

Electronic Supplementary Information (ESI)

Cobalt-phosphate (Co-Pi) catalyst modified Mo-doped BiVO₄ photoelectrodes for solar water oxidation

Satyananda Kishore Pilli,^{*a} Thomas E. Furtak,^b Todd G. Deutsch,^c John A. Turner,^c Logan Brown,^a and Andrew M. Herring^{*a}

Experimental detail

The Photoelectrochemical setup for the action spectra consisted of a 100 W tungsten arc lamp housing with an LPS-220 lamp power supply (Photon Technology International), a model SID-101 monochromator (Photon Technology International), and an EG&G model 263A potentiostat/galvanostat (Princeton Applied Research). Light exiting the monochromator was chopped at 37 Hz, and an optical filter was used to block any second-order diffraction. The signal was detected using a Stanford Research Systems model SR830 DSP lock-in amplifier using the chopping frequency as a reference. The spectral output of the lamp and monochromator was measured using a thermopile detector in place of the electrochemical cell. The voltage output from the thermopile is proportional to the incident power and it was plotted vs photon energy. To obtain photon flux from this plot, the photocurrent signal at each wavelength was divided by the energy (eV) of that wavelength. The semiconductor photocurrent spectrum was divided by the photon flux spectra to get a normalized spectrum. In these tests the electrode was held at 0.7 V vs. Ag/AgCl (3M KCl) and the photocurrent at each monochromatic light was registered

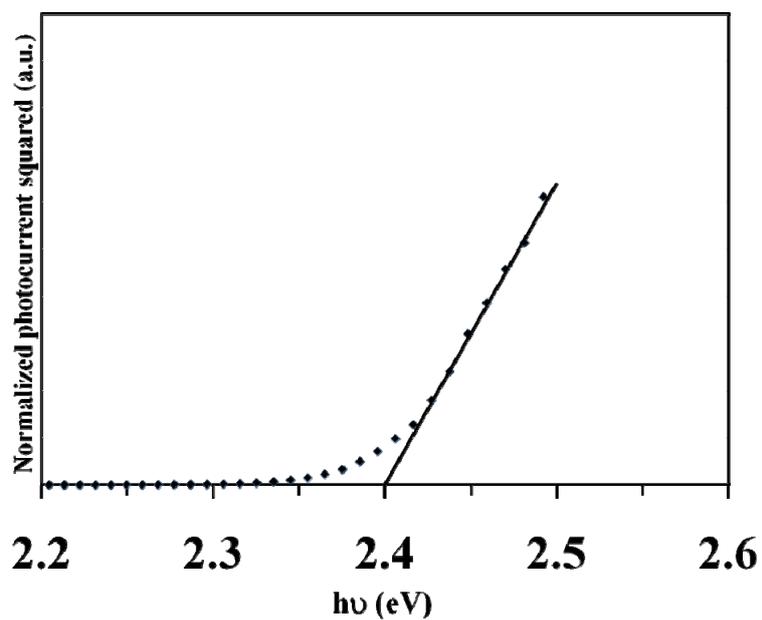


Fig. S1 Photocurrent action spectrum of $\text{BiV}_{0.98}\text{Mo}_{0.02}\text{O}_4$ electrode measured in 1 mM Na_2SO_3 aqueous solution (pH7 phosphate buffered), applied bias 0.7 V vs. Ag/AgCl.

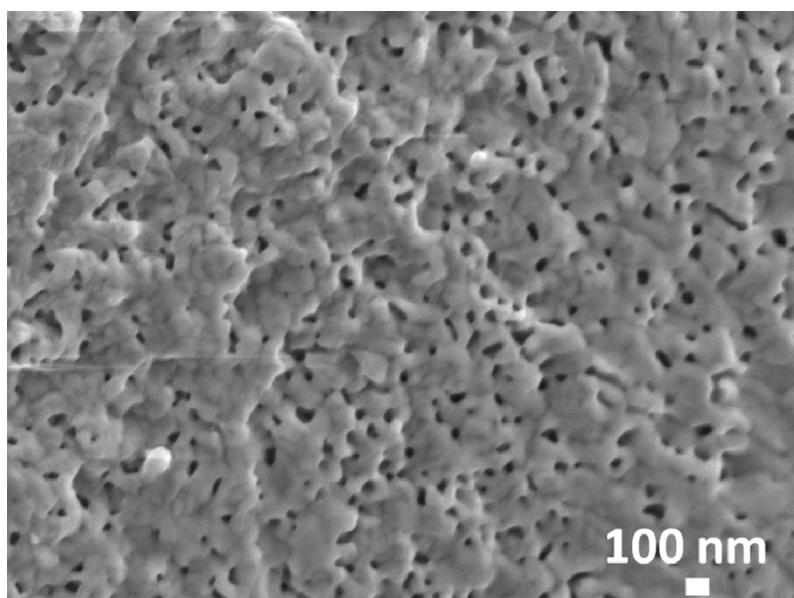


Fig. S2 SEM image of porous BiV_{0.98}Mo_{0.02}O₄ electrode at X30000 resolution.

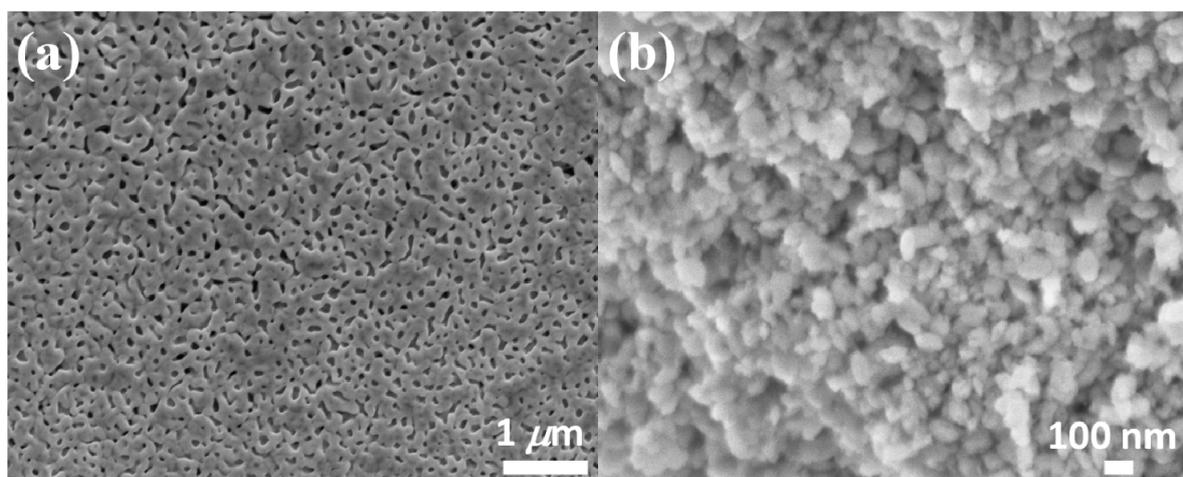


Fig. S3 SEM images of undoped BiVO₄ electrodes porous (a) and nonporous(b) prepared in the presence and absence of SDS.

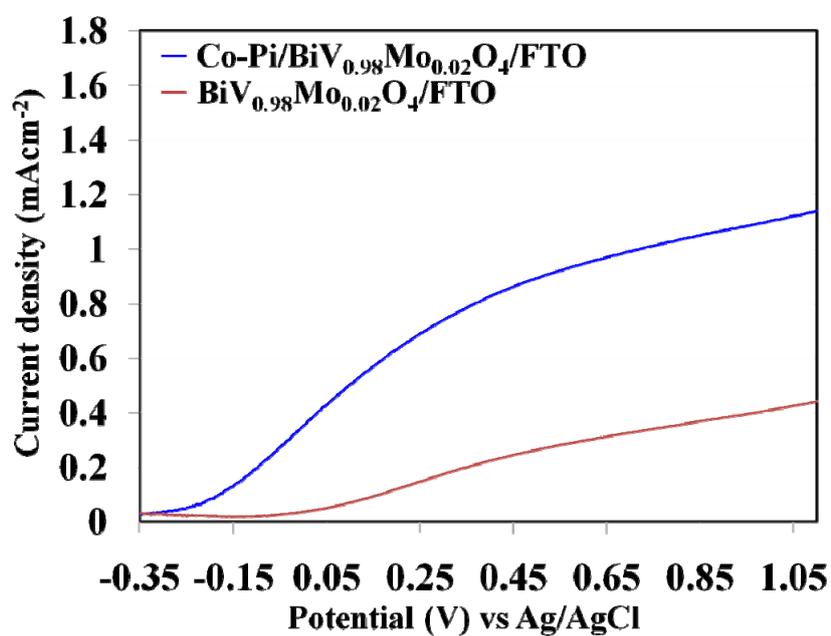


Fig. S4 Linear sweep voltammograms of BiV_{0.98}Mo_{0.02}O₄ (red) and Co-Pi/ BiV_{0.98}Mo_{0.02}O₄ (blue) electrodes measured (scan rate, 25 mV/s) with continuous light (100 mW/cm²) in 0.5M Na₂SO₄ (pH7) solution.

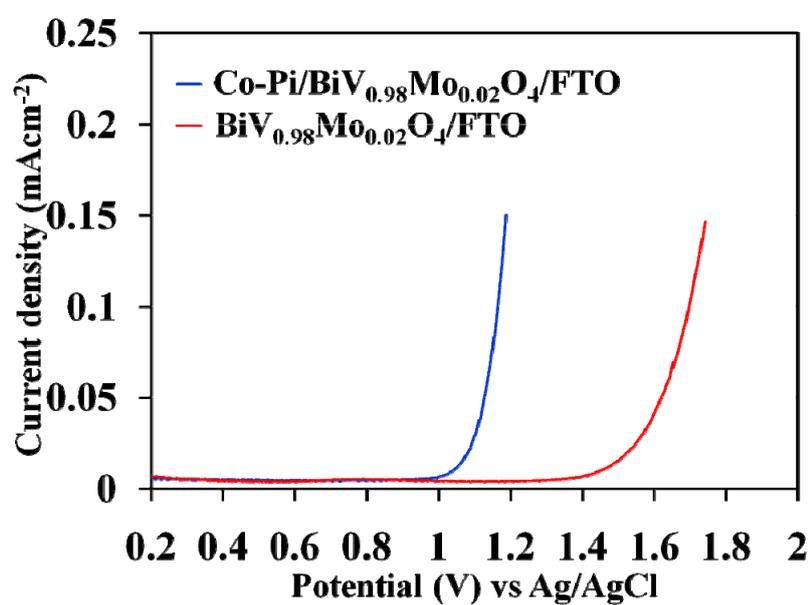


Fig. S5 Linear sweep voltammograms of BiV_{0.98}Mo_{0.02}O₄ (red) and Co-Pi/ BiV_{0.98}Mo_{0.02}O₄ (blue) electrodes measured (scan rate, 25 mV/s) under dark using 0.5M Na₂SO₄ aqueous solution in phosphate buffer (pH7) electrolyte.

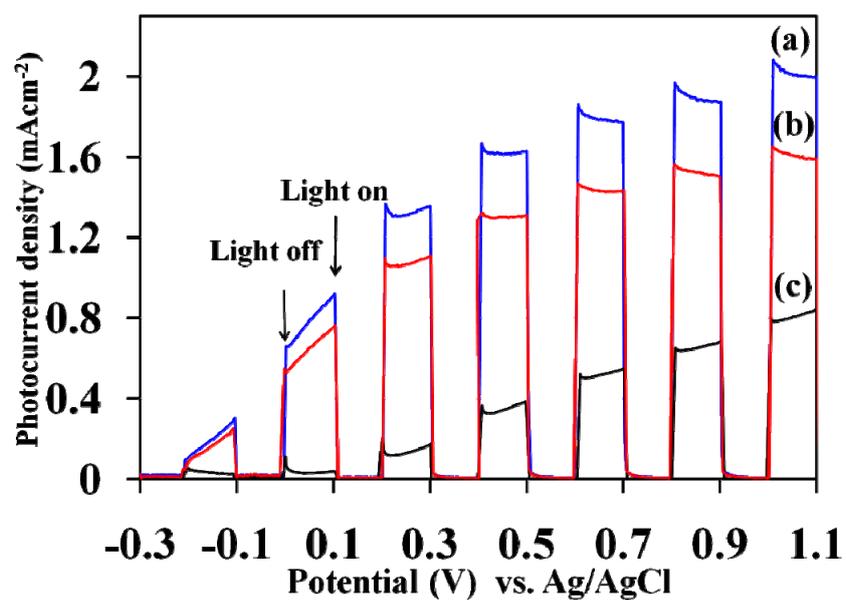


Fig. S6 Photocurrent-potential characteristics of BiV_{0.98}Mo_{0.02}O₄ electrodes modified with photo-assisted electrodeposited Co-Pi OEC (a); electrodeposited Co-Pi OEC (b) and unmodified BiV_{0.98}Mo_{0.02}O₄ electrode (c) measured in 0.5 M Na₂SO₄.

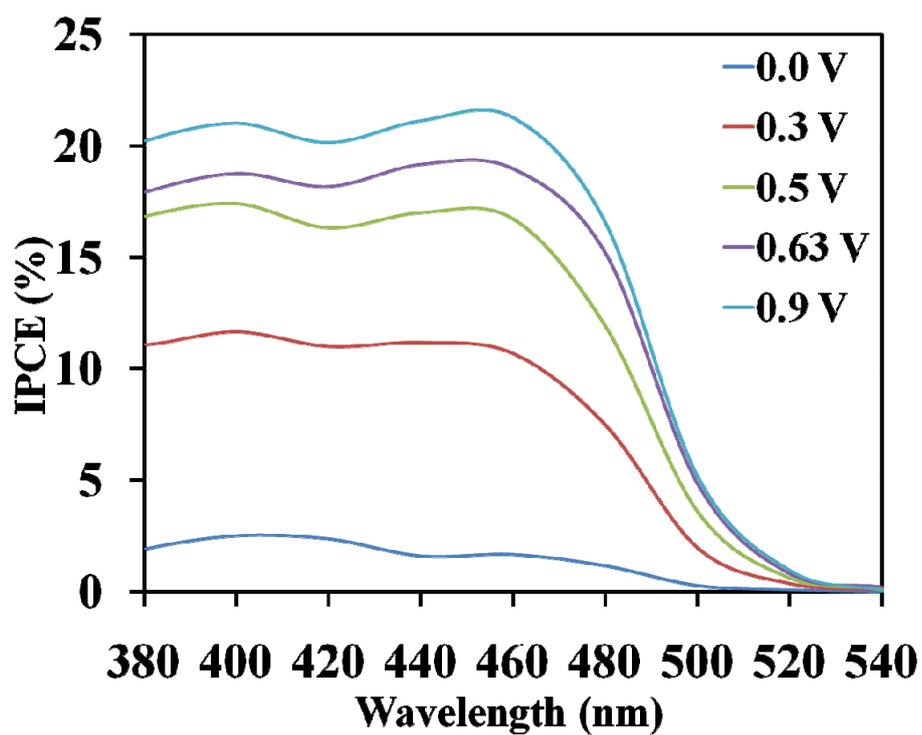


Fig S7 IPCE plot of $\text{BiV}_{0.98}\text{Mo}_{0.02}\text{O}_4$ electrode measured at 0.0, 0.3, 0.5, 0.63, 0.9 V vs. Ag/AgCl in 0.5 M Na_2SO_4 (pH 7 phosphate buffered).

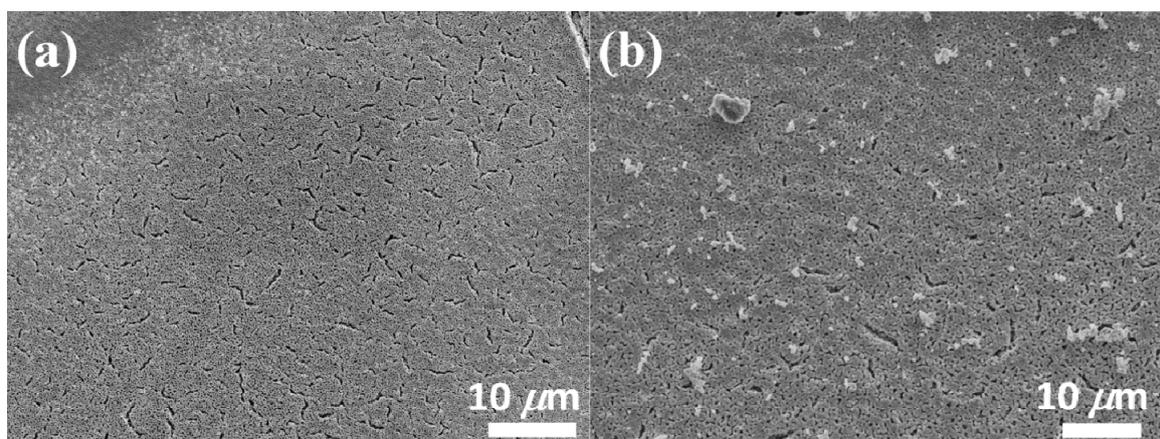


Fig S8 SEM images of $\text{BiV}_{0.98}\text{Mo}_{0.02}\text{O}_4$ (a) and $\text{Co-Pi/BiV}_{0.98}\text{Mo}_{0.02}\text{O}_4$ electrodes after 24 h of durability test.