

SUPPORTING INFORMATION

Solvent free base Catalysis and Transesterification over basic functionalised Metal-Organic Frameworks

M. Savonnet, S. Aguado, U.Ravon, D. Bazer-Bachi, V. Lecocq, N. Bats, C. Pinel, D. Farrusseng*

IRCELYON, Institut de recherches sur la catalyse et l'environnement de Lyon, 2 avenue Albert Einstein, F-69626 Villeurbanne

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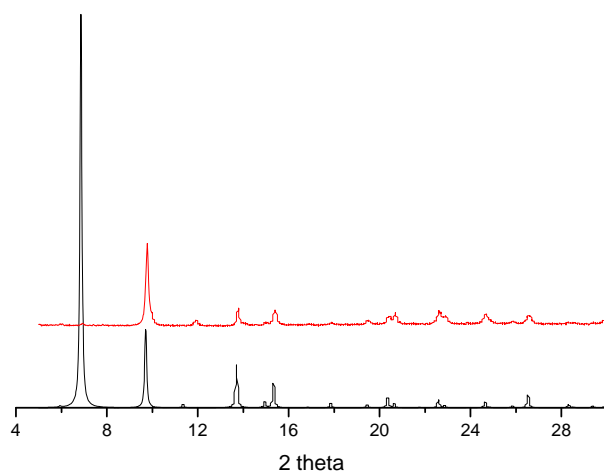


Figure 1. Powder X-Ray diffraction patterns of IRMOF-3 **1a**. Experimental result (top), simulated pattern from IRMOF-3.cif file (bottom).

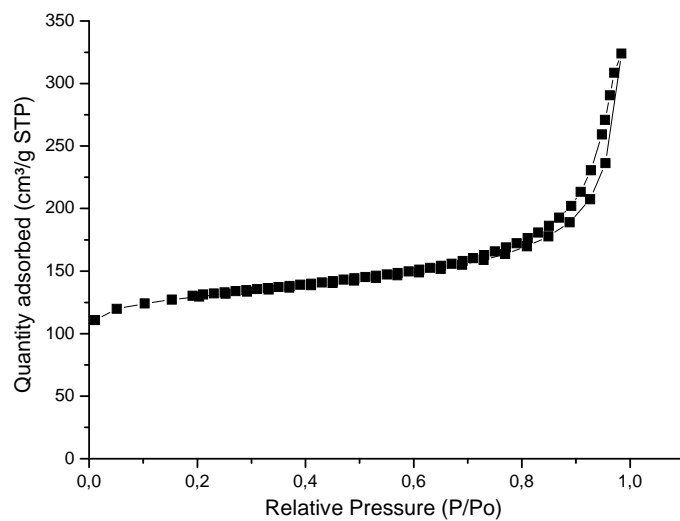


Figure 2. N₂ isotherm at 77K of IRMOF-3 **1a**

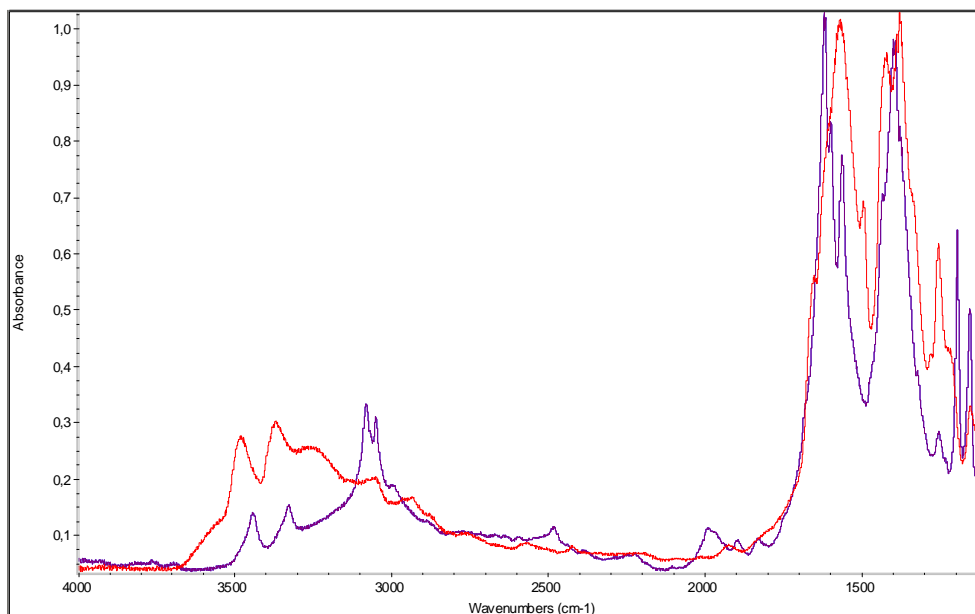
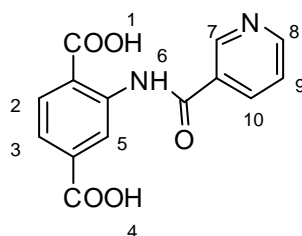
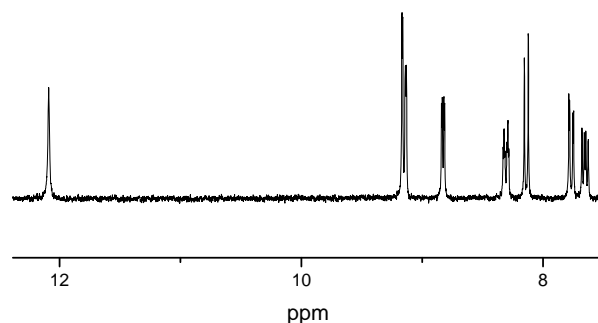


Figure 3. FT-IR spectrum of as-synthesized (red) **1a** and modified IRMOF-3 **1b** (blue).

The functionalised ligand was obtained by acylation of dimethylaminoterephthalate (1.5 g) with nicotinoyl chloride (3 g) in a dichloromethane solution containing triethylamine (9 mL) and DMAP (0.7 g) at room temperature for 5 h. The ligand is obtained after saponification, followed by acidification on Amberlite IR-120 resin.



$^1\text{H RMN}$: (DMSO, $\text{D}_2\text{O}/\text{DCI}$, 250 MHz): δ (ppm) 13.44 (s, 2H, H_1 , H_4), 12.09 (s, 1H, H_6), 9.16 (d, 1H, $^4J_{\text{HH}} = 1.58$ Hz, H_5), 9.13 (d, 1H, $^4J_{\text{HH}} = 1.9$ Hz, H_7), 8.82 (dd, $^4J_{\text{HH}} = 1.58$ Hz, $^3J_{\text{HH}} = 4.9$ Hz, H_8), 8.30 (ddd, $^4J_{\text{HH}} = 1.9$ Hz, $^4J_{\text{HH}} = 1.58$ Hz, $^3J_{\text{HH}} = 8$ Hz, H_{10}), 8.14 (d, 1H, $^1J_{\text{HH}} = 8$ Hz, H_2), 7.76 (dd, 1H, $^3J_{\text{CH}} = 1.58$ Hz, $^3J_{\text{CH}} = 8$ Hz, H_3), 7.64 (dd, 1H, $^3J_{\text{CH}} = 4.9$ Hz, $^3J_{\text{CH}} = 8$ Hz, H_9)

Figure 4. ^1H -NMR results of the functionalised ligand.

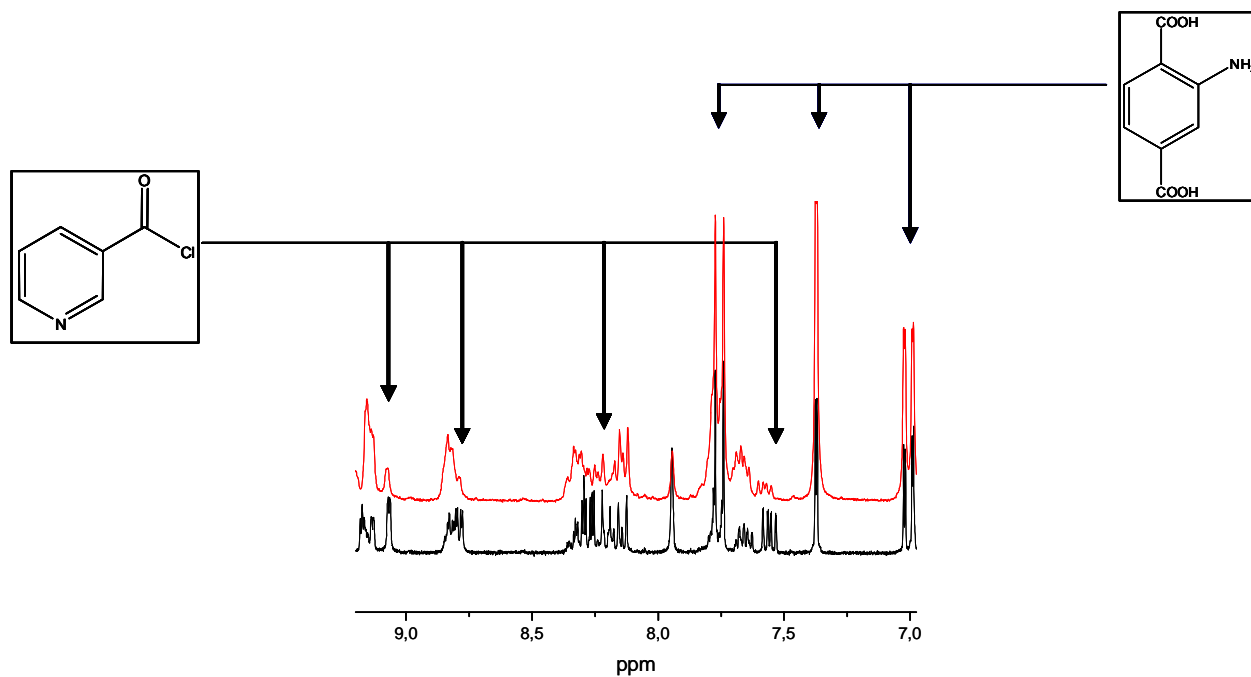


Figure 5. $^1\text{H-NMR}$ results of digested modified IRMOF-3 **1b** without washing (black) and digested modified IRMOF-3 **1b** with repeated washings (red).

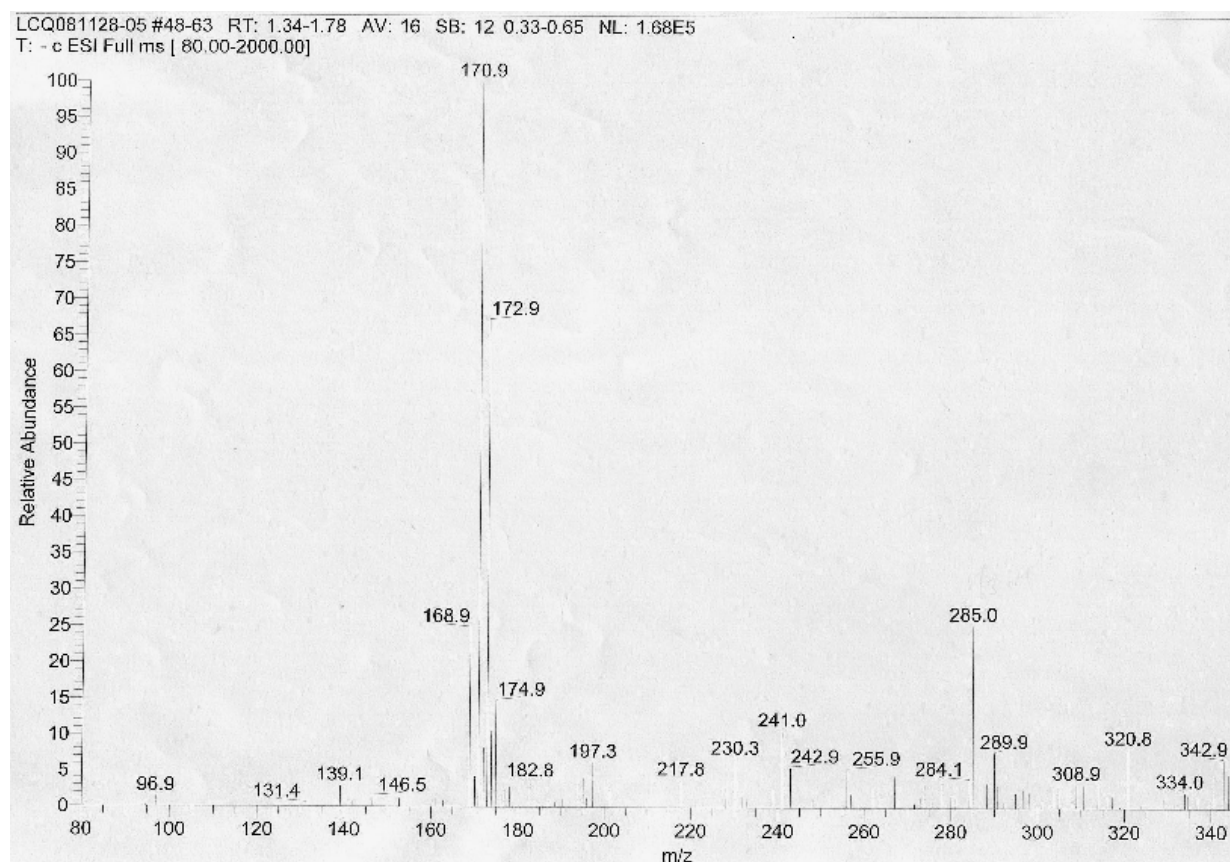


Figure 6. Negative ion mode ESI-MS of digested modified IRMOF-3 **1b**

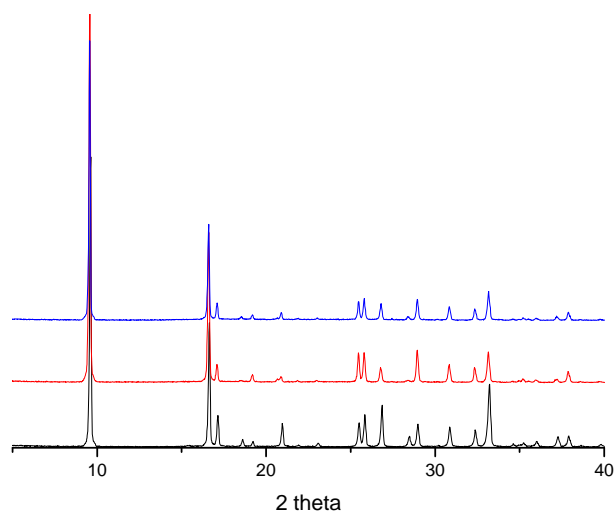


Figure 7. Powder X-Ray diffraction patterns of ZnF(Am₂TAZ) 2a (black), post-functionalized ZnF(Am₂TAZ) 2b (red) and post-functionalised ZnF(Am₂TAZ) 2b after reaction.