

Selective Aerobic Oxidative Dibromination of Alkenes with Aqueous HBr and Sodium Nitrite as a Catalyst

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Supplementary Information

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NMR spectra of (2,3-dibromo-1,1,1-trifluoropropan-2-yl)benzene 20

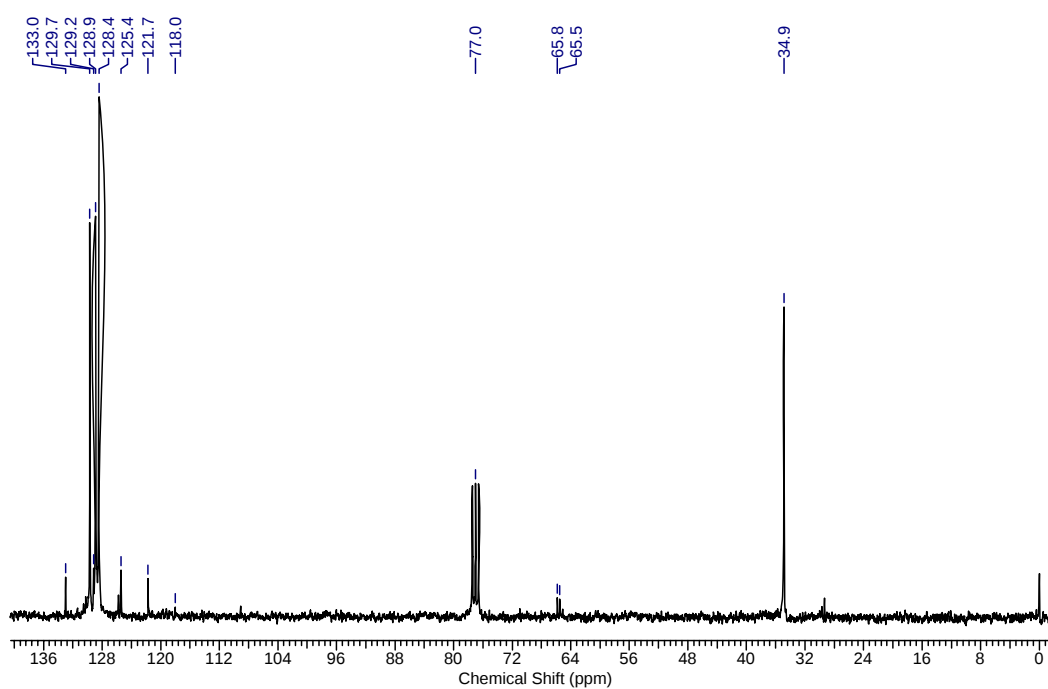
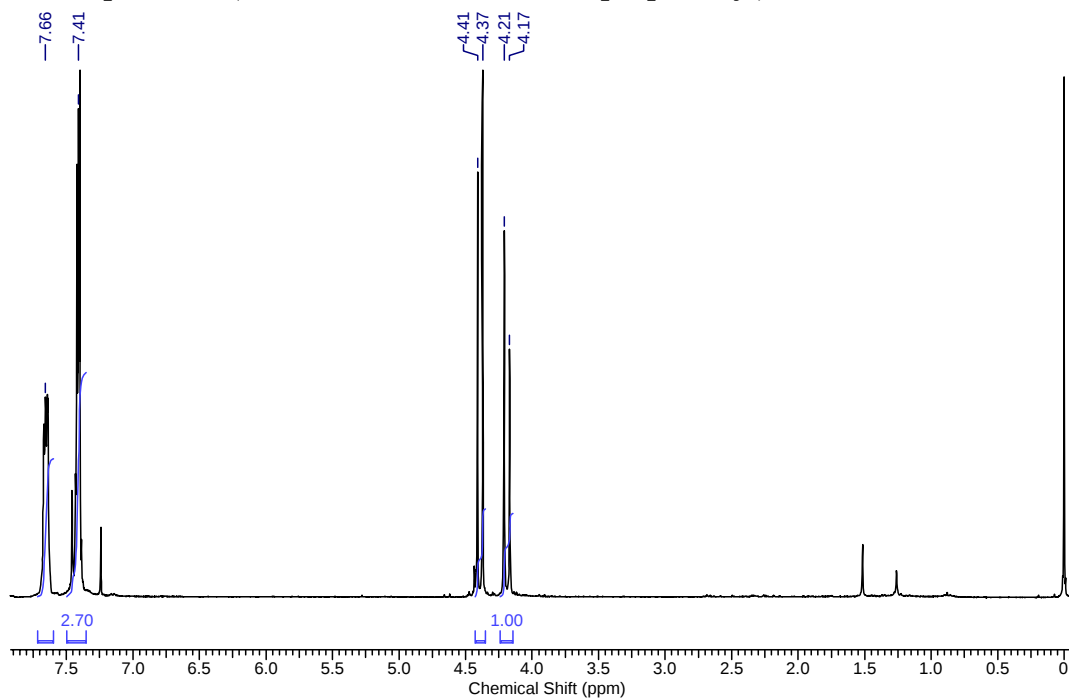
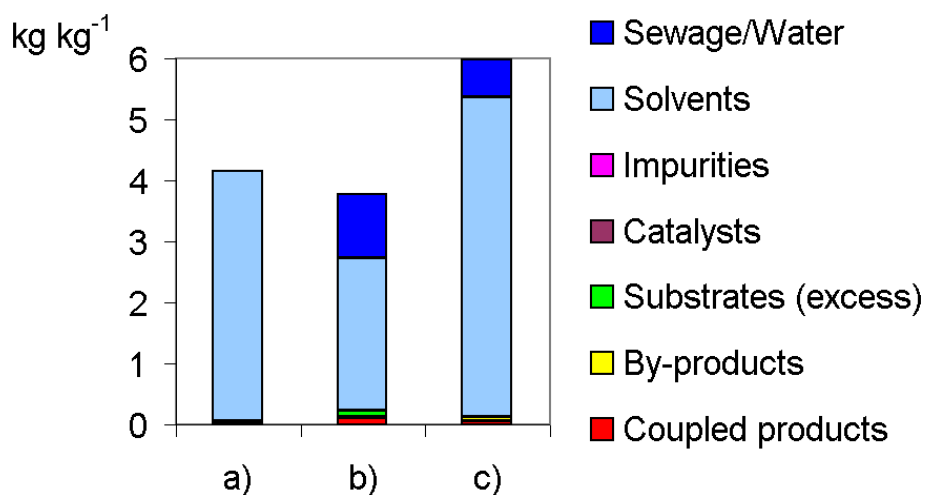
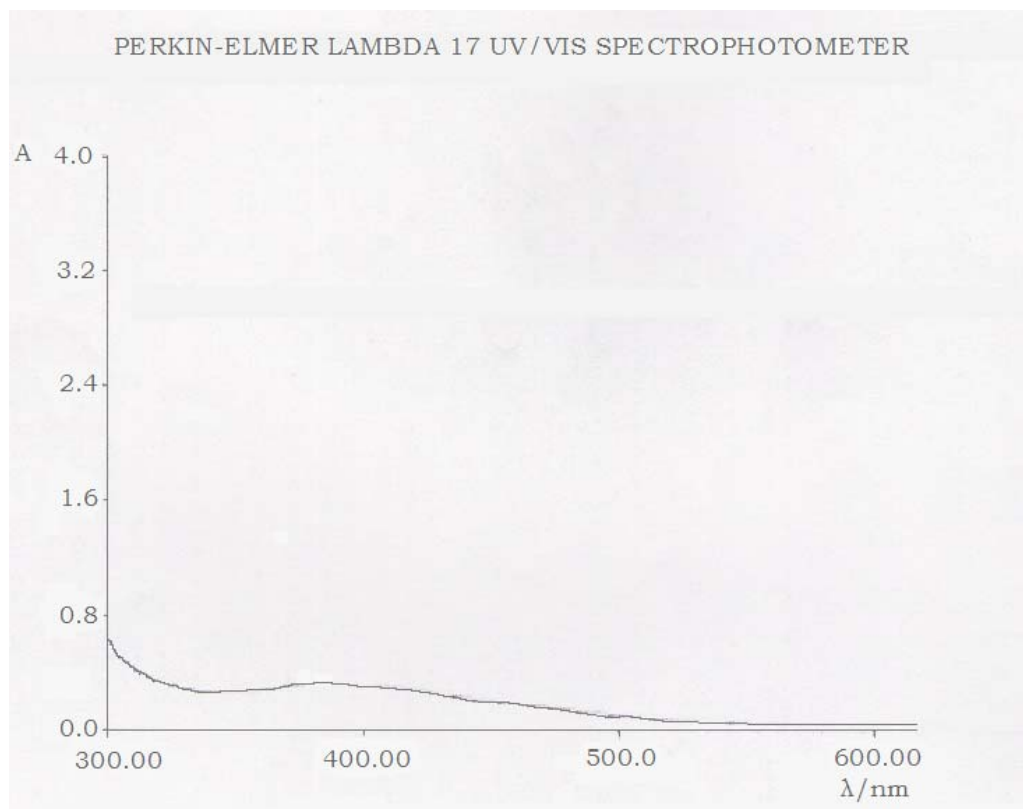


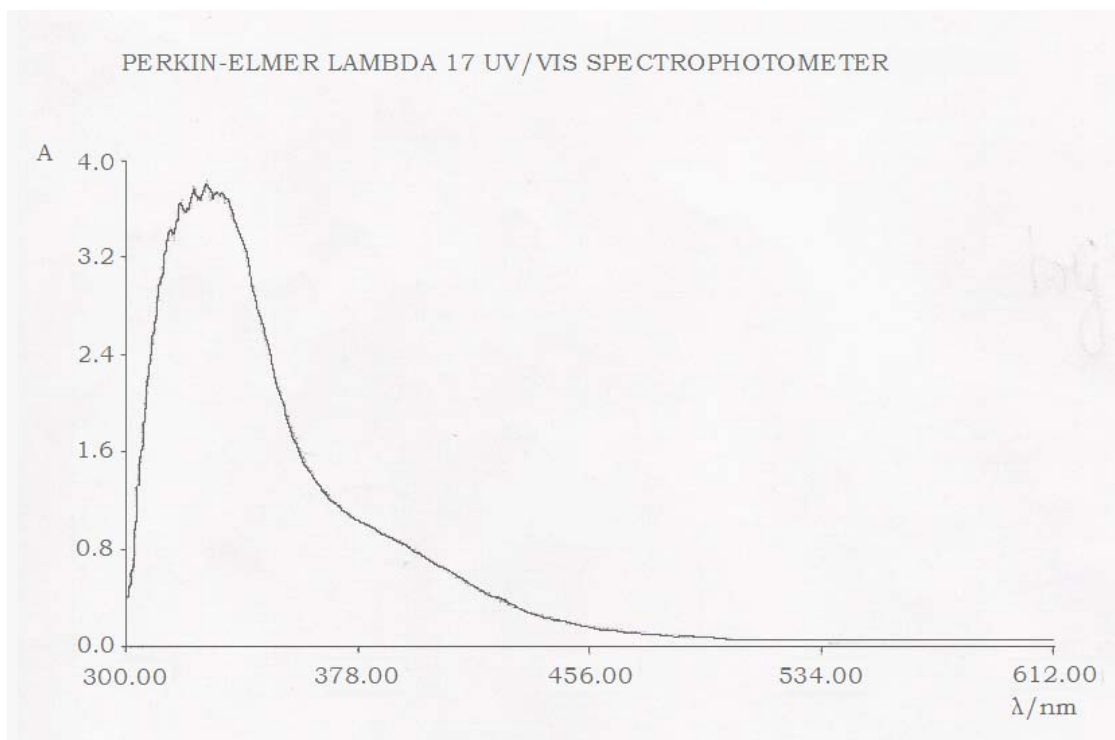
Figure 1: Environmental factor E of different bromination methods shown in Scheme 6 using the software EATOS.¹ For details see Table 2. Work-up procedures have not been described in detail^{2,3} and, thus, were not considered here: (a) Recrystallization; b) Washing with water, brine and drying over sodium sulfate; c) Pouring into 10 mL of water



UV-vis spectrum of Br₂ in CH₃CN (2·10⁻³ M)



UV-vis spectrum of NOBr (NaNO₂ and HBr in CH₃CN (2·10⁻³ M))



References

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- 3 H. G. O. Becker, in *Organikum: Organisch-chemisches Grundpraktikum*, Wiley-VCH, Weinheim, 21st ed., 2001, p. 299.