

Electronic Supplementary Information

Formation of a long-lived electron-transfer state of a naphthalene-quinolinium ion dyad and the π -dimer radical cation

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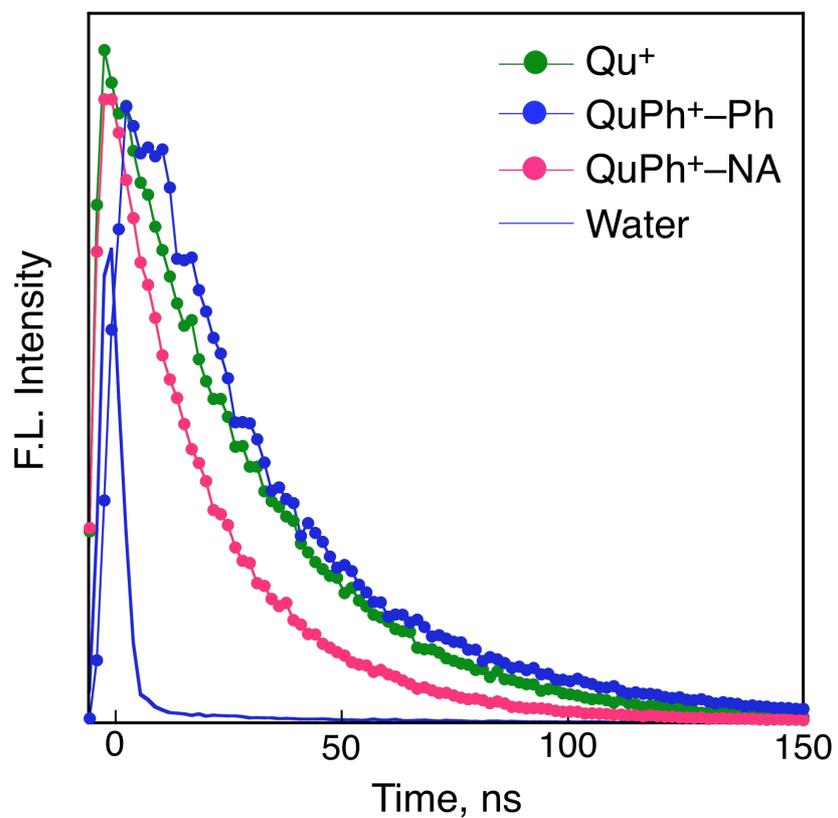


Fig. S1 Fluorescence decay time profiles for Qu^+ (1.3×10^{-4} M) observed at 396 nm ($\lambda_{\text{ex}} = 330$ nm), $\text{QuPh}^+\text{-Ph}$ (1.3×10^{-4} M) observed at 430 nm ($\lambda_{\text{ex}} = 330$ nm), and $\text{QuPh}^+\text{-NA}$ (1.3×10^{-4} M) observed at 545 nm ($\lambda_{\text{ex}} = 330$ nm).

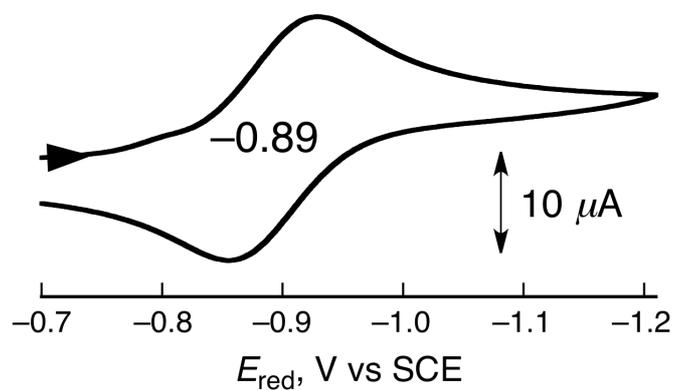


Fig. S2 Cyclic voltammogram of QuPh⁺-Ph ($1.0 \times 10^{-3} \text{ M}$) in deaerated MeCN containing TBAP (0.1 M) at 298 K; sweep rate (100 mV s^{-1}).

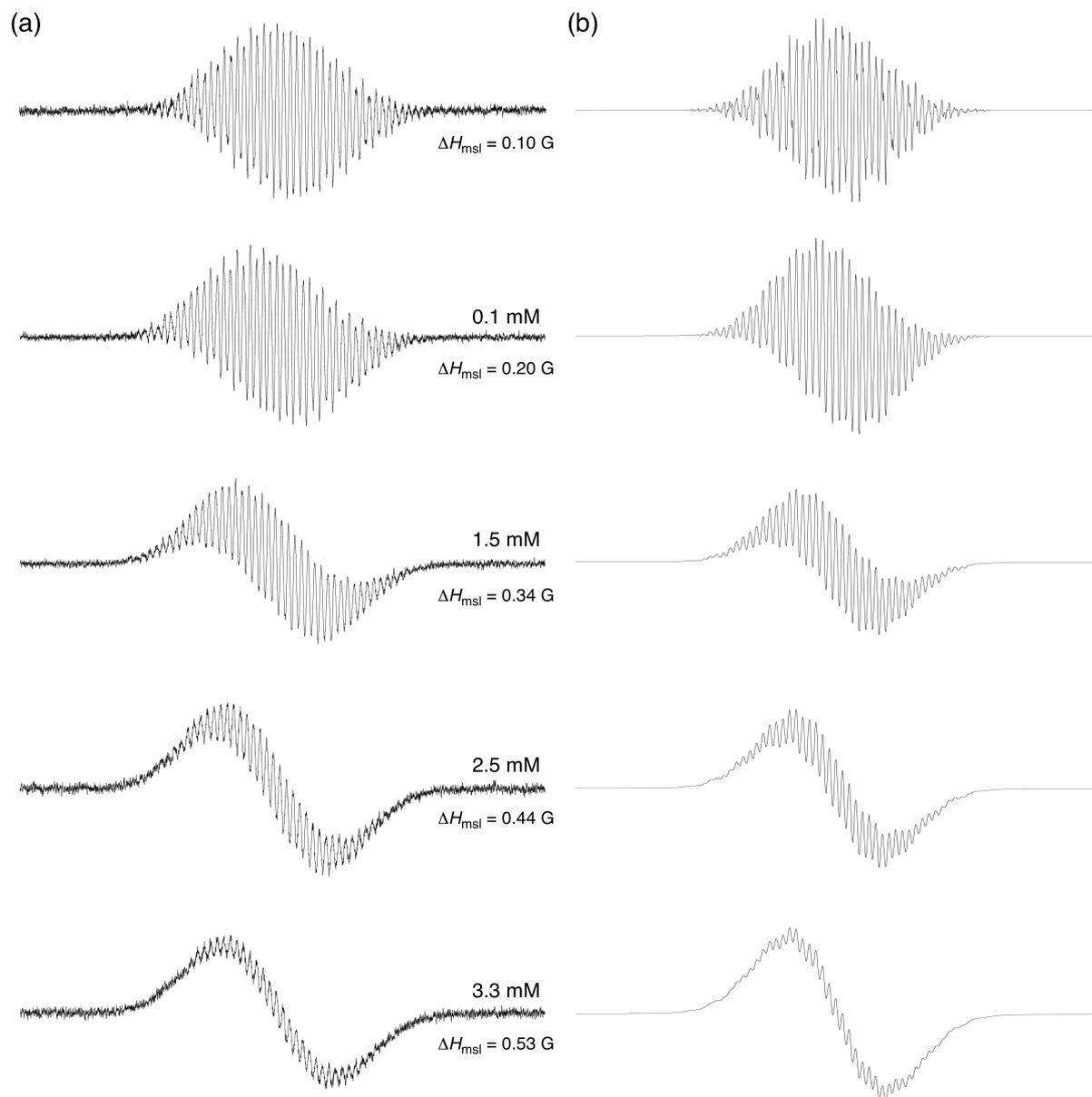


Fig. S3 (a) ESR spectrum of QuPh•-NA produced by the photochemical reduction of QuPh⁺-NA ($1.3 \times 10^{-4} - 3.3 \times 10^{-3}$ M) with (BNA)₂ (7.0×10^{-5} M) in deaerated MeCN at 298 K and (b) the computer simulation spectrum.

Table S1 | X-ray Crystallographic data for QuPh⁺-NA

Empirical formula	C ₂₆ H ₂₀ ClNO ₄
Formula weight	445.88
Crystal system, space group	Orthorhombic, <i>P</i> 2 ₁ 2 ₁ 2 ₁
Temperature / K	123
Wavelength / Å	0.71070 Å
Unit cell dimensions	$\alpha = 9.2173(5) \text{ \AA}, \alpha = 90^\circ$ $\beta = 9.9916(8) \text{ \AA}, \beta = 90^\circ$ $\gamma = 22.5665(17) \text{ \AA}, \gamma = 90^\circ$
Volume / Å ³	2078.3(3)
<i>Z</i>	4
Reflections collected / unique	7640 / 4422
GoF on <i>F</i> ²	1.109
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0590
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.1852

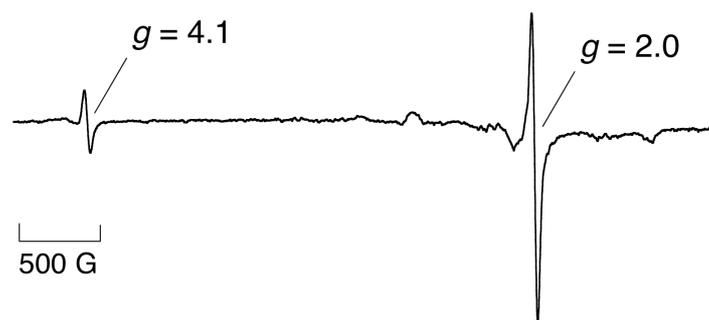


Fig. S4 ESR spectrum of $\text{QuPh}^{\bullet-}\text{NA}^+$ produced by the photoirradiation (high-pressure mercury lamp) of QuPh^+NA (1.5×10^{-3} M) in deaerated MeCN at 77 K.

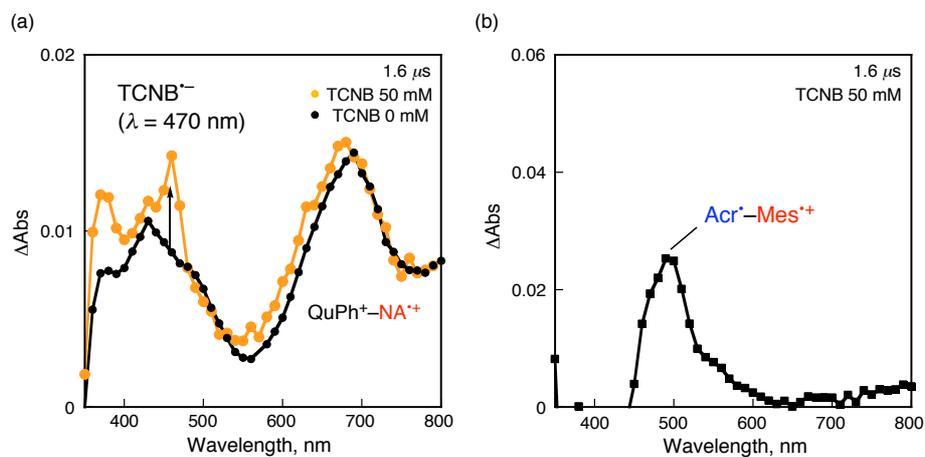


Fig. S5 Transient absorption spectra of (a) QuPh⁺-NA (1.0 × 10⁻⁴ M) and (b) Acr⁺-Mes (1.0 × 10⁻⁴ M) with TCNB (5.0 × 10⁻² M) in deaerated MeCN at 298 K taken at 1.6 μs after laser excitation at 355 nm.