## Supporting Information (SI) For Architectural control of urea in supramolecular 1D strontium vanadium oxide chains

#### 1. Instrumentation

Elemental analysis: Euro Vector Euro EA 3000 elemental analyzer.

**Single-crystal X-Ray Diffraction (XRD):** Nonius Kappa CCD single-crystal X-ray diffractometer  $(\lambda(Mo-K_{\alpha}) = 0.71073 \text{ Å})$  equipped with a graphite monochromator.

**UV-Vis spectroscopy:** UV-Vis spectra from 200 to 1000 nm were collected on a Shimadzu UV-2401 PC UV-Vis spectrophotometer in transmission mode using quartz cuvettes with 1.0 cm optical path length. UV-Vis spectra from 200 to 1600 nm were collected on a Shimadzu UV-3600 UV-Vis-NIR spectrophotometer in transmission mode using quartz cuvettes with 1.0 cm optical path length.

**Fourier-transform infrared (FT-IR) spectroscopy:** Unless stated otherwise, the materials were prepared as KBr pellets. FT-IR spectra were collected in transmission mode using a Shimadzu FT-IR-8400S spectrometer. FT-IR spectroscopy of compound **2** was performed on a Shimadzu FT-IRPrestige-21 spectrometer including a Golden Gate ATR-unit. Signals are given in wavenumbers (cm<sup>-1</sup>), intensities are denoted as vs = very strong, s = strong, m = medium, w = weak, b = broad.

**Mass spectrometry:** Electrospray-ionization mass spectrometry (ESI-MS) was performed using *ultra-high resolution* time-of-flight (UHR-TOF) Bruker Daltonik maXis mass spectrometer in negative ion detection mode. Measurement conditions: source voltage: 4 kV, sample flow rate: 500  $\mu$ I/h, drying gas temperature (N<sub>2</sub>): 180 °C. Before each series of measurements the spectrometer was calibrated with Agilent "ESI-TOF low concentration tuning mixture".

**General remarks:** All chemicals were purchased from Sigma Aldrich, ABCR or ACROS and were of reagent grade. The chemicals were used without further purification unless stated otherwise.

#### 2. Synthetic section

## 2.1. Synthesis of 1: ${[Sr(dmf)_3(CON_2H_4)_2][Sr(dmf)_2(CON_2H_4)_2][H_2V_{10}O_{28}]}$

 $(^{n}Bu_{4}N)_{3}[H_{3}V_{10}O_{28}]$  (0.20 g, 0.12 mmol), SrBr<sub>2</sub> · 6 H<sub>2</sub>O (0.09 g, 0.25 mmol) and CON<sub>2</sub>H<sub>4</sub> (0.06 g, 1.07 mmol) were suspended in *N*,*N*-dimethyl formamide (DMF, 6 ml) and the reaction mixture was heated to 70 °C for 2 h. Thereby the color changed from orange to reddish orange and slight turbidity occurred. After the addition of H<sub>2</sub>O (250 µl) the solution turned clear. Diffusion crystallization with acetone as diffusion solvent resulted in yellow crystalline plates. The crude product was washed with acetone and was air-dried.

Yield: 10.2 mg (5.87 µmol, 4.89 % based on V)

**Elemental analysis in wt.-% for Sr<sub>2</sub>V<sub>10</sub>O<sub>37</sub>C<sub>19</sub>N<sub>13</sub>H<sub>53</sub> (calcd.): C 12.83 (13.11), N 10.26 (10.46), H 2.96 (3.07).** 

**Characteristic IR-bands (in cm<sup>-1</sup>):** 3325 (m, b), 3210 (m, b), 2933 (w), 1646 (vs), 1624 (vs), 1566 (s), 1495 (s), 1432 (s), 1250 (m), 981 (s), 940 (vs), 826 (vs).

### 2.2. Synthesis of 2: {[Sr(dmf)<sub>4</sub>]<sub>2</sub>[H<sub>2</sub>V<sub>10</sub>O<sub>28</sub>]}

 $(^{n}Bu_{4}N)_{3}[H_{3}V_{10}O_{28}]$  (0.409 g, 0.24 mmol) and  $SrBr_{2} \cdot 6 H_{2}O$  (0.173 g, 0.49 mmol) were dissolved in DMF (5 ml). After the addition of  $H_{2}O$  (400 µl) the orange solution was heated to 70 °C for 2 h. Diffusion crystallization with acetone as diffusion solvent yielded single orange crystalline blocks. The crude product was washed with acetone and was air-dried.

Yield: 155.1 mg (0.09 mmol, 75 % based on V)

Elemental analysis in wt.-% for  $Sr_2V_{10}O_{36}C_{24}N_8H_{58}$  (calcd.): C 16.83 (16.77), N 6.08 (6.52), H 3.42 (3.40).

**Characteristic IR-bands (in cm<sup>-1</sup>):** 2934 (w), 2353 (m), 1663 (vs), 1385 (m), 1109 (m), 972 (vs), 837 (s), 755 (m).

#### 3. UV-Vis spectroscopy



Figure 1: Room temperature UV-Vis spectra of {V<sub>10</sub>}, 1, 2 recorded in methanolic solution (c =  $2.5 \times 10^{-4}$  M).

#### 4. Mass spectrometry

## 4.1. {[ $Sr(dmf)_3(CON_2H_4)_2$ ][ $Sr(dmf)_2(CON_2H_4)_2$ ][ $V_{10}O_{28}$ ]} (1)

Found m/z	Simulated m/z	Charge	Assignment
280.7903	280.7918	2-	$[V_6O_{16}]^{2}$
413.1878	413.1899	2-	$[V_{8}^{V}V_{023}]^{2}$
454.6567	454.6568	2-	$[V_{8}^{V}V_{2}^{V}O_{25}]^{2}$
462.6532	462.6543	2-	[V <sub>10</sub> O <sub>26</sub> ] <sup>2-</sup>
480.6623	480.6648	2-	$[H_4V_{10}O_{28}]^{2-}$

 Table 1: Cluster units and fragments of compound 1 assigned by analysis of negative-mode ESI-MS spectra.



**Figure 2:** Negative-mode high resolution ESI mass spectrum of **1** in DMF ( $c = 5 \times 10^{-5}$  M).



**Figure 3: Left:** Experimental and simulated ESI mass spectrum of  $[V_6O_{16}]^{2^-}$  observed at m/z = 280.7903. **Right:** Experimental and simulated ESI mass spectrum of  $[V_8^V V^V O_{23}]^{2^-}$  observed at m/z = 413.1878.



**Figure 4: Top Left:** Experimental and simulated ESI mass spectrum of  $[V_8^V V_2^V O_{25}]^{2-}$  observed at m/z = 454.6567. **Top Right:** Experimental and simulated ESI mass spectrum of  $[V_{10}O_{26}]^{2-}$  observed at m/z = 462.6532. **Bottom Left:** Experimental and simulated ESI mass spectrum of the principal building block  $[H_4V_{10}O_{28}]^{2-}$  observed at m/z = 480.6623.

Found m/z	Simulated m/z	Charge	Assignment
280.7917	280.7918	2-	$[V_6O_{16}]^{2}$
330.7603	330.7600	2-	${{H[V_7O_{19}]}^{2-}}$
371.7229	371.7230	2-	$[V_8O_{21}]^{2-}$
380.7277	380.7283	1-	${H[V_4O_{11}]}^{-}$
423.6718	423.6733	2-	[SrV <sub>8</sub> O <sub>22</sub> ] <sup>2-</sup>
462.6529	462.6543	2-	$[V_{10}O_{26}]^{2}$
480.6630	480.6648	2-	${H_4[V_{10}O_{28}]}^{2-}$
514.6045	514.6046	2-	$[SrV_{10}O_{27}]^{2-}$
519.1053	519.1072	4-	${H_2[SrV_{10}O_{28}][SrV_{10}O_{27}]}^{4-}$
531.9197	531.9213	3-	$\{[SrV_{10}O_{28}][SrV_5O_{13}]\}^{3-1}$
540.5796	540.5798	4-	$\{[Sr_2V_{10}O_{28}][SrV_{10}O_{27}]\}^{4-1}$
686.4734	686.4752	3-	${H[SrV_{10}O_{28}][SrV_{10}O_{26}]}^{3-}$

#### 4.2. {[Sr(dmf)<sub>4</sub>]<sub>2</sub>[ $V_{10}O_{28}$ ]} (2)

692.4791	692.4787	3-	${H_3[SrV_{10}O_{28}][SrV_{10}O_{27}]}^{3-}$
721.1093	721.1088	3-	${H[Sr_2V_{10}O_{28}][SrV_{10}O_{27}]}^{3-}$
748.4152	748.4174	2-	$\{[SrV_{10}O_{28}][SrV_4O_{10}]\}^{2}$
793.8828	793.8831	4-	$\{[SrV_{10}O_{26}][Sr_2V_{10}O_{28}][SrV_{10}O_{27}]\}^{4-}$
798.3834	798.3858	4-	${H_2[SrV_{10}O_{27}][Sr_2V_{10}O_{28}][SrV_{10}O_{27}]}^{4-}$
802.8888	802.8884	4-	${H_4[SrV_{10}O_{28}][Sr_2V_{10}O_{28}][SrV_{10}O_{27}]}^{4-}$
807.3891	807.3910	4-	${H_6[V_{10}O_{28}][Sr_2V_{10}O_{28}][Sr_2V_{10}O_{28}]\}^{4-}}$
819.8577	819.8583	4-	$\{[SrV_{10}O_{26}][Sr_2V_{10}O_{28}][Sr_2V_{10}O_{28}]\}^{4-}$
824.3614	824.3609	4-	$\left\{H_{2}[SrV_{10}O_{27}][Sr_{2}V_{10}O_{28}][Sr_{2}V_{10}O_{28}]\right\}^{4^{-}}$
876.9840	876.9840	3-	${H[SrV_4O_{11}][Sr_2V_{10}O_{28}][SrV_{10}O_{27}]}^{3-}$

**Table 2:** Cluster fragments and poly-cluster assemblies of compound **2** assigned by analysis of negative-mode

 ESI-MS spectra.



Figure 5: Negative-mode high resolution ESI mass spectrum of 2 in methanol (c =  $5 \times 10^{-5}$  M).



**Figure 6: Top Left:** Experimental and simulated ESI mass spectrum of  $[V_6O_{16}]^{2^{-}}$  observed at m/z = 280.7917. **Top Right:** Experimental and simulated ESI mass spectrum of  $\{H[V_4O_{11}]\}$  observed at m/z = 380.7277. **Middle Left:** Experimental and simulated ESI mass spectrum of  $[V_{10}O_{26}]^{2^{-}}$  observed at m/z = 462.6529. **Middle Right:** Experimental and simulated ESI mass spectrum of  $\{H_4[V_{10}O_{28}]\}^{2^{-}}$  observed at m/z = 480.6630. **Bottom Left:** Experimental and simulated ESI mass spectrum of  $[SrV_{10}O_{27}]^{2^{-}}$  observed at m/z = 514.6045. **Bottom Right:** Experimental and simulated ESI mass spectrum of  $\{[Sr_2V_{10}O_{28}][SrV_{10}O_{27}]\}^{4^{-}}$  observed at m/z = 540.5796.



**Figure 7:** Experimental and simulated ESI mass spectrum of  $\{H[Sr_2V_{10}O_{28}][SrV_{10}O_{27}]\}^{3-}$  observed at m/z = 721.1093.

# 5. Crystallographic Section

# 5.1. ${[Sr(dmf)_3(CON_2H_4)_2][Sr(dmf)_2(CON_2H_4)_2][H_2V_{10}O_{28}]}(1)$

CCDC code	1027675
Empirical formula	$C_{19}H_{53}N_{13}O_{37}Sr_2V_{10}$
Formula weight	1740.36
Temperature/K	150.15
Crystal system	monoclinic
Space group	Cc
a/Å	10.8044(5)
b/Å	23.2995(14)
c/Å	22.2340(11)
α/°	90
β/°	100.681(4)
$\gamma/^{\circ}$	90
Volume/Å <sup>3</sup>	5500.1(5)
Ζ	4
$\rho_{calc}g/cm^3$	2.099
$\mu/\text{mm}^{-1}$	3.644
F(000)	3432.0
Crystal size/mm <sup>3</sup>	$0.25\times0.2\times0.15$
Radiation	MoKa ( $\lambda = 0.71073$ )
2@ range for data collection/°	12.47 to 53.002
Index ranges	$\text{-13} \leq h \leq \text{13},  \text{-29} \leq k \leq \text{29},  \text{-27} \leq l \leq \text{27}$
Reflections collected	38057
Independent reflections	10852 [ $R_{int} = 0.0227$ , $R_{sigma} = 0.0259$ ]
Data/restraints/parameters	10852/652/741
Goodness-of-fit on F <sup>2</sup>	1.054
Final R indexes [I>=2 $\sigma$ (I)]	$R_1 = 0.0198, wR_2 = 0.0458$
Final R indexes [all data]	$R_1 = 0.0229, wR_2 = 0.0467$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.69/-0.38
Flack parameter	-0.0033(16)



# 5.2. ${[Sr(dmf)_4]_2[H_2V_{10}O_{28}]}$ (2)

CCDC code	1027674
Empirical formula	$C_{24}H_{58}N_8O_{36}Sr2V_{10}$
Formula weight	17191.4
Temperature/K	150
Crystal system	triclinic
Space group	P-1
a/Å	10.9593(3)
b/Å	11.6409(7)
c/Å	12.2782(7)
$\alpha/^{\circ}$	104.697(4)
β/°	99.497(4)
$\gamma/^{\circ}$	109.603(3)
Volume/Å <sup>3</sup>	1372.51(13)
Z	2
$\rho_{calc}g/cm^3$	2.078
$\mu/\text{mm}^{-1}$	3.646
Crystal size/mm <sup>3</sup>	$0.25\times0.2\times0.15$
Radiation	MoKα ( $\lambda$ = 0.71073)
$2\Theta$ range for data collection/°	12.438 to 53
Index ranges	$-13 \le h \le 13, -14 \le k \le 14, -15 \le l \le 15$
Reflections collected	34533
Independent reflections	5609 [ $R_{int} = 0.0232$ , $R_{sigma} = 0.0150$ ]
Data/restraints/parameters	5609/562/398
Goodness-of-fit on F <sup>2</sup>	1.047
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0280, wR_2 = 0.0699$
Final R indexes [all data]	$R_1 = 0.0342, wR_2 = 0.0747$
Largest diff. peak/hole / e Å <sup>-3</sup>	1.31/-1.07

