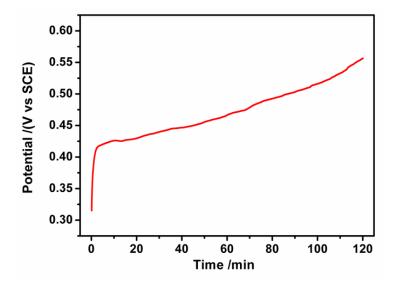
## **Supporting Information**

## Facile Synthesis of Large-area Manganese Oxide Nanorod Arrays as High-Performance Electrochemical Supercapacitor

Xihong Lu, $^a$  Dezhou Zheng, $^a$  Teng Zhai, $^a$  Zhaoqing Liu, $^{a,b}$  Yunyun Huang, $^a$  Shilei Xie, $^a$  Yexiang Tong $^{*a}$ 

<sup>a</sup>MOE Laboratory of Bioinorganic and Synthetic Chemistry, School of Chemistry and Chemical Engineering, Institute of Optoelectronic and Functional Composite Materials, Sun Yat-sen University, Guangzhou 510275, P. R. China

<sup>&</sup>lt;sup>b</sup>School of Chemistry and Chemical Engineering, Guangzhou University, Guangzhou 510006, China



**Fig. S1** Potential-time curve of MONRAs grown on FTO substrates prepared in solution of 0.01 M MnAc<sub>2</sub> + 0.02 M NH<sub>4</sub>Ac + 10% DMSO with a current density of 0.2 mA cm<sup>-2</sup> for 120 min at 70  $^{\circ}$ C.

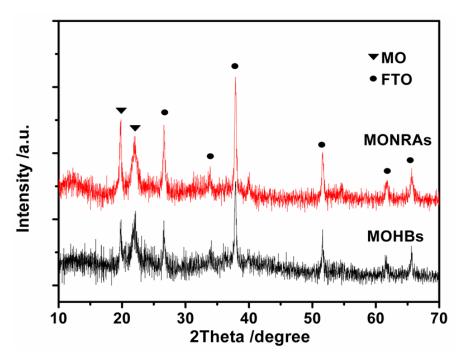


Fig. S2 XRD patterns of MONRAs and MOHBs.

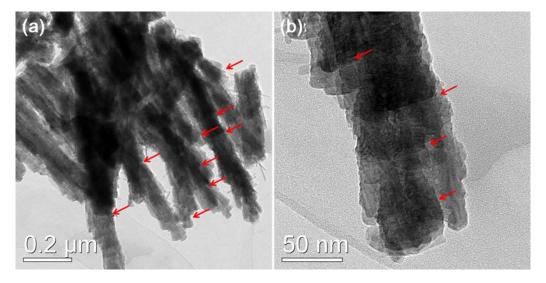
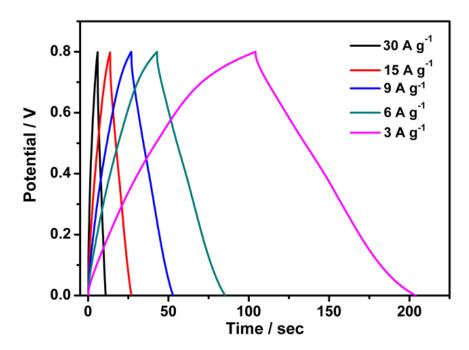
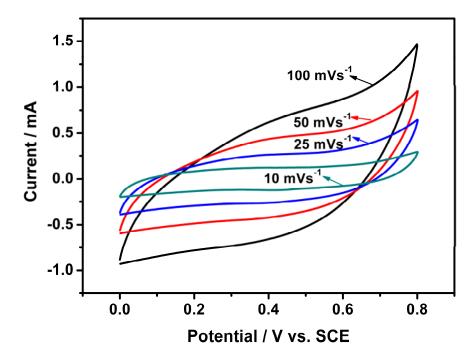


Fig. S3 TEM images of the MONRs.



**Fig. S4** Galvanostatic current charge/discharge curves of the as-prepared MOHBs at different current density.



**Fig. S5** Cyclic voltammetry curves of the MONRs grown on Ti substrates at different scan rates in 0.5 M Na<sub>2</sub>SO<sub>4</sub> aqueous solution.