

Supplementary Material (ESI) for Energy & Environmental Science

Three-dimensional porous V₂O₅ cathode with a ultra high rate capability

By Suqing Wang¹, Sirong Li¹, Yi Sun¹, Xuyong Feng¹, Chunhua Chen*¹

¹CAS Key Laboratory of Materials for Energy Conversion, Department of Materials Science and Engineering, University of Science and Technology of China, Hefei, Anhui, 230026 China

*To whom correspondence should be addressed. Emails: cchchen@ustc.edu.cn

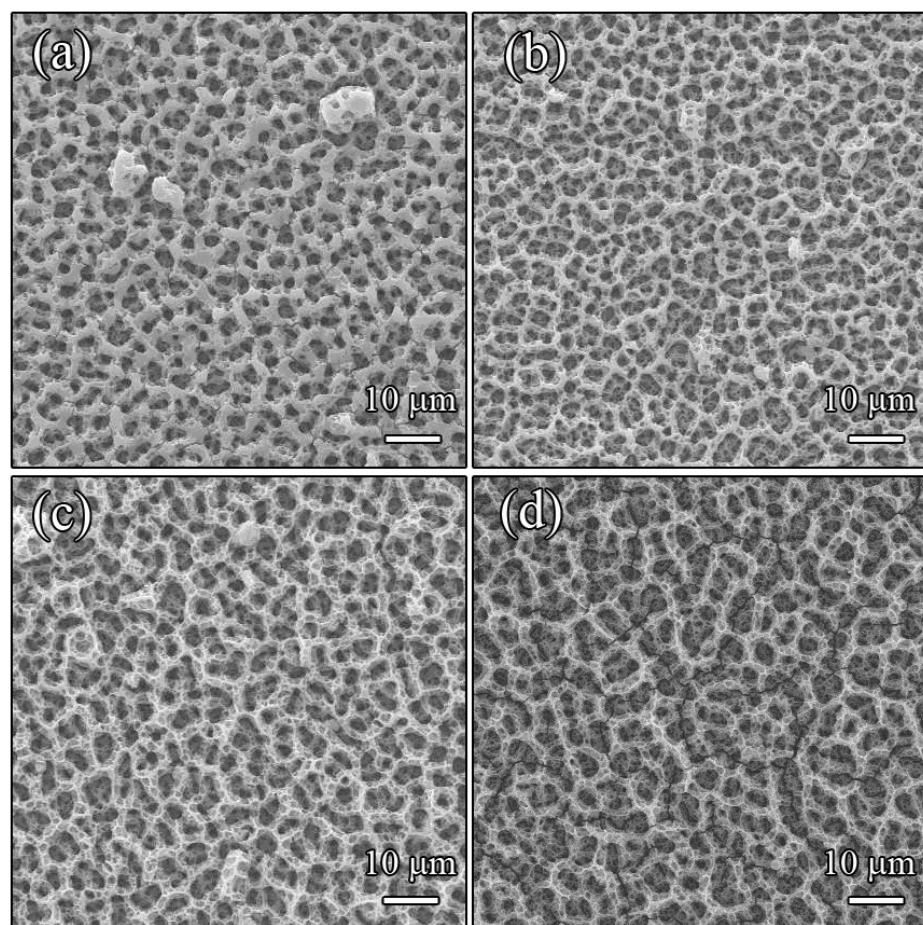


Figure S1. Low magnification SEM images of the thin films in different deposit temperatures: (a) 230°C, (b) 245°C, (c) 260°C, (d) 275°C.

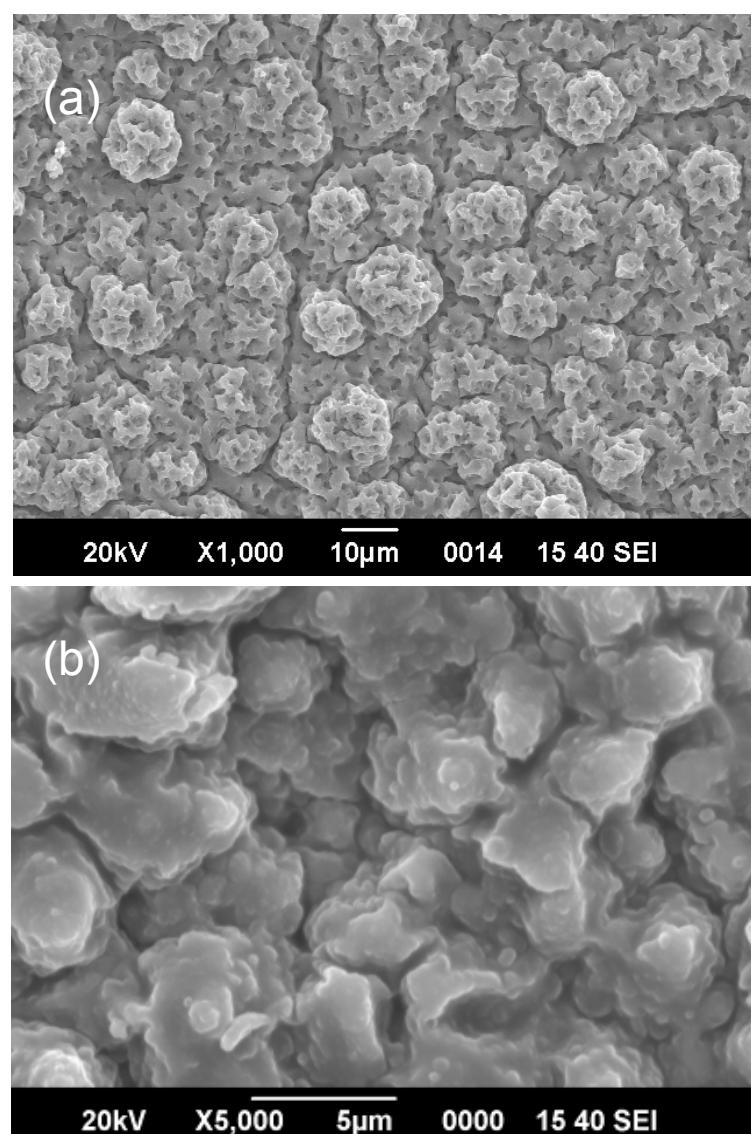


Figure S2. SEM image of the vanadium oxide thin film with different solvent: (a) use 1, 2-propylene glycol /Acetone/H₂O mixture as solvent; (b) the ratio of 1, 2-propylene glycol :ethanol:H₂O=75:15:10.

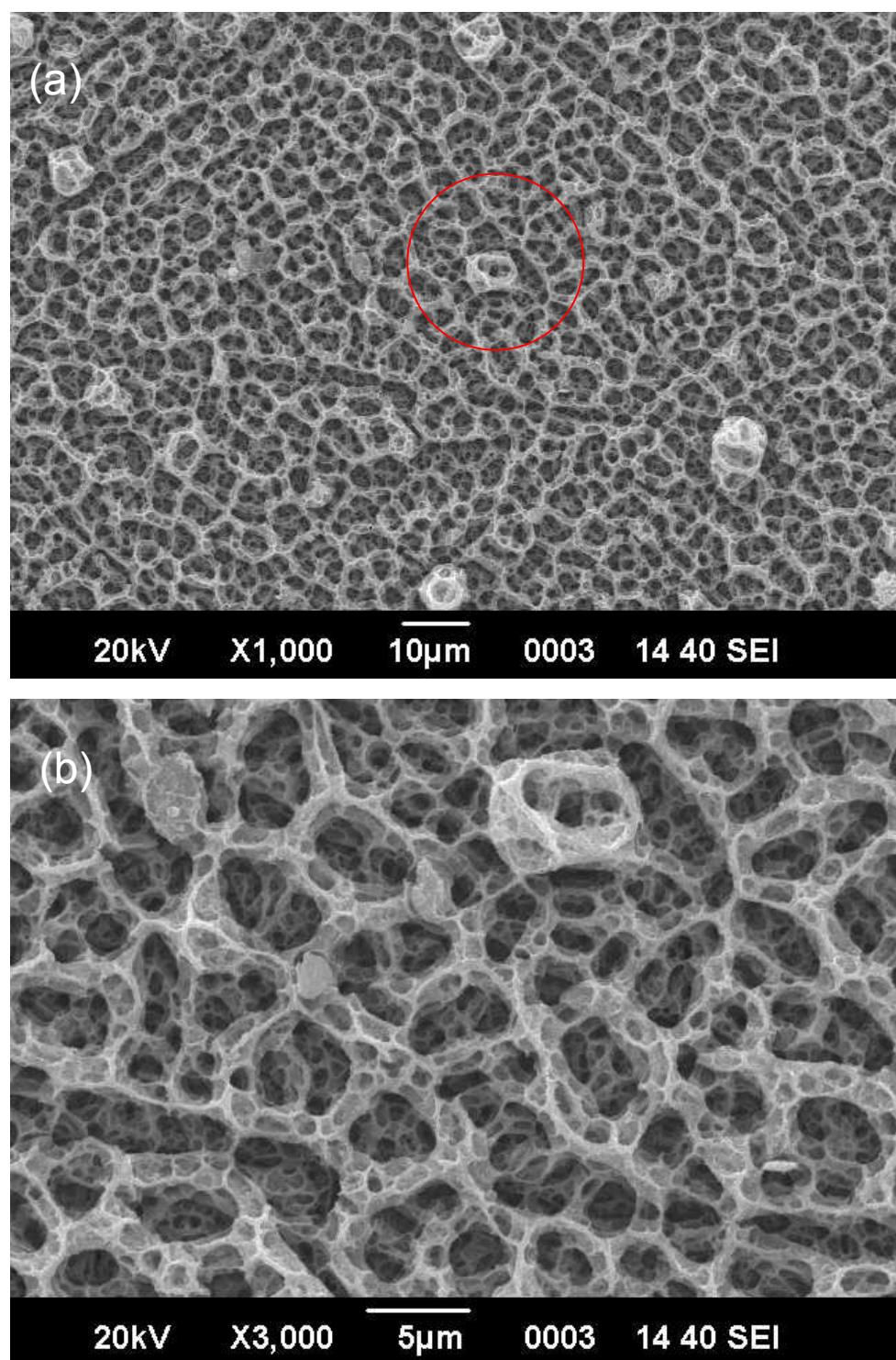


Figure S3. SEM images of V_2O_5 thin films

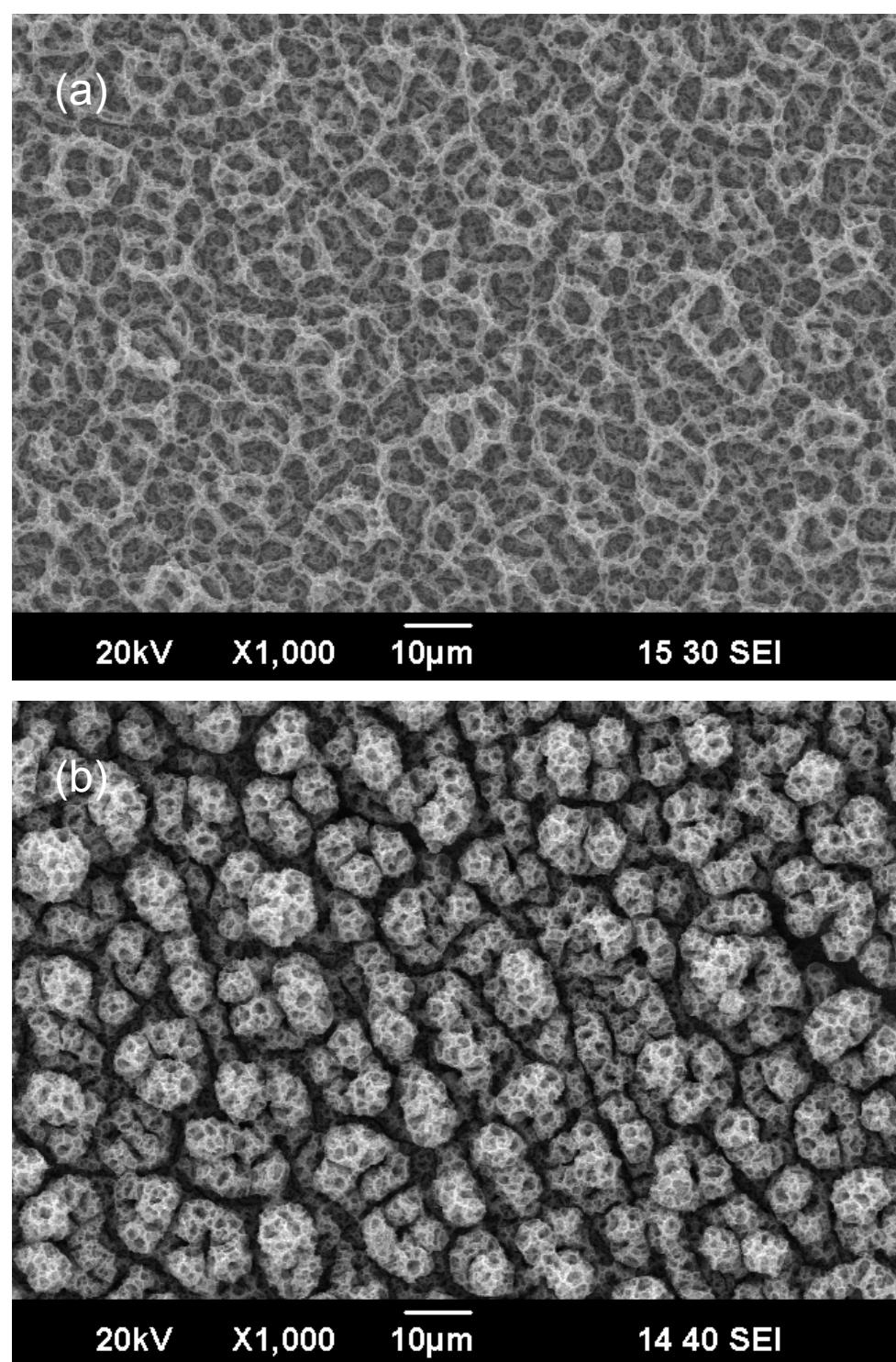


Figure S4. SEM images of V_2O_5 thin films after annealing at 350°C for 2h: (a) the deposit time=2h; (b) the deposit time=4h.

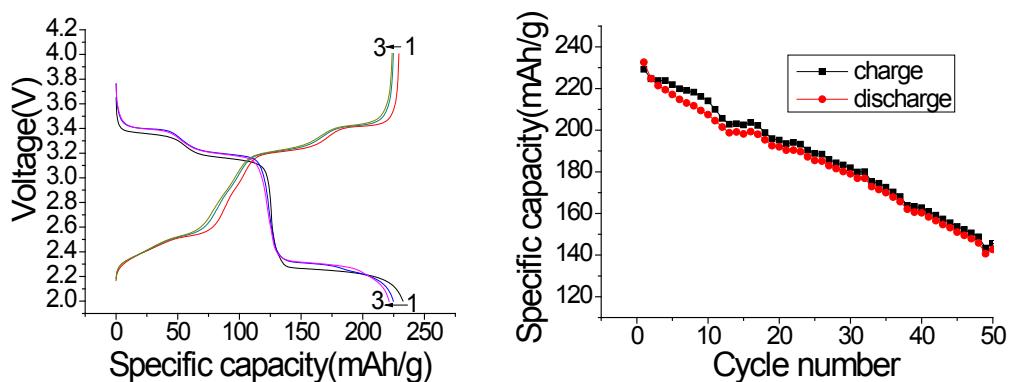


Figure S5. Galvanostatic cycling results of 3d-porous V_2O_5 thin film in the range of 1.5-4.0V: (a) first three discharge curves at the voltage range of 2.0-4.0 V, (b) cycling performance.

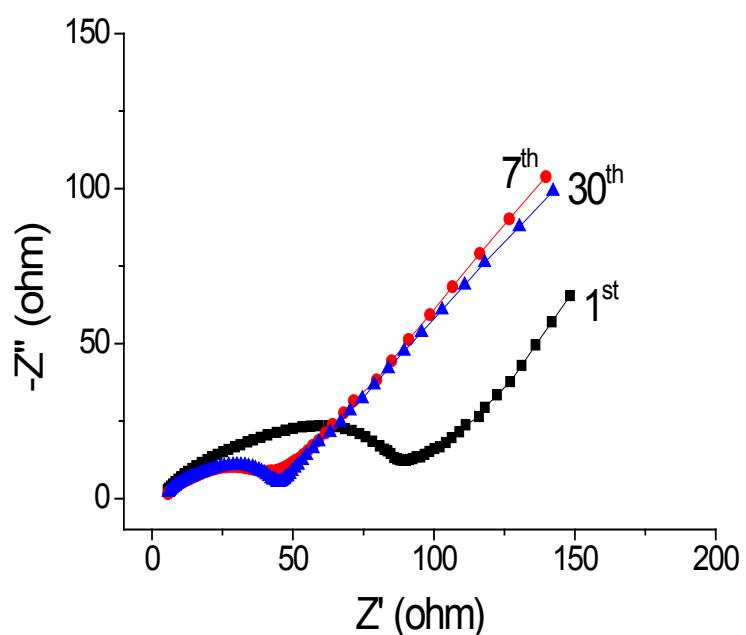


Figure S6. AC impedance spectra of V_2O_5 /Li cells after charged to 4.0 V in different cycles (1st, 7th, and 30th).