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Supporting information



Figure S1. Highest occupied NBO of 9-nickelafluorenyl anion



Figure S2. Lowest unoccupied NBO of imidazolium part of 4.



Figure S3. Lowest unoccupied NBO of imidazolium part of 3.

Compound	1	2	3	4	5	7
Empirical formula	C46H56N2NiO2	C44H52N2NiO	C42H46N2NiO	$C_{33}H_{36}N_2Ni$	$C_{38}H_{40}N_2Ni$	C ₆₉ H ₉₃ Cl ₁₇ N ₆ Ni
Formula weight	727.63	683.59	653.52	519.35	583.43	1667.85
Temperature/K	100.00(10)	100(2)	100(2)	100(2)	100.00(10)	100.0(2)
Crystal system	orthorhombic	monoclinic	monoclinic	monoclinic	monoclinic	trigonal
Space group	Pca2 ₁	$P2_1/c$	$P2_1/n$	$P2_1/c$	$P2_1/n$	R3c
a/Å	9.0713(2)	17.1430(5)	13.0464(6)	10.6985(1)	13.4758(2)	15.3884(4)
b/Å	24.610(2)	13.1907(2)	17.1568(8)	15.4023(1)	13.6096(2)	15.3884(4)
<i>c</i> /Å	16.8975(5)	17.5420(4)	15.5211(6)	16.7954(1)	17.1452(3)	61.781(2)
a/°	90	90.00	90.00	90.00	90	90
β /°	90	114.528(3)	101.257(2)	96.035(1)	105.054(2)	90
γ/°	90	90.00	90.00	90.00	90	120
Volume/Å ³	3772.2(3)	3608.8(1)	3407.3(3)	2752.23(4)	3036.52(9)	12669.9(8)
Z	4	4	4	4	4	6
$\rho_{\rm calc}{ m mg}/{ m mm^3}$	1.281	1.258	1.274	1.253	1.276	1.312
µ/mm ⁻¹	0.555	0.574	0.605	0.728	0.668	0.808
<i>F</i> (000)	1560.0	1464.0	1392.0	1104.0	1240.0	5196.0
Crystal size/mm ³	$0.412 \times 0.131 \times 0.043$	$0.519 \times 0.182 \times 0.076$	$0.35 \times 0.15 \times 0.1$	$0.57 \times 0.41 \times 0.23$	$0.63 \times 0.42 \times 0.29$	0.25 imes 0.25 imes 0.05
Radiation	MoK α ($\lambda = 0.71073$)	Mo $K\alpha$ ($\lambda = 0.71073$)	$MoK\alpha (\lambda = 0.71073)$	MoKa ($\lambda = 0.71073$)	MoKa ($\lambda = 0.71073$)	MoKa ($\lambda = 0.71073$)
2 Θ range for data collection	5.85 to 59.026°	5.96 to 62.68°	5.12 to 46.5°	6.44 to 69.9°	6.474 to 66.864°	5.652 to 53.462°
Index ranges	$-10 \le h \le 12, -31 \le k \le 33, -23 \le l \le 21$	$-23 \le h \le 24, -19 \le k \le 18, -25 \le l \le 21$	$-14 \le h \le 14, -19 \le k \le 19, -17 \le l \le 17$	$-16 \le h \le 17, -24 \le k \le 24, -26 \le l \le 23$	$-20 \le h \le 12, -20 \le k \le 15, -26 \le l \le 26$	$-19 \le h \le 16, -19 \le k \le 19, -78 \le l \le 78$
Reflections collected	14432	41020	22387	43738	24629	51307
Independent reflections	8281 [$R_{int} = 0.0679, R_{sigma} = 0.1078$]	10889 [$R_{int} = 0.0359, R_{sigma} = 0.0849$]	4859 [$R_{int} = 0.094$, $R_{sigma} = 0.1326$]	11418 [$R_{int} = 0.0201, R_{sigma}$ = 0.0493]	10953 [$R_{int} = 0.0256, R_{sigma} = 0.0338$]	5990 [$R_{int} = 0.0777, R_{sigma} = 0.0309$]
Data/restraints/parameters	8281/7/466	10889/0/437	4859/0/421	11418/0/329	10953/0/376	5990/1/297
Goodness-of-fit on F ²	1.025	1.041	1.080	1.061	1.021	1.067
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0756, wR_2 = 0.1611$	$R_1 = 0.0371, wR_2 = 0.0831$	$R_1 = 0.0730, wR_2 = 0.1240$	$R_1 = 0.0322, wR_2 = 0.0820$	$R_1 = 0.0340, wR_2 = 0.0860$	$R_1 = 0.0618, wR_2 = 0.1636$
Final R indexes [all data]	$R_1 = 0.1023, wR_2 = 0.1803$	$R_1 = 0.0497, wR_2 = 0.0890$	$R_1 = 0.1220, wR_2 = 0.1400$	$R_1 = 0.0413, wR_2 = 0.0877$	$R_1 = 0.0422, wR_2 = 0.0911$	$R_1 = 0.0691, wR_2 = 0.1693$
Largest diff. peak/hole / e Å ⁻³	1.41/-0.84	0.46/-0.41	0.40/-0.37	0.51/-0.40	0.47/-0.32	0.65/-0.45

Table S1 Crystal data and structure refinement.

Compound	Number of imaginary frequencies	Energy	Zero-point correction	Sum of electronic and thermal free energies	Frontier orbital energy	Solvent	Temperature
9-nickelafluorenyl anion	0	-2164.00477952	0.240784	-2163.806439	HO NBO -0.10342	THF	298.15 K
<u>1</u> +	0	-925.799700206	0.433427	-925.421814	LU NBO -0.19496	THF	298.15 K
<u>2</u> +	0	-1004.40634788	0.494026	-1003.969709	LU NBO -0.19958	THF	298.15 K
<u>3</u> +	0	-924.596950906	0.409945	-924.242930	LU NBO -0.11345	THF	298.15 K
<u>4</u> +	0	-732.863787531	0.360553	-732.553397	LU NBO -0.11288	THF	298.15 K
<u>5</u>	0	-3089.865128	0.676751	-3089.260086	-	THF	298.15 K
9-nickelafluorenyl anion	0	-2163.98047085	0.240496	-2163.798230	-	1,4-dioxane	374.2 K
<u>1</u> +	0	-925.779389352	0.433495	-925.423723	_	1,4-dioxane	374.2 K
<u>5</u>	0	-3089.86060321	0.677100	-3089.285612	-	1,4-dioxane	374.2 K

 Table S2: Exact energies (Hartree/particle), zero-point corrections (Hartree/particle), number of imaginary frequencies, sum of electronic and thermal

 Free Energies (Hartree/particle) and frontier orbital energies (Hartree).