

Supplementary Information for

Lithium electrochemistry and cycling behaviour of ionic liquids of the cyano-anions

H. Yoon^{a,b}, G. H. Lane^{a,b}, Y. Shekibi^b, P. C. Howlett^c, M. Forsyth^c, A. S. Best^{*b} and D. R. MacFarlane^a

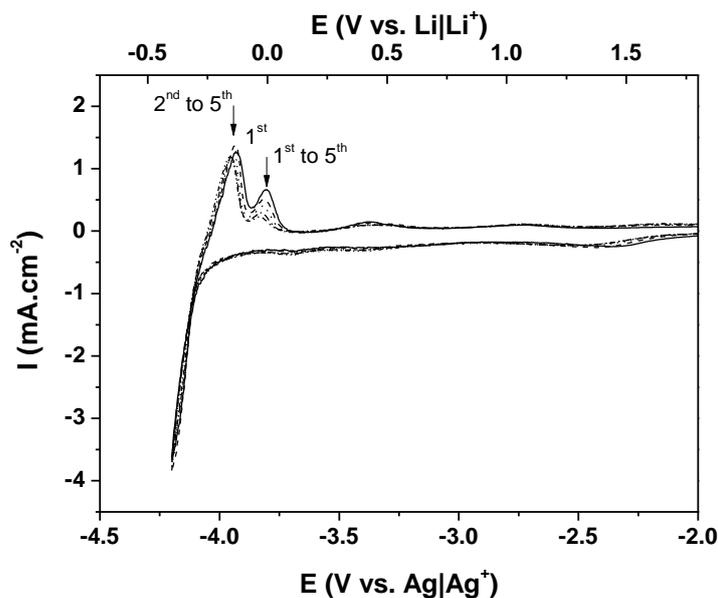
^aSchool of Chemistry, Monash University, 3800 Victoria, Australia.

^bCommonwealth Scientific and Industrial Research Organisation (CSIRO) Division of Energy Technology, Bayview Ave, Clayton. 3169 Victoria. Australia.; Tel: (+61)3 9545 8660; E-mail: adam.best@csiro.au

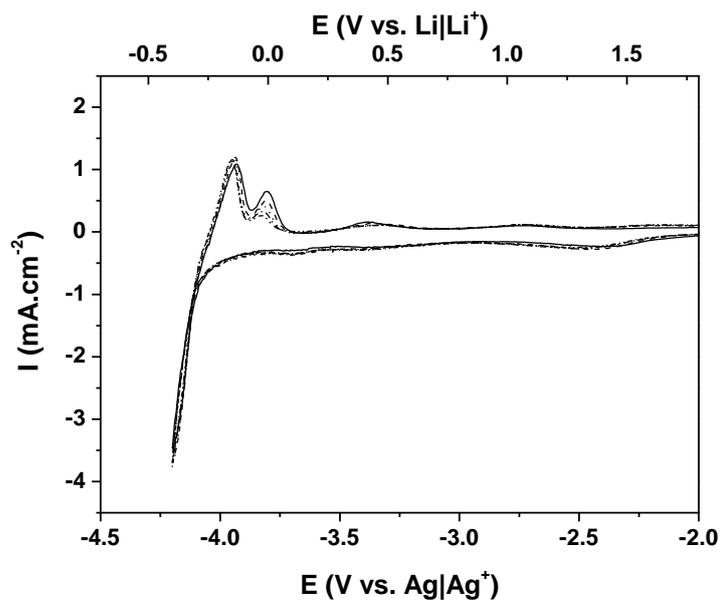
^cInstitute for Technology Research and Innovation (ITRI) Deakin University, Burwood. 3125 Victoria, Australia.

Cycling stability of C₄mpyr DCA with different moisture content

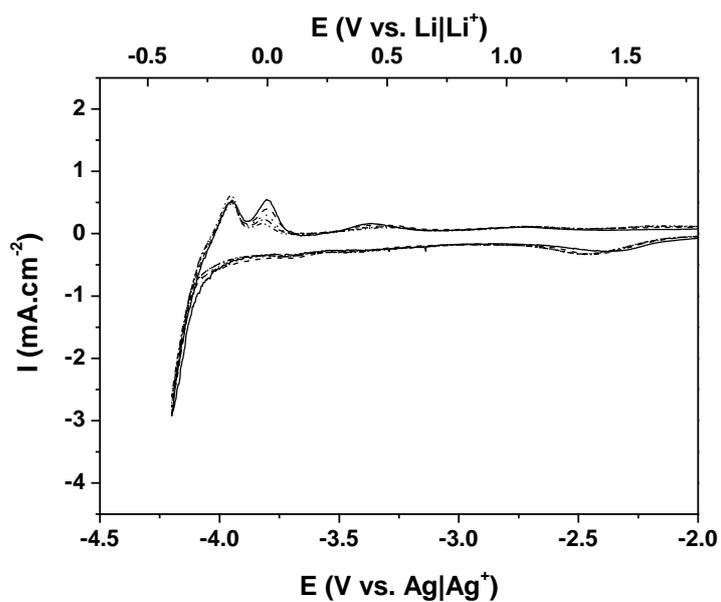
The figures below are in support of Figure 3 in the main text. Supplementary figures 1, 2 and 3 show the effect of moisture content, 233, 311 and 630 ppm respectively on the lithium plating and stripping in C₄mpyr DCA + 0.5 mol.kg LiDCA. The effect of the moisture decreases the maximum peak currents observed as well as to affect the electrochemical kinetics.



Supplement Fig.1. 5 CV scans of 0.5 mol kg⁻¹ LiDCA in C₄mpyr DCA at ambient temperature with 233 ppm of H₂O (solid line is the first scan)

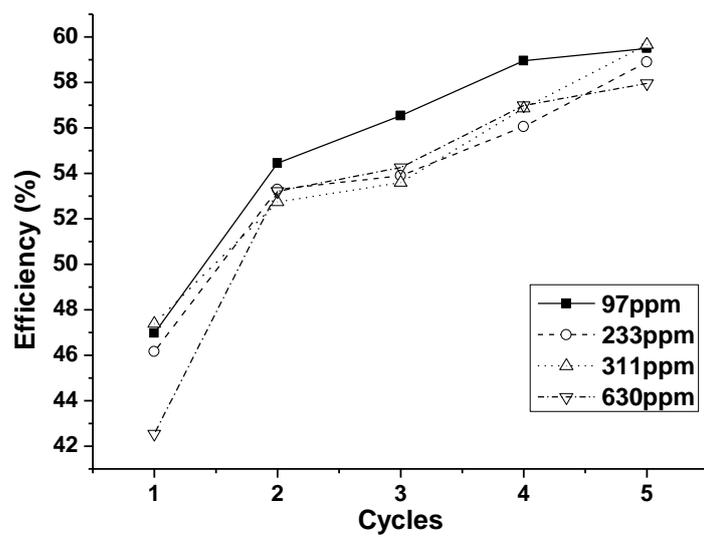


Supplement Fig.2. 5 CV scans of 0.5 mol kg⁻¹ LiDCA in C₄mpyr DCA at ambient temperature with 311 ppm of H₂O (solid line is the first scan)



Supplement Fig.3. 5 CV scans of 0.5 mol kg⁻¹ LiDCA in C₄mpyr DCA at ambient temperature with 630 ppm of H₂O (solid line is the first scan)

Supplementary figure 5 shows the effect of moisture on the coulombic efficiency of Lithium plating and stripping for the first 5 cycles. In all cases, independent of moisture content, the cycling efficiency of the .5 mol kg⁻¹ LiDCA in C₄mpyr DCA reaches a maximum of 60 %.



Supplement Fig.4. Coulombic efficiencies of 0.5 mol kg⁻¹ LiDCA in C₄mpyr DCA electrolytes with different amount of H₂O.