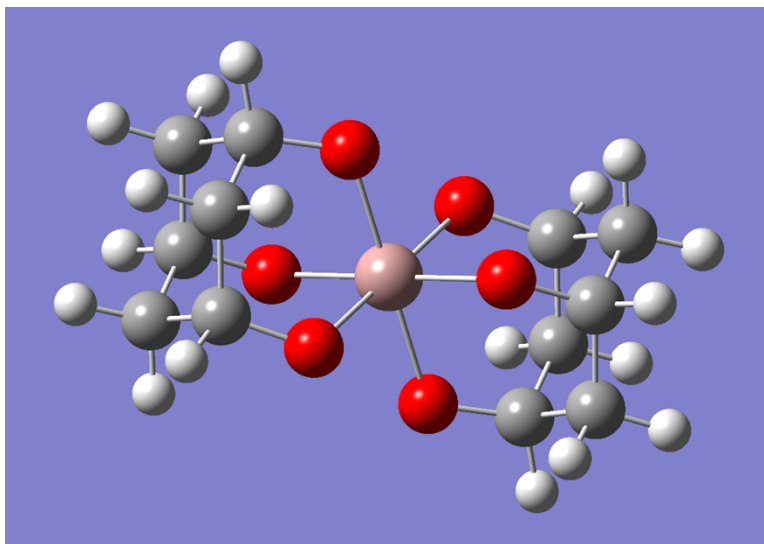


Fig. S14 Calculated structure of $[\text{Ga}(\text{L}_3)_2]^{3-}$.