## Supplemental document

Thermodynamic analysis on energy densities of batteries

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Supplementary data caption:

Table S1. All chemical reactions calculated.

Table S2. Thermodynamic data of typical battery systems. The volumetric energy densities of the batteries containing gas components are not calculated.

Table S3. Thermodynamic data of selected typical lithium battery systems.

Table S4. Thermodynamic data of all lithium battery systems calculated.

Table S5. Thermodynamic data of lithium-ion battery systems with high gravimetric energy densities.

Table S6. Capacity values of typical lithium-free cathodes.

Table S7. Capacity values of lithium-contained cathodes.

Table S8. Capacity values of typical anodes.

Table S9. Capacity values calculated in all.

Table S10. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and lithium.

Table S11. Calculated EMF (V) of conversion reactions between fluorides and

Li, Na, Mg or Al.

Table S12. Key elements in the battery assembling and the toxicity.

Figure S1. Capacity values of typical lithium-free cathodes.

Figure S2. Capacity values of lithium-contained cathodes.

Figure S3. Capacity values of typical anodes.

Figure S4. Calculated EMF values (V) of conversion reactions between transition mental salts and lithium.

Figure S5. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and sodium.

Figure S6. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and magnesium.

Figure S7. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and aluminum.

Figure S8. Calculated EMF (V) of conversion reactions between oxides and Li, Na, Mg or Al.

Figure S9. Calculated EMF (V) of conversion reactions between sulphides and Li, Na, Mg or Al.

Figure S10. Calculated EMF (V) of conversion reactions between nitrides and Li, Na, Mg or Al.

Figure S11. Calculated EMF (V) of conversion reactions between fluorides and Li, Na, Mg or Al.

Figure S12. Elements abundance in the earth.

Table S1. All chemical r	reactions calculated.
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			Li			
$M_a X_b$	n	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )
F						
BeF <sub>2</sub>	2.00	-196.0	1.016	1140.3	894.1	1125.3
NaF	1.00	-41.4	0.429	638.3	235.0	409.2
MgF <sub>2</sub>	2.00	-104.3	0.540	860.4	380.3	632.8
AIF <sub>3</sub>	3.00	-332.0	1.147	957.5	880.0	1395.5
KF	1.00	-49.9	0.517	461.3	213.1	380.5
CaF <sub>2</sub>	2.00	0.2	-0.001	686.6	-0.6	-1.1
ScF <sub>3</sub>	3.00	-223.3	0.771	788.7	505.1	775.8
TiF <sub>3</sub>	3.00	-401.1	1.386	766.8	886.5	1501.9
TiF <sub>4</sub>	4.00	-791.8	2.052	865.5	1450.6	2284.9
VF <sub>3</sub>	3.00	-536.4	1.853	744.9	1157.3	2096.1
CrF <sub>3</sub>	3.00	-659.7	2.279	737.7	1411.5	2707.6
$MnF_2$	2.00	-368.3	1.909	576.8	957.8	2073.2
$MnF_3$	3.00	-763.3	2.637	718.3	1597.1	3002.6
FeF <sub>2</sub>	2.00	-506.8	2.626	571.2	1306.8	2876.5
FeF <sub>3</sub>	3.00	-791.1	2.733	712.5	1644.1	3224.4
CoF <sub>2</sub>	2.00	-528.2	2.737	553.0	1324.1	3074.0
CoF <sub>3</sub>	3.00	-1044.1	3.607	693.6	2120.8	4211.1
NiF <sub>2</sub>	2.00	-571.3	2.961	554.4	1435.2	3407.8
CuF	1.00	-327.7	3.396	324.7	1017.2	3696.7
CuF <sub>2</sub>	2.00	-683.4	3.541	527.9	1644.7	3796.5
$ZnF_2$	2.00	-462.1	2.395	518.5	1094.6	2725.5
GaF₃	3.00	-677.8	2.342	634.5	1276.1	2795.8
RbF	1.00	-59.2	0.613	256.6	147.5	388.0
$YF_3$	3.00	-118.4	0.409	551.1	197.3	435.8
$ZrF_2$	2.00	-262.4	1.360	414.8	509.3	-
ZrF <sub>3</sub>	3.00	-437.5	1.512	542.5	719.0	1647.2
ZrF <sub>4</sub>	4.00	-540.9	1.402	641.1	770.6	1674.3

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NbF <sub>5</sub>	5.00	-1239.5	2.569	713.2	1546.7	2558.3
RuF₃	3.00	-	-	508.7	-	-
$RuF_5$	5.00	-2157.2	4.472	683.5	2596.7	5198.8
AgF	1.00	-400.7	4.153	211.3	831.8	3209.8
SnF <sub>2</sub>	2.00	-573.9	2.974	342.1	934.6	2644.5
LaF <sub>3</sub>	3.00	-139.3	0.481	410.4	178.6	536.0
CeF₃	3.00	-207.1	0.715	407.9	264.0	810.2
0						
BeO-α	2.00	18.9	-0.098	2143.1	-135.0	-153.0
$B_2O_3$	3.00	-244.7	0.845	2309.7	1221.5	1290.9
Na <sub>2</sub> O	1.00	-92.9	0.962	864.9	680.0	967.8
MgO						
microcrystal	2.00	8.1	-0.042	1329.9	-41.5	-61.0
Al <sub>2</sub> O <sub>3</sub>	3.00	-50.7	0.175	1577.2	196.0	271.7
SiO <sub>2 quartz</sub>	4.00	-266.1	0.689	1784.3	841.4	983.1
SiO <sub>2 high</sub>						
cristobalite	4.00	-268.8	0.696	1784.3	850.0	960.5
P <sub>4</sub> O <sub>10</sub>	5.00	-722.2	1.497	1888.2	1898.3	2040.5
K <sub>2</sub> O	1.00	-119.6	1.239	569.1	614.5	1005.1
CaO	2.00	42.1	-0.218	955.9	-167.2	-273.3
Sc <sub>2</sub> O <sub>3</sub>	3.00	67.9	-0.235	1166.0	-210.1	-331.8
TiO	2.00	-66.2	0.343	839.3	236.5	472.7
Ti <sub>2</sub> O <sub>3</sub>	3.00	-124.7	0.431	1118.8	373.7	629.6
Ti <sub>3</sub> O <sub>5</sub>	3.33	-162.9	0.506	1198.6	463.2	742.8
TiO <sub>2</sub> -R	4.00	-233.3	0.605	1342.3	602.2	911.1
TiO <sub>2</sub>	4.00	-233.6	0.605	1342.3	602.9	912.1
TiO <sub>2</sub> -A	4.00	-239.5	0.621	1342.3	618.1	917.9
VO	2.00	-157.0	0.814	800.7	539.6	1259.5
$V_2O_3$	3.00	-272.2	0.940	1072.9	789.4	1390.1
VO <sub>2</sub>	4.00	196.1	-0.508	1292.6	-492.0	-765.9
V <sub>3</sub> O <sub>5</sub>	3.33	-334.3	1.040	1151.2	921.8	-
V <sub>2</sub> O <sub>5</sub>	5.00	-693.3	1.437	1473.6	1532.6	2090.0
CrO <sub>2</sub>	4.00	-577.5	1.496	1276.3	1435.4	2319.2
Cr <sub>2</sub> O <sub>3</sub>	3.00	-312.8	1.080	1058.0	897.3	1622.2
Cr <sub>3</sub> O <sub>4 (Δ</sub>		3.2.0				
H=-1513 kJ						
-1 mol )	2.67	-237.9	0.925	974.7	719.7	1415.8

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	i.					
CrO <sub>3</sub>	6.00	-1171.0	2.023	1608.2	2296.6	2828.0
MnO <sub>2</sub>	4.00	-657.3	1.703	1233.1	1591.8	2642.1
Mn <sub>2</sub> O <sub>3</sub>	3.00	-401.3	1.386	1018.6	1117.3	2034.6
MnO	2.00	-198.3	1.028	755.6	649.4	1405.0
Mn <sub>3</sub> O <sub>4</sub>	2.67	-320.5	1.246	937.1	939.4	1765.9
FeO	2.00	-309.8	1.605	746.1	1003.8	2266.4
Fe <sub>2</sub> O <sub>3</sub>						
hematite	3.00	-470.7	1.626	1007.0	1298.8	2412.2
Fe <sub>3</sub> O <sub>4</sub>	2.67	400.9	1 502	0.06 1	1100 6	2205 5
	2.07	-409.0	1.595	920.1 715 3	109.0	2295.5
	2.00	400.2	1.730	000.4	1000.0	2001.4
	2.07	-490.3	1.905	090.4 717.6	1006 1	2649.2
	2.00	040.0	1.011	072.2	1000.1	2000.1
$C_{10}$	2.00	431.5	- 2 236	972.3 673.9	- 1282 9	- 3105.0
	1.00	-207.6	2 152	374.6	734.7	2313.8
	2 00	-240.7	1 247	658.6	701.8	1649 7
Ga.O.	3.00	-342.7	1 184	857.9	830.9	1742.8
	2 00	-324 0	1.104	604 7	877.9	-
GeOa	4 00	-601.0	1 557	1024 5	1260.9	2179 1
As <sub>4</sub> O <sub>6</sub>	1.00	001.0	1.007	1021.0	1200.0	2170.1
octahedral	3.00	-553.7	1.913	812.8	1284.4	2365.2
As <sub>2</sub> O <sub>5</sub>	5.00	-1011.9	2.097	1166.1	1878.5	3068.7
SeO <sub>2</sub>	4.00	-950.9	2.464	966.2	1904.1	3298.4
Y <sub>2</sub> O <sub>3</sub>	3.00	66.5	-0.230	712.1	-138.2	-300.8
ZrO <sub>2</sub>	4.00	-79.6	0.206	870.0	146.4	300.1
NbO	2.00	-169.3	0.877	492.2	382.9	1149.1
NbO <sub>2</sub>	4.00	-381.9	0.990	858.3	694.9	1450.0
Nb <sub>2</sub> O <sub>5</sub>	5.00	-520.1	1.078	1008.3	862.0	1525.2
MoO <sub>2</sub>	4.00	-589.4	1.527	837.9	1051.5	2281.3
MoO <sub>3</sub>	6.00	-1015.6	1.754	1117.2	1520.1	2597.4
RuO <sub>2</sub>	4.00	-842.4	2.183	805.6	1454.9	3301.9
RuO <sub>4</sub>	8.00	-2092.6	2.711	1298.9	2635.0	3770.7
AgO	2.00	-575.0	2.980	432.7	1159.6	3757.3
Ag <sub>2</sub> O	1.00	-275.0	2.850	231.3	622.0	2625.8
Ag <sub>2</sub> O <sub>3</sub>	3.00	-902.5	3.118	609.7	1641.9	-
La <sub>2</sub> O <sub>3</sub>	3.00	11.1	-0.038	493.6	-16.8	-48.2

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CeO <sub>2</sub>	4.00	-97.8	0.253	622.9	135.9	358.2
Ce <sub>2</sub> O <sub>3</sub>	3.00	11.3	-0.039	489.9	-17.0	-47.9
WO <sub>2</sub>	4.00	-588.5	1.525	496.7	671.1	2271.2
WO <sub>3</sub>	6.00	-919.6	1.589	693.6	934.0	2318.3
Na <sub>2</sub> O <sub>2</sub>	1.00	-61.6	0.639	687.4	372.7	636.4
K <sub>2</sub> O <sub>2</sub>	1.00	-72.9	0.756	486.4	326.5	522.9
Ag <sub>2</sub> O <sub>2</sub>	1.00	-299.3	3.102	216.4	635.5	2804.1
NaO <sub>2</sub>	1.00	-	-	487.4	-	-
KO <sub>2</sub>	1.00	-	-	377.0	-	-
CaO <sub>2</sub>	1.00	-	-	371.8	-	-
S						
BeS	2.00	-206.0	1.068	1304.9	1041.2	1318.4
$B_2S_3$	3.00	-534.7	1.847	1364.8	1862.8	2016.7
Na <sub>2</sub> S	1.00	-44.6	0.462	686.8	269.5	364.1
MgS	2.00	-97.2	0.504	950.9	384.3	574.1
$AI_2S_3$	3.00	-338.5	1.169	1070.9	980.4	1234.6
SiS <sub>2</sub>	4.00	-665.4	1.724	1162.5	1540.5	1901.6
$P_4S_3$	1.50	-289.5	2.000	730.6	1229.0	1725.6
P <sub>4</sub> S <sub>7</sub>	3.50	-693.5	2.054	1077.1	1729.4	2259.3
$P_2S_5$	5.00	-1027.9	2.131	1205.8	1957.7	2384.6
K <sub>2</sub> S	1.00	-37.5	0.389	486.1	167.8	233.1
CaS	2.00	38.4	-0.199	743.0	-124.0	-198.1
$Sc_2S_3$	3.00	-	-	864.1	-	-
TiS <sub>2</sub>	4.00	-475.8	1.233	957.2	945.7	1550.8
TiS	2.00	-168.9	0.875	670.6	500.2	1003.6
Ti <sub>2</sub> S <sub>3</sub>	3.00	-	-	837.8	-	-
VS	2.00	-	-	645.8	-	-
$V_2S_3$	3.00	-	-	811.8	-	-
VS <sub>2</sub>	4.00	-	-	931.6	-	-
$V_2S_5$	5.00	-	-	1022.1	-	-
CrS	2.00	-264.6	1.371	637.7	750.3	-
$Cr_2S_3$	3.00	-574.6	1.985	803.3	1319.9	2435.0
MnS	2.00	-220.6	1.143	616.1	607.4	1210.5
MnS <sub>2</sub>	4.00	-653.0	1.692	900.4	1235.4	-
FeS	2.00	-338.6	1.755	609.7	924.0	2104.1
FeS <sub>2(marcasite</sub>	4.00	-721.9	1.870	893.6	1357.3	2642.3

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FeS <sub>2(pvrite)</sub>	4.00	-711.1	1.843	893.6	1337.0	2602.7
CoS	2.00	-	-	589.0	-	-
$Co_2S_3$	3.00	-	-	751.2	-	-
CoS <sub>2</sub>	4.00	-732.4	1.898	871.1	1348.7	2523.6
Co <sub>3</sub> S <sub>4</sub>	2.67	-423.0	1.644	702.8	977.6	-
NiS	2.00	-359.5	1.863	590.6	954.3	2349.8
Ni <sub>3</sub> S <sub>4</sub>	2.67	-488.0	1.897	704.5	1130.1	2423.8
$Ni_3S_2$	1.33	-222.7	1.731	446.3	692.4	-
NiS <sub>2</sub>	4.00	-753.3	1.952	872.8	1389.5	-
CuS	2.00	-385.4	1.997	560.6	977.7	2323.1
Cu <sub>2</sub> S	1.00	-176.4	1.828	336.8	566.3	1800.9
ZnS	2.00	-237.7	1.232	550.0	593.0	1321.3
$Ga_2S_3$	3.00	-405.6	1.401	682.4	812.7	1590.7
GaS	2.00	-234.3	1.214	526.6	562.7	1242.9
GeS	2.00	-367.5	1.904	511.9	860.8	1980.9
GeS <sub>2</sub>	4.00	-723.4	1.874	783.8	1221.3	2062.4
$As_2S_3$	3.00	-574.2	1.984	653.6	1108.8	2139.5
Se <sub>2</sub> S <sub>6</sub>	6.00	-	-	918.1	-	-
Se <sub>4</sub> S <sub>4</sub>	2.00	-	-	482.8	-	-
RuS <sub>2</sub>	4.00	-	-	648.9	-	-
$Y_2S_3$	3.00	-	-	586.9	-	-
ZrS <sub>2</sub>	4.00	-308.0	0.798	690.1	467.2	928.5
NbS <sub>2</sub>	4.00	-	-	690.1	-	-
MoS <sub>2</sub>	4.00	-652.1	1.690	669.7	964.3	2166.0
$Mo_2S_3$	3.00	-519.2	1.794	558.2	874.8	2276.0
Ag <sub>2</sub> S(argent						
ite)	1.00	-199.2	2.064	216.3	422.8	1835.8
LaS	2.00	12.5	-0.065	313.5	-18.8	-61.5
$La_2S_3$	3.00	-54.3	0.188	430.0	72.6	195.6
CeS	2.00	12.5	-0.065	311.3	-18.7	-62.9
WS <sub>2</sub>	4.00	-628.1	1.627	432.3	632.8	2061.9
$Na_2S_2$	1.00	-	-	486.8	-	-
N						
Be <sub>3</sub> N <sub>2</sub>	2.00	91.9	-0.476	2921.1	-792.0	-779.1
BN	3.00	99.8	-0.345	3239.7	-607.4	-550.3
(CN)	3.00	-	-	3090.4	-	-

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Na₃N	1.00	-	-	969.0	-	-
$Mg_3N_2$	2.00	47.9	-0.248	1593.3	-280.0	-346.4
AIN	3.00	158.4	-0.547	1961.6	-711.8	-852.9
Si <sub>3</sub> N <sub>4</sub>	4.00	42.7	-0.111	2292.6	-159.3	-177.9
K₃N	1.00	-	-	612.4	-	-
Ca <sub>3</sub> N <sub>2</sub>	2.00	36.9	-0.191	1084.7	-162.1	-230.5
ScN	3.00	155.2	-0.536	1363.6	-540.2	-
TiN	3.00	115.2	-0.398	1299.5	-387.0	-629.1
VN	3.00	62.5	-0.216	1238.0	-202.3	-350.0
CrN	3.00	-35.6	0.123	1218.2	113.9	197.1
Cr <sub>2</sub> N	1.50	-13.2	0.091	681.4	52.8	130.2
Mn₄N	0.75	-5.9	0.082	344.0	25.8	-
$Mn_5N_2$	1.20	-20.4	0.177	531.2	82.4	-
Fe <sub>2</sub> N	1.50	-	-	639.7	-	-
Fe₄N	0.75	-31.2	0.431	338.7	134.3	-
Co <sub>3</sub> N	1.00	-31.4	0.326	421.4	123.7	380.4
NiN	3.00	-	-	1106.0	-	-
Cu₃N	1.00	-	-	392.9	-	-
Zn <sub>3</sub> N <sub>2</sub>	2.00	-98.8	0.512	717.3	309.9	722.3
GaN	3.00	-50.9	0.176	960.3	135.1	268.0
Ge <sub>3</sub> N <sub>4</sub>	4.00	-	-	1174.0	-	-
As <sub>3</sub> N <sub>5</sub>	5.00	-	-	1363.7	-	-
SeN <sub>2</sub>	6.00	-	-	1503.3	-	-
YN	3.00	140.0	-0.484	781.3	-314.2	-677.7
ZrN	3.00	208.1	-0.719	764.1	-458.6	-1073.7
NbN	3.00	77.3	-0.267	752.1	-168.1	-416.0
MoN	3.00	-	-	731.3	-	-
RuN	3.00	-	-	698.7	-	-
Ag₃N	1.00	-	-	238.2	-	-
LaN	3.00	142.4	-0.492	525.8	-227.7	-641.1
CeN	3.00	166.4	-0.575	521.7	-264.1	-789.5
WN <sub>2</sub>	6.00	-	-	759.1	-	-
NaN <sub>3</sub>	1.00	-	-	179.8	-	-
KN <sub>3</sub>	1.00	-	-	162.3	-	-
AgN <sub>3</sub>	1.00	-	-	114.6	-	-
Р		-				
Be <sub>3</sub> P <sub>2</sub>	2.00	-	-	1807.2	-	-

BP	3.00	-	-	1924.2	-	-
Na₃P	1.00	-	-	804.5	-	-
Mg <sub>3</sub> P <sub>2</sub>	2.00	-	-	1192.4	-	-
AIP	3.00	-	-	1387.3	-	-
Si <sub>3</sub> P <sub>4</sub>	4.00	-	-	1545.1	-	-
K <sub>3</sub> P	1.00	-	-	542.3	-	-
Ca <sub>3</sub> P <sub>2</sub>	2.00	-	-	882.7	-	-
ScP	3.00	-	-	1058.9	-	-
TiP	3.00	-	-	1019.8	-	-
(VP)	3.00	-	-	981.6	-	-
CrP	3.00	-	-	969.1	-	-
MnP	3.00	-	-	935.9	-	-
Mn <sub>2</sub> P	1.50	-	-	570.9	-	-
FeP	3.00	-	-	926.1	-	-
Fe <sub>2</sub> P	1.50	-	-	563.6	-	-
Fe₃P	1.00	-	-	405.0	-	-
Co <sub>2</sub> P	1.50	-	-	540.2	-	-
Ni <sub>2</sub> P	1.50	-	-	542.0	-	-
Ni <sub>3</sub> P	1.00	-	-	388.3	-	-
Ni <sub>5</sub> P <sub>2</sub>	1.20	-	-	452.5	-	-
CuP <sub>2</sub>	6.00	-	-	1281.4	-	-
Cu₃P	1.00	-	-	362.8	-	-
Zn <sub>3</sub> P <sub>2</sub>	2.00	-	-	623.0	-	-
GaP	3.00	-	-	798.5	-	-
GeP	3.00	-	-	776.0	-	-
Ge <sub>3</sub> P <sub>4</sub>	4.00	-	-	940.9	-	-
As <sub>3</sub> P <sub>5</sub>	5.00	-	-	1059.0	-	-
Se(P) <sub>2</sub>	6.00	-	-	1141.2	-	-
YP	3.00	-	-	670.7	-	-
ZrP <sub>2</sub>	6.00	-	-	1049.9	-	-
NbP	3.00	-	-	649.0	-	-
MoP	3.00	-	-	633.5	-	-
RuP	3.00	-	-	608.9	-	-
AgP <sub>2</sub>	6.00	-	-	947.0	-	-
AgP₃	9.00	-	-	1201.3	-	-
LaP	3.00	-	-	473.3	-	-
Ce <sub>3</sub> P <sub>4</sub>	4.00	-	-	590.9	-	-

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WP <sub>2</sub>	6.00	-	-	654.3	-	-
Cl						
BeCl <sub>2</sub>	2.00	-323.2	1.675	670.7	957.1	1319.1
PCl₅	5.00	-1617.1	3.352	643.5	1849.0	2736.5
NaCl	1.00	-0.3	0.003	458.6	1.3	2.1
MgCl <sub>2</sub>	2.00	-177.0	0.917	563.0	450.7	734.4
AICI <sub>3</sub>	3.00	-524.4	1.812	603.0	944.9	1570.3
KCI	1.00	24.1	-0.250	359.5	-82.1	-132.6
CaCl <sub>2</sub>	2.00	-20.0	0.104	483.0	44.5	71.6
ScCl <sub>3</sub>	3.00	-327.3	1.131	531.4	528.2	891.0
TiCl <sub>2</sub>	2.00	-304.4	1.577	451.3	637.4	1322.4
TiCl <sub>3</sub>	3.00	-499.7	1.726	521.3	793.0	1424.9
VCl <sub>2</sub>	2.00	-362.8	1.880	439.9	742.5	1581.6
VCl <sub>3</sub>	3.00	-641.9	2.218	511.2	1001.0	1950.2
CrCl <sub>2</sub>	2.00	-412.8	2.139	436.1	838.3	1669.8
CrCl <sub>3</sub>	3.00	-667.1	2.305	507.7	1034.2	1922.9
MnCl <sub>2</sub>	2.00	-328.3	1.701	425.9	652.7	1335.8
FeCl <sub>2</sub>	2.00	-466.5	2.417	422.9	921.4	1964.0
FeCl <sub>3</sub>	3.00	-819.2	2.830	495.7	1243.3	2397.2
CoCl <sub>2</sub>	2.00	-499.0	2.586	412.8	964.4	2144.4
NiCl <sub>2</sub>	2.00	-509.8	2.642	413.6	987.0	2250.7
CuCl	1.00	-264.5	2.741	270.7	693.5	1990.5
CuCl <sub>2</sub>	2.00	-593.1	3.074	398.7	1110.7	2513.7
ZnCl <sub>2</sub>	2.00	-399.4	2.070	393.3	738.8	1522.3
GaCl₃	3.00	-698.4	2.413	456.6	985.2	1759.1
AsCl <sub>5</sub>	5.00	-	-	531.4	-	-
SeCl <sub>4</sub>	4.00	-	-	485.6	-	-
YCl <sub>3</sub>	3.00	-225.5	0.779	411.8	289.8	550.3
ZrCl <sub>2</sub>	2.00	-382.8	1.984	330.6	604.1	1375.5
ZrCl <sub>3</sub>	3.00	-507.2	1.752	406.9	645.1	1357.6
ZrCl <sub>4</sub>	4.00	-647.7	1.678	460.0	689.9	1330.6
NbCl <sub>5</sub>	5.00	-1238.8	2.568	496.0	1128.7	2121.9
NbCl <sub>4</sub>	4.00	-932.1	2.415	456.7	986.4	2065.6
MoCl <sub>4</sub>	4.00	-1135.6	2.942	450.9	1188.0	-
MoCl <sub>5</sub>	5.00	-1499.0	3.107	490.5	1352.3	2631.5

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MoCl <sub>6</sub>	6.00	-1915.4	3.309	521.0	1518.8	-
RuCl₃	3.00	-993.5	3.432	387.6	1209.0	2605.8
AgCl	1.00	-274.6	2.846	187.0	507.6	1967.2
SnCl <sub>2</sub>	2.00	-482.6	2.501	282.7	658.7	1811.5
LaCl₃	3.00	-	-	327.8	-	-
CeCl <sub>3</sub>	3.00	-168.4	0.582	326.2	175.0	462.8
WCl <sub>2</sub>	2.00	-548.8	2.844	210.4	567.5	2093.4
WCl <sub>4</sub>	4.00	-1177.6	3.051	329.2	925.6	2670.7
WCl <sub>5</sub>	5.00	-1520.0	3.151	371.1	1066.7	2671.3
WCl <sub>6</sub>	6.00	-1850.4	3.196	405.5	1173.0	2696.1
BO <sub>2</sub> <sup>-</sup>						
NaBO <sub>2</sub>	1.00	-56.7	0.588	407.3	216.6	396.8
Mg(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	487.6	-	-
AI (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	517.4	-	-
KBO <sub>2</sub>	1.00	-39.5	0.409	327.2	123.4	-
Sc(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	463.7	-	-
Ti (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	489.3	-	-
V(BO <sub>2</sub> ) <sub>5</sub>	5.00	-	-	505.7	-	-
Cr(BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	520.6	-	-
Mn (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	438.5	-	-
Fe (BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	378.9	-	-
Co (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	429.1	-	-
Ni(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	371.4	-	-
Cu(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	359.3	-	-
Zn(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	355.0	-	-
Ga(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	405.8	-	-
Ge(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	439.6	-	-
Y (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	369.9	-	-
Zr (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	408.5	-	-
Nb BO <sub>2</sub>	1.00	-	-	197.5	-	-
MoBO <sub>2</sub>	1.00	-	-	193.2	-	-
Ru(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	287.1	-	-
Ag BO <sub>2</sub>	1.00	-	-	177.9	-	-
La (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	300.8	-	-
Ce(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	344.3	-	-

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W (BO <sub>2</sub> ) <sub>6</sub>	6.00	-		-		364.9	-		-	
Be <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-		-		1111.7	-		-	
Ca <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-		-		676.1	-		-	
CO3 <sup>2-</sup>										
Be CO <sub>3</sub>	2.00	-		-		776.6	-		-	
$B_2(CO_3)_3$	3.00	-		-		797.5	-		-	
Na <sub>2</sub> CO <sub>3</sub>	1.00		-43.9		0.455	505.7		203.3		359.8
Mg CO <sub>3</sub>	2.00		-120.0		0.622	635.8		339.5		617.3
Al <sub>2</sub> ( CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		687.2	-		-	
K <sub>2</sub> CO <sub>3</sub>	1.00		-34.3		0.356	387.8		125.3		220.7
CaCO <sub>3</sub>	2.00		-3.3		0.017	535.6		8.1		14.8
Sc 2(CO3)3	3.00	-		-		595.7	-		-	
Ti (CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-		638.6	-		-	
V <sub>2</sub> (CO <sub>3</sub> ) <sub>5</sub>	5.00	-		-		666.8	-		-	
Cr (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-		693.1	-		-	
MnCO <sub>3</sub>	2.00		-315.4		1.635	466.3		680.1		1535.4
Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		551.2	-		-	
FeCO <sub>3</sub>	2.00		-465.4		2.412	450.7		973.3		2302.3
Co CO <sub>3</sub>	2.00		-495.3		2.567	450.7		1035.9		2533.2
Ni CO <sub>3</sub>	2.00		-514.2		2.665	451.6		1077.4		2693.1
CuCO <sub>3</sub>	2.00	-		-		433.8	-		-	
Zn CO <sub>3</sub>	2.00		-400.6		2.076	427.5		798.8		2049.9
Ga <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		503.4	-		-	
Ge(CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-		556.5	-		-	
$Y_2(CO_3)_3$	3.00	-		-		449.4	-		-	
ZrCO <sub>3</sub>	2.00	-		-		354.4	-		-	
Nb <sub>2</sub> CO <sub>3</sub>	1.00	-		-		218.1	-		-	
Mo <sub>2</sub> CO <sub>3</sub>	1.00	-		-		212.8	-		-	
Ru <sub>2</sub> CO <sub>3</sub>	1.00	-		-		204.5	-		-	
Ag <sub>2</sub> CO <sub>3</sub>	1.00		-347.7		3.603	194.4		666.9		2706.2
La 2(CO3)3	3.00	-		-		351.2	-		-	
$Ce_2(CO_3)_3$	3.00	-		-		349.4	-		-	
W (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-		441.9	-		-	
PO <sub>4</sub> <sup>3-</sup>										

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Be <sub>3(</sub> PO <sub>4)2</sub>	2.00	-		-		741.1	-		-	
Na <sub>3</sub> PO <sub>4</sub>	1.00		-102.3		1.060	490.4		461.4		823.5
Mg <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00		-217.6		1.128	611.8		595.5		918.3
AI PO <sub>4</sub>	3.00		-477.8		1.651	659.3		929.6		1532.0
K <sub>3</sub> PO <sub>4</sub>	1.00	-		-		378.8	-		-	
Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00		-102.3		0.530	518.4		242.2		482.1
Sc PO <sub>4</sub>	3.00	-		-		574.6	-		-	
Ti <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	4.00	-		-		614.4	-		-	
V <sub>3</sub> (PO <sub>4</sub> ) <sub>5</sub>	5.00	-		-		640.5	-		-	
CrPO <sub>4</sub>	3.00	-		-		547.1	-		-	
Mn PO₄	3.00	-		-		536.4	-		-	
Mn3( PO <sub>4 )2</sub>	2.00	-		-		453.3	-		-	
Fe PO <sub>4</sub>	3.00		-908.2		3.138	533.1		1469.8		2755.8
Co3(PO <sub>4)2</sub>	2.00	-		-		438.5	-		-	
Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-				439.3	-		-	
$Cu_3(PO_4)_2$	2.00	-		-		422.5	-		-	
Zn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00		-509.3		2.639	416.5		992.1		2431.8
Ga PO <sub>4</sub>	3.00	-		-		488.2	-		-	
Ge <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	4.00	-		-		538.0	-		-	
Y PO <sub>4</sub>	3.00	-		-		437.3	-		-	
Zr <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-		-		346.9	-		-	
Nb <sub>3</sub> PO <sub>4</sub>	1.00	-		-		215.2	-		-	
Mo <sub>3</sub> PO <sub>4</sub>	1.00	-		-		210.0	-		-	
Ru <sub>3</sub> PO <sub>4</sub>	1.00	-		-		201.9	-		-	
Ag <sub>3</sub> PO <sub>4</sub>	1.00	-		-		192.1	-		-	
La PO₄	3.00	-		-		343.8	-		-	
CePO <sub>4</sub>	3.00	-		-		342.0	-		-	
W (PO <sub>4</sub> ) <sub>2</sub>	6.00	-		-		430.2	-		-	
SO4 <sup>2-</sup>										
Be SO <sub>4</sub>	2.00		-232.3		1.204	510.1		542.4		948.6
Na <sub>2</sub> SO <sub>4</sub>	1.00		-25.8		0.267	377.4		91.7		182.0
Mg SO <sub>4</sub>	2.00		-151.1		0.783	445.3		312.6		589.1
Al <sub>2</sub> ( SO <sub>4</sub> ) <sub>3</sub>	3.00		-229.1		0.791	470.0		331.6	-	
AIK(SO <sub>4</sub> ) <sub>2</sub>	4.00		-403.3		1.045	415.2		391.7	-	

K <sub>2</sub> SO <sub>4</sub>	1.00	-0.1	0.002	307.6	0.4	0.9
CaSO <sub>4</sub>	2.00	-12.6	0.065	393.7	23.3	48.6
Sc 2(SO4)3	3.00	-	-	425.3	-	-
Ti (SO <sub>4</sub> ) <sub>2</sub>	4.00	-	-	446.7	-	-
$V_2(SO_4)_5$	5.00	-	-	460.3	-	-
Cr 2(SO <sub>4</sub> ) <sub>3</sub>	3.00	-674.0	2.328	410.0	863.1	1831.0
MnSO <sub>4</sub>	2.00	-364.3	1.888	355.0	613.7	1396.5
FeSO <sub>4</sub>	2.00	-500.9	2.596	352.9	839.2	2057.8
$Fe_2(SO_4)_3$	3.00	-851.2	2.941	402.1	1071.0	2284.7
Co SO <sub>4</sub>	2.00	-539.3	2.795	345.8	887.1	2210.4
Ni SO4	2.00	-562.0	2.912	346.4	925.7	2417.0
CuSO <sub>4</sub>	2.00	-659.5	3.418	335.8	1055.9	2604.7
Zn SO <sub>4</sub>	2.00	-450.2	2.333	332.0	713.2	1826.1
$Y_2(SO_4)_3$	3.00	-	-	345.1	-	-
Zr(SO <sub>4)2</sub>	4.00	-	-	378.3	-	-
$Nb_2SO_4$	1.00	-	-	190.2	-	-
$Mo_2SO_4$	1.00	-	-	186.2	-	-
$Ru_2SO_4$	1.00	-	-	179.8	-	-
Ag <sub>2</sub> SO <sub>4</sub>	1.00	-351.7	3.645	171.9	599.9	2347.9
La <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	3.00	-	-	284.1	-	-
$Ce_2(SO_4)_3$	3.00	-181.1	0.626	282.9	164.9	-
W (SO <sub>4</sub> ) <sub>3</sub>	6.00	-	-	340.7	-	-

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			Na			
$M_a X_b$	n	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )
F						
BeF <sub>2</sub>	2.00	-113.2	0.587	1140.3	338.2	450.6
LiF	1.00	41.4	-0.429	1033.2	-235.0	-343.0
$MgF_2$	2.00	-21.5	0.111	860.4	55.2	88.9
AIF <sub>3</sub>	3.00	-207.8	0.718	957.5	377.4	587.9
KF	1.00	-8.5	0.088	461.3	29.1	50.1
CaF <sub>2</sub>	2.00	83.0	-0.430	686.6	-185.9	-320.4

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1	1	1	i -			
ScF <sub>3</sub>	3.00	-99.1	0.342	788.7	161.0	245.6
TiF <sub>3</sub>	3.00	-276.9	0.957	766.8	442.5	723.6
TiF <sub>4</sub>	4.00	-626.2	1.623	865.5	806.0	1250.8
VF <sub>3</sub>	3.00	-412.2	1.424	744.9	647.3	1109.6
CrF <sub>3</sub>	3.00	-535.5	1.850	737.7	835.8	1490.6
MnF <sub>2</sub>	2.00	-285.5	1.479	576.8	570.9	1120.9
MnF <sub>3</sub>	3.00	-639.1	2.208	718.3	981.3	1728.2
FeF <sub>2</sub>	2.00	-424.0	2.197	571.2	842.3	1674.3
FeF <sub>3</sub>	3.00	-666.9	2.304	712.5	1018.9	1847.7
CoF <sub>2</sub>	2.00	-445.4	2.308	553.0	865.7	1789.6
CoF <sub>3</sub>	3.00	-919.9	3.178	693.6	1382.0	2530.5
NiF <sub>2</sub>	2.00	-488.5	2.531	554.4	951.1	1996.3
CuF	1.00	-286.3	2.967	324.7	753.6	2251.2
CuF <sub>2</sub>	2.00	-600.6	3.112	527.9	1130.9	2336.4
ZnF <sub>2</sub>	2.00	-379.3	1.966	518.5	705.4	1538.1
GaF <sub>3</sub>	3.00	-553.6	1.913	634.5	785.8	1546.3
RbF	1.00	-17.8	0.184	256.6	38.7	93.0
YF <sub>3</sub>	3.00	5.8	-0.020	551.1	-7.5	-15.0
ZrF <sub>2</sub>	2.00	-179.6	0.931	414.8	284.8	-
$ZrF_3$	3.00	-313.3	1.083	542.5	0.0	0.0
ZrF <sub>4</sub>	4.00	-375.3	0.972	641.1	402.2	786.5
NbF <sub>5</sub>	5.00	-1032.5	2.140	713.2	947.0	1524.8
RuF <sub>3</sub>	3.00	-	-	508.7	-	-
RuF₅	5.00	-1950.2	4.043	683.5	1741.8	3209.8
AgF	1.00	-359.3	3.724	211.3	666.0	2199.3
SnF <sub>2</sub>	2.00	-491.1	2.545	342.1	673.1	1670.0
LaF <sub>3</sub>	3.00	-15.1	0.052	410.4	15.9	40.3
$CeF_3$	3.00	-82.9	0.286	407.9	86.5	223.3
0						
BeO-α	2.00	204.6	-1.060	2143.1	-800.6	-1020.1
B <sub>2</sub> O <sub>3</sub>	3.00	33.9	-0.117	2309.7	-90.7	-111.1
Li <sub>2</sub> O	1.00	92.9	-0.962	1793.9	-680.0	-828.7
MgO	0.00	400.0	4 00 1	4000.0	000.0	000 5
microcrystal	2.00	193.8	-1.004	1329.9	-623.9	-923.5
Al <sub>2</sub> O <sub>3</sub>	3.00	227.9	-0.787	1577.2	-527.7	-754.4
SiO <sub>2 quartz</sub>	4.00	105.3	-0.273	1784.3	-192.4	-247.9

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SiO <sub>2 high</sub>						
cristobalite	4.00	102.6	-0.266	1784.3	-187.4	-236.4
P <sub>4</sub> O <sub>10</sub>	5.00	-257.9	0.535	1888.2	385.4	471.9
K <sub>2</sub> O	1.00	-26.7	0.277	569.1	105.8	169.6
CaO	2.00	227.8	-1.180	955.9	-620.0	-985.8
Sc <sub>2</sub> O <sub>3</sub>	3.00	346.5	-1.197	1166.0	-697.7	-1081.9
TiO	2.00	119.5	-0.619	839.3	-302.2	-550.5
Ti <sub>2</sub> O <sub>3</sub>	3.00	153.9	-0.532	1118.8	-303.4	-490.5
Ti <sub>3</sub> O <sub>5</sub>	3.33	146.6	-0.456	1198.6	-269.5	-421.7
TiO <sub>2</sub> -R	4.00	138.1	-0.358	1342.3	-223.2	-336.5
TiO <sub>2</sub>	4.00	137.8	-0.357	1342.3	-222.8	-335.9
TiO <sub>2</sub> -A	4.00	131.9	-0.342	1342.3	-213.3	-317.9
VO	2.00	28.7	-0.149	800.7	-70.6	-142.3
$V_2O_3$	3.00	6.4	-0.022	1072.9	-12.4	-20.6
VO <sub>2</sub>	4.00	567.5	-1.470	1292.6	-901.2	-1383.7
V <sub>3</sub> O <sub>5</sub>	3.33	-24.8	0.077	1151.2	44.7	-
V <sub>2</sub> O <sub>5</sub>	5.00	-229.0	0.475	1473.6	309.0	436.7
CrO <sub>2</sub>	4.00	-206.1	0.534	1276.3	325.4	511.3
Cr <sub>2</sub> O <sub>3</sub>	3.00	-34.2	0.118	1058.0	65.5	110.9
Cr <sub>3</sub> O <sub>4 ( Δ</sub>						
H=-1513 kJ -1	0.07	0.7	0.000	0747	10.0	05.7
mol )	2.67	9.7	-0.038	974.7	-19.9	-35.7
CrO <sub>3</sub>	6.00	-613.9	1.061	1608.2	716.7	951.5
MnO <sub>2</sub>	4.00	-285.9	0.741	1233.1	443.9	709.6
Mn <sub>2</sub> O <sub>3</sub>	3.00	-122.7	0.424	1018.6	230.4	392.3
MnO	2.00	-12.6	0.065	755.6	29.9	57.7
Mn <sub>3</sub> O <sub>4</sub>	2.67	-72.9	0.283	937.1	147.3	256.6
FeO	2.00	-124.1	0.643	746.1	292.6	580.6
	3 00	-192.2	0 664	1007.0	358 7	618.4
Fe <sub>3</sub> O <sub>4</sub>	0.00	102.2	0.001	1001.0	000.1	010.1
magnetite	2.67	-162.2	0.630	926.1	325.3	576.7
CoO	2.00	-161.3	0.836	715.3	370.6	758.9
Co <sub>3</sub> O <sub>4</sub>	2.67	-242.7	0.943	890.4	476.1	883.0
NiO	2.00	-163.8	0.849	717.6	377.1	792.3
Ni <sub>2</sub> O <sub>3</sub>	3.00	-	-	972.3	-	-
CuO	2.00	-245.8	1.274	673.9	543.9	1137.8
Cu <sub>2</sub> O	1.00	-114.8	1.189	374.6	337.2	894.7

ZnO	2.00	-55.0	0.285	658.6	119.9	246.7
$Ga_2O_3$	3.00	-64.1	0.221	857.9	109.4	205.3
GeO	2.00	-138.3	0.717	604.7	285.4	-
GeO <sub>2</sub>	4.00	-229.6	0.595	1024.5	324.4	534.0
As <sub>4</sub> O <sub>6</sub>		<i>(</i>				
octahedral	3.00	-275.1	0.950	812.8	455.2	786.8
As <sub>2</sub> O <sub>5</sub>	5.00	-547.6	1.135	1166.1	661.7	1048.3
SeO <sub>2</sub>	4.00	-579.5	1.502	966.2	793.3	1309.9
$Y_2O_3$	3.00	345.1	-1.192	712.1	-527.0	-1024.6
ZrO <sub>2</sub>	4.00	291.8	-0.756	870.0	-376.7	-695.8
NbO	2.00	16.4	-0.085	492.2	-29.5	-73.3
NbO <sub>2</sub>	4.00	-10.5	0.027	858.3	13.4	25.1
Nb <sub>2</sub> O <sub>5</sub>	5.00	-55.9	0.116	1008.3	62.6	104.7
MoO <sub>2</sub>	4.00	-218.0	0.565	837.9	275.4	528.5
MoO <sub>3</sub>	6.00	-458.5	0.792	1117.2	451.8	736.9
RuO <sub>2</sub>	4.00	-471.0	1.220	805.6	581.4	1150.9
RuO₄	8.00	-1349.8	1.749	1298.9	1074.4	1563.7
AgO	2.00	-389.3	2.018	432.7	636.7	1692.0
Ag <sub>2</sub> O	1.00	-182.1	1.888	231.3	364.4	1271.5
$Ag_2O_3$	3.00	-624.0	2.156	609.7	863.0	-
$La_2O_3$	3.00	289.7	-1.001	493.6	-347.0	-837.0
CeO <sub>2</sub>	4.00	273.6	-0.709	622.9	-287.8	-640.5
$Ce_2O_3$	3.00	289.9	-1.001	489.9	-345.4	-825.2
WO <sub>2</sub>	4.00	-217.1	0.563	496.7	195.9	525.4
WO <sub>3</sub>	6.00	-362.5	0.626	693.6	272.3	577.4
Li <sub>2</sub> O <sub>2</sub>	1.00	61.6	-0.639	1168.3	-372.7	-537.0
$K_2O_2$	1.00	-11.3	0.117	486.4	40.2	63.5
$Ag_2O_2$	1.00	-237.7	2.463	216.4	449.5	1636.0
LiO <sub>2</sub>	1.00	-	-	688.3	-	-
KO <sub>2</sub>	1.00	21.0	-0.218	377.0	-62.0	-103.0
	1.00	385.7	-3.998	371.8	-1127.1	-2206.8
S						
BeS	2.00	-116.8	0.605	1304.9	372.7	500.6
$B_2S_3$	3.00	-400.9	1.385	1364.8	870.8	1053.0
Li <sub>2</sub> S	1.00	44.6	-0.462	1166.6	-269.5	-343.4
MgS	2.00	-8.0	0.041	950.9	21.7	32.5

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$AI_2S_3$	3.00	-204.7	0.707	1070.9	394.7	525.2
SiS <sub>2</sub>	4.00	-487.0	1.262	1162.5	734.5	966.2
$P_4S_3$	1.50	-222.6	1.538	730.6	690.8	986.9
$P_4S_7$	3.50	-537.4	1.591	1077.1	890.9	1216.3
$P_2S_5$	5.00	-804.9	1.668	1205.8	988.9	1290.5
K <sub>2</sub> S	1.00	7.1	-0.074	486.1	-25.2	-35.6
CaS	2.00	127.6	-0.661	743.0	-300.1	-471.0
$Sc_2S_3$	3.00	-	-	864.1	-	-
TiS <sub>2</sub>	4.00	-297.4	0.771	957.2	405.1	645.3
TiS	2.00	-79.7	0.413	670.6	175.9	324.9
Ti <sub>2</sub> S <sub>3</sub>	3.00	-	-	837.8	-	-
VS	2.00	-	-	645.8	-	-
$V_2S_3$	3.00	-	-	811.8	-	-
VS <sub>2</sub>	4.00	-	-	931.6	-	-
$V_2S_5$	5.00	-	-	1022.1	-	-
CrS	2.00	-175.4	0.909	637.7	374.6	-
$Cr_2S_3$	3.00	-440.8	1.523	803.3	724.2	1253.8
MnS	2.00	-131.4	0.681	616.1	274.5	506.8
$MnS_2$	4.00	-474.6	1.230	900.4	624.7	-
FeS	2.00	-249.4	1.292	609.7	517.4	1048.0
FeS <sub>2(marcasite</sub>						
)	4.00	-543.5	1.408	893.6	712.3	1271.8
FeS <sub>2(pyrite)</sub>	4.00	-532.7	1.380	893.6	698.2	1246.6
CoS	2.00	-	-	589.0	-	-
$Co_2S_3$	3.00	-	-	751.2	-	-
CoS <sub>2</sub>	4.00	-554.0	1.435	871.1	715.6	1246.7
$Co_3S_4$	2.67	-304.1	1.182	702.8	518.3	-
NiS	2.00	-270.3	1.401	590.6	549.1	1175.0
$Ni_3S_4$	2.67	-369.1	1.435	704.5	629.9	1213.7
$Ni_3S_2$	1.33	-163.2	1.269	446.3	409.4	-
NiS <sub>2</sub>	4.00	-574.9	1.490	872.8	743.5	-
CuS	2.00	-296.2	1.535	560.6	581.1	1219.1
Cu <sub>2</sub> S	1.00	-131.8	1.366	336.8	356.9	965.7
ZnS	2.00	-148.5	0.770	550.0	287.6	577.9
Ga <sub>2</sub> S <sub>3</sub>	3.00	-271.8	0.939	682.4	404.3	733.5
GaS	2.00	-145.1	0.752	526.6	272.8	546.4
GeS	2.00	-278.3	1.442	511.9	513.0	1059.9

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GeS <sub>2</sub>	4.00	-545.0	1.412	783.8	661.9	1079.5
$As_2S_3$	3.00	-440.4	1.521	653.6	637.2	1147.0
Se <sub>2</sub> S <sub>6</sub>	6.00	-	-	918.1	-	-
Se <sub>4</sub> S <sub>4</sub>	2.00	-	-	482.8	-	-
RuS <sub>2</sub>	4.00	-	-	648.9	-	-
$Y_2S_3$	3.00	-	-	586.9	-	-
ZrS <sub>2</sub>	4.00	-129.6	0.336	690.1	145.5	266.7
NbS <sub>2</sub>	4.00	-	-	682.7	-	-
MoS <sub>2</sub>	4.00	-473.7	1.227	669.7	522.1	1040.7
$Mo_2S_3$	3.00	-385.4	1.331	558.2	502.6	1121.3
Ag <sub>2</sub> S <sub>(argentite)</sub>	1.00	-154.6	1.602	216.3	292.3	1051.3
LaS	2.00	101.7	-0.527	313.5	-130.2	-362.7
$La_2S_3$	3.00	79.5	-0.275	430.0	-86.2	-202.0
CeS	2.00	101.7	-0.527	311.3	-129.5	-368.9
WS <sub>2</sub>	4.00	-449.7	1.165	432.3	367.5	980.3
$Li_2S_2$	1.00	-	-	376.6	-	-
N						
Be <sub>3</sub> N <sub>2</sub>	2.00	240.1	-1.244	2921.1	-1036.9	-1231.3
BN	3.00	322.2	-1.113	3239.7	-954.2	-1084.9
(CN)	3.00	-	-	3090.4	-	-
Li <sub>3</sub> N	1.00	74.1	-0.768	2308.5	-595.1	-
$Mg_3N_2$	2.00	196.1	-1.016	1593.3	-684.3	-910.9
AIN	3.00	380.8	-1.315	1961.6	-961.9	-1263.7
Si <sub>3</sub> N <sub>4</sub>	4.00	339.2	-0.879	2292.6	-679.3	-860.1
K₃N	1.00	-	-	612.4	-	-
Ca <sub>3</sub> N <sub>2</sub>	2.00	185.2	-0.960	1084.7	-539.2	-780.4
ScN	3.00	377.5	-1.304	1363.6	-819.7	-
TiN	3.00	337.6	-1.166	1299.5	-716.6	-1130.0
VN	3.00	284.8	-0.984	1238.0	-590.8	-968.5
CrN	3.00	186.8	-0.645	1218.2	-384.4	-630.4
Cr <sub>2</sub> N	1.50	98.0	-0.677	681.4	-291.1	-615.4
Mn₄N	0.75	49.7	-0.687	344.0	-182.4	-
$Mn_5N_2$	1.20	68.5	-0.592	531.2	-215.9	-
Fe <sub>2</sub> N	1.50	-	-	639.7	-	-
	·					
Fe4N	0.75	24.4	-0.337	338.7	-88.4	-

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NiN	3.00	-		-	1106.0	-		-
Cu₃N	1.00	-		-	392.9	-		-
Zn <sub>3</sub> N <sub>2</sub>	2.00		49.4	-0.256	717.3		-113.7	-230.9
GaN	3.00		171.5	-0.592	960.3		-312.0	-561.6
Ge <sub>3</sub> N <sub>4</sub>	4.00	-		-	1174.0	-		-
As <sub>3</sub> N <sub>5</sub>	5.00	-		-	1363.7	I		-
SeN <sub>2</sub>	6.00	-		-	1503.3	-		-
YN	3.00		362.3	-1.252	781.3		-585.6	-1124.8
ZrN	3.00		430.5	-1.487	764.1		-686.4	-1391.3
NbN	3.00		299.7	-1.035	752.1		-473.3	-994.2
MoN	3.00	-		-	731.3	-		-
RuN	3.00	-		-	698.7	-		-
Ag₃N	1.00	-		-	238.2	-		-
LaN	3.00		364.8	-1.260	525.8		-456.7	-1080.0
CeN	3.00		388.7	-1.343	521.7		-484.0	-1191.3
WN <sub>2</sub>	6.00	-		-	759.1	-		-
LiN <sub>3</sub>	1.00	-		-	201.5	-		-
KN <sub>3</sub>	1.00	-		-	162.3	I		-
AgN₃	1.00	-		-	114.6	-		-
•								
Р								
P Be <sub>3</sub> P <sub>2</sub>	2.00	-		-	1807.2	-		-
P Be <sub>3</sub> P <sub>2</sub> BP	2.00 3.00	-		-	1807.2 1924.2	-		-
P Be <sub>3</sub> P <sub>2</sub> BP Li <sub>3</sub> P	2.00 3.00 1.00	- - -		-	1807.2 1924.2 1552.3	-		
P Be <sub>3</sub> P <sub>2</sub> BP Li <sub>3</sub> P Mg <sub>3</sub> P <sub>2</sub>	2.00 3.00 1.00 2.00	- - -		- - -	1807.2 1924.2 1552.3 1192.4			- - -
$\begin{array}{c} P \\ Be_3P_2 \\ BP \\ Li_3P \\ Mg_3P_2 \\ AIP \end{array}$	2.00 3.00 1.00 2.00 3.00	- - - -		- - - -	1807.2 1924.2 1552.3 1192.4 1387.3			- - - -
$\begin{array}{c} P \\ Be_3P_2 \\ BP \\ Li_3P \\ Mg_3P_2 \\ AIP \\ Si_3P_4 \end{array}$	2.00 3.00 1.00 2.00 3.00 4.00	- - - - -		- - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1	- - - -		- - - - -
$\begin{array}{c} P \\ Be_3P_2 \\ BP \\ Li_3P \\ Mg_3P_2 \\ AIP \\ Si_3P_4 \\ K_3P \end{array}$	2.00 3.00 1.00 2.00 3.00 4.00 1.00	- - - - - - -		- - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3	- - - -		- - - - -
$\begin{array}{c} P \\ Be_3P_2 \\ BP \\ Li_3P \\ Mg_3P_2 \\ AIP \\ Si_3P_4 \\ K_3P \\ Ca_3P_2 \end{array}$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00	- - - - - -		- - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7			- - - - - -
$\begin{array}{c} P \\ Be_3P_2 \\ BP \\ Li_3P \\ Mg_3P_2 \\ AIP \\ Si_3P_4 \\ K_3P \\ Ca_3P_2 \\ ScP \end{array}$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00	- - - - - - - - - - - - -		- - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9	- - - - - - -		- - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00	- - - - - - - - - - - -		- - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8	- - - - - - - - -		- - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00	- - - - - - - - - - - - -		- - - - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6	- - - - - - - - - - - - -		- - - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00	- - - - - - - - - - - - - - - - -		- - - - - - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1			- - - - - - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$ $MnP$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00 3.00 3.00	- - - - - - - - - - - - - - - - -		- - - - - - - - - - - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1 935.9	- - - - - - - - - - - - - - - - - - -		- - - - - - - - - - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$ $MnP$ $Mn_2P$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00 3.00 1.50	- - - - - - - - - - - - - - - - - - -		- - - - - - - - - - - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1 935.9 570.9			- - - - - - - - - - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$ $MnP$ $Mn_2P$ $FeP$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00 3.00 1.50 3.00	- - - - - - - - - - - - - - - - - - -			1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1 935.9 570.9 926.1			- - - - - - - - - - - - - - - - - - -
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$ $Mn_2P$ $FeP$ $Fe_2P$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00 3.00 1.50 3.00 1.50			- - - - - - - - - - - - - - - - - - -	1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1 935.9 570.9 926.1 563.6			
P $Be_3P_2$ $BP$ $Li_3P$ $Mg_3P_2$ $AIP$ $Si_3P_4$ $K_3P$ $Ca_3P_2$ $ScP$ $TiP$ $(VP)$ $CrP$ $MnP$ $Mn_2P$ $FeP$ $Fe_3P$	2.00 3.00 1.00 2.00 3.00 4.00 1.00 2.00 3.00 3.00 3.00 3.00 3.00 1.50 3.00 1.50 1.00				1807.2 1924.2 1552.3 1192.4 1387.3 1545.1 542.3 882.7 1058.9 1019.8 981.6 969.1 935.9 570.9 926.1 563.6 405.0			

	1	I	I		I	1
NI <sub>2</sub> P	1.50	-	-	542.0	-	-
Ni <sub>3</sub> P	1.00	-	-	388.3	-	-
Ni <sub>5</sub> P <sub>2</sub>	1.20	-	-	452.5	-	-
CuP <sub>2</sub>	6.00	-	-	1281.4	-	-
Cu₃P	1.00	-	-	362.8	-	-
Zn <sub>3</sub> P <sub>2</sub>	2.00	-	-	623.0	-	-
GaP	3.00	-	-	798.5	-	-
GeP	3.00	-	-	776.0	-	-
Ge <sub>3</sub> P <sub>4</sub>	4.00	-	-	940.9	-	-
As <sub>3</sub> P <sub>5</sub>	5.00	-	-	1059.0	-	-
Se(P) <sub>2</sub>	6.00	-	-	1141.2	-	-
YP	3.00	-	-	670.7	-	-
ZrP <sub>2</sub>	6.00	-	-	1049.9	-	-
NbP	3.00	-	-	649.0	-	-
MoP	3.00	-	-	633.5	-	-
RuP	3.00	-	-	608.9	-	-
AgP <sub>2</sub>	6.00	-	-	947.0	-	-
AgP <sub>3</sub>	9.00	-	-	1201.3	-	-
LaP	3.00	-	-	473.3	-	-
Ce <sub>3</sub> P <sub>4</sub>	4.00	-	-	590.9	-	-
WP <sub>2</sub>	6.00	-	-	654.3	-	-
Cl						
BeCl <sub>2</sub>	2.00	-322.6	1.672	670.7	711.8	1001.6
PCI <sub>5</sub>	5.00	-1615.6	3.349	643.5	1388.6	2061.8
LiCl	1.00	0.3	-0.003	632.2	-1.3	-1.9
MgCl <sub>2</sub>	2.00	-176.4	0.914	563.0	347.0	554.6
AICI <sub>3</sub>	3.00	-523.5	1.809	603.0	718.8	1164.6
KCI	1.00	24.4	-0.253	359.5	-69.5	-110.7
CaCl <sub>2</sub>	2.00	-19.4	0.101	483.0	34.3	54.4
ScCl <sub>3</sub>	3.00	-326.4	1.128	531.4	411.6	675.9
TiCl <sub>2</sub>	2.00	-303.8	1.574	451.3	512.2	988.8
TiCl <sub>3</sub>	3.00	-498.8	1.723	521.3	620.8	1069.8
VCl <sub>2</sub>	2.00	-362.2	1.877	439.9	599.5	1181.9
VCl <sub>3</sub>	3.00	-641.0	2.215	511.2	786.9	1441.3
CrCl <sub>2</sub>	2.00	-412.2	2.136	436.1	678.0	1271.2
CrCl <sub>3</sub>	3.00	-666.2	2.302	507.7	814.1	1440.4

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MnCl <sub>2</sub>	2.00	-327.7	1.698	425.9	529.8	1015.1
FeCl <sub>2</sub>	2.00	-465.9	2.414	422.9	749.2	1481.0
FeCl <sub>3</sub>	3.00	-818.3	2.827	495.7	983.3	1789.3
CoCl <sub>2</sub>	2.00	-498.4	2.583	412.8	787.4	1609.0
NiCl <sub>2</sub>	2.00	-509.2	2.639	413.6	805.6	1677.4
CuCl	1.00	-264.2	2.738	270.7	601.6	1541.3
CuCl <sub>2</sub>	2.00	-592.5	3.070	398.7	912.2	1892.9
ZnCl <sub>2</sub>	2.00	-398.8	2.067	393.3	607.7	1174.9
GaCl₃	3.00	-697.5	2.410	456.6	790.7	1360.7
AsCl <sub>5</sub>	5.00	-	-	531.4	-	-
SeCl <sub>4</sub>	4.00	-	-	485.6	-	-
YCl <sub>3</sub>	3.00	-224.6	0.776	411.8	236.1	427.5
ZrCl <sub>2</sub>	2.00	-382.2	1.981	330.6	510.1	1075.6
ZrCl <sub>3</sub>	3.00	-506.3	1.749	406.9	527.6	1035.0
ZrCl <sub>4</sub>	4.00	-646.5	1.675	460.0	552.6	1008.7
NbCl <sub>5</sub>	5.00	-1237.3	2.565	496.0	892.4	1593.5
NbCl <sub>4</sub>	4.00	-930.9	2.412	456.7	791.5	1537.7
MoCl <sub>4</sub>	4.00	-1134.4	2.939	450.9	955.7	-
MoCl <sub>5</sub>	5.00	-1497.5	3.104	490.5	1071.7	1964.5
MoCl <sub>6</sub>	6.00	-1913.6	3.306	521.0	1190.2	-
RuCl <sub>3</sub>	3.00	-992.6	3.429	387.6	997.5	1997.7
AgCl	1.00	-274.3	2.843	187.0	458.1	1540.0
SnCl <sub>2</sub>	2.00	-482.0	2.498	282.7	568.3	1403.3
LaCl₃	3.00	-	-	327.8	-	-
CeCl <sub>3</sub>	3.00	-167.5	0.579	326.2	147.5	349.3
WCl <sub>2</sub>	2.00	-548.2	2.841	210.4	506.4	1616.1
WCl <sub>4</sub>	4.00	-1176.4	3.048	329.2	782.5	1977.0
WCl <sub>5</sub>	5.00	-1518.5	3.148	371.1	886.0	1993.7
WCl <sub>6</sub>	6.00	-1848.6	3.193	405.5	960.7	2014.8
BO <sub>2</sub> -						
LiBO <sub>2</sub>	1.00	56.7	-0.588	538.7	-216.6	-
Mg(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	487.6	-	-
AI (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	517.4	-	-
KBO <sub>2</sub>	1.00	-	-	327.2	-	-
Sc(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	463.7	-	-

1	1	1	1	I	I	1
Ti (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	489.3	-	-
V(BO <sub>2</sub> ) <sub>5</sub>	5.00	-	-	505.7	-	-
Cr(BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	520.6	-	-
Mn (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	438.5	-	-
Fe (BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	378.9	-	-
Co (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	429.1	-	-
Ni(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	371.4	-	-
Cu(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	359.3	-	-
Zn(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	355.0	-	-
Ga(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	405.8	-	-
Ge(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	439.6	-	-
Y (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	369.9	-	-
Zr (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	408.5	-	-
Nb BO <sub>2</sub>	1.00	-	-	197.5	-	-
MoBO <sub>2</sub>	1.00	-	-	193.2	-	-
Ru(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	287.1	-	-
Ag BO <sub>2</sub>	1.00	-	-	177.9	-	-
La (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	300.8	-	-
Ce(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	344.3	-	-
W (BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	364.9	-	-
Be <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-	-	1111.8	-	-
Ca <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-	-	676.1	-	-
CO3 <sup>2-</sup>						
Be CO <sub>3</sub>	2.00	-	-	776.6	-	-
B <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-	-	797.5	-	-
Li <sub>2</sub> CO <sub>3</sub>	1.00	43.9	-0.455	725.4	-203.3	-
Mg CO <sub>3</sub>	2.00	-32.3	0.167	635.8	68.9	119.0
Al <sub>2</sub> ( CO <sub>3</sub> ) <sub>3</sub>	3.00	-	-	687.2	-	-
K <sub>2</sub> CO <sub>3</sub>	1.00	9.5	-0.099	387.8	-28.8	-49.2
CaCO <sub>3</sub>	2.00	84.4	-0.437	535.6	-160.5	-280.4
Sc 2(CO3)3	3.00	-	-	595.7	-	-
Ti (CO <sub>3</sub> ) <sub>2</sub>	4.00	-	-	638.6	-	-
V <sub>2</sub> (CO <sub>3</sub> ) <sub>5</sub>	5.00	-	-	666.8	-	-
Cr (CO <sub>3</sub> ) <sub>3</sub>	6.00	-	-	693.1	-	-

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MnCO32.00 $-227.7$ 1.180466.3333.0880.1Fe2(CO3)33.00551.2FeCO32.00 $-377.7$ 1.957450.76636.21352.7Co CO32.00-407.62.112450.76686.61495.3Ni CO32.00-426.52.210451.6719.41591.5CuCO32.00433.8Zn CO32.00-312.81.621427.5507.01148.2Ga2(CO3)33.00556.5Y2(CO3)33.00354.4Y2(CO3)33.00354.4Nb2 CO31.00212.8Nb2 CO31.00214.8Nb2 CO31.00214.8Nb2 CO31.00351.2Ru2 CO31.00351.2Ru2 CO33.00351.2Ru2 CO33.00349.4524.61819.2La 2(CO3)33.00349.4524.61819.2La 2(CO3)33.00349.4524.61819.2La 2(CO3)33.00349.4524.6399.7N(CO3)33.00349.		i.	i.						i.			
Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         551.2         -         -           FeCO <sub>3</sub> 2.00         -377.7         1.957         450.7         636.2         1352.7           Co CO <sub>3</sub> 2.00         -407.6         2.112         450.7         686.6         1495.3           Ni CO <sub>3</sub> 2.00         -426.5         2.210         451.6         719.4         1591.5           CuCO <sub>3</sub> 2.00         -426.5         2.210         450.7         507.0         1148.2           Ga(CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         503.4         -         -           Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         556.5         -         -         -           Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         354.4         -         -         -           Mo <sub>2</sub> CO <sub>3</sub> 1.00         -         -         212.8         -         -         -           Mo <sub>2</sub> CO <sub>3</sub> 1.00         -         -         204.5         -         -         -           Mo <sub>2</sub> CO <sub>3</sub> 3.00         -         -         351.2         -         -         -           Mo	MnCO <sub>3</sub>	2.00		-227.7		1.180	4	66.3		393.0		806.1
FeCO <sub>3</sub> 2.00        377.7         1.957         450.7         636.2         1352.7           Co CO <sub>3</sub> 2.00         -407.6         2.112         450.7         686.6         1495.3           Ni CO <sub>3</sub> 2.00         -426.5         2.210         451.6         719.4         1591.5           CuCO <sub>3</sub> 2.00         -312.8         1.621         427.5         507.0         1148.2           Ga <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         556.5         -         -           Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         3554.4         -         -           Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 1.00         -         -         218.1         -         -           Nb <sub>2</sub> CO <sub>3</sub> 1.00         -         -         218.1         -         -           Nu <sub>2</sub> CO <sub>3</sub> 1.00         -         -         218.1         -         -           Nu <sub>2</sub> CO <sub>3</sub> 1.00         -303.8         3.149         194.4         524.6         1819.2           La <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub> 3.00         -         -         3351.2         -         -           M <sub>2</sub> CO <sub>3</sub> 1.00         -         - </td <td>Fe<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub></td> <td>3.00</td> <td>-</td> <td></td> <td>-</td> <td></td> <td>5</td> <td>51.2</td> <td>-</td> <td></td> <td>-</td> <td></td>	Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		5	51.2	-		-	
Co $CO_3$ 2.00-407.62.112450.7686.61495.3Ni $CO_3$ 2.00-426.52.210451.6719.41591.5Cu $CO_3$ 2.00-312.81.621427.5507.01148.2Ga_2( $CO_3)_3$ 3.00556.5Ge( $CO_3)_2$ 4.00556.5Y_2( $CO_3)_3$ 3.00354.4Tr $C_3$ 1.00218.1Mo_2 $CO_3$ 1.00218.1Nb_2 CO_31.00204.5Ru_2 $CO_3$ 1.00204.5Mo_2 $CO_3$ 1.00204.5Ru_2 $CO_3$ 1.00204.5Mo_2 $CO_3$ 1.00204.5Ru_2 $CO_3$ 3.00204.5Ru_2 $CO_3$ 3.00351.2Ru_2 $CO_3$ 3.00349.4Ru_2 $CO_3$ 3.00349.4Ru_2 $CO_3$ 3.00Ru_2 $CO_3$ 3.00Ru_2 $CO_3$ 3.00Ru_2 $CO_3$ 3.00<	FeCO <sub>3</sub>	2.00		-377.7		1.957	4	50.7		636.2		1352.7
Ni CO32.00-426.52.210451.6719.41591.5CuCO32.00433.8Zn CO32.00-312.81.621427.5507.01148.2Ga2(CO3)33.00556.5Ge(CO3)24.00556.5Y2(CO3)33.00354.4TrO31.00218.1Mo2CO31.002212.8Ru2CO31.00204.5Ag2 CO31.00351.4Mo2CO31.00351.2Ru2CO33.00351.2Ag2 CO31.00349.4MQCO3)33.00351.2Ru2CO33.00349.4Mg2 CO3)33.00349.4Mg2 CO3)33.00349.4Mg2 CO3)33.00349.4Mg2 CO3)33.00Ha 2 CO3)33.00Ha 2 CO3)33.00Ha 2 CO3)3<	Co CO <sub>3</sub>	2.00		-407.6		2.112	4	50.7		686.6		1495.3
CuCO32.00-433.8Zn CO32.00-312.81.621427.5507.01148.2Ga2(CO3)33.00503.4Ge(CO3)24.00556.5Y2(CO3)33.00449.4TCO32.00354.4Nb2 CO31.00218.1Mo2CO31.00204.5Ru2CO31.00204.5Ag2 CO31.00351.2Ce2(CO3)33.00349.4V (CO3)36.00349.4PQ31.00349.4W (CO3)36.00441.9PQ431.00441.9H=686.61.00Mg3(PO4)22.00AlPO43.00-170.90.590659.3248.6399.7K3 PO41.00AlPO43.00AlPO43.00AlPO43.00AlPO43.00	Ni CO <sub>3</sub>	2.00		-426.5		2.210	4	51.6		719.4		1591.5
Zn CO32.00 $-312.8$ 1.621427.5507.01148.2Ga2(CO3)33.00 $  503.4$ $ -$ Ge(CO3)24.00 $  556.5$ $ -$ Y2(CO3)33.00 $  449.4$ $ -$ Y2(CO3)33.00 $  354.4$ $ -$ Nb2 CO31.00 $  218.1$ $ -$ Mo2CO31.00 $  204.5$ $ -$ Ag2 CO31.00 $  204.5$ $ -$ Ag2 CO31.00 $  351.2$ $ -$ Ce2(CO3)33.00 $  349.4$ $ -$ V (CO3)36.00 $  349.4$ $ -$ PO431.00 $  349.4$ $ -$ H=698.6 $M_1$ $    -$ Mg(PO4)22.00 $    -$ Al PO4 $3.00$ $    -$ Mg3(PO4)22.00 $    -$ Al PO4 $3.00$ $    -$ <td< td=""><td>CuCO<sub>3</sub></td><td>2.00</td><td>-</td><td></td><td>-</td><td></td><td>4</td><td>33.8</td><td>-</td><td></td><td>-</td><td></td></td<>	CuCO <sub>3</sub>	2.00	-		-		4	33.8	-		-	
Ga2(CO3)33.00503.4Ge(CO3)24.00556.5Y2(CO3)33.00449.4Y2(CO3)33.00354.4Nb2 CO31.00218.1Mo2CO31.00212.8Ru2CO31.00204.5Ag2 CO31.00351.2Ag2 CO33.00349.4Ce2(CO3)33.00349.4V (CO3)36.00349.4PQ4 <sup>3-0</sup> 4.00349.4H=688.6 bi mo1 <sup>1</sup> 1.00741.1H=688.6 bi mo1 <sup>1</sup> 1.00363.43.767694.41639.12595.3Mg3(PO4)22.00170.90.590669.3248.6399.7K3 PO41.00170.90.590659.3248.6399.7K3 PO43.00170.90.590659.3248.6399.7K3 PO43.00170.90.590518.4-190.3-353.9Sc PO43.00102.4-0.530518.4-190.3-353.9Sc PO43.00102.4-0.530.1516.4Ya (PO4)55.00 <t< td=""><td>Zn CO<sub>3</sub></td><td>2.00</td><td></td><td>-312.8</td><td></td><td>1.621</td><td>4</td><td>27.5</td><td></td><td>507.0</td><td></td><td>1148.2</td></t<>	Zn CO <sub>3</sub>	2.00		-312.8		1.621	4	27.5		507.0		1148.2
Ge(CO3)24.00556.5 $Y_2(CO_3)_3$ 3.00449.4 $ZrCO_3$ 2.00354.4 $Nb_2 CO_3$ 1.00218.1 $Mo_2CO_3$ 1.00212.8 $Ru_2CO_3$ 1.00204.5 $Ag_2 CO_3$ 1.00351.2 $Ag_2 CO_3$ 3.00351.2 $La_2(CO_3)_3$ 3.00349.4 $V(CO_3)_3$ 6.00349.4 $PQ_4^{3-0}$ 441.9 $W(CO_3)_3$ 6.00441.9 $PQ_4^{3-0}$ 441.9 $W(CO_3)_3$ 6.00 $M_1 = 698.6 M_1$ $M_1 = 698.6 M_2$ $M_1 = 0^{-1}$ 1.00 $M_2 = 0^{-1}$ $M_3 = 0^{-1}$ $M_3 = 0^{-1}$ $M_1 = 0^{-1}$ $M_3 = 0^{-1}$ $M_3 = 0^{-1}$ <t< td=""><td><math>Ga_2(CO_3)_3</math></td><td>3.00</td><td>-</td><td></td><td>-</td><td></td><td>5</td><td>03.4</td><td>-</td><td></td><td>-</td><td></td></t<>	$Ga_2(CO_3)_3$	3.00	-		-		5	03.4	-		-	
$Y_2(CO_3)_3$ $3.00$ $  449.4$ $  ZrCO_3$ $2.00$ $  354.4$ $  Nb_2 CO_3$ $1.00$ $  218.1$ $  Mo_2CO_3$ $1.00$ $  212.8$ $  Ru_2CO_3$ $1.00$ $  204.5$ $  Ag_2 CO_3$ $1.00$ $  351.2$ $  Ag_2 CO_3$ $3.00$ $  351.2$ $  La_2(CO_3)_3$ $3.00$ $  349.4$ $  V(CO_3)_3$ $6.00$ $  349.4$ $  W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $6.00$ $     W(CO_3)_3$ $0.00$ $     W(CO_3)_3$ $1.00$ $     W(CO_3)_3$ <	Ge(CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-		5	56.5	-		-	
ZrCO32.00354.4Nb2 CO31.00218.1Mo2CO31.00212.8Ru2CO31.00204.5Ag2 CO31.00204.5Ag2 CO31.00351.2La 2(CO3)33.00351.2Ce2(CO3)33.00349.4W (CO3)36.00441.9PQ43741.1Be3(PO4)22.00741.1H=696.6 kJ mol <sup>-1</sup> 1.00-363.43.767694.41639.12595.3Mg3(PO4)22.00-113.00.067611.827.0411.3AI PO43.00-170.90.590659.3248.6399.7K <sub>3</sub> PO41.00378.8Ca3(PO4)22.00102.4-0.530518.4-190.3-353.9Sc PO43.00574.6Ti $_3(PO4)4$ 4.00640.5 $V_3$ (PO4)55.00536.4 $V_3$ (PO4)43.00536.4 $V_3$ (PO4)55.00536.4 </td <td>Y<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub></td> <td>3.00</td> <td>-</td> <td></td> <td>-</td> <td></td> <td>4</td> <td>49.4</td> <td>-</td> <td></td> <td>-</td> <td></td>	Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		4	49.4	-		-	
Nb2 CO31.00218.1Mo2CO31.00212.8Ru2CO31.00204.5Ag2 CO31.00-303.83.149194.4524.61819.2La 2(CO3)33.00351.2Ce2(CO3)33.00349.4W (CO3)36.00441.9PQ43*6.00741.1Be3(PO4)22.00741.1Li3PO4( $^{\wedge}$ 741.1mo1 -1.00-363.43.767694.41639.12595.3Mg3(PO4)22.00-170.90.590659.3248.6399.7K3 PO41.00-170.90.590518.4-190.3-353.9Sc PO43.00378.8Ti 3(PO4)44.00574.6V3 (PO4)55.00640.5Ti 3(PO4)43.00536.4Mn PO43.00536.4Mn3(PO4)22.00453.3Fe PO43.00453.3Fe PO43.00453.5	ZrCO <sub>3</sub>	2.00	-		-		3	54.4	-		-	
$Mo_2CO_3$ 1.00212.8 $Ru_2CO_3$ 1.00204.5 $Ag_2 CO_3$ 1.00303.83.149194.4524.61819.2 $La_2(CO_3)_3$ 3.00351.2 $Ce_2(CO_3)_3$ 3.00349.4 $W(CO_3)_3$ 6.00441.9 $W(CO_3)_3$ 6.00441.9 $PO_4^{3-}$ 6.00741.1 $PO_4^{3-}$ 7.00741.1 $Be_3(PO_4)_2$ 2.00741.1 $I_{13}PO_4( \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	Nb <sub>2</sub> CO <sub>3</sub>	1.00	-		-		2	18.1	-		-	
Ru2CO31.00204.5Ag2 CO31.00-303.83.149194.4524.61819.2La 2(CO3)33.00351.2Ce2(CO3)33.00349.4W (CO3)36.00441.9PQ436.00441.9Be3(PO4)22.00741.1Li3PO4( $^{-6}$ 741.1H=698.6M31.00-363.43.767694.41639.12595.3Mg3(PO4)22.00-13.00.067611.827.041.3AI PO43.00-170.90.590659.3248.6399.7K3 PO41.00378.8Ca3(PO4)22.00102.4-0.530518.4-190.3-353.9Sc PO43.00640.5Ti 3(PO4)44.00640.5V3 (PO4)55.00536.4Mn PO43.00536.4Mn PO43.00536.4Fe PO43.00435.3Fe PO43.00438.5	Mo <sub>2</sub> CO <sub>3</sub>	1.00	-		-		2	12.8	-		-	
Ag2 CO31.00303.83.149194.4524.61819.2La 2(CO3)33.00351.2Ce2(CO3)33.00349.4W (CO3)36.00441.9PO43*Image: Composition of the	Ru <sub>2</sub> CO <sub>3</sub>	1.00	-		-		2	04.5	-		-	
La $_2(CO_3)_3$ 3.00351.2 $Ce_2(CO_3)_3$ 3.00349.4W (CO_3)_36.00441.9 $PO_4^{3-}$ 441.9 $PO_4^{3-}$ 741.1 $Be_3(PO_4)_2$ 2.00741.1 $Li_3PO_4( \ A \ M^{-1})^{-1}$ 1.00-363.43.767694.41639.12595.3 $Mg_3(PO_4)_2$ 2.00-13.00.067611.827.041.3Al PO_43.00-170.90.590659.3248.6399.7 $K_3 PO_4$ 1.00378.8 $Ca_3(PO_4)_2$ 2.00102.4-0.530518.4-190.3-353.9Sc PO_43.006414.4 $V_3 (PO_4)_5$ 5.00640.5 $V_3 (PO_4)_5$ 5.00536.4 $Mn PO_4$ 3.00536.4 $Mn_3(PO_4)_2$ 2.00453.3 $Fe PO_4$ 3.00438.5	Ag <sub>2</sub> CO <sub>3</sub>	1.00		-303.8		3.149	1	94.4		524.6		1819.2
$\begin{array}{c c c c c c c } \hline Ce_2(CO_3)_3 & 3.00 & - & - & 349.4 & - & - & - & \\ \hline W (CO_3)_3 & 6.00 & - & - & 441.9 & - & - & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & V & \\ \hline PO_4^{3-} & V & V & \\ \hline PO_4^{3-} & V & V & V & \\ \hline PO_4^{3-} & V & \\ \hline PO_4^{3-} & V $	La 2(CO3)3	3.00	-		-		3	51.2	-		-	
W (CO3)36.00441.9PO43-3-KKKKKKKKBe3(PO4)22.00741.1Li3PO4( $^{A}$ K741.1H=-698.6KJKJ694.41639.12595.3Mg3(PO4)22.0013.00.067611.827.041.3Al PO43.00-170.90.590659.3248.6399.7K3 PO41.00378.8Ca3(PO4)22.00102.4-0.530518.4-190.3-353.9Sc PO43.00614.4Ti $_3(PO4)_4$ 4.00640.5V3 (PO4)55.00640.5Mn PO43.00536.4Mn3(PO4)22.00453.3Fe PO43.00438.5	$Ce_2(CO_3)_3$	3.00	-		-		3	49.4	-		-	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	W (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-		4	41.9	-		-	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	PO4 <sup>3-</sup>											
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$Be_3(PO_4)_2$	2.00	-		-		7	41.1	-		-	
$\begin{array}{c c c c c c c c c c c c c c c c c c c $												
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	H=-698.6 kJ -1	1 00		-363 4		3 767	6	94 4		1639 1		2595.3
Mg3(1 $O_{4/2}$ 2.0010.00.0000.1000.1100.110AI PO43.00-170.90.590659.3248.6399.7K3 PO41.00378.8Ca3(PO4)22.00102.4-0.530518.4-190.3-353.9Sc PO43.00574.6Ti $_3(PO_4)_4$ 4.00614.4V3 (PO4)55.00640.5CrPO43.00536.4Mn PO43.00453.3Fe PO43.00-601.32.077533.1759.91350.7Co3(PO4)22.00438.5	$Ma_2(PO_4)_2$	2 00		-13.0		0.067	6	11.8		27.0		41.3
$K_3 PO_4$ 1.00378.8 $Ca_3(PO_4)_2$ 2.00102.4-0.530518.4-190.3-353.9 $Sc PO_4$ 3.00574.6 $Ti_3(PO_4)_4$ 4.00614.4 $V_3 (PO_4)_5$ 5.00640.5 $CrPO_4$ 3.00536.4 $Mn PO_4$ 3.00453.3 $Mn_3(PO_4)_2$ 2.00438.5 $Fe PO_4$ 3.00438.5		3.00		-170.9		0.590	6	59.3		248.6		399.7
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	K₂ PO₄	1 00	_	110.0	_	0.000	3	78.8	_	210.0	_	000.1
Sc PO4 $3.00$ $574.6$ Ti $_3(PO_4)_4$ $4.00$ $614.4$ $V_3$ (PO4)_5 $5.00$ $640.5$ CrPO4 $3.00$ $547.1$ Mn PO4 $3.00$ $536.4$ Mn_3(PO4)_2 $2.00$ $453.3$ Fe PO4 $3.00$ $438.5$	$Ca_2(PO_4)_2$	2 00		102 4		-0 530	5	18.4		-190.3		-353.9
Ti $_3(PO_4)_4$ 4.00614.4 $V_3 (PO_4)_5$ 5.00640.5CrPO_43.00547.1Mn PO_43.00536.4Mn_3(PO_4)_22.00453.3Fe PO_43.00-601.32.077533.1759.91350.7Co3(PO_4)_22.00438.5	$Sc PO_4$	3.00	_		_	0.000	5	74 6	_		_	00010
$V_3 (PO_4)_5$ 5.00640.5 $CrPO_4$ 3.00547.1 $Mn PO_4$ 3.00536.4 $Mn_3 (PO_4)_2$ 2.00453.3Fe PO_43.00-601.32.077533.1759.91350.7Co3(PO_4)_22.00438.5		4 00	_		_		6	14 4	_		_	
CrPO4 $3.00$ $547.1$ Mn PO4 $3.00$ $536.4$ Mn <sub>3</sub> (PO4) <sub>2</sub> $2.00$ $453.3$ Fe PO4 $3.00$ -601.3 $2.077$ $533.1$ $759.9$ $1350.7$ Co3(PO4) <sub>2</sub> $2.00$ $438.5$	$V_2 (PO_4)_5$	5.00	_		_		6	40.5	_		_	
Mn PO <sub>4</sub> 3.00       -       -       536.4       -       -         Mn <sub>3</sub> ( PO <sub>4</sub> ) <sub>2</sub> 2.00       -       -       453.3       -       -         Fe PO <sub>4</sub> 3.00       -601.3       2.077       533.1       759.9       1350.7         Co3(PO <sub>4)2</sub> 2.00       -       -       438.5       -       -	CrPO <sub>4</sub>	3.00	-		-		5	47 1	_		-	
Mn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> 2.00         -         -         453.3         -         -           Fe PO <sub>4</sub> 3.00         -601.3         2.077         533.1         759.9         1350.7           Co3(PO <sub>4)2</sub> 2.00         -         -         438.5         -         -	Mn PO₄	3.00	-		-		5	36.4	_		-	
Fe PO <sub>4</sub> 3.00         -601.3         2.077         533.1         759.9         1350.7           Co3(PO <sub>4)2</sub> 2.00         -         -         438.5         -         -	$Mn_3(PO_4)_2$	2.00	-		-		4	53.3	_		-	
Co3(PO <sub>4)2</sub> 2.00 438.5	Fe PO <sub>4</sub>	3.00		-601 3		2.077	5	33.1		759.9		1350 7
	Co3(PO <sub>4)2</sub>	2.00	_		-		4	38.5	_		-	

Supplementa	ry Material (ESI	I) for Energ	gy & Environr	nental	Science
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1	1	i.	1	I.	i.	1
Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-	-	439.3	-	-
Cu <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-	-	422.5	-	-
Zn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-304.7	1.579	416.5	484.5	1063.4
Ga PO <sub>4</sub>	3.00	-	-	488.2	-	-
Ge <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	4.00	-	-	538.0	-	-
Y PO <sub>4</sub>	3.00	-	-	437.3	-	-
Zr <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-	-	346.9	-	-
Nb <sub>3</sub> PO <sub>4</sub>	1.00	-	-	215.2	-	-
Mo <sub>3</sub> PO <sub>4</sub>	1.00	-	-	210.0	-	-
Ru <sub>3</sub> PO <sub>4</sub>	1.00	-	-	201.9	-	-
Ag <sub>3</sub> PO <sub>4</sub>	1.00	-	-	192.1	-	-
La PO <sub>4</sub>	3.00	-	-	343.8	-	-
CePO <sub>4</sub>	3.00	-	-	342.0	-	-
W (PO <sub>4</sub> ) <sub>2</sub>	6.00	-	-	430.2	-	-
SO4 <sup>2-</sup>						
Be SO <sub>4</sub>	2.00	-180.8	0.937	510.1	332.5	561.6
Li <sub>2</sub> SO <sub>4</sub>	1.00	25.8	-0.267	487.5	-91.7	-162.3
Mg SO <sub>4</sub>	2.00	-99.6	0.516	445.3	166.3	298.6
Al <sub>2</sub> ( SO <sub>4</sub> ) <sub>3</sub>	3.00	-151.8	0.524	470.0	175.7	-
AIK(SO <sub>4</sub> ) <sub>2</sub>	4.00	-300.3	0.778	415.2	238.2	-
K <sub>2</sub> SO <sub>4</sub>	1.00	25.6	-0.265	307.6	-64.6	-126.0
CaSO <sub>4</sub>	2.00	38.9	-0.202	393.7	-59.3	-115.7
Sc 2(SO4)3	3.00	-	-	425.3	-	-
Ti (SO <sub>4</sub> ) <sub>2</sub>	4.00	-	-	446.7	-	-
V <sub>2</sub> (SO <sub>4</sub> ) <sub>5</sub>	5.00	-	-	460.3	-	-
Cr 2(SO <sub>4</sub> ) <sub>3</sub>	3.00	-596.7	2.062	410.0	625.4	1233.7
MnSO <sub>4</sub>	2.00	-312.8	1.621	355.0	441.1	925.6
FeSO <sub>4</sub>	2.00	-449.4	2.329	352.9	630.8	1402.3
Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	3.00	-774.0	2.674	402.1	799.5	1585.5
Co SO <sub>4</sub>	2.00	-487.8	2.528	345.8	674.2	1519.4
Ni SO4	2.00	-510.5	2.645	346.4	706.4	1649.0
CuSO <sub>4</sub>	2.00	-608.0	3.151	335.8	821.5	1841.0
Zn SO <sub>4</sub>	2.00	-398.7	2.066	332.0	533.9	1232.1
$Y_{2}(SO_{4})_{3}$	3.00	-	-	345.1	-	-

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Zr(SO <sub>4)2</sub>	4.00	-	-		378.3	-		-	
$Nb_2SO_4$	1.00	-	-		190.2	-		-	
$Mo_2SO_4$	1.00	-	-		186.2	-		-	
$Ru_2SO_4$	1.00	-	-		179.8	-		-	
Ag <sub>2</sub> SO <sub>4</sub>	1.00	-32	25.9	3.378	171.9		506.1		1730.7
La <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	3.00	-	-		284.1	-		-	
$Ce_2(SO_4)_3$	3.00	-10	)3.8	0.359	282.9		81.7	-	
W (SO <sub>4</sub> ) <sub>3</sub>	6.00	-	-		340.7	-		-	

	Мд									
M <sub>a</sub> X <sub>b</sub>	n	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )				
F										
BeF <sub>2</sub>	2.00	-91.7	0.475	1140.3	357.2	700.4				
NaF	1.00	10.8	-0.111	638.3	-55.2	-135.1				
LiF	1.00	52.2	-0.540	1033.2	-380.3	-951.0				
AIF <sub>3</sub>	3.00	-175.6	0.606	957.5	404.9	1014.5				
KF	1.00	2.3	-0.023	461.3	-8.9	-20.5				
CaF <sub>2</sub>	2.00	104.5	-0.542	686.6	-283.5	-753.3				
ScF₃	3.00	-66.8	0.231	788.7	134.1	299.7				
TiF <sub>3</sub>	3.00	-244.7	0.845	766.8	480.9	1210.0				
TiF <sub>4</sub>	4.00	-583.2	1.511	865.5	939.3	2242.6				
VF <sub>3</sub>	3.00	-380.0	1.313	744.9	731.0	1988.8				
CrF₃	3.00	-503.2	1.738	737.7	961.0	2814.8				
$MnF_2$	2.00	-264.0	1.368	576.8	625.5	1964.2				
$MnF_3$	3.00	-606.8	2.097	718.3	1136.0	3204.9				
FeF <sub>2</sub>	2.00	-402.5	2.086	571.2	946.3	3027.6				
FeF <sub>3</sub>	3.00	-634.7	2.193	712.5	1180.8	3516.4				
CoF <sub>2</sub>	2.00	-423.9	2.197	553.0	971.3	3296.7				
CoF <sub>3</sub>	3.00	-887.7	3.067	693.6	1618.1	4848.5				
NiF <sub>2</sub>	2.00	-467.0	2.420	554.4	1072.1	3753.9				
CuF	1.00	-275.6	2.856	324.7	808.3	4111.1				
CuF <sub>2</sub>	2.00	-579.1	3.001	527.9	1278.2	4234.3				

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	i.			i.		i i
$ZnF_2$	2.00	-357.8	1.854	518.5	778.3	2832.9
GaF <sub>3</sub>	3.00	-521.4	1.801	634.5	887.5	2936.0
RbF	1.00	-7.0	0.073	256.6	16.7	53.7
YF <sub>3</sub>	3.00	38.1	-0.131	551.1	-58.0	-184.0
$ZrF_2$	2.00	-158.1	0.819	414.8	286.1	-
$ZrF_3$	3.00	-281.1	0.971	542.5	422.8	1400.1
ZrF <sub>4</sub>	4.00	-332.3	0.861	641.1	427.7	1404.6
NbF <sub>5</sub>	5.00	-978.8	2.029	713.2	1093.4	2600.4
$RuF_3$	3.00	-	-	508.7	-	-
RuF₅	5.00	-1896.5	3.931	683.5	2051.2	6180.7
AgF	1.00	-348.6	3.612	211.3	696.4	3376.9
SnF <sub>2</sub>	2.00	-469.6	2.434	342.1	720.7	2702.3
LaF <sub>3</sub>	3.00	17.1	-0.059	410.4	-20.5	-87.8
$CeF_3$	3.00	-50.6	0.175	407.9	60.2	265.5
0						
BeO-α	2.00	10.8	-0.056	2143.1	-60.8	-134.6
B <sub>2</sub> O <sub>3</sub>	3.00	-256.8	0.887	2309.7	1000.9	2060.0
Na <sub>2</sub> O	1.00	-96.9	1.004	864.9	623.9	1303.8
Li <sub>2</sub> O	1.00	-4.0	0.042	1793.9	41.5	78.0
Al <sub>2</sub> O <sub>3</sub>	3.00	-62.8	0.217	1577.2	199.6	516.4
SiO <sub>2 quartz</sub>	4.00	-282.3	0.731	1784.3	721.4	1532.7
SiO <sub>2 high</sub>	1 00	005.0	0 700	4704.0	700.0	1170.0
cristobalite	4.00	-285.0	0.738	1/84.3	/28.3	14/3.9
P <sub>4</sub> O <sub>10</sub>	5.00	-742.4	1.539	1888.2	1565.5	3020.3
K <sub>2</sub> O	1.00	-123.6	1.281	569.1	579.5	1270.0
CaO	2.00	34.0	-0.176	955.9	-117.5	-306.9
	3.00	-74.3	-0.193	1166.0	-146.9	-398.9
Ti-O-	2.00	136.0	0.303	1118.8	350.0	1027.5
	2.00	-130.9	0.473	1102.6	425.0	11027.3
	3.33	-170.4	0.040	1190.0	420.9	1190.2
	4.00	-249.5	0.047	1342.3	539.5	1471.0
	4.00	-249.8	0.647	1342.3	540.1	14/2.6
TIO <sub>2</sub> -A	4.00	-255.7	0.662	1342.3	552.8	1466.0
VO	2.00	- 100.1	0.000	1072.0	709.0	2020.1
$v_2 U_3$	3.00	-284.3	0.982	1072.9	/08.9	21/1./
VO <sub>2</sub>	4.00	179.9	-0.466	1292.6	-379.8	-1061.1

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1	1			1		
V <sub>3</sub> O <sub>5</sub>	3.33	-347.8	1.082	1151.2	818.0	-
$V_2O_5$	5.00	-713.5	1.479	1473.6	1306.5	3191.2
CrO <sub>2</sub>	4.00	-593.7	1.538	1276.3	1243.7	3653.0
Cr <sub>2</sub> O <sub>3</sub>	3.00	-324.9	1.122	1058.0	802.6	2539.7
Cr <sub>3</sub> O <sub>4 ( Δ</sub>						
H=-1513 kJ						
-1 mol )	2.67	-248.7	0.967	974.7	653.4	2253.0
CrO <sub>3</sub>	6.00	-1195.3	2.065	1608.2	1920.3	4203.6
MnO <sub>2</sub>	4.00	-673.5	1.745	1233.1	1380.2	4149.8
Mn <sub>2</sub> O <sub>3</sub>	3.00	-413.4	1.428	1018.6	995.1	3123.5
MnO	2.00	-206.4	1.070	755.6	602.0	2108.3
Mn <sub>3</sub> O <sub>4</sub>	2.67	-331.3	1.288	937.1	846.9	2675.2
FeO	2.00	-317.9	1.647	746.1	918.4	3401.8
Fe <sub>2</sub> O <sub>3</sub>						
hematite	3.00	-482.9	1.668	1007.0	1153.3	3706.6
Fe <sub>3</sub> O <sub>4</sub>						
magnetite	2.67	-420.6	1.635	926.1	1066.2	3479.9
CoO	2.00	-355.1	1.840	715.3	994.0	3850.1
Co <sub>3</sub> O <sub>4</sub>	2.67	-501.1	1.947	890.4	1235.3	4379.3
NiO	2.00	-357.6	1.853	717.6	1003.4	4137.1
Ni <sub>2</sub> O <sub>3</sub>	3.00	-	-	972.3	-	-
CuO	2.00	-439.6	2.278	673.9	1175.8	4592.2
Cu <sub>2</sub> O	1.00	-211.7	2.194	374.6	702.4	3108.0
ZnO	2.00	-248.8	1.289	658.6	653.9	2423.4
Ga <sub>2</sub> O <sub>3</sub>	3.00	-354.8	1.226	857.9	757.1	2693.0
GeO	2.00	-332.1	1.721	604.7	816.8	-
GeO <sub>2</sub>	4.00	-617.2	1.599	1024.5	1118.7	3260.0
As <sub>4</sub> O <sub>6</sub>						
octahedral	3.00	-565.8	1.955	812.8	1161.0	3343.5
$As_2O_5$	5.00	-1032.1	2.139	1166.1	1631.9	4656.9
SeO <sub>2</sub>	4.00	-967.1	2.506	966.2	1683.6	4792.1
$Y_2O_3$	3.00	54.4	-0.188	712.1	-101.1	-347.7
ZrO <sub>2</sub>	4.00	-95.8	0.248	870.0	154.9	535.8
NbO	2.00	-177.4	0.919	492.2	369.8	1704.5
NbO <sub>2</sub>	4.00	-398.1	1.032	858.3	637.3	2250.4
Nb <sub>2</sub> O <sub>5</sub>	5.00	-540.4	1.120	1008.3	775.0	2320.1
MoO <sub>2</sub>	4.00	-605.6	1.569	837.9	952.8	3523.5
MoO <sub>3</sub>	6.00	-1039.9	1.796	1117.2	1332.1	3980.0

RuO <sub>2</sub>	4 00	-858.6	2 225	805.6	1312.8	50914
RuO4	8.00	-2125.0	2 753	1298.9	2250.5	5562.9
AqO	2.00	-583.1	3.022	432.7	1093.2	5310.8
Aq <sub>2</sub> O	1.00	-279.0	2.892	231.3	605.5	3357.7
Ag <sub>2</sub> O <sub>3</sub>	3.00	-914.7	3.160	609.7	1509.4	_
$La_2O_3$	3.00	-1.0	0.004	493.6	1.5	6.3
CeO <sub>2</sub>	4.00	-114.0	0.295	622.9	143.5	611.1
	3.00	-0.8	0.003	489.9	1.2	5.0
WQ <sub>2</sub>	4 00	-604 7	1 567	496 7	635.2	3502.8
WO <sub>3</sub>	6.00	-943.9	1.630	693.6	860.3	3535.8
	1 00	-	-	687.4	-	-
K <sub>2</sub> O <sub>2</sub>	1 00	-	_	486.4	-	-
	1 00	-	_	216.4	_	_
S	1.00			210.1		
BeS	2.00	-108.8	0.564	1304.9	462.2	962.8
$B_2S_3$	3.00	-388.9	1.344	1364.8	1132.8	1941.9
Na <sub>2</sub> S	1.00	4.0	-0.041	686.8	-21.7	-39.7
Li <sub>2</sub> S	1.00	48.6	-0.504	1166.6	-384.3	-867.4
Al <sub>2</sub> S <sub>3</sub>	3.00	-192.7	0.666	1070.9	479.9	920.6
SiS <sub>2</sub>	4.00	-471.0	1.220	1162.5	929.0	1788.0
$P_4S_3$	1.50	-216.6	1.497	730.6	821.4	1600.5
P <sub>4</sub> S <sub>7</sub>	3.50	-523.4	1.550	1077.1	1121.5	2263.1
$P_2S_5$	5.00	-784.9	1.627	1205.8	1268.3	2430.3
K <sub>2</sub> S	1.00	11.1	-0.115	486.1	-45.8	-79.7
CaS	2.00	135.6	-0.703	743.0	-390.5	-900.3
$Sc_2S_3$	3.00	-	-	864.1	-	-
TiS <sub>2</sub>	4.00	-281.4	0.729	957.2	486.7	1277.2
TiS	2.00	-71.7	0.372	670.6	191.1	573.4
Ti <sub>2</sub> S <sub>3</sub>	3.00	-	-	837.8	-	-
VS	2.00	-	-	645.8	-	-
$V_2S_3$	3.00	-	-	811.8	-	-
VS <sub>2</sub>	4.00	-	-	931.6	-	-
$V_2S_5$	5.00	-	-	1022.1	-	-
CrS	2.00	-167.4	0.867	637.7	429.0	-
$Cr_2S_3$	3.00	-428.8	1.481	803.3	872.2	2506.0
MnS	2.00	-123.4	0.639	616.1	308.0	887.8

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MnS <sub>2</sub>	4.00	-458.6	1.188	900.4	759.7	-
FeS	2.00	-241.4	1.251	609.7	597.6	2051.3
FeS <sub>2(marcasite</sub>						
)	4.00	-527.5	1.367	893.6	869.2	2825.0
FeS <sub>2(pyrite)</sub>	4.00	-516.7	1.339	893.6	851.4	2767.1
CoS	2.00	-	-	589.0	-	-
Co <sub>2</sub> S <sub>3</sub>	3.00	-	-	751.2	-	-
CoS <sub>2</sub>	4.00	-538.0	1.394	871.1	870.4	2640.6
$Co_3S_4$	2.67	-293.4	1.140	702.8	607.8	-
NiS	2.00	-262.3	1.359	590.6	633.2	2390.0
Ni <sub>3</sub> S <sub>4</sub>	2.67	-358.4	1.393	704.5	743.8	2494.4
$Ni_3S_2$	1.33	-157.9	1.227	446.3	455.5	-
NiS <sub>2</sub>	4.00	-558.9	1.448	872.8	905.6	-
CuS	2.00	-288.2	1.493	560.6	667.6	2349.7
Cu <sub>2</sub> S	1.00	-127.8	1.325	336.8	387.0	1674.3
ZnS	2.00	-140.5	0.728	550.0	320.5	1028.2
$Ga_2S_3$	3.00	-259.8	0.898	682.4	467.9	1366.5
GaS	2.00	-137.1	0.711	526.6	302.1	943.8
GeS	2.00	-270.3	1.401	511.9	582.0	1899.8
GeS <sub>2</sub>	4.00	-529.0	1.371	783.8	792.7	2001.8
$As_2S_3$	3.00	-428.4	1.480	653.6	746.2	2105.0
Se <sub>2</sub> S <sub>6</sub>	6.00	-	-	918.1	-	-
Se <sub>4</sub> S <sub>4</sub>	2.00	-	-	482.8	-	-
RuS <sub>2</sub>	4.00	-	-	648.9	-	-
$Y_2S_3$	3.00	-	-	586.9	-	-
ZrS <sub>2</sub>	4.00	-113.6	0.294	690.1	154.7	463.2
NbS <sub>2</sub>	4.00	-683.6	1.771	682.7	923.4	2982.9
MoS <sub>2</sub>	4.00	-457.7	1.186	669.7	609.2	2133.1
$Mo_2S_3$	3.00	-373.4	1.290	558.2	574.6	2287.2
Ag <sub>2</sub> S(argent						
ite)	1.00	-150.6	1.561	216.3	307.4	1733.3
LaS	2.00	109.7	-0.568	313.5	-156.0	-685.4
$La_2S_3$	3.00	91.5	-0.316	430.0	-113.7	-429.6
CeS	2.00	109.7	-0.568	311.3	-155.1	-705.9
WS <sub>2</sub>	4.00	-433.7	1.124	432.3	406.2	1988.1
$Na_2S_2$	1.00	-	-	486.8	-	-
N						

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$Be_3N_2$	2.00	44.0	-0.228	2921.1	-286.5	-588.9
BN	3.00	28.0	-0.097	3239.7	-126.7	-239.9
(CN)	3.00	-	-	3090.4	-	-
Na₃N	1.00	-	-	969.0	-	-
Li <sub>3</sub> N	1.00	-24.0	0.248	2308.5	280.0	590.0
AIN	3.00	86.6	-0.299	1961.6	-310.4	-716.2
Si <sub>3</sub> N <sub>4</sub>	4.00	-53.1	0.137	2292.6	154.6	345.1
K <sub>3</sub> N	1.00	-	-	612.4	-	-
Ca <sub>3</sub> N <sub>2</sub>	2.00	-11.0	0.057	1084.7	41.3	93.7
ScN	3.00	83.3	-0.288	1363.6	-242.5	-
TiN	3.00	43.4	-0.150	1299.5	-122.5	-366.5
VN	3.00	-9.4	0.032	1238.0	25.7	82.4
CrN	3.00	-107.5	0.371	1218.2	291.3	928.0
Cr <sub>2</sub> N	1.50	-49.1	0.339	681.4	176.7	712.1
Mn₄N	0.75	-23.9	0.330	344.0	98.1	-
$Mn_5N_2$	1.20	-49.2	0.425	531.2	181.8	-
Fe <sub>2</sub> N	1.50	-	-	639.7	-	-
Fe₄N	0.75	-49.2	0.680	338.7	199.6	-
Co₃N	1.00	-55.4	0.574	421.4	203.0	908.4
NiN	3.00	-	-	1106.0	-	-
Cu₃N	1.00	-	-	392.9	-	-
Zn <sub>3</sub> N <sub>2</sub>	2.00	-146.7	0.760	717.3	411.6	1567.9
GaN	3.00	-122.7	0.424	960.3	283.6	982.2
$Ge_3N_4$	4.00	-	-	1174.0	-	-
$As_3N_5$	5.00	-	-	1363.7	-	-
SeN <sub>2</sub>	6.00	-	-	1503.3	-	-
YN	3.00	68.1	-0.235	781.3	-135.8	-480.8
ZrN	3.00	136.3	-0.471	764.1	-267.1	-1056.6
NbN	3.00	5.5	-0.019	752.1	-10.6	-45.1
MoN	3.00	-	-	731.3	-	-
RuN	3.00	-	-	698.7	-	-
Ag₃N	1.00	-	-	238.2	-	-
LaN	3.00	70.6	-0.244	525.8	-103.5	-448.7
CeN	3.00	94.5	-0.326	521.7	-137.7	-648.0
WN <sub>2</sub>	6.00	-	-	759.1	-	-
Р						
Be <sub>3</sub> P <sub>2</sub>	2.00	-	-	1807.2	-	-
BP	3.00	-	-	1924.2	-	-

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	Na₃P	1.00	-	-	804.5	-	-
	Li₃P	1.00	-	-	1552.3	-	-
	AIP	3.00	-	-	1387.3	-	-
	Si <sub>3</sub> P <sub>4</sub>	4.00	-	-	1545.1	-	-
	K₃P	1.00	-	-	542.3	-	-
	$Ca_3P_2$	2.00	-	-	882.7	-	-
	ScP	3.00	-	-	1058.9	-	-
	TiP	3.00	-	-	1019.8	-	-
	(VP)	3.00	-	-	981.6	-	-
	CrP	3.00	-	-	969.1	-	-
	MnP	3.00	-	-	935.9	-	-
	Mn₂P	1.50	-	-	570.9	-	-
	FeP	3.00	-	-	926.1	-	-
	Fe <sub>2</sub> P	1.50	-	-	563.6	-	-
	Fe₃P	1.00	-	-	405.0	-	-
	Co <sub>2</sub> P	1.50	-	-	540.2	-	-
	Ni <sub>2</sub> P	1.50	-	-	542.0	-	-
	Ni₃P	1.00	-	-	388.3	-	-
	$Ni_5P_2$	1.20	-	-	452.5	-	-
	CuP <sub>2</sub>	6.00	-	-	1281.4	-	-
	Cu₃P	1.00	-	-	362.8	-	-
	$Zn_3P_2$	2.00	-	-	623.0	-	-
	GaP	3.00	-	-	798.5	-	-
	GeP	3.00	-	-	776.0	-	-
	Ge <sub>3</sub> P <sub>4</sub>	4.00	-	-	940.9	-	-
	$As_3P_5$	5.00	-	-	1059.0	-	-
	Se(P) <sub>2</sub>	6.00	-	-	1141.2	-	-
	YP	3.00	-	-	670.7	-	-
	ZrP <sub>2</sub>	6.00	-	-	1049.9	-	-
	NbP	3.00	-	-	649.0	-	-
	MoP	3.00	-	-	633.5	-	-
	RuP	3.00	-	-	608.9	-	-
	AgP <sub>2</sub>	6.00	-	-	947.0	-	-
	AgP <sub>3</sub>	9.00	-	-	1201.3	-	-
	LaP	3.00	-	-	473.3	-	-
	Ce <sub>3</sub> P <sub>4</sub>	4.00	-	-	590.9	-	-
	WP <sub>2</sub>	6.00	-	-	654.3	-	-
ļ	-		1	1		1	1

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CI						
BeCl <sub>2</sub>	2.00	-146.2	0.758	670.7	389.7	724.6
PCl₅	5.00	-1174.6	2.435	643.5	1212.9	2432.7
NaCl	1.00	88.2	-0.914	458.6	-347.0	-722.2
LiCl	1.00	88.5	-0.917	632.2	-450.7	-974.5
AICI <sub>3</sub>	3.00	-258.9	0.894	603.0	423.5	962.2
KCI	1.00	112.6	-1.167	359.5	-360.7	-703.0
CaCl <sub>2</sub>	2.00	157.0	-0.814	483.0	-322.4	-664.8
ScCl <sub>3</sub>	3.00	-61.8	0.214	531.4	91.4	204.3
TiCl <sub>2</sub>	2.00	-127.4	0.660	451.3	247.3	681.5
TiCl <sub>3</sub>	3.00	-234.2	0.809	521.3	341.2	819.4
VCI <sub>2</sub>	2.00	-185.8	0.963	439.9	353.1	998.1
VCI <sub>3</sub>	3.00	-376.4	1.300	511.2	539.6	1424.3
CrCl <sub>2</sub>	2.00	-235.8	1.222	436.1	445.0	1156.0
CrCl <sub>3</sub>	3.00	-401.6	1.387	507.7	572.6	1423.8
MnCl <sub>2</sub>	2.00	-151.3	0.784	425.9	279.9	747.1
FeCl <sub>2</sub>	2.00	-289.5	1.500	422.9	532.4	1490.1
FeCl <sub>3</sub>	3.00	-553.7	1.913	495.7	774.2	1999.8
CoCl <sub>2</sub>	2.00	-322.0	1.669	412.8	580.3	1699.6
NiCl <sub>2</sub>	2.00	-332.8	1.725	413.6	600.7	1815.9
CuCl	1.00	-176.0	1.824	270.7	439.8	1581.9
CuCl <sub>2</sub>	2.00	-416.1	2.156	398.7	728.1	2159.3
ZnCl <sub>2</sub>	2.00	-222.4	1.153	393.3	384.7	1014.9
GaCl <sub>3</sub>	3.00	-432.9	1.496	456.6	565.8	1303.3
AsCl <sub>5</sub>	5.00	-	-	531.4	-	-
SeCl <sub>4</sub>	4.00	-	-	485.6	-	-
YCl <sub>3</sub>	3.00	40.0	-0.138	411.8	-48.0	-116.1
ZrCl <sub>2</sub>	2.00	-205.8	1.066	330.6	306.6	875.6
ZrCl <sub>3</sub>	3.00	-241.7	0.835	406.9	286.9	782.9
ZrCl <sub>4</sub>	4.00	-293.7	0.761	460.0	289.7	733.7
NbCl <sub>5</sub>	5.00	-796.3	1.651	496.0	668.4	1673.9
NbCl <sub>4</sub>	4.00	-578.1	1.498	456.7	566.7	1584.8
MoCl <sub>4</sub>	4.00	-781.6	2.025	450.9	758.2	-
MoCl <sub>5</sub>	5.00	-1056.5	2.190	490.5	878.7	2289.1
MoCl <sub>6</sub>	6.00	-1384.4	2.391	521.0	1007.8	-

	I	1	I	1	I	I
RuCl <sub>3</sub>	3.00	-728.0	2.515	387.6	829.1	2300.8
AgCl	1.00	-186.1	1.929	187.0	332.5	1577.5
SnCl <sub>2</sub>	2.00	-305.6	1.584	282.7	396.8	1369.3
LaCl <sub>3</sub>	3.00	-	-	327.8	-	-
CeCl <sub>3</sub>	3.00	97.1	-0.335	326.2	-95.3	-324.7
WCl <sub>2</sub>	2.00	-371.8	1.927	210.4	370.1	1698.4
WCl <sub>4</sub>	4.00	-823.6	2.134	329.2	611.3	2323.6
WCI <sub>5</sub>	5.00	-1077.5	2.234	371.1	709.5	2337.8
WCl <sub>6</sub>	6.00	-1319.4	2.279	405.5	780.7	2370.5
BO <sub>2</sub> <sup>-</sup>						
NaBO <sub>2</sub>	1.00	-	-	407.3	-	-
LiBO <sub>2</sub>	1.00	-	-	538.7	-	-
AI (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	517.4	-	-
KBO <sub>2</sub>	1.00	-	-	327.2	-	-
Sc(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	463.7	-	-
Ti (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	489.3	-	-
V(BO <sub>2</sub> ) <sub>5</sub>	5.00	-	-	505.7	-	-
Cr(BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	520.6	-	-
Mn (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	438.5	-	-
Fe (BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	378.9	-	-
Co (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	429.1	-	-
Ni(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	371.4	-	-
Cu(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	359.3	-	-
Zn(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	355.0	-	-
Ga(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	405.8	-	-
Ge(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	439.6	-	-
Y (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	369.9	-	-
Zr (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	408.5	-	-
Nb BO <sub>2</sub>	1.00	-	-	197.5	-	-
MoBO <sub>2</sub>	1.00	-	-	193.2	-	-
Ru(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	287.1	-	-
Ag BO <sub>2</sub>	1.00	-	-	177.9	-	-
La (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	300.8	-	-
Ce(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	344.3	-	-
W (BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	364.9	-	-

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Be <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-		-		-	1111.8	-		-	
Ca <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	2.00	-		-			676.1	-		-	
CO3 <sup>2-</sup>				-							
Be CO <sub>3</sub>	2.00	-		-			776.6	-		-	
B <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-			797.5	-		-	
Na <sub>2</sub> CO <sub>3</sub>	1.00		16.2		-0.167		505.7		-68.9		-161.0
Li <sub>2</sub> CO <sub>3</sub>	1.00	-		-			725.4	-		-	
Al <sub>2</sub> ( CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-			687.2	-		-	
K <sub>2</sub> CO <sub>3</sub>	1.00		25.7		-0.266		387.8		-87.9		-192.1
CaCO <sub>3</sub>	2.00		116.7		-0.605		535.6		-260.6		-645.9
Sc 2(CO3)3	3.00	-		-			595.7	-		-	
Ti (CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-			638.6	-		-	
V <sub>2</sub> (CO <sub>3</sub> ) <sub>5</sub>	5.00	-		-			666.8	-		-	
Cr (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-			693.1	-		-	
MnCO₃	2.00		-195.4		1.013		466.3		389.8		1204.8
Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-			551.2	-		-	
FeCO <sub>3</sub>	2.00		-345.4		1.790		450.7		669.8		2173.5
CoCO <sub>3</sub>	2.00		-375.3		1.945		450.7		727.8		2464.4
Ni CO <sub>3</sub>	2.00		-394.2		2.043		451.6		765.7		2668.9
CuCO <sub>3</sub>	2.00	-		-			433.8	-		-	
Zn CO <sub>3</sub>	2.00		-280.5		1.454		427.5		520.5		1843.7
Ga <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-			503.4	-		-	
Ge(CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-			556.5	-		-	
Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-			449.4	-		-	
ZrCO <sub>3</sub>	2.00	-		-			354.4	-		-	
Nb <sub>2</sub> CO <sub>3</sub>	1.00	-		-			218.1	-		-	
Mo <sub>2</sub> CO <sub>3</sub>	1.00	-		-			212.8	-		-	
Ru <sub>2</sub> CO <sub>3</sub>	1.00	-		-			204.5	-		-	
Ag <sub>2</sub> CO <sub>3</sub>	1.00		-287.7		2.981		194.4		532.6		2692.2
La 2(CO3)3	3.00	-		-			351.2	-		-	
$Ce_2(CO_3)_3$	3.00	-		-			349.4	-		-	
W (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-			441.9	-		-	
PO4 <sup>3-</sup>											
Be <sub>3(</sub> PO <sub>4)2</sub>	2.00	-		-			741.1	-		-	

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Na <sub>3</sub> PO <sub>4</sub> Li <sub>3</sub> PO <sub>4( Δ</sub>	1.00	6.5	-0.067	490.4	-27.0	-63.2
H=-698.6 kJ -1	1 00	250.0	2 000	CO 4 4	4050.0	4040.0
mol)	1.00	-356.9	3.699	694.4	1953.6	4040.8
	3.00	-151.4	0.523	059.3	265.5	612.9
$K_3 PU_4$	1.00	-	-	3/8.8	-	-
$Ca_3(PO_4)_2$	2.00	115.3	-0.598	518.4	-250.9	-682.9
	3.00	-	-	574.6	-	-
$11_3(PO_4)_4$	4.00	-	-	614.4	-	-
V <sub>3</sub> (PO <sub>4</sub> ) <sub>5</sub>	5.00	-	-	640.5	-	-
	3.00	-	-	547.1	-	-
Mn PO₄	3.00	-	-	536.4	-	-
Mn <sub>3</sub> ( PO <sub>4 )2</sub>	2.00	-	-	453.3	-	-
Fe PO <sub>4</sub>	3.00	-581.8	2.010	533.1	863.0	2198.0
Co <sub>3</sub> (PO <sub>4)2</sub>	2.00	-	-	438.5	-	-
Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-	-	439.3	-	-
$Cu_3(PO_4)_2$	2.00	-	-	422.5	-	-
$Zn_3(PO_4)_2$	2.00	-291.7	1.512	416.5	529.5	1755.2
Ga PO <sub>4</sub>	3.00	-	-	488.2	-	-
Ge <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	4.00	-	-	538.0	-	-
Y PO <sub>4</sub>	3.00	-	-	437.3	-	-
Zr <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-	-	346.9	-	-
Nb <sub>3</sub> PO <sub>4</sub>	1.00	-	-	215.2	-	-
Mo <sub>3</sub> PO <sub>4</sub>	1.00	-	-	210.0	-	-
Ru <sub>3</sub> PO <sub>4</sub>	1.00	-	-	201.9	-	-
Ag <sub>3</sub> PO <sub>4</sub>	1.00	-	-	192.1	-	-
La PO₄	3.00	-	-	343.8	-	-
CePO <sub>4</sub>	3.00	-	-	342.0	-	-
W (PO <sub>4</sub> ) <sub>2</sub>	6.00	-	-	430.2	-	-
SO4 <sup>2-</sup>						
Be SO <sub>4</sub>	2.00	-81.2	0.421	510.1	174.3	402.7
Na <sub>2</sub> SO <sub>4</sub>	1.00	49.8	-0.516	377.4	-166.3	-415.5
Li <sub>2</sub> SO <sub>4</sub>	1.00	75.6	-0.783	487.5	-312.6	-758.8
Al <sub>2</sub> ( SO <sub>4</sub> ) <sub>3</sub>	3.00	-2.4	0.008	470.0	3.2	-
AIK(SO <sub>4</sub> ) <sub>2</sub>	4.00	-101.1	0.262	415.2	91.5	-
K <sub>2</sub> SO <sub>4</sub>	1.00	75.4	-0.781	307.6	-211.0	-526.9
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CaSO <sub>4</sub>	2.00	138.5	-0.718	393.7	-239.8	-641.4
Sc 2(SO4)3	3.00	-	-	425.3	-	-
Ti (SO <sub>4</sub> ) <sub>2</sub>	4.00	-	-	446.7	-	-
$V_2(SO_4)_5$	5.00	-	-	460.3	-	-
Cr 2(SO4)3	3.00	-447.3	1.545	410.0	534.3	1475.2
MnSO <sub>4</sub>	2.00	-213.2	1.105	355.0	337.8	979.7
FeSO <sub>4</sub>	2.00	-349.8	1.813	352.9	551.4	1747.5
$Fe_2(SO_4)_3$	3.00	-624.6	2.158	402.1	733.9	2029.7
Co SO <sub>4</sub>	2.00	-388.2	2.012	345.8	601.4	1933.8
Ni SO4	2.00	-410.9	2.129	346.4	637.4	2170.9
CuSO <sub>4</sub>	2.00	-508.4	2.635	335.8	767.9	2421.5
Zn SO <sub>4</sub>	2.00	-299.1	1.550	332.0	447.3	1471.2
$Y_2(SO_4)_3$	3.00	-	-	345.1	-	-
Zr(SO <sub>4)2</sub>	4.00	-	-	378.3	-	-
$Nb_2SO_4$	1.00	-	-	190.2	-	-
$Mo_2SO_4$	1.00	-	-	186.2	-	-
$Ru_2SO_4$	1.00	-	-	179.8	-	-
Ag <sub>2</sub> SO <sub>4</sub>	1.00	-276.1	2.862	171.9	456.4	2154.5
La <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	3.00	-	-	284.1	-	-
$Ce_2(SO_4)_3$	3.00	45.6	-0.157	282.9	-39.5	-
W (SO <sub>4</sub> ) <sub>3</sub>	6.00	-	-	340.7	-	-

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AI									
$M_a X_b$	n	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )			
F									
BeF <sub>2</sub>	2.00	25.3	-0.131	1140.3	-108.3	-242.3			
NaF	1.00	69.3	-0.718	638.3	-377.4	-1043.9			
MgF <sub>2</sub>	2.00	117.0	-0.606	860.4	-404.9	-1229.2			
LiF	1.00	110.7	-1.147	1033.2	-880.0	-2628.3			
KF	1.00	60.8	-0.630	461.3	-251.6	-630.9			
CaF <sub>2</sub>	2.00	221.5	-1.148	686.6	-640.6	-1971.8			

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1	1	1 1	1	1	1	1	
ScF <sub>3</sub>	3.00	108.7	-0.376	788.7	-234.3	-593.1	
TiF₃	3.00	-69.1	0.239	766.8	145.6	424.9	
TiF <sub>4</sub>	4.00	-349.1	0.905	865.5	606.8	1684.2	
VF <sub>3</sub>	3.00	-204.4	0.706	744.9	420.9	1349.4	
CrF <sub>3</sub>	3.00	-327.7	1.132	737.7	669.4	2353.8	
$MnF_2$	2.00	-147.0	0.762	576.8	368.0	1360.4	
MnF <sub>3</sub>	3.00	-431.3	1.490	718.3	862.4	2879.5	
FeF <sub>2</sub>	2.00	-285.5	1.479	571.2	709.1	2678.8	
FeF <sub>3</sub>	3.00	-459.1	1.586	712.5	912.1	3258.0	
CoF <sub>2</sub>	2.00	-306.9	1.590	553.0	741.8	3002.5	
CoF <sub>3</sub>	3.00	-712.1	2.460	693.6	1384.1	4962.0	
NiF <sub>2</sub>	2.00	-350.0	1.814	554.4	847.7	3570.1	
CuF	1.00	-217.0	2.249	324.7	658.6	4031.4	
CuF <sub>2</sub>	2.00	-462.1	2.395	527.9	1073.8	4185.9	
$ZnF_2$	2.00	-240.8	1.248	518.5	551.0	2409.5	
GaF <sub>3</sub>	3.00	-345.8	1.195	634.5	625.0	2505.7	
RbF	1.00	51.5	-0.534	256.6	-126.1	-437.4	
YF <sub>3</sub>	3.00	213.6	-0.738	551.1	-343.2	-1277.1	
ZrF <sub>2</sub>	2.00	-41.1	0.213	414.8	77.5	-	
ZrF <sub>3</sub>	3.00	-105.5	0.000	542.5	167.3	654.7	
ZrF <sub>4</sub>	4.00	-98.2	0.255	641.1	134.3	534.4	
NbF <sub>5</sub>	5.00	-686.2	1.422	713.2	818.5	2210.3	
RuF <sub>3</sub>	3.00	-	-	508.7	-	-	
RuF₅	5.00	-1603.9	3.325	683.5	1848.4	6658.1	
AgF	1.00	-290.0	3.006	211.3	593.0	3221.6	
SnF <sub>2</sub>	2.00	-352.6	1.827	342.1	560.7	2391.9	
LaF <sub>3</sub>	3.00	192.7	-0.666	410.4	-240.1	-1239.2	
CeF <sub>3</sub>	3.00	124.9	-0.432	407.9	-154.8	-826.1	
0							
BeO-α	2.00	52.7	-0.273	2143.1	-340.3	-977.7	
B <sub>2</sub> O <sub>3</sub>	3.00	-194.0	0.670	2309.7	872.0	2279.7	
Na <sub>2</sub> O MgO	1.00	-76.0	0.787	864.9	527.7	1242.6	
microcrystal	2.00	41.9	-0.217	1329.9	-199.6	-662.8	
Li <sub>2</sub> O	1.00	16.9	-0.175	1793.9	-196.0	-436.4	
SiO <sub>2 quartz</sub>	4.00	-198.5	0.514	1784.3	574.1	1510.7	

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SiO <sub>2 high</sub>							
cristobalite	4.00	-201.2	0.521	1784.3	581.9	1431.3	
P <sub>4</sub> O <sub>10</sub>	5.00	-637.7	1.322	1888.2	1527.9	3545.6	
K <sub>2</sub> O	1.00	-102.7	1.064	569.1	508.4	1220.2	
CaO	2.00	75.9	-0.393	955.9	-284.6	-898.9	
Sc <sub>2</sub> O <sub>3</sub>	3.00	118.6	-0.410	1166.0	-343.3	-1183.4	
TiO	2.00	-32.4	0.168	839.3	110.0	460.5	
Ti <sub>2</sub> O <sub>3</sub>	3.00	-74.0	0.256	1118.8	208.1	790.8	
Ti <sub>3</sub> O <sub>5</sub>	3.33	-106.6	0.331	1198.6	283.3	1032.4	
TiO <sub>2</sub> -R	4.00	-165.8	0.430	1342.3	397.5	1418.4	
TiO <sub>2</sub>	4.00	-166.0	0.430	1342.3	398.2	1420.7	
TiO <sub>2</sub> -A	4.00	-171.9	0.445	1342.3	412.3	1413.2	
VO	2.00	-123.2	0.639	800.7	403.0	2239.2	
$V_2O_3$	3.00	-221.5	0.765	1072.9	603.6	2424.7	
VO <sub>2</sub>	4.00	263.6	-0.683	1292.6	-615.8	-2258.0	
V <sub>3</sub> O <sub>5</sub>	3.33	-278.0	0.865	1151.2	717.9	-	
V <sub>2</sub> O <sub>5</sub>	5.00	-608.8	1.262	1473.6	1244.3	3862.0	
CrO <sub>2</sub>	4.00	-509.9	1.321	1276.3	1180.7	4645.7	
Cr <sub>2</sub> O <sub>3</sub>	3.00	-262.1	0.905	1058.0	707.0	2966.1	
Cr <sub>3</sub> O <sub>4</sub> ( $^{\triangle}$							
H=-1513 kJ							
mol )	2.67	-192.9	0.750	974.7	550.6	2564.1	
CrO <sub>3</sub>	6.00	-1069.7	1.848	1608.2	1930.0	5212.4	
MnO <sub>2</sub>	4.00	-589.7	1.528	1233.1	1332.8	5383.8	
$Mn_2O_3$	3.00	-350.6	1.211	1018.6	919.4	3778.5	
MnO	2.00	-164.5	0.853	755.6	513.9	2300.3	
Mn <sub>3</sub> O <sub>4</sub>	2.67	-275.5	1.071	937.1	763.3	3106.5	
FeO	2.00	-276.0	1.430	746.1	853.5	4115.3	
Fe <sub>2</sub> O <sub>3</sub>							
hematite	3.00	-420.0	1.451	1007.0	1092.2	4631.1	
Fe <sub>3</sub> O <sub>4</sub>	0.07	004.0	4 440	000.4	4004.0	4050 F	
magnetite	2.67	-364.8	1.418	926.1	1001.6	4256.5	
00	2.00	-313.2	1.623	715.3	936.4	4756.4	
Co <sub>3</sub> O <sub>4</sub>	2.67	-445.2	1.730	890.4	1186.3	5618.3	
NiO	2.00	-315.7	1.636	717.6	946.3	5256.9	
Ni <sub>2</sub> O <sub>3</sub>	3.00	-	-	972.3	-	-	
CuO	2.00	-397.7	2.061	673.9	1132.7	5735.2	
Cu <sub>2</sub> O	1.00	-190.7	1.977	374.6	657.8	3473.1	

ZnO	2.00	-206.9	1.072	658.6	578.4	2712.4
Ga <sub>2</sub> O <sub>3</sub>	3.00	-292.0	1.009	857.9	672.0	3167.5
GeO	2.00	-290.2	1.504	604.7	756.1	-
GeO <sub>2</sub>	4.00	-533.4	1.382	1024.5	1053.8	3906.1
As <sub>4</sub> O <sub>6</sub>						
octahedral	3.00	-503.0	1.738	812.8	1109.8	3879.3
As <sub>2</sub> O <sub>5</sub>	5.00	-927.4	1.922	1166.1	1611.2	5957.1
SeO <sub>2</sub>	4.00	-883.4	2.289	966.2	1670.0	5926.4
Y <sub>2</sub> O <sub>3</sub>	3.00	117.2	-0.405	712.1	-232.7	-1003.5
ZrO <sub>2</sub>	4.00	-12.0	0.031	870.0	21.0	95.6
NbO	2.00	-135.5	0.702	492.2	296.6	1744.3
NbO <sub>2</sub>	4.00	-314.3	0.814	858.3	542.8	2532.1
Nb <sub>2</sub> O <sub>5</sub>	5.00	-435.7	0.903	1008.3	680.3	2609.5
MoO <sub>2</sub>	4.00	-521.8	1.352	837.9	884.4	4380.9
MoO <sub>3</sub>	6.00	-914.3	1.579	1117.2	1283.3	5019.4
RuO <sub>2</sub>	4.00	-774.8	2.008	805.6	1273.2	6686.5
RuO <sub>4</sub>	8.00	-1957.5	2.536	1298.9	2294.1	7079.9
AgO	2.00	-541.3	2.805	432.7	1059.9	6488.1
Ag <sub>2</sub> O	1.00	-258.1	2.675	231.3	574.2	3691.6
Ag <sub>2</sub> O <sub>3</sub>	3.00	-851.8	2.943	609.7	1489.6	-
La <sub>2</sub> O <sub>3</sub>	3.00	61.8	-0.213	493.6	-90.4	-490.1
CeO <sub>2</sub>	4.00	-30.2	0.078	622.9	40.4	226.1
Ce <sub>2</sub> O <sub>3</sub>	3.00	62.0	-0.214	489.9	-90.1	-472.1
WO <sub>2</sub>	4.00	-520.9	1.350	496.7	574.7	4345.6
WO <sub>3</sub>	6.00	-818.3	1.413	693.6	795.3	4356.8
S						
BeS	2.00	19.7	-0.102	1304.9	-92.5	-227.0
$B_2S_3$	3.00	-196.2	0.678	1364.8	634.5	1220.9
Na₂S	1.00	68.2	-0.707	686.8	-394.7	-778.3
MgS	2.00	128.5	-0.666	950.9	-479.9	-1288.7
Li <sub>2</sub> S	1.00	112.8	-1.169	1166.6	-980.4	-2131.9
SiS <sub>2</sub>	4.00	-214.1	0.555	1162.5	463.9	1016.1
$P_4S_3$	1.50	-120.3	0.831	730.6	487.5	1040.7
P <sub>4</sub> S <sub>7</sub>	3.50	-298.5	0.884	1077.1	699.4	1612.9
$P_2S_5$	5.00	-463.7	0.961	1205.8	825.1	1804.3
K <sub>2</sub> S	1.00	75.3	-0.781	486.1	-326.3	-597.7

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CaS 2.00 264.1 -1.368 743.0 -813.8 -2125.4  $Sc_2S_3$ 3.00 864.1 \_ 4.00 TiS<sub>2</sub> -24.5 0.063 957.2 46.0 146.1 TiS 2.00 56.7 -0.294 670.6 -161.0 -574.8 Ti<sub>2</sub>S<sub>3</sub> 3.00 837.8 \_ \_ \_ \_ VS 2.00 \_ 645.8 --\_  $V_2S_3$ 3.00 \_ \_ 811.8 \_ \_  $VS_2$ 4.00 931.6 -\_ -\_ 5.00 1022.1  $V_2S_5$ \_ --\_ CrS 2.00 -38.9 0.202 637.7 105.9 \_ 3.00 -236.1 0.816 516.0  $Cr_2S_3$ 803.3 1794.8 MnS 2.00 -0.026 -45.0 5.1 616.1 -13.4 MnS<sub>2</sub> 4.00 -201.7 0.523 900.4 361.3 \_ FeS 2.00 -112.9 0.585 609.7 296.2 1236.9 FeS<sub>2</sub>(marca 4.00 -270.6 0.701 893.6 481.9 2019.6 site) FeS<sub>2</sub>(pyrite) 4.00 -259.8 0.673 893.6 462.7 1939.0 CoS 2.00 589.0 ---- $Co_2S_3$ 3.00 --751.2 -\_ CoS<sub>2</sub> 4.00 -281.0 0.728 871.1 490.8 1861.5  $Co_3S_4$ -122.1 2.67 0.475 702.8 270.0 \_ NiS 2.00 -133.8 0.694 590.6 341.9 1605.3  $Ni_3S_4$ 2.67 -187.1 0.727 704.5 414.4 1724.4 -72.2 446.3  $Ni_3S_2$ 1.33 0.561 217.9 \_ 4.00 -302.0 NiS<sub>2</sub> 0.782 872.8 528.2 CuS 2.00 -159.7 0.828 560.6 390.6 1659.1  $Cu_2S$ 1.00 336.8 199.4 1006.7 -63.6 0.659 ZnS 2.00 -12.0 0.062 550.0 29.0 109.1  $Ga_2S_3$ 3.00 -67.1 0.232 682.4 128.8 445.9 GaS 2.00 -8.6 0.045 526.6 20.1 72.7 GeS 2.00 -141.8 0.735 511.9 321.1 1223.7 GeS<sub>2</sub> 4.00 -272.1 0.705 783.8 437.5 1286.3 3.00 -235.7 0.814 436.5 1437.7  $As_2S_3$ 653.6  $Se_2S_6$ 6.00 \_ 918.1 \_ -\_  $Se_4S_4$ 2.00 \_ 482.8 \_ \_ \_ \_  $RuS_2$ 4.00 \_ 648.9 \_ \_ \_  $Y_2S_3$ 3.00 -586.9 \_ \_

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$ZrS_2$	4.00	143.3	-0.371	690.1	-208.1	-744.9
NbS <sub>2</sub>	4.00	-	-	682.7	-	-
MoS <sub>2</sub>	4.00	-200.8	0.520	669.7	284.5	1240.7
$Mo_2S_3$	3.00	-180.7	0.624	558.2	293.5	1460.9
Ag <sub>2</sub> S(argent						
ite)	1.00	-86.3	0.895	216.3	180.5	1171.8
LaS	2.00	238.2	-1.234	313.5	-350.1	-1781.6
$La_2S_3$	3.00	284.2	-0.982	430.0	-368.9	-1639.3
CeS	2.00	238.2	-1.234	311.3	-347.9	-1845.9
WS <sub>2</sub>	4.00	-176.8	0.458	432.3	172.9	1068.8
N						
Be <sub>3</sub> N <sub>2</sub>	2.00	-13.7	0.071	2921.1	104.7	283.4
BN	3.00	-58.6	0.202	3239.7	314.2	761.7
(CN)	3.00	-	-	3090.4	-	-
Na₃N	1.00	-	-	969.0	-	-
Mg₃N2	2.00	-57.7	0.299	1593.3	310.4	840.4
Li <sub>3</sub> N	1.00	-52.8	0.547	2308.5	711.8	2127.0
Si <sub>3</sub> N <sub>4</sub>	4.00	-168.5	0.437	2292.6	565.6	1667.4
K₃N	1.00	-	-	612.4	-	-
Ca <sub>3</sub> N <sub>2</sub>	2.00	-68.7	0.356	1084.7	283.0	757.9
ScN	3.00	-3.2	0.011	1363.6	10.5	-
TiN	3.00	-43.2	0.149	1299.5	135.1	548.9
VN	3.00	-95.9	0.331	1238.0	289.8	1294.6
CrN	3.00	-194.0	0.670	1218.2	579.5	2545.2
Cr₂N	1.50	-92.4	0.638	681.4	354.1	1877.7
Mn₄N	0.75	-45.5	0.629	344.0	193.9	-
$Mn_5N_2$	1.20	-83.8	0.724	531.2	326.3	-
Fe <sub>2</sub> N	1.50	-	-	639.7	-	-
Fe₄N	0.75	-70.8	0.979	338.7	297.6	-
Co <sub>3</sub> N	1.00	-84.2	0.873	421.4	322.2	1763.3
NiN	3.00	-	-	1106.0	-	-
Cu₃N	1.00	-	-	392.9	-	-
$Zn_3N_2$	2.00	-204.4	1.059	717.3	612.5	3041.6
GaN	3.00	-209.3	0.723	960.3	525.0	2451.4
Ge <sub>3</sub> N <sub>4</sub>	4.00	-	-	1174.0	-	-
$As_3N_5$	5.00	-	-	1363.7	-	-
SeN <sub>2</sub>	6.00	-	-	1503.3	-	-

YN	3.00	-18.4	0.064	781.3	39.4	180.5
ZrN	3.00	49.7	-0.172	764.1	-104.4	-556.1
NbN	3.00	-81.1	0.280	752.1	168.3	996.4
MoN	3.00	-	-	731.3	-	-
RuN	3.00	-	-	698.7	-	-
Ag₃N	1.00	-	-	238.2	-	-
LaN	3.00	-16.0	0.055	525.8	24.7	135.6
CeN	3.00	8.0	-0.027	521.7	-12.2	-74.8
WN <sub>2</sub>	6.00	-	-	759.1	-	-
NaN <sub>3</sub>	1.00	-	-	179.8	-	-
KN <sub>3</sub>	1.00	-	-	162.3	-	-
AgN₃	1.00	-	-	114.6	-	-
Р						
Be <sub>3</sub> P <sub>2</sub>	2.00	-	-	1807.2	-	-
BP	3.00	-84.9	0.293	1924.2	342.8	-
Na₃P	1.00	-	-	804.5	-	-
$Mg_3P_2$	2.00	-	-	1192.4	-	-
Li₃P	1.00	-	-	1552.3	-	-
Si <sub>3</sub> P <sub>4</sub>	4.00	-	-	1545.1	-	-
K <sub>3</sub> P	1.00	-	-	542.3	-	-
Ca <sub>3</sub> P <sub>2</sub>	2.00	55.3	-0.287	882.7	-195.2	-498.0
ScP	3.00	-	-	1058.9	-	-
TiP	3.00	-	-	1019.8	-	-
(VP)	3.00	-	-	981.6	-	-
CrP	3.00	-	-	969.1	-	-
MnP	3.00	-47.2	0.163	935.9	116.2	511.6
Mn <sub>2</sub> P	1.50	4.5	-0.031	570.9	-15.0	-75.4
FeP	3.00	-	-	926.1	-	-
Fe <sub>2</sub> P	1.50	-	-	563.6	-	-
Fe <sub>3</sub> P	1.00	-	-	405.0	-	-
Co <sub>2</sub> P	1.50	11.5	-0.080	540.2	-36.4	-192.6
Ni <sub>2</sub> P	1.50	9.6	-0.067	542.0	-30.5	-177.0
Ni <sub>3</sub> P	1.00	18.2	-0.188	388.3	-64.7	-
Ni <sub>5</sub> P <sub>2</sub>	1.20	21.1	-0.182	452.5	-71.6	-
CuP <sub>2</sub>	6.00	-204.8	0.354	1281.4	316.9	1140.9
Cu₃P	1.00	-4.2	0.044	362.8	14.2	81.9
Zn <sub>3</sub> P <sub>2</sub>	2.00	-	-	623.0	-	-
GaP	3.00	-66.5	0.230	798.5	144.8	538.5

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GeP	3.00	-140.9	0.487	776.0	299.6	-
Ge <sub>3</sub> P <sub>4</sub>	4.00	-	-	940.9	-	-
As <sub>3</sub> P <sub>5</sub>	5.00	-	-	1059.0	-	-
Se(P) <sub>2</sub>	6.00	-	-	1141.2	-	-
YP	3.00	-	-	670.7	-	-
ZrP <sub>2</sub>	6.00	-	-	1049.9	-	-
NbP	3.00	-	-	649.0	-	-
MoP	3.00	-	-	633.5	-	-
RuP	3.00	-	-	608.9	-	-
AgP <sub>2</sub>	6.00	-914.8	1.580	947.0	766.1	-
AgP <sub>3</sub>	9.00	-1369.2	1.577	1201.3	857.3	-
LaP	3.00	-	-	473.3	-	-
Ce <sub>3</sub> P <sub>4</sub>	4.00	-	-	590.9	-	-
WP <sub>2</sub>	6.00	-	-	654.3	-	-
CI						
BeCl <sub>2</sub>	2.00	26.4	-0.137	670.7	-74.9	-150.5
PCI <sub>5</sub>	5.00	-743.1	1.540	643.5	815.2	1782.5
NaCl	1.00	174.5	-1.809	458.6	-718.8	-1601.8
MgCl <sub>2</sub>	2.00	172.6	-0.894	563.0	-423.5	-1007.1
LiCl	1.00	174.8	-1.812	632.2	-944.9	-2377.5
KCI	1.00	198.9	-2.061	359.5	-661.3	-1353.2
CaCl <sub>2</sub>	2.00	329.6	-1.708	483.0	-709.9	-1571.0
ScCl <sub>3</sub>	3.00	197.1	-0.681	531.4	-307.1	-749.6
TiCl <sub>2</sub>	2.00	45.2	-0.234	451.3	-91.8	-281.5
TiCl₃	3.00	24.7	-0.085	521.3	-37.9	-100.3
VCl <sub>2</sub>	2.00	-13.2	0.068	439.9	26.2	82.6
VCI <sub>3</sub>	3.00	-117.5	0.406	511.2	177.1	522.9
CrCl <sub>2</sub>	2.00	-63.2	0.328	436.1	124.6	355.9
CrCl <sub>3</sub>	3.00	-142.7	0.493	507.7	213.9	588.5
MnCl <sub>2</sub>	2.00	21.3	-0.110	425.9	-41.1	-120.9
FeCl <sub>2</sub>	2.00	-116.9	0.606	422.9	224.4	696.2
FeCl <sub>3</sub>	3.00	-294.8	1.018	495.7	432.8	1242.3
CoCl <sub>2</sub>	2.00	-149.4	0.774	412.8	280.7	916.1
NiCl <sub>2</sub>	2.00	-160.2	0.830	413.6	301.5	1021.1
CuCl	1.00	-89.7	0.930	270.7	230.7	914.7
CuCl <sub>2</sub>	2.00	-243.5	1.262	398.7	443.7	1464.0

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ZnCl <sub>2</sub>	2.00	-49.8	0.258	393.3	89.7	258.4
GaCl₃	3.00	-174.0	0.601	456.6	238.0	594.7
AsCl <sub>5</sub>	5.00	-	-	531.4	-	-
SeCl <sub>4</sub>	4.00	-	-	485.6	-	-
YCl <sub>3</sub>	3.00	298.9	-1.033	411.8	-373.6	-979.2
ZrCl <sub>2</sub>	2.00	-33.2	0.172	330.6	51.2	159.1
ZrCl <sub>3</sub>	3.00	17.2	-0.059	406.9	-21.3	-63.9
ZrCl <sub>4</sub>	4.00	51.5	-0.133	460.0	-53.2	-148.2
NbCl <sub>5</sub>	5.00	-364.8	0.756	496.0	321.5	890.2
NbCl <sub>4</sub>	4.00	-232.9	0.603	456.7	239.0	746.4
MoCl <sub>4</sub>	4.00	-436.4	1.131	450.9	442.9	-
MoCl <sub>5</sub>	5.00	-625.0	1.296	490.5	545.6	1579.9
MoCl <sub>6</sub>	6.00	-866.6	1.497	521.0	663.8	-
RuCl <sub>3</sub>	3.00	-469.1	1.621	387.6	555.9	1694.4
AgCl	1.00	-99.8	1.034	187.0	182.0	952.5
SnCl <sub>2</sub>	2.00	-133.0	0.689	282.7	177.9	675.8
LaCl <sub>3</sub>	3.00	-	-	327.8	-	-
CeCl <sub>3</sub>	3.00	356.0	-1.230	326.2	-361.6	-1372.1
WCl <sub>2</sub>	2.00	-199.2	1.032	210.4	202.9	1034.7
WCl <sub>4</sub>	4.00	-478.4	1.240	329.2	367.5	1585.8
WCl <sub>5</sub>	5.00	-646.0	1.339	371.1	441.9	1635.6
WCl <sub>6</sub>	6.00	-801.6	1.385	405.5	494.2	1678.9
BO <sub>2</sub> -						
NaBO <sub>2</sub>	1.00	-	-	407.3	-	-
Mg(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	487.6	-	-
LiBO <sub>2</sub>	1.00	-	-	538.7	-	-
KBO <sub>2</sub>	1.00	-	-	327.2	-	-
Sc(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	463.7	-	-
Ti (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	489.3	-	-
V(BO <sub>2</sub> ) <sub>5</sub>	5.00	-	-	505.7	-	-
Cr(BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	520.6	-	-
Mn (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	438.5	-	-
Fe (BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	378.9	-	-
Co (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	429.1	-	-
Ni(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	371.4	-	-

Cu(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	359.3	-	-
Zn(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	355.0	-	-
Ga(BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	405.8	-	-
Ge(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	439.6	-	-
Y (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	369.9	-	-
Zr (BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	408.5	-	-
Nb BO <sub>2</sub>	1.00	-	-	197.5	-	-
MoBO <sub>2</sub>	1.00	-	-	193.2	-	-
Ru(BO <sub>2</sub> ) <sub>2</sub>	2.00	-	-	287.1	-	-
Ag BO <sub>2</sub>	1.00	-	-	177.9	-	-
La (BO <sub>2</sub> ) <sub>3</sub>	3.00	-	-	300.8	-	-
Ce(BO <sub>2</sub> ) <sub>4</sub>	4.00	-	-	344.3	-	-
W (BO <sub>2</sub> ) <sub>6</sub>	6.00	-	-	364.9	-	-
$Be_3(BO_3)_2$	2.00	-	-	1111.8	-	-
$Ca_3(BO_3)_2$	2.00	-	-	676.1	-	-
CO3 <sup>2-</sup>						
Be CO <sub>3</sub>	2.00	-	-	776.6	-	-
$B_2(CO_3)_3$	3.00	-	-	797.5	-	-
Na <sub>2</sub> CO <sub>3</sub>	1.00	-	-	505.7	-	-
Mg CO <sub>3</sub>	2.00	-	-	635.8	-	-
Li <sub>2</sub> CO <sub>3</sub>	1.00	-	-	725.4	-	-
K <sub>2</sub> CO <sub>3</sub>	1.00	-	-	387.8	-	-
CaCO <sub>3</sub>	2.00	-	-	535.6	-	-
Sc 2(CO3)3	3.00	-	-	595.7	-	-
Ti (CO <sub>3</sub> ) <sub>2</sub>	4.00	-	-	638.6	-	-
V <sub>2</sub> (CO <sub>3</sub> ) <sub>5</sub>	5.00	-	-	666.8	-	-
Cr (CO <sub>3</sub> ) <sub>3</sub>	6.00	-	-	693.1	-	-
MnCO₃	2.00	-	-	466.3	-	-
$Fe_2(CO_3)_3$	3.00	-	-	551.2	-	-
FeCO <sub>3</sub>	2.00	-	-	450.7	-	-
CoCO <sub>3</sub>	2.00	-	-	450.7	-	-
Ni CO <sub>3</sub>	2.00	-	-	451.6	-	-
CuCO <sub>3</sub>	2.00	-	-	433.8	-	-
Zn CO <sub>3</sub>	2.00	-	-	427.5	-	-

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1	1	1		i.							
Ga <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		50	3.4	-		-	
Ge(CO <sub>3</sub> ) <sub>2</sub>	4.00	-		-		55	6.5	-		-	
Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		44	9.4	-		-	
ZrCO <sub>3</sub>	2.00	-		-		35	4.4	-		-	
Nb <sub>2</sub> CO <sub>3</sub>	1.00	-		-		21	8.1	-		-	
Mo <sub>2</sub> CO <sub>3</sub>	1.00	-		-		21	2.8	-		-	
Ru <sub>2</sub> CO <sub>3</sub>	1.00	-		-		20	4.5	-		-	
Ag <sub>2</sub> CO <sub>3</sub>	1.00	-		-		19	4.4	-		-	
La <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	3.00	-		-		35	1.2	-		-	
$Ce_{2}(CO_{3})_{3}$	3.00	-		-		34	9.4	-		-	
W (CO <sub>3</sub> ) <sub>3</sub>	6.00	-		-		44	1.9	-		-	
PO4 <sup>3-</sup>											
Be <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00	-		-		74	1.1	-		-	
Na₃PO₄	1.00		57.0	-	0.590	49	0.4		-248.6		-636.8
Mg <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	2.00		100.9	-	0.523	61	1.8		-265.5		-603.2
Li <sub>3</sub> PO <sub>4( Δ</sub>											
H=-698.6 kJ -1	1 00		206 5		2 176	60	лл		1700 7		1625 1
	1.00		-300.5		5.170	27	4.4 0 0		1700.7		4025.1
$\Gamma_3 = \Gamma_4$	2.00	-	216.2	-	1 1 2 1	51	0.0 0 1	-	404.0	-	1517 6
$Ca_3(FU_4)_2$	2.00		210.3	-	·1.1Z1	57	0.4 1 G		-494.9	-	-1317.0
	3.00	-		-		57	4.0 1 1	-		-	
$11_3(PO_4)_4$	4.00 5.00	-		-		64	4.4 0 5	-		-	
$V_3 (F \cup_4)_5$	3.00	-		-			0.5	-		-	
	3.00	-		-		52	7.1 6.4	-		-	
$M_{\rm P}({\rm PO})$	2.00	-		-		55	0.4 2.2	-		-	
$\frac{1}{1} \frac{1}{1} \frac{1}{3} \frac{1}{1} \frac{1}{2}$	2.00	-	420.4	-	1 107	40 52	3.3 2.1	-	672.4	-	1011 0
	2.00		-430.4		1.407	12	Э. I 0 Б		072.4		1911.0
$O_3(PO_4)_2$	2.00	-		-		43	0.0 0.2	-		-	
$\frac{NI_3(PO_4)_2}{CU(PO_4)}$	2.00	-		-		43	9.3 2.5	-		-	
$\overline{CU_3(PO_4)_2}$	2.00	-	100.7	-	0.000	42	2.5	-	264.2	-	1004 4
$2n_3(PO_4)_2$	2.00		-190.7		0.988	41	0.0		301.2		1304.4
	3.00	-		-		48	0.Z	-		-	
	4.00	-		-		53	ö.U	-		-	
	3.00	-		-		43	1.3	-		-	
Zr 3(PO4)2	2.00	-		-		34	6.9	-		-	

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Nb <sub>3</sub> PO <sub>4</sub>	1.00	-	-		215.2	-		-	
Mo <sub>3</sub> PO <sub>4</sub>	1.00	-	-		210.0	-		-	
Ru <sub>3</sub> PO <sub>4</sub>	1.00	-	-		201.9	-		-	
$Ag_3 PO_4$	1.00	-	-		192.1	-		I	
La PO₄	3.00	-	-		343.8	-		-	
CePO <sub>4</sub>	3.00	-	-		342.0	-		-	
W (PO <sub>4</sub> ) <sub>2</sub>	6.00	-	-		430.2	-		-	
SO4 <sup>2-</sup>									
Be SO <sub>4</sub>	2.00	-79	.6	0.412	510.1		179.7		454.1
Na <sub>2</sub> SO <sub>4</sub>	1.00	50	.6	-0.524	377.4		-175.7		-474.3
Mg SO <sub>4</sub>	2.00	1	.6	-0.008	445.3		-3.2		-8.6
Li <sub>2</sub> SO <sub>4</sub>	1.00	76	.4	-0.791	487.5		-331.6	-	
AIK(SO <sub>4</sub> ) <sub>2</sub>	4.00	-97	.9	0.254	415.2		92.4	-	
K <sub>2</sub> SO <sub>4</sub>	1.00	76	.2	-0.790	307.6		-220.2		-586.6
CaSO <sub>4</sub>	2.00	140	.1	-0.726	393.7		-252.5		-739.1
Sc 2(SO4)3	3.00	-	-		425.3	-		I	
Ti (SO <sub>4</sub> ) <sub>2</sub>	4.00	-	-		446.7	-		-	
$V_2(SO_4)_5$	5.00	-	-		460.3	-		-	
$Cr_2(SO_4)_3$	3.00	-444	.9	1.537	410.0		554.1		1687.5
MnSO <sub>4</sub>	2.00	-211	.6	1.096	355.0		347.8		1106.4
FeSO <sub>4</sub>	2.00	-348	.2	1.804	352.9		569.3		2003.5
$Fe_2(SO_4)_3$	3.00	-622	.2	2.149	402.1		761.6		2320.3
Co SO <sub>4</sub>	2.00	-386	.6	2.003	345.8		620.8		2217.2
Ni SO4	2.00	-409	.3	2.121	346.4		658.2		2512.6
CuSO <sub>4</sub>	2.00	-506	.8	2.626	335.8		792.7		2760.7
Zn SO <sub>4</sub>	2.00	-297	.5	1.542	332.0		460.5		1681.5
$Y_2(SO_4)_3$	3.00	-	-		345.1	-		-	
Zr(SO <sub>4)2</sub>	4.00	-	-		378.3	-		-	
$Nb_2SO_4$	1.00	-	-		190.2	-		I	
$Mo_2SO_4$	1.00	-	-		186.2	-		-	
$Ru_2SO_4$	1.00	-	-		179.8	-		I	
Ag <sub>2</sub> SO <sub>4</sub>	1.00	-275	.3	2.853	171.9		463.8		2394.7
La 2(SO4)3	3.00	-	-		284.1	-		-	
$Ce_2(SO_4)_3$	3.00	48	.0	-0.166	282.9		-42.8	-	

	W (SO <sub>4</sub> ) <sub>3</sub>	6.00	-	-	340.7	-	-
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		Li			
х	Chemical Reaction	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )
F <sub>2</sub>	2Li+F <sub>2</sub> =2LiF	-1175.4	6.091	1410.7	6293.5
O <sub>2</sub>	4Li+O <sub>2</sub> =2Li <sub>2</sub> O	-1122.4	2.908	3350.3	5216.9
N <sub>2</sub>	6Li+N <sub>2</sub> =2Li <sub>3</sub> N	-257.2	0.444	5740.4	1025.6
S	2Li+S=Li₂S	-439.0	2.275	1671.6	2654.0
В	3Li+B=Li₃B	-	-	7436.6	-
Si	4Li+Si=Li <sub>4</sub> Si	-	-	3817.1	-
Р	3Li+P=Li₃P	-	-	2595.9	-
Cl <sub>2</sub>	2Li+Cl <sub>2</sub> =2LiCl	-768.8	3.984	756.0	2518.7
SF <sub>4</sub>	4Li+SF <sub>4</sub> =4LiF+S	-1628.8	4.220	992.1	3331.1
SF <sub>6</sub>	6Li+SF <sub>6</sub> =6LiF+S	-2409.7	4.162	1101.0	3566.1
$S_2F_{10}$	10Li+S <sub>2</sub> F <sub>10</sub> =10LiF+2S	-4016.0	4.162	1054.7	3448.1
SCl <sub>2</sub>	2Li+SCl <sub>2</sub> =2LiCl+S	-740.3	3.836	520.6	1759.8
$S_2Cl_2$	2Li+S <sub>2</sub> Cl <sub>2</sub> =2LiCl+2S	-729.8	3.782	396.9	1361.3
CIF	2Li+CIF=LiCI+LiF	-920.3	4.769	984.4	3741.1
BrF	2Li+BrF=LiF+LiBr	-875.8	4.538	542.0	2156.9
BrCl	2Li+BrCl=LiCl+LiBr	-780.7	4.046	464.7	1678.0
CIF <sub>3</sub>	4Li+CIF <sub>3</sub> =3LiF+LiCI	-2024.5	5.246	1159.6	4678.1
$BrF_3$	4Li+BrF <sub>3</sub> =3LiF+LiBr	-1919.9	4.975	783.1	3238.7
$CIF_5$	6Li+ClF₅=5LiF+LiCl	-3175.9	5.486	1232.8	5126.3
$BrF_5$	6Li+BrF₅=5LiF+LiBr	-2983.9	5.154	919.5	3827.7
$IF_5$	6Li+IF₅=LiI(s)+5LiF	-	-	724.7	2532.0
SOF <sub>2</sub>	4Li+2SOF <sub>2</sub> =4LiF+S+SO 2	-1646.9	4.267	622.8	2288.7
SOCI <sub>2</sub>	$4Li+2SOCl_2=4LiCl+S+S$ $O_2$ $4Li+2SOBr_2=4LiBr+S+S$	-1441.1	3.734	450.6	1506.6
SOBr <sub>2</sub>	0 <sub>2</sub>	-	-	257.9	-
$SO_2F_2$	2Li+SO <sub>2</sub> F <sub>2</sub> =2LiF+SO <sub>2</sub>	-1475.5	7.646	525.2	3535.1
SO <sub>2</sub> CI	2Li+SO <sub>2</sub> CIF=LiCI+LiF+S	-759.2	3.934	452.3	1592.9

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F	O <sub>2</sub>				
SO <sub>2</sub> Cl <sub>2</sub>	2Li+SO <sub>2</sub> Cl <sub>2</sub> =2LiCl+SO <sub>2</sub>	-748.9	3.881	397.1	1397.6
		Na			
х	Chemical Reaction	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )
F <sub>2</sub>	2Na+F <sub>2</sub> =2NaF	-1092.6	5.662	1410.7	3614.1
O <sub>2</sub>	4Na+O <sub>2</sub> =2Na <sub>2</sub> O	-751.0	1.946	3350.3	1682.9
N <sub>2</sub>	6Na+N <sub>2</sub> =2Na <sub>3</sub> N	187.5	-0.32	5740.4	-313.9
S	2Na+S=Na <sub>2</sub> S	-349.8	1.813	1671.6	1245.0
В	3Na+B=Na₃B	-	-	7436.6	-
Si	4Na+Si=Na <sub>4</sub> Si	-	-	3817.1	-
Р	3Na+P=Na₃P	-	-	2595.9	-
Cl <sub>2</sub>	2Na+Cl <sub>2</sub> =2NaCl	-768.2	3.981	756.0	1825.6
SF <sub>4</sub>	4Na+SF <sub>4</sub> =4NaF+S	-1463.2	3.791	992.1	2032.0
SF <sub>6</sub>	6Na+SF <sub>6</sub> =6NaF+S	-2161.3	3.733	1101.0	2114.0
$S_2F_{10}$	10Na+S <sub>2</sub> F <sub>10</sub> =10NaF+2S	-3602.0	3.733	1054.7	2067.2
SCI <sub>2</sub>	2Na+SCl <sub>2</sub> =2NaCl+S	-739.7	3.833	520.6	1379.5
$S_2Cl_2$	2Na+S2Cl2=2NaCl+2S	-729.2	3.779	396.9	1119.0
CIF	2Na+CIF=NaCI+NaF	-878.6	4.553	984.4	2430.1
BrF	2Na+BrF=NaF+NaBr	-786.1	4.074	542.0	1507.2
BrCl	2Na+BrCl=NaCl+NaBr	-732.1	3.794	464.7	1260.5
CIF <sub>3</sub>	4Na+CIF <sub>3</sub> =3NaF+NaCl	-1900.0	4.923	1159.6	2862.0
$BrF_3$	4Na+BrF <sub>3</sub> =3NaF+NaBr	-1747.4	4.528	783.1	2120.9
$CIF_5$	6Na+ClF₅=5NaF+NaCl	-2968.6	5.128	1232.8	3072.5
$BrF_5$	6Na+BrF₅=5NaF+NaBr	-2728.6	4.713	919.5	2422.8
$IF_5$	6Na+IF₅=Nal(s)+5NaF	-	-	724.7	2532.0
SOF <sub>2</sub>	4Na+2SOF <sub>2</sub> =4NaF+S+ SO <sub>2</sub> 4Na+2SOCl <sub>2</sub> =4NaCl+S	-1481.3	3.838	622.8	1558.1
SOCI <sub>2</sub>	+SO <sub>2</sub>	-1439.9	3.731	450.6	1212.4
SOBr <sub>2</sub>	4Na+2SOBr <sub>2</sub> =4NaBr+S +SO <sub>2</sub>	-	-	257.9	-
$SO_2F_2$	$2Na+SO_2F_2=2NaF+SO_2$	-1392.7	7.217	525.2	2613.3
SO <sub>2</sub> CI	2Na+SO <sub>2</sub> CIF=NaCI+Na	-717.5	3.718	452.3	1211.7

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F	F+SO <sub>2</sub>				
SO <sub>2</sub> Cl <sub>2</sub>	2Na+SO <sub>2</sub> Cl <sub>2</sub> =2NaCl+S O <sub>2</sub>	-748.3	3.878	397.1	1148.8
		Mg			
х	chemical reaction	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )
$F_2$	Mg+F <sub>2</sub> =MgF <sub>2</sub>	-1071.1	5.551	1410.7	4775.6
O <sub>2</sub>	2Mg+O <sub>2</sub> =2MgO	-1138.6	2.950	3350.3	3923.6
N <sub>2</sub>	$3Mg+N_2=Mg_3N_2$	-400.9	0.693	5740.4	1103.4
S	Mg+S=MgS	-341.8	1.771	1671.6	1684.3
В	3/2Mg+B=1/2Mg <sub>3</sub> B <sub>2</sub>	-	-	7436.6	-
Si	2Mg+Si=Mg <sub>2</sub> Si	-	-	3817.1	-
Р	3/2Mg+P=1/2Mg <sub>3</sub> P <sub>2</sub>	-	-	2595.9	-
Cl <sub>2</sub>	Mg+Cl <sub>2</sub> =MgCl <sub>2</sub>	-591.8	3.067	756.0	1726.6
SF <sub>4</sub>	2Mg+SF <sub>4</sub> =2MgF <sub>2</sub> +S	-1420.2	3.680	992.1	2518.0
SF <sub>6</sub>	3Mg+SF <sub>6</sub> =3MgF <sub>2</sub> +S	-2096.8	3.622	1101.0	2659.9
$S_2F_{10}$	5Mg+S <sub>2</sub> F <sub>10</sub> =5MgF <sub>2</sub> +2S	-3494.5	3.622	1054.7	2584.1
SCl <sub>2</sub>	Mg+SCl <sub>2</sub> =MgCl <sub>2</sub> +S	-563.3	2.919	520.6	1229.4
$S_2Cl_2$	Mg+S <sub>2</sub> Cl <sub>2</sub> =MgCl <sub>2</sub> +2S	-552.8	2.865	396.9	963.7
CIF	Mg+CIF=1/2MgCl <sub>2</sub> +1/2 MgF <sub>2</sub>	-779.7	4.040	984.4	2749.9
BrF	$aBr_2$	-678.3	3.515	542.0	1529.2
BrCl	Mg+BrCl=1/2MgCl <sub>2</sub> +1/2 MgBr <sub>2</sub>	-546.8	2.834	464.7	1087.6
CIF <sub>3</sub>	2Mg+CIF <sub>3</sub> =3/2MgF <sub>2</sub> +1/2 MgCl <sub>2</sub>	-1779.6	4.611	1159.6	3504.4
BrF₃	2Mg+BrF <sub>3</sub> =3/2MgF <sub>2</sub> +1/2 MgBr <sub>2</sub>	-1618.1	4.192	783.1	2422.8
$CIF_5$	$3Mg+CIF_5=5/2MgF_2+1/2$ $MgCl_2$	-2826.7	4.883	1232.8	3861.0
BrF₅	3NIG+Br <sub>5</sub> =5/2MGF <sub>2</sub> +1/2 MgBr <sub>2</sub>	-2577.8	4.453	919.5	2889.5
IF₅	2MgF <sub>2</sub>		0.000	724.7	2532.0
$SOF_2$	$2Mg+2SOF_2=2MgF_2+S+$	-1438.3	3.727	622.8	1810.0

	SO <sub>2</sub>				
	2Mg+2SOCl <sub>2</sub> =2MgCl <sub>2</sub> +				
SOCI <sub>2</sub>	S+SO <sub>2</sub>	-1087.1	2.817	450.6	1053.8
	2Mg+2SOBr <sub>2</sub> =2MgBr <sub>2</sub> +				
SOBr <sub>2</sub>	S+SO <sub>2</sub>	-	-	257.9	-
$SO_2F_2$	$Mg+SO_2F_2=MgF_2+SO_2$	-1371.2	7.106	525.2	3014.2
SO <sub>2</sub> CI	Mg+SO <sub>2</sub> CIF=1/2MgCl <sub>2</sub> +				
F	$1/2MgF_2+SO_2$	-618.6	3.206	452.3	1203.1
$SO_2CI_2$	Mg+SO <sub>2</sub> Cl <sub>2</sub> =MgCl <sub>2</sub> +SO <sub>2</sub>	-571.9	2.964	397.1	997.5

		AI			
х	Chemical Reaction	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (m A h g <sup>-1</sup> )	Cal. Energy density (W h kg <sup>-1</sup> )
$F_2$	2/3AI+F <sub>2</sub> =2/3AIF <sub>3</sub>	-954.1	4.944	1410.7	4733.8
O <sub>2</sub>	4/3AI+O <sub>2</sub> =2/3AI <sub>2</sub> O <sub>3</sub>	-1054.8	2.733	3350.3	4310.7
$N_2$	2AI+N <sub>2</sub> =2AIN	-574.0	0.992	5740.4	1945.0
S	2/3AI+S=1/3AI <sub>2</sub> S <sub>3</sub>	-213.3	1.106	1671.6	1183.9
В	AI+B=AIB	-	-	7436.6	-
Si	$4/3AI+Si=1/3AI_4Si_3$	-	-	3817.1	-
Р	AI+P=AIP	-157.9	0.545	2595.9	756.6
Cl <sub>2</sub>	2/3AI+CI <sub>2</sub> =2/3AICI <sub>3</sub>	-419.2	2.172	756.0	1309.9
SF <sub>4</sub>	4/3AI+SF <sub>4</sub> =4/3AIF <sub>3</sub> +S	-1186.1	3.073	992.1	2287.5
$SF_6$	$2AI+SF_6=2AIF_3+S$	-1745.7	3.015	1101.0	2424.3
$S_2F_{10}$	10/3Al+S <sub>2</sub> F <sub>10</sub> =10/3AlF <sub>3</sub> + 2S	-2909.3	3.015	1054.7	2348.9
SCl <sub>2</sub>	2/3AI+SCI2=2/3AICI3+S	-390.7	2.025	520.6	897.2
$S_2Cl_2$	$2/3AI+S_2CI_2=2/3AICI_3+2$ S	-380.2	1.970	396.9	690.2
CIF	2/3AI+CIF=1/3AICI <sub>3</sub> +1/3 AIF <sub>3</sub>	-634.9	3.290	984.4	2434.4
BrF	2/3Al+BrF=1/3AlF <sub>3</sub> +1/3 AlBr <sub>3</sub>	-530.7	2.750	542.0	1261.1
BrCl	2/3Al+BrCl=1/3AlCl <sub>3</sub> +1/ 3AlBr <sub>3</sub>	-371.5	1.925	464.7	773.8
$CIF_3$	4/3AI+CIF <sub>3</sub> =AIF <sub>3</sub> +1/3AIC	-1517.7	3.932	1159.6	3282.8

	$4/3AI+BrF_3=AIF_3+1/3AIB$				
BrF₃	r <sub>3</sub>	-1353.4	3.507	783.1	2174.7
	2AI+CIF <sub>5</sub> =5/3AIF <sub>3</sub> +1/3AI				
$CIF_5$	Cl <sub>3</sub>	-2447.8	4.228	1232.8	3687.1
	2AI+BrF <sub>5</sub> =5/3AIF <sub>3</sub> +1/3AI				
$BrF_5$	Br <sub>3</sub>	-2196.1	3.794	919.5	2665.5
	2AI+IF <sub>5</sub> =1/3AII <sub>3</sub> (s)+5/3AI				
$IF_5$	F <sub>3</sub>	-	-	724.7	2532.0
	4/3AI+2SOF <sub>2</sub> =4/3AIF <sub>3</sub> +				
SOF <sub>2</sub>	S+SO <sub>2</sub>	-1204.3	3.120	622.8	1607.5
	4/3AI+2SOCI <sub>2</sub> =4/3AICI <sub>3</sub>				
SOCI <sub>2</sub>	+S+SO <sub>2</sub>	-741.9	1.922	450.6	752.4
	4/3AI+2SOBr <sub>2</sub> =4/3AIBr <sub>3</sub>				
SOBr <sub>2</sub>	+S+SO <sub>2</sub>	-	-	257.9	-
	$2/3AI+SO_2F_2=2/3AIF_3+S$				
$SO_2F_2$	O <sub>2</sub>	-1254.2	6.499	525.2	2902.0
SO <sub>2</sub> CI	2/3AI+SO <sub>2</sub> CIF=1/3AICI <sub>3</sub>				
F	+1/3AIF <sub>3</sub> +SO <sub>2</sub>	-473.8	2.455	452.3	964.1
	2/3AI+SO <sub>2</sub> CI <sub>2</sub> =2/3AICI <sub>3</sub> +				
$SO_2Cl_2$	SO <sub>2</sub>	-399.3	2.069	397.1	725.2

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FUEL	∆ <sub>r</sub> G (kJ mol⁻¹)	n	EMF(V )	Q=∆H (kJ mol⁻¹)	Cal. Energy density (W h kg <sup>-1</sup> )
В	-592.000	3.00	2.045	-636.750	4723.91
Si	-805.000	4.00	2.086	-910.700	3721.63
V	-567.000	3.00	1.959	-609.400	2101.67
Ті	-853.000	4.00	2.210	-944.000	2966.78
TiB <sub>2</sub>	-1737.000	10.00	1.800	-1937.500	3227.69
VB <sub>2</sub>	-	9.00			
AIB <sub>2</sub>	-	9.00			
Li	-280.600	1.00	2.908	-293.950	5216.92
	-578.900	4.00	1.500	-634.300	3504.86
Na	-187.750	1.00	1.946	-207.100	1682.92
Mg	-569.300	2.00	2.950	-601.600	3923.61
AI	-791.150	3.00	2.733	-837.850	4310.74

Zn	-320.500	2.00	1.661	-350.500	1093.85
H <sub>2</sub>	-228.600	2.00	1.185	-285.830	3524.78
СО	-257.184	2.00	1.333	-282.964	1623.28
coal <sub>(C; graphite)</sub>	-394.364	4.00	1.022	-393.505	2489.13
$C_2H_2$	-1235.070	10.00	1.280	-1299.571	3235.51
C <sub>2</sub> H <sub>4</sub>	-1331.430	12.00	1.150	-1411.137	2981.40
$_{(allenic)}C_{3}H_{4}$	-1859.802	16.00	1.205	-1944.304	3073.99
$_{(propine)}C_3H_4$	-1851.861	16.00	1.200	-1937.610	3060.86
$_{(cyclopropane)}C_{3}H_{6}$	-1998.793	18.00	1.151	-2091.225	2983.86
$_{(propylene)}C_{3}H_{6}$	-1957.334	18.00	1.127	-2058.423	2921.97
CH <sub>4</sub>	-817.889	8.00	1.060	-890.292	2838.47
C <sub>8</sub> H <sub>18</sub>	-5305.899	50.00	1.100	-5512.063	2866.24
C <sub>8</sub> H <sub>18</sub>	-5295.888	50.00	1.098	-5470.558	2860.83
(methanol)CH3OH	-702.046	6.00	1.213	-725.965	2436.44
$_{(ethanol)}C_2H_5OH$	-1331.951	12.00	1.150	-1409.694	2604.35
$_{(ethanol)}C_2H_5OH$	-1326.160	12.00	1.145	-1367.519	2593.02
$(dimethyl ether)C_2H_6O$	-1387.551	12.00	1.198	-1460.400	2713.06
C <sub>7</sub> H <sub>7</sub> NO <sub>2</sub>	-	70.00	-	-3320.945	-
TNT	-	33.00	-	-3353.825	-
$N_2H_4$	-521.058	12.00	0.450	-556.070	1130.40
(CH <sub>3</sub> ) <sub>2</sub> NNH <sub>2</sub>	-	24.00	-	-1957.040	-
(CH <sub>3</sub> ) <sub>2</sub> NNH <sub>3</sub>	-	24.00	-	-1916.840	-
Si <sub>2</sub> H <sub>6</sub>	-2551.426	14.00	1.889	-2759.504	4068.15
B <sub>2</sub> H <sub>6</sub>	-1990.987	12.00	1.720	-2165.032	4472.07
(CH <sub>3</sub> CH <sub>2</sub> ) <sub>3</sub> N	-4119.840	43.00	0.993	-4447.860	2570.66

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Typical Battery System								
Battery Type	Δ <sub>r</sub> G (kJ mol <sup>-1</sup> )	EMF (V)	Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )	Pra./Cal (max) (%)	*		
secondary battery								
Pd-acid	-393.7	2.04	170.2	768.3	29.4			

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Na/S	-405.2	2.10	791.7	1178.7	21.5	14
Na/NiCl <sub>2</sub>	-509.2	2.64	805.6	1231.4	14.3	14
H <sub>2</sub> -NiOOH	-132.8	1.38	397.9	1850.7	16.1	
Cd-NiOOH	-248.9	1.29	208.4	778.2	26.4	
Zn-NiOOH	-273.0	1.41	266.2	891.8	31.9	
Fe-NiOOH	-985.3	1.40	281.2	1082.8	19.6	
Fe-O <sub>2</sub>	-505.7	1.31	781.6	2796.1	10.2	
	-1015.4	1.32	1218.2	13232.8	6.6	
Fe/AgO	-266.7	1.38	374.4	1738.8	28.0	
LaNi <sub>5</sub> (H <sub>2</sub> )-NiOOH	-125.4	1.30	240.0	-	37.5	14
primary battery						
Li-O <sub>2</sub>	-1122.4	2.91	5216.9	5996.6	6.9	17
	-571.0	2.96	3456.8	6100.9	10.5	17
Li/S	-439.0	2.27	2654.0	2901.4	0.0	
Li/I <sub>2</sub>	-270.3	2.80	561.0	1938.2	48.1	
Li/MnO <sub>2</sub>	-327.8	3.40	969.8	3023.5	23.7	
Li/(CF) <sub>n</sub>	-299.1	3.10	2189.3	3393.6	26.9	
Li-SO <sub>2</sub>	-598.2	3.10	1170.1	6392.0	22.2	
Li-SOCI <sub>2</sub>	-1441.1	3.73	1506.3	2021.7	22.6	
Li-SO <sub>2</sub> Cl <sub>2</sub>	-748.9	3.88	1397.3	1938.4	34.4	
Li/CuO	-431.5	2.24	1282.2	3118.6	21.8	
Li/CuS	-385.4	2.00	977.8	2288.5		
Li/FeS	-338.6	1.75	924.1	2122.7		
Li/FeS <sub>2</sub>	-721.9	1.87	1358.0	2622.6	19.1	
Li/AgV <sub>2</sub> O <sub>5.5</sub>	-1094.1	3.24	943.9	-	28.6	
Li/V <sub>2</sub> O <sub>5</sub>	-328.0	3.40	482.5	1434.5	12.4	
Li/Lil(Al2O3)/Pbl						
	-367.0	1.90185	214.7	1011.0		
PbS,Pb	-367.0	1.90185	214.7	1011.0		
, -	-340.3	1.76	373.4	1632.7		
AI-O <sub>2</sub>	-3585.4	3.10	3192.0	6727.8	12.5	
Zn-O <sub>2</sub>	-320.5	1.66	1093.9	9721.0	33.8	
Zn-Br <sub>2</sub>	-355.0	1.84	437.8	an	17.1	
Zn/HgO	-262.0	1.36	258.1	2527.6	40.7	

			1			
Zn/Ag <sub>2</sub> O	-309.3	1.60	289.2	2078.0	76.1	
Zn/MnO <sub>2</sub>	-271.4	1.41	315.1	1737.7	0.0	
Zn/KOH/MnO <sub>2</sub>	-308.8	1.60	311.6	1080.4	46.5	14
Cd/HgO	-142.0	0.91	113.7	746.9	48.4	
Cd/AgO	-214.3	1.11	234.1	1168.4		
$H_2-O_2$	-228.6	1.18	3524.8	g	11.3	
Mg/AgCl	-497.5	2.58	444.4	2108.6	33.8	
Mg/CuCl	-352.0	1.82	439.8	1581.9	18.2	
Mg/PbCl <sub>2</sub>	-403.6	2.09	370.7	1822.2	21.6	
Mg/Cu <sub>2</sub> I <sub>2</sub>	-419.1	2.17	287.3	1434.4	27.8	
Mg/CuSCN	-250.9	1.30	260.4	700.6	30.7	
Mg/MnO <sub>2</sub>	-511.5	2.65	657.2	2207.0	17.5	

Li-ion battery									
Battery Type	Battery Type $ \begin{array}{ c c c } \Delta_r G & E_{aver} & Capacity \\ (kJ \text{ mol}^{-1}) & (V) & (m \text{ A h g}^{-1}) \\ (m \text{ A h g}^{-1}) & (W) & (m \text{ A h g}^{-1}) \end{array} $		Cal. Energy density (W h kg <sup>-1</sup> )	Cal. Energy density (W h L <sup>-1</sup> )	*				
Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub> -Li	-964.9	2.500	589.4	1278.43	-	18			
V <sub>2</sub> O <sub>5</sub> -Li	-781.5	2.700	442.1	1070.98	2327.14	4			
LiCoO <sub>2</sub> -Li	-357.0	3.700	273.8	1013.20	5228.13	1-3			
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub> - Li	-347.3	3.600	277.8	1000.25	4756.17	1,15			
MnO <sub>2</sub> -Li	-327.8	3.397	308.3	969.79	3023.47	4			
LiMnO <sub>2</sub> -Li	-327.8	3.397	285.5	969.79	4095.42	4			
V <sub>2</sub> O <sub>5</sub> -Li <sub>4.4</sub> Si	-1128.9	2.659	442.1	963.70	2425.72	8			
LiNiO <sub>2</sub> - Li	-337.7	3.500	274.5	960.79	4592.57	1,11			
LiCoO <sub>2</sub> -Si	-1553.4	3.659	273.8	940.65	4517.78	8,11			
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub> - Si	-1510.9	3.559	277.8	927.50	4142.65	8,15			
0.3Li <sub>2</sub> MnO <sub>3</sub> -0.7Li[ MnNiCo] <sub>0.333</sub> O <sub>2</sub> / Li	-337.7	3.500	261.3	914.46	4066.59	21,2 2			
MnO <sub>2</sub> -Li <sub>4.4</sub> Si	-1424.7	3.356	308.3	897.10	3167.55	8			
LiMnO <sub>2</sub> -Si	-1424.7	3.356	285.5	897.10	3602.15	8			

LiNiO <sub>2</sub> - Si	-1468.5	3.459	274.5	891.28	4002.09	8,11
0.3Li <sub>2</sub> MnO <sub>3</sub> -0.7Li[						21,2
MnNiCo] <sub>0.333</sub> O <sub>2</sub> / Si	-333.7	3.450	261.3	851.00	3593.6	2
LiCoPO <sub>4</sub> -Li	-453.5	4.700	166.6	783.15	2954.83	5,12
LiCoPO <sub>4</sub> -Si	-1977.9	4.659	166.6	746.70	2752.24	8
LiCoO <sub>2</sub> -Sn	-1393.0	3.281	273.8	704.35	3878.54	9,11
LiMnPO <sub>4</sub> -Li	-395.6	4.100	170.9	700.58	2473.74	5
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>	-1350 5	3 181	277 8	690 70	3553 65	9,15
LiNio = Mn1 = O4- Li	-449.6	4 660	146 7	683.63	3042 17	6
	-1723.2	4 059	170.9	666.46	2306 78	8
LiNiO - Sn	-1308 1	3 081	274 5	662.69	3422 12	9.11
LiMnO <sub>2</sub> -Sn	-1264.3	2 978	285.5	660.42	3077 44	9
MnO <sub>2</sub> -Li <sub>4</sub> Sn	-1264.3	2 978	308.3	660.42	2629 58	9
LiNio 5Mn4 504- Si	-1960.9	4 619	146 7	654 75	2826.81	8
V <sub>2</sub> O <sub>5</sub> -Li <sub>4</sub> 4Sn	-968.4	2 281	442.1	646.65	1950.99	9
	-1817 5	4 281	166.6	610.90	2476 27	9
	-385.4	3 995	148.2	592.09	2534 14	11
LiFePO <sub>4</sub> -Li	-332.9	3 4 5 0	169.9	586.12	2107 10	5
	002.0	0.100	100.0	000.12	2101.10	7,11
LiCoO <sub>2</sub> -C <sub>6</sub>	-347.4	3.600	273.8	567.79	1901.04	,16
LiMn <sub>2</sub> O <sub>4</sub> - Si	-1678.4	3.954	148.2	566.04	2355.43	8,11
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>	-337 7	3 500	277 8	556 65	1801 43	7,15
LiFePO₄-Si	-1447.3	3.409	169.9	556.64	1959.75	8
LiNi <sub>0 5</sub> Mn <sub>1 5</sub> O <sub>4</sub> - Sn	-1800.5	4.241	146.7	542.13	2539.51	9
LiNiO <sub>2</sub> - C <sub>6</sub>	-328.1	3.400	274.5	537.01	1745.31	7,11
LiMnPO₄-Sn	-1562.8	3.681	170.9	536.70	2050.02	9
MnO <sub>2</sub> -LiC <sub>6</sub>	-318.1	3.297	308.3	532.51	1669.57	7
LiMnO <sub>2</sub> -C <sub>6</sub>	-318.1	3.297	285.5	532.51	1635.84	7
LiCoPO <sub>4</sub> -C <sub>6</sub>	-443.8	4.600	166.6	529.35	1656.71	7
V <sub>2</sub> O <sub>5</sub> -LiC <sub>6</sub>	-752.6	2.600	442.1	499.08	1292.64	7
LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub> - C <sub>6</sub>	-440.0	4.560	146.7	479.75	1677.84	7
LiMnPO <sub>4</sub> -C <sub>6</sub>	-386.0	4.000	170.9	468.34	1406.78	7
LiMn <sub>2</sub> O <sub>4</sub> - Sn	-1518.0	3.576	148.2	461.19	2085.50	9,11
LiFePO₄-Sn	-1286.8	3.031	169.9	439.76	1707.14	9

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LiMn <sub>2</sub> O <sub>4</sub> - C <sub>6</sub>	-375.8	3.895	148.2	412.78	1409.91	7,11
LiFePO <sub>4</sub> -C <sub>6</sub>	-323.2	3.350	169.9	390.69	1186.57	7
Li <sub>2</sub> C <sub>6</sub> H <sub>4</sub> O <sub>4</sub> -Li	-162.1	1.400	209.0	277.53	-	19
LiCoPO <sub>4</sub> -Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-303.9	3.150	166.6	268.97	975.53	10
Li <sub>2</sub> C <sub>8</sub> H <sub>4</sub> O <sub>4</sub> -Li	-177.5	0.800	346.5	254.36	-	19
LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub> - Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-300.1	3.110	146.7	248.28	981.01	10,1 3
Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub> -Li <sub>6</sub> C <sub>6</sub> O <sub>6</sub>	-347.3	1.800	294.7	246.44	-	20
						10,1
LiCoO <sub>2</sub> -Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-207.4	2.150	273.8	229.66	916.41	1
LiMnPO <sub>4</sub> -Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-246.0	2.550	170.9	220.55	773.71	10
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>						10,1
- Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-197.8	2.050	277.8	220.22	855.84	5
						10,1
LiNiO <sub>2</sub> - Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-188.1	1.950	274.5	208.50	812.33	1
MnO <sub>2</sub> -Li <sub>7/3</sub> Ti <sub>5/3</sub> O <sub>4</sub>	-178.5	1.850	308.3	200.81	812.49	10
$LiMnO_2$ - $Li_4Ti_5O_{12}$	-178.5	1.850	285.5	200.81	749.65	10
						10,1
LiMn <sub>2</sub> O <sub>4</sub> - Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-235.4	2.440	148.2	195.89	759.02	1
LiFePO <sub>4</sub> -Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	-183.3	1.900	169.9	163.85	580.03	10
V <sub>2</sub> O <sub>5</sub> - Li <sub>7/3</sub> Ti <sub>5/3</sub> O <sub>4</sub>	-332.9	1.150	442.1	139.72	497.05	

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Table S2. Thermodynamic data of typical battery systems. The volumetric energy densities of the batteries containing gas components are not calculated.

				Cal.	Cal.
Battery	Electrophomical Bassian	ΔG	EMF	Energy	Energy
Туре	Electrochemical Reaction	(kJ mol⁻¹)	(V)	Density	Density
				(W h kg⁻¹)	(W h L⁻¹)
Li/ F <sub>2</sub>	2Li+F <sub>2</sub> =2LiF	-1175	6.090	6294	-
Li/ O <sub>2</sub>	4Li +O <sub>2</sub> =2Li <sub>2</sub> O	-1122	2.910	5217	-
Li/ CIF₅(g)	6Li+ClF₅=5LiF+LiCl	-3176	5.490	5126	-
Mg/ F <sub>2</sub>	Mg+F <sub>2</sub> =MgF <sub>2</sub>	-1071	5.551	4776	-
Al/ F <sub>2</sub>	2/3AI+F <sub>2</sub> =2/3AIF <sub>3</sub>	-954.1	4.944	4734	-
B <sub>2</sub> H <sub>6</sub> (g)/ O <sub>2</sub>	$B_2H_6+3O_2=B_2O_3+3H_2O(I)$	-1991	1.720	4472	-
Al/ O <sub>2</sub>	4/3Al+O <sub>2</sub> =2/3Al <sub>2</sub> O <sub>3</sub>	-1055	2.733	4311	-
Mg/ O <sub>2</sub>	2Mg+O <sub>2</sub> =2MgO	-1139	2.950	3924	-
Na/ F <sub>2</sub>	2Na+F <sub>2</sub> =2NaF	-1093	5.660	3614	-
Li/ SO <sub>2</sub> F <sub>2</sub> (I)	2Li+SO <sub>2</sub> F <sub>2</sub> =2LiF+SO <sub>2</sub>	-1476	7.650	3535	4630
H <sub>2</sub> / O <sub>2</sub>	$H_2 + 1/2O_2 = H_2O$	-228.6	1.185	3525	-
	C <sub>8</sub> H <sub>18</sub> +25/2O <sub>2</sub> =8CO <sub>2</sub> +9H <sub>2</sub>	5206	1 100	2866	
C <sub>8</sub> H <sub>18</sub> / O <sub>2</sub>	O(I)	-3300	1.100		-
CH <sub>4</sub> / O <sub>2</sub>	$CH_4+2O_2=CO_2+2H_2O(I)$	-817.9	1.060	2839	-
Li/ S	2Li+S=Li <sub>2</sub> S	-439.0	2.275	2654	2856
$C_2H_5OH(g)$	C <sub>2</sub> H <sub>6</sub> O	1332	1 150	2604	
/ O <sub>2</sub>	+3O <sub>2</sub> =2CO <sub>2</sub> +3H <sub>2</sub> O(I)	-1332	1.150		-
Li/ RuF₅	5Li+RuF₅=5LiF+Ru	-2157	4.472	2597	5199
Li/ CrO <sub>3</sub>	6Li+CrO <sub>3</sub> =3Li <sub>2</sub> O+Cr	-1171	2.023	2297	2828
Li/ (CF) <sub>n</sub>	nLi+(CF) <sub>n</sub> =nLiF+nC	-299.1	3.100	2189	3394
Li/ B <sub>2</sub> S <sub>3</sub>	6Li+B <sub>2</sub> S <sub>3</sub> =3Li <sub>2</sub> S+2B	-534.7	1.847	1863	2017
Li/ Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub>	$Li_2C_6O_6$ +4Li= $Li_6C_6O_6$	-964.9	2.500	1278	-
C <sub>6</sub> / LiCoO <sub>2</sub>	$CoO_2$ +LiC <sub>6</sub> =LiCoO <sub>2</sub> +C <sub>6</sub>	-347.4	3.600	567.8	1901

Battery type	Electrochemical reaction	ΔG (kJ mol <sup>-1</sup> )	EMF (V)	Energy Density (Wh kg <sup>-1</sup> )	Capacity (+) (mAh g <sup>-1</sup> )
Li/ F <sub>2</sub>	2Li+F <sub>2</sub> =2LiF	-1175	6.090	6294	1411
Li/ O <sub>2</sub>	4Li+O <sub>2</sub> =2Li <sub>2</sub> O	-1122	2.910	5217	3350
Li/ CIF <sub>5</sub>	6Li+CIF <sub>5</sub> =5LiF+LiCl	-3176	5.490	5126	1233
Li/ S	2Li+S=Li <sub>2</sub> S	-439.0	2.275	2654	1672
Li/ RuF₅	5Li+RuF₅=5LiF+Ru	-2157	4.472	2597	683.5
Li/ Cl <sub>2</sub>	2Li+Cl <sub>2</sub> =2LiCl	-768.8	3.980	2519	756.0
Li/ CrO <sub>3</sub>	6Li+CrO <sub>3</sub> =3Li <sub>2</sub> O+Cr	-1171	2.023	2297	1608
Li/ (CF) <sub>n</sub>	nLi+(CF) <sub>n</sub> =nLiF+nC	-299.1	3.100	2189	864.6
Li/ CoF <sub>3</sub>	3Li+CoF <sub>3</sub> =3LiF+Co	-1044	3.607	2121	693.6
Li/ P <sub>2</sub> S <sub>5</sub>	5Li+1/2 P <sub>2</sub> S <sub>5</sub> =5/2 Li <sub>2</sub> S+P	-1208	2.131	1958	1206
Li/ P <sub>4</sub> O <sub>10</sub>	5Li+1/4 P <sub>4</sub> O <sub>10</sub> =5/2 Li <sub>2</sub> O+P	-722.2	1.497	1898	1888
Li/ B <sub>2</sub> S <sub>3</sub>	3Li+1/2B <sub>2</sub> S <sub>3</sub> =3/2Li <sub>2</sub> S+B	-534.7	1.847	1863	1365
Li/ CuF <sub>2</sub>	2Li+CuF <sub>2</sub> =2LiF+Cu	-683.4	3.542	1645	527.9
Li/ FeF <sub>3</sub>	3Li+FeF <sub>3</sub> =3LiF+Fe	-791.1	2.733	1644	712.6
Li/ MnO <sub>2</sub>	4Li+MnO <sub>2</sub> =2Li <sub>2</sub> O+Mn	-657.3	1.703	1592	1233
Li/ V <sub>2</sub> O <sub>5</sub>	5Li+1/2V <sub>2</sub> O <sub>5</sub> =5/2Li <sub>2</sub> O+V	-693.3	1.437	1533	1474
Li/ Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub>	$Li_2C_6O_6$ +4Li= $Li_6C_6O_6$	-964.9	2.500	1278	589.4
Li/ SO <sub>2</sub>	2Li+2SO <sub>2</sub> =Li <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	-598.2	3.100	1170	418.8
Li/ AgO	2Li+AgO=Li <sub>2</sub> O+Ag	-575.0	2.980	1160	432.7
Li/ MnO <sub>2</sub>	Li+MnO <sub>2</sub> =LiMnO <sub>2</sub>	-327.8	3.400	969.8	308.1
C <sub>6</sub> / LiCoO <sub>2</sub>	$CoO_2$ +Li $C_6$ =Li $CoO_2$ + $C_6$	-347.4	3.600	567.8	237.8

Table S3. Thermodynamic data of selected typical lithium battery systems.

	Metallic lithium battery							
Battery Type	Electrochemical Reaction	∆G <sub>r</sub> (kJ mol⁻¹)	EMF (V)	Energy Density (Wh kg <sup>-1</sup> )	Capacity (+) (mAh g <sup>-1</sup> )			
Li/ F <sub>2</sub>	2Li+F <sub>2</sub> =2LiF	-1175.4	6.090	6293.5	1410.7			
Li/ O <sub>2</sub>	4Li+O <sub>2</sub> =2Li <sub>2</sub> O	-1122.4	2.910	5216.9	3350.3			
Li/ CIF <sub>5</sub>	6Li+ClF₅=5LiF+LiCl	-3175.9	5.490	5126.3	1232.8			
Li/CIF <sub>3</sub>	4Li+CIF <sub>3</sub> =3LiF+LiCl	-2024.5	5.250	4678.1	1159.6			
Li/ SO <sub>2</sub> F <sub>2</sub>	$2Li+SO_2F_2=2LiF+SO_2$	-1475.5	7.650	3535.1	525.2			
Li/BrF₅	6Li+BrF₅=5LiF+LiBr	-2983.9	5.150	3827.7	919.5			
Li/(g)CIF	2Li+CIF=LiCI+LiF	-920.3	4.770	3741.2	984.4			
Li/(g)SF <sub>6</sub>	6Li+SF <sub>6</sub> =6LiF+S	-2409.7	4.160	3566.1	1101.0			
Li/ S <sub>2</sub> F <sub>10</sub>	10Li+S <sub>2</sub> F <sub>10</sub> =10LiF+2S	-4016.0	4.160	3448.1	1054.7			
Li/ (g)SF <sub>4</sub>	4Li+SF <sub>4</sub> =4LiF+S	-1628.8	4.220	3331.1	992.1			
Li/ BrF <sub>3</sub>	4Li+BrF <sub>3</sub> =3LiF+LiBr	-1919.9	4.970	3238.7	783.1			
Li/ S	2Li+S=Li <sub>2</sub> S	-439.0	2.270	2654.0	1671.6			
Li/ RuF <sub>5</sub>	5Li+RuF₅=5LiF+Ru	-2157.2	4.472	2596.7	683.5			
Li/ Cl <sub>2</sub>	2Li+Cl <sub>2</sub> =2LiCl	-768.8	3.980	2518.7	756.0			
Li/ CrO <sub>3</sub>	6Li+CrO <sub>3</sub> =3Li <sub>2</sub> O+Cr	-1171.0	2.023	2296.6	1608.2			
Li/ SOF <sub>2</sub>	4Li+2SOF <sub>2</sub> =4LiF+S+S O <sub>2</sub>	-1646.9	4.270	2288.7	622.8			
Li/ (CF) <sub>n</sub>	nLi+(CF) <sub>n</sub> =nLiF+nC	-299.1	3.100	2189.3	864.6			
Li/ BrF	2Li+BrF=LiF+LiBr	-875.8	4.540	2156.9	542.0			
Li/ CoF <sub>3</sub>	3Li+CoF <sub>3</sub> =3LiF+Co	-1044.1	3.607	2120.8	693.6			
Li/ P <sub>2</sub> S <sub>5</sub>	5Li+1/2P₂S₅=5/2 Li₂S+P	-1207.9	2.131	1957.7	1205.8			
Li/ SeO <sub>2</sub>	4Li+SeO <sub>2</sub> =2Li <sub>2</sub> O+Se	-950.9	2.464	1904.1	966.2			
Li/ P <sub>4</sub> O <sub>10</sub>	5Li+1/4P₄O <sub>10</sub> =5/2 Li₂O+P	-722.2	1.497	1898.3	1888.2			
Li/ As <sub>2</sub> O <sub>5</sub>	5Li+1/2As <sub>2</sub> O <sub>5</sub> =5/2Li <sub>2</sub> O +As	-1011.9	2.097	1878.5	1166.1			
Li/ B <sub>2</sub> S <sub>3</sub>	3Li+1/2B <sub>2</sub> S <sub>3</sub> =3/2Li <sub>2</sub> S+ B	-534.7	1.847	1862.8	1364.8			
Li/ SCl <sub>2</sub>	2Li+SCl <sub>2</sub> =2LiCl+S	-740.3	3.840	1759.8	520.6			

Table S4. Thermodynamic data of all lithium battery systems calculated.

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Li/ BrCl	2Li+BrCl=LiCl+LiBr	-780.7	4.050	1678.0	464.7
Li/ CuF <sub>2</sub>	2Li+CuF <sub>2</sub> =2LiF+Cu	-683.4	3.542	1644.7	527.9
Li/ FeF <sub>3</sub>	3Li+FeF <sub>3</sub> =3LiF+Fe	-791.1	2.733	1644.1	712.6
Li/ Ag <sub>2</sub> O <sub>3</sub>	3Li+1/2Ag <sub>2</sub> O <sub>3</sub> =3/2Li <sub>2</sub> O +Ag	-902.5	3.118	1641.9	609.7
Li/ MnF <sub>3</sub>	3Li+MnF <sub>3</sub> =3LiF+Mn	-763.1	2.636	1596.7	718.3
Li/ SO <sub>2</sub> CIF(g)	2Li+SO <sub>2</sub> CIF=LiCI+LiF+ SO <sub>2</sub>	-759.2	3.930	1592.9	452.3
Li/ MnO <sub>2</sub>	4Li+MnO <sub>2</sub> =2Li <sub>2</sub> O+Mn	-657.3	1.703	1591.8	1233.1
Li/ NbF5	5Li+NbF <sub>5</sub> =5LiF+Nb	-1239.5	2.569	1546.7	713.2
Li/ SiS <sub>2</sub>	4Li+SiS <sub>2</sub> =2Li <sub>2</sub> S+Si	-665.4	1.724	1540.5	1162.5
Li/ V <sub>2</sub> O <sub>5</sub>	5Li+1/2V <sub>2</sub> O <sub>5</sub> =5/2Li <sub>2</sub> O+ V	-693.3	1.437	1532.6	1473.6
Li/ MoO <sub>3</sub>	6Li+MoO <sub>3</sub> =3Li <sub>2</sub> O+Mo	-1015.6	1.754	1520.1	1117.2
Li/ MoCl <sub>6</sub>	6Li+MoCl <sub>6</sub> =6LiCl+Mo	-1915.4	3.309	1518.8	521.0
Li/ SOCl <sub>2</sub>	4Li+2SOCl <sub>2</sub> =4LiCl+S+ SO <sub>2</sub>	-1441.1	3.730	1506.6	450.6
Li/ RuO <sub>2</sub>	4Li+RuO <sub>2</sub> =2Li <sub>2</sub> O+Ru	-842.4	2.183	1454.9	805.6
Li/ TiF <sub>4</sub>	4Li+TiF <sub>4</sub> =4LiF+Ti	-791.8	2.052	1450.6	865.5
Li/ CrO <sub>2</sub>	4Li+CrO <sub>2</sub> =2Li <sub>2</sub> O+Cr	-577.5	1.496	1435.4	1276.3
Li/ NiF <sub>2</sub>	2Li+NiF <sub>2</sub> =2LiF+Ni	-571.3	2.961	1435.2	554.4
Li/ CrF <sub>3</sub>	3Li+CrF <sub>3</sub> =3LiF+Cr	-659.7	2.279	1411.5	737.7
Li/ SO <sub>2</sub> Cl <sub>2</sub>	2Li+SO <sub>2</sub> Cl <sub>2</sub> =2LiCl+SO <sub>2</sub>	-748.9	3.880	1397.6	397.2
Li/ NiS <sub>2</sub>	4Li+NiS <sub>2</sub> =2Li <sub>2</sub> S+Ni	-753.3	1.952	1389.5	872.8
Li/ Co <sub>3</sub> O <sub>4</sub>	8/3Li+1/3Co <sub>3</sub> O <sub>4</sub> =4/3Li <sub>2</sub> O+Co	-490.3	1.906	1378.7	890.4
Li/ FeS <sub>2</sub>	4Li+FeS <sub>2</sub> =2Li <sub>2</sub> S+Fe	-721.9	1.870	1358.0	893.4
Li/ MoCl <sub>5</sub>	5Li+MoCl <sub>5</sub> =5LiCl+Mo	-1499.0	3.107	1352.3	490.5
Li/ CoS <sub>2</sub>	4Li+CoS <sub>2</sub> =2Li <sub>2</sub> S+Co	-732.4	1.898	1348.7	871.1
Li/ FeS <sub>2(pyrite)</sub>	4Li+FeS <sub>2</sub> =2Li <sub>2</sub> S+Fe	-711.1	1.843	1337.0	893.6
Li/ CoF <sub>2</sub>	2Li+CoF <sub>2</sub> =2LiF+Co	-528.2	2.737	1324.1	533.0
Li/ Cr <sub>2</sub> S <sub>3</sub>	3Li+1/2Cr <sub>2</sub> S <sub>3</sub> =3/2 Li <sub>2</sub> S+Cr	-574.6	1.985	1319.9	803.3
Li/ FeF <sub>2</sub>	2Li+FeF₂=2LiF+Fe	-506.8	2.626	1306.8	571.2
Li/ Fe <sub>2</sub> O <sub>3</sub>	3Li+1/2Fe <sub>2</sub> O <sub>3</sub> =3/2 Li <sub>2</sub> O+Fe	-470.7	1.626	1298.8	1007.0
Li/ As <sub>4</sub> O <sub>6</sub>	6Li+1/2	-533.7	1.913	1284.4	812.8

	As <sub>2</sub> O <sub>6</sub> =3Li <sub>2</sub> O+As				
Li/ CuO	2Li+CuO=Li <sub>2</sub> O+Cu	-431.5	2.236	1282.9	673.9
Li/ Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub>	Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub> +4Li=Li <sub>6</sub> C <sub>6</sub> O <sub>6</sub>	-964.9	2.500	1278.4	589.4
Li/ GaF <sub>3</sub>	3Li+GaF₃=3LiF+Ga	-677.8	2.342	1276.1	634.5
Li/ GeO <sub>2</sub>	4Li+GeO <sub>2</sub> =2Li <sub>2</sub> O+Ge	-601.0	1.557	1260.9	1024.5
Li/ Cu <sub>2</sub> O	Li+1/2Cu <sub>2</sub> O=1/2Li <sub>2</sub> O+ Cu	-354.5	3.675	1254.8	374.6
Li/ FeCl <sub>3</sub>	3Li+FeCl <sub>3</sub> =3LiCl+Fe	-819.2	2.830	1243.3	495.7
Li/ MnS <sub>2</sub>	4Li+MnS <sub>2</sub> =2Li <sub>2</sub> S+Mn	-653.0	1.692	1235.4	900.4
Li/ P <sub>4</sub> S <sub>3</sub>	3/2Li+1/4P <sub>4</sub> S <sub>3</sub> =3/4Li <sub>2</sub> S +P	-289.5	2.000	1229.0	730.6
Li/ B <sub>2</sub> O <sub>3</sub>	3Li+1/2B <sub>2</sub> O <sub>3</sub> =3/2Li <sub>2</sub> O+ B	-244.7	0.845	1221.5	2309.7
Li/ GeS <sub>2</sub>	4Li+GeS <sub>2</sub> =2Li <sub>2</sub> S+Ge	-723.4	1.874	1221.3	783.8
Li/ RuCl <sub>3</sub>	3Li+RuCl <sub>3</sub> =3LiCl+Ru	-993.5	3.432	1209.0	387.6
Li/ Fe <sub>3</sub> O <sub>4</sub>	8/3Li+1/3Fe <sub>3</sub> O <sub>4</sub> =4/3Li <sub>2</sub> O+Fe	-409.8	1.593	1189.6	926.1
Li/ MoCl <sub>4</sub>	4Li+MoCl <sub>4</sub> =4LiCl+Mo	-1135.6	2.942	1188.0	450.9
Li/ WCl <sub>6</sub>	6Li+WCl <sub>6</sub> =6LiCl+W	-1850.4	3.196	1173.0	405.5
Li/ SO <sub>2</sub>	2Li+2SO <sub>2</sub> =Li <sub>2</sub> S <sub>2</sub> O <sub>4</sub>	-598.2	3.100	1170.1	418.8
Li/ AgO	2Li+AgO=Li <sub>2</sub> O+Ag	-575.0	2.980	1159.6	432.7
Li/ VF <sub>3</sub>	3Li+VF <sub>3</sub> =3LiF+V	-536.4	1.853	1157.3	744.9
Li/ Ni <sub>3</sub> S <sub>4</sub>	8/3Li+1/3Ni <sub>3</sub> S <sub>4</sub> =4/3Li <sub>2</sub> S +Ni	-488.0	1.897	1130.1	704.5
Li/ NbCl <sub>5</sub>	5Li+NbCl₅=5LiCl+Nb	-1238.8	2.568	1128.7	496.0
Li/ Mn <sub>2</sub> O <sub>3</sub>	3Li+1/2Mn <sub>2</sub> O <sub>3</sub> =3/2Li <sub>2</sub> O +Mn	-401.3	1.386	1117.3	1018.6
Li/ CuCl <sub>2</sub>	2Li+CuCl <sub>2</sub> =2LiCl+Cu	-593.1	3.074	1110.7	398.7
Li/ As <sub>2</sub> S <sub>3</sub>	3Li+1/2As <sub>2</sub> S <sub>3</sub> =3/2 Li <sub>2</sub> S+As	-574.2	1.984	1108.8	653.6
Li/ NiO	2Li+NiO=Li <sub>2</sub> O+Ni	-349.5	1.811	1096.1	717.6
Li/ ZnF <sub>2</sub>	$2Li+ZnF_2=2LiF+Zn$	-462.1	2.395	1094.6	518.5
Li/ CoO	2Li+CoO=Li <sub>2</sub> O+Co	-347.0	1.798	1085.3	715.4
Li/ Ni CO3	2Li+NiCO <sub>3</sub> =Li <sub>2</sub> CO <sub>3</sub> +Ni	-514.2	2.665	1077.4	451.6
Li/ Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	3Li+1/2 Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> =3/2Li <sub>2</sub> SO <sub>4</sub> + Fe	-851.2	2.941	1071.0	402.1
Li/ WCl <sub>5</sub>	5Li+WCl <sub>5</sub> +5LiCl+W	-1520.0	3.151	1066.7	371.1
Li/ CuSO <sub>4</sub>	2Li+CuSO <sub>4</sub> =Li <sub>2</sub> SO <sub>4</sub> +C	-659.5	3.418	1055.9	335.8

	u				
Li/ MoO <sub>2</sub>	4Li+MoO <sub>2</sub> =2Li <sub>2</sub> O+Mo	-589.4	1.527	1051.5	837.9
Li/ BeS	2Li+BeS=Li₂S+Be	-206.0	1.068	1041.2	1304.9
Li/ Co CO <sub>3</sub>	2Li+CoCO <sub>3</sub> =Li <sub>2</sub> CO <sub>3</sub> +C	-495.3	2.567	1035.9	450.7
	0				
Li/ CrCl <sub>3</sub>	3Li+CrCl <sub>3</sub> =3LiCl+Cr	-667.1	2.305	1034.2	507.8
Li/ N <sub>2</sub>	6Li+N <sub>2</sub> =2Li <sub>3</sub> N	-257.2	0.440	1025.6	5740.4
Li/ CuF	Li+CuF=LiF+Cu	-327.7	3.396	1017.2	324.7
Li/ FeO	2Li+FeO=Li <sub>2</sub> O+Fe	-309.8	1.605	1003.8	746.1
Li/ VCI <sub>3</sub>	3Li+VCl <sub>3</sub> =3LiCl+V	-641.9	2.218	1001.0	511.2
Li/ NiCl <sub>2</sub>	2Li+NiCl <sub>2</sub> =2LiCl+Ni	-509.8	2.6	987.0	413.6
Li/ NbCl <sub>4</sub>	4Li+NbCl <sub>4</sub> =4LiCl+Nb	-932.1	2.4	986.4	456.7
Li/ GaCl <sub>3</sub>	3Li+GaCl <sub>3</sub> =3LiCl+Ga	-698.4	2.4	985.2	456.6
	$3Li+1/2Al_2S_3=3/2Li_2S+$	-338.5	1.2	980.4	1070.9
Li/ Al <sub>2</sub> S <sub>3</sub>	AI				
Li/ CuS	2Li+CuS=Li <sub>2</sub> S+Cu	-385.4	2.0	977.7	560.6
	8/3Li+1/3Co <sub>3</sub> S <sub>4</sub> =4/3	-423.0	1.6	977.6	702.8
Li/ Co <sub>3</sub> S <sub>4</sub>	Li <sub>2</sub> S+Co				
Li/ FeCO <sub>3</sub>	2Li+FeCO <sub>3</sub> =Li <sub>2</sub> CO <sub>3</sub> +Fe	-465.4	2.4	973.3	450.7
Li/ MnO <sub>2</sub>	Li+MnO <sub>2</sub> =LiMnO2	-327.8	3.4	969.8	308.1

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Battery Type	ΔG (kJ mol <sup>-1</sup> )	E <sub>aver</sub> (V)	Capacity (+) (mAh g <sup>-1</sup> )	Capacity (-) (mAh g <sup>-1</sup> )	Cal. Energy Density (Wh kg <sup>-1</sup> )	Cal. Energy Density (Wh L <sup>-1</sup> )
V <sub>2</sub> O <sub>5</sub> -Li <sub>4.4</sub> Si	-1128.9	2.660	442.1	2011.5	963.7	2425.7
LiCoO <sub>2</sub> -Si	-1553.4	3.660	273.8	4198.8	940.7	4517.8
Li(CoNiMn) <sub>1/3</sub> O <sub>2</sub> - Si	-1510.9	3.560	277.8	4198.8	927.5	4142.7
MnO <sub>2</sub> -Li <sub>4.4</sub> Si	-1424.7	3.360	308.3	2011.5	897.1	3167.6
0.3Li <sub>2</sub> MnO <sub>3</sub> -0.7Li[ MnNiCo] <sub>0.333</sub> O <sub>2</sub> - Si	-333.75	3.450	261.3	4198.8	851.0	3593.6
LiCoPO₄-Si	-1977.9	4.660	166.6	4198.8	746.7	2752.2
LiCoO <sub>2</sub> -Sn	-1393.0	3.280	273.8	993.40	704.4	3878.5
Li(CoNiMn) <sub>1/3</sub> O <sub>2</sub> - Sn	-1350.5	3.180	277.8	993.40	690.7	3553.7
LiMnPO <sub>4</sub> -Si	-1723.2	4.060	170.9	4198.8	666.5	2306.8
LiMnO <sub>2</sub> -Sn	-1264.3	2.980	285.5	993.4	660.4	3077.4

Table S5. Thermodynamic data of lithium-ion battery systems with high gravimetric energy densities.

Formula	Theoretical Capacity (mAh g <sup>-1</sup> )	Practical Capacity (mAh g <sup>-1</sup> )	Charge Transfer (max)	Voltage (V) (vs.Li+/Li)
CrO <sub>3</sub>	1608	-	6.0	2.023
$P_2S_5$	1206	-	10.0	2.131
SeO <sub>2</sub>	966.2	-	4.0	2.464
TiF <sub>4</sub>	865.5	-	4.0	2.052
(CF) <sub>n</sub>	864.3	220.0	1.0	3.100
Li <sub>1.2</sub> V <sub>3</sub> O <sub>8</sub>	816.0	250.0	8.8	2.500 <sup>*</sup>
RuO <sub>2</sub>	805.6	-	4.0	2.183
CrF <sub>3</sub>	737.7	-	3.0	2.279
$MnF_3$	718.3	-	3.0	2.636
FeF <sub>3</sub>	712.6	-	3.0	2.733
CoF <sub>3</sub>	693.6	-	3.0	3.607
RuF₅	683.5	-	5.0	4.472
CuO	673.9	187.0	2.0	2.236
Cr <sub>8</sub> O <sub>21</sub>	642.0	265.0	18.0	3.000*
$Li_2C_6O_6$	589.0	300.0	4.0	2.500 <sup>*</sup>
NiF <sub>2</sub>	554.4	-	2.0	2.961
CuF <sub>2</sub>	527.9	-	2.0	3.542
VCI <sub>3</sub>	511.2	-	3.0	2.218
Cu <sub>2.33</sub> V <sub>4</sub> O <sub>11</sub>	475.0	270.0	9.3	2.700 <sup>*</sup>
NiCO <sub>3</sub>	451.6	-	2.0	2.665
V <sub>2</sub> O <sub>5</sub>	442.1	400.0	3.0	2.700
MnO <sub>2</sub>	308.3	150.0	1.0	3.400

Table S6. Capacity values of typical lithium-free cathodes.

Formula	Theoretical Capacity (mAh g <sup>-1</sup> )	Practical Capacity (mAh g <sup>-1</sup> )	Charge Transfer (max)	Voltage (V) (vs.Li+/Li)
0.3Li <sub>2</sub> MnO <sub>3</sub> -0.7Li[Mn <sub>0.333</sub> Ni <sub>0.333</sub> Co <sub>0.333</sub> ]O <sub>2</sub>	261.3	250.0	1.0	3.500*
LiMnO <sub>2</sub>	285.5	190.0	1.0	3.400*
LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>	277.8	118.0	1.0	3.60
LiNiO <sub>2</sub>	274.5	170.0	1.0	3.50
LiCoO <sub>2</sub>	273.8	140.0	1.0	3.70
LiMnPO <sub>4</sub>	170.9	168.0	1.0	4.100
LiFePO <sub>4</sub>	169.9	160.0	1.0	3.450
LiCoPO <sub>4</sub>	166.7	100.0	1.0	4.700
LiMn <sub>2</sub> O <sub>4</sub>	148.2	110.0	1.0	3.990
LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub>	146.7	110.0	1.0	4.660
LiCo <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub>	146.6	130.0	1.0	5.100
LiCoMnO <sub>4</sub>	145.0	130.0	1.0	5.100
Li <sub>3</sub> CrO <sub>3</sub>	665.5	-	3.0	-
$Li_3CrO_2F_2$	563.0	-	3.0	-
Li <sub>2</sub> NiO <sub>2</sub>	512.6	200.0	2.0	-
Li <sub>3</sub> CrF <sub>6</sub>	430.4	-	3.0	-
Li <sub>2</sub> MnBO <sub>3</sub> F	365.6	-	2.0	-
Li <sub>2</sub> NiF <sub>4</sub>	360.8	-	2.0	-
Li <sub>2</sub> MnSiO <sub>4</sub>	333.1	150.0	2.0	3.000*
Li <sub>3</sub> Cr(PO <sub>4</sub> ) <sub>2</sub>	305.0	-	3.0	-
Li <sub>2</sub> CrPO <sub>4</sub> O	302.0	-	2.0	-
Li <sub>2</sub> CrMnO <sub>4</sub>	290.0	-	2.0	-
LiNi <sub>0.5</sub> Ti <sub>1.5</sub> O <sub>4</sub>	155.7	-	1.0	-
LiCrMnO <sub>4</sub>	150.7	75.0	1.0	4.800
$LiCr_{0.5}Mn_{1.5}O_4$	149.4	125.0	1.0	4.800
LiNiVO <sub>4</sub>	148.4	40.0	1.0	4.800
$LiFe_{0.5}Mn_{1.5}O_4$	147.9	125.0	1.0	4.900
Li <sub>1.01</sub> Cu <sub>0.32</sub> Mn <sub>1.67</sub> O <sub>4</sub>	147.8	71.0	1.0	4.900

Table S7. Capacity values of lithium-contained cathodes.

Formula	Capacity (mA h g <sup>-1</sup> )	Charge Transfer (max)	Voltage (V) (vs.Li+/Li)
Si	4200	4.4	0.04000
BN	3240	3.0	-0.3448
AIN	1962	3.0	-0.5472
P <sub>4</sub> O <sub>10</sub>	1888	20.0	1.497
SiO <sub>2</sub>	1784	4.0	0.6895
Al <sub>2</sub> O <sub>3</sub>	1577	6.0	0.1750
V <sub>2</sub> O <sub>5</sub>	1474	10.0	1.437
$B_2S_3$	1365	6.0	1.847
TiO <sub>2</sub> -R	1342	1.0	0.6046
TiN	1299	3.0	-0.3980
VO <sub>2</sub>	1293	4.0	-0.5080
CrO <sub>2</sub>	1276	4.0	1.496
MnO <sub>2</sub>	1233	4.0	1.703
Sn	993.4	4.4	0.5040
TiO	839.3	2.0	0.3431
VO	800.8	2.0	0.8136
Al <sub>2</sub> ( SO <sub>4</sub> ) <sub>3</sub>	470.0	6.0	0.7913
LiTiS <sub>2</sub>	225.4	1.0	0.5000*
LiVS <sub>2</sub>	219.7	1.0	1.000*
Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	175.1	3.0	1.550*

Table S8. Capacity values of typical anodes.

Table S9. Capacity values calculated in all.

Capacity						
Chemical formula	Capacity (m A h g <sup>-1</sup> )	Chemical formula	Capacity (m A h g <sup>-1</sup> )	Chemical formula	Capacity (m A h g <sup>-1</sup> )	
LiF	1033.23	Li <sub>2</sub> S	1166.6	LiBO <sub>2</sub>	538.7	
BeF <sub>2</sub>	1140.27	BeS	1304.90	NaBO <sub>2</sub>	407.31	
NaF	638.31	$B_2S_3$	1364.84	Mg(BO <sub>2</sub> ) <sub>2</sub>	487.62	
$MgF_2$	860.37	Na <sub>2</sub> S	686.81	AI (BO <sub>2</sub> ) <sub>3</sub>	517.35	
AIF <sub>3</sub>	957.46	MgS	950.89	KBO <sub>2</sub>	327.21	
KF	461.32	$AI_2S_3$	1070.91	Sc(BO <sub>2</sub> ) <sub>3</sub>	463.72	
CaF <sub>2</sub>	686.56	SiS <sub>2</sub>	1162.53	Ti (BO <sub>2</sub> ) <sub>4</sub>	489.28	
ScF <sub>3</sub>	788.65	$P_4S_3$	730.64	V(BO <sub>2</sub> ) <sub>5</sub>	505.70	
TiF₃	766.76	$P_4S_7$	1077.11	Cr(BO <sub>2</sub> ) <sub>6</sub>	520.65	
TiF <sub>4</sub>	865.53	$P_2S_5$	1205.76	Mn (BO <sub>2</sub> ) <sub>3</sub>	438.48	
VF <sub>3</sub>	744.92	$K_2S$	486.14	Fe (BO <sub>2</sub> ) <sub>2</sub>	378.91	
CrF₃	737.71	CaS	743.00	Co (BO <sub>2</sub> ) <sub>3</sub>	429.13	
$MnF_2$	576.78	$Sc_2S_3$	864.05	Ni(BO <sub>2</sub> ) <sub>3</sub>	429.68	
$MnF_3$	718.32	TiS <sub>2</sub>	957.20	Cu(BO <sub>2</sub> ) <sub>2</sub>	359.35	
FeF <sub>2</sub>	571.20	TiS	670.60	Zn(BO <sub>2</sub> ) <sub>2</sub>	354.96	
FeF <sub>3</sub>	712.55	$Ti_2S_3$	837.84	Ga(BO <sub>2</sub> ) <sub>3</sub>	405.76	
CoF <sub>2</sub>	553.00	VS	645.76	Ge(BO <sub>2</sub> ) <sub>4</sub>	439.58	
CoF <sub>3</sub>	693.57	$V_2S_3$	811.83	Y (BO <sub>2</sub> ) <sub>3</sub>	369.95	
$NiF_2$	554.38	VS <sub>2</sub>	931.63	Zr (BO <sub>2</sub> ) <sub>4</sub>	408.45	
CuF	324.69	$V_2S_5$	1022.12	Nb BO <sub>2</sub>	197.48	
CuF <sub>2</sub>	527.88	CrS	637.66	MoBO <sub>2</sub>	193.16	
$ZnF_2$	518.47	$Cr_2S_3$	803.28	Ru(BO <sub>2</sub> ) <sub>2</sub>	287.12	
GaF₃	634.51	MnS	616.10	Ag BO <sub>2</sub>	177.87	
RbF	256.56	MnS <sub>2</sub>	900.36	La (BO <sub>2</sub> ) <sub>3</sub>	300.76	
YF <sub>3</sub>	551.09	FeS	609.74	Ce(BO <sub>2</sub> ) <sub>4</sub>	344.31	
$ZrF_2$	414.82	FeS <sub>2(marcasite)</sub>	893.55	W (BO <sub>2</sub> ) <sub>6</sub>	364.89	

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ZrF <sub>3</sub>	542.47	FeS <sub>2(pyrite)</sub>	893.55	Be <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	1111.65
ZrF <sub>4</sub>	641.11	CoS	589.05	Ca <sub>3</sub> (BO <sub>3</sub> ) <sub>2</sub>	676.08
$NbF_5$	713.19	$Co_2S_3$	751.21	Li <sub>2</sub> CO <sub>3</sub>	725.43
RuF₃	508.68	CoS <sub>2</sub>	871.13	Be CO <sub>3</sub>	776.61
RuF₅	683.49	$Co_3S_4$	702.84	$B_2(CO_3)_3$	797.46
AgF	211.26	NiS	590.60	Na <sub>2</sub> CO <sub>3</sub>	505.74
SnF <sub>2</sub>	342.06	$Ni_3S_4$	704.50	Mg CO <sub>3</sub>	653.75
LaF <sub>3</sub>	410.43	$Ni_3S_2$	446.30	Al <sub>2</sub> ( CO <sub>3</sub> ) <sub>3</sub>	687.24
CeF <sub>3</sub>	407.91	NiS <sub>2</sub>	872.83	K <sub>2</sub> CO <sub>3</sub>	387.85
Li <sub>2</sub> O	1793.85	CuS	560.63	CaCO <sub>3</sub>	535.56
BeO	2143.12	Cu <sub>2</sub> S	336.79	Sc 2(CO3)3	595.72
B <sub>2</sub> O <sub>3</sub>	2309.73	ZnS	550.02	Ti (CO <sub>3</sub> ) <sub>2</sub>	638.57
Na <sub>2</sub> O	864.85	$Ga_2S_3$	682.42	V <sub>2</sub> (CO <sub>3</sub> ) <sub>5</sub>	666.82
MgO	1329.95	GaS	526.61	Cr (CO <sub>3</sub> ) <sub>3</sub>	693.07
$AI_2O_3$	1577.15	GeS	511.94	MnCO <sub>3</sub>	466.33
SiO <sub>2</sub>	4704.05			E (00.)	
quartz SiOa	1/84.25	GeS <sub>2</sub>	783.83	$Fe_2(CO_3)_3$	551.25
high					
cristobalite	1784.25	$As_2S_3$	653.58	FeCO <sub>3</sub>	450.66
P <sub>4</sub> O <sub>10</sub>	1888.16	$Se_2S_6$	918.08	Co CO <sub>3</sub>	450.66
K <sub>2</sub> O	569.06	Se <sub>4</sub> S <sub>4</sub>	482.79	Ni CO <sub>3</sub>	451.57
CaO	955.87	$RuS_2$	648.94	CuCO <sub>3</sub>	433.84
Sc <sub>2</sub> O <sub>3</sub>	1166.04	$Y_2S_3$	586.87	Zn CO₃	427.46
TiO	839.30	$ZrS_2$	690.06	Ga <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	503.36
Ti <sub>2</sub> O <sub>3</sub>	1118.81	NbS <sub>2</sub>	690.06	Ge(CO <sub>3</sub> ) <sub>2</sub>	556.46
Ti <sub>3</sub> O <sub>5</sub>	1198.64	$MoS_2$	669.73	Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	449.39
TiO <sub>2</sub> -R	1342.32	$Mo_2S_3$	558.21	ZrCO <sub>3</sub>	354.44
TiO2-B	1342.32	$Ag_2S_{(argentite)}$	216.31	Nb <sub>2</sub> CO <sub>3</sub>	218.06
TiO <sub>2</sub> -A	1342.32	LaS	313.52	Mo <sub>2</sub> CO <sub>3</sub>	212.80
VO	800.75	$La_2S_3$	429.96	Ru <sub>2</sub> CO <sub>3</sub>	204.47
V <sub>2</sub> O <sub>3</sub>	1072.91	CeS	311.31	Ag <sub>2</sub> CO <sub>3</sub>	194.39
VO <sub>2</sub>	1292.56	WS <sub>2</sub>	432.33	La 2(CO3)3	351.23
$V_3O_5$	1151.16	$Na_2S_2$	486.80	$Ce_2(CO_3)_3$	349.39
V <sub>2</sub> O <sub>5</sub>	1473.58	Li₃N	2308.49	W (CO <sub>3</sub> ) <sub>3</sub>	441.94

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CrO <sub>2</sub>	1276.33	$Be_3N_2$	2921.14	Li <sub>3</sub> PO <sub>4</sub>	694.37
Cr <sub>2</sub> O <sub>3</sub>	1058.01	BN	3239.66	Be <sub>3(</sub> PO <sub>4)2</sub>	741.12
Cr <sub>3</sub> O <sub>4</sub>	947.66	(CN)	3090.39	Na <sub>3</sub> PO <sub>4</sub>	490.45
CrO <sub>3</sub>	1608.17	Na₃N	969.00	$Mg_3(PO_4)_2$	611.77
MnO <sub>2</sub>	1233.14	Mg₃N2	1593.29	AI PO <sub>4</sub>	659.31
$Mn_2O_3$	1018.58	AIN	1961.64	K <sub>3</sub> PO <sub>4</sub>	378.79
MnO	755.63	Si <sub>3</sub> N <sub>4</sub>	2292.62	Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	518.44
Mn <sub>3</sub> O <sub>4</sub>	937.06	K₃N	612.36	Sc PO <sub>4</sub>	574.61
FeO	746.10	$Ca_3N_2$	1084.73	Ti <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	614.37
Fe <sub>2</sub> O <sub>3</sub>	1007.01	ScN	1363.65	V <sub>3</sub> (PO <sub>4</sub> ) <sub>5</sub>	640.49
Fe <sub>3</sub> O <sub>4</sub>	926.05	TiN	1299.49	CrPO <sub>4</sub>	547.09
CoO	715.35	VN	1237.97	Mn PO <sub>4</sub>	536.35
Co <sub>3</sub> O <sub>4</sub>	890.42	CrN	1218.19	Mn3( PO <sub>4 )2</sub>	453.29
NiO	717.64	Cr <sub>2</sub> N	681.40	Fe PO <sub>4</sub>	533.13
Ni <sub>2</sub> O <sub>3</sub>	972.33	Mn₄N	343.96	Co3(PO <sub>4)2</sub>	438.48
CuO	673.86	$Mn_5N_2$	531.24	Ni <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	439.34
Cu <sub>2</sub> O	374.61	Fe <sub>2</sub> N	639.67	Cu <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	422.53
ZnO	658.60	Fe₄N	338.71	Zn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	416.48
Ga <sub>2</sub> O <sub>3</sub>	857.90	Co <sub>3</sub> N	421.39	Ga PO <sub>4</sub>	488.20
GeO	604.73	NiN	1105.97	Ge <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	538.00
GeO <sub>2</sub>	1024.53	Cu₃N	392.90	Y PO <sub>4</sub>	437.27
As <sub>4</sub> O <sub>6</sub>	812.81	$Zn_3N_2$	717.31	Zr <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	346.27
$As_2O_5$	1166.09	GaN	960.28	Nb <sub>3</sub> PO <sub>4</sub>	215.16
SeO <sub>2</sub>	966.17	Ge <sub>3</sub> N <sub>4</sub>	1174.01	Mo <sub>3</sub> PO <sub>4</sub>	210.05
$Y_2O_3$	712.14	$As_3N_5$	1363.71	Ru <sub>3</sub> PO <sub>4</sub>	201.93
ZrO <sub>2</sub>	870.01	SeN <sub>2</sub>	1503.26	Ag <sub>3</sub> PO <sub>4</sub>	192.09
NbO	492.19	YN	781.29	La PO <sub>4</sub>	343.79
NbO <sub>2</sub>	858.30	ZrN	764.08	CePO <sub>4</sub>	342.02
$Nb_2O_5$	1008.29	NbN	752.05	W (PO <sub>4</sub> ) <sub>2</sub>	430.22
MoO <sub>2</sub>	837.94	MoN	731.30	Li <sub>2</sub> SO <sub>4</sub>	487.54
MoO <sub>3</sub>	1117.20	RuN	698.70	Be SO <sub>4</sub>	510.13
RuO <sub>2</sub>	805.64	Ag <sub>3</sub> N	238.16	Na <sub>2</sub> SO <sub>4</sub>	377.37
RuO <sub>4</sub>	1298.93	LaN	525.82	Mg SO <sub>4</sub>	445.32
AgO	432.74	CeN	521.69	Al <sub>2</sub> ( SO <sub>4</sub> ) <sub>3</sub>	469.99

	004.04		750.05		445.40
Ag <sub>2</sub> O	231.31	WN <sub>2</sub>	759.05	$AIK(SO_4)_2$	415.19
Ag <sub>2</sub> O <sub>3</sub>	609.74	NaN <sub>3</sub>	179.81	K <sub>2</sub> SO <sub>4</sub>	307.60
$La_2O_3$	493.57	KN <sub>3</sub>	162.28	CaSO <sub>4</sub>	393.73
CeO <sub>2</sub>	622.87	LiCl	632.20	Sc <sub>2</sub> (SO <sub>4)3</sub>	425.30
$Ce_2O_3$	489.93	BeCl <sub>2</sub>	670.72	Ti (SO <sub>4</sub> ) <sub>2</sub>	446.70
WO <sub>2</sub>	496.69	PCI <sub>5</sub>	643.53	V <sub>2</sub> (SO <sub>4</sub> ) <sub>5</sub>	460.35
WO <sub>3</sub>	693.62	NaCl	458.59	Cr 2(SO <sub>4</sub> ) <sub>3</sub>	410.03
Na <sub>2</sub> O <sub>2</sub>	687.41	MgCl <sub>2</sub>	562.99	MnSO <sub>4</sub>	354.98
$K_2O_2$	486.43	AICI <sub>3</sub>	603.00	FeSO <sub>4</sub>	352.86
$Ag_2O_2$	216.37	KCI	359.50	$Fe_2(SO_4)_3$	402.14
$Li_2O_2$	1168.31	CaCl <sub>2</sub>	482.98	Co SO <sub>4</sub>	345.83
NaO <sub>2</sub>	487.40	ScCl <sub>3</sub>	531.37	Ni SO4	346.37
KO <sub>2</sub>	376.97	TiCl <sub>2</sub>	451.30	CuSO <sub>4</sub>	335.84
CaO <sub>2</sub>	371.84	TiCl₃	521.34	Zn SO <sub>4</sub>	332.00
Li₃P	1552.30	VCl <sub>2</sub>	439.92	Y <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	345.08
$Be_3P_2$	1807.16	VCl <sub>3</sub>	511.15	Zr(SO <sub>4)2</sub>	378.35
BP	1924.20	CrCl <sub>2</sub>	436.14	$Nb_2SO_4$	190.16
Na₃P	804.50	CrCl <sub>3</sub>	507.75	Mo <sub>2</sub> SO <sub>4</sub>	186.16
$Mg_3P_2$	1192.38	MnCl <sub>2</sub>	425.95	$Ru_2SO_4$	179.75
AIP	1387.35	FeCl <sub>2</sub>	422.90	Ag <sub>2</sub> SO <sub>4</sub>	171.91
Si <sub>3</sub> P <sub>4</sub>	1545.11	FeCl₃	495.70	La <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	284.11
K₃P	542.29	CoCl <sub>2</sub>	412.84	$Ce_2(SO_4)_3$	282.90
$Ca_3P_2$	882.68	NiCl <sub>2</sub>	413.60	W (SO <sub>4</sub> ) <sub>3</sub>	340.67
ScP	1058.93	CuCl	270.72	F <sub>2</sub>	1410.72
TiP	1019.83	CuCl <sub>2</sub>	398.68	O <sub>2</sub>	3350.30
(VP)	981.55	ZnCl <sub>2</sub>	393.68	N <sub>2</sub>	5740.41
CrP	969.08	GaCl₃	456.63	S	1671.64
MnP	935.89	AsCl <sub>5</sub>	531.38	В	7436.57
$Mn_2P$	570.85	SeCl <sub>4</sub>	485.59	Si	3817.11
FeP	926.12	YCl <sub>3</sub>	411.77	Р	2595.88
Fe <sub>2</sub> P	563.59	ZrCl <sub>2</sub>	330.62	Cl <sub>2</sub>	755.97
Fe₃P	405.04	ZrCl <sub>3</sub>	406.94	(g)SF <sub>4</sub>	992.10
Co <sub>2</sub> P	540.20	ZrCl <sub>4</sub>	460.04	(g)SF <sub>6</sub>	1101.00
Ni <sub>2</sub> P	541.95	NbCl <sub>5</sub>	496.01	$S_2F_{10}$	1054.69

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	i -	I.	1	I.	i i
Ni₃P	388.32	NbCl <sub>4</sub>	456.74	SCl <sub>2</sub>	520.56
$Ni_5P_2$	452.45	MoCl <sub>4</sub>	450.91	S <sub>2</sub> Cl <sub>2</sub>	396.95
$CuP_2$	1281.41	MoCl <sub>5</sub>	490.50	(g)CIF	984.41
Cu3P	362.82	MoCl <sub>6</sub>	520.99	BrF	541.98
$Zn_3P_2$	623.00	RuCl₃	387.62	BrCl	464.67
GaP	798.48	AgCl	187.00	CIF <sub>3</sub>	1159.63
GeP	776.00	SnCl <sub>2</sub>	282.69	BrF₃	783.10
$Ge_3P_4$	940.91	LaCl₃	327.83	(g)CIF <sub>5</sub>	1232.77
$As_3P_5$	1058.97	CeCl <sub>3</sub>	326.22	BrF₅	919.45
Se(P) <sub>2</sub>	1141.23	WCl <sub>2</sub>	210.42	$IF_5$	724.70
YP	670.71	WCI <sub>4</sub>	329.20	SOF <sub>2</sub>	622.84
$ZrP_2$	1049.86	WCI <sub>5</sub>	371.10	(g)SOCl <sub>2</sub>	450.55
NbP	649.05	WCl <sub>6</sub>	405.51	SOBr <sub>2</sub>	257.86
		0.3Li <sub>2</sub> MnO <sub>3</sub> -0.			
MoP	633 53	7LILIVININICOJ <sub>1/3</sub>	262.00	SO2F2	645 33
	000.00	MnO <sub>2(intercalat</sub>	202.00		010.00
RuP	608.92	ion)	308.30	(g)SO <sub>2</sub> CIF	452.28
	0 4 0 0 0	$V_2O_{5(intercalati}$			
AgP2	946.96	on)	442.10	SO <sub>2</sub> Cl <sub>2</sub>	397.15
AgP3	1201.32	Li <sub>2</sub> C <sub>6</sub> O <sub>6</sub>	589.40	LiCoO <sub>2</sub>	273.80
LaP	473.30	Li <sub>6</sub> C <sub>6</sub> O <sub>6</sub>	294.70	LiNiO <sub>2</sub>	274.50
$Ce_3P_4$	590.94	$Li_2C_6H_4O_4$	209.00	LiMnO <sub>2</sub>	285.50
WP <sub>2</sub>	654.26	$Li_2C_8H_4O_4$	346.50	LiMn <sub>2</sub> O <sub>4</sub>	148.20
-	-	C <sub>6</sub>	371.90	LiFePO <sub>4</sub>	169.90
-	-	Si	4198.80	LiCoPO <sub>4</sub>	166.60
-	-	Sn	993.40	LiMnPO₄	170.90
				$LiNi_{0.5}Mn_{1.5}$	
-	-	Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	175.10	O <sub>4</sub>	146.70
-	-			LICO <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>	277.80

bLi +1/a M <sub>a</sub> X <sub>b</sub> =b/aLi <sub>a</sub> X+M							
M <sub>a</sub> X <sub>b</sub>	ΔG (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (+) (mAh g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )			
TiF <sub>3</sub>	-401.1	1.386	766.8	886.5			
TiF <sub>4</sub>	-791.8	2.052	865.5	1451			
VF <sub>3</sub>	-536.4	1.853	744.9	1157			
CrF <sub>3</sub>	-659.7	2.279	737.7	1412			
MnF <sub>2</sub>	-368.3	1.909	576.8	957.8			
MnF₃	-763.1	2.636	718.3	1597			
FeF <sub>2</sub>	-506.8	2.626	571.2	1307			
FeF <sub>3</sub>	-791.0	2.733	712.6	1644			
CoF <sub>2</sub>	-528.2	2.737	553.0	1324			
CoF <sub>3</sub>	-1044	3.607	693.6	2121			
NiF <sub>2</sub>	-571.3	2.961	554.4	1435			
CuF	-327.7	3.396	324.6	1017			
CuF <sub>2</sub>	-683.4	3.542	527.9	1645			
TiO	-66.20	0.3430	839.3	236.5			
Ti <sub>2</sub> O <sub>3</sub>	-124.0	0.4310	1119	373.7			
Ti <sub>3</sub> O <sub>5</sub>	-162.9	0.5060	1199	463.2			
TiO <sub>2</sub> -R	-233.3	0.6050	1342	602.2			
TiO <sub>2</sub>	-233.6	0.6050	1342	602.9			
TiO <sub>2</sub> -A	-239.5	0.6210	1342	618.1			
VO	-157.0	0.8140	800.8	539.8			
$V_2O_3$	-272.2	0.9400	1073	789.4			
$V_3O_5$	-334.3	1.040	1151	921.8			
VO <sub>2</sub>	196.1	-0.5080	1293	-			
V <sub>2</sub> O <sub>5</sub>	-693.3	1.437	1474	1533			
CrO <sub>2</sub>	-577.5	1.496	1276	1435			
Cr <sub>2</sub> O <sub>3</sub>	-312.8	1.081	1058	897.3			

Table S10. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and lithium.

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Cr <sub>3</sub> O <sub>4</sub>	-237.9	0.9250	974.7	719.7
CrO <sub>3</sub>	-1171	2.023	1608	2297
MnO <sub>2</sub>	-657.3	1.703	1233	1592
Mn <sub>2</sub> O <sub>3</sub>	-401.3	1.386	1019	1117
MnO	-198.3	1.028	755.6	649.4
Mn <sub>3</sub> O <sub>4</sub>	-320.5	1.246	937.1	939.4
FeO	-309.8	1.605	746.1	1004
Fe <sub>2</sub> O <sub>3</sub>	-470.7	1.626	1007	1299
Fe <sub>3</sub> O <sub>4</sub>	-409.8	1.593	926.1	1190
CoO	-347.0	1.798	715.4	1086
Co <sub>3</sub> O <sub>4</sub>	-490.3	1.906	890.4	1379
NiO	-349.5	1.811	717.6	1096
CuO	-431.5	2.236	673.9	1283
Cu <sub>2</sub> O	-207.6	2.152	374.6	734.7
TiS <sub>2</sub>	-475.8	1.233	957.2	945.7
TiS	-168.9	0.8750	670.6	500.2
CrS	-264.6	1.371	637.7	750.3
$Cr_2S_3$	-574.6	1.985	803.3	1320
MnS	-220.6	1.143	616.1	607.4
MnS <sub>2</sub>	-653.0	1.692	900.4	1235
FeS	-338.6	1.755	609.7	924.0
FeS <sub>2(marcasite)</sub>	-721.9	1.871	893.6	1357
FeS <sub>2(pyrite)</sub>	-711.1	1.843	893.6	1337
CoS <sub>2</sub>	-732.4	1.898	871.1	1349
$Co_3S_4$	-423.0	1.644	702.5	977.6
NiS	-359.5	1.863	590.6	954.3
Ni <sub>3</sub> S <sub>4</sub>	-488.0	1.897	704.5	1130
$Ni_3S_2$	-222.7	1.731	446.3	692.4
NiS <sub>2</sub>	-753.3	1.952	872.8	1390
CuS	-385.4	1.997	560.6	977.7
Cu <sub>2</sub> S	-176.4	1.828	336.8	566.3
TiN	115.2	-0.3980	1299	-
VN	62.48	-0.2160	1238	-
CrN	-35.60	0.1230	1218	113.9
Cr <sub>2</sub> N	-13.20	0.09100	681.4	52.84
Mn₄N	-5.900	0.08200	344.0	25.75
Mn <sub>5</sub> N <sub>2</sub>	-20.44	0.1770	531.2	82.44

Fe₄N	-31.22	0.4310	338.7	134.3
Co <sub>3</sub> N	-31.52	0.3270	421.4	124.2

Table S11. Calculated EMF (V) of conversion reactions between fluorides and Li, Na, Mg or Al.

Li				Na					
bLi +1/a M <sub>a</sub> X <sub>b</sub> =b/aLi <sub>a</sub> X+M				bNa+1/a M <sub>a</sub> X <sub>b</sub> =b/aNa <sub>a</sub> X+M					
MX	∆G (kJ mol <sup>-1</sup> )	EMF (V)	Capacity (+) (mAh g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )	nergy ensity Wh kg <sup>-1</sup> )		EMF (V)	Capacity (+) (mAh g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )
TiF₃	-401.1	1.386	766.8	886.5	TiF₃	-276.9	0.9570	766.8	442.5
TiF₄	-791.8	2.052	865.5	1451	TiF₄	-626.2	1.623	865.5	806.0
VF <sub>3</sub>	-536.4	1.853	744.9	1157	$VF_3$	-412.2	1.424	744.9	647.3
CrF <sub>3</sub>	-659.7	2.279	737.7	1412	CrF₃	-535.5	1.850	737.7	835.8
$MnF_2$	-368.3	1.909	576.8	957.8	MnF <sub>2</sub>	-285.5	1.480	576.8	570.9
$MnF_3$	-763.1	2.636	718.3	1597	MnF₃	-638.9	2.207	718.3	981.0
FeF <sub>2</sub>	-506.8	2.626	571.2	1307	FeF <sub>2</sub>	-424.0	2.197	571.2	842.3
FeF <sub>3</sub>	-791.0	2.733	712.6	1644	FeF₃	-666.9	2.304	712.6	1019
CoF <sub>2</sub>	-528.2	2.737	553.0	1324	CoF <sub>2</sub>	-445.4	2.308	553.0	865.7
$CoF_3$	-1044	3.607	693.6	2121	CoF₃	-919.9	3.178	693.6	1382
$NiF_2$	-571.3	2.961	554.4	1435	$NiF_2$	-488.5	2.532	554.4	951.1
CuF	-327.7	3.396	324.6	1017	CuF	-286.3	2.967	324.7	753.6
CuF <sub>2</sub>	-683.4	3.542	527.9	1645	CuF <sub>2</sub>	-600.6	3.112	527.9	1131
Mg				AI					
b/2 Mg+1/a M <sub>a</sub> X <sub>b</sub> =b/2a Mg <sub>a</sub> X <sub>2</sub> +M					b/3 Al+1	/a M <sub>a</sub> X <sub>b</sub> =ł	o/3a Al <sub>a</sub> X₃+I	M	
MX	∆G (kJ mol⁻¹)	EMF (V)	Capacity (+) (mAh g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )	MX	∆G (kJ mol⁻¹)	EMF (V)	Capacity (+) (mAh g <sup>-1</sup> )	Energy Density (Wh kg <sup>-1</sup> )
TiF₃	-244.7	0.8450	766.8	480.9	TiF₃	-69.10	0.2390	766.8	145.6
TiF₄	-583.2	1.511	865.5	939.3	TiF₄	-349.1	0.9050	865.5	606.8
$VF_3$	-380.0	1.313	744.9	731.0	VF <sub>3</sub>	-204.4	0.7060	744.9	420.9
$CrF_3$	-503.2	1.739	737.7	961.0	CrF₃	-327.7	1.132	737.7	669.4
$MnF_2$	-264.0	1.368	576.8	625.5	$MnF_2$	-147.0	0.7620	576.8	368.0

-606.7 862.0 MnF<sub>3</sub> 2.096 718.3 1136 MnF₃ -431.1 1.489 718.3 FeF<sub>2</sub> -402.5 2.086 571.2 946.3  $FeF_2$ -285.5 1.479 571.2 709.1 -634.7 2.193 712.6 FeF<sub>3</sub> -459.1 1.586 712.6 FeF<sub>3</sub> 1181 912.1  $CoF_2$ -423.9 2.197 553.0 971.3  $\mathsf{CoF}_2$ -306.9 1.590 553.0 741.8  $CoF_3$ -887.7 3.067 693.6 1618  $\mathsf{CoF}_3$ -712.1 2.460 693.6 1384  $NiF_2$ -467.0 2.420 554.4 1072  $NiF_2$ -350.0 1.184 554.4 847.7 CuF -275.6 2.856 324.7 808.3 CuF -217.0 2.249 324.7 658.6 -579.1 3.001 527.9  $CuF_2$ -462.1 2.395 527.9 1074  $CuF_2$ 1278

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Elements Abundance& Toxicity							
No.	Atomic	Name	Sym.	Earth	Toxicity		
	Weight			crust(%)			
14	28.0855	Silicon	Si	27.69	Relatively nontoxic		
13	26.9815	Aluminum	AI	8.07	Neurotoxicity (high doses); allergy		
26	55.845	Iron	Fe	5.05	Free iron in the cell is detrimental		
					to DNA, proteins, lipids, and other		
					cellular components. 20 milligrams		
					of iron for every kilogram of mass		
					is toxic, and 60 milligrams per		
					kilogram is considered a lethal		
					dose.		
11	22.9897	Sodium	Na	2.75	Relatively nontoxic		
19	39.0983	Potassium	К	2.58	Relatively nontoxic		
12	24.305	Manganese	Mg	2.08	Relatively nontoxic		
22	47.867	Titanium	Ti	0.62	Non-toxic		
15	30.9738	Phosphorus	Р	0.13	White phosphorus is toxic (causing		
					severe liver damage on ingestion		
6	12.0107	Carbon	С	0.094	Relatively nontoxic		
25	54.938	Manganese	Mn	0.09	Neurotoxicity		
24	51.9961	Chromium	Cr	0.035	Acute toxicity; Genotoxicity;		
					Sensitization & irritation;		
					Carcinogenicity		
28	58.6934	Nickel	Ni	0.019	Acute toxicity; Genotoxicity;		
					Carcinogenicity; Reproductive		
					toxicity		
23	50.9415	Vanadium	V	0.016	Vanadium compounds are toxic		
29	63.546	Copper	Cu	0.006	Free copper is toxic, similar to iron		
27	58.9332	Cobalt	Со	0.002	Lung toxicity; Genotoxicity;		
					Carcinogenicity		
3	6.941	Lithium	Li	0.002	Neurotoxicity (toxic amounts on		
					blood levels >1.5 mmol/l)		
50	118.71	Tin	Sn	0.00022	Acute toxicity; Neurotoxicity		
51	121.76	Antimony	Sb	0.00002	Antimony and many of its		
					compounds are toxic, similar to		
					arsenic		

## Table S12. Key elements in the battery assembling and the toxicity.





Figure S2. Capacity values of lithium-contained cathodes.







Figure S4. Calculated EMF values (V) of conversion reactions between transition mental salts and lithium.



Figure S5. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and sodium.



Figure S6. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and magnesium.



Figure S7. Calculated EMF values (V) of conversion reactions between binary transition mental compounds and aluminum.



Figure S8. Calculated EMF (V) of conversion reactions between oxides and Li, Na, Mg or Al.



Figure S9. Calculated EMF (V) of conversion reactions between sulphides and Li, Na, Mg or Al.



Figure S10. Calculated EMF (V) of conversion reactions between nitrides and Li, Na, Mg or Al.



Figure S11. Calculated EMF (V) of conversion reactions between fluorides and Li, Na, Mg or Al.



Figure S12. Elements abundance in the earth.



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