

Submitted to *Energy & Environmental Science*

October 26, 2010

Electronic Supporting Information

To be published in the themed issue on **Carbon Nanostructures for Energy**
Guest Edited by Professors Dirk Guldi, Andreas Hirsch and Nazario Martin

Photoinduced Processes of the Supramolecularly Functionalized Semi-Conductive SWCNTs with Porphyrins via Ion Pair Interactions

Sushanta K. Das,^a Navaneetha K. Subbaiyan,^a Francis D'Souza^{a,*} Atula S. D. Sandanayaka,^{b,c} Taku Hasobe,^{b,*} and Osamu Ito^{d,*}

^a*Department of Chemistry, Wichita State University, 1845 Fairmount, Wichita, KS 67260-0051, USA, FAX: 316-978-3431, E-mail: Francis.DSouza@wichita.edu)*

^b *School of Materials Science, Japan Advanced Institute of Science and Technology (JAIST), Nomi, Ishikawa, 923-1292 Japan.*

^{c^b} *Department of Chemistry, Faculty of Science and Technology, Keio University, Yokohama 223-8522 and PRESTO, Japan Science and Technology Agency (JST), Saitama, 332-0012, Japan; E-mail: hasobe@chem.keio.ac.jp*

^d *Fullerene Group, NIMS, Namiki, Tsukuba, and CarbonPhotoScience Lab., Kita-Nakayama, Izumi-ku, Sendai, 981-3215, and NIMS, Namiki, Tsukuba, Japan. E-mail: ito@tagen.tohoku.ac.jp*

Received

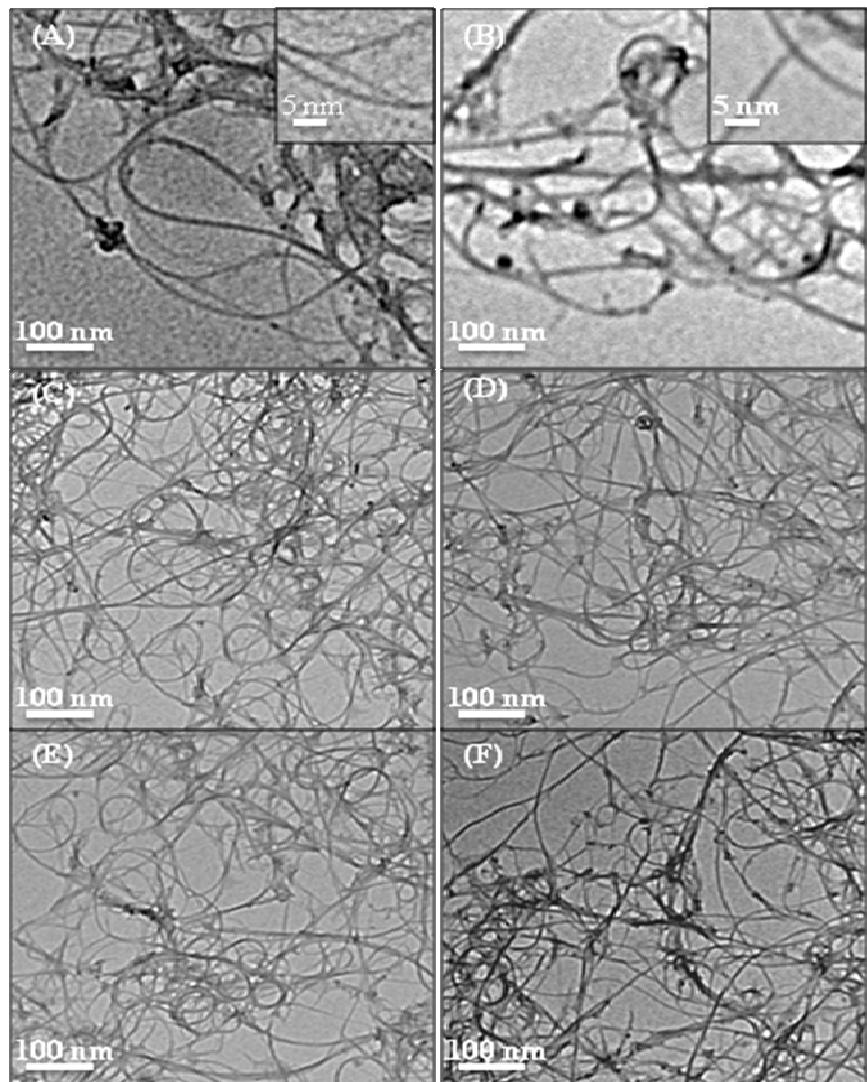


Fig. S1. TEM images for dried samples prepared from homogenous DMF solution. (A) SWCNT(6,5)/PyrCOO[□], (B) SWCNT(7,6)/PyrCOO[□], (C) SWCNT(6,5)/PyrCOO[□]/(TMPyP⁺)H₂, (D) SWCNT(7,6)/PyrCOO[□]/(TMPyP⁺)H₂, (E) SWCNT(6,5)/PyrCOO[□]/(TMPyP⁺)Zn, and (F) SWCNT(7,6)/ PyrCOO[□]/(TMPyP⁺)Zn.

Estimation of Molecular Stoichiometry of the Nanohybrids

Estimation of ratio of Pyr-R to SWCNT

1) MW of Pyr-R \approx 400

mol of 4 mg of Pyr-R \approx $0.004/400 \approx 1 \times 10^{-5}$ mol:

2) MW of SWCNT(ϕ 1 nm \times 800 nm) \approx 1×10^6 SWCNT

mol of 2 mg of SWCNT \approx $0.002/1 \times 10^6 \approx 2 \times 10^{-9}$ mol.

3) Pyr-R/SWCNT \approx $1 \times 10^{-5}/2 \times 10^{-9} = 5000$ mols per one mol of 800 nm SWCNT.

4) length of Pyr-R \approx 0.5 nm, length of SWCNT \approx 800 nm,

$5000 \times 0.5 = 2500$ nm/800 nm = ca. 3 (Pyr-R every 3 molecules on 1 nm of SWCNT).

Estimation of ratio of MP to SWCNT

5) Porphyrins; Abs. = 1 (1 cm cell, from Fig. 2), molar extinction coefficient = $10^5 \text{ M}^{-1} \text{ cm}^{-1}$. Concentration = $1/10^5 = 1 \times 10^{-5}$ mol/L;

6) In 15 mL, $1 \times 10^{-5} \times 1.5 \times 10^{-3}$ mol = 15×10^{-9} mol

7) 15 molecules on SWCNT; one MP every 75 nm of SWCNT.

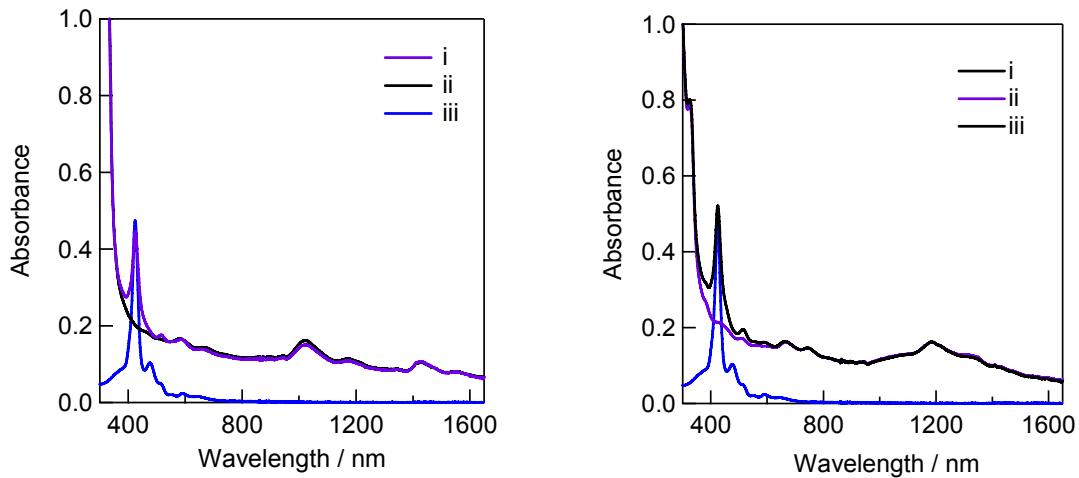


Fig. S2-1. Absorption spectra in DMF: **(Left)** (i) SWCNT(6,5)/PyrCOO⁻/(TMPyP⁺)H₂, (ii) SWCNT(6,5)/PyrCOO⁻, and (iii) (TMPyP⁺)H₂, and **(Right)** (i) SWCNT(7,6)/PyrCOO⁻/(TMPyP⁺)H₂, (ii) SWCNT(7,6)/PyrCOO⁻, and (iii) (TMPyP⁺)H₂.

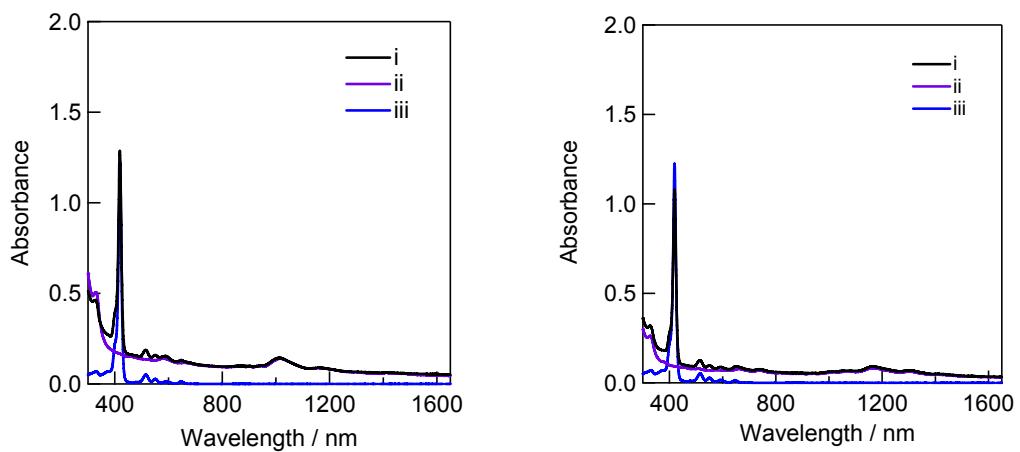


Fig. S2-2. Absorption spectra in DMF: **(Left)** (i) SWCNT(6,5)/PyrNH₃⁺/(TPPS⁻)H₂, (ii) SWCNT(6,5)/PyrNH₃⁺, and (iii) (TPPS⁻)H₂. **(Right)** SWCNT(7,6)/PyrNH₃⁺/(TPPS⁻)H₂, (ii) SWCNT(7,6)/PyrNH₃⁺, and (iii) PyrNH₃⁺.

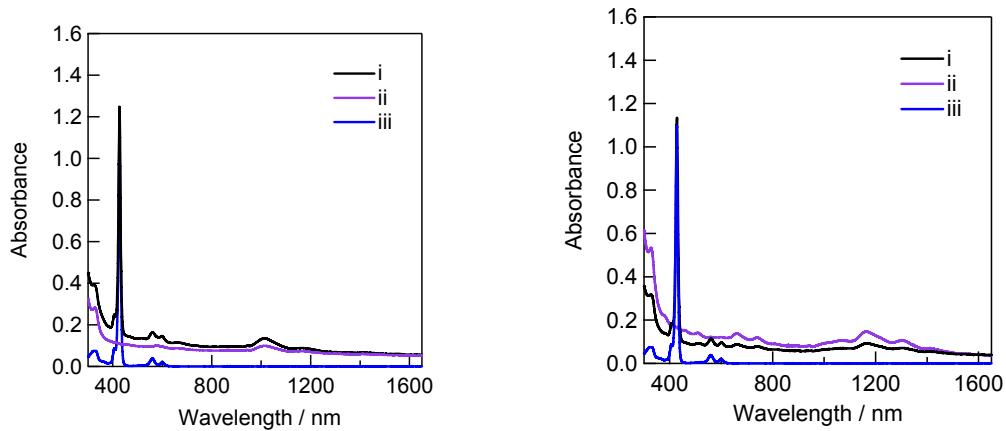


Fig. S2-3. Absorption spectra in DMF: (**Left**) (i) SWCNT(6,5)/PyrNH₃⁺(TPPS⁻)Zn, (ii) SWCNT(6,5)/PyrNH₃⁺, and (iii) (TPPS⁻)Zn. (**Right**) SWCNT(7,6)/PyrNH₃⁺(TPPS⁻)Zn, (ii) SWCNT(7,6)/PyrNH₃⁺, and (iii) PyrNH₃⁺.

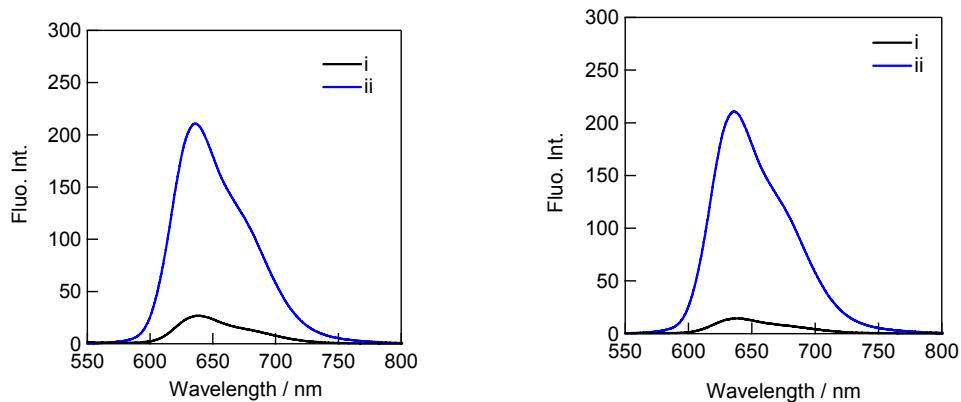


Fig. S3-1. Steady-state fluorescence spectra in DMF($\lambda_{\text{ex}} = 425$ nm). **Left** (i) SWCNT(6,5)/PyrCOO[□](TMPyP⁺)Zn and (ii) (TMPyP⁺)Zn. **Right** (i) SWCNT(7,6)/PyrCOO[□](TMPyP⁺)Zn and (ii) (TMPyP⁺)Zn.

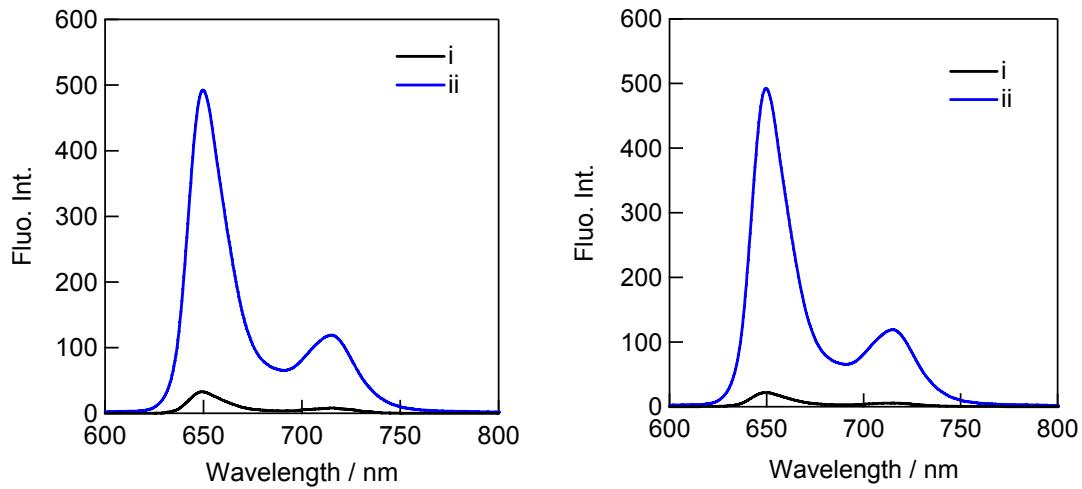


Fig. S3-2. Steady-state fluorescence spectra in DMF ($\lambda_{\text{ex}} = 428$ nm). **(Left)** SWCNT(6,5)/PyrNH₃⁺/(TPPS⁻)H₂ and (ii) (TPPS⁻)H₂. (TMPPyP⁺)Zn (i) SWCNT(7,6)/PyrNH₃⁺/(TPPS⁻)H₂ and (ii) (TPPS⁻)H₂.

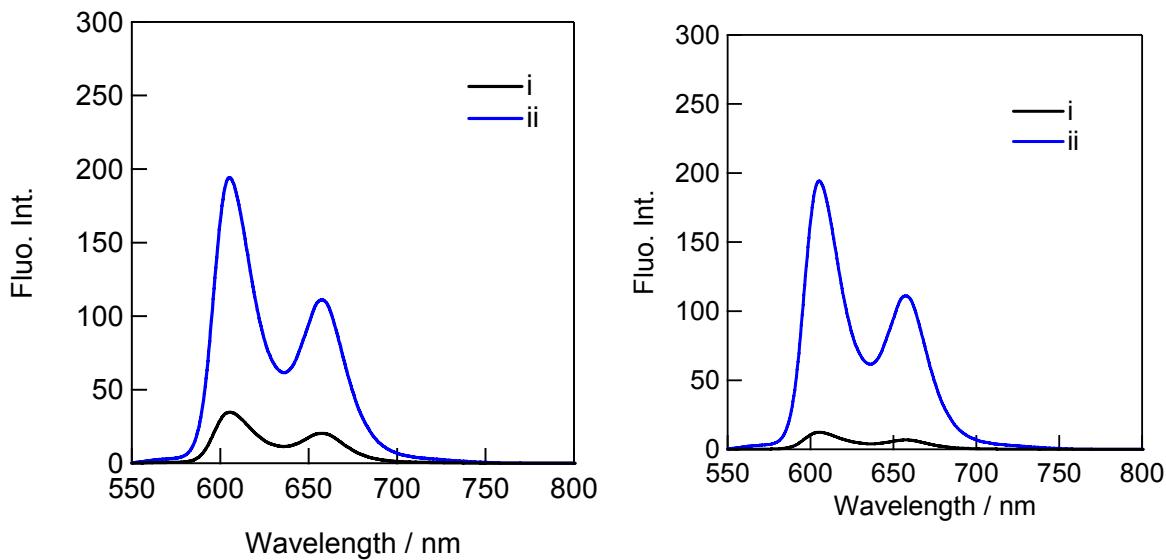


Fig. S3-3. Steady-state fluorescence spectra in DMF ($\lambda_{\text{ex}} = 428$ nm). **(Left)** SWCNT(6,5)/PyrNH₃⁺/(TPPS⁻)Zn and (ii) (TPPS⁻)Zn. (TMPyP⁺)Zn (i) SWCNT(7,6)/PyrNH₃⁺/(TPPS⁻)Zn and (ii) (TPPS⁻)Zn.

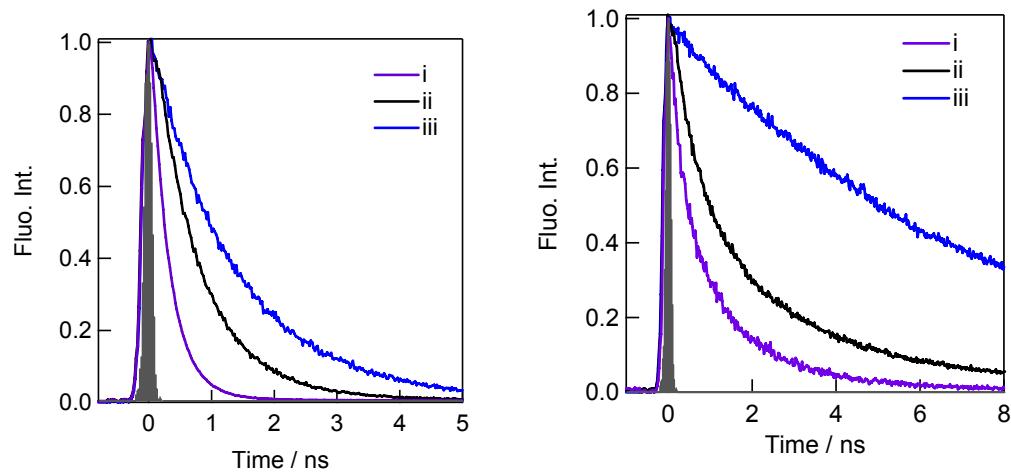


Fig. S4-1 Fluorescence decays in DMF. $\lambda_{\text{ex}} = 408$ nm. **(Left)** (i) SWCNT(7,6)/PyrNH₃⁺/(TPPS⁻)Zn, (ii) SWCNT(6,5)/PyrNH₃⁺/(TPPS⁻)Zn and (iii) (TPPS⁻)Zn. **(Right)** (i) SWCNT(7,6)/PyrNH₃⁺/(TPPS⁻)H₂, (ii) SWCNT(6,5)/PyrNH₃⁺/(TPPS⁻)H₂, and (iii) (TPPS⁻)H₂.

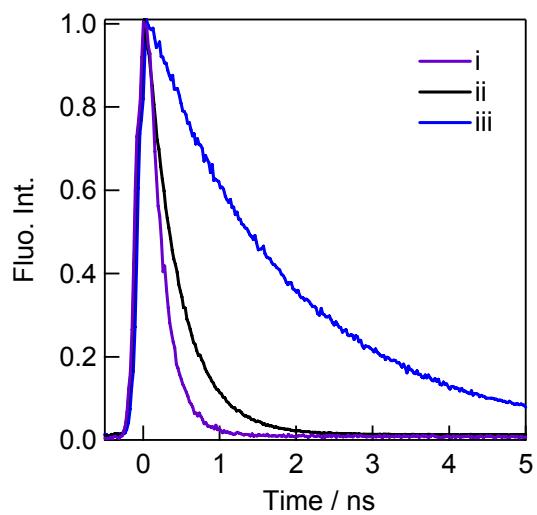


Fig. S4-2. Fluorescence decays of (i) SWCNT(7,6)/PyrCOO[□]/(TPMPyP⁺)Zn (ii) SWCNT(6,5)/PyrCOO[□]/(TPMPyP⁺)Zn and (iii) (TPMPyP⁺)Zn in DMF. $\lambda_{\text{ex}} = 408 \text{ nm}$.

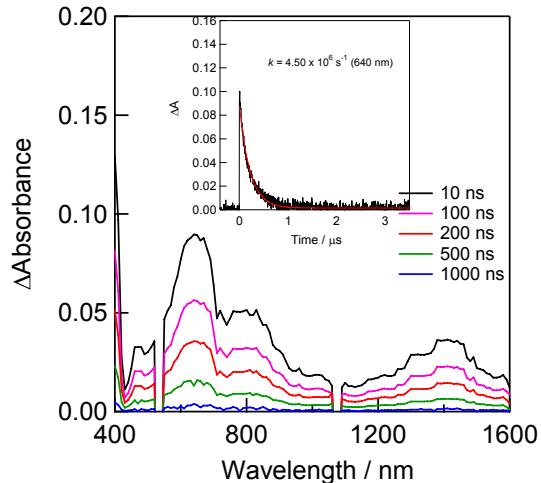


Fig. S5-1. Nanosecond transient absorption spectra of SWCNT(6,5)/PyrNH₃⁺(TPPS⁻)Zn observed by 532 nm (ca. 3 mJ/ pulse) laser irradiation in DMF. Inset: Absorption-time profile.

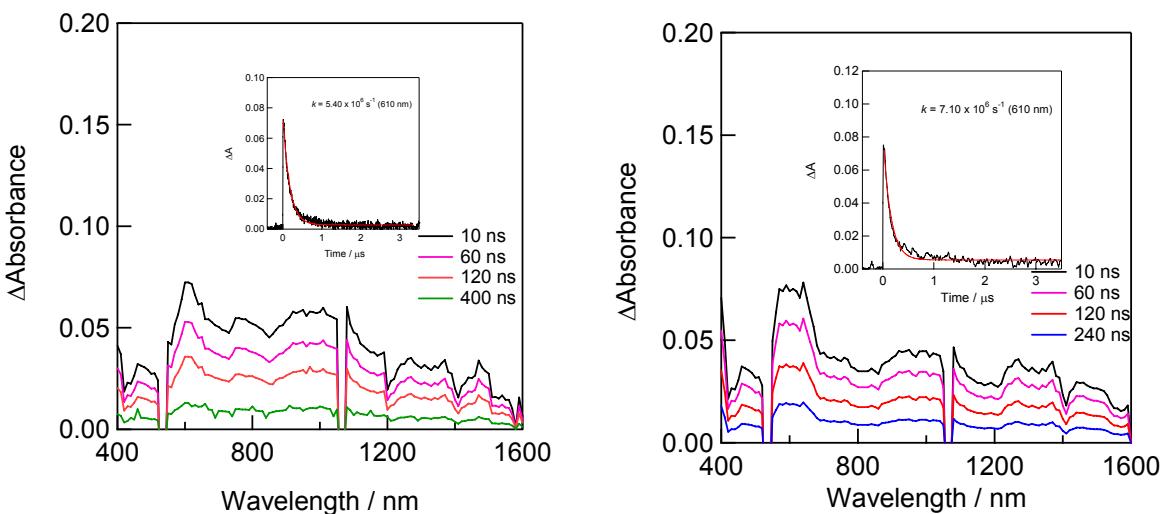


Fig. S5-2. Nanosecond transient absorption spectra of SWCNT(6,5)/PyrNH₃⁺(TPPS⁻)H₂, SWCNT(7,6)/PyrNH₃⁺(TPPS⁻)H₂ observed by 532 nm (ca. 3 mJ/ pulse) laser irradiation in DMF. Inset: Absorption-time profile.

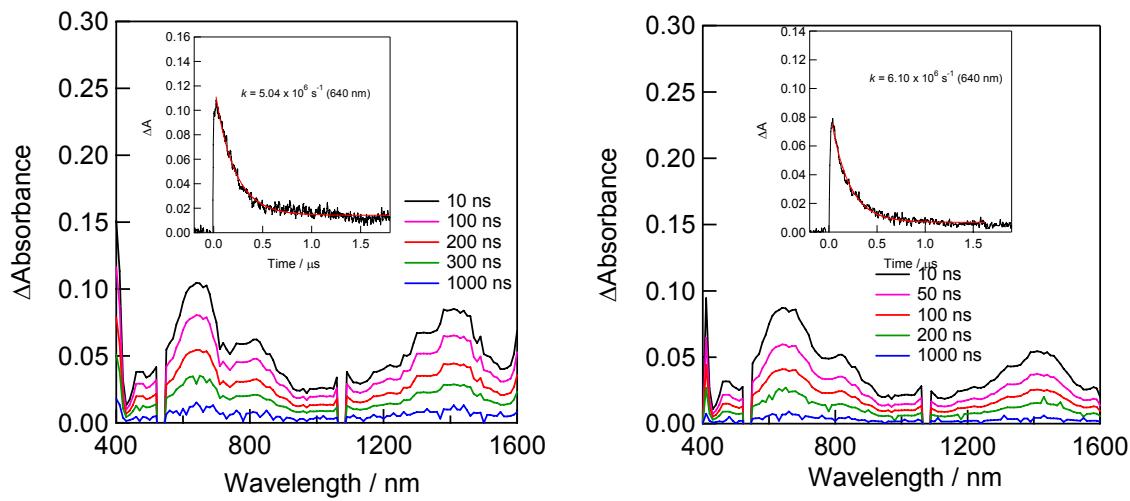


Fig. S5-3. Nanosecond transient absorption spectra of SWCNT(6,5)/PyrCOO[−]/(TMPyP⁺)Zn and SWCNT(7,6)/PyrCOO[−]/(TMPyP⁺)Zn observed by 532 nm (ca. 3 mJ/ pulse) laser irradiation in DMF. Inset: Absorption-time profile.

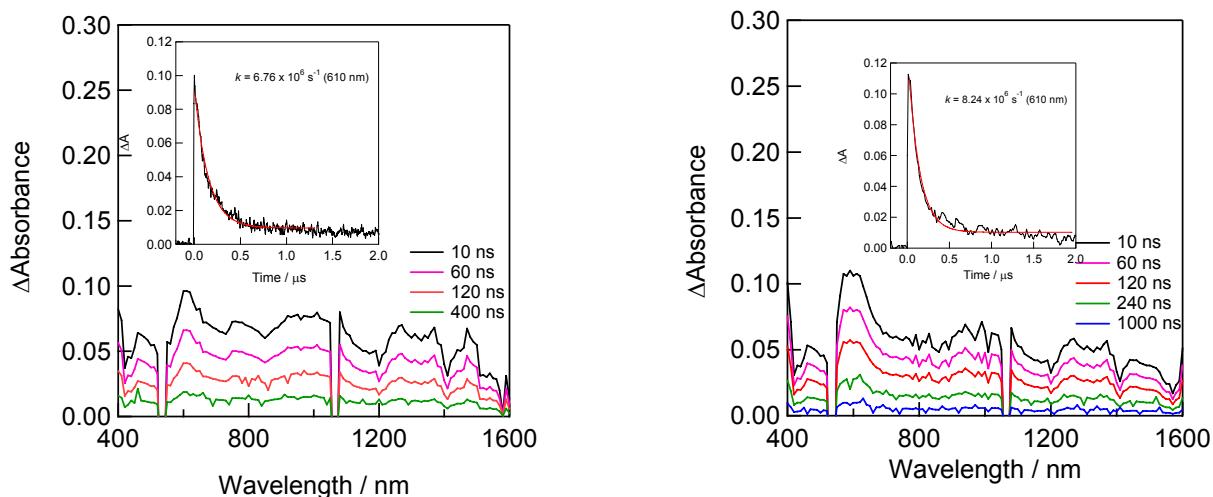


Fig. S5-4. Nanosecond transient absorption spectra of SWCNT(6,5)/PyrCOO[−]/(TMPyP⁺)H₂ and SWCNT(7,6)/PyrCOO[−]/(TMPyP⁺)H₂ observed by 532 nm (ca. 3 mJ/ pulse) laser irradiation in DMF. Inset: Absorption-time profile.

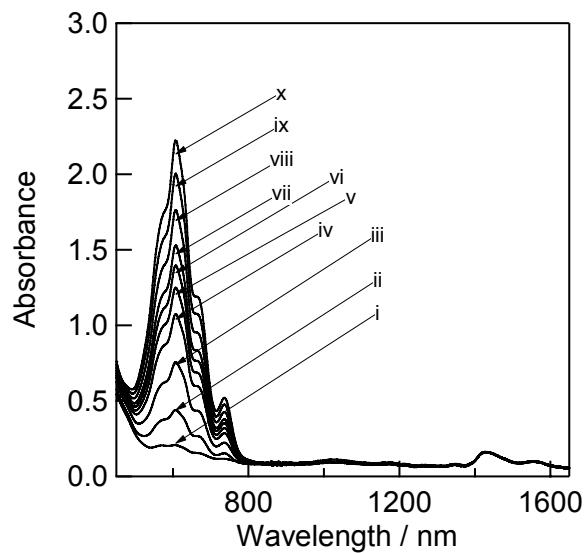


Fig. S6-1. Steady-state absorption spectra of SWCNT(6,5)/PyrCOO \square /(TMPyP $^+$)Zn in Ar-saturated DMF solution measured after 5 laser-shots with laser light (6-ns pulse width) at 532-nm in the presence of HV $^{2+}$ (0.5 mM) and BNAH (i) 0, (ii) 0.5, (iii) 1.0, (iv) 1.5, (v) 2.0 (vi) 2.5 (vii) 3.0 (viii) 3.5 and (ix) 4.0 mM.

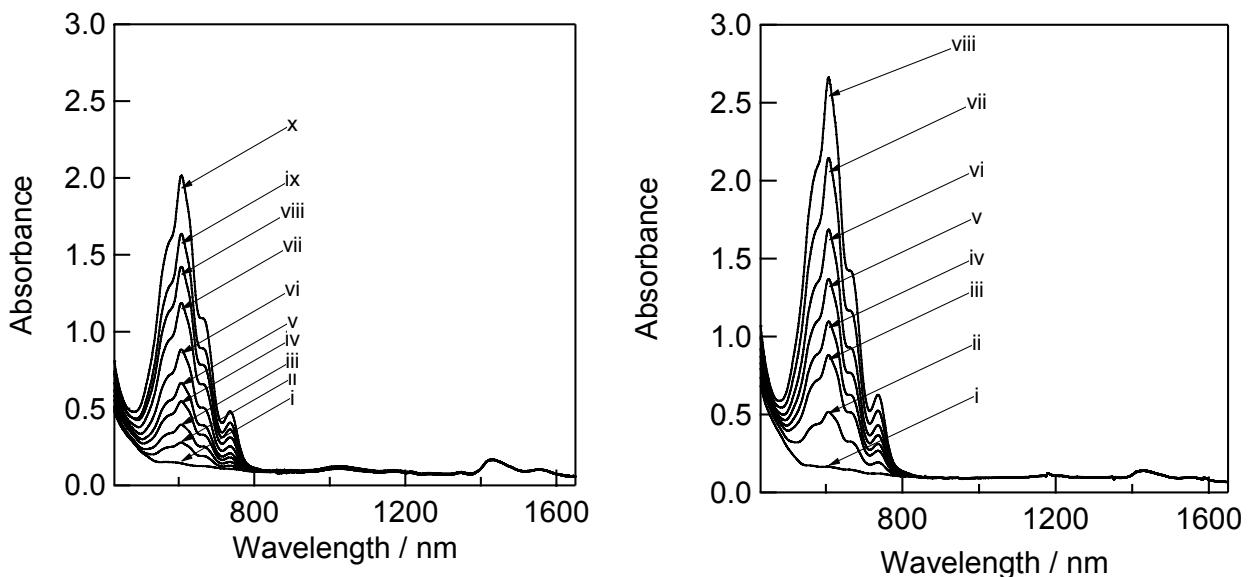


Fig. S6-2. Steady-state absorption spectra of (Left) SWCNT(6,5)/PyrCOO \square /(TMPyP $^+$)H $_2$ and (Right) SWCNT(7,6)/PyrCOO \square /(TMPyP $^+$)H $_2$ in Ar-saturated DMF solution measured after 5 laser-shots with laser light (6-ns pulse width) at 532-nm in the presence of HV $^{2+}$ (0.5 mM) and BNAH (i) 0, (ii) 0.5, (iii) 1.0, (iv) 1.5, (v) 2.0 (vi) 2.5 (vii) 3.0 (viii) 3.5 and (ix) 4.0 mM.

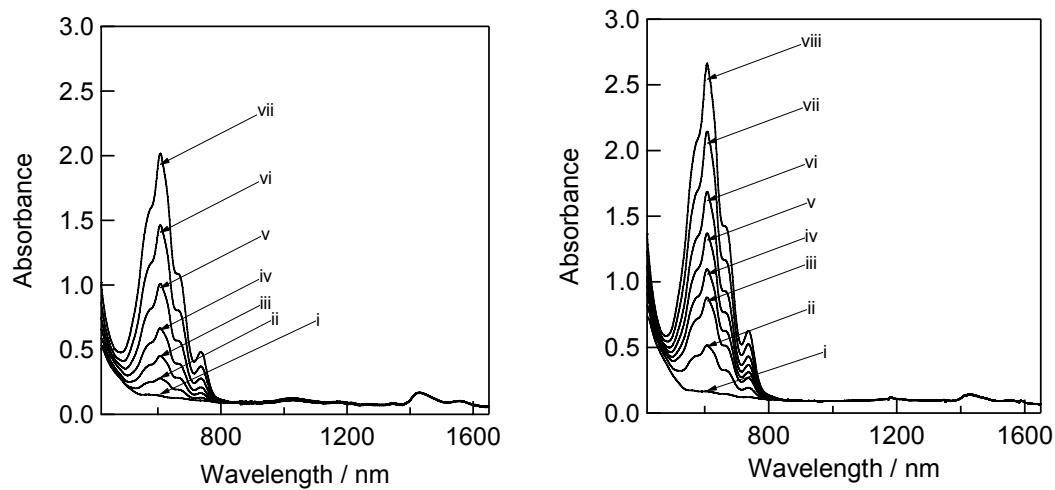


Fig. S6-3. Steady-state absorption spectra of (Left) SWCNT(6,5)/PyrNH₃⁺(TPPS⁻)H₂ and (Right) SWCNT(7,6)/PyrNH₃⁺(TPPS⁻)H₂ in Ar-saturated DMF solution measured after 5 laser-shots with laser light (6-ns pulse width) at 532-nm in the presence of HV²⁺ (0.5 mM) and BNAH (i) 0, (ii) 0.5, (iii) 1.0, (iv) 1.5, (v) 2.0 (vi) 2.5 (vii) 3.0 (viii) 3.5 and (ix) 4.0 mM.

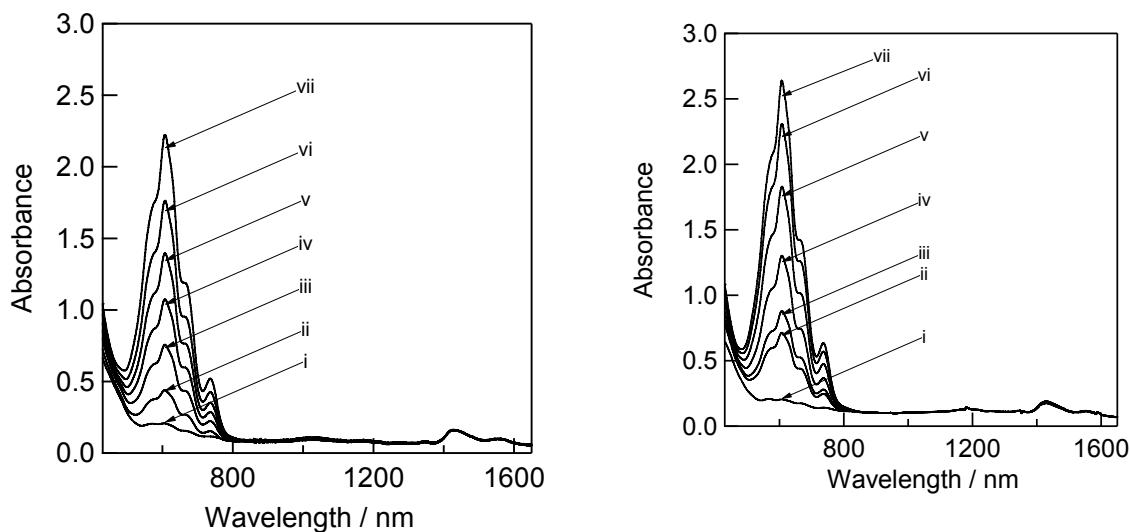


Fig. S6-4. Steady-state absorption spectra of (Left) SWCNT(6,5)/PyrNH₃⁺(TPPS⁻)Zn and (Right) SWCNT(7,6)/PyrNH₃⁺(TPPS⁻)Zn in Ar-saturated DMF solution measured after 5 laser-shots with laser light (6-ns pulse width) at 532-nm in the presence of HV²⁺ (0.5 mM) and BNAH (i) 0, (ii) 0.5, (iii) 1.0, (iv) 1.5, (v) 2.0 (vi) 2.5 (vii) 3.0 (viii) 3.5 and (ix) 4.0 mM.