

## Supporting Information

### Transamidation of amides with amines in solvent-free condition by CeO<sub>2</sub> catalyst

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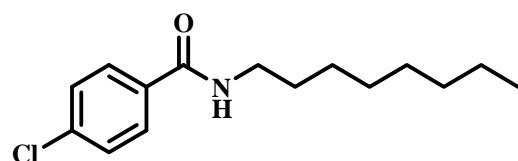
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#### Characterization of known compounds

Products N-octyl-benzamide (Table2-1),<sup>1</sup> N-octyl-4-pyridinecarboxamide (Table2-8)<sup>2</sup>, N-octyl-pyradinecarboxamide (Table2-9)<sup>3</sup>, N-octyl-acetamide (Table2-11),<sup>4</sup> N-benzylhexanamide (Table3-3),<sup>5</sup> N-(4-methoxybenzil)-hexanamide (Table3-5),<sup>6</sup> 1-morpholin-4-ylhexan-1-one (Table3-7)<sup>7</sup> and are known compounds and were identified by comparison of their NMR features with the respective reported data.

#### Characterization of new compounds.

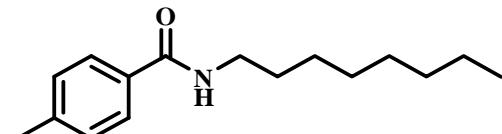
N-Octyl-4-chrolobenzamide (Table2-2)



<sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.72-7.68(m, 2H), 7.41-7.37(m, 2H), 6.21(br, s, 1H), 3.45-3.40(m, 2H), 1.64-1.56(m, 2H), 1.36-1.27(m, 10H), 0.88(t, J=6.9Hz, 3H)

MS: *m/z* (relative intensity) 267(M<sup>+</sup>, 4), 139(100).

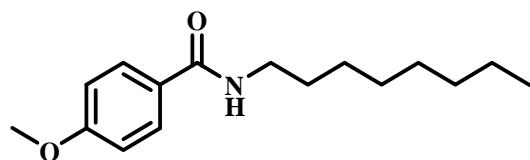
N-Octyl-4-methyl-benzlamide (Table2-3)



<sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.65 (d, J=8.2Hz, 2H), 7.22(d, J=7.8Hz, 2H), 6.09(br, s, 1H), 3.46-3.41(m, 2H), 2.39(s, 3H), 1.64-1.57(m, 2H), 1.39-1.27(m, 10H), 0.88(t, J=7.1Hz, 3H)

MS: *m/z* (relative intensity) 247(M<sup>+</sup>, 12), 119(100).

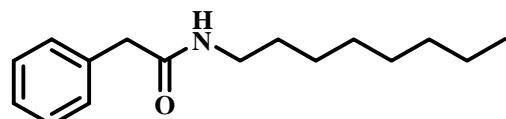
N-Octyl-4-methoxybenzamide (Table2-4)



<sup>1</sup>H NMR ( $\text{CDCl}_3$ ):  $\delta$  7.73(d,  $J=9.2\text{Hz}$ , 2H), 6.92(d,  $J=8.7\text{Hz}$ , 2H), 6.08(br, s, 1H), 3.84(s, 3H), 3.45-3.40(m, 2H), 1.64-1.56(m, 2H), 1.38-1.27(m, 10H), 0.88(t,  $J=7.1\text{Hz}$ , 3H)

MS:  $m/z$  (relative intensity) 263( $M^+$ , 3), 135(100).

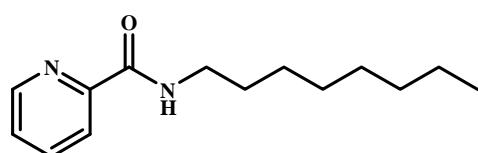
N-Octyl-benzylamide (Table2-5)



<sup>1</sup>H NMR ( $\text{CDCl}_3$ ):  $\delta$  7.35-7.24(m, 5H), 5.40(br, s, 1H), 3.56(s, 2H), 3.21-3.16(m, 2H), 1.44-1.37(m, 2H), 1.30-1.22(m, 10H), 0.87(t,  $J=7.1\text{Hz}$ , 3H)

MS:  $m/z$  (relative intensity) 247( $M^+$ , 16), 92(100).

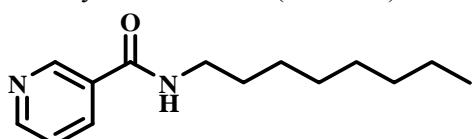
N-Octyl-2-picolinamide (Table2-6)



<sup>1</sup>H NMR ( $\text{CDCl}_3$ ):  $\delta$  8.55 (d,  $J=5.5\text{Hz}$ , 1H), 8.22-8.20(m, 1H), 8.06(br, s, 1H), 7.89-7.82(m, 1H), 7.43-7.40(m, 1H), 3.49-3.44(m, 2H), 1.66-1.60(m, 2H), 1.40-1.27(m, 10H), 0.88(t,  $J=6.9\text{Hz}$ , 3H)

MS:  $m/z$  (relative intensity) 234( $M^+$ , 9), 128(100).

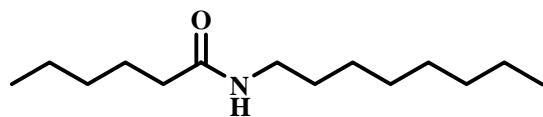
N-Octyl-nicotinamide (Table2-7)



<sup>1</sup>H NMR ( $\text{CDCl}_3$ ):  $\delta$  8.73-8.71(m, 2H), 7.61-7.60(m, 2H), 6.44(br, s, 1H), 3.48-3.43(m, 2H), 1.66-1.58(m, 2H), 1.39-1.27(m, 10H), 0.88(t,  $J=6.9\text{Hz}$ , 3H)

MS:  $m/z$  (relative intensity) 234( $M^+$ , 10), 106(100).

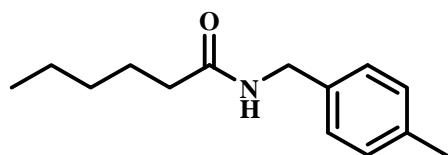
N-Octyl-hexanamide (Table2-12, 3-2)



$^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  5.48(br, s, 1H), 3.26-3.21(m, 2H), 2.15(t,  $J=7.7\text{Hz}$ , 2H), 1.69-1.59(m, 2H), 1.50-1.45(m, 2H), 1.35-1.27(m, 14H), 0.91-0.86(m, 6H)

MS:  $m/z$  (relative intensity) 227( $\text{M}^+$ , 5), 43(100).

N-(4-methoxybenzil)-hexanamide (Table3-4)



$^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  7.18-7.13(m, 4H), 5.71(br, s, 1H), 4.40(d,  $J=5.5\text{Hz}$ , 2H), 2.33(s, 3H), 2.24-2.17(m, 2H), 1.69-1.60(m, 2H), 1.38-1.26(m, 4H), 0.92-0.87(m, 3H)

MS:  $m/z$  (relative intensity) 219( $\text{M}^+$ , 16), 105(100).

Table S1 Transamidation by various metal oxides<sup>a</sup>.

Catalyst	$S_{\text{BET}} / \text{m}^2 \text{ g}^{-1}$	$t / \text{h}$	Yield (%)	$V^b / \text{mmol h}^{-1} \text{ g}^{-1}$	$V^c / \text{mmol h}^{-1} \text{ m}^{-2}$
CeO <sub>2</sub>	81	0.5	21.1	42.2	0.52
TiO <sub>2</sub>	47	1	14.3	14.3	0.30
Nb <sub>2</sub> O <sub>5</sub>	54	1	15.3	15.3	0.28
ZrO <sub>2</sub>	73	1	12.9	12.9	0.18
Al <sub>2</sub> O <sub>3</sub>	124	1	10.3	10.3	0.08
Sc <sub>2</sub> O <sub>3</sub>	24	1	6.1	6.1	0.25
SiO <sub>2</sub>	300	2	10.1	5.0	0.02
La <sub>2</sub> O <sub>3</sub>	18	1.5	2.4	1.6	0.09
Er <sub>2</sub> O <sub>3</sub>	9	1	1.6	1.6	0.18
MgO	19	2	2.2	1.1	0.06
CaO	59	2	1.5	0.8	0.01
Blank	-	1	0	-	-

<sup>a</sup> Reaction conditions: picolinamide (2.5 mmol), n-octylamine (5.0 mmol), metal oxide (25 mg),  $T = 160^\circ\text{C}$ . Yield of N-octyl picolinamide was determined by GC.

<sup>b</sup> Formation rate of N-octyl picolinamide per catalyst weight measured under the condition in which yield of N-octyl picolinamide was below 30%.

<sup>c</sup> Formation rate of N-octyl picolinamide per catalyst surface area.

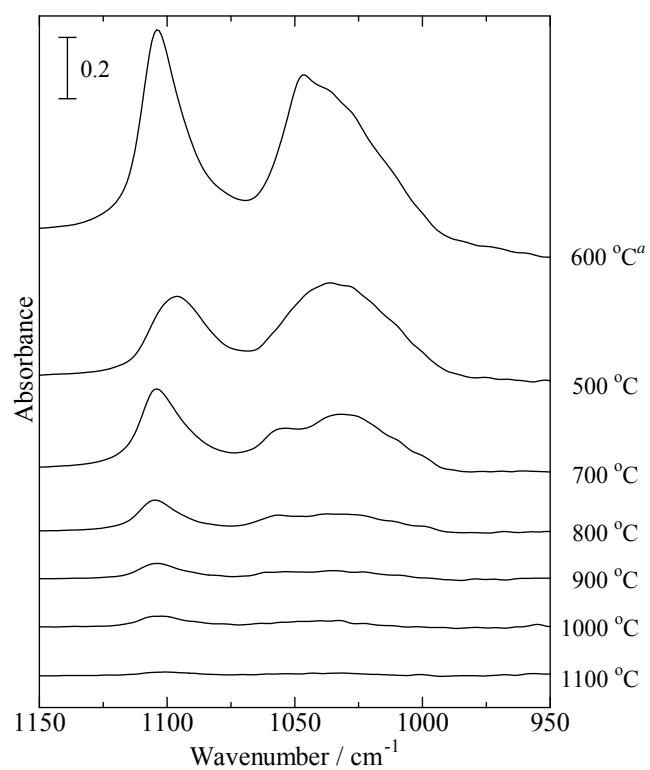


Fig. S1. Spectra of methanol complexes adsorbed to CeO<sub>2</sub> at different calcination temperature.  
Numbers in the graph are calcination temperatures (<sup>a</sup> JRC-CEO3).

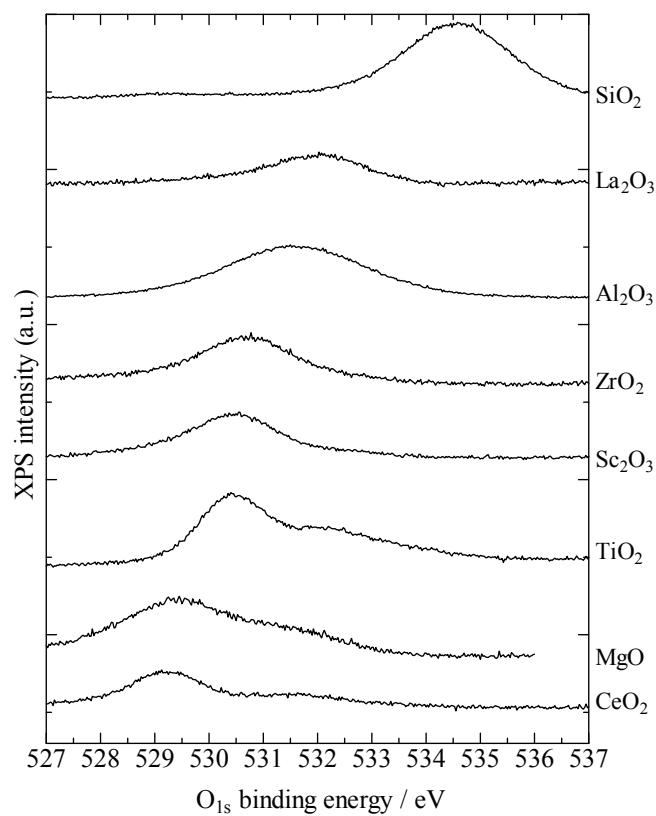


Fig. S2. XPS spectra of various metal oxides in the region of  $\text{O}_{1s}$  binding energy.

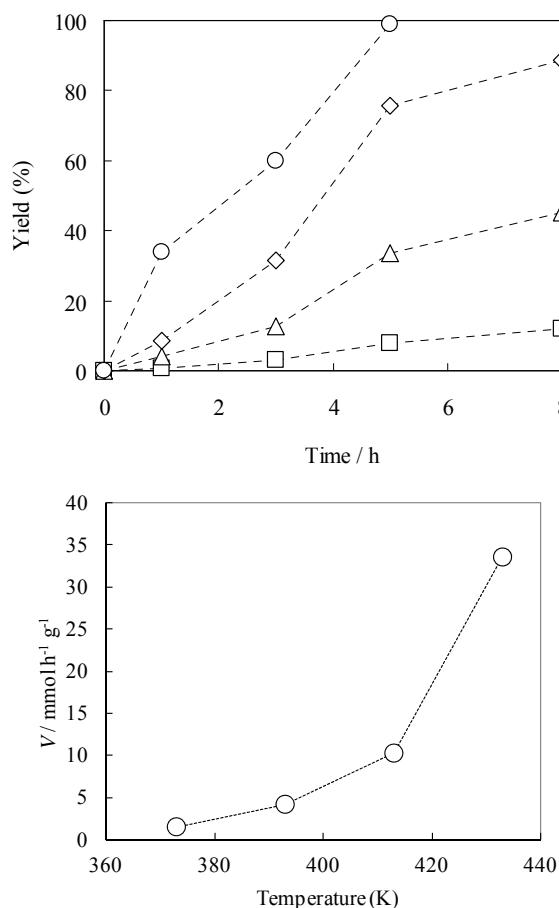


Fig. S3. Effect of reaction temperature. Conditions: picolinamide (2.5 mmol), n-octylamine (5.0 mmol), CeO<sub>2</sub> (25 mg).

## Reference

- 1 N. Iranpoor, H. Firouzabadi and D. Khalili, *Bull. Chem. Soc. Jpn.*, 2010, **83**, 923-924.
- 2 K. Motoshima, Y. Hiwasa, M. Yoshikawa, K. Fujimorto, A. Tai, H. Kakuta and K. Sasaki, *ChemMedChem*, 2007, **2**, 1527-1532.
- 3 Spectral data of N-octylpyradinecarboxamide were obtained from Wiley Subscription Services, Inc.
- 4 Y. Furuya, K. Ishihara and H. Yamamoto, *J. Am. Chem. Soc.*, 2005, **127**, 11240-11241.
- 5 L. U. Nordstrøm, H. Vogt and R. Madsen, *J. Am. Chem. Soc.*, 2008, **130**, 17672-17673.
- 6 M. Shi and S. C. Cui, *Syn. Commun.*, 2005, **35**, 2847-2858.
- 7 M. Badioli, R. Ballini, M. Bartolacci, G. Bosica, E. Torregiani and E. Marcantoni, *J. Org. Chem.*, 2002, **67**, 8938-8942.