

Supplementary material

1. Time courses for the photodegradation of the herbicides

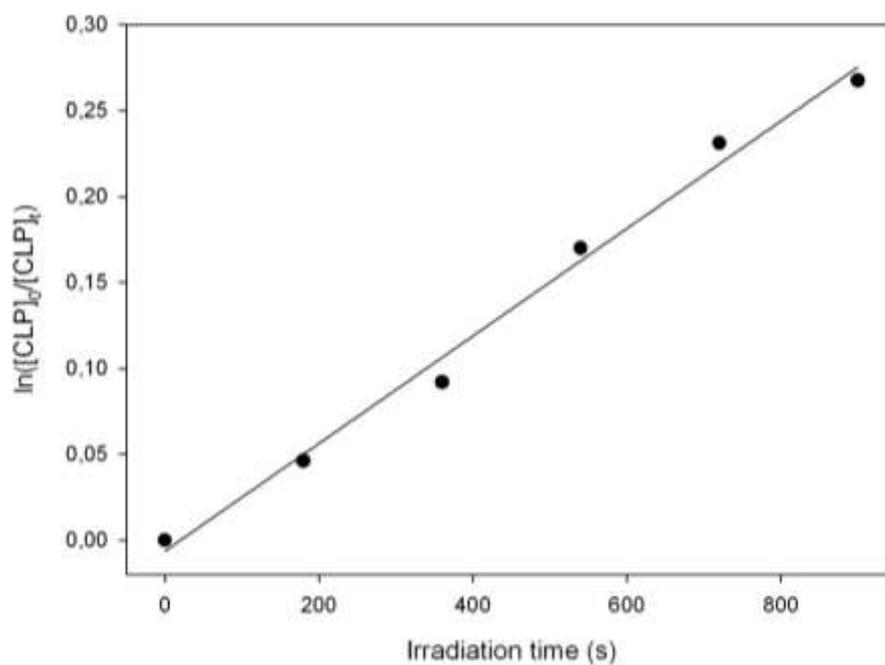


Fig.1SM. Photodegradation of chlorpropham

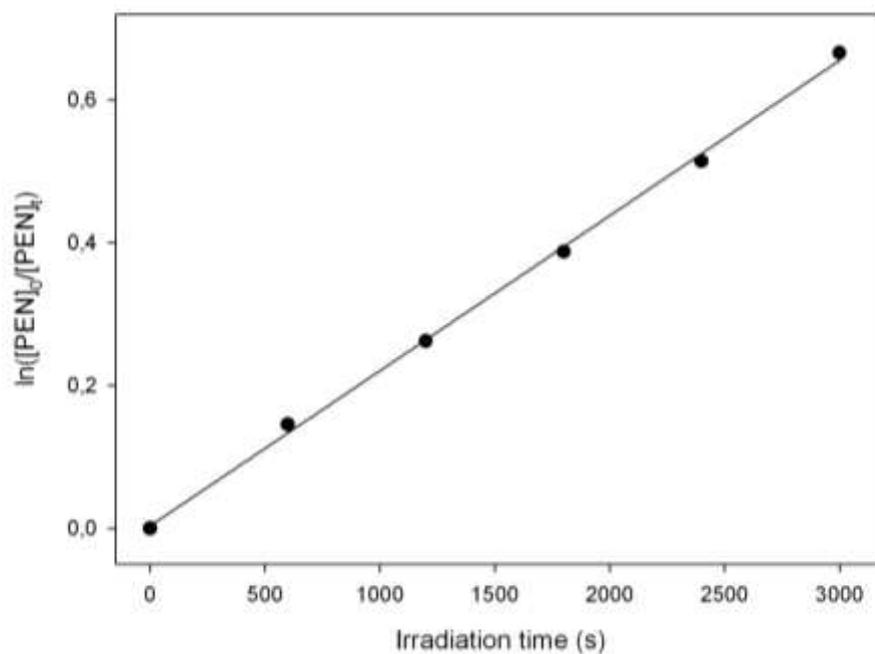


Fig.2SM. Photodegradation of phenisopham

2. Isolation and identification of photoproducts

Procedures for isolation and characterization of phototransformation products from chlorpropham (**1a**)

A 1×10^{-3} M solution of chlorpropham (30 mg) in 140 mL of H₂O/CH₃CN (9:1 v/v) was irradiated with UV-B lamps for 1 h. Then, the solvents were evaporated under vacuum and the residue was separated by preparative TLC. Elution with CH₂Cl₂/AcOEt (9:1 v/v) gave **1a** (5 mg) and photoproduct **2a** (22 mg) at decreasing R_fs. Compound **2a** was identified by spectroscopic data.²³

Isopropyl 3-hydroxyphenylcarbamate (2a): UV λ_{\max} (H₂O/CH₃CN 9:1 v/v) nm 235 (log ϵ 3.7), 280 nm (log ϵ 3.1); EI-MS m/z 195, 153, 136, 109; IR ν_{\max} (CHCl₃) 3576, 3425, 1719, 1519, 1214 cm⁻¹; ¹H NMR (CDCl₃) δ 7.33 (1H, br s), 7.13, (1H, t, J=8.1 Hz), 6.65 (1H, d, J=8.1 Hz), 6.63 (1H, s), 6.56 (1H, d, J=8.0 Hz), 6.38 (1H, br s), 5.01 (1H, sept, J=6.5 Hz), 1.30 (6H, d, J=6.5 Hz); ¹³C NMR (CDCl₃) δ 157.1, 153.7, 139.2, 130.2, 110.8, 110.6, 106.0, 69.4, 22.3.

Procedures for isolation and characterization of phototransformation products from phenisopham(**1b**)

A 1×10^{-3} M solution of phenisopham (50 mg) in 146 mL of H₂O/CH₃CN (7:3 v/v) was irradiated in open quartz tubes with UV-B lamps for 2 h. Then, the solvents were evaporated under vacuum and the residue was separated by preparative TLC. Elution with CH₂Cl₂/AcOEt (9:1 v/v) gave phenisopham **1b** (15 mg), photoproducts **5b** (2 mg), **2b** (<1 mg), **6b** (4 mg) and **4b** (25 mg) at decreasing R_fs. Product **2b** (=2a) was identified as above. Products **4b-6b** were spectroscopically characterized.

Isopropyl 2-(ethyl(phenyl)carbamoyl)-3-hydroxyphenylcarbamate (4b): UV λ_{\max} (H₂O/CH₃CN 9:1 v/v) nm 230 (log ϵ 3.0), 270 nm (log ϵ 2.4); EI-MS m/z 342, 180, 162, 121, 106; IR ν_{\max} (CHCl₃) 3561, 3411, 1731, 1685, 1635, 1627, 1210 cm⁻¹; ¹H NMR (CDCl₃) δ 7.19-7.17 (4H, m), 7.07 (1H, t, J=8.0 Hz), 7.01 (2H, d, J=9.0 Hz), 6.90 (1H, brs), 6.81 (1H, brs), 6.50 (1H, d, J=8.0 Hz), 4.96 (1H, sept, J=6.5 Hz), 4.10 (1H, m), 3.86 (1H, m), 1.31 (6H, m), 1.20 (3H, t, J=7.1 Hz). ¹³C NMR (CDCl₃) δ 171.0, 154.8, 154.7, 152.6, 141.1, 135.5, 131.5, 128.6, 127.3, 126.6, 112.3, 111.4, 68.9, 45.2, 22.1, 14.15.

Isopropyl 4-(ethyl(phenyl)carbamoyl)-3-hydroxyphenylcarbamate (5b): UV λ_{max} (H₂O/CH₃CN 9:1 v/v) 232 nm (log ϵ 1.9), 273 nm (log ϵ 2.4); EI-MS m/z 342, 222, 180, 148, 121; IR ν_{max} (CHCl₃) 3628, 3424, 1730, 1630, 1626, 1209 cm⁻¹; ¹H NMR (CDCl₃) δ 11.57 (1H, s), 7.33 (2H, t, J=9.0 Hz), 7.27 (1H, m), 7.10 (2H, d, J=9.0 Hz), 6.86 (1H, d, J=2.0 Hz), 6.54 (1H, d, J=9.0 Hz), 6.43-6.35 (2H, brs), 4.96 (1H, sept, J=6.5 Hz), 3.93 (2H, q, J=7.0 Hz), 1.25 (6H, d, J=6.5 Hz), 1.21 (3H, t, J=7.0 Hz); ¹³C NMR (CDCl₃) δ 170.7, 162.5, 152.9, 152.7, 152.4, 141.9, 131.5, 129.6, 127.5, 127.2, 107.5, 106.1, 69.0, 46.5, 21.9, 12.5.

Isopropyl 2-(ethyl(phenyl)carbamoyl)-5-hydroxyphenylcarbamate (6b): UV λ_{max} (H₂O/CH₃CN 9:1 v/v) nm 233 (log ϵ 3.3), 278 nm (log ϵ 2.7); EI-MS m/z 342, 180, 162, 121, 106; IR ν_{max} (CHCl₃) 3583, 3300, 1722, 1648, 1627, 1214 cm⁻¹; ¹H NMR (CDCl₃) δ 9.29 (1H, s), 7.78 (1H, d, J=1.5Hz), 7.26-7.22 (2H, m), 7.16 (1H, t, J=8.0 Hz), 6.99 (2H, d, J=8.0 Hz), 6.74 (1H, d, J=8.0 Hz), 6.09 (1H, dd, J=8.0, 1.5Hz), 6.01 (1H, br s), 4.99 (1H, sept, J=6.5 Hz), 3.98 (2H, q, J=7.0 Hz), 1.31 (6H, d, J=6.5 Hz), 1.24 (3H, t, J=7.0 Hz). ¹³C NMR (CDCl₃) δ 170.4, 158.4, 156.9, 153.7, 143.9, 140.9, 132.4, 129.7, 127.6, 126.9, 108.9, 106.4, 69.4, 46.0, 22.5, 13.4.

3. HPLC chromatograms

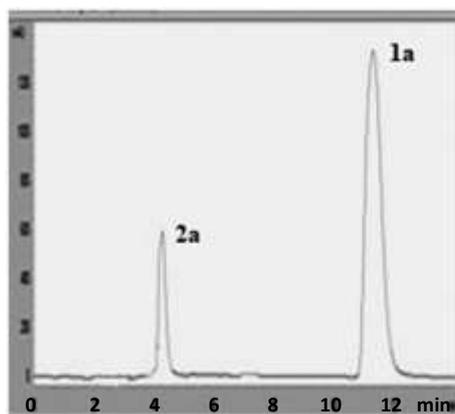


Fig.3SM HPLC profile of the irradiation mixture of chlorpropham (**1a**) in H₂O/CH₃CN (9 : 1 v/v, 5 x 10⁻⁵ M) after 15 min of UV-B irradiation.

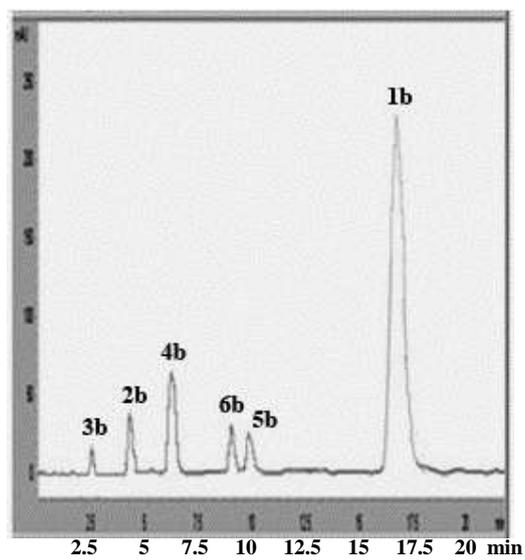


Fig.4SM HPLC profile of the irradiation mixture of phenisopham (**1b**) in H₂O/CH₃CN (9:1 v/v, 5 x 10⁻⁵ M) after 30 min of UV-B irradiation.