

Supplementary data

Atomic coordinates for PET monomers as well as topology and parameter files needed to perform MD simulations of PET polymers are provided. Topology files can be used to generate PET polymers with the plugin *psfgen* available in the VMD program (1). Parameter files comply with the functional form of CHARMM force field (2).

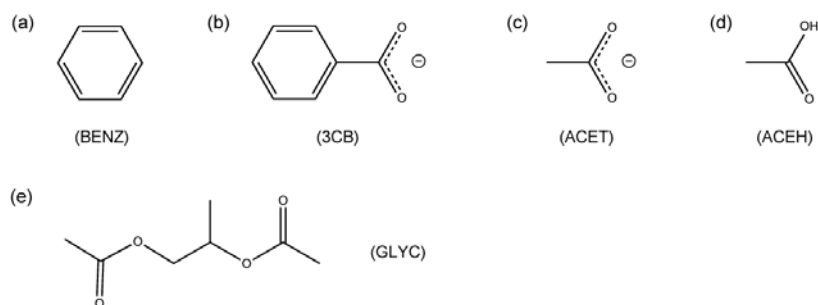
(1) W. Humphrey *et al.*, *J. Mol. Graphics*, 1996, **14**, 3338.

(2) A. MacKerell Jr. *et al.*, *J. Phys. Chem. B*, 1998, **102**, 35863616.

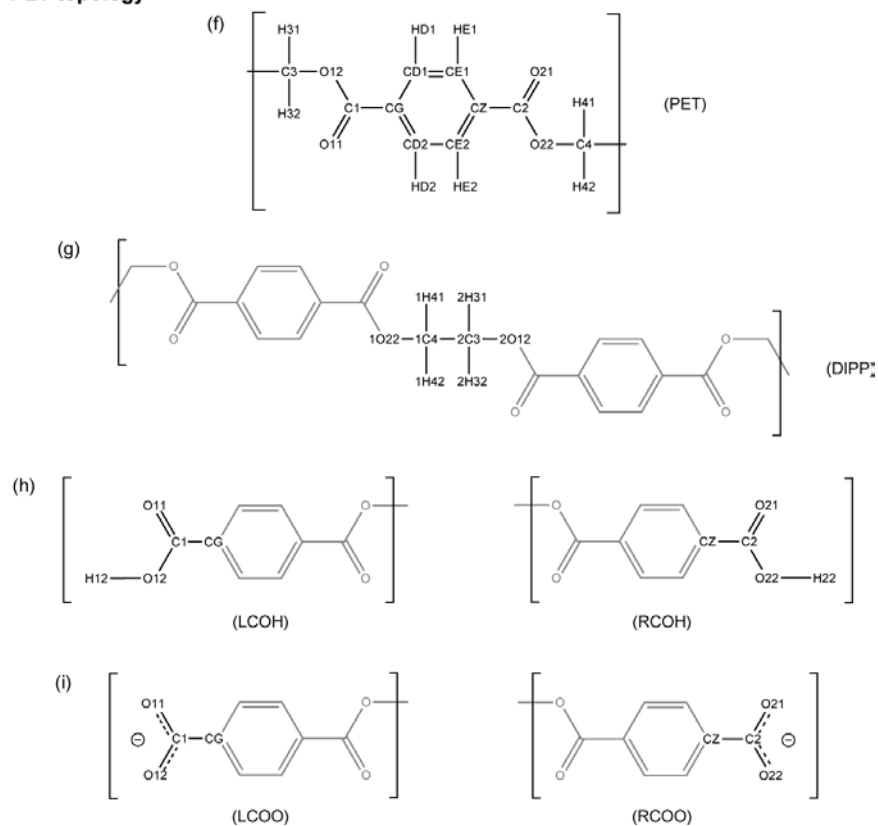
PET atomic model

Figure S1. The figure shows the CHARMM compound models (a) benzene, (b) deprotonated benzoic acid, (c) deprotonated acetic acid, (d) protonated acetic acid, and (e) propane-1,2-diyl diacetate. These compound models were used to build models for (f) PET monomer, (g) juncture between PET monomers, (h) protonated terminal residues, and (i) deprotonated terminal residues. CHARMM abbreviations are shown in parenthesis.

Compound models from CHARMM force field



PET topology



Atomic coordinates for PET monomer

REMARK	Coordinates for PET monomer in TRANS conformation											
REMARK	For CIS isomer, change the following coordinates											
REMARK	O12				27.653	-4.733	0.000					
REMARK	O11				30.840	-4.733	0.000					
REMARK	C3				27.653	-2.893	0.000					
REMARK	H31				28.223	-2.564	0.931					
REMARK	H32				28.107	-2.564	-1.012					
ATOM	1	CD1	PET	T	1	30.840	-8.412	0.000	1.00	0.00	TR	C
ATOM	2	HD1	PET	T	1	31.794	-7.862	0.000	1.00	0.00	TR	H
ATOM	3	CE1	PET	T	1	30.840	-10.252	0.000	1.00	0.00	TR	C
ATOM	4	HE1	PET	T	1	31.794	-10.803	0.000	1.00	0.00	TR	H
ATOM	5	CD2	PET	T	1	27.653	-8.412	0.000	1.00	0.00	TR	C
ATOM	6	HD2	PET	T	1	26.699	-7.862	0.000	1.00	0.00	TR	H
ATOM	7	CE2	PET	T	1	27.653	-10.252	0.000	1.00	0.00	TR	C
ATOM	8	HE2	PET	T	1	26.699	-10.803	0.000	1.00	0.00	TR	H
ATOM	9	CG	PET	T	1	29.246	-7.492	0.000	1.00	0.00	TR	C
ATOM	10	C1	PET	T	1	29.246	-5.653	0.000	1.00	0.00	TR	C
ATOM	11	O12	PET	T	1	30.840	-4.733	0.000	1.00	0.00	TR	O
ATOM	12	O11	PET	T	1	27.653	-4.733	0.000	1.00	0.00	TR	O
ATOM	13	C3	PET	T	1	30.840	-2.893	0.000	1.00	0.00	TR	C
ATOM	14	H31	PET	T	1	31.700	-2.564	0.931	1.00	0.00	TR	H
ATOM	15	H32	PET	T	1	31.700	-2.564	-1.012	1.00	0.00	TR	H
ATOM	16	CZ	PET	T	1	29.246	-11.172	0.000	1.00	0.00	TR	C
ATOM	17	C2	PET	T	1	29.246	-13.012	0.000	1.00	0.00	TR	C
ATOM	18	O21	PET	T	1	30.840	-13.932	0.000	1.00	0.00	TR	O
ATOM	19	O22	PET	T	1	27.653	-13.932	0.000	1.00	0.00	TR	O
ATOM	20	C4	PET	T	1	27.653	-15.772	0.000	1.00	0.00	TR	C
ATOM	21	H41	PET	T	1	27.153	-16.101	0.931	1.00	0.00	TR	H
ATOM	22	H42	PET	T	1	27.153	-16.034	-1.012	1.00	0.00	TR	H

PET topology

* <CHARMM22 topology file for PET>

27 1

MASS	1	H	1.008	H	! nonpolar H
MASS	2	HAL2	1.008	H	! nonpolar H
MASS	3	HP	1.008	H	! aromatic H
MASS	4	CAP	12.011	C	! aromatic C
MASS	5	CC	12.011	C	! carbonyl C, asn,asp,gln,glu,cter,ct2
MASS	6	CD	12.011	C	! carbonyl C, pres aspp,glup,ct1
MASS	7	CL	12.011	C	! carbonyl C (acetic acid/methyl acetate)
MASS	8	CTL2	12.011	C	! aliphatic sp3 C for CH2
MASS	9	OC	15.999	O	! carboxylate oxygen
MASS	10	OB	15.999	O	! carbonyl oxygen in acetic acid
MASS	11	OBL	15.999	O	! acetic acid carboxyl oxygen (e. to protein OB)
MASS	12	OH1	15.999	O	! hydroxyl oxygen
MASS	13	OSL	15.999	O	! ester oxygen

AUTO ANGLES DIHE

RESI PET		0.0	! topology for PET			
GROUP						
ATOM	CD1	CAP	-0.115			
ATOM	HD1	HP	0.115			
GROUP						
ATOM	CE1	CAP	-0.115			
ATOM	HE1	HP	0.115			
GROUP						
ATOM	CD2	CAP	-0.115			
ATOM	HD2	HP	0.115			
GROUP						
ATOM	CE2	CAP	-0.115			
ATOM	HE2	HP	0.115			
GROUP						
ATOM	CG	CAP	0.10			
ATOM	C1	CL	0.63			
ATOM	O11	OBL	-0.52			
ATOM	O12	OSL	-0.34			
ATOM	C3	CTL2	-0.05			
ATOM	H31	HAL2	0.09			
ATOM	H32	HAL2	0.09			
GROUP						
ATOM	CZ	CAP	0.10			
ATOM	C2	CL	0.63			
ATOM	O21	OBL	-0.52			
ATOM	O22	OSL	-0.34			
ATOM	C4	CTL2	-0.05			
ATOM	H41	HAL2	0.09			
ATOM	H42	HAL2	0.09			
BOND	C3	H31	C3	H32	C3	O12
BOND	O12	C1	C1	CG		
BOND	C4 H41	C4	H42	C4	O22	
BOND	O22	C2	C2	CZ		
BOND	CD1	HD1	CD2	HD2		
BOND	CE1	HE1	CE2	HE2		
BOND	CD1	CE1	CG	CD2	CE2	CZ
DOUBLE	O11	C1	C2	O21		
DOUBLE	CG	CD1	CE1	CZ	CD2	CE2
IMPR	C1	CG	O12	O11		
IMPR	C2	CZ	O22	O21		

PRES DIPP 0.0 ! patch for two PET molecules. Patch must be 1-PET and 2-PET
 ! use in a patch statement
 ! follow with AUTOgenerate ANGLEs DIHEdrals command

BOND 1C4 2C3
 ANGLE 1H41 1C4 2C3 1H42 1C4 2C3
 ANGLE 2H31 2C3 1C4 2H32 2C3 1C4
 ANGLE 2O12 2C3 1C4 1O22 1C4 2C3
 DIHE 1H41 1C4 2C3 2H31
 DIHE 1H41 1C4 2C3 2H32
 DIHE 1H41 1C4 2C3 2O12
 DIHE 1O22 1C4 2C3 2H31
 DIHE 1O22 1C4 2C3 2H32
 DIHE 1O22 1C4 2C3 2O12
 DIHE 1H42 1C4 2C3 2H31
 DIHE 1H42 1C4 2C3 2H32
 DIHE 1H42 1C4 2C3 2O12

PRES LCOH 0.0 ! protonated carboxyl group

GROUP
 ATOM C1 CD 0.75
 ATOM O11 OB -0.55
 ATOM O12 OH1 -0.61
 ATOM CG CAP -0.03
 ATOM H12 H 0.44
 DELETE ATOM H31
 DELETE ATOM H32
 DELETE ATOM C3
 BOND O12 H12
 ANGLE H12 O12 C1
 DIHE H12 O12 C1 O11
 DIHE H12 O12 C1 CG
 IC H12 O12 C1 O11 0.0000 0.0000 0.0000 0.0000 0.0000

PRES RCOH 0.0 ! protonated carboxyl group

GROUP
 ATOM C2 CD 0.75
 ATOM O21 OB -0.55
 ATOM O22 OH1 -0.61
 ATOM CZ CAP -0.03
 ATOM H22 H 0.44
 DELETE ATOM H41
 DELETE ATOM H42
 DELETE ATOM C4
 BOND O22 H22
 ANGLE H22 O22 C2
 DIHE H22 O22 C2 O21
 DIHE H22 O22 C2 CZ
 IC H22 O22 C2 O21 0.0000 0.0000 0.0000 0.0000 0.0000

PRES LCOO		-1.0	! deprotonated carboxyl group
GROUP			
ATOM	C1	CC	0.62
ATOM	O11	OC	-0.76
ATOM	O12	OC	-0.76
ATOM	CG	CAP	-0.10
DELETE	ATOM	H31	
DELETE	ATOM	H32	
DELETE	ATOM	C3	

PRES RCOO		-1.0	! deprotonated carboxyl group
GROUP			
ATOM	C2	CC	0.62
ATOM	O21	OC	-0.76
ATOM	O22	OC	-0.76
ATOM	CZ	CAP	-0.10
DELETE	ATOM	H41	
DELETE	ATOM	H42	
DELETE	ATOM	C4	

PET parameters

BONDS

$$!V(\text{bond}) = Kb(b - b0)**2$$

$$!Kb: \text{kcal/mole/\AA}**2$$

$$!b0: \text{\AA}$$

atom types		Kb	b0	
CAP	CAP	370.000	1.3600	! from 3CB
CAP	HP	340.000	1.0800	! from 3CB
CAP	CC	200.000	1.5000	! from 3CB
CAP	CD	200.000	1.5000	! from 3CB
CAP	CL	200.000	1.5000	! from 3CB
CC	OC	525.000	1.2600	! from ACET
CD	OB	750.000	1.2200	! from ACEH
CD	OH1	230.000	1.4000	! from ACEH
CL	OBL	750.000	1.2200	! from GLYC
CL	OSL	150.000	1.3340	! from GLYC
CTL2	CTL2	222.500	1.5300	! from GLYC
CTL2	HAL2	309.000	1.1110	! from GLYC
CTL2	OSL	340.000	1.4300	! from GLYC
H	OH1	545.000	0.9600	! from ACEH

ANGLES

!V(angle) = Ktheta(Theta - Theta0)**2

!Ktheta: kcal/mole/rad**2

!Theta0: degrees

!Kub: kcal/mole/A**2 (Urey-Bradley)

!S0: A

!atom types			Ktheta	Theta0	Kub	S0	
CAP	CAP	CAP	90.00	122.00	30.00	2.38800	! from 3CB
CAP	CAP	HP	30.00	119.00	22.00	2.15250	! from 3CB
CAP	CAP	CC	40.00	119.00	35.00	2.41620	! from 3CB
CAP	CAP	CD	40.00	119.00	35.00	2.41620	! from 3CB
CAP	CAP	CL	40.00	119.00	35.00	2.41620	! from 3CB
CAP	CC	OC	40.00	118.00	50.00	2.38800	! from 3CB
CAP	CD	OB	70.000	125.00	20.00	2.44200	! from ACEH
CAP	CD	OH1	55.000	110.50			! from ACEH
CAP	CL	OBL	70.000	125.00	20.00	2.44200	! from GLYC
CAP	CL	OSL	55.000	109.00	20.00	2.32600	! from GLYC
CD	OH1	H	55.000	115.00			! from ACEH
CL	OSL	CTL2	40.000	109.60	30.00	2.26510	! from GLYC
CTL2	CTL2	HAL2	26.500	110.10	22.53	2.17900	! from GLYC
CTL2	CTL2	OSL	75.700	110.10			! from GLYC
HAL2	CTL2	HAL2	35.500	109.00	5.40	1.80200	! from GLYC
HAL2	CTL2	OSL	60.000	109.50			! from GLYC
OBL	CL	OSL	90.000	125.90	160.00	2.25760	! from GLYC
OB	CD	OH1	50.000	123.00	210.00	2.26200	! from ACEH
OC	CC	OC	100.000	124.00	70.00	2.22500	! from ACET

DIHEDRALS

$$!V(\text{dihedral}) = K\text{chi}(1 + \cos(n(\text{chi}) - \text{delta}))$$

!Kchi: kcal/mole

!n: multiplicity

!delta: degrees

!atom types

				Kchi	n	delta	
CAP	CAP	CAP	CAP	1.20	2	180.00	! from 3CB
CC	CAP	CAP	CAP	3.10	2	180.00	! from 3CB
CD	CAP	CAP	CAP	3.10	2	180.00	! from 3CB
CL	CAP	CAP	CAP	3.10	2	180.00	! from 3CB
OC	CC	CAP	CAP	3.10	2	180.00	! from 3CB
OSL	CL	CAP	CAP	3.10	2	180.00	! from 3CB
OBL	CL	CAP	CAP	3.10	2	180.00	! from 3CB
OH1	CD	CAP	CAP	3.10	2	180.00	! from 3CB
OB	CD	CAP	CAP	3.10	2	180.00	! from 3CB
HP	CAP	CAP	CAP	4.20	2	180.00	! from 3CB
HP	CAP	CAP	CC	4.20	2	180.00	! from 3CB
HP	CAP	CAP	CD	4.20	2	180.00	! from 3CB
HP	CAP	CAP	CL	4.20	2	180.00	! from 3CB
HP	CAP	CAP	HP	1.90	2	180.00	! from 3CB
OBL	CL	OSL	CTL2	3.85	2	180.00	! from GLYC
X	CD	OH1	X	2.05	2	180.00	! from ACEH
X	CL	OSL	X	2.05	2	180.00	! from GLYC
CTL2	CTL2	OSL	CL	0.07	1	180.00	! from GLYC
X	CTL2	OSL	X	0.00	3	0.00	! from GLYC
X	CTL2	CTL2	X	0.20	3	0.00	! from GLYC

IMPROPERS

$$!V(\text{improper}) = K\text{psi}(\text{psi} - \text{psi0})^{**2}$$

!Kpsi: kcal/mole/rad**2

!psi0: degrees

!note that the second column of numbers (0) is ignored

!atom types

				Kpsi	psi0	
OC	X	X	CC	96.00	0	0.00 ! from ACET
OBL	X	X	CL	100.00	0	0.00 ! from GLYC
OB	X	X	CD	100.00	0	0.00 ! from ACEH

NONBONDED

!V(Lennard-Jones) = Eps,i,j[(Rmin,i,j/ri,j)**12 - 2(Rmin,i,j/ri,j)**6]

!epsilon: kcal/mole, Eps,i,j = sqrt(eps,i * eps,j)

!Rmin/2: A, Rmin,i,j = Rmin/2,i + Rmin/2,j

!atom ignored epsilon Rmin/2 ignored eps,1-4 Rmin/2,1-4

H	0.0000	-0.0460	0.2245			
HAL2	0.0000	-0.0280	1.3400			
HP	0.0000	-0.0300	1.3582	0.0000	-0.0300	1.3582
CAP	0.0000	-0.0700	1.9924			
CC	0.0000	-0.0700	2.0000			
CD	0.0000	-0.0700	2.0000			
CL	0.0000	-0.0700	2.0000			
CTL2	0.0000	-0.0560	2.0100	0.0000	-0.0100	1.9000
OC	0.0000	-0.1200	1.7000			
OB	0.0000	-0.1200	1.7000	0.0000	-0.1200	1.4000
OBL	0.0000	-0.1200	1.7000	0.0000	-0.1200	1.4000
OH1	0.0000	-0.1521	1.7700			
OSL	0.0000	-0.1521	1.7700			
