Hardin and Sarpong, Supporting Information 1

Supplemental material for:

Remarkable Facilitation of Hetero-Cycloisomerizations with Water: Metal-Free Synthesis of Indolizines

Alison R. Hardin Narayan and Richmond Sarpong*

Department of Chemistry, University of California, Berkeley, California 94720

E-mail: rsarpong@berkeley.edu

Table of Contents

General	S2
Materials and Methods	S2
Representative Procedure	S2
Additional Studies 1. Temperature studies 2. Deuterium incorporation	S4 S5
1H NMR Spectra of Products	S6- S11

General

The majority of the compounds reported herein were prepared previously in our laboratory. For detailed procedures describing the synthesis of substrates as well as full characterization data for both starting materials and heterocyclic products, see C. R. Smith, E. M. Bunnelle, A. J. Rhodes, R. Sarpong, *Org. Lett.* **2007**, *9*, 1169-1171.

Materials and Methods

Unless otherwise stated, reactions were performed in 4 mL glass vials sealed with Teflon® lined caps. The water used in all reactions was deionized and distilled using a Milli-Q® purification system. Reaction temperatures above 23 °C were controlled by an IKA[®] temperature modulator. Reactions were monitored by thin layer chromatography using Sorbent Technologies silica gel XHL precoated plates (0.25 mm), which were visualized using UV irradiation or anisaldehyde stain. SiliCycle Silica-P silica gel (particle size 40-63 µm) was used for flash chromatography. ¹H NMR were recorded on Bruker AVB-400, AV-500 or DRX-500 MHz spectrometers with ¹³C operating frequencies of 100, 125 and 125 MHz, respectively, in benzene-*d*₆ at 23 °C.

Representative Procedures for the Formation of Indolizines

Scheme S1 The hetero-cycloisomerization of propargylic ester 4 to indolizine 5.



A heterogeneous mixture of pyridine propargylic ester **4** (54.7 mg, 0.200 mmol) and distilled deionized water (200 μ L) were heated at 100 °C for 2 h at which point TLC analysis indicated that the starting material had been consumed. The reaction mixture was cooled to 23 °C and the solvent removed to provide the crude product, indolizine **5** as a brown oil (52.0 mg, 95% yield). The crude product was >90% purity by ¹H NMR. ¹H NMR (500 MHz, C₆D₆) δ 7.30 (d, J = 9.01 Hz, 1H), 7.04 (d, J= 7.10 Hz, 1H), 6.76 (s,

1H), 6.32 (dd, J = 8.99, 6.42 Hz, 1H), 6.06 (t, J = 6.76 Hz, 1H), 2.26 (t, J= 7.65 Hz, 2H), 1.40-1.32 (m, 2H), 1.27 (s, 9H), 1.18-1.10 (m, 2H), 0.74 (t, J = 7.34 Hz, 3H); ¹³C NMR (125 MHz, C₆D₆) δ 175.7, 127.1, 121.3, 120.9, 120.7, 116.1, 114.0, 109.6, 105.0, 38.9, 29.0, 27.0, 25.2, 22.4, 13.6; IR (film) v_{max} 2958, 2931, 2871, 1749, 1278, 1120, 728 cm⁻¹; HRMS (EI) calcd for [C₁₇H₂₃NO₂]⁺: m/z 273.1729, found 273.1732.

Hardin and Sarpong, Supporting Information 4

Additional Studies

1. Temperature studies

In order to maximize product formation on an acceptable time scale (~18 h) using 4, it was important to heat the reaction mixture to 100 °C (Table S1). No product was detected when 4 was vigorously stirred with water at room temperature (entry 4). However, indolizine product 5 was detected in reactions conducted at both 50 °C and 75 °C (entries 3 and 2, respectively) although these reactions failed to reach completion after 18 h.

OPiv OPiv н H₂O (1mL/mmol 4), `*n*-Bu temp. 'n-Bu 5 Δ temp (°C) time (h) result (5:4) entry 1 2 >95:5 100 2 75 18 7:1 3 50 18 1:15 23 4 18 <5:95

Table S1 Temperature study for the hetero-cycloisomerization of 4.

2. Deuterium incorporation study

When the cycloisomerization of **4** was carried out in D_2O , deuterium incorporation was observed at the 2 position of the indolizine product (see, **10** in Scheme S2). ¹H NMR clearly indicates 45% deuterium incorporation at C2 (see attached spectrum). In an effort to determine if deuterium incorporation occurs during the cycloisomerization of **4** or after the indolizine product had formed, indolizine **5** (Scheme S3) in D_2O was heated to 100 °C for 2 h and then worked up according to the representative procedure. No deuterium incorporation was observed as judged by ¹H NMR, suggesting that deuterium incorporation takes place prior to indolizine formation.

Scheme S2 Hetero-cycloisomerization conducted in D₂O.



Scheme S3 Indolizine 5 in D₂O.













