## **Supplementary Information**

## Gold Nanoparticles Inlaid TiO<sub>2</sub> Photoanodes: A Superior Candidate for High-Efficiency Dye-Sensitized Solar Cells

*Yan Li*,<sup>†,‡</sup> *Hong Wang*,<sup>†,⊥</sup> *Quanyou Feng*,<sup>†,⊥</sup> *Gang Zhou*,<sup>†</sup> *and Zhong-Sheng Wang*\*<sup>†</sup>

<sup>†</sup> Department of Chemistry, Lab of Advanced Materials, Fudan University, Shanghai 200438, P. R. China

<sup>‡</sup> Present address: Key Laboratory of Atomic and Molecular Physics & Functional Materials of Gansu Province, College of Physics and Electronic Engineering, Northwest Normal University, Lanzhou 730070, P. R. China

\* E-mail: zs.wang@fudan.edu.cn



Figure S1. Synthetic route of the cross-linked polymer (CLP).



*Figure S2*. TEM image of the as-synthesized polymer stabilized Au nanoparticles dispersed in methanol.



*Figure S3*. UV-vis absorption spectrum of the as-synthesized polymer stabilized Au nanoparticles dispersed in methanol.



*Figure S4.* XPS spectrum of the Au-TiO<sub>2</sub> (Au/TiO<sub>2</sub> = 0.800wt%) composite film.

Electronic Supplementary Material (ESI) for Energy & Environmental Science This journal is O The Royal Society of Chemistry 2013



*Figure S5.* (a) The chemical structure of **FNE29** and (b) the UV-vis absorption spectrum of **FNE29** in THF.



*Figure S6.* Efficiency evolution of the best DSSC fabricated with 0.168 wt% Au-TiO<sub>2</sub> photoanode subjected to aging.



*Figure S7.* The multiples in enhancement of maximum absorbance of **FNE29** caused by the plasma effect as a function of Au amount. Assuming that the absorbance of dye-loaded TiO<sub>2</sub> film is proportional to the dye amount without regard to the plasma effect, this suppositional absorbance  $(A'(Au-TiO_2))$  for each dye-loaded Au-TiO<sub>2</sub> film is estimated from the absorbance of dye-loaded pure TiO<sub>2</sub> film  $(A(TiO_2))$  times the dye amount in each film  $(c(Au-TiO_2))$  divided by the dye amount in pure TiO<sub>2</sub> film  $(c(TiO_2))$ . Therefore, multiple enhancement =  $A(Au-TiO_2)/A'(Au-TiO_2)$ , where  $A(Au-TiO_2)$  is the measured absorbance for the dye-loaded Au-TiO<sub>2</sub> film.



Figure S8. Electron lifetime as a function of charge density at open circuit.



Figure S9. Structures of dyes MK1 (ref. 1), MK3 (ref.1) and FNE31 (ref.2).

Electronic Supplementary Material (ESI) for Energy & Environmental Science This journal is  $\ensuremath{\mathbb{O}}$  The Royal Society of Chemistry 2013

Dye	Photoanode	$V_{\rm oc}~({\rm mV})$	$J_{\rm sc}$ (mA cm <sup>-2</sup> )	FF	η (%)
MK1	TiO <sub>2</sub>	782	7.53	0.64	3.77
	Au-TiO <sub>2</sub>	815	10.21	0.68	5.66
MK3	$TiO_2$	713	7.90	0.75	4.22
	Au-TiO <sub>2</sub>	749	11.95	0.76	6.80
FNE31	TiO <sub>2</sub>	746	6.11	0.75	3.42
	Au-TiO <sub>2</sub>	774	8.73	0.76	5.14

Table S1. Photovoltaic performance parameters for DSSCs based on various dyes and 4 µm films<sup>a</sup>.

<sup>a</sup> The Au/TiO<sub>2</sub> weight ratio in the Au-TiO<sub>2</sub> film is 0.104wt%.

## References

- Wang, Z.-S.; Koumura, N.; Cui, Y.; Takahashi, M.; Sekiguchi, H.; Mori, A.; Kubo, T.; Furube, A.; Hara, K. *Chem. Mater.* 2008, 20, 3993.
- Tomas, K. R. J.; Hsu, Y.-C.; Lin, J. T.; Lee, K.-M.; Ho, K.-C.; Lai, C.-H.; Cheng, Y.-M.; Chou, P.-T. *Chem. Mater.* 2008, 20, 1830.