

Electronic Supplementary Information

Synthesis, characterisation and antibacterialactivity of [(*p*-cym)RuX(L)]⁺²⁺ (X= Cl, H₂O; L = bpmo, bpms) complexes

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Fig. S1 UV-Vis spectra of complexes $[1](\text{ClO}_4)$ – $[4](\text{PF}_6)_2$ recorded in CH_3CN at room temperature.

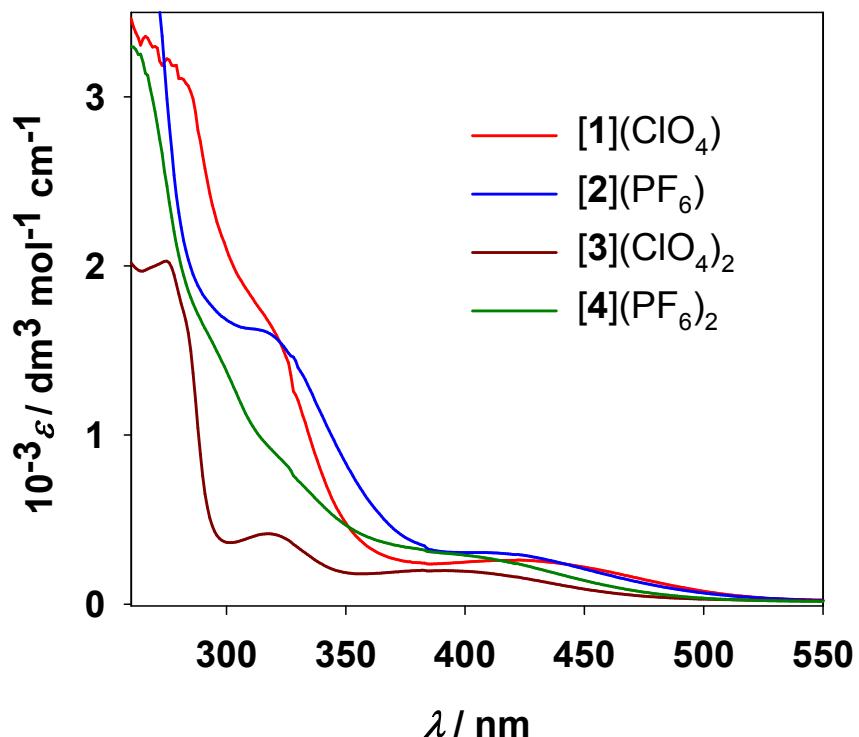


Fig. S2 Cyclic voltammograms of $[1]\text{ClO}_4$ and $[2]\text{PF}_6$ recorded in $\text{CH}_3\text{CN}/0.1\text{mol dm}^{-3}$

Et_4NClO_4 versus $\text{Hg}/\text{Hg}_2\text{Cl}_2$ (scan rate 50 mV s⁻¹).

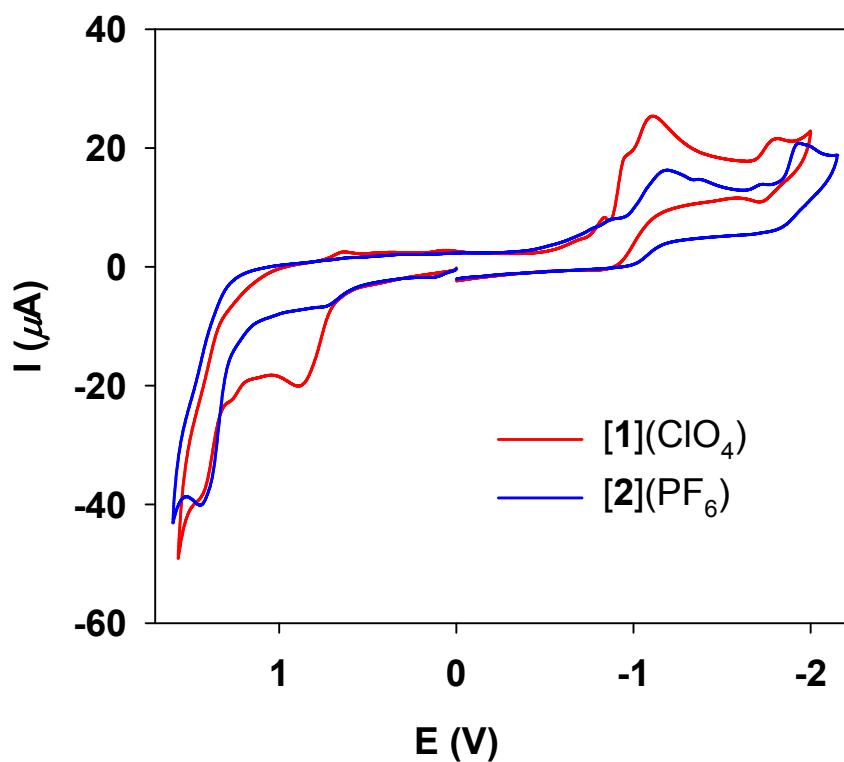


Fig. S3 Emission spectra of the ethidium bromide (EB) bound CT DNA in aqueous buffer

solution in the absence and presence of increasing amount of complexes. $\lambda_{\text{ex}} = 520 \text{ nm}$, $[\text{EB}] = 0.33 \mu\text{M}$, $[\text{DNA}] = 10 \mu\text{M}$, $[\text{complexes}] (\mu\text{M})$ $0 - 120$ for $[\mathbf{1}](\text{ClO}_4)$, $0 - 100$ for $[\mathbf{2}](\text{PF}_6)$, $0 - 80$ for $[\mathbf{3}](\text{ClO}_4)_2$, and $0 - 60$ for $[\mathbf{4}](\text{PF}_6)_2$. $T = 298 \text{ K}$. Inset : Stern-Volmer plot.

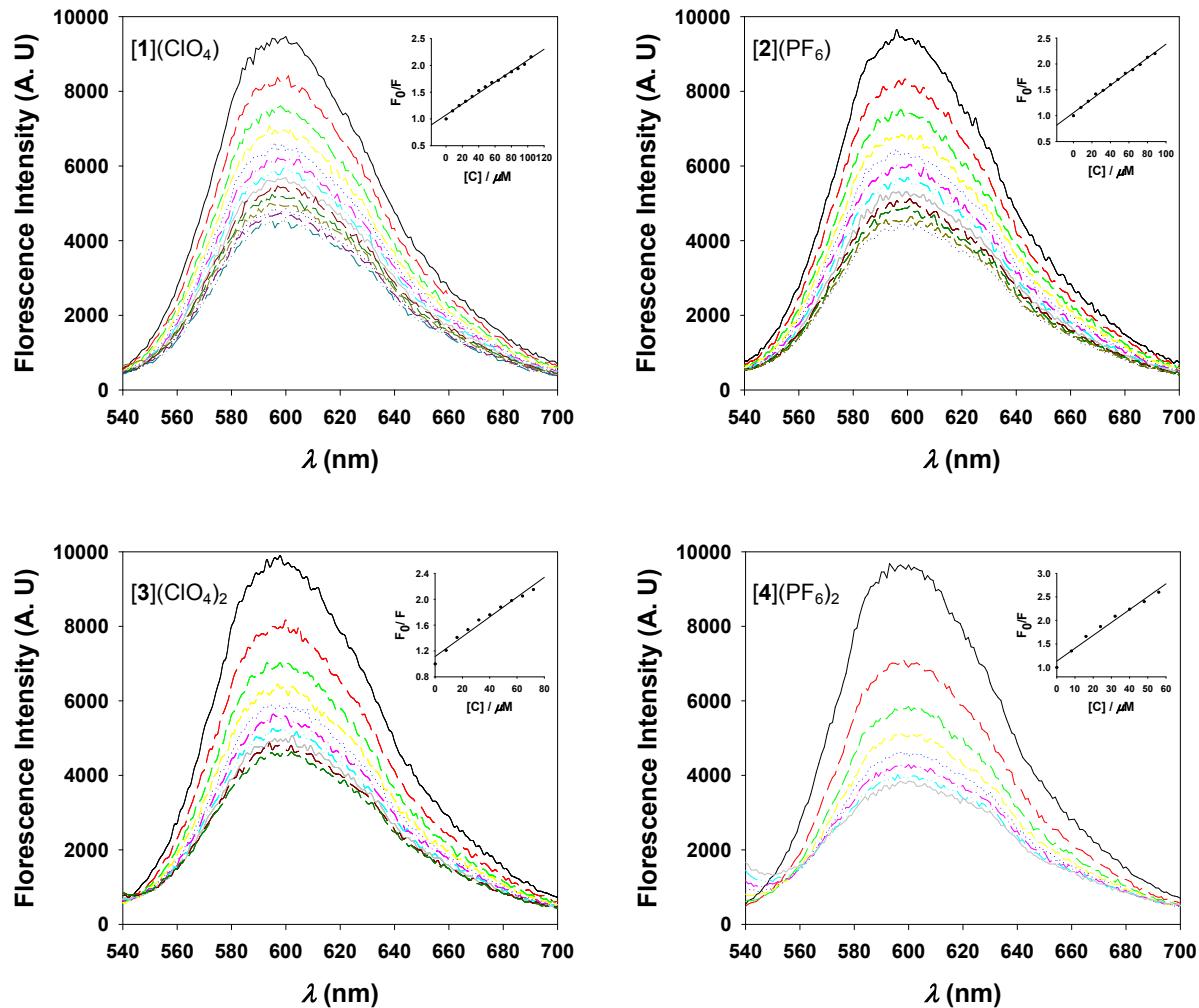


Fig. S4 (a) Effect of increasing amount of complexes on the relative viscosities of CT DNA at 300 K in tris-HCl buffer solution (pH 7.4) Conditions: CT DNA 200 μM , $[\text{complexes}]$

(μM) 0 – 60 for [1](ClO₄), 0 – 80 for [2](PF₆), 0 – 70 for [3](ClO₄)₂, and 0– 60 for [4](PF₆)₂.

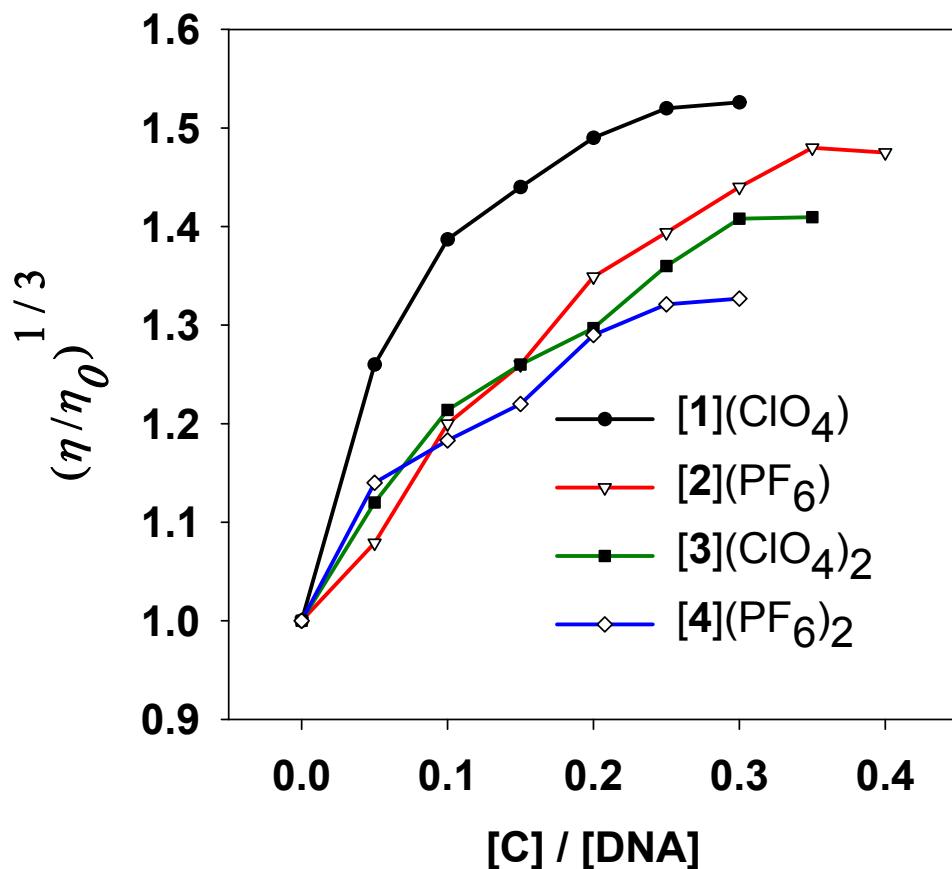


Fig. S5 (a) DIC and light microscopic images (stained with FM-4-64 and DAPI dye) and (b) changes in cellular morphology of control and [3](ClO₄)₂ and [4](PF₆)₂ treated *B. subtilis*

strains.

