

1 Electronic Supplementary Information for:

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3 **Surface and Subsurface Attenuation of Trenbolone Acetate Metabolites and Manure-**

4 **Derived Constituents in Irrigation Runoff on Agro-Ecosystems**

Gerrad D. Jones^{1,2}, Peter V. Benchetler¹, Kenneth W. Tate³, Edward P. Kolodziej^{1*}

⁸ ¹Department of Civil and Environmental Engineering, University of Nevada-Reno, MS 0258,
⁹ Reno, Nevada, USA, 89557.

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11 ²Current Address: Eawag, Swiss Federal Institute of Aquatic Science and Technology,
12 Überlandstrasse 133, 8600 Dübendorf, Switzerland.
13
14 ³Department of Plant Sciences, University of California-Davis, MS 1, Davis, California, USA,

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19 *Corresponding author contact information:
20 Kolodziej, E.P.; Telephone: (775) 682-5553; fax: (775) 784-1390; email: koloj@unr.edu
21 Department of Civil and Environmental Engineering, MS 258, 1664. N. Virginia St, Reno, NV
22 89557

23 Contents: 11 Pages, 5 Figures

24 *Surface Transport*- We evaluated the transport of manure-derived constituents on VFS on
25 four different dates (May 24, May 31, June 8, and July 3, 2012). In general, all experiments
26 were conducted as described within in the main text under “Surface Transport”. In short, each
27 VFS (strips #1, 2, 3) was saturated with irrigation water for 30 minutes to promote runoff during
28 the experiment and to minimize the leachate volume required to generate runoff. After 30
29 minutes, leachate was applied from a 1500 L tank simultaneously to the top of each VFS. On
30 May 24, samples were collected at t = 0, 30, and 60 minutes. In order to promote mass transfer
31 and increase concentrations in the leachate on the experiment conducted on May 31, the manure
32 was broken apart and thoroughly mixed within the tank. The leachate was then applied
33 simultaneously to each the VFS, but because of the high solids content, samples were very
34 difficult to filter and were composited for convenience. The composite samples was first
35 centrifuged then filtered as described within the main text. From both experiments, only TBA
36 metabolites were analyzed. On June 8, samples were collected at t = 0 (i.e., first flush), 15, 30,
37 45, and 60 minutes while on 3 July, samples were collected at t = 15, 30, 45, 60, and 75 minutes.
38 Otherwise, the later experiments were conducted identically. For these experiments, TBA
39 metabolites, total ammonia, nitrate, nitrite, orthophosphate, and DOC were analyzed and for the
40 last experiment, total coliforms and *E. coli* were measured but from composite samples only.

41

42 *Colloidal and Soil Partitioning*- We calculated the dissolved fraction of steroid in water
43 (F_D) from physical and chemical properties of the soil based on the following mass balance:

$$44 \quad F_D = \frac{C_w V_w}{C_w V_w + C_s M_s + C_c M_c} = \frac{1}{1 + K_D r + K_{OC} DOC} \quad (1)$$

45 where C_w , C_s , and C_c are the TBA metabolite concentration in water, on soil, and on organic
46 carbon, respectively, V_w is the volume of water within a sample , and M_s and M_c are the mass of

47 soil and the mass of organic carbon within a sample, respectively. In practice, the right hand side
 48 of equation 1 represents the reduced form of the mass balance and is more suitable for solving
 49 for the dissolved fraction of steroids. It is derived from the soil water partitioning coefficient
 50 (K_D , L/kg), the organic carbon partitioning coefficient (K_{OC} , L/kg), the dissolved organic carbon
 51 concentration (DOC, kg/L), and the solids to water ratio (r , unitless), which calculated as
 52 follows:

$$r = \frac{\rho_b(1-\varphi)}{\varphi} \quad (2)$$

53 where ρ_b is the bulk density (kg/L) and φ is the soil porosity (unitless). The steroid fraction
 54 sorbed to soil (F_S) and organic carbon (F_C) can be described as follows:

$$F_S = \frac{C_S M_S}{C_w V_w + C_S M_S + C_C M_C} = \frac{K_D r}{1 + K_D r + K_{OC} DOC} \quad (3)$$

$$F_C = \frac{C_C M_C}{C_w V_w + C_S M_S + C_C M_C} = \frac{K_{OC} DOC}{1 + K_D r + K_{OC} DOC} \quad (4)$$

55 In addition to the equilibrium mass fraction sorbed, we estimated the actual mass sorbed.
 56 Using a simplified mass balance, the mass of 17 α -TBOH that partitions to the soil within the
 57 subsurface can be described mathematically as the product of the flow, concentration, and time
 58 as follows:

$$\int_0^t (Q C_o) dt - \int_0^t (Q C) dt = M_s \quad (5)$$

59 where Q is the flow (L/min) entering and leaving the system, C_o and C are the influent (tank) and
 60 discharge concentrations (ng/L), respectively, and M_s is the mass of steroid sorbed to soil (ng).

61 Equation 5 can be simplified to the following:

$$Q C_o \int_0^t \left(1 - \frac{C}{C_o}\right) dt = M_s \quad (6)$$

62 Within the trench, the relative discharge concentration (C/C_o , unitless) of 17 α -TBOH varied with
 63 time and increased linearly throughout the duration of the experiment. Fitting a linear regression

69 model to the breakthrough data for 17 α -TBOH (Figure 2b, Results and Discussion, Subsurface
70 Transport), C/C_o can be described as follows:

71
$$\frac{C}{C_o} = 0.0043t + 0.31, R^2 = 0.98 \quad (7)$$

72 Using equation 7, complete breakthrough (i.e., C/C_o = 1) is expected to occur at 160 minutes. In
73 other words, the 17 α -TBOH sorption capacity of the soil is expected to be exhausted at 160
74 minutes under experimental conditions. Substituting equation 7 into equation 6 and integrating
75 yields the following:

76
$$M_s = QC_o(-0.0022t^2 + 0.69t) \quad (8)$$

77 The equilibrium partitioning coefficient (K_D, L/kg) for 17 α -TBOH and soil can be estimated
78 from equation 8 as follows:

79
$$K_D = \frac{C_s}{C_o} = \frac{QC_o(-0.0022t^2 + 0.69t)}{V\rho_b C_o} \quad (9)$$

80 Where C_s is concentration of 17 α -TBOH on soil (ng/kg), C_o is the leachate concentration (36
81 ng/L), V is the volume of soil in contact with water (L), t is time (evaluated at 160 minutes), and
82 ρ_b is the bulk density of the soil (1.5 kg/L). In equation 9, the only unknown variables are Q and
83 V, and depending on the control volume used, different values can be used for both parameters to
84 the estimate the value of K_D for the experimental system.

85 We applied 3 different scenarios to equation 9. In scenario A, we assumed that only the
86 water that discharged into the tray (width = 100 cm) within the AB horizon (Q = 0.63 L/min)
87 passed through the upslope soil column (width = 100 cm, length = 400, depth = 30 cm; V =
88 1,200 L). The remaining water (14.37 L/m) immediately exited the soil column. While the soil
89 volume was well defined, the latter assumption likely underestimates the flow through the soil
90 column resulting in a low estimate of K_D. Within the soil trench, the subsurface wetting-front
91 spread at least 8.5 m laterally over the 4 m travel distance from the point of application upslope

92 to the point of discharge. Assuming that the flow was approximately uniform across this 8.5 m
93 distance, the estimated discharge of the wetting front within the AB horizon is ~5.4 L/min. For
94 scenario B, we assumed that only this portion of the flow ($Q = 5.4$ L/min) passed through the soil
95 column (wetted area = 19.5 m², depth = 0.3 m; $V = 5,900$ L), and the remaining water (9.6 L/m)
96 immediately exited the soil column. While the estimated wetted soil volume is reasonable, the
97 latter assumption likely underestimates the flow through the soil column resulting in a low
98 estimate of K_D , albeit to a lesser extent than in scenario A. Finally, in scenario C, we assumed
99 that all of the applied water ($Q = 15$ L/min) passed through the observed wetted zone ($V = 5,900$
100 L). Because some flow is expected to be lost through infiltration, this assumption likely over
101 estimates the flow through the soil and therefore, K_D . Evaluating equation 9 at $t = 160$ minutes
102 estimates the mass of 17 α -TBOH that can be sorbed to soil once complete breakthrough occurs
103 under the experimental conditions.

104 For scenarios A, B, and C, the estimated K_D values for 17 α -TBOH are 0.02, 0.03, and
105 0.09 L/kg (Table 1 of main text). Based on soil characteristics, the calculated K_D is 10 L/kg
106 (Table 2 of main text) suggesting that the sorption capacity of the soil during the experiment was
107 <1% of the equilibrium partitioning capacity of the soil (based on the ratio of the system K_D
108 [calculated above] and the K_D derived from soil characteristics). This indicates that non-
109 equilibrium processes dominate the transport of 17 α -TBOH in the subsurface over short time
110 scales. Although several assumptions were made during this analysis, each was reasonable, and
111 despite the fact that values used for flow and soil mass varied by factors of 20 and 5,
112 respectively, the partitioning estimates varied by a factor of <5. Even with this variability, the
113 conclusion from each estimate is the same: far less mass partitioned to soil than was predicted by
114 equilibrium partitioning.

115 A similar approach can be taken for the ammonia, orthophosphate, and DOC. The linear
116 increase in concentration was modeled through time as follows:

117 $\frac{c}{c_o} (NH_3) = 0.0069t + 0.39, R^2 = 0.89 \quad (10)$

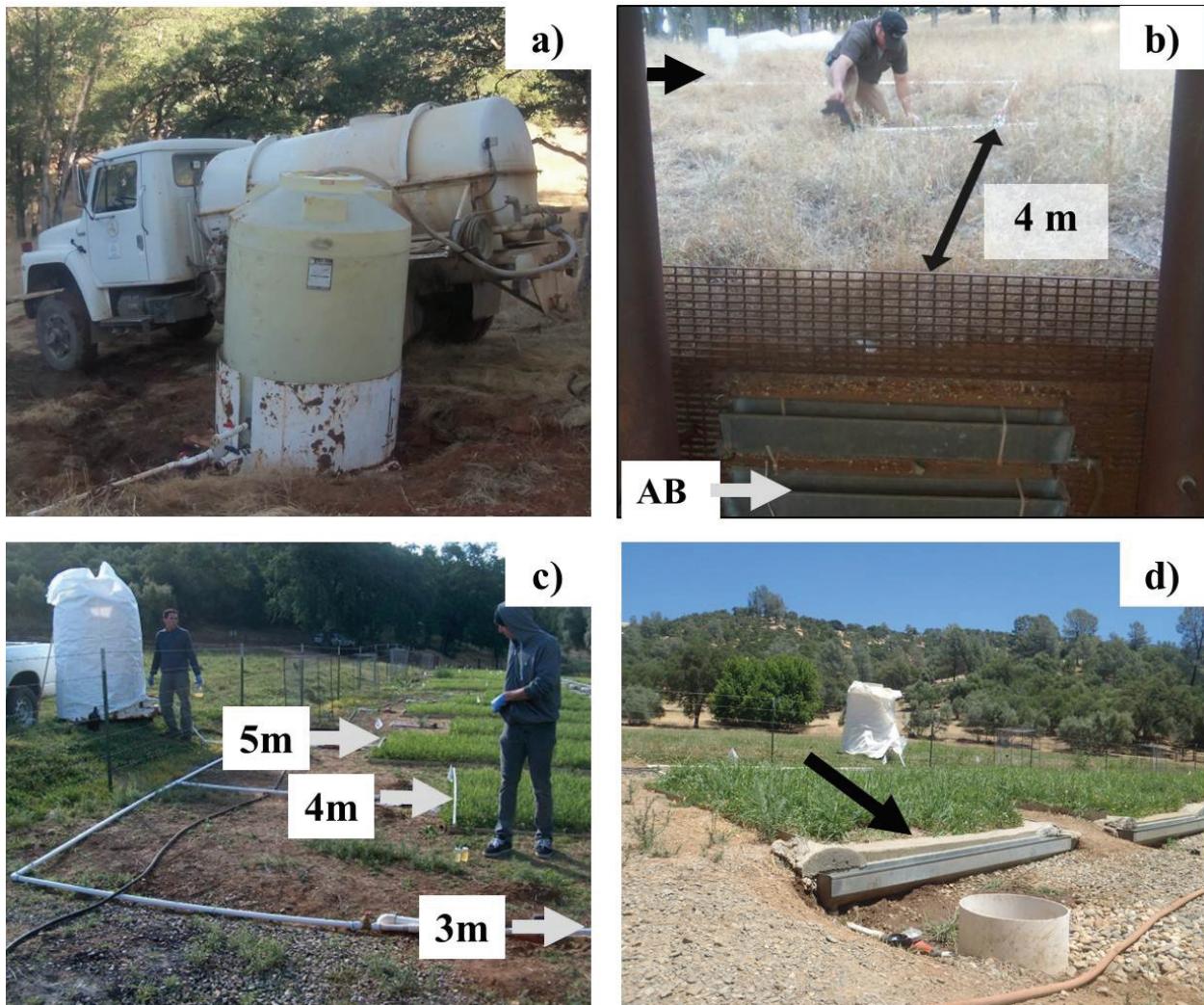
118 $\frac{c}{c_o} (PO_4) = 0.0072t + 0.50, R^2 = 0.91 \quad (11)$

119 $\frac{c}{c_o} (DOC) = 0.0035t + 0.89, R^2 = 0.99 \quad (12)$

120 Based on equations 10-12, the time to complete breakthrough of ammonia, orthophosphate, and
121 DOC is ~90, 70, and 30 minutes, respectively. By substituting these equations into equation 6
122 and integrating (as was done for 17 α -TBOH), the estimated increase in soil concentration of each
123 constituent once the sorption capacity is exhausted is 0.07-0.32 mg NH₃-N/kg, 0.02-0.07 mg
124 PO₄-P/kg, and 0.04-0.18 mg C/kg.

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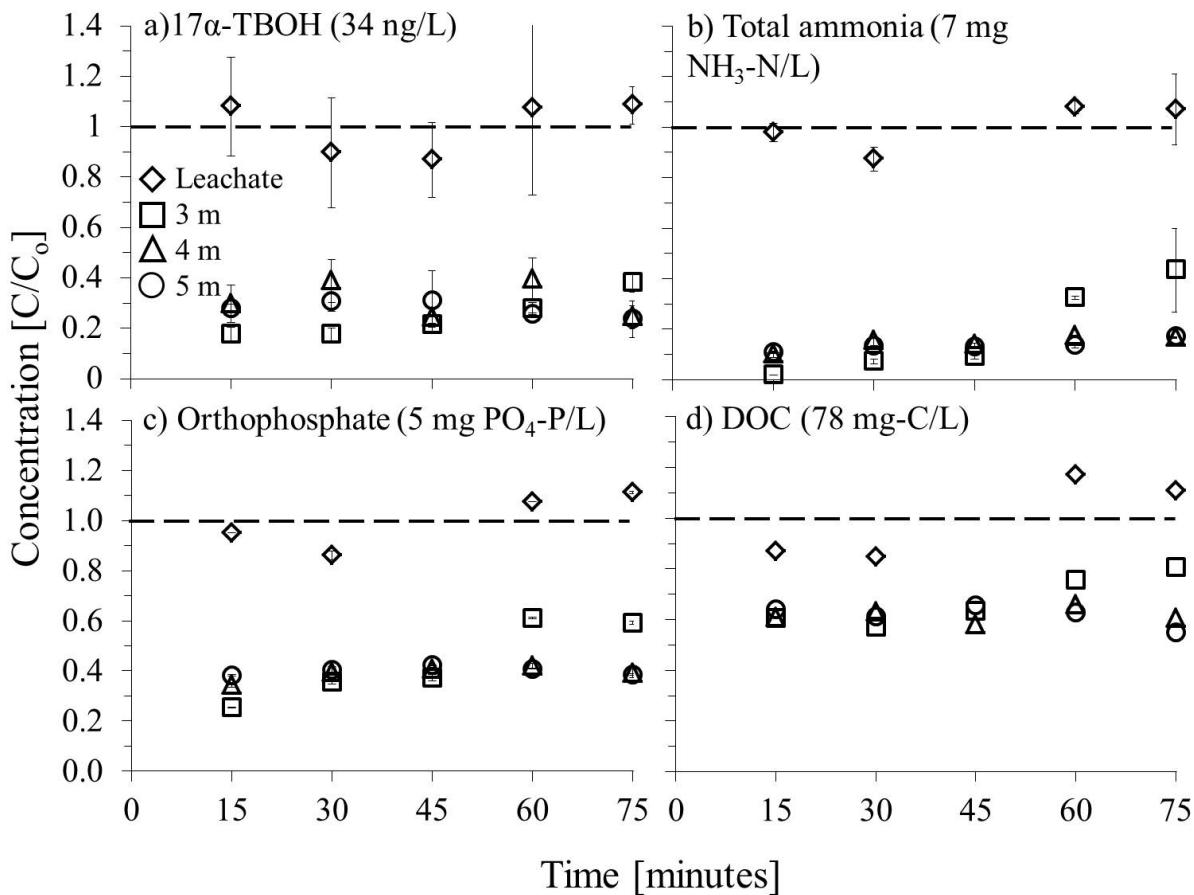
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128 **Figure S1.** Leachate was made onsite during (a, b) subsurface and (c, d) surface transport
129 experiments. Leachate was piped to (b) 4 m upslope the soil trench and to (c) 3, 4, and 5 m
130 vegetative filter strips (#s 1, 2, and 3, respectively). Subsurface discharge was collected from the
131 (b) AB soil horizon (30 cm below the surface) while surface runoff was collected from the (d)
132 outfall of the vegetative filter strips, which was constructed of concrete and aluminum trays.

133



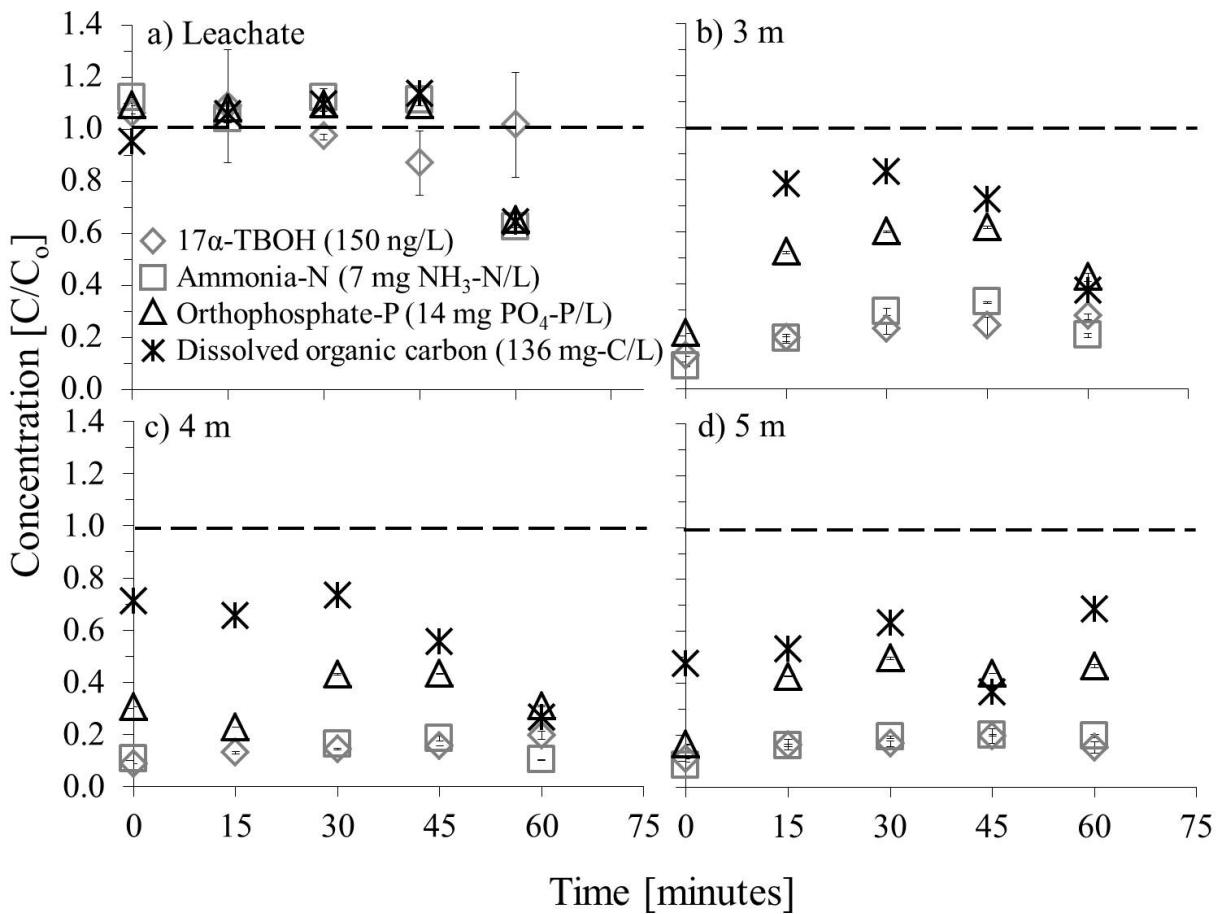
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135 **Figure S2.** Relative concentrations of (a) 17 α -TBOH, (b) total ammonia, (c) orthophosphate, and
 136 (d) dissolved organic carbon (DOC) in leachate and runoff from 3, 4, and 5 m vegetative filter
 137 strips. All concentrations were normalized to the average leachate concentration, which is
 138 provided in each panel. Data were collected on July 3, 2012. Error bars represent 95%
 139 confidence intervals ($n = 3$) and are not present for DOC.

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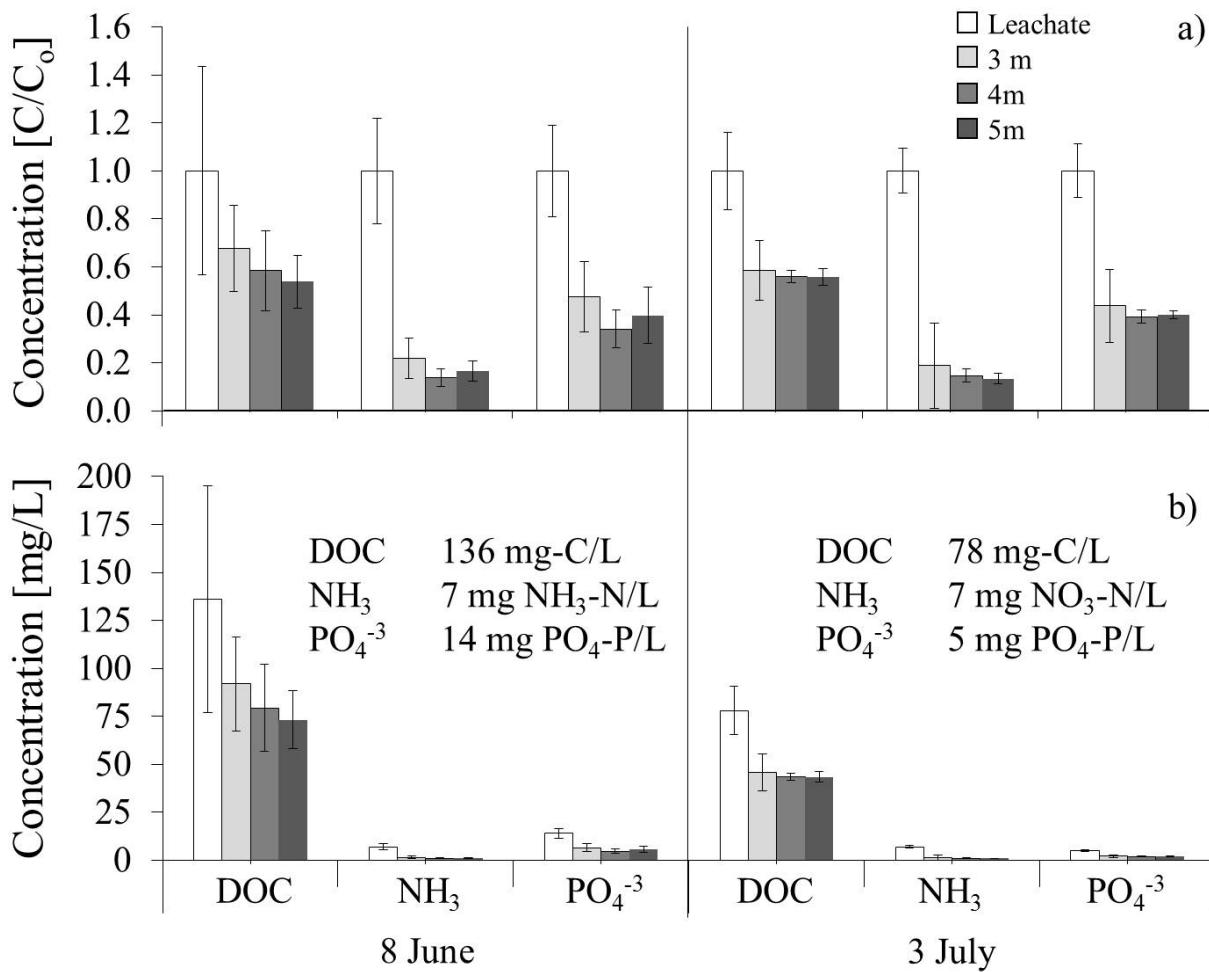
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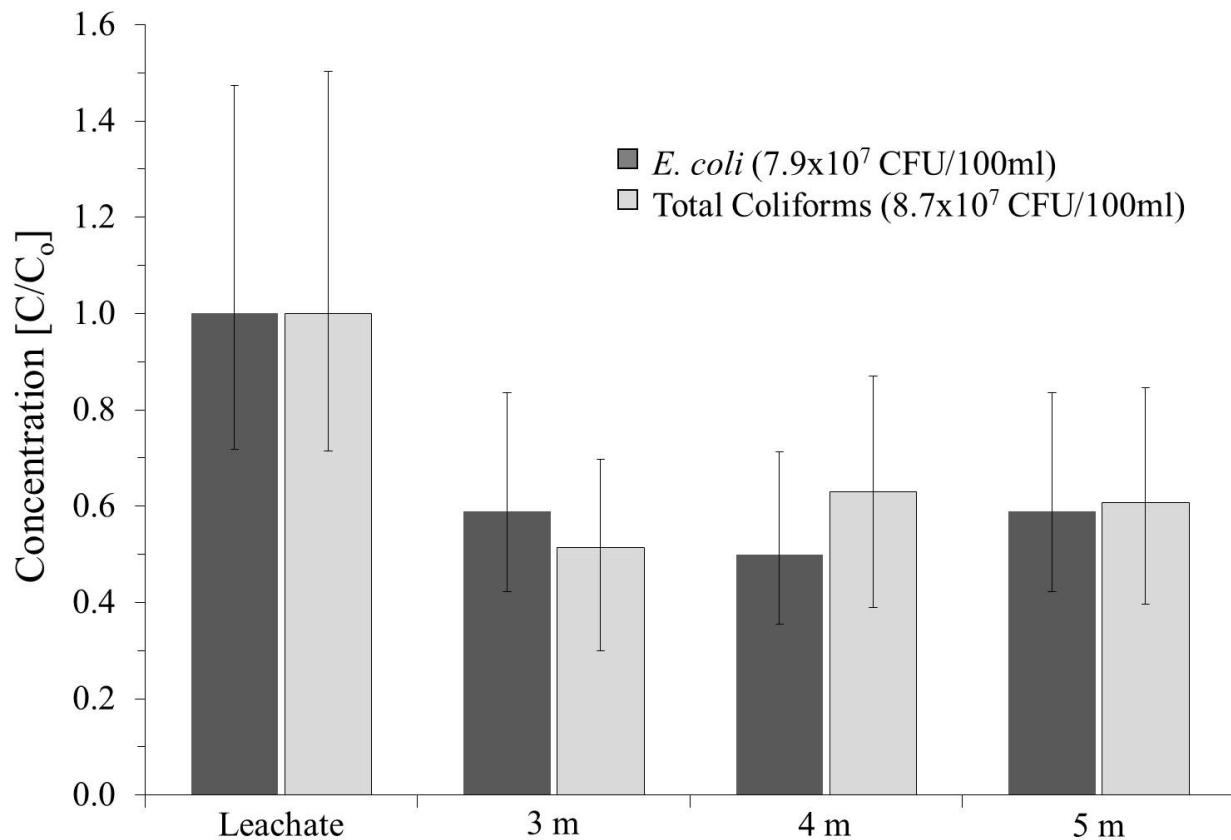


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144 **Figure S3.** Relative concentrations of manure derived contaminants in (a) leachate, and surface
145 runoff from (b) 3 m, (c) 4 m, and (d) 5 m vegetative filter strips. All concentrations were
146 normalized to the average leachate concentration, which is provided for each constituent in the
147 legend. Values reported at $t = 0$ represent first flush concentrations. Data was collected on June
148 8, 2012. Error bars represent 95% confidence intervals ($n = 3$) and are not present for DOC.
149



150
151 **Figure S4.** Average (a) normalized runoff concentrations (i.e., C/C_o) and (b) actual runoff
152 concentrations of dissolved organic carbon (DOC), total ammonia (NH₃), and orthophosphate
153 (PO₄⁻³) during two vegetative filter strip (VFS) runoff experiments conducted in June 8 and 3
154 July, 2012. Relative concentrations were normalized based on the average contaminant
155 concentration within leachate (see legend). Error bars represent 95% confidence intervals (n = 6-
156 30).
157
158



161 **Figure S5.** Concentrations of total coliforms and *E. coli* in leachate and runoff from 3, 4,
 162 and 5 m vegetative filter strips. All concentrations were normalized to the average leachate
 163 concentration provided in legend. Data was collected on July 3, 2012. Large error bars are due
 164 to inherent uncertainties within the analytical method. Error bars represent 95% confidence
 165 intervals.