Supplementary Information

On the structure of H₂TiO₃—a short discussion on "Lithium recovery from salt lake brine by H₂TiO₃"

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Fig. S1 Simulated XRD patterns of Li₂TiO₃ (left column) and the relative intensity variation of peak (133) and peak (206) by substitution of Li⁺ with H⁺ (right column):
(a) H⁺ substituting Li1⁺ from 0% to 100%; (b) H⁺ substituting Li2⁺ from 0% to 100%; (c) H⁺ substituting Li3⁺ from 0% to 100%.



Fig. S2 The structure of H_2TiO_3 with reasonable OH bond length (a), the small orange, large blue and medium red balls correspond to H, Ti, and O atoms; and the simulated XRD pattern (b). Note that the intensity of peak ($\overline{1}33$) and that of peak ($\overline{2}06$) decrease by 19 % and 7 % compared with that of Li₂TiO₃, respectively.

Atom	x	У	Z	$U_{ m eq}$
H1	0.13958	0.90588	0.03568	0.0147(10)
H2	0	0.26356	0.4434	0.0161(15)
H3	0.10555	0.58444	0.23604	0.0040(8)
Ti1	0	0.41749	0.25	0.00581(6)
Ti2	0	0.74971(6)	0.25	0.00575(7)
01	0.14116(15)	0.26356(12)	0.13720(7)	0.0054(2)
02	0.10555(14)	0.58444(16)	0.13719(7)	0.00549(18)
03	0.13958(15)	0.90588(14)	0.13454(7)	0.0056(2)

Table S1. Atomic coordinates and equivalent isotropic displacement parameters (Å²) for H_2TiO_3 .