Supporting Information

Semi-transparent polymer solar cells with 6% PCE, 25% average visible transmittance and color rendering index close to 100 for power generating window applications

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Material preparation, device fabrication, and characterizations

PBDTTT-C-T is synthesized according to the procedure reported earlier.¹ PBDTTT-C-T (10 mg ml⁻¹) and PC₇₁BM (American Dye Source, 15 mg ml⁻¹) were dissolved in 1,2-dichlorobenzene with 3 vol% of 1,8-diiodooctaneas the processing additive. 1,2-dichlorobenzene (Aldrich) and 1,8-diiodooctane (Aldrich) were used without further purification. The PBDTTT-C-T:PC₇₁BM blend solution was heatedfor homogenous mixing but was cooled and filtered with 0.45 μ m PTFE filter before use. The ZnO precursor was prepared following the procedure reported by Sun et al.²Zinc acetate dihydrate (Aldrich, 99.9%, 1 g) is dissolved in and ethanolamine (Aldrich, 99.5%, 0.28 g) are dissolved in 2-methoxyethanol (Aldrich, 99.8%, 10 mL) under vigorous stirring for a least 12 h and filtered (0.45 μ m PTFE) before use. C₆₀-SAM molecules were dissolved in chlorobenzene (Aldrich, 50 vol%) and tetrahydrofuran (THF, Aldrich, 50 vol%) co-solvent at the concentration of 1 wt%.

ITO-coated glass substrates were cleaned sequentially by sonication in detergent, de-ionized water, acetone, and isopropyl alcohol, and dried in a nitrogen stream, followed by 30 s of oxygen plasma treatment. The deposition of each layer is described as followed. ZnO precursor solution was first spin-coated on the ITO-coated glass substrates to deposit a thin ZnO layer (ca. 33 nm) and baked at 200°C for 1 h under ambient condition. The substrates were then transferred into a nitrogen-filled glovebox for the following processing. The self-assembly of the C₆₀-SAM was performed by spin-coating of the C₆₀-SAM solution on top of the ZnO surface. The substrates were washed with THF to remove unbound C₆₀-SAM molecules. Subsequently, the PBDTTT-C-T:PC₇₁BM active layer (ca. 90 nm) was spin-coated on the ZnO layer from filtered solution. No further post-treatments such as thermal annealing were performed on the devices. The substrates pumped down to high vacuum (< 2×10^{-6} Torr) in the evaporator chamber, and then MoO₃ (5 nm) and silver (6 nm - 60 nm) weresequentially evaporated onto the active layer through shadow masks to define the active area of the devices.

The current density-voltage (*J-V*) characteristics of photovoltaic devices were measured in an inert environment using a Keithley 2400 source-measurement unit under simulated AM 1.5 global illumination from an 300 Watt Solar Lightsolar simulator. A Hamamatsu silicon solar cell with a KG5 color filter, which is traced to the National Renewable Energy Laboratory (NREL),

was used as the reference cell to calibrate the light intensity to 1 sun (100 mW/cm^2). The spectral external quantum efficiency was recorded under chopped illumination in ambient atmosphere. The photocurrent was recorded with a lock-in-amplifier (Stanford Research Systems SR830).

Table S1. Special color rendering indices (CRIs) of the transmitted light of the STOPV devices under AM1.5G illumination.

	0 nm	6 nm	12 nm	18 nm	30 nm	60 nm
Special CRI - TCS01	97.7	95.0	96.3	97.5	97.9	97.6
Special CRI - TCS02	98.7	96.9	97.6	97.8	96.8	94.8
Special CRI - TCS03	99.5	99.3	98.9	98.3	96.0	91.2
Special CRI - TCS04	98.2	96.1	97.1	97.7	96.6	93.0
Special CRI - TCS05	98.1	95.6	96.9	98.0	98.4	98.2
Special CRI - TCS06	98.8	96.6	97.5	97.6	96.5	94.5
Special CRI - TCS07	98.1	97.8	97.2	97.2	96.2	93.7
Special CRI - TCS08	95.1	92.8	93.1	94.6	95.5	95.6







Figure S1. Simulations for the electric field intensity profile $|E|^2$ in the studied STOPV devices with (a) 6 nm, (b) 18 nm and (c) 30 nm of Ag electrode.



Figure S2. Simulations of exciton generation rate in the studied STOPV devices with (a) 6 nm, (b) 18 nm, and (c) 30 nm of Ag electrode.



Figure S3. (a) Drumheller Fountain in the University of Washington in a sunny day. (b) Look of the fountain through the STOPV devices with different thickness of silver electrodes. Blank ITO and glass substrate are also compared.

- 1. L. Huo, S. Zhang, X. Guo, F. Xu, Y. Li and J. Hou, Angew. Chem. Int. Ed., 2011, 50, 9697-9702.
- 2. Y. Sun, J. Seo, C. Takacs, J. Seifter and A. Heeger, Adv. Mater., 2011, 23, 1679-1683.