## **Electronic Supplementary Information**

## Catalytic decarbonylation of biomass-derived carboxylic acids as an efficient route to commodity monomers

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GC-MS yields were calculated as follows; representative GC traces are included as Figures S1-S5 to illustrate typical results. A solution containing a known mass of the isolated distillate was injected on the GC-MS instrument. Each trace was analyzed using the Agilent GC/MSD ChemStation Software package and the area under the each peak of interest in the trace was determined. Previously generated standard curves were used to determine the concentration and mass ratio of each component in the mixture from the peak areas. The percent yield was calculated using these determined mass ratios compared to the total mass of distillate collected. The yields were calculated from averages of multiple (3-5) GC-MS runs.



## **Figure S1.** Representative GC-MS trace of distillate mixture using mono-methyl succinate as the substrate and PPh<sub>3</sub> as the ligand.



**Figure S2.** Representative GC-MS trace of distillate mixture using mono-*n*-butyl succinate as the substrate and DPEphos as the ligand.



**Figure S3.** Representative GC-MS trace of distillate mixture using mono-*t*-butyl succinate as the substrate and DPEphos as the ligand.



**Figure S4.** GC-MS trace of the sealed reaction between mono-methyl succinate and pivalic anhydride at 190 °C for 2 hours.



**Figure S5.** GC-MS trace of the sealed reaction between mono-*t*-butyl succinate and pivalic anhydride at 190 °C for 2 hours.



Figure S6. Representative GC-MS trace of distillate mixture using hydrocinnamic acid as the substrate and DPEphos as the ligand.



Figure S7. Representative GC-MS trace of distillate mixture using 3-cyanopropanoic acid as the substrate and PPh<sub>3</sub> as the ligand.