SUPPLEMENTARY INFORMATION

Ionic liquids via efficient, solvent-free anion metathesis

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General Considerations. NMR spectra were recorded using a Varian Unity Plus 300 or 400 spectrometer. Chemical shifts are reported in delta (δ) units, expressed in parts per million (ppm) downfield from tetramethylsilane using residual protio solvent as an internal standard (CDCl₃, ¹H: 7.26 ppm, ¹³C: 77.0 ppm; CD_2Cl_2 ¹H: 5.32 ppm, ¹³C: 54.0 ppm). ³¹P NMR spectra were recorded at 121 MHz and were externally referenced to H₃PO₄. ¹³C and ³¹P NMR spectra were routinely run with broadband decoupling. CH₂Cl₂ was distilled from CaH₂ under N₂ atmosphere prior to use. 1-Butvl-3methylimidazolium ([BMIM]) halides were prepared via the reaction of 1-methylimidazole (1.0 equiv) and butyl halide (1.05 equiv) in a sealed vessel at 130 °C for 24 h, followed by removal of volatile components. Similarly, 1-(3-hydroxypropyl)-3-methylimidazolium bromide ([HPMIM]Br) was obtained from 1-methylimidazole and 3-bromo-1-propanol. 1-Butylpyridinium ([BPY]) halides were prepared in an analogous fashion from pyridine and the corresponding 1-haloalkane. All other reagents were purchased from commercial suppliers and used as received. All organic halide salts were stored in a desiccator prior to use. Iodide salts were protected from light using aluminum foil. Due to the hygroscopic nature of the salts discussed herein, reactions were routinely protected with a drying tube containing Drierite, unless otherwise noted. All reactions were conducted in a ventilated fume hood. Note: For solvent-free reactions involving solid azolium halides and trialkyloxonium salts, gradual liquefaction occurs over time. For comparative analyses, [BMIM][MeSO₄], [BPY][MeSO₄], [BMIM][OTs], [BMIM]BF₄, [BMIM]PF₆, [BPY]BF₄, trihexyl(tetradecyl)phosphonium tetrafluoroborate $([Hx_3PC_{14}]BF_4)$, and 1,3-dimesitylimidazolium tetrafluoroborate ($[IMes]BF_4$) were purchased from commercial suppliers. Caution: Dimethyl sulfate is toxic and should be handled with care.

Entry	Trapping Reagent	Organic Halide Salt	Product
1	Me ₂ SO ₄	[BMIM]CI	[BMIM][MeSO ₄]
2	Me_2SO_4	[BMIM]Br	[BMIM][MeSO ₄]
3	Me ₂ SO ₄	[BMIM]I	[BMIM][MeSO ₄]
4	Me ₂ SO ₄	[BPY]CI	[BPY][MeSO ₄]
5	Me_2SO_4	[BPY]I	[BPY][MeSO ₄]
6	Me ₂ SO ₄	[Hx ₃ PC ₁₄]Cl	[Hx ₃ PC ₁₄][MeSO ₄]
7	Me ₂ SO ₄	[Hx ₃ PC ₁₄]Br	[Hx ₃ PC ₁₄][MeSO ₄]
8	MeOTs	[BMIM]Br	[BMIM][OTs]
9	MeOTs	[BMIM]I	[BMIM][OTs]
10	MeOTs	[BPY]CI	[BPY][OTs]
11	MeOTs	[BPY]I	[BPY][OTs]
12	Me ₃ PO ₄	[BPY]CI	[BPY][Me ₂ PO ₄]
13	Me ₃ O•BF ₄	[BMIM]CI	[BMIM]BF ₄
14	Me_3O-BF_4	[BMIM]Br	[BMIM]BF ₄
15	Me_3O-BF_4	[BMIM]I	[BMIM]BF ₄
16	Et ₃ O-BF ₄	[BMIM]Br	[BMIM]BF ₄
17	Et_3O-PF_6	[BMIM]CI	[BMIM]PF ₆
18	Et_3O-BF_4	[BPY]CI	[BPY]BF ₄
19	Et_3O-BF_4	[Hx ₃ PC ₁₄]Cl	[Hx ₃ PC ₁₄]BF ₄
20	Et_3O-BF_4	[Hx ₃ PC ₁₄]Br	[Hx ₃ PC ₁₄]BF ₄
21 ^b	Et_3O-BF_4	[IMes]Cl	[IMes]BF ₄
22 ^c	Et_3O-BF_4	[BMIM]Br	[BMIM]BF ₄
23 ^b	Et_3O-BF_4	[HPMIM]Br	[HPMIM]BF ₄

Table 1. Summary of anion metathesis reactions reported in the accompanying manuscript.^a

^a Unless otherwise noted, all reactions were performed without the aid of solvent. ^b Reaction was performed in dichloromethane. ^c Reaction was performed in water.

Entry 1. A flask was charged with a magnetic stirbar and 1-butyl-3-methylimidazolium chloride (521 mg, 2.98 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.28 mL, 2.98 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 747 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 2. A flask was charged with a magnetic stirbar and 1-butyl-3-methylimidazolium bromide (639 mg, 2.90 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.28 mL, 2.90 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 727 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 3. A flask was charged with a magnetic stirbar and 1-butyl-3-methylimidazolium iodide (541 mg, 2.02 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.19 mL, 2.02 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 509 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 4. A flask was charged with a magnetic stirbar and 1-butylpyridinium chloride (603 mg, 3.51 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.33 mL, 3.51 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 868 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 5. A flask was charged with a magnetic stirbar and 1-butylpyridinium iodide (528 mg, 2.01 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.19 mL, 2.01 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 496 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 6. A flask was charged with a magnetic stirbar and trihexyl(tetradecyl)phosphonium chloride (141 mg, 0.27 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.03 mL, 0.27 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 162 mg (>99% yield) of the desired product as a clear viscous liquid. The product was analytically pure as determined by ¹H, ¹³C, and ³¹P NMR spectroscopy; spectra are provided below.

Entry 7. A flask was charged with a magnetic stirbar and trihexyl(tetradecyl)phosphonium bromide (152 mg, 0.27 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. Dimethyl sulfate (0.03 mL, 0.27 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 162 mg (>99% yield) of the desired product as a clear viscous liquid. The product was analytically pure as determined by ¹H, ¹³C, and ³¹P NMR spectroscopy.

Entry 8. A flask was charged with a magnetic stirbar and 1-butyl-3-methylimidazolium bromide (534 mg, 2.43 mmol), methyl tosylate (452 mg, 2.43 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. The mixture was then placed in an oil bath at 50 °C and vigorously stirred for 5 min, then placed under vacuum to provide 753 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 9. A flask was charged with a magnetic stirbar and 1-butyl-3-methylimidazolium iodide (548 mg, 2.04 mmol), methyl tosylate (381 mg, 2.04 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 635 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 10. A flask was charged with a magnetic stirbar and 1-butylpyridinium chloride (556 mg, 3.24 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. The flask was then placed in an oil bath at 120 °C and methyl tosylate (608 mg, 3.27 mmol) was injected via syringe through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 1.00 g (>99% yield) of the desired product as a beige solid. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy; spectra are provided below.

Entry 11. A flask was charged with a magnetic stirbar and 1-butylpyridinium iodide (699 mg, 2.66 mmol), methyl tosylate (495 mg, 2.66 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. The mixture was then vigorously stirred for 5 min, then placed under vacuum to provide 817 mg (>99% yield) of the desired product as a beige solid. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy.

Entry 12. A flask was charged with a magnetic stirbar and 1-butylpyridinium chloride (362 mg, 2.11 mmol), and then sealed with a rubber septum. A drying tube containing Drierite was fitted with a needle and inserted through the septum. The flask was then placed in an oil bath as 120 °C and trimethyl phosphate (0.25 mL, 2.12 mmol) was then injected via syringe through the septum. The mixture was then vigorously stirred for 90 min to provide 555 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H, ¹³C, and ³¹P NMR spectroscopy; spectra are provided below.

Entry 13. A flask was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium chloride (530 mg, 3.03 mmol), and trimethyloxonium tetrafluoroborate (448 mg, 3.03 mmol), and then fitted with a drying tube containing Drierite. The mixture was then vigorously stirred for 2 h to provide 685 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 14. A flask was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium bromide (1.0 g, 4.56 mmol), and trimethyloxonium tetrafluoroborate (675 mg, 4.56 mmol), and then fitted with a drying tube containing Drierite. The mixture was then vigorously stirred for 2 h to provide 1.02 g (99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 15. A flask was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium iodide (3.24 g, 12.2 mmol), and trimethyloxonium tetrafluoroborate (1.80 g, 12.2 mmol), and then fitted with a drying tube containing Drierite. After vigorously stirring for 2 h, the reaction mixture was placed under vacuum to provide 2.75 g (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 16. A flask was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium bromide (28.8 g, 132 mmol), and triethyloxonium tetrafluoroborate (25.0 g, 132 mmol), and then fitted with a drying tube containing Drierite. The mixture was then vigorously stirred for 2 h. After stirring was stopped, the top layer (determined by ¹H NMR spectroscopy to be EtBr and Et₂O) of the resulting biphasic mixture was decanted leaving 29.6 g (>99% yield) of the desired product. In a separate reaction (same scale), a flask was fitted with a distillation head in lieu of the drying tube. Upon completion, the biphasic mixture was placed in an oil bath and the EtBr and Et₂O were codistilled leaving the desired product in quantitative yield. (Note: The EtBr and Et₂O distillate was collected in 97% yield; a ¹H NMR spectroscopy in comparison with an authentic sample.

Entry 17. A flask was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium chloride (185 mg, 1.06 mmol), and triethyloxonium hexafluorophosphate (263 mg, 1.06 mmol), and then fitted with a drying tube containing Drierite. After vigorously stirring the mixture for 2 h, the reaction was placed under vacuum to give 299 mg (99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 18. A flask was charged with a magnetic stirbar, 1-butylpyridinium chloride (621 mg, 3.62 mmol), and triethyloxonium tetrafluoroborate (688 mg, 3.62 mmol), and then fitted with a drying tube containing Drierite. After vigorously stirring the mixture for 2 h, the reaction was placed under vacuum to provide 807 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 19. A flask was charged with a magnetic stirbar, trihexyl(tetradecyl)phosphonium chloride (500 mg, 0.96 mmol), and triethyloxonium tetrafluoroborate (183 mg, 0.96 mmol), and then fitted with a drying tube containing Drierite. After vigorously stirring the mixture for 2 h, the reaction was placed under vacuum to provide 546 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 20. A flask was charged with a magnetic stirbar, trihexyl(tetradecyl)phosphonium bromide (330 mg, 0.59 mmol), and triethyloxonium tetrafluoroborate (111 mg, 0.59 mmol), and then fitted with a drying tube containing Drierite. After vigorously stirring the mixture for 2 h, the reaction was placed under vacuum to provide 336 mg (>99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample.

Entry 21. A vial was charged with 1,3-dimesitylimidazolium chloride (900 mg, 2.64 mmol), CD_2Cl_2 (9 mL), and triethyloxonium tetrafluoroborate (502 mg, 2.64 mmol). The reaction was monitored by ¹H NMR spectroscopy (a spectrum of an aliquot is shown below). The mixture was then concentrated under vacuum to give 1.03 g (99% yield) of the desired product. The product was analytically pure as determined by ¹H and ¹³C NMR spectroscopy in comparison with an authentic sample; spectra are provided below.

Entry 22. A vial was charged with a magnetic stirbar, 1-butyl-3-methylimidazolium bromide (1.08 g, 4.93 mmol), and distilled H₂O (2 mL). Triethyloxonium tetrafluoroborate (936 mg, 4.93 mmol) was added in one portion and the vial was capped and shaken. Upon standing, a biphasic mixture formed. The layers were separated, providing an aqueous solution of 1-butyl-3-methylimidazolium tetrafluoroborate and an organic layer comprised of EtBr and Et₂O in ca. 1:1 molar ratio, as determined by ¹H NMR spectroscopy (see NMR spectra below). The combined EtBr and Et₂O products (897 mg) were recovered in 99% yield. The aqueous layer was extracted with CD₂Cl₂ and found to contain 1-butyl-3-methylimidazolium tetrafluoroborate with trace amounts of diethyl ether. The solution was concentrated and corresponding NMR spectra (CD₂Cl₂) of the isolated product, [BMIM]BF₄, are shown below.

Entry 23. A vial was charged with a magnetic stirbar, 1-(3-hydroxypropyl)-3-methylimidazolium bromide (1.14 g, 5.13 mmol), and CH_2Cl_2 (10 mL). The resulting suspension was vigorously stirred and triethyloxonium tetrafluoroborate (975 mg, 5.13 mmol) was added in one portion. The resulting mixture was stirred for ca. 10 min, then concentrated under vacuum to give 1.17 g (>99% yield) of the desired product. The ¹H and ¹³C NMR were consistent with previously reported spectral data;¹ corresponding NMR spectra (D₂O) of the isolated product, [HPMIM]BF₄, are shown below.

Inductively-Coupled Mass Spectroscopy. Measurements performed on aqueous samples using standard additions of KBr. Mass spectrometry (ICP-MS) was performed on a GBC OptiMass 8000, assaying for total ⁸¹Br presence. Experiments were conducted in triplicate. A representative example is as follows: A product sample solution was prepared with concentration of [BMIM]BF₄ = 1.11 wt % in 100 mL H₂O (MilliPore). From this solution, 10 mL solutions were prepared containing standard additions of 0, 11, 22, and 33 µL of 3336 ppm KBr. The solutions were analyzed in triplicate via ICP-MS to provide the graph shown below. Analysis of the data using a linear regression (R² = 0.9999) gave slope = 176502 counts/ppm, Y-int = 215007 counts, and X-int = -1.22 ppm. The concentration of Br in the initial [BMIM]BF₄ aqueous solution is the absolute value of the X-int (1.22 ppm). The background correction for the concentrations of Br in the H₂O was 0.27 ppm. This corresponds to an initial concentration of Br in [BMIM]BF₄ of 85 ± 8 ppm.



¹ A. Lesimple, O. Mamer, W. Miao, T. H. Chan. J. Am. Soc. Mass Spectrom., 2006, 17, 85.













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Entry 21 - Isolated Product

