

# **Photochemistry as a green synthetic method.**

## **Environmental assessment of C-C bond forming reactions**

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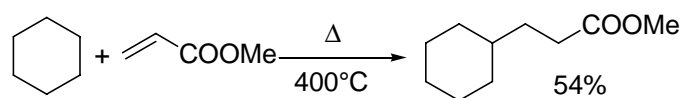
## 1. GENERAL

**EATOS** (**E**nvironmental **A**ssessment **T**ool for **O**rganic **S**ynthesis) software was downloaded from Professor J. O. Metzger web site (<http://www.chemie.uni-oldenburg.de/oc/metzger/eatos/>). Physical and environmental information (including risk and safety phrases, toxicity and ecological parameters) of considered compounds were obtained from the safety data sheets available on Sigma Aldrich web site ([www.sigma-aldrich.com](http://www.sigma-aldrich.com)). The price of used compounds and materials (of reasonable purity) were obtained from Sigma Aldrich web site too by choosing the largest quantity offered. The software affords plots of relevant parameters subdivided for the components (attached as .zip file). The figures are reported here as obtained from the application of the software, whereas in the manuscript these are rounded off.

## 2. TABLE OF PARAMETERS FOR REACTIONS 1-15.

### ALKYLATION PROCESSES

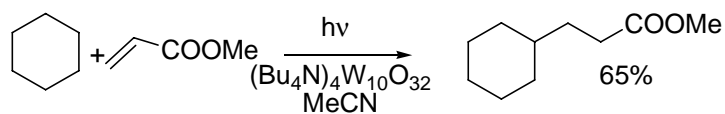
#### 2.1.1. Reaction 1, $\Delta^1$



Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
Methyl acrylate	0.90	0.80	4.70	2.80	16
Cyclohexane	0.90	-	2.30	-	37
<b>Reagents</b>	<b>1.80</b>	<b>0.80</b>	<b>7.00</b>	<b>2.80</b>	<b>53</b>
Cyclohexane <sup>a</sup>	44.7	44.7	111.7	119	857
<b>Solvents</b>	<b>44.7</b>	<b>44.7</b>	<b>111.7</b>	<b>119</b>	<b>857</b>
Methyl 3-cyclohexylpropanoate	-	-	-	-	-
<b>Coupled products</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>
<b>Total value of parameters</b>	<b>46.6</b>	<b>45.6</b>	<b>119</b>	<b>122</b>	<b>910</b>

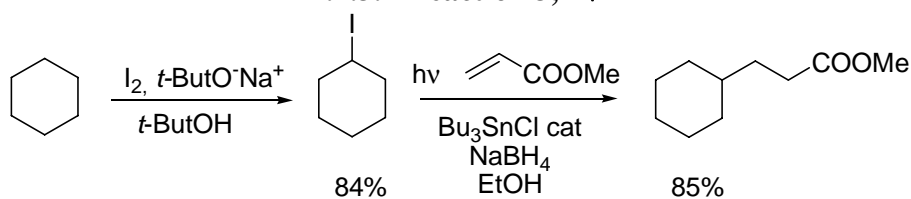
<sup>a</sup> Non stoichiometric excess of cyclohexane was considered as solvent.

### 2.1.2. Reaction 2, $h\nu^2$



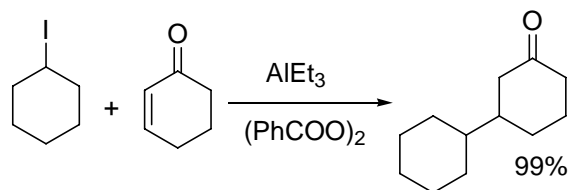
Compounds	$\text{S}^{-1}$	E	$\text{E}_{\text{in}}$	$\text{E}_{\text{out}}$	€Kg
Methyl acrylate	0.80	0.50	3.90	2.20	14
Cyclohexane	3.80	3.00	9.50	8.10	73
<b>Reagents</b>	<b>4.60</b>	<b>3.50</b>	<b>13.4</b>	<b>10.3</b>	<b>87</b>
Acetonitrile	142	142	426	284	2150
<b>Solvent</b>	<b>142</b>	<b>142</b>	<b>426</b>	<b>284</b>	<b>2150</b>
Methyl 3-cyclohexylpropanoate	-	-	-	-	-
<b>Coupled products</b>	-	-	-	-	-
Tetrabutylammonium decatungstate	-	0.50	1.40	0.50	195
<b>Catalyst</b>	-	<b>0.50</b>	<b>1.40</b>	<b>0.50</b>	<b>195</b>
<b>Total value of parameters</b>	<b>147</b>	<b>146</b>	<b>441</b>	<b>295</b>	<b>2432</b>

### 2.1.3. Reaction 3, $h\nu^{3,4}$



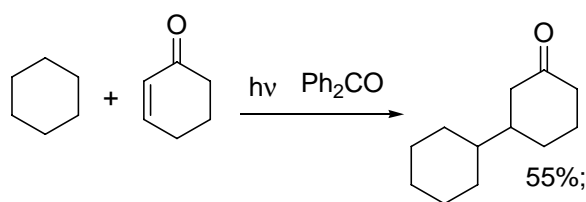
Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€Kg
Cyclohexane	5.80	5.20	5.80	20.9	24
Sodium <i>t</i> -butylate	0.70	-	3.00	-	129
Iodine	1.70	0.60	2.60	3.50	340
Methyl acrylate	2.50	1.90	2.50	10.0	44
NaBH <sub>4</sub>	0.30	-	1.40	0.20	94
<b>Reagents</b>	<b>11.0</b>	<b>7.70</b>	<b>15.3</b>	<b>34.6</b>	<b>631</b>
Ethanol	4.60	4.60	9.10	13.7	90
<b>Solvents</b>	<b>4.60</b>	<b>4.60</b>	<b>9.10</b>	<b>13.7</b>	<b>90</b>
Bu <sub>3</sub> SnCl	0.40	0.40	0.80	2.60	65
<b>Catalyst</b>	<b>0.40</b>	<b>0.40</b>	<b>0.80</b>	<b>2.60</b>	<b>65</b>
NaS <sub>2</sub> O <sub>3</sub> 2% aq	1.80	1.80	1.80	1.80	14
MgSO <sub>4</sub>	0.30	0.30	0.30	2.20	16
<b>Auxiliaries (reaction)</b>	<b>2.10</b>	<b>2.10</b>	<b>2.10</b>	<b>4.00</b>	<b>30</b>
NaI	-	0.90	-	2.20	-
<i>t</i> -Butanol	-	0.40	-	0.40	-
NaBH <sub>3</sub> I	-	1.00	-	1.00	-
<b>Product/Coupled Product</b>	<b>-</b>	<b>2.30</b>	<b>-</b>	<b>3.60</b>	<b>-</b>
<b>Total value of parameters</b>	<b>18.1</b>	<b>16.7</b>	<b>27.3</b>	<b>58.5</b>	<b>816</b>

2.1.4. Reaction 4,  $\Delta^5$ .



Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
Iodocyclohexane	11.8	10.6	29.6	106	10768
2-Cyclohexen-1-one	0.50	-	2.7	-	746
Triethylaluminum	3.5	2.8	17.4	2.8	1289
<b>Reagents</b>	<b>15.8</b>	<b>13.4</b>	<b>49.7</b>	<b>109</b>	<b>12803</b>
Diethyl ether	39.9	39.9	139.6	120	1223
<b>Solvent</b>	<b>39.9</b>	<b>39.9</b>	<b>139.6</b>	<b>120</b>	<b>1223</b>
Benzoyl peroxide	2.7	2.7	15.0	10.9	185
<b>Catalyst</b>	<b>2.7</b>	<b>2.7</b>	<b>15.0</b>	<b>10.9</b>	<b>185</b>
HCl 37% aq	5.0	5.0	9.9	24.8	45
CH <sub>2</sub> Cl <sub>2</sub>	14.9	14.9	29.9	99.5	101
<b>Auxiliaries (Reaction)</b>	<b>19.9</b>	<b>19.4</b>	<b>39.8</b>	<b>124</b>	<b>146</b>
3-Cyclohexyl-methylpropanoate	-	-	-	-	-
Et <sub>3</sub> AlI	-	1.3	-	1.3	-
<b>Products-coupled products</b>	<b>-</b>	<b>1.3</b>	<b>-</b>	<b>1.3</b>	<b>-</b>
<b>Total value of parameters</b>	<b>78.3</b>	<b>76.7</b>	<b>244.1</b>	<b>365</b>	<b>14357</b>

### 2.1.5. Reaction 5, hv<sup>6</sup>



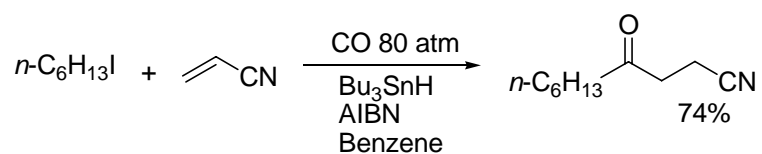
Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
Cyclohexane	0.84	-	2.10	-	16
2-Cyclohexen-1-one	1.0	0.0	3.40	-	1069
<b>Reagents</b>	<b>2.80</b>	<b>76.7</b>	<b>5.50</b>	-	<b>1085</b>
Cyclohexane <sup>a</sup>	76.7	76.7	191.7	307	1206
<b>Solvents</b>	76.7	76.7	191.7	307	1206
Benzophenone	1.8	1.8	1.8	9.4	95
<b>Catalyst<sup>b</sup></b>	<b>1.8</b>	<b>1.8</b>	<b>1.8</b>	<b>9.4</b>	<b>95</b>
3-Cyclohexyl-methylpropanoate	-	-	-	-	-
<b>Products-coupled products</b>	-	-	-	-	-
<b>Total value of parameters</b>	<b>80.3</b>	<b>78.5</b>	<b>199</b>	<b>316</b>	<b>2386</b>

<sup>a</sup> Non stoichiometric excess of cyclohexane was considered as solvent. <sup>b</sup> Photoproducts of benzophenone considered.



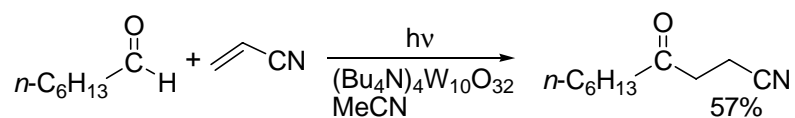
## 2.2. ACYLATION PROCESSES

### 2.2.1. Reaction 6, $\Delta^7$



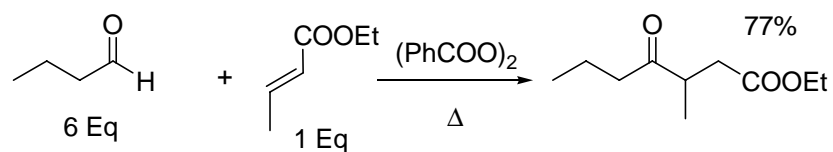
Compounds	$S^{-1}$	E	$E_{in}$	$E_{out}$	€/Kg
Acrylonitrile	1.7	1.3	9.4	11.2	17
1-Iodohexane	1.7	-	5.1	-	613
Tributylhydride	3.3	0.90	13.2	9.4	3541
Carbon monoxide	98.7	98.4	543	984	2435
<b>Reagents</b>	<b>105</b>	<b>101</b>	<b>571</b>	<b>1005</b>	<b>6606</b>
Benzene	426	426	2344	3196	7795
<b>Solvents</b>	<b>426</b>	<b>426</b>	<b>2344</b>	<b>3196</b>	<b>7795</b>
AIBN	-	-	-	-	16
<b>Catalyst</b>	-	-	-	-	<b>16</b>
Tributyltin iodide	-	2.5	-	2.5	-
4-keto-decanitrile	-	0,0	-	-	-
<b>Products-coupled products</b>	-	<b>2.5</b>	-	<b>2.5</b>	-
<b>Total value of parameters</b>	<b>531</b>	<b>530</b>	<b>2915</b>	<b>4203</b>	<b>14417</b>

### 2.2.2. Reaction 7, hv<sup>8</sup>



Compounds	$S^{-1}$	E	$E_{\text{in}}$	$E_{\text{out}}$	€Kg
Acrylonitrile	1.20	0.80	1.80	2.20	12
Heptanal	0.60	-	3.10	-	62
<b>Reagents</b>	<b>1.80</b>	<b>0.80</b>	<b>4.90</b>	<b>2.20</b>	<b>74</b>
Acetonitrile	82.4	82.4	206	165	1248
<b>Solvent</b>	<b>82.4</b>	<b>82.4</b>	<b>206</b>	<b>165</b>	<b>1248</b>
Tetrabutylammonium decatungstate	0.70	0.70	1.40	0.70	286
<b>Catalyst</b>	<b>0.70</b>	<b>0.70</b>	<b>1.40</b>	<b>0.70</b>	<b>286</b>
4-keto-decanitrile	-	-	-	-	-
<b>Products-coupled products</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>
<b>Total value of parameters</b>	<b>84.9</b>	<b>83.9</b>	<b>212</b>	<b>168</b>	<b>1608</b>

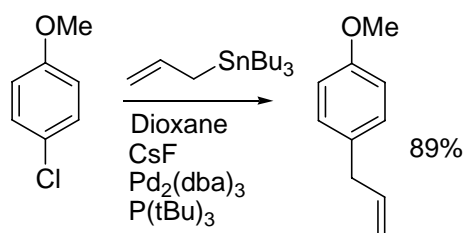
### 2.2.3. Reaction 8, $\Delta^9$



Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€Kg
Ethyl crotonate	0.8	0.3	2.8	0.8	204
Butanal	3.0	2.6	4.6	14.1	110
<b>Reagents</b>	<b>3.8</b>	<b>2.9</b>	<b>7.4</b>	<b>14.9</b>	<b>314</b>
Benzoyl peroxide	0.2	0.2	0.9	0.2	9
<b>Catalyst</b>	<b>0.2</b>	<b>0.2</b>	<b>0.9</b>	<b>0.2</b>	<b>9</b>
Diethyl ether	1.0	1.0	3.4	2.9	30
Sodium bicarbonate 5% <sub>aq</sub>	3.4	3.4	3.4	3.8	1
<b>Auxiliaries (isolation)</b>	<b>4.4</b>	<b>4.4</b>	<b>6.8</b>	<b>6.7</b>	<b>31</b>
4-Oxo-3-methylethyl-heptanoate	0.0	0.0	0.0	0.0	0
<b>Product-coupled products</b>	<b>0.0</b>	<b>0.0</b>	<b>0.0</b>	<b>0.0</b>	<b>0</b>
<b>Total value of parameters</b>	<b>8.4</b>	<b>7.5</b>	<b>15.1</b>	<b>21.8</b>	<b>354</b>

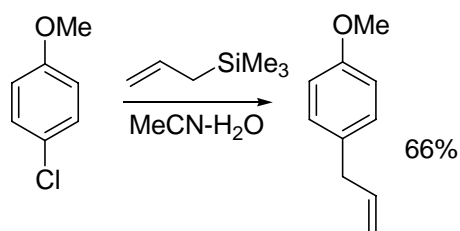
## 2.3 ARYLATION PROCESSES.

### 2.3.1. Reaction 9, $\Delta^{10}$



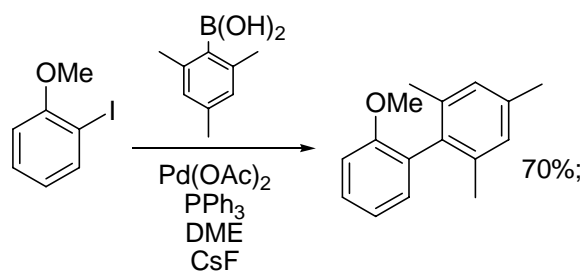
Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
4-Chloroanisole	1.1	0.2	4.3	0.2	1378
Allyltributyltin	2.7	0.4	14.6	3.8	7872
<b>Reagents</b>	<b>3.8</b>	<b>0.6</b>	<b>18.9</b>	<b>4.0</b>	<b>9250</b>
1,4-dioxane	10.2	10.2	51.0	40.8	95
<b>Solvent</b>	<b>10.2</b>	<b>10.2</b>	<b>51.0</b>	<b>40.8</b>	<b>95</b>
Tri( <i>t</i> -butyl)phosphine	0.1	0.1	0.7	0.1	2922
Pd <sub>2</sub> (dba) <sub>3</sub>	0.1	0.1	0.8	0.1	3009
<b>Catalyst</b>	<b>0.2</b>	<b>0.2</b>	<b>1.5</b>	<b>0.2</b>	<b>5931</b>
CsF	2.6	2.6	11.5	25.6	5458
<b>Auxiliaries materials (reaction)</b>	<b>2.6</b>	<b>2.6</b>	<b>11.5</b>	<b>25.6</b>	<b>5458</b>
Silica gel	3.8	3.8	3.8	3.8	113
Ethyl acetate	34.0	34.0	68.0	68.0	377
<b>Auxiliaries materials (isolation)</b>	<b>37.8</b>	<b>37.8</b>	<b>71.8</b>	<b>71.8</b>	<b>490</b>
4-Allylanisole	-	-	-	-	-
tributylchlorotin	-	2.20	-	18.7	-
<b>Product-coupled products</b>	<b>-</b>	<b>2.20</b>	<b>-</b>	<b>18.7</b>	<b>-</b>
<b>Total value of parameters</b>	<b>54.6</b>	<b>53.6</b>	<b>155</b>	<b>161</b>	<b>21224</b>

### 2.3.2. Reaction 10, hv<sup>11</sup>



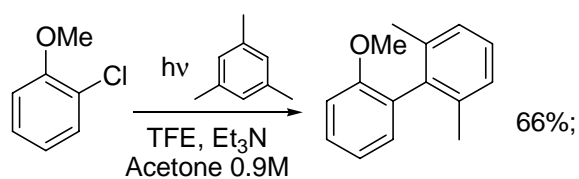
Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
4-Chloroanisole	1.5	0.9	6.6	8.9	1618
Allyltrimethylsilane	5.8	4.7	40.9	46.7	19916
<b>Reagents</b>	<b>7.3</b>	<b>5.6</b>	<b>47.5</b>	<b>55.6</b>	<b>21534</b>
MeCN	66.4	66.4	199.3	133	1013
Water	17.1	17.1	17.1	17.1	1
<b>Solvents</b>	<b>83.5</b>	<b>83.5</b>	<b>216.4</b>	<b>151</b>	<b>1014</b>
CH <sub>2</sub> Cl <sub>2</sub>	40.6	40.6	81.3	271	277
<b>Auxiliaries (isolation)</b>	<b>40.6</b>	<b>40.6</b>	<b>81.3</b>	<b>271</b>	<b>277</b>
4-Allylanisole	-	-	-	-	-
Trimethylsilylchloride	-	0.70	-	0.70	-
<b>Product-coupled products</b>	<b>-</b>	<b>0.70</b>	<b>-</b>	<b>0.70</b>	<b>-</b>
<b>Total value of parameters</b>	<b>131</b>	<b>130</b>	<b>345</b>	<b>478</b>	<b>22800</b>

### 2.3.3. Reaction 11, $\Delta^{12}$



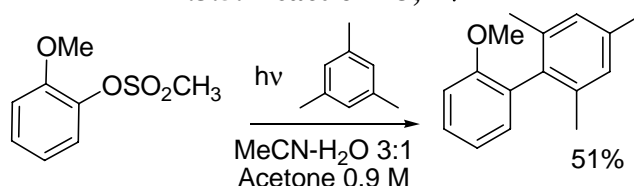
Compounds	$S^{-1}$	E	$E_{in}$	$E_{out}$	€/Kg
2-Iodoanisole	1.0	-	3.7	-	1676
Mesitylboronic acid	1.5	0.8	7.2	7.5	18386
<b>Reagents</b>	<b>2.5</b>	<b>0.8</b>	<b>10.9</b>	<b>7.5</b>	<b>20062</b>
Dimethoxyethane (DME)	5.2	5.2	28.7	28.7	295
<b>Solvent</b>	<b>5.2</b>	<b>5.2</b>	<b>28.7</b>	<b>28.7</b>	<b>295</b>
$PPh_3$	0.20	0.15	0.2	0.6	12
$Pd(OAc)_2$	0.05	0.05	0.2	-	2607
<b>Catalyst</b>	<b>0.2</b>	<b>0.2</b>	<b>0.4</b>	<b>0.6</b>	<b>2630</b>
CsF	2.9	2.9	12.9	24.4	4343
<b>Auxiliaries Matherials</b>	<b>2.9</b>	<b>2.9</b>	<b>12.9</b>	<b>24.4</b>	<b>4343</b>
2'-4'-6'-Trimethyl-2-methoxybiphenyl	-	-	-	-	-
$IB(OH)_2$	-	0.8	-	0.80	-
<b>Product-coupled products</b>	-	<b>0.8</b>	-	<b>0.80</b>	-
<b>Total value of parameters</b>	<b>10.8</b>	<b>9.9</b>	<b>52.9</b>	<b>62.0</b>	<b>27321</b>

### 2.3.4. Reaction 12, hv<sup>13</sup>



Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€Kg
2-Chloroanisole	1.0	-	1.0	0.20	780
Mesitylene	16.1	15.3	24.1	61.2	1257
<b>Reagents</b>	<b>17.1</b>	<b>15.3</b>	<b>25.1</b>	<b>61.4</b>	<b>2037</b>
TFE	179	179	718	359	63012
Acetone	10.6	10.6	21.1	24.7	36
<b>Solvents</b>	<b>190</b>	<b>190</b>	<b>739</b>	<b>384</b>	<b>63048</b>
Triethylamine	0.7	0.70	2.4	0.7	38
<b>Auxiliaries (reaction)</b>	<b>0.7</b>	<b>0.70</b>	<b>2.4</b>	<b>0.7</b>	<b>38</b>
2'-4'-6'-Trimethyl-2-methoxybiphenyl	-	-	-	-	-
Chloridric Acid	-	0.60	-	-	-
<b>Product-coupled products</b>	<b>-</b>	<b>0.60</b>	<b>-</b>	<b>-</b>	<b>-</b>
<b>Total value of parameters</b>	<b>208</b>	<b>206</b>	<b>766</b>	<b>446</b>	<b>65123</b>

### 2.3.5. Reaction 13, hv<sup>14</sup>

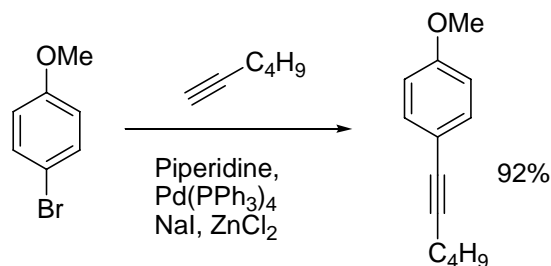


Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
Guaiacol	1.10	-	1.60	-	83
Mesitylene	5.60	4.60	8.40	18.3	440
Methanesulfonylchloride	1.10	0.10	1.10	0.80	37
Triethylamine	1.30	0.40	4.60	0.40	73
<b>Reagents</b>	<b>9.10</b>	<b>5.10</b>	<b>15.7</b>	<b>19.5</b>	<b>633</b>
Acetonitrile	55.2	55.2	166	138	835
CH <sub>2</sub> Cl <sub>2</sub>	2.70	2.70	5.40	14.2	18
Water	23.4	23.4	23.4	23.4	1
Acetone	7.40	7.40	14.8	17.3	25
<b>Solvents</b>	<b>88.7</b>	<b>88.7</b>	<b>210</b>	<b>193</b>	<b>879</b>
2'-4'-6'-Trimethyl-2-methoxybiphenyl	-	-	-	-	-
Methanesulfonic acid	-	1.60	-	4.00	-
Trimethylammonium chloride	-	1.40	-	4.60	-
<b>Product-coupled products</b>	-	<b>3.00</b>	-	<b>8.60</b>	-
<b>Total value of parameters</b>	<b>97.8</b>	<b>96.8</b>	<b>226</b>	<b>221</b>	<b>1512</b>

As hinted in the text, reactions 11-13 make use of different starting compounds (2-iodoanisole, 2-chloroanisole and guaiacol). However, the input parameters of the three compounds are quite similar and the output parameters are small or zero. Therefore, this difference has no major effect on the results, apart from an economical advantages in using less expensive guaiacol instead of halides. Anyway, major contribution to environmental and economical cost of reactions is given by nucleophile (eq. 11) or solvent (eq. 12 and 13) rather than the choice of aromatic derivative.

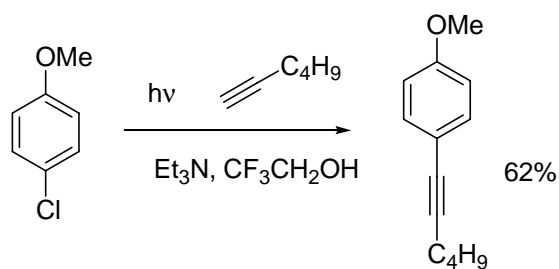


### 2.3.6. Reaction 14, $\Delta^{15}$



Compounds	$S^{-1}$	E	$E_{in}$	$E_{out}$	€Kg
4-Bromoanisole	1.1	-	4.0	-	170
1-Hexyne	0.6	-	2.2	-	523
<b>Reagents</b>	<b>1.7</b>	-	<b>6.2</b>	-	<b>693</b>
Piperidine	6.7	6.7	23.3	26.6	344
<b>Solvents</b>	<b>6.7</b>	<b>6.7</b>	<b>23.3</b>	<b>26.6</b>	<b>344</b>
$Pd(PPh_3)_4$	0.4	0.4	2.8	0.4	15023
NaI	0.2	0.2	0.3	0.6	34
$ZnCl_2$	0.2	0.2	0.3	1.2	14
<b>Catalysts</b>	<b>0.8</b>	<b>0.8</b>	<b>3.4</b>	<b>2.2</b>	<b>15100</b>
$NH_4Cl$ aq	30.8	30.8	33.8	63.8	152
$CH_2Cl_2$	81.8	81.8	164	546	559
<b>Auxiliaries (isolation)</b>	<b>113</b>	<b>113</b>	<b>198</b>	<b>610</b>	<b>711</b>
4-(1-Hexynyl)-anisole	-	-	-	-	-
HBr	-	0.4	-	2.1	-
<b>Product-coupled products</b>	-	<b>0.4</b>	-	<b>2.1</b>	-
<b>Total value of parameters</b>	<b>122</b>	<b>121</b>	<b>231</b>	<b>641</b>	<b>16800</b>

### 2.3.7. Reaction 15, hv<sup>16</sup>



Compounds	S <sup>-1</sup>	E	E <sub>in</sub>	E <sub>out</sub>	€/Kg
4-Chloroanisole	1.2	-	5.5	-	1356
1-Hexyne	7.0	6.3	28.1	44.3	6584
<b>Reagents</b>	<b>8.2</b>	<b>6.3</b>	<b>33.6</b>	<b>44.3</b>	<b>7940</b>
2,2,2-Trifluoroethanol	238	238	715	1668	60140
<b>Solvents</b>	<b>238</b>	<b>238</b>	<b>715</b>	<b>1668</b>	<b>60140</b>
Triethylamine	0.9	-	3.0	2.0	20
<b>Auxiliaries (Reaction)</b>	<b>0.9</b>	<b>-</b>	<b>3.0</b>	<b>2.0</b>	<b>20</b>
4-(1-Hexynyl-anisole)	-	-	-	-	-
Chloridric acid	-	0.2	-	7.4	-
<b>Product-coupled products</b>	<b>-</b>	<b>0.2</b>	<b>0</b>	<b>7.4</b>	<b>-</b>
<b>Total value of parameters</b>	<b>247</b>	<b>245</b>	<b>752</b>	<b>1724</b>	<b>68100</b>

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