"A low cost synthesis method for functionalised iron oxide nanoparticles for magnetic hyperthermia from readily available materials"

Joseph C. Bear^[1], Bin Yu^[1], Cristina Blanco-Andujar^[2], Paul D. McNaughter^[3], Paul Southern^[2], Marc-Krystelle Mafina^[4], Quentin A. Pankhurst^[2] and Ivan P. Parkin^[1]*

[1] Materials Chemistry Research Centre, Department of Chemistry, University College London, 20 Gordon Street, London, WC1H 0AJ, UK.

[2] UCL Healthcare Biomagnetics Laboratories, Royal Institution of Great Britain, London, W1S 4BS, UK.

[3] School of Chemistry, University of East Anglia, Norwich Research Park, Norwich, NR4 7TJ, UK.

[4] School of Engineering and Materials Science, Queen Mary, University of London, Mile End Road, London, E1 4NS, UK.

Supporting Information

Table of Contents

1.	TEM, EDX and HRTEM analysis for iron oxide nanomaterials:	.2
2.	Differential Scanning Calorimetry:	.6
3.	ATR-FTIR spectra	.7
4.	Table of costs	.8
5.	Mössbauer analysis of iron palmitate decomposed in 1-octadecene	.9
6.	XRD patterns and references	10
7.	References	12

1. TEM,EDX and HRTEM analysis for iron oxide nanomaterials:



Figure S1.1 Transmission electron microscope (TEM) image of rhenanite nanorods synthesised with oleic acid (in a molar ratio of 1:1 iron palmitate to oleic acid) in shark liver oil.



Figure S1.2 High resolution transmission electron microscope (HRTEM) image of Sample (II), displaying <311> magnetite crystal planes.



Figure S1.3 High resolution transmission electron microscope (HRTEM) image of magnetite nanoparticles synthesised according to a method adapted from Park *et al.*¹



Figure S1.4 Energy dispersive X-ray spectrum of iron oxide nanoparticles synthesised according to Park *et al.*¹



Figure S1.5 Selected area electron diffractogram (SAED) of iron oxide nanoparticles synthesised according to Park *et al.*¹



Figure S1.6 Labelled SAED of figure S2.5.



Figure S1.7 Zoomed EDX spectrum of iron oxide nanoparticles from the decomposition of iron palmitate in shark liver oil.

2. Differential Scanning Calorimetry:



Figure S2.1 Differential Scanning Calorimetry (DSC) curve of iron-oleate.



Figure S2.2 DSC curve of iron-palmitate.

3. ATR-FTIR spectra:



Figure S3.1 Fourier transform infrared spectrum of Sainsbury's Basics Soap, with the C=O stretch of sodium palmitate and -OH stretches from adsorbed water clearly visible.



Figure S3.2 Fourier transform infrared spectrum of annealed (350 °C in air) iron oxide nanoparticles synthesised *via* the decomposition of iron palmitate in shark liver oil, with the Fe-O stretch of the iron oxide visible at 565 cm⁻¹.

4. Table of costs

Reagent	Vendor	Price per gram (£ GBP)	High-street reagent	Vendor	Price per gram (£ GBP)
Iron(III) chloride hexahydrate (ACS reagent grade)	VWR International Ltd.	0.08*	Iron tablets	Boots Ltd.	0.12
Sodium oleate (≥82% fatty acid content)	Sigma-Aldrich	0.03*	"Basics soap"	J. Sainsbury Ltd.	0.0013
Oleic acid (techn. grade 90%)	Sigma-Aldrich	0.03*	Olive oil	J. Sainsbury Ltd.	0.0038
1-octadecene (techn. grade 90%)	Sigma-Aldrich	0.54*	Shark Liver oil	Shark Liver oil UK Ltd.	0.23

*The prices listed above do not include delivery, VAT and other associated costs.

For the synthesis of 1 g of nanoparticles exclusive of ligands and based on iron content of Fe_2O_3 , the cost of the iron oleate and iron palmitate syntheses was £18.66* and £15.81 respectively.

5. Mössbauer analysis of iron palmitate decomposed in 1-octadecene



Figure S5 ⁵⁷Fe Mössbauer spectra of: iron oxide nanoparticles from iron palmitate decomposed in
1-octadecene (Sample II). The red trace represents the fit with the thick black line the Fe³⁺ doublet component and the thin black line the Fe²⁺ doublet component.

6. XRD patterns and references



Figure S6.1 XRD pattern of annealed iron oxide nanoparticles prepared from the decomposition of iron palmitate in shark liver oil (Sample (III)). The <611>, <020> and <031>peaks of Rhenanite (β -NaCaPO₄) are visible between 38 and 42° (2 θ).



Figure S6.2 Standard XRD pattern of β-NaCaPO₄ (rhenanite).²



Figure S6.2 Standard XRD pattern of Fe₃O₄ (magnetite).³

7. References

- 1. J. Park, K. An, Y. Hwang, J.-G. Park, H.-J. Noh, J.-Y. Kim, J.-H. Park, N.-M. Hwang, and T. Hyeon, *Nat. Mater.*, 2004, **3**, 891–895.
- 2. ICSD no. 35629.
- 3. ICSD no. 64829.