

Supplementary data: Adsorption isotherms

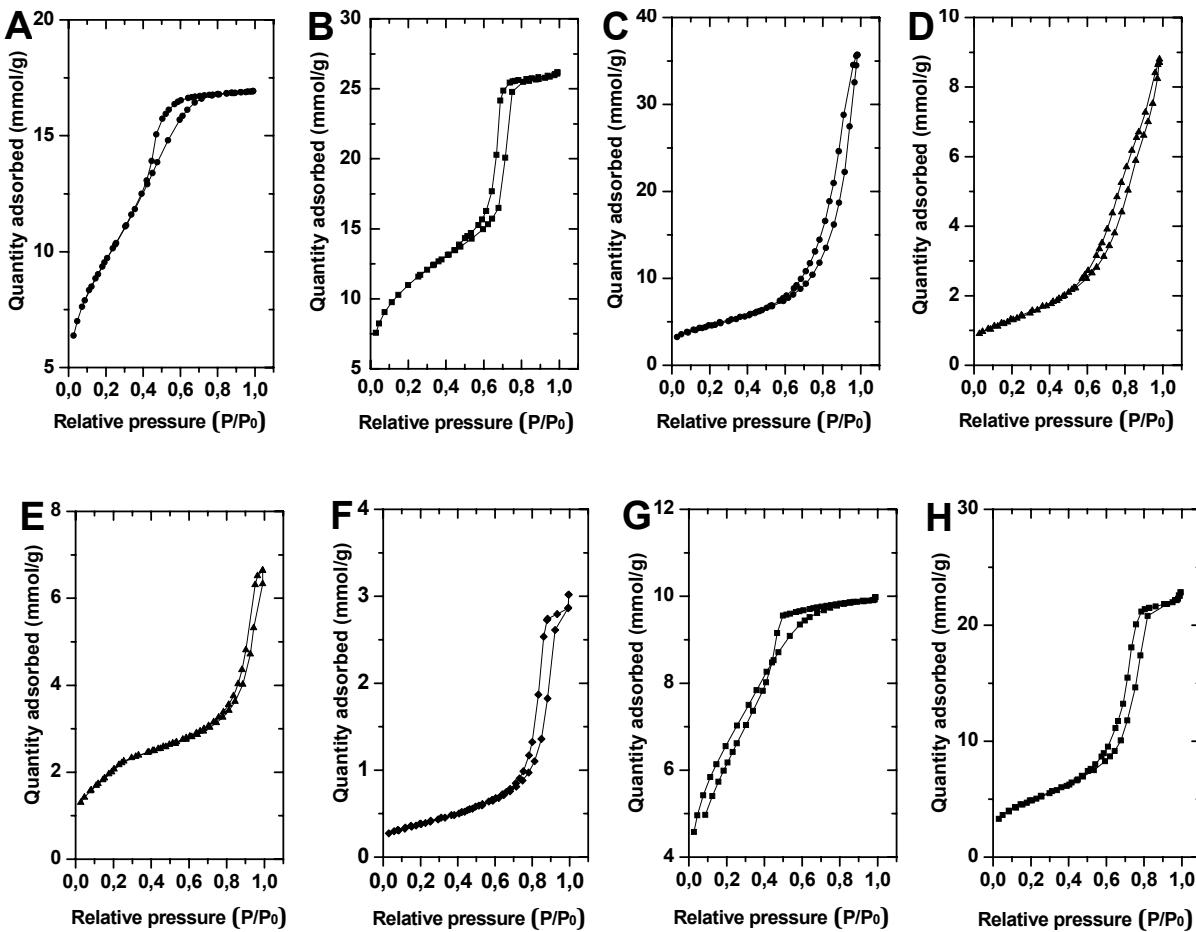


Figure 1. Nitrogen adsorption isotherms and pore size distributions derived from the adsorption branch. (A): $\text{SiO}_2\text{-A}$; (B) $\text{SiO}_2\text{-B}$; (C) Al_2O_3 ; (D) TiO_2 ; (E) $\text{Al}_2\text{O}_3/\text{SiO}_2$; (F) WO_3 ; (G) $\text{WO}_3/\text{SiO}_2\text{-3}$ and (H) WO_4SiO_2 .

ICP-AES Analysis

Performed by ALS Scandinavia AB.

Webb: www.alsglobal.se

Email: info.lu@alsglobal.com

Phone: +46 920 28 99 00

Fax: +46 920 28 99 40

Address: Aurorum 10, 977 75 Luleå, Sweden.

Method

0.1 g was melted with 0.375 g LiBO₂ and was dissolved in HNO₃. LOI (loss of ignition) was performed at 1000°C. Analysis has been performed according to EPA-method (modified) 200.7(ICP-AES) and 200.8 (ICP-QMS).

Result

ICP-AES analysis for W in the WO₄/SiO₂ material: 9880 mg/kg.

Characterization and analysis techniques

¹H-NMR was used for quantitative analysis of the products obtained after each run. The spectra were recorded at 400 MHz using a JEOL, model Eclipse FT-NMR Oxford instrument. The olefin bond in cyclohexene at 5.60 ppm was monitored and quantified using benzene as internal standard. The α and β methylene groups in adipic acid, which appear at 2.20 ppm and 1.50 ppm, respectively were used for quantification of the yield. The two α methine groups in cyclohexanediol at 3.25 ppm were used to determine the amount of this compound in the reaction mixture. The ¹H-NMR runs were made at 25°C using either CDCl₃ or DMSO-d₆ as solvent. The concentration of added standard (benzene) was controlled by weight with an accuracy of ± 1mg.

Determination of the specific surface area was performed on an ASAP 2010 instrument, using nitrogen adsorption and the Brunauer-Emmett-Teller (BET) method.²² The pore size distribution was calculated from the isotherms using the Barret-Joyner-Halenda (BJH) procedure.²³ All samples were dried at 225°C in a vacuum oven for approximately 3 h before measurement.

Samples for the transmission electron microscopy (TEM), run on a JEOL 1200 EX II instrument at 120 kV, were prepared by placing a drop of an ethanol dispersion of the mesoporous material onto a copper Holey grid.

Scanning electron microscopy (SEM) was performed with a LEO Ultra 55 FEG equipped with an Oxford Inca x-sight EDX system, operated at 6.0 kV with WD = 3 mm. Specimen were prepared by placing the material onto a carbon tape with a silver glue.

Small angle X-ray scattering (SAXS) was performed with a Kratky compact small angle system on a HECUS Mbraun, Graz instrument. All runs were performed under vacuum at 50 kV and 40 mA. The runs were performed with monochromatic CuKα1 radiation using a Ni-filter. The samples were prepared by moulding the mesoporous oxide and then placing the particles in a paste holder with thin mica windows.

Low angle X-ray powder diffraction (XRD) was performed on a LynxEye AXS D8 ADVANCE 0/20 diffractometer, linear detector. The runs were performed at 40 kV and 40 mA, in monochromatic mode with G(111) CuKα1 radiation ($\lambda=1.5406 \text{ \AA}$, step size 0.050, step time 366 s and primary slit width 0.2 mm).

X-ray photoelectron spectroscopy (XPS or ESCA) was performed on a Perkin-Elmer PHI 5000C spectrometer equipped with a pre-treatment reactor cell. The runs were performed under a base pressure in the analysis chamber of $1 \times 10^{-8} \text{ Pa}$, in a monochromatic mode with MgKα radiation at 187.85 eV. A small

amount of mesoporous material was placed onto a tape with adhesives on both sides. The information depth was approximately 4-5 nm. The surface composition is given as atomic percentage of the elements.

Supplementary Material (ESI) for Green Chemistry

This journal is (c) The Royal Society of Chemistry 2010