

Supporting Information

Selective Reduction of Aldehydes and Ketones to Alcohols with Ammonia Borane in Neat Water

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I. General Remarks:

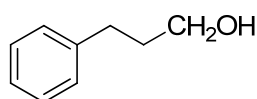
All reagents were purchased from commercial sources and used without further treatment, unless otherwise indicated. Ammonia borane was prepared according to the known methods.¹⁻² The nucleoside **1e** was obtained from prof. Xueseng Fan lab in Henan Normal University. ¹H NMR spectra were recorded on a Bruker Avance/400 and TMS as internal standard. Data are represented as follows: chemical shift, integration, multiplicity (br = broad, s = singlet, d = doublet, dd = double doublet, t = triplet, q = quartet, m = multiplet), coupling constants in Hertz (Hz). Melting points were measured on a YuHua X-5 apparatus.. All reactions were monitored by TLC with GF254 silica gel coated plates. Flash column chromatography was carried out using 200-300 mesh silica gel at increased pressure.

Preparation of Ammonia Borane (AB) using Sodium Borohydride and Ammonium Sulfate in THF: A mixture of sodium borohydride (1.00 g, 26.3 mmol) and powdered dry ammonium sulfate (3.48 g, 26.3 mmol) in THF (156 mL) was vigorously stirred at 40 °C under nitrogen. The reaction was monitored by ¹¹B NMR spectroscopy. Upon completion, the reaction mixture was cooled to room temperature and filtered. The filtrate was concentrated under vacuum to produce AB (690 mg, 85%) in >96% chemical purity, as determined by hydride analysis.² The solvent was recovered and reused for the preparation of subsequent batches of AB without further purification. The product AB was obtained as a white powder: mp. 108.9-110 °C (110 °C in literature³); ¹H NMR (CDCl₃, 400 MHz) δ : 3.40 (t, *J*_{N-H} = 48 Hz, 3H, NH₃), 1.62 (q, *J*_{B-H} = 96 Hz, 3H, BH₃); ¹H NMR (DMSO-*d*₆, 400 MHz) δ : 4.42 (br., 3H, NH₃), 1.90-0.50 (m, 3H, BH₃).

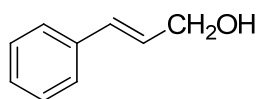
Caution: A large amount of hydrogen is generated during the preparation of AB, which is a potential fire hazard. The reaction must be carried out in a well-ventilated hood and that the reaction vessel outlet be led directly into the hood exhaust.

References:

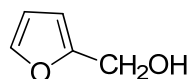
1. (S. G. Shore, X. Chen, "Methods for Synthesizing Ammonia Borane": WO 2007120511 A2 2007; US 20090104102 A1 2009).
2. P. V. Ramachandran and P. D. Gagare, *Inorg. Chem.*, **2007**, 46, 7810-7817.
3. D. J. Heldebrant, A. Karkamkar, J. C. Linehan and T. Autrey, *Energy. Envir. Sci.*, **2008**, 1, 156-160.



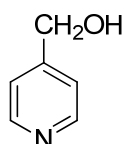
3-Phenyl propanol (2a): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.18-7.30 (m, 5H), 3.64-3.67 (t, J = 6.4 Hz, 2H), 2.68-2.72 (t, J = 7.6 Hz, 2H), 1.85-1.92 (m, 2H), 1.75 (s, 1H)



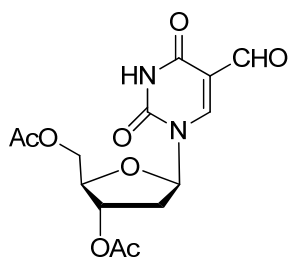
Cinnamyl alcohol (2b): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.40-7.42 (d, J = 8 Hz, 2H), 7.32-7.36 (m, 2H), 7.23-7.29 (m, 1H), 6.62-6.66 (d, J = 16 Hz, 1H), 6.36-6.42 (m, 1H), 4.34-4.35 (ds, J = 1.2 Hz, 4.4 Hz, 2H) 1.69 (br, 1H)



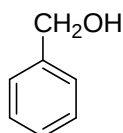
(Furan-2-yl)methanol (2c): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.39 (t, J = 0.8 Hz, 1H), 6.32-6.33 (m, 1H), 6.26-6.27 (d, J = 3.2 Hz, 1H), 4.56 (s, 2H), 2.61 (s, 1H).



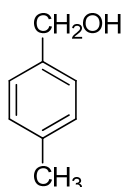
(Pyridine-4-yl)methanol (2d): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 8.41-8.42 (d, J = 6.0 Hz, 2H) 7.30-7.31 (d, J = 5.6 Hz, 2H), 4.72 (s, 2H), 2.48-2.51 (t, J = 6.4 Hz, 1H)



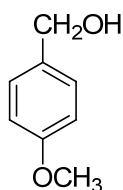
3',4'-O-Diacetyl-2'-deoxy-5-formyl-uridine (2e): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 9.69 (s, 1H), 7.58 (s, 1H) 6.36-6.41 (m, 1H), 5.31 (s, 1H), 4.27-4.40 (m, 4H), 2.48-2.53 (m, 1H), 2.11-2.14 (m, 7H).



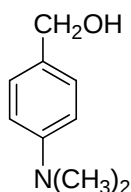
Benzyl alcohol (2f): Colorless liquid, ^1H NMR (DMSO- d_6 , 400 MHz) δ : 7.32-7.20 (m, 5H), 5.20-5.18 (t, J = 5.6 Hz, 1H), 4.51-4.49 (d, J = 8.0 Hz, 2H).



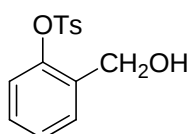
4-Methylbenzyl alcohol (2g): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.22-7.24 (d, J = 8.0 Hz, 2H), 7.15-7.17 (d, J = 8.0 Hz, 2H), 4.61 (s, 2H), 2.44-2.47 (t, J = 6.0 Hz, 1H), 2.34 (s, 3H).



4-Methoxybenzyl alcohol (2h): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.26-7.28 (d, J = 8.4 Hz, 2H), 6.87-6.89 (d, J = 8.4 Hz, 2H), 4.59 (s, 2H), 3.80 (s, 3H), 1.96 (s, 1H).

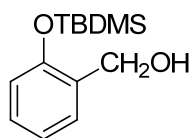


4-N,N-Dimethylbenzyl alcohol (2i): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.07-7.04 (d, J = 8.4 Hz, 2H), 6.70-6.67 (d, J = 8.4 Hz, 2H), 3.81 (s, 1H), 2.90 (s, 6H), 1.56 (s, 2H).

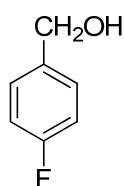


4-Tosyloxyl benzyl alcohol (2j): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.77-7.79 (d, J = 8.0 Hz, 2H), 7.51-7.53 (d, J = 7.6 Hz, 1H), 7.19-7.38 (m, 4H), 6.84-6.86 (d, J = 8.0 Hz,

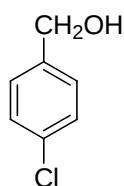
1H), 4.59-4.60 (d, J = 5.2 Hz, 2H), 2.48 (s, 3H), 2.29-2.32 (m, 1H).



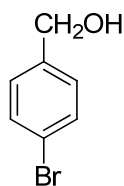
(2-((tert-Butyldimethylsilyl)oxy)benzyl alcohol (2k): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 6.80-7.31 (m, 4H), 4.66-4.68 (d, J = 6.4 Hz, 2H), 2.09-2.12 (m, 1H), 1.01 (s, 9H), 0.26 (s, 6H)



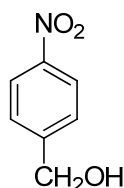
4-Fluorobenzyl alcohol (2l): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.30-7.33 (m, 2H), 7.01-7.05 (m, 2H), 4.63 (s, 2H), 2.07 (s, 1H).



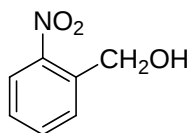
4-Chlorobenzyl alcohol (2m): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.30-7.36 (m, J = 8.8 Hz, 4H), 4.67 (s, 2H), 1.73-1.70 (t, J = 5.6 Hz, 1H).



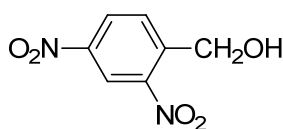
4-Bromobenzyl alcohol (2n): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.51-7.49 (d, J = 8.0 Hz, 2H), 7.27-7.26 (d, J = 8.8 Hz, 4H), 4.68 (s, 2H), 1.27 (s, 1H).



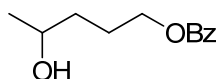
4-Nitrobenzyl alcohol (2o): Straw yellow solid, ^1H NMR (DMSO, 400 MHz) δ : 8.22-8.20(d, $J=8.8\text{Hz}$, 2H), 7.60-7.58(d, $J=8.4\text{Hz}$, 2H), 5.56-5.53(t, 1H), 4.65-4.63(d,, $J=5.6\text{Hz}$, 2H).



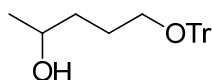
2-Nitrobenzyl alcohol (2p): White solid, ^1H NMR (DMSO, 400 MHz) δ : 8.05-8.03(d, $J=8.4\text{Hz}$, 1H), 7.84-7.74(m, 2H), 7.55-7.51(t, 1H), 5.56-5.53(t, 1H), 4.83-4.81(d, $J=5.6\text{Hz}$, 2H).



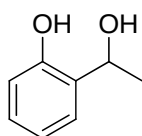
2, 4-Dinitrobenzyl alcohol (2q): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 8.97-8.98 (d, $J = 1.2\text{ Hz}$, 1H), 8.52-8.55 (dd, $J = 1.6\text{ Hz}$, 6.8 Hz, 1H), 8.14-8.17 (d, $J = 8.8\text{ Hz}$, 1H), 5.19 (s, 2H), 1.61-1.66 (br, 1H).



1-O-Benzoyl-1,4-pentanediol (2r): White solid, ^1H NMR (DMSO, 400 MHz) δ :8.01-8.03(m, 2H), 7.51-7.56(m,1H), 5.28(s, 1H), 7.40-7.44(m, 2H), 4.32-4.35(t, $J = 6.8\text{ Hz}$, 2H), 3.84-3.89(m, 1H), 1.78-1.87(m,2H), 1.56-1.65(m, 2H), 1.21-1.23(d, $J = 6.0\text{Hz}$, 2H).

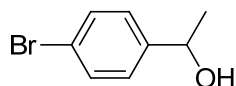


1-O-Trityl-1,4-pentanediol (2s): White solid, ^1H NMR (DMSO, 400 MHz) δ :7.20-7.45(m, 15H), 5.27-5.28(d, $J = 2.0\text{ Hz}$, 1H), 3.76-3.80(m, 1H), 3.09-3.12(m, 2H), 1.64-1.75(m, 2H), 1.52-1.54(m, 2H), 1.10-1.18(dd, $J = 2.8\text{ Hz}$, 6.0 Hz, 3H).

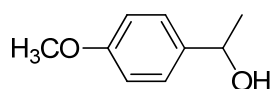


2-(1-Hydroxyethyl)phenol (2t): White solid, ^1H NMR (CDCl_3 , 400 MHz) δ : 6.81-7.18 (m,

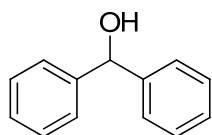
5H), 5.02-5.07 (dd, $J = 6.4$ Hz, 6.8 Hz, 1H), 2.04 (s, 1H), 1.56-1.58 (d, $J = 6.8$ Hz, 3H), 1.42-1.43 (br, 1H).



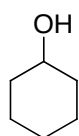
1-(4-Bromophenyl) ethanol (2u): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.44-7.47 (d, $J = 8.4$ Hz, 2H), 7.22-7.24 (d, $J = 8.4$ Hz, 2H), 4.81-4.86 (q, $J = 6.4$ Hz, 1H), 2.13-2.18 (br, 1H), 1.44-1.46 (d, $J = 6.4$ Hz, 3H).



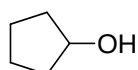
1-(4-Methoxyphenyl)ethanol (2v): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.29-7.31 (d, $J = 8.8$ Hz, 2H), 6.88-6.90 (d, $J = 8.4$ Hz, 2H), 4.85-4.88 (t, $J = 6.4$ Hz, 1H), 3.81 (s, 3H), 1.71 (s, 1H).



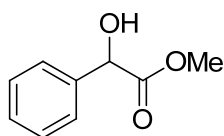
Diphenyl methanol (2W): Colorless liquid, 86% yield, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.25-7.39 (m, 10H), 5.84-5.85 (d, $J = 3.2$ Hz, 1H), 2.24-2.25 (d, $J = 3.6$ Hz, 1H).



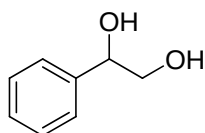
Cyclohexanol (2x): Colourless liquid, ^1H NMR (400 MHz, CDCl_3): 3.61-3.59 (m, 1H), 1.88-1.75 (m, 3H), 1.73-1.69 (m, 2H), 1.53-1.50 (m, 1H), 1.30-1.14 (m, 5H).



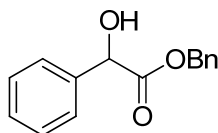
Cyclopentanol (2y): Colourless liquid, ^1H NMR (400 MHz, CDCl_3): δ 4.33-4.32 (m, 1H), 1.77 (s, 5H), 1.56 (s, 4H).



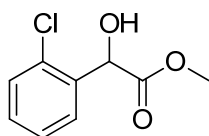
Methyl mandelate (2A): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz) δ : 7.31-7.43 (m, 5H), 5.17-5.19 (d, $J = 5.2$ Hz, 1H), 3.76 (s, 3H), 3.40-3.42 (d, $J = 5.2$ Hz, 1H)



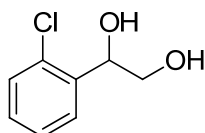
1-Phenyl glycol (3A) ^1H NMR ($\text{DMSO}-d_6$, 400 MHz): 7.34-7.20 (m, 5H), 5.17 (d, $J = 3.6$ Hz, 1H), 4.67 (t, $J = 5.6$ Hz, 1H), 4.53 (q, $J = 5.6$ Hz, 1H), 3.43 (t, $J = 6.0$ Hz, 2H).



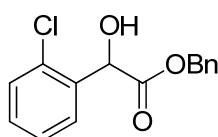
Benzyl mandelate (2B): ^1H NMR (CDCl_3 , 400 MHz): 7.37-7.30 (m, 10H), 5.03-6.01 (m, 2H), 4.92-4.89 (m, 1H), 5.15 (s, 2H).



Methyl 2-chloromandlate (2C): ^1H NMR ($\text{DMSO}-d_6$, 400 MHz): 7.51-7.36 (m, 4H), 6.38-6.36 (d, $J = 8.0$ Hz, 1H), 5.43-5.41 (d, $J = 8.0$ Hz, 1H), 3.62 (s, 3H).

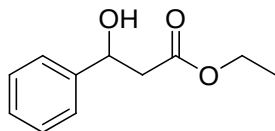


2-Chlorophenyl glycol (3C): ^1H NMR ($\text{DMSO}-d_6$, 400 MHz): 7.57-7.56 (d, $J = 7.2$ Hz, 1H), 7.38-7.24 (m, 3H), 5.48 (s, 1H), 4.93-4.88 (m, 2H), 3.53-3.47 (m, 1H), 3.34-3.28 (m, 1H).

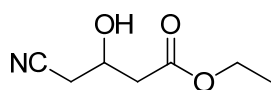


Phenyl 2-chloromandlate (2D): ^1H NMR ($\text{DMSO}-d_6$, 400 MHz): 7.56-7.25 (m, 9H),

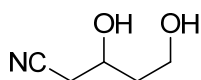
6.47-6.46 (d, $J = 5.6$ Hz, 1H), 5.51-5.23 (d, $J = 6.0$ Hz, 1H), 5.15 (s, 2H).



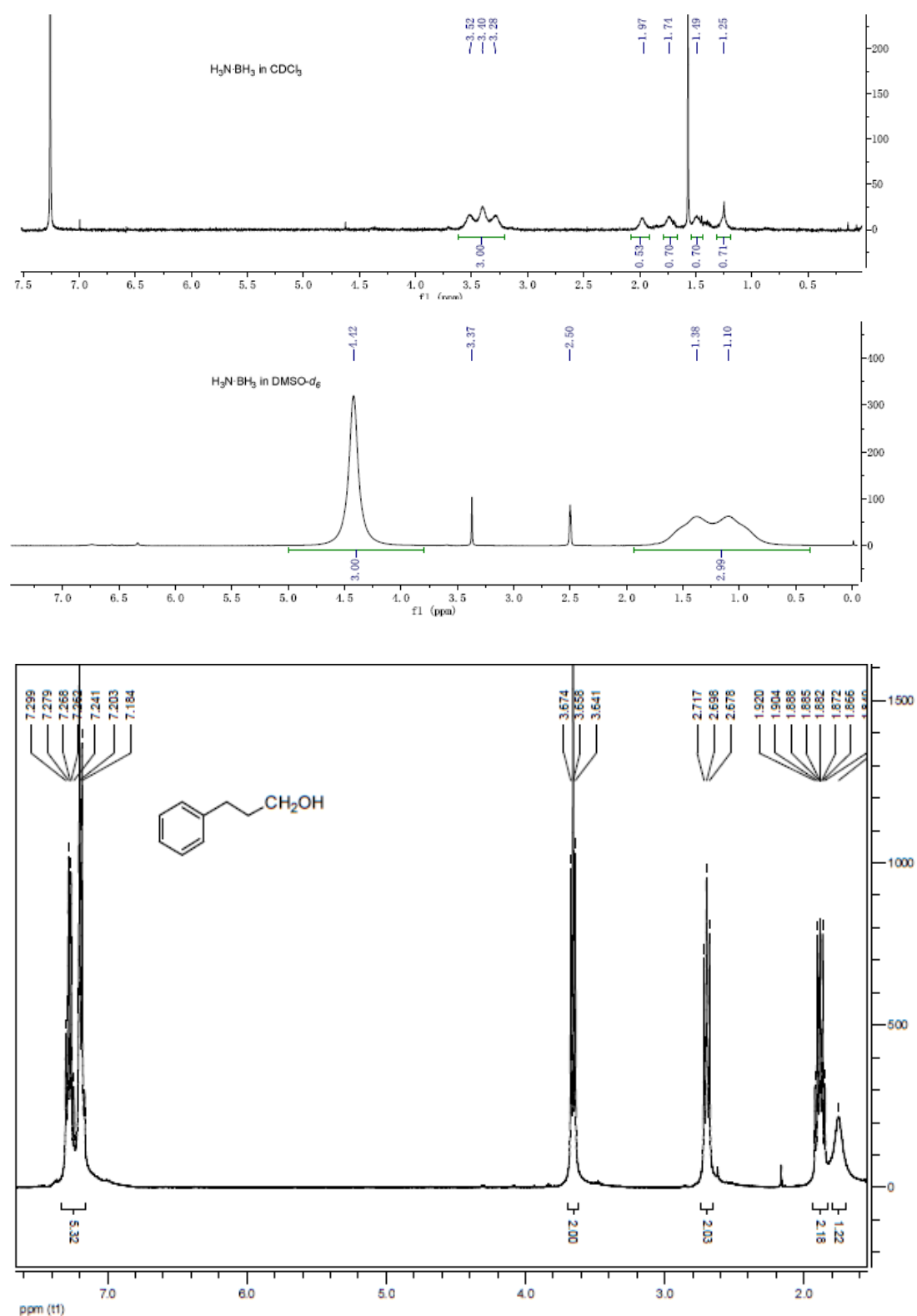
Ethyl 3-hydroxy-3-phenylpropionate (2E): Colorless liquid, ^1H NMR (CDCl_3 , 400 MHz)
 δ : 7.27-7.39 (m, 5H), 5.12-5.15 (m, 1H), 4.15-4.21 (q, $J = 7.2$ Hz, 2H), 3.32 (s, 1H),
2.67-2.79 (m, 2H), 1.24-1.28 (t, $J = 7.2$ Hz, 3H).

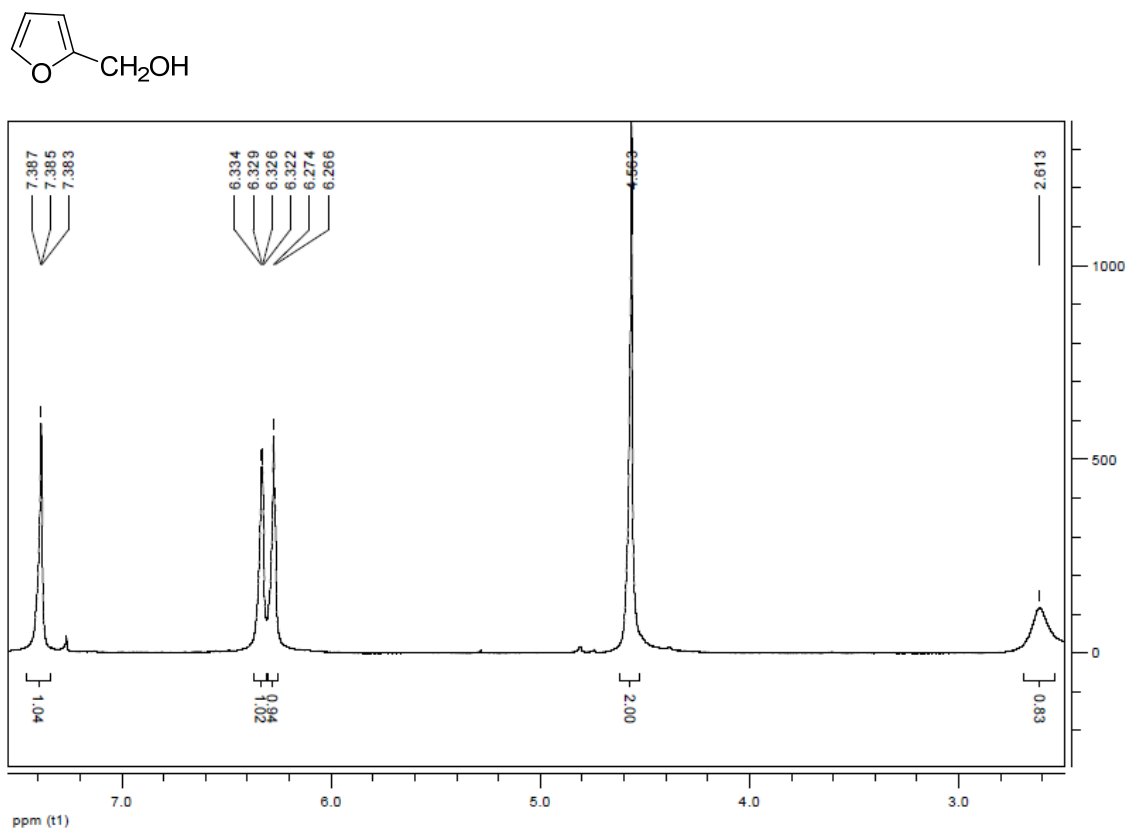
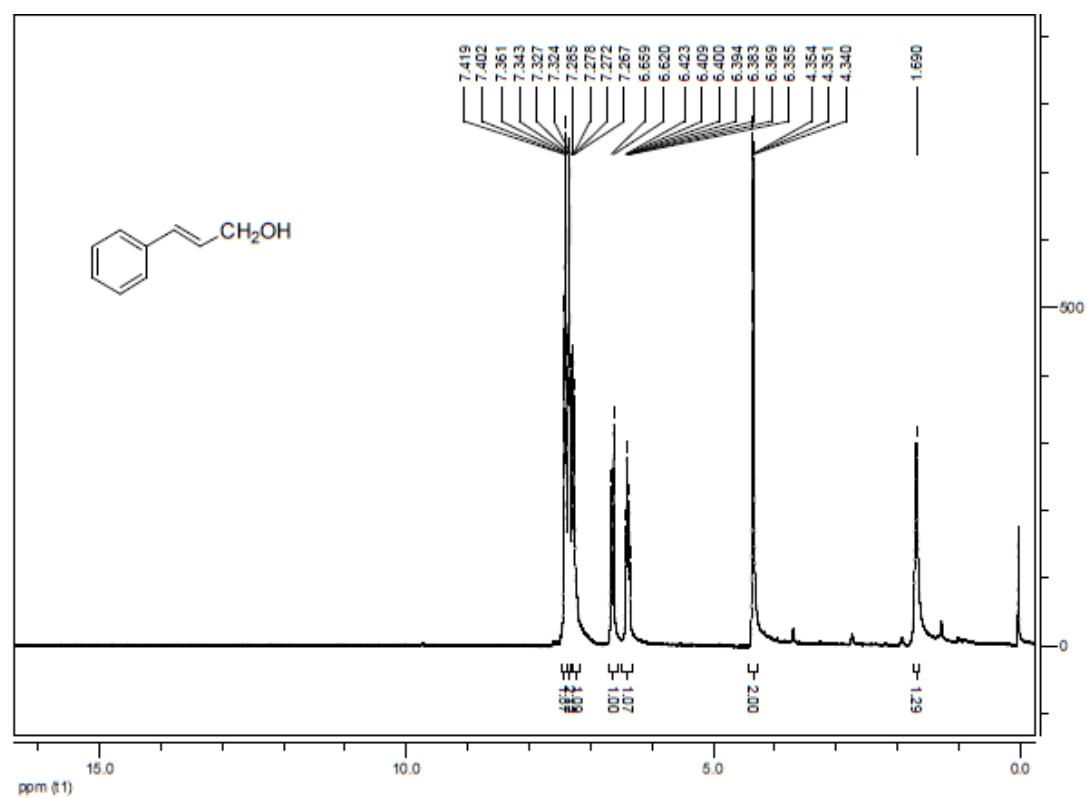


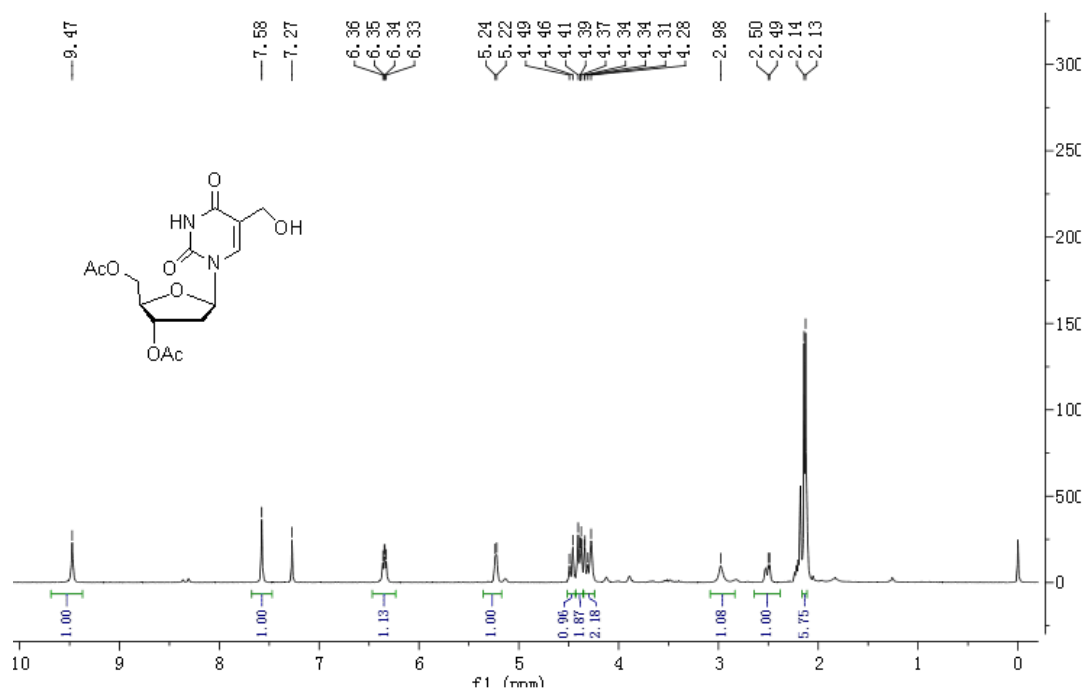
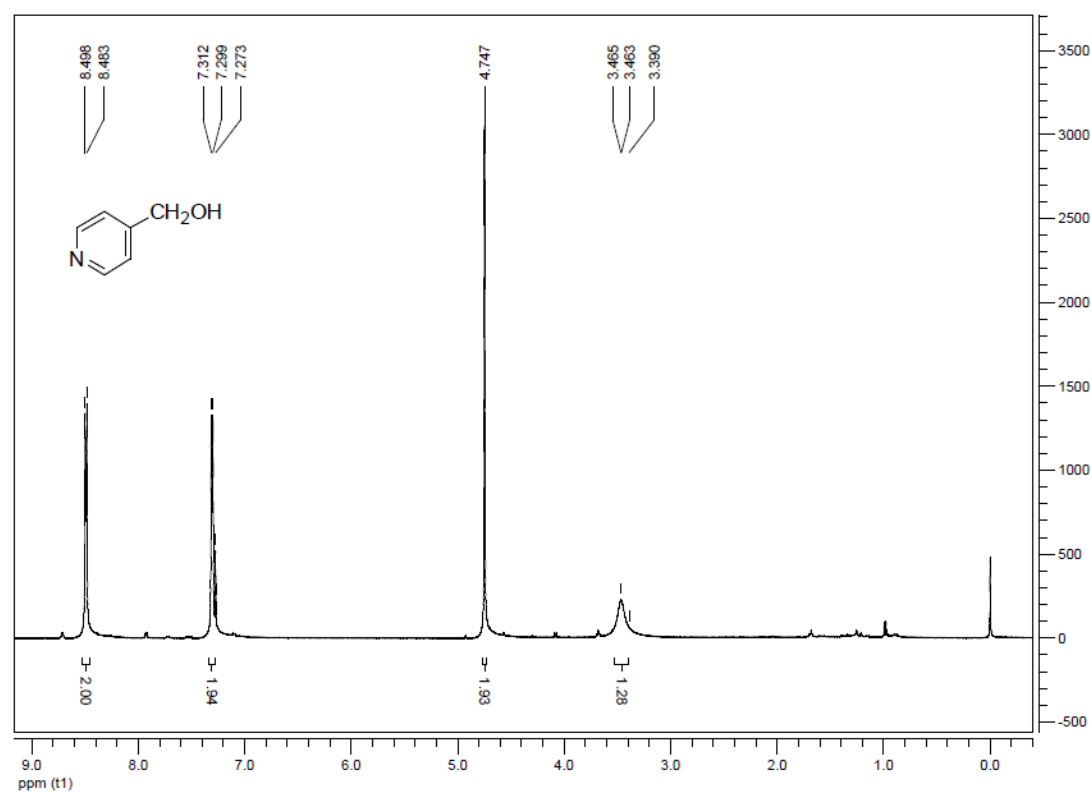
Ethyl 3-cyano-3-hydroxypropionate (2F): ^1H NMR (CDCl_3 , 400 MHz): 4.39-4.31 (m, 1H),
4.22-4.17 (q, $J = 7.2$ Hz, 2H), 3.64-3.63 (d, $J = 4.0$ Hz, 1H), 2.69-2.58 (m, 4H), 1.30-1.26 (t, J
 $= 7.2$ Hz, 3H).

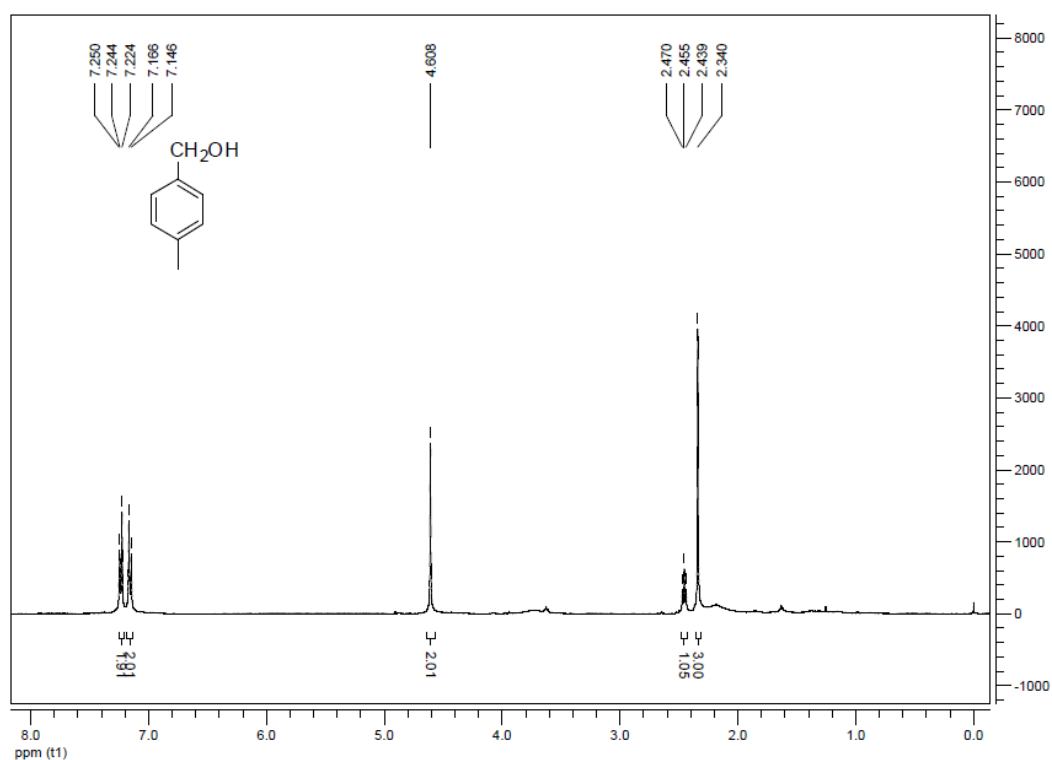
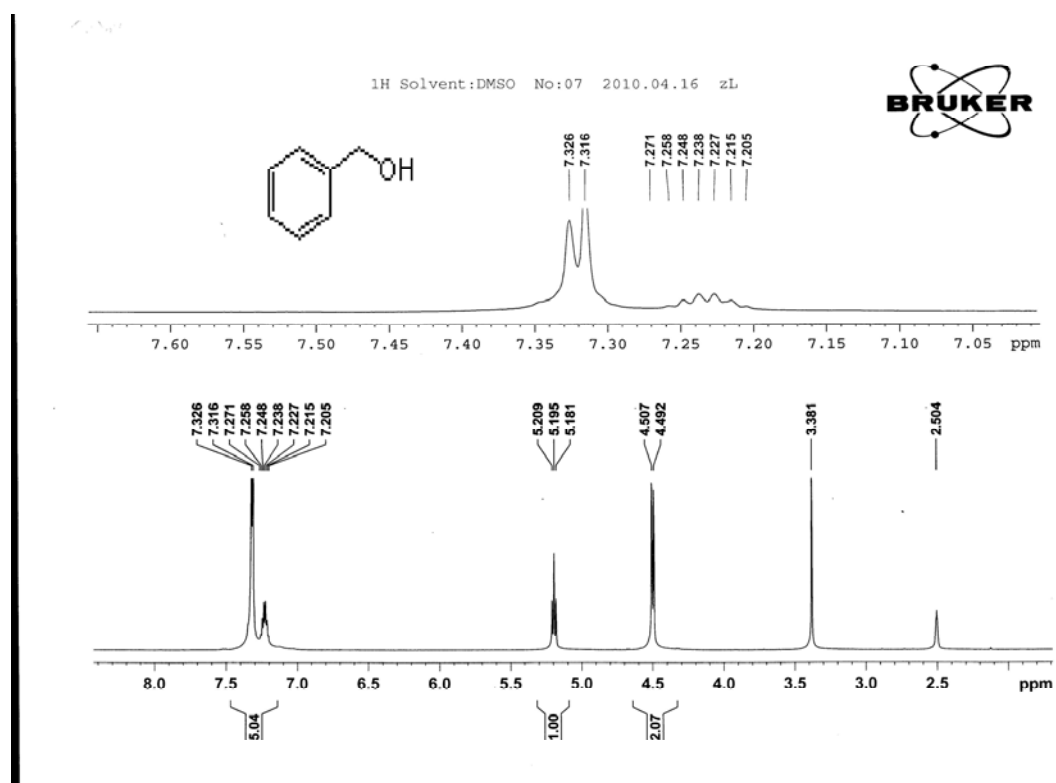


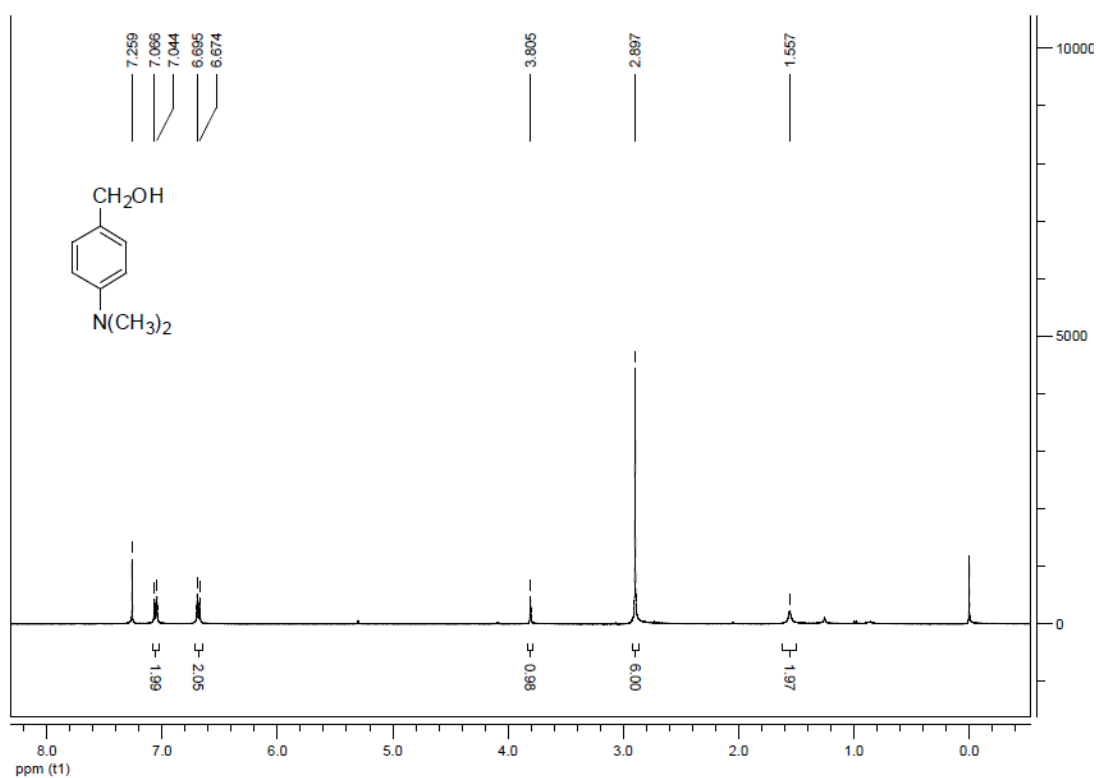
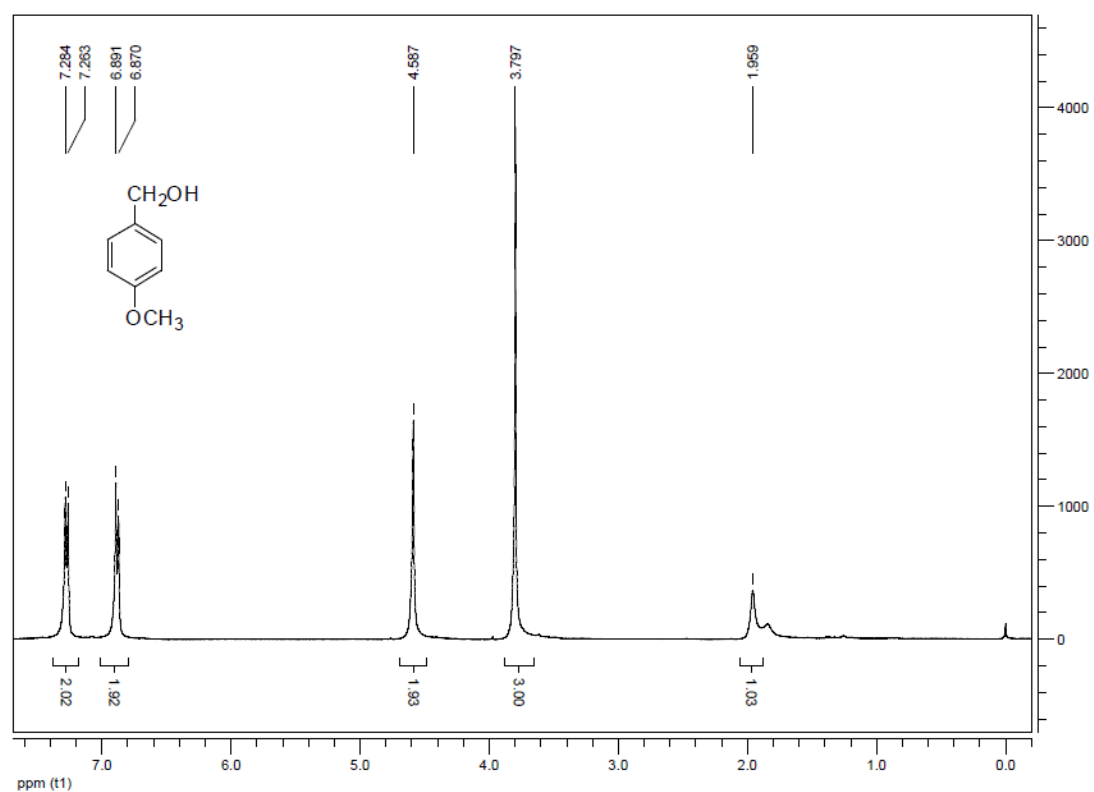
4-Cyano-butane-1,3-diol (3F): ^1H NMR (CDCl_3 , 400 MHz): 4.27-4.21 (m, 1H), 4.02-3.89
(m, 2H), 3.51 (br, 1H), 2.61-2.60 (d, $J = 5.6$ Hz, 2H), 1.94-1.84 (m, 2H), 0.89 (m, 1H).

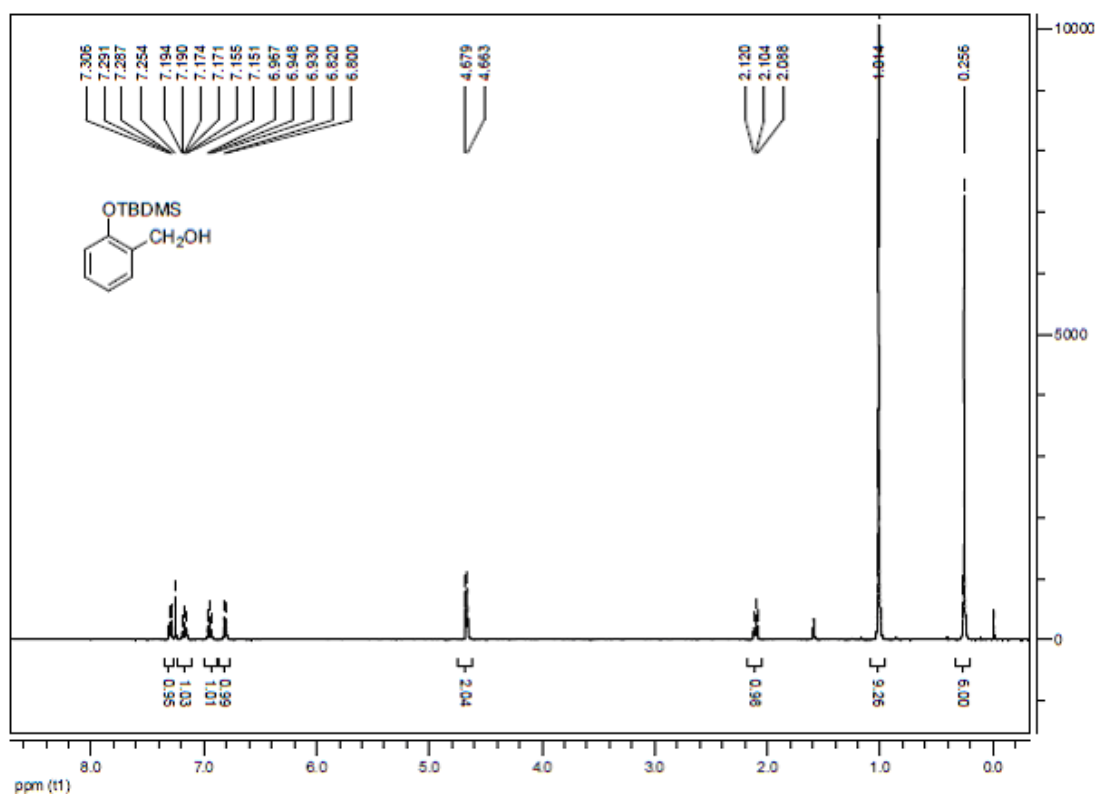
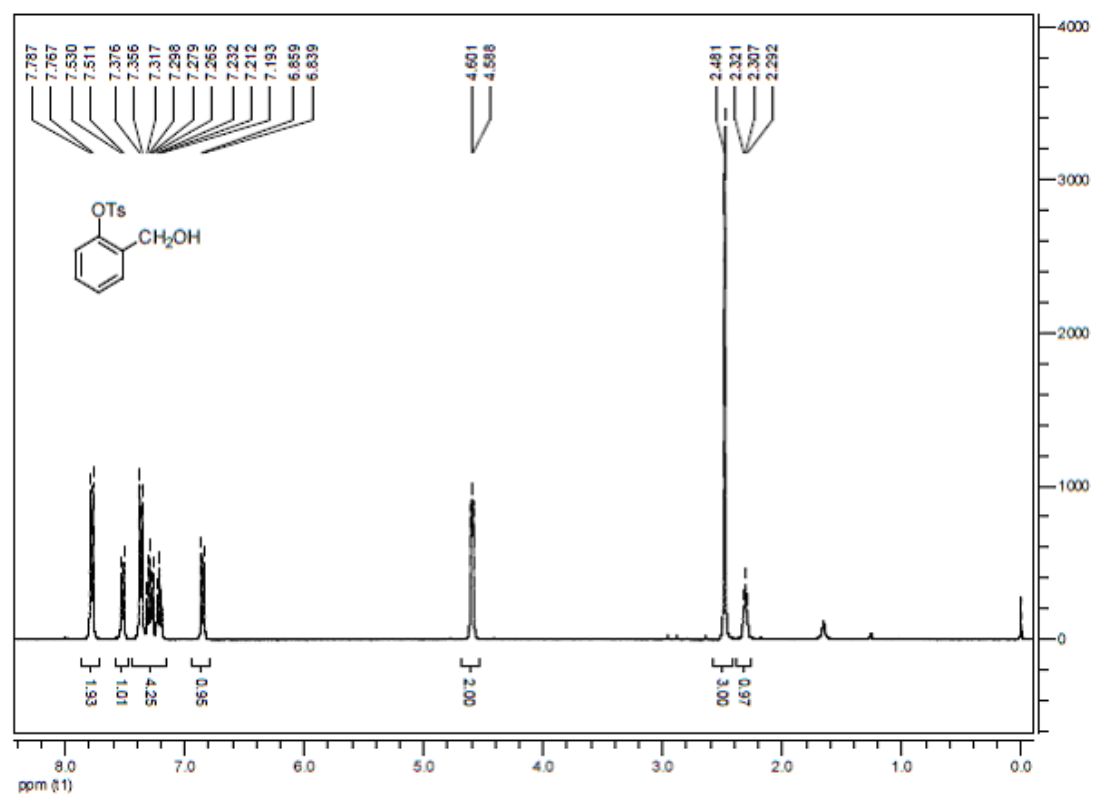


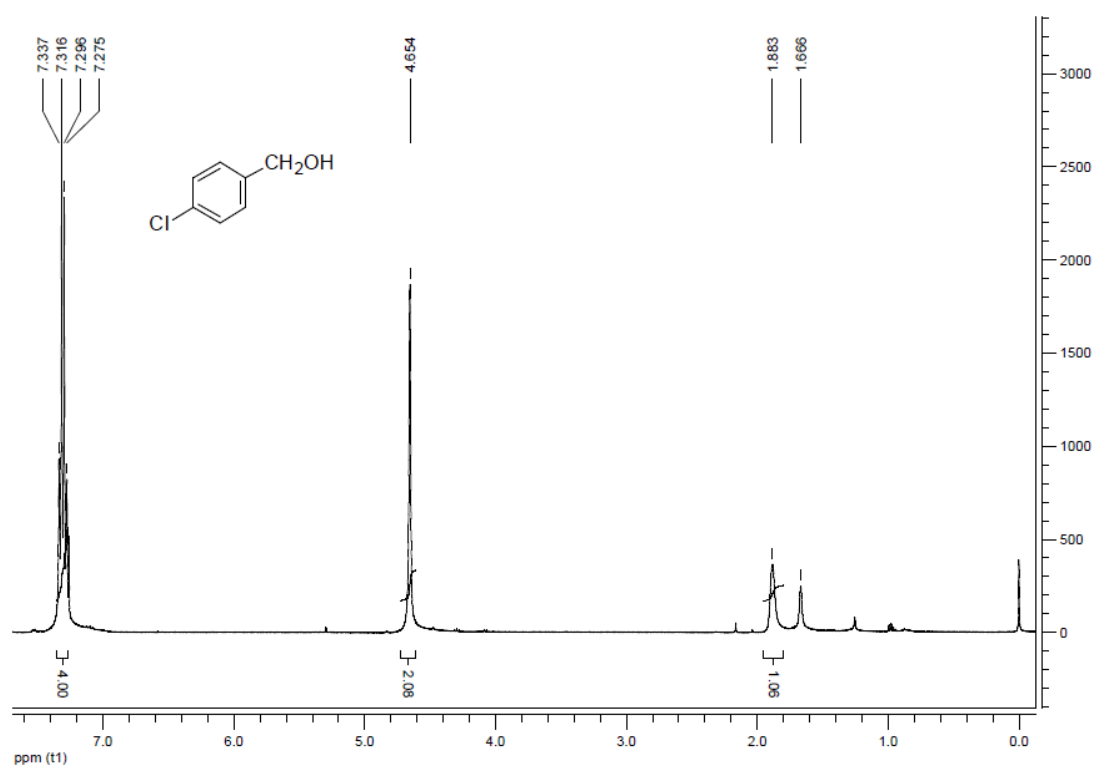
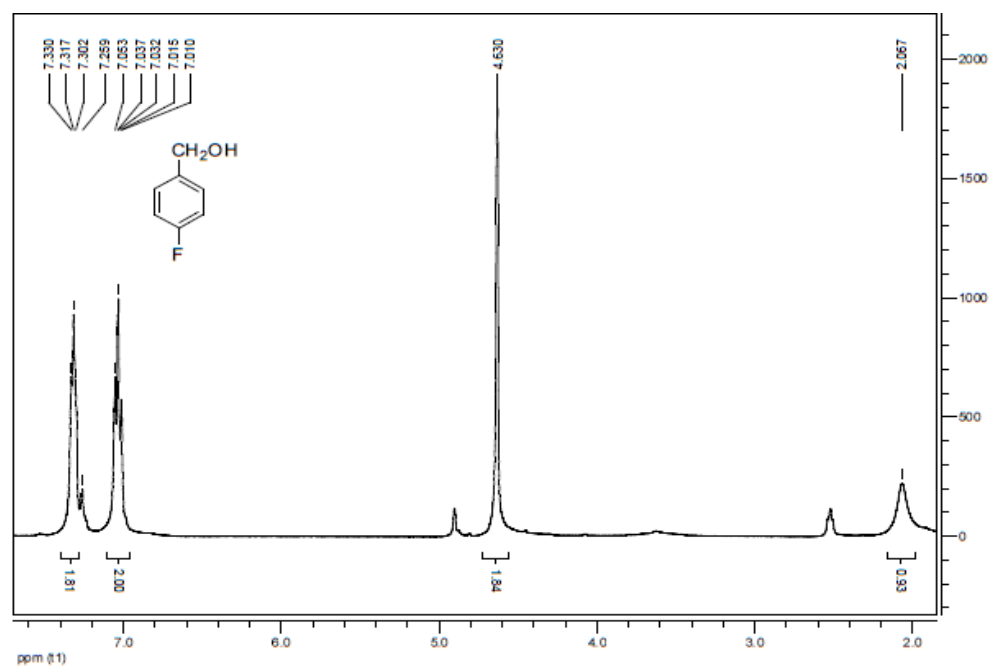


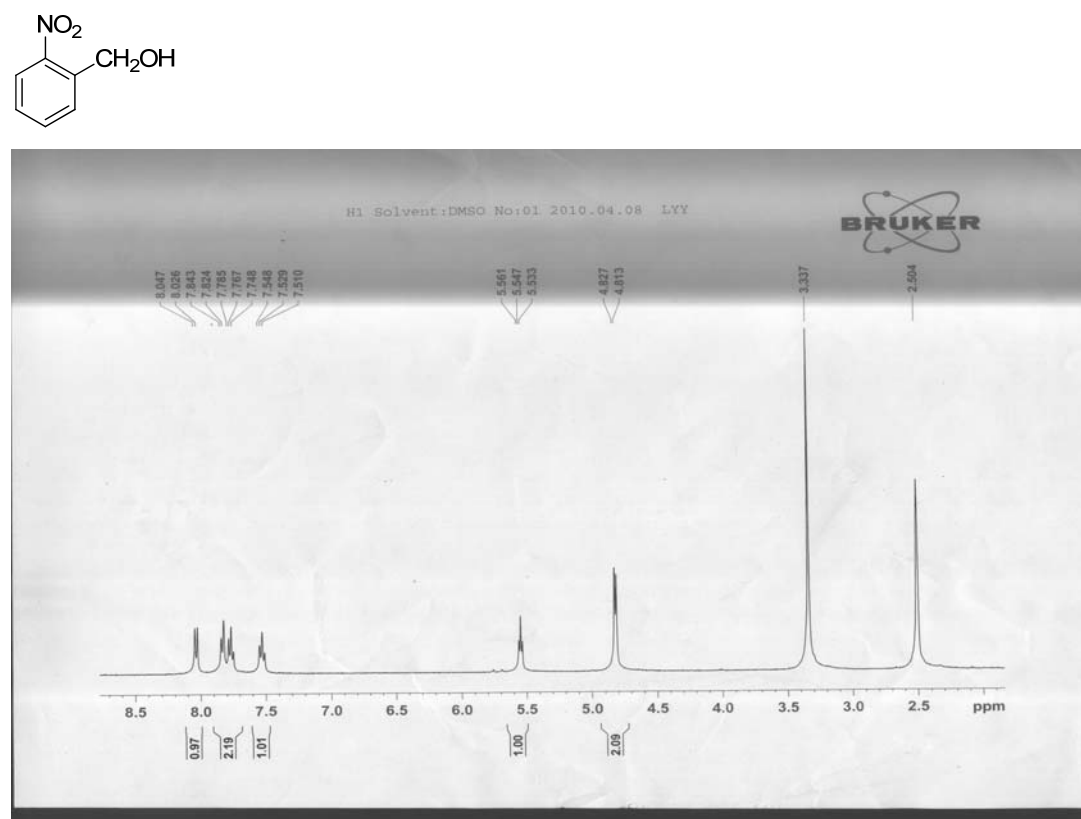
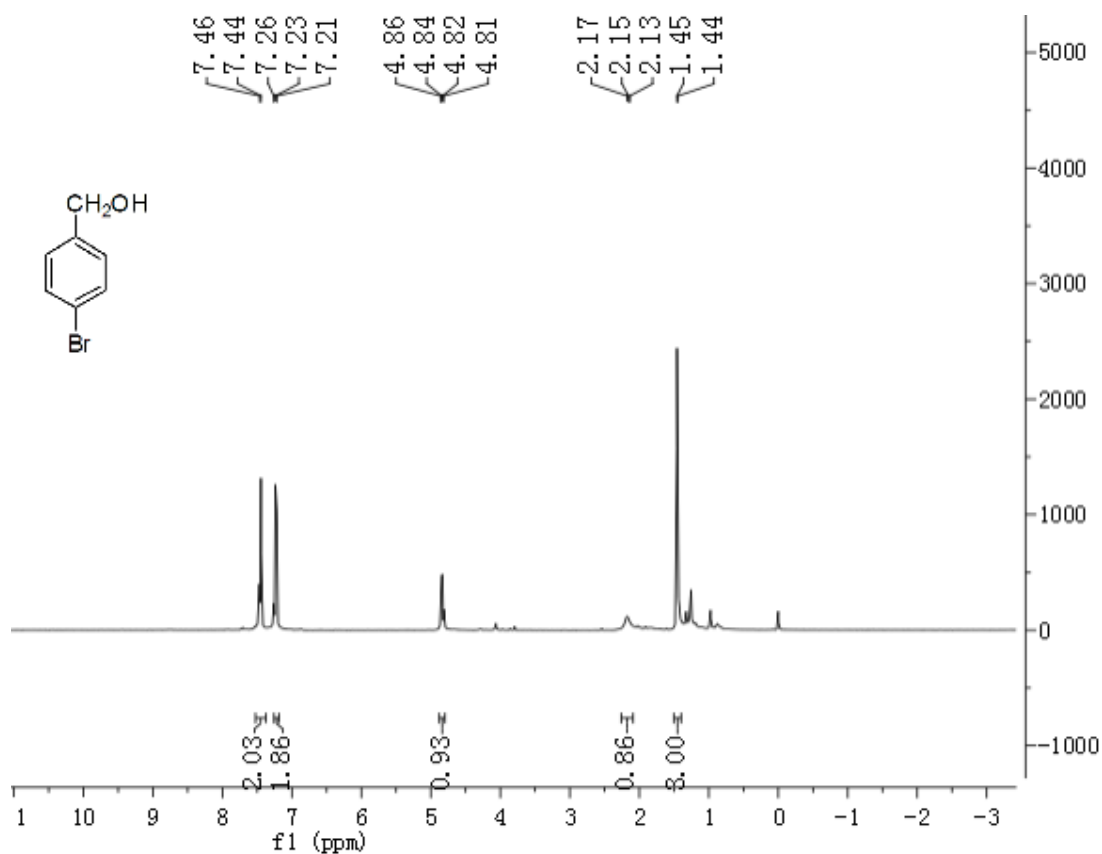


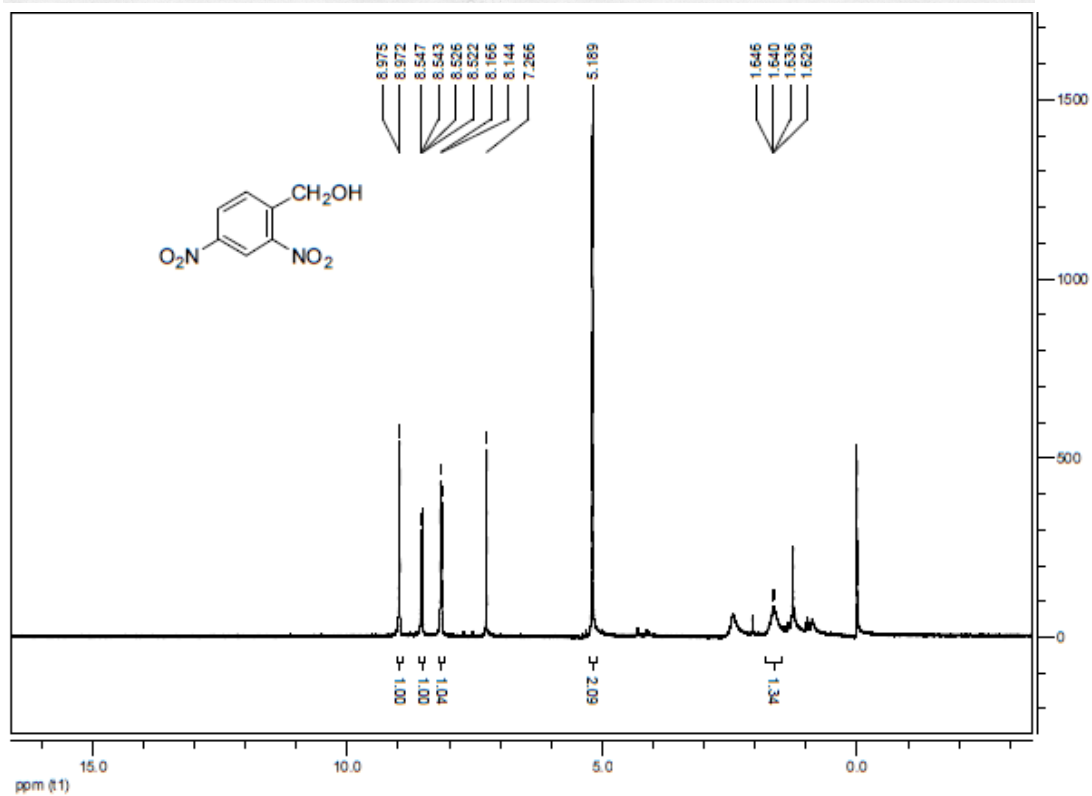
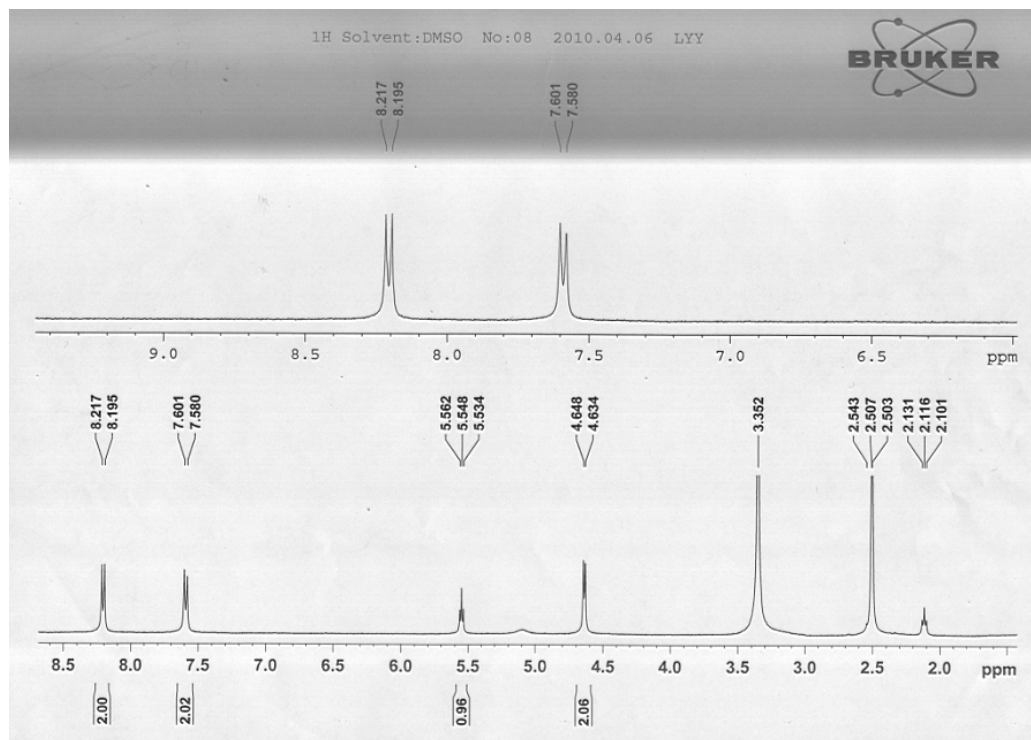
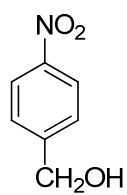


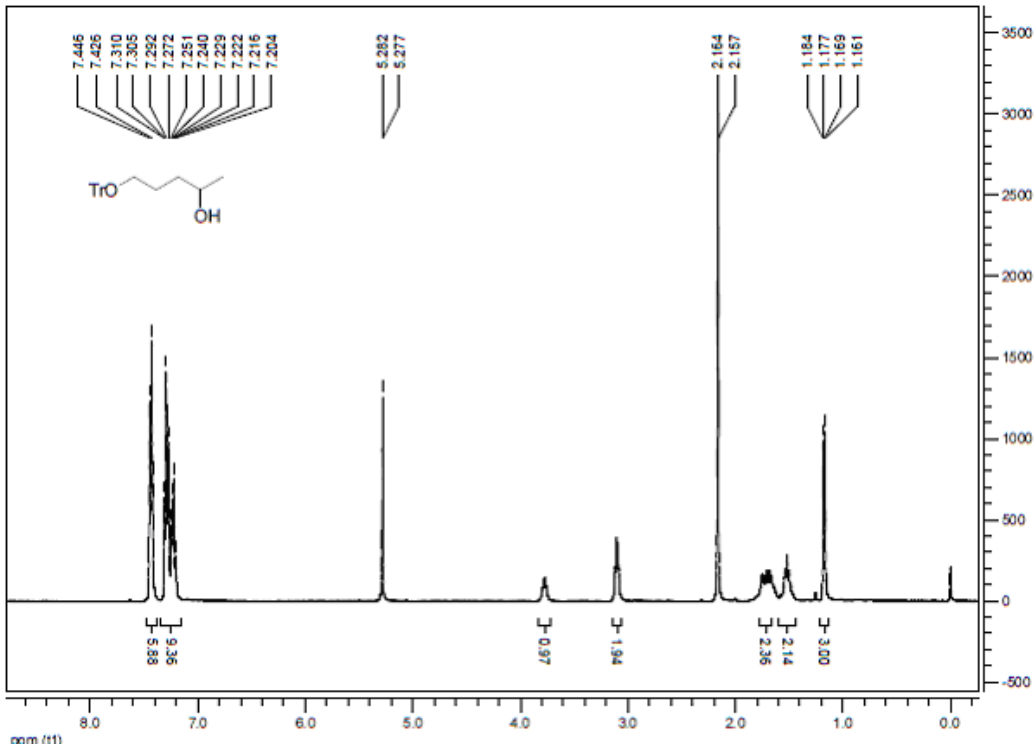
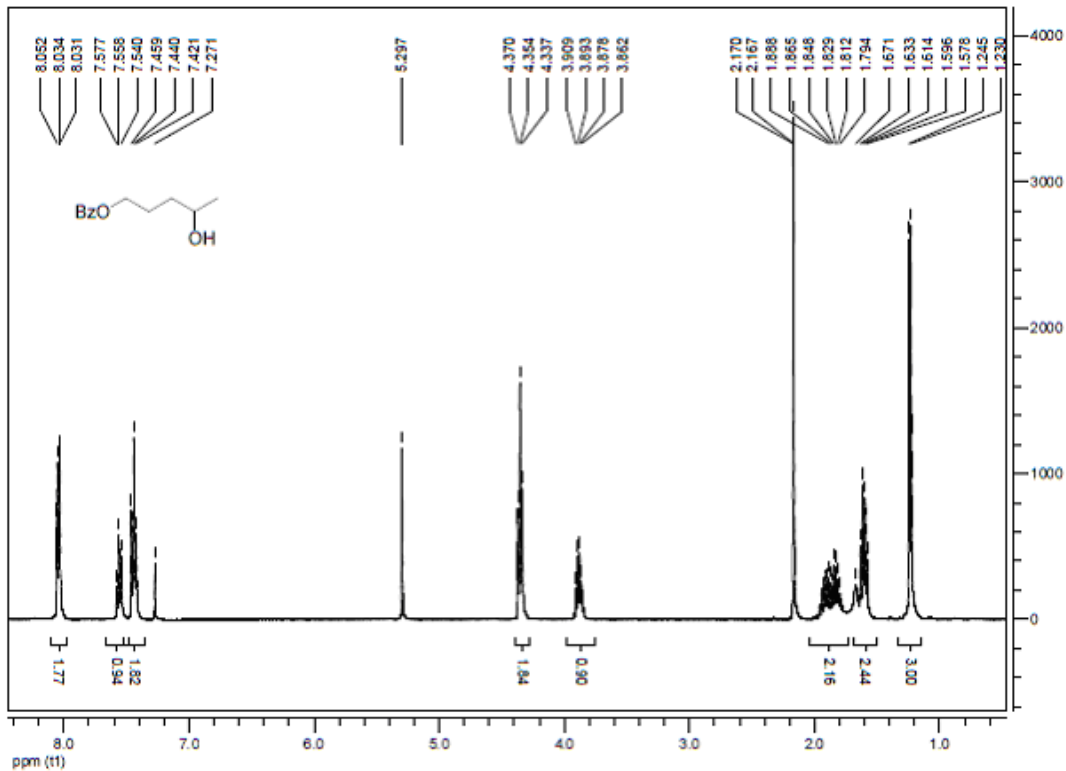


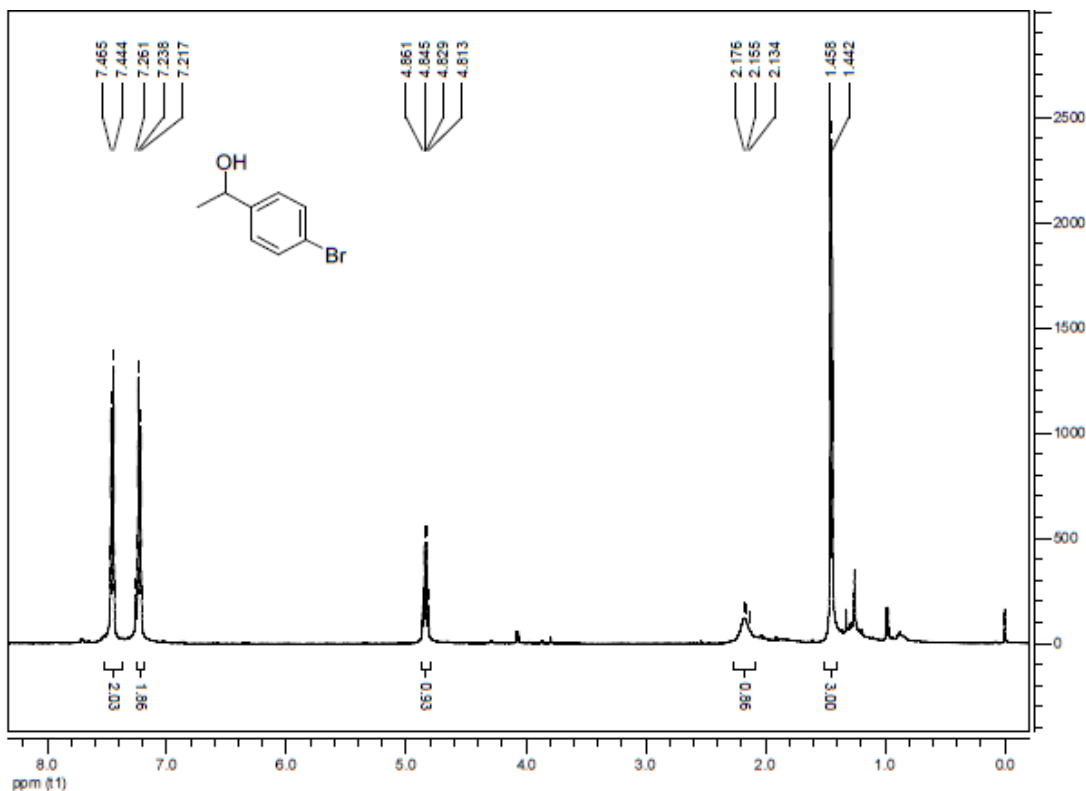
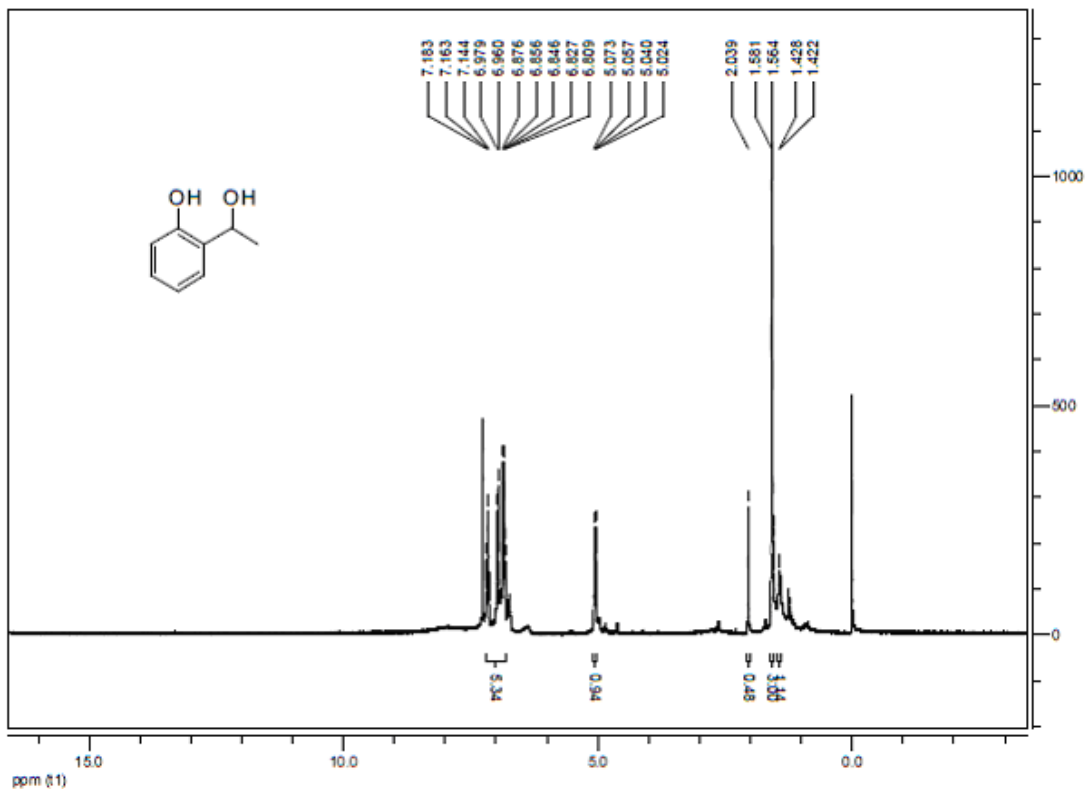


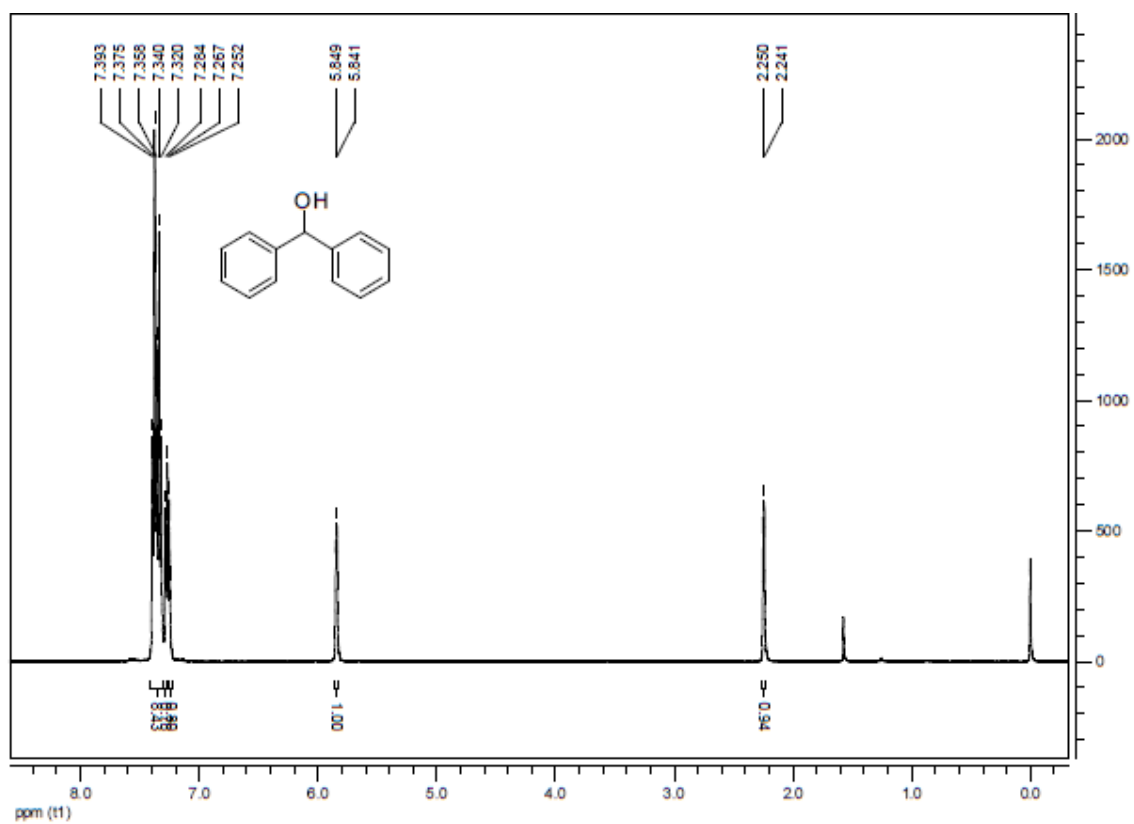
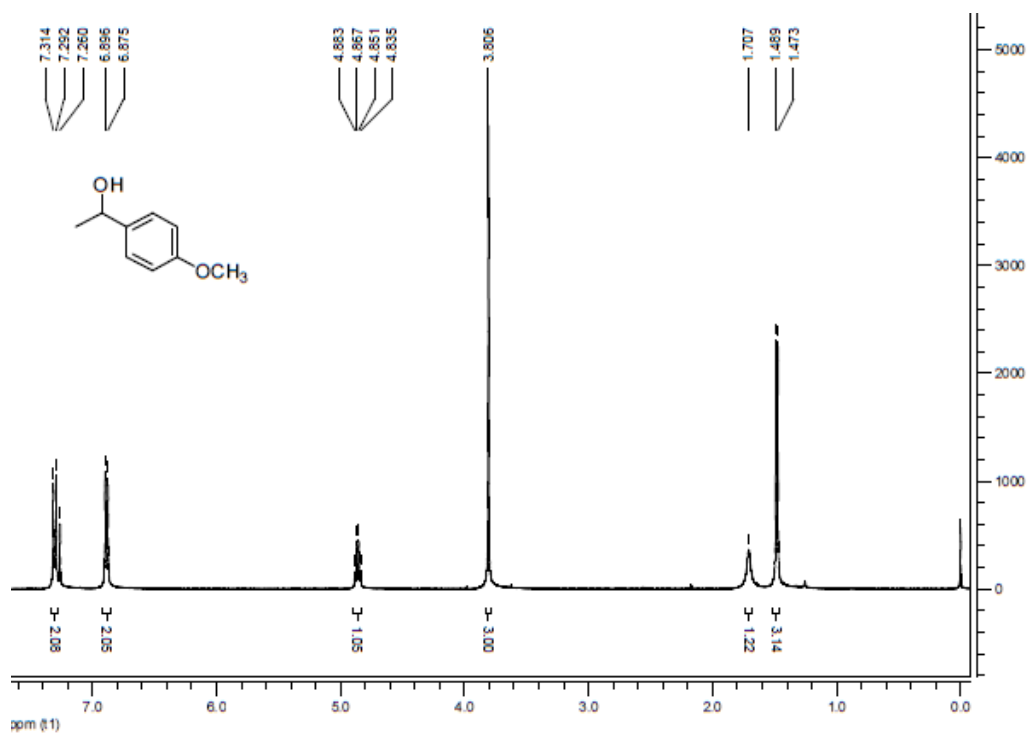




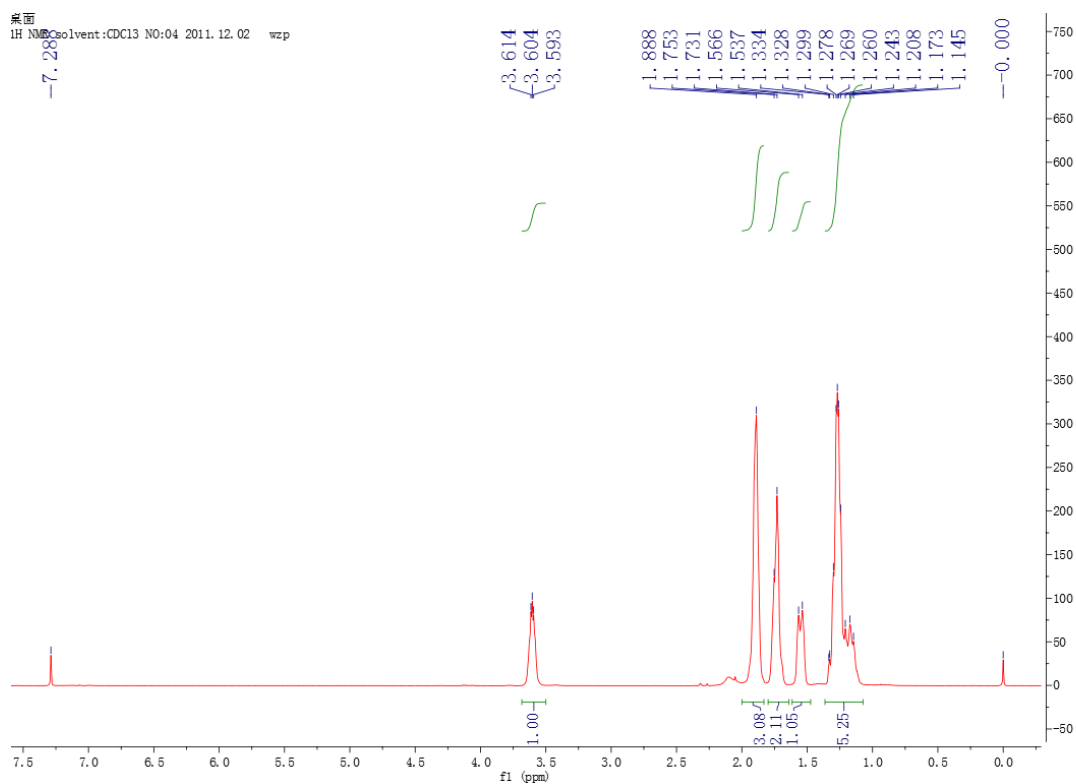








Cyclohexanol



Cyclopentanol

