

Supporting Information

Microwave Assisted Conversion of Carbohydrates and Biopolymers to 5-Hydroxymethylfurfural with Aluminum Chloride Catalyst in Water

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Table S1 Optimization of reaction time for dehydration of 30 wt% of fructose catalyzed by AlCl₃ in biphasic water-MIBK (1/4 volume ratio).

Entry	t (min)	HMF Yield (%)
1	30	27.7
2	60	32.8
3	75	33.3
4	90	33.9

Table S2 Results of fructose dehydration to HMF catalyzed by AlCl₃ in different reaction time.

Entry	Fructose (wt%)	AlCl ₃ (mol%)	t (min)	HMF Yield (%)
1.	5	50	0.5	47.8
2.	5	50	2	52.3
3.	5	50	5	53.9
4.	5	50	10	54.6
5.	5	50	15	55.1
6.	5	50	20	55.7

Solvent = water (2 mL), T = 120 °C, MW

Figure S1 Representative ^1H NMR spectrum of HMF product obtained from dehydration reaction of fructose with AlCl_3 in water. This spectrum recorded in CDCl_3 at room temperature for determining HMF yield using mesitylene as an internal standard.

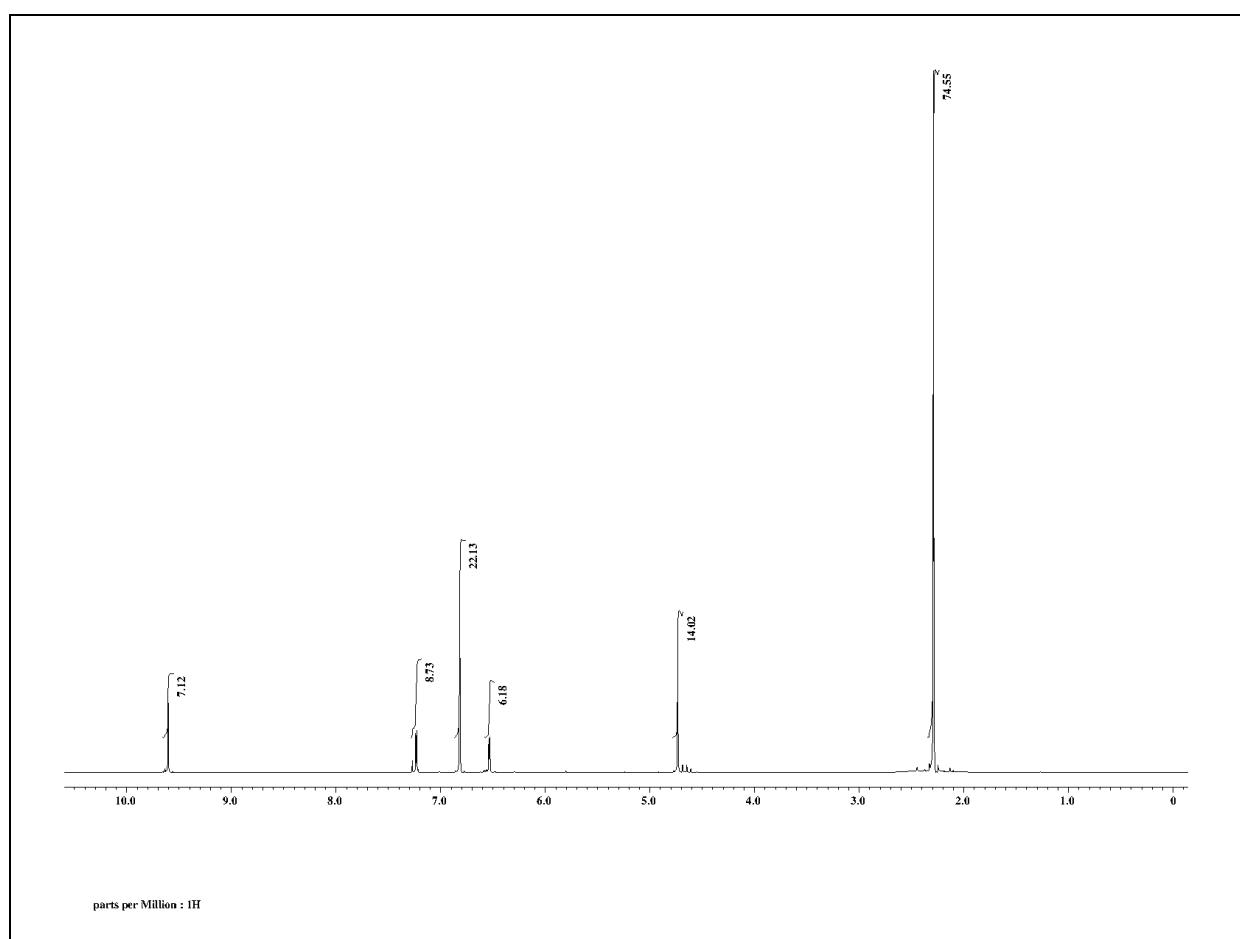


Figure S2 Representative UV-Vis spectrum of HMF product solution obtained from dehydration reaction of fructose with AlCl₃ in water.

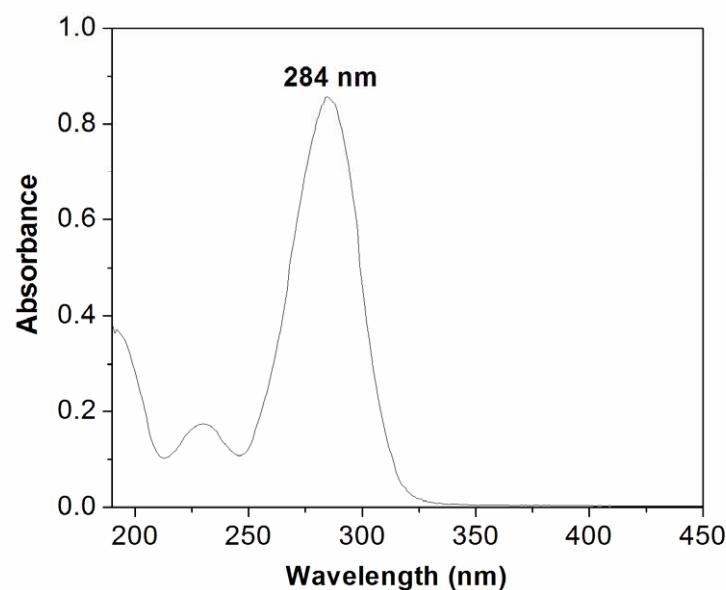


Figure S3 Representative ^1H NMR (in CDCl_3) spectrum of the reaction product collected from MIBK phase of fructose (30 wt%) dehydration reaction with 50 mol% AlCl_3 catalyst in biphasic water-MIBK solvent under microwave heating at $T = 120^\circ\text{C}$ for 5 min.

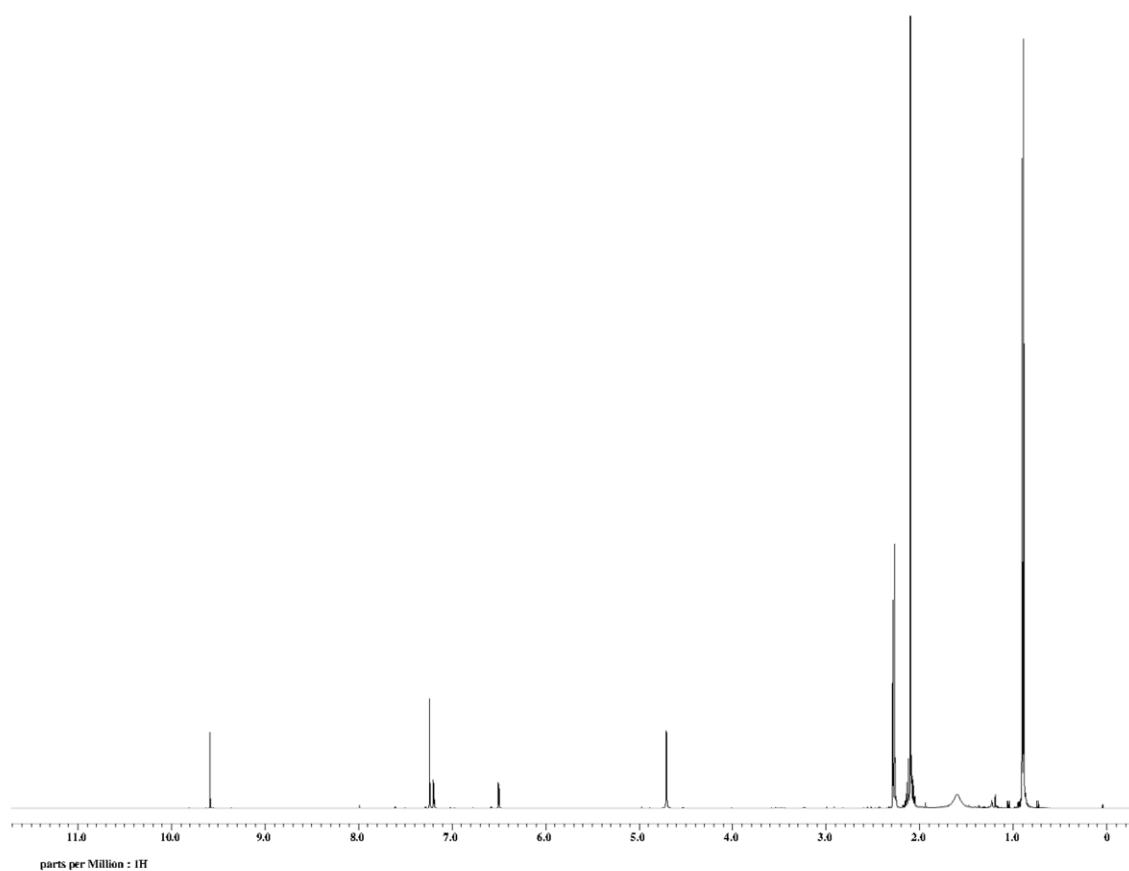


Figure S4 Representative ^1H NMR ($\text{DMSO}-d_6$) spectrum of the insoluble brown material obtained in fructose (30 wt%) dehydration with AlCl_3 in water under microwave heating at 120 °C for 5 min.

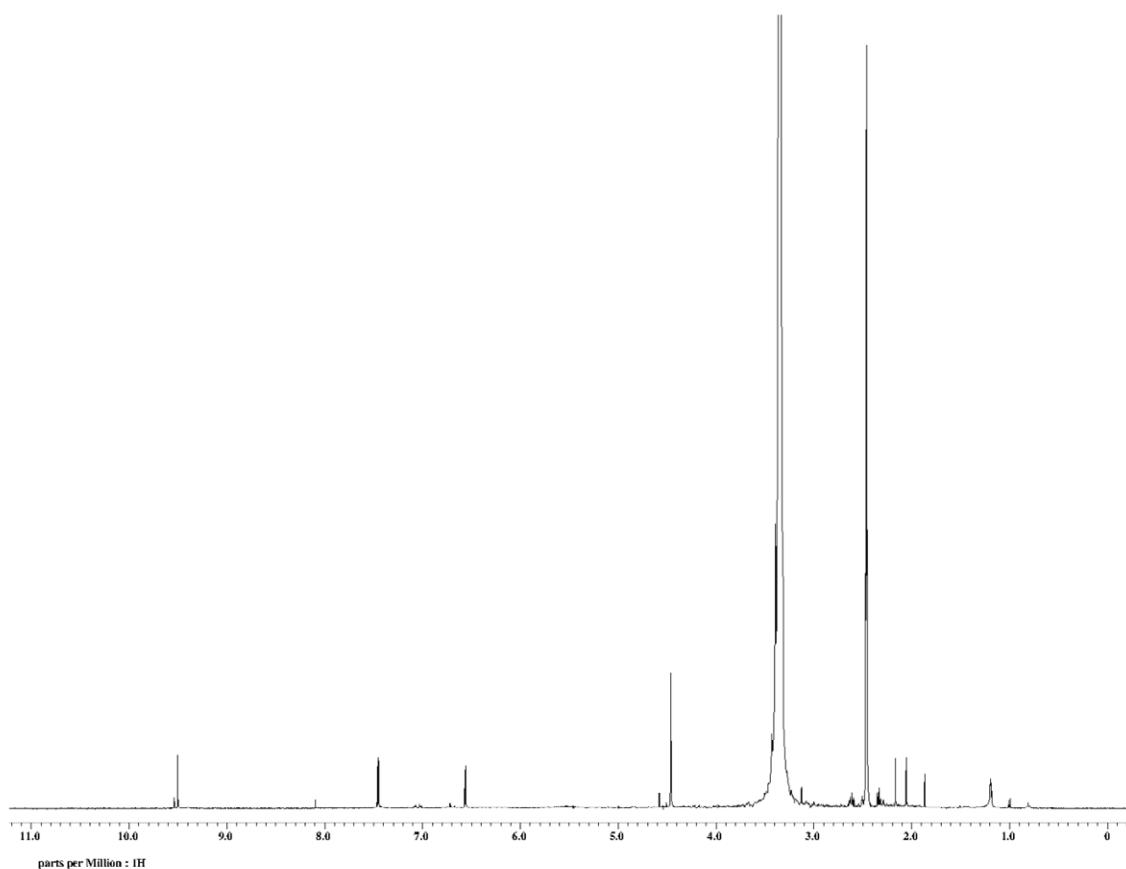


Figure S5 Representative ^1H NMR spectrum for a mixture of glucose and AlCl_3 in $\text{DMSO}-d_6$ at room temperature to 100 °C; (a) ^1H NMR spectrum of glucose (25 mg) at room temperature in $\text{DMSO}-d_6$. (b) ^1H NMR spectrum of glucose (25 mg) + AlCl_3 (9.3 mg) (2:1 molar ratio) at room temperature displaying broad $H_{\text{O}-\text{H}}$ signals of glucose due to the interaction of Cl of AlCl_3 with OH of glucose in $\text{DMSO}-d_6$. (c) ^1H NMR spectrum of glucose at 100 °C heated for 40 min. (d) ^1H NMR spectrum of glucose + AlCl_3 (2:1 molar ratio) at 100 °C heated for 40 min. Data presented for δ 4 to 5 ppm with full spectrums (δ 1 to 10 ppm) as inset.

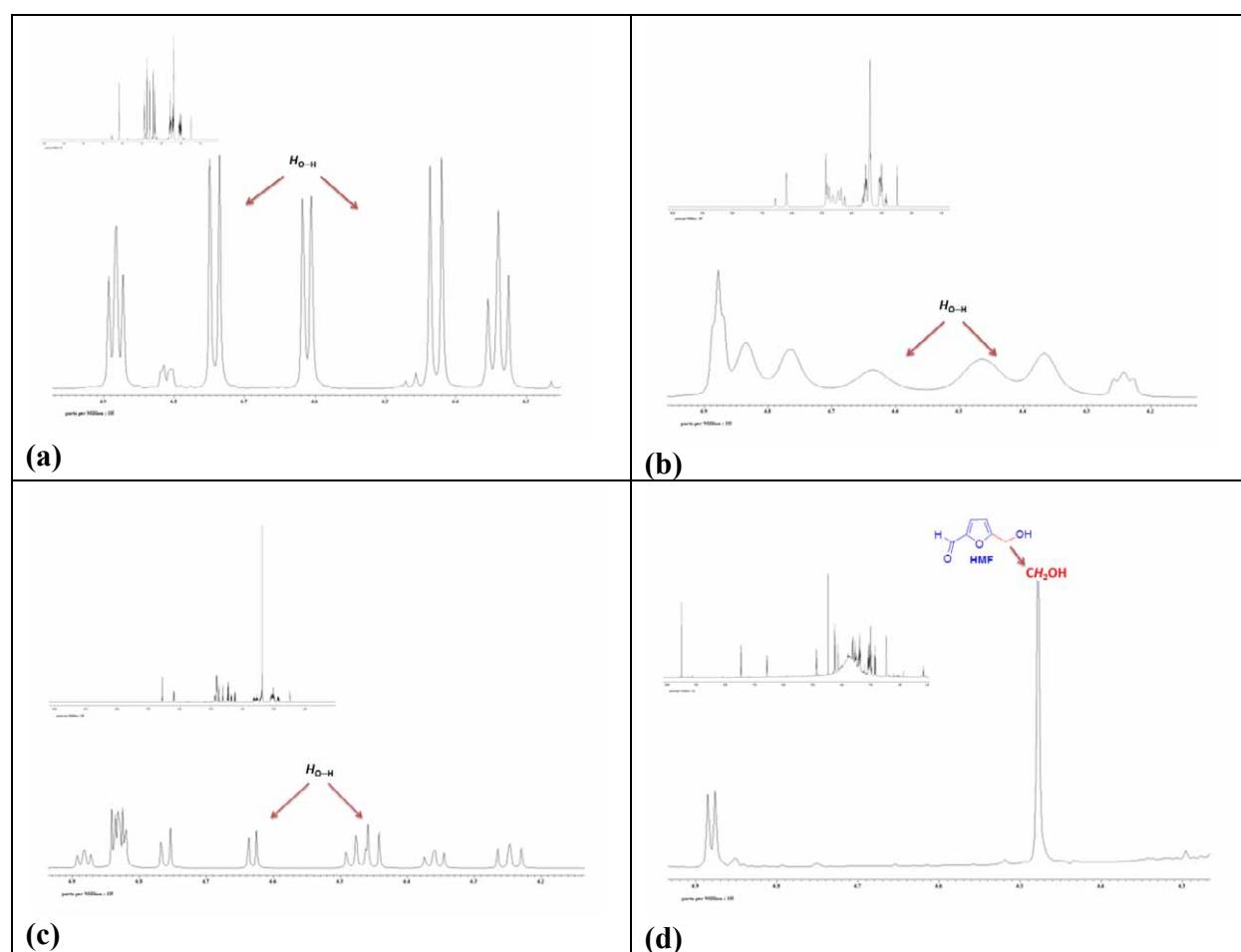


Figure S6 Representative ^1H NMR spectrum for a mixture of glucose and SnCl_4 in $\text{DMSO}-d_6$ at room temperature to 120 °C; (a) ^1H NMR spectrum of glucose (25 mg) at room temperature in $\text{DMSO}-d_6$. (b) ^1H NMR spectrum of glucose (25 mg) + SnCl_4 (9 mg) (4:1 molar ratio) at room temperature displaying disappearance of $H_{\text{O-H}}$ signals of glucose due to the interaction of SnCl_4 with OH of glucose in $\text{DMSO}-d_6$. (c) ^1H NMR spectrum of glucose + SnCl_4 (2:1 molar ratio) at 80 °C heated for 40 min. (d) ^1H NMR spectrum of glucose + SnCl_4 (2:1 molar ratio) at 120 °C heated for 30 min. Full spectrums (δ 0 to 11 ppm) for each steps presented as inset.

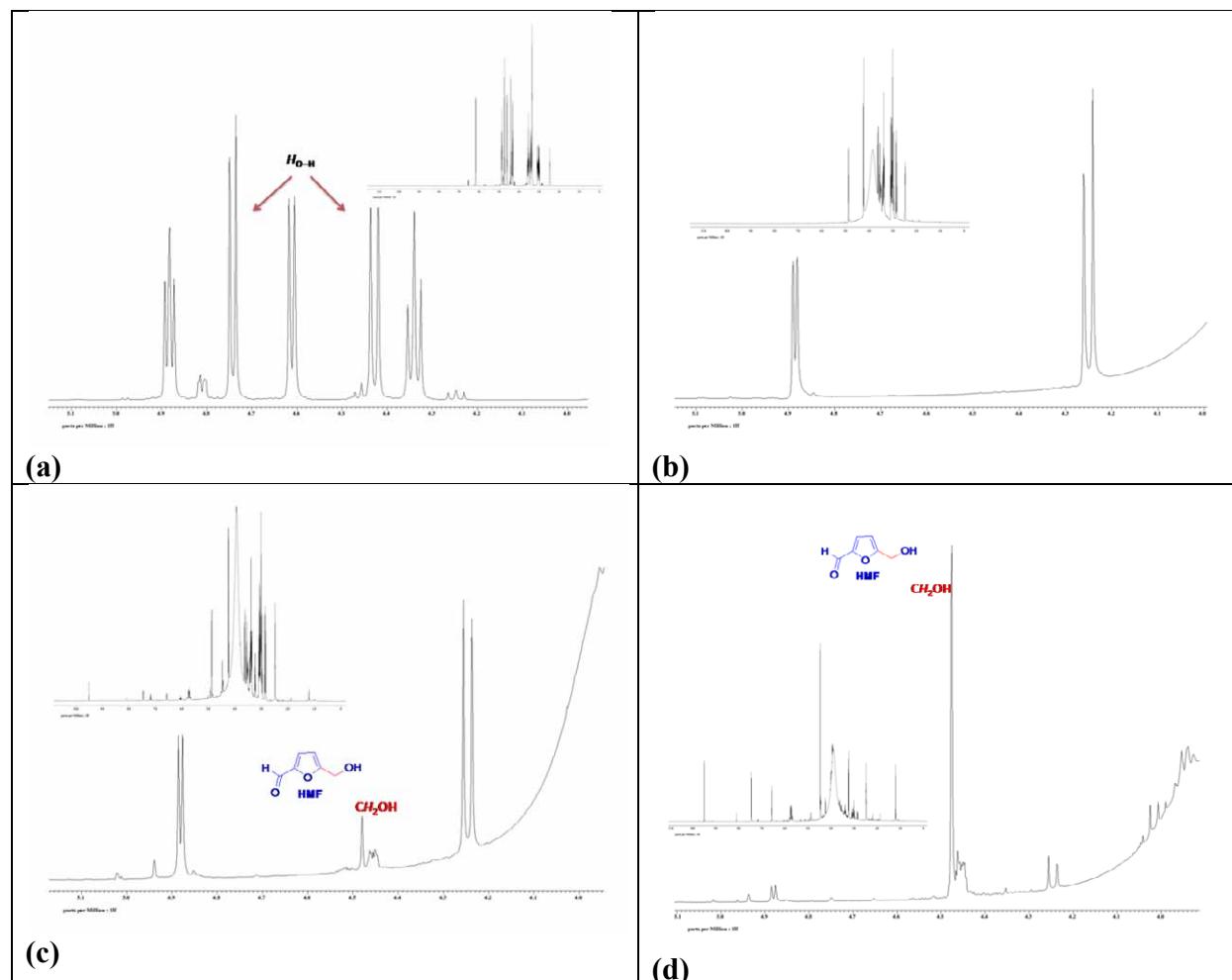


Figure S7 Representative ^1H NMR spectrum for a mixture of glucose and $\text{Yb}(\text{OTf})_3$ in $\text{DMSO}-d_6$ at room temperature to 120 °C; (a) ^1H NMR spectrum of glucose (25 mg) at room temperature in $\text{DMSO}-d_6$. (b) ^1H NMR spectrum of glucose (25 mg) + $\text{Yb}(\text{OTf})_3$ (9.3 mg) (2:1 molar ratio) at 120 °C heated for 40 min in $\text{DMSO}-d_6$. Full spectrums (δ 0 to 11 ppm) for each steps presented as inset.

