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Supporting Information

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Sodium insertion in carboxylate based materials and their application in 3.6 V full sodium cells

Ali Abouimrane*, Wei Weng, Hussameldin Eltayeb, Yanjie Cui, Jens Niklas, Oleg Poluektov, and Khalil Amine*

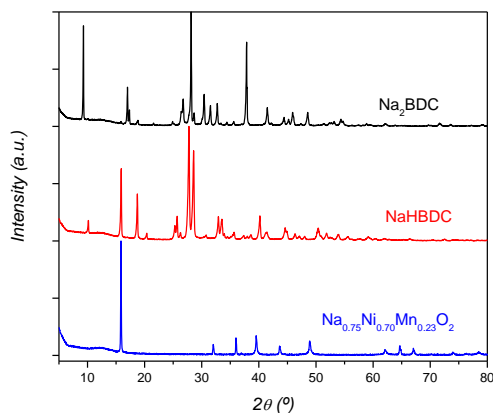


Figure S1. XRD of Na₂BDC, NaHBDC and Na_{0.75}Mn_{0.7}Ni_{0.23}O₂ pristine materials.

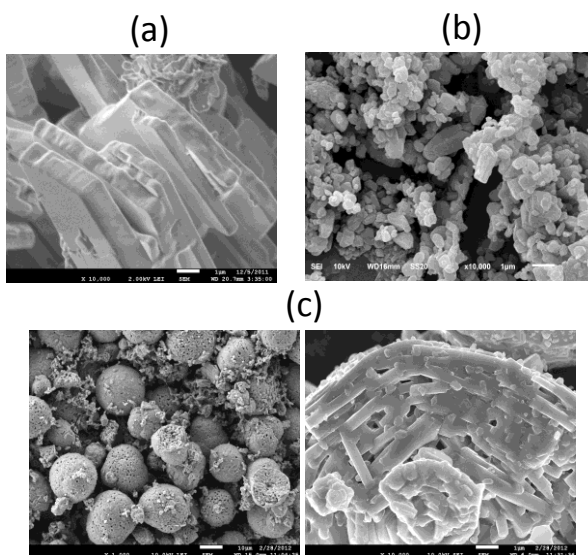


Figure S2. SEM pictures of Na₂BDC (a), NaHBDC (b) and Na_{0.75}Mn_{0.7}Ni_{0.23}O₂ (c). For Na_{0.75}Mn_{0.7}Ni_{0.23}O₂ material, SEM images shows the formation of ~ 20 μm particle size spherical secondary particles.

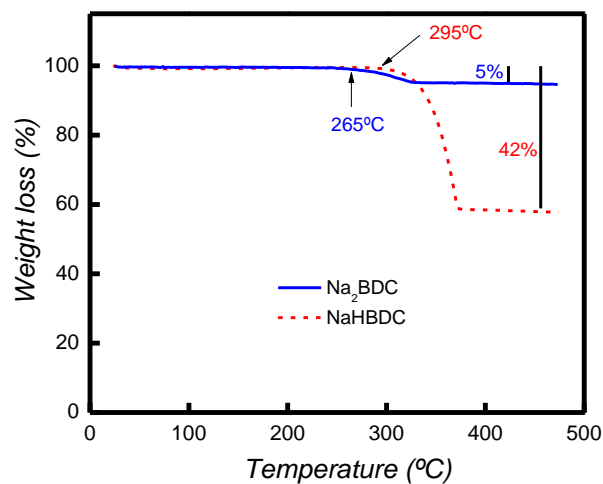
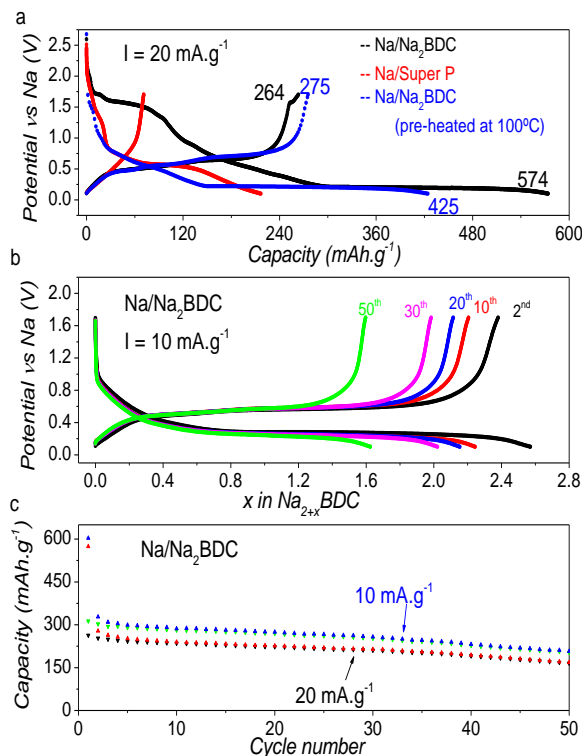


Figure S3. TGA of Na₂BDC, and NaHBDC showing good thermal stability until ~ 480°C for Na₂BDC and until ~ 300°C for NaHBDC.



10 **Figure S4.** Voltage profile and capacity retention of Na/Na₂BDC cell for 50 cycles under 10 and 20 mA/g current density.

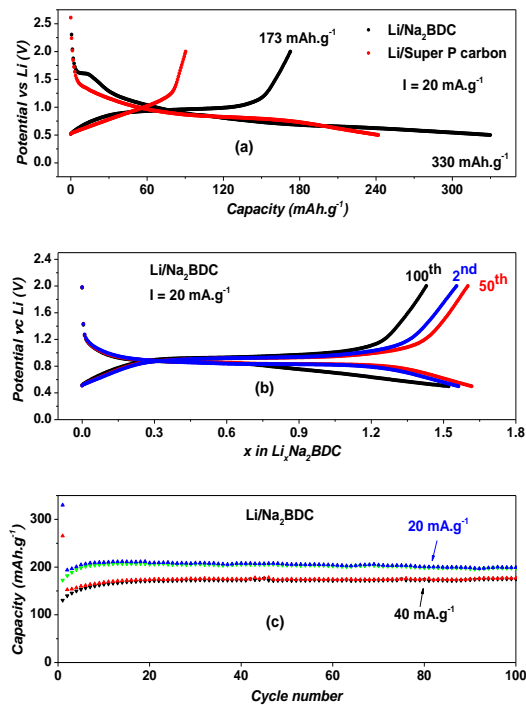


Figure S5. Voltage profile and capacity retention of Li/Na₂BDC cell for 100 cycles under 20 and 40 mA/g current density showing good cycleability.

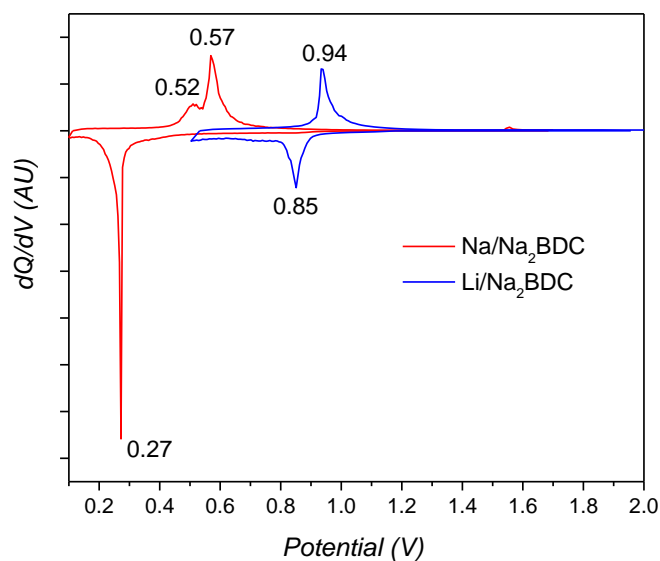
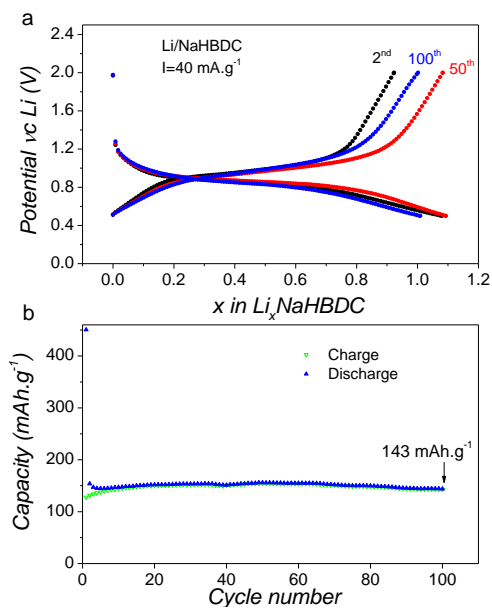
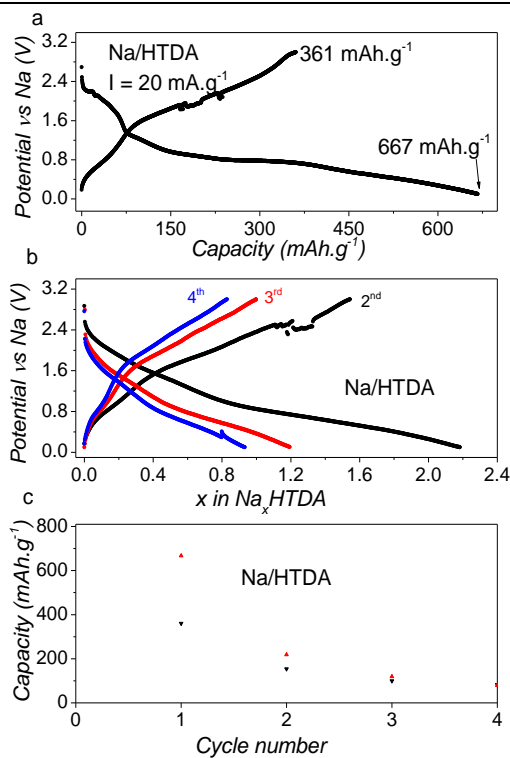


Figure S6. 2nd cycle dQ/dV curves of Na/Na₂BDC and Li/Na₂BDC systems under 20 mA/g current density.



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Figure S7. Voltage profile and capacity retention of Li/NaHBDC cell for 100 cycles under 40 mA/g current density.



10 **Figure S8.** Voltage profile and Capacity retention of Na/HTDA half-cell showing poor cycleability.

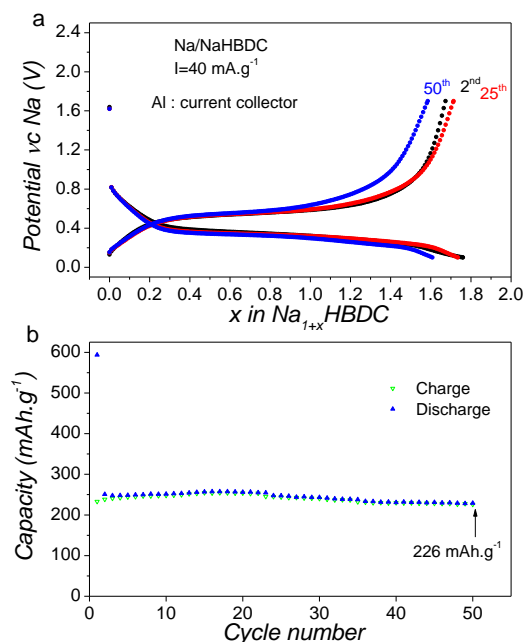


Figure S9. (a) Voltage profile and (b) Capacity retention of Na/NaHBDC half-cell for 50 cycles under the current rate of 40 mA.g^{-1} between 0.1V and 1.7V demonstrates that this anode can be cycled using Al current collector contrary to lithium batteries as no Al-Na was predicted by their phase diagram.