



**Fig. S1** NH<sub>3</sub>-DRIFTS spectra of ZrPO<sub>x</sub> after adsorption of ammonia at 373 K followed by degassing under He flow at various temperatures. Shown is (A) the O-H/N-H stretching region (4000- 2000 cm<sup>-1</sup>) and (B) the region 4000-1350 cm<sup>-1</sup> including the HNH bending region.

The Research Octane Number (RON) of the gasoline range products were estimated according to method introduced in the Automotive Fuel Reference Book by making a weighted average using measured values for each of the components (listed in Table S1), as shown below for a three-component blend<sup>1</sup>.

$$PB = V1P1 + V2P2 + V3P3$$

Where PB is the property (such as RON, MON) of the blend, P1, P2, P3 are the properties of the components and V1, V2, V3 are the volume fractions.

**Table S1.** Physical properties and research octane number of different gasoline range products.

	Molar mass (g mol <sup>-1</sup> )	Boiling point (K)	Density (g mL <sup>-1</sup> )	Research octane number (RON)	Water solubility (g L <sup>-1</sup> )
Pentane	72.15	309	0.626	62 <sup>2</sup>	0.1
Hexane	86.18	342	0.655	26 <sup>2</sup>	0.01
Ethanol	40.07	351	0.789	130 <sup>3</sup>	Miscible
Propanol	60.1	370	0.803	118 <sup>3</sup>	Miscible
Acetone	58.08	330	0.793	118 <sup>3</sup>	Miscible
Butanol	74.12	390	0.810	96 <sup>3</sup>	77
Butanone	72.11	353	0.805	96 <sup>3</sup>	290
Tetrahydrofuran	72.11	339	0.889	73 <sup>4</sup>	301
Butanoic acid	88.11	435	0.860	96 <sup>3</sup>	Miscible
Pentanol	88.15	409	0.814	84 <sup>3</sup>	27
Pantanone	86.13	376	0.809	84 <sup>3</sup>	0.04
2-Methyl-tetrahydro-furan	86.13	352	0.86	86 <sup>5</sup>	140
Tetrahydropyran	86.13	361	0.880	73 <sup>4</sup>	80
Pentanoic acid	102.13	459	0.930	97 <sup>3</sup>	50
Tetrahydrofurfuryl alcohol	102.13	451	1.054	108 <sup>4</sup>	Miscible
Hexanol	102.17	431	0.814	56 <sup>6</sup>	6
Hexanone	100.16	400	0.811	56 <sup>6</sup>	14
2,5-Dimethyl-tetrahydrofuran	100.16	364	0.833	82 <sup>6</sup>	Miscible
2-Methyl-tetrahydropyran	100.16	352	0.863	87 <sup>5</sup>	39
Hexanoic acid	116.16	475	0.920	56 <sup>6</sup>	11
Tetrahydropyran alcohol	116.16	460	1.027	108 <sup>4</sup>	Miscible

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**Table S2** Hydrogenation of mono-, di- and polysaccharides.

Saccharide	Catalyst	Conditions	Conversion	Yield	Ref.
glucose	Ru	T=393 K, total pressure = 120 bar, glucose concentration: 40 wt % glucose in aqueous solution (120 mL), reaction time 5 h, stirring at 850 rpm	~100% conversion	~100% selectivity sorbitol	1
glucose	Ru/C	5%Ru/C catalyst was studied in a semi-batch slurry autoclave, 373–403K 4.0–7.5MPa H <sub>2</sub>	100% >300min	~100% selective to sorbitol	2
glucose	Ru/MWNT	5wt% Ru 393 K; time 120 min; initial H <sub>2</sub> pressure 4 MPa	62.5%		3
fructose	CuO-ZnO	35–65 bar and between 90 and 130°C	~100% conv. and selectivity after 250min	mannitol	4
Glucose/fructose	Raney Ni	85 to 130°C, 50 to 1000 psi		Fructose not hydrogenated	5
Glucose/fructose	Rh				5
Glucose/fructose	Ru			Most effective	5
xylose	Raney Ni	90-100°C, 40-60bar, 50-60wt% xylose	200min	>90% xylitol	6
maltose	Ru-P	maltose aqueous solution (40 wt%, 50 mL), T = 363 K, PH <sub>2</sub> = 2.0 MPa, reaction time = 2 h, stirring rate = 1200 rpm	98%	100% selectivity maltitol	7
lactose	Ni sponge,	300 mL, Parr Co., operating at 20-70 bar and between 110 and 130 °C	100% 200min	90-99% selectivity lactitol, small amounts of lactobionic acid, lactulose, lactulitol, sorbitol, and galactitol	8
lactose	Ni sponge , Ru/C	120°C, 20-70 Bar H <sub>2</sub>	~100% after 150 min Ni 120 min Ru/C	Lactitol ~100%	9
sucrose	Raney Ni	3 h of reaction at 130 °C and 750 psia 1wt% cat, 40 g/L sucrose	~60% 3h		10
sucrose	Ru/Al <sub>2</sub> O <sub>3</sub>	3 h of reaction at 130°C and 750 psia 1wt% cat, 40 g/L sucrose	~100% 3h	100%, Sorbitol, mannitol	10
sucrose	Ru/NaY, Ru/USY Ru/CaY	135°C~ and 12 atm H <sub>2</sub> 2.5h	100% after 3h	Sorbitol, mannitol	11
Inulin	Ru/C(activated)	1% Ru:C, 80 mL aqueous solution of 1 g inulin and 0.100 g 1% Ru:SX1GNS, 100 °C, 100 bar H <sub>2</sub> .	100% 300min	inulin to D-mannitol and D-glucitol	12
cellulose	Ru/CNT	cellulose (0.16 g, equivalent to 1 mmol C <sub>6</sub> H <sub>10</sub> O <sub>5</sub> unit) and the catalyst (0.050 g) with H <sub>2</sub> O (20 mL H <sub>2</sub> of 5 MPa, 185°C, for 24 h with stirring		69 sorbitol 4.0 mannitol 5 Erythritol 5 Glycerol	13
Cellulose	Ru/C	245°C, 4 wt% Ru/C, 6 MPa H <sub>2</sub>	85.8% 30min	34.6 sorbitol, 11.4mannitol, 13.4 sorbitan	14
cellobiose	Pd	120 °C,	Pd pH 2 100%	100% glucose	15
	Rh	40 bar H <sub>2</sub> 12 h	Rh pH2 100%	6.9 sorb 66.9 glu	
	Pt	cellobiose (7.31 mmol), using H <sub>2</sub> O (30 g)	Pt pH2 100%	18.5 sorb 42.6 glu	
	Ru	HCl to adjust pH	Ru pH2 100%	100% sorbitol	
	Ru		Ru pH7 87.8%	26.4 sorb 1.6 glu 64.8 alc	
	Ru/C		Ru pH10 75. c	7.2 other	
			Ru/C pH7 100%	24.0 sorb 3.2 glu 55.7 alc	
				17.1 other	
				>99% alc	
cellulose	Ru/C	170°C, 1wt% H <sub>2</sub> SO <sub>4</sub>	95% 4h		16,17
biomass	Ru/C	15% load, <190°C for 3-5h, 0.8% H <sub>3</sub> PO <sub>4</sub>	>99%	~100%	16,17
		5wt% Ru/C, 10 wt% cat/biomass			

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