

An efficient and recyclable dendritic catalyst able to dramatically reduce palladium leaching in Suzuki couplings

Michel Keller,^{a,b} Aurélien Hameau,^{a,b} Grégory Spataro,^{a,b} Sonia Ladeira,^{a,b} Anne-Marie Caminade,^{a,b*} Jean-Pierre Majoral,^{a,b*} Armelle Ouali.^{a,b*}

Supporting Information

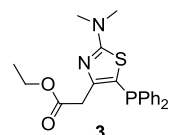
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1. Complete characterisation of the ligands with attribution of NMR signals

The preparation and characterization of compounds **1**, **2**, **3**, **4** have been previously described. The latter were identified in each case by ¹H NMR and ³¹P NMR (if applicable) spectroscopies and the data obtained matched literature values. Remarkably, in the case of **3**, a slightly modified procedure was used and crystals were grown from this compound thus allowing to obtain its structure by X-ray crystallography. For the preparation of compounds **3**, **4**, **5-OH**, **5-OMe**, **5-G₁**, **5-G₃**, **6-G₁**, **6-G₃**, all solvents were degassed before using.

Compound **3**:¹



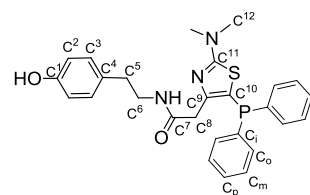
Triethylamine (3.3 mL, 23.6 mmol) was added to a solution of compound **2**¹ (2.81 g, 13.11 mmol) in anhydrous pyridine (12 mL). The mixture was cooled to -25°C and chlorodiphenylphosphine was added dropwise (15 min). The mixture was warmed up to room temperature overnight and heated to 40°C for 13 days (monitoring by ³¹P NMR). At the end of the reaction, 70 mL of toluene were added to the mixture. The salts were eliminated by filtration under argon and the brownish solution was evaporated to dryness. Column chromatography (CH₂Cl₂/AcOEt (90:10) as eluent) of the crude oil gave 2.45 g of the expected compound as a yellow oil (47% yield).

³¹P{¹H} NMR (CDCl₃, 121.50 MHz): δ = -28.89 ppm.

¹H NMR (CDCl₃, 300.13 MHz): δ = 1.18 (t, ³J_{H-H} = 7.2 Hz, 3H, CH₃), 3.07 (s, 6H, (CH₃)₂N), 3.98 (s, 2H, CH₂-thiazolyl), 4.11 (q, ³J_{H-H} = 7.2 Hz, 2H, CH₂O), 7.30-7.65 (m, 10H, H_{arom}).

This compound was crystallized from CHCl₃ by slow evaporation and the structure determined by X-ray crystallography (see section 3).

Compound **5-OH**:



To a solution of **4** (315 mg, 8.05.10⁻¹ mmol) in *N,N*-dimethylformamide [DMF] (6.5 mL) were added at 0°C hydroxybenzotriazole [HoBt] (0.156 g, 1.02 mmol) and *N,N'*-dicyclohexylcarbodiimide [DCC] (0.211 g, 1.02 mmol). After one hour at 0°C, a solution of tyramine (0.233 g, 1.7 mmol) in DMF (3 mL) was added over 5 minutes. After 2 hours at 0°C, the mixture was warmed up to room temperature and stirred for 7 days (monitoring by ³¹P NMR). The mixture was filtered and the filtrate was freeze-dried. A first column chromatography (CH₂Cl₂/EtOH = 93/7 as eluent) of the crude product permitted to eliminate the excess of tyramine, DCU and a large part of the oxidized compound. A second column chromatography (CH₂Cl₂/AcOEt (80:20) as eluent) gave 196 mg of the expected product as a white powder (47% yield).

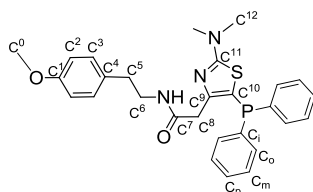
³¹P{¹H} NMR (CD₂Cl₂, 121.50 MHz): δ = -29.79 ppm. (impurity : minus 2%; δ = 21.9 ppm : oxide).

¹H NMR (CD₂Cl₂, 300.13 MHz): δ = 2.62 (t, ³J_{H-H} = 6.8 Hz, 2H, C⁵-H), 2.98 (s, 6H, C¹²-H), 3.42 (m, 2H, C⁶-H), 3.84 (d, ⁴J_{H-P} = 1.2 Hz, 2H, C⁸-H), 6.47 (s, 1H, OH), 6.70 (m, 2H, C²-H), 6.90 (m, 2H, C³-H), 7.06 (br t, ³J_{H-H} = 5.7 Hz, 1H, NH), 7.25-7.50 (m, 10H, H_{pph2}).

¹³C{¹H} NMR (CD₂Cl₂, 75.47 MHz): δ = 34.56 (C⁵), 38.27 (d, ³J_{C-P} = 16.2 Hz, C⁸), 39.93 (C¹²), 40.79 (C⁶), 113.64 (d, C¹⁰, ¹J_{C-P} = 33.2 Hz), 115.36 (C²), 128.45 (d, ³J_{C-P} = 6.8 Hz, C_m), 128.64 (C_p), 129.63 (C³), 130.06 (C⁴), 132.68 (d, ²J_{C-P} = 19.6 Hz, C_o), 138.04 (d, ¹J_{C-P} = 6.8 Hz, C_i), 155.19 (C¹), 156.09 (d, ²J_{C-P} = 31.7 Hz, C⁹), 169, 56 (C⁷), 174.47 (C¹¹).

DCI-MS (NH₃) m/z: 490.2 [M+H]⁺. This compound was crystallized from CH₂Cl₂ by slow evaporation and the structure determined by X-ray crystallography (see section 3).

Compound 5-OMe :



To a solution of **4**¹ (200 mg, 5.39.10⁻¹ mmol) in DMF (5 mL) was added at 0°C 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride [EDC] (0.127 g, 6.28.10⁻¹ mmol) and HoBt (0.088 g, 6.28.10⁻¹ mmol). After two hours at 0°C, 2-*p*-methoxy-ethylamine (160 µL, 1.07 mmol) was added dropwise and after 1.5 hours at 0°C, the mixture was warmed up to room temperature and stirred during 3 days (monitoring by ³¹P NMR). The mixture was filtered and the filtrate was freeze-dried. The crude oil was dissolved in dichloromethane (25 mL) and washed twice with 15 mL of water. The organic phase was dried over MgSO₄, filtered, and evaporated to dryness. Column chromatography (CH₂Cl₂/Acetone (80:20) as eluent, R_f = 0.6) of the crude product gave 225 mg of the expected compound as a white powder (83% yield).

M. p. = 113.6 °C.

³¹P{¹H} NMR (CD₂Cl₂, 121.50 MHz): δ = -29.65 ppm.

¹H NMR (CD₂Cl₂, 300.13 MHz): δ = 2.65 (t, ³J_{H-H} = 6.9 Hz, 2H, C⁵-H), 2.98 (s, 6H, C¹²-H), 3.41 (m, 2H, C⁶-H), 3.77 (s, 3H, C⁰-H), 3.79 (d, ⁴J_{H-P} = 1.2 Hz, 2H, C⁸-H), 6.75 (m, 2H, C²-H), 6.86 (br s, 1H, NH), 7.00 (m, 2H, C³-H), 7.30-7.50 (m, 10H, H_{PPh₂}).

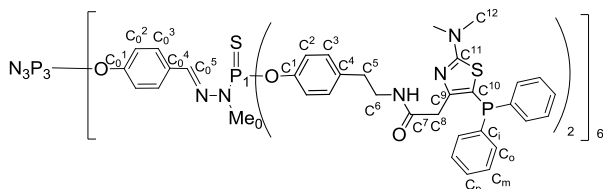
¹³C{¹H} NMR (CD₂Cl₂, 75.47 MHz): δ = 34.58 (C⁵), 38.52 (d, ³J_{C-P} = 15.9 Hz, C⁸), 39.86 (C¹²), 40.52 (C⁶), 55.08 (C⁰), 113.17 (d, C¹⁰, ¹J_{C-P} = 35.2 Hz), 113.68 (C²), 128.41 (d, ³J_{C-P} = 6.8 Hz, C_m), 128.60 (C_p), 129.59 (C³), 131.24 (C⁴), 132.67 (d, ²J_{C-P} = 19.4 Hz, C_o), 138.14 (d, ¹J_{C-P} = 6.6 Hz, C_i), 156.60 (d, ²J_{C-P} = 31.0 Hz, C⁹), 158.06 (C¹), 168.63 (C⁷), 174.32 (C¹¹).

DCI-MS (CH₄) m/z: 504.2 [M+H]⁺.

113.6 °C.

This compound was crystallized from CH₂Cl₂ by slow evaporation and the structure determined by X-ray crystallography (see section 3).

Compound 5-G₁ :



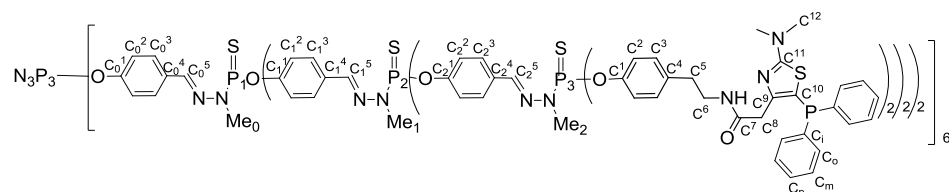
Cesium carbonate (349 mg, 1.07.10⁻³ mol) was added at 0°C to a solution of **G**₁ (89 mg, 4.87.10⁻² mmol) and **5-OH** (310 mg, 6.33.10⁻¹ mmol) in THF (10 mL). The mixture was stirred two hours at 0°C and warmed up to room temperature. The progress of the reaction was monitored by ³¹P NMR. After complete conversion (36 hrs) the crude mixture was centrifuged and the supernatant collected. The solvent was removed in *vacuum* and the crude product purified by column chromatography (first with dichloromethane/acetone (70:30) as eluent to eliminate the excess of free phenol, R_f = 0.6 and then with dichloromethane/ethanol (93:7) as eluent, R_f = 0.2) to give 278 mg of the desired compound as a white powder (80% yield).

³¹P{¹H} NMR (CD₂Cl₂, 121.50 MHz): δ = -29.68 (PPh₂), 8.12 (N₃P₃), 62.78 (P₁) (impurity : minus 3%; δ = 21.2 ppm: oxide).

¹H NMR (CD₂Cl₂, 300.13 MHz): δ = 2.62 (t, ³J_{H-H} = 6.6 Hz, 24H, C⁵-H), 2.92 (s, 72H, C¹²-H), 3.23 (d, ²J_{H-P} = 10.2 Hz, 18H, Me₀), 3.34 (m, 24H, C⁶-H), 3.77 (br s, 24H, C⁸-H), 6.82 (br t, 12NH), 6.90-7.12 (m, 60H, 12 C₀²-H, 24 C²-H, 24 C³-H), 7.19-7.49 (m, 120H, H_{PPh₂}), 7.58-7.70 (m, 18H, 6 C⁵-H, 12 C₀³-H).

¹³C{¹H} NMR (CD₂Cl₂, 75.47 MHz): δ = 33.07 (d, ²J_{C-P} = 11.9 Hz, Me₀), 34.98 (C⁵), 38.48 (d, ³J_{C-P} = 15.8 Hz, C⁸), 40.12 (C¹²), 40.49 (C⁶), 114.00 (d, C¹⁰, ¹J_{C-P} = 33.2 Hz), 121.12 (d, ³J_{C-P} = 4.5 Hz, C²), 121.33 (C₀²), 128.28 (C₀³), 128.48 (d, ³J_{C-P} = 6.8 Hz, C_m), 128.84 (C_p), 129.80 (C³), 132.26 (C₀⁴), 132.72 (d, ²J_{C-P} = 18.9 Hz, C_o), 136.45 (C⁴), 137.96 (d, ¹J_{C-P} = 6.6 Hz, C_i), 138.69 (d, ³J_{C-P} = 14.3 Hz, C₀⁵), 148.99 (d, ²J_{C-P} = 7.5 Hz, C¹), 151.23 (m, C₀¹), 156.08 (d, ²J_{C-P} = 31.7 Hz, C⁹), 169.22 (C⁷), 174.32 (C¹¹).

Compound 5-G₃ :

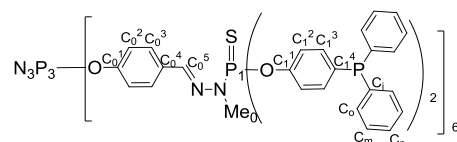


Cesium carbonate (348 mg, 1.07 mmol) was added at 0°C to a solution of **G₃** (130 mg, 1.21.10⁻² mmol) and **5-OH** (310 mg, 6.33.10⁻¹ mmol) in THF (10 mL). The mixture was stirred two hours at 0°C and warmed up to room temperature. The progress of the reaction was monitored by ³¹P NMR. After complete conversion (72 hrs), the mixture was centrifuged and the supernatant collected. The solvent was removed in *vacuum* and the crude product purified by size exclusion chromatography (THF as eluent) to eliminate the excess of phenol and to give 420 mg of the desired compound as white powder (75% yield). ³¹P{¹H} NMR (CD₂Cl₂, 121.50 MHz): δ = -29.67 (PPh₂), 7.65 (N₃P₃), 62.77 (P₂+ P₃), 62.90 (P₁) (impurity : minus 5%; δ = 19.9 ppm: oxide).

¹H NMR (CD₂Cl₂, 300.13 MHz): δ = 2.61 (m, 96H, C⁵-H), 2.88 (s, 288H, C¹²-H), 3.10-3.50 (m, 222H: 126 [Me₀ + Me₁ + Me₂] + 96 C⁶-H), 3.62-4.02 (br s, 96H, C⁸-H), 6.75-7.12 (m, 252H: 48NH + 12 C₀²-H + 96 C²-H + 96 C³-H), 7.13-7.52 (m, 552H : 24 C₁²-H + 48 C₂²-H + 480 H_{PPh₂}), 7.53-7.90 (m, 126H : 12 C₀³-H + 24 C₁³-H + 48 C₂³-H + 6 C₀⁵-H + 12 C₁⁵-H + 24 C₂⁵-H).

¹³C{¹H} NMR (CD₂Cl₂, 75.47 MHz): δ = 32.53 (m, Me₀, Me₁), 32.99 (d, ²J_{C-P} = 12.5 Hz, Me₁), 34.90 (C⁵), 38.57 (d, ³J_{C-P} = 15.8 Hz, C⁸), 39.91 (C¹²), 40.40 (C⁶), 113.35 (d, C¹⁰, ¹J_{C-P} = 32.5 Hz), 121.18 (d, ³J_{C-P} = 4.4 Hz, C²), 121.83 (m, C₀², C₁² C₂²), 128.33 (m, C₀³, C₁³ C₂³), 128.47 (d, ³J_{C-P} = 6.8 Hz, C_m), 128.65 (C_p), 129.81 (C³), 132.44 (C₀⁴, C₁⁴ C₂⁴), 132.65 (d, ²J_{C-P} = 19.3 Hz, C₀), 136.63 (C⁴), 138.07 (d, ¹J_{C-P} = 6.5 Hz, C_i), 138.95 (d, ³J_{C-P} = 14.4 Hz, C₂⁵), 139.54 (m, C₀⁵, C₁⁵), 149.02 (d, ²J_{C-P} = 7.1 Hz, C¹), 151.38 (m, C₀¹, C₁¹, C₂¹), 156.60 (d, ²J_{C-P} = 31.5 Hz, C⁹), 168.73 (C⁷), 174.28 (C¹¹).

Compound 6-G₁ :



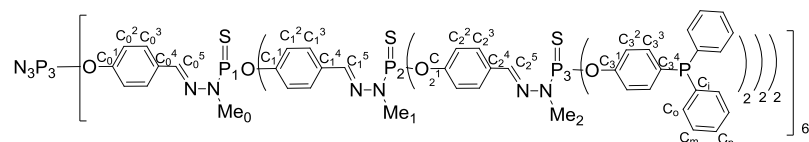
Cesium carbonate (356.5 mg, 1.09 mmol) was added to a solution of 4-(diphenylphosphino)phenol **6²** (198 mg, 0.71 mmol) and **G₁** (100 mg, 0.053 mmol) in THF (10 mL). The mixture was stirred overnight at room temperature, then filtered under argon and the filtrate was evaporated to dryness. The resulting oil was dissolved in CH₂Cl₂ (1 mL) and this solution was added to a mixture of 50 mL *n*-pentane/Et₂O (10:1) to allow dendrimer **6-G₁** to precipitate. **6-G₁** was obtained as a white powder in 71 % yield (186 mg, 0.039 mmol).

¹H NMR (300 MHz, CD₂Cl₂, 25°C) δ (ppm): 3.27 (d, ³J_{H-P} = 10.5 Hz, 18H, Me₀), 6.96-6.98 (m, 12H, C₀²-H), 7.14-7.30 (m, 168H, C₁²-H, C₁³-H, PPh₂), 7.49-7.52 (m, 18H, C₀³-H, C₀⁵-H);

³¹P-{¹H} NMR (121.5 MHz, CD₂Cl₂, 25°C) δ (ppm): -6.67 (s, PPh₂), 8.12 (s, N₃P₃), 61.67 (s, P₁);

¹³C-{¹H} NMR (75 MHz, CD₂Cl₂, 25°C) δ (ppm): 32.83 (d, ²J_{C-P} = 12.5 Hz, Me₀), 121.12 (s, C₀²), 121.35 (dd, ³J_{C-P} = 7.1 Hz, ³J_{C-P} = 5.0 Hz, C₁²), 128.15 (s, C₀³), 128.54 (d, ³J_{C-P} = 7.1 Hz, C_m), 128.82 (s, C_p), 132.01 (s, C₀⁴), 133.56 (d, ²J_{C-P} = 19.7 Hz, C₀), 134.46 (dd, ¹J_{C-P} = 12.3 Hz, ⁵J_{C-P} = 1.7 Hz, C₁⁴), 135.49 (d, ²J_{C-P} = 20.8 Hz, C₁³), 136.99 (d, ¹J_{C-P} = 11.3 Hz, C_i), 138.94 (d, ³J_{C-P} = 15.0 Hz, C₀⁵), 151.11-151.21 (m, C₀¹, C₁¹).

Compound 6-G₃ :



Cesium carbonate (365 mg, 1.12 mmol) was added to a solution of 4-(diphenylphosphino)phenol **6²** (203 mg, 0.73 mmol) and **G₃** (150 mg, 0.014 mmol) in THF (10 mL). The mixture was stirred overnight at room temperature, then filtered under argon and the filtrate was evaporated to dryness. The resulting oil was dissolved in CH₂Cl₂ (2 mL) and this solution was added to a mixture of 50 mL *n*-pentane /Et₂O (10:1 then 5:5 then 5:15) to allow dendrimer **6-G₃** to precipitate. **6-G₃** was obtained as a white powder in 65 % yield (184 mg, 0.005 mmol).

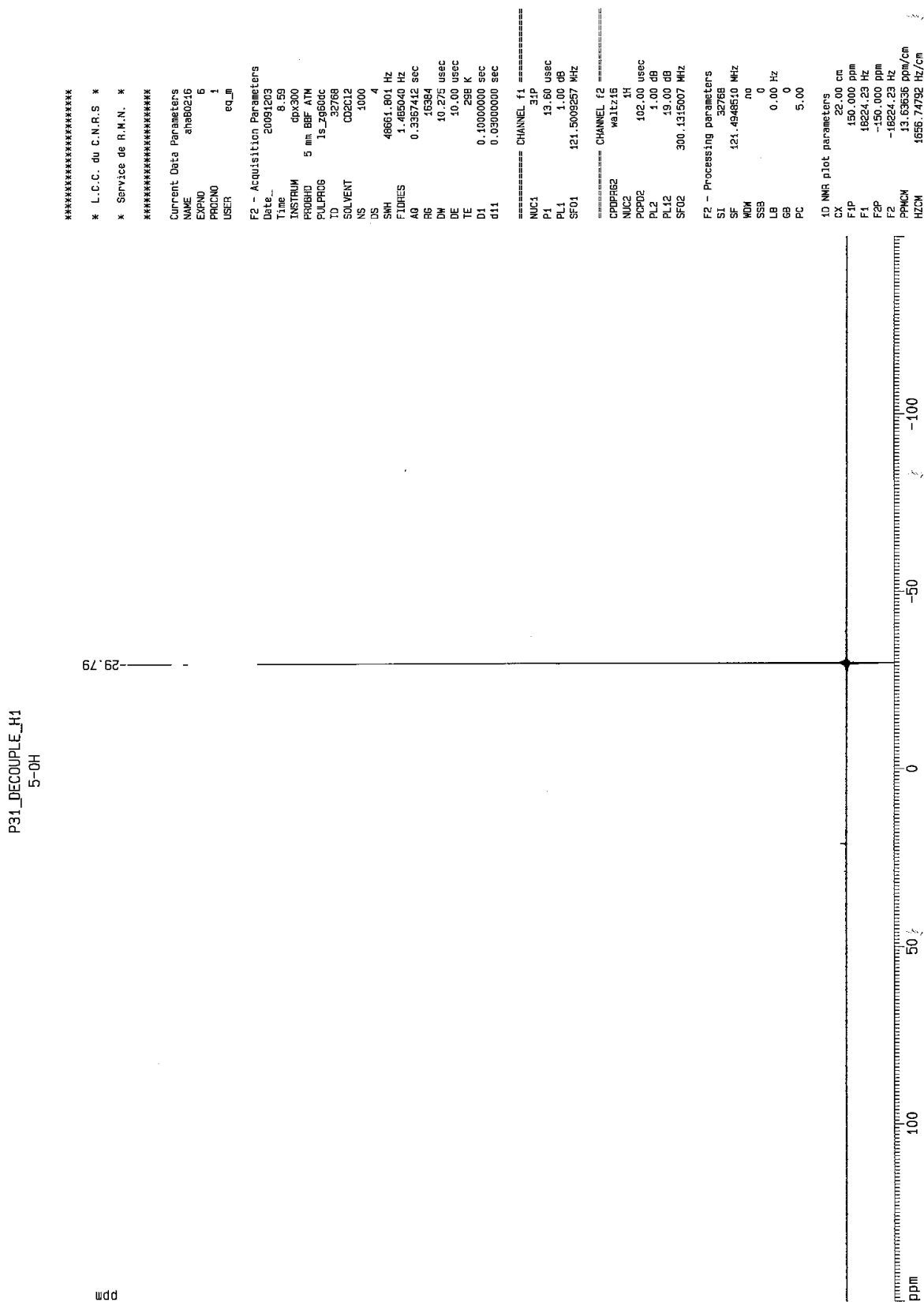
³¹P-{¹H} NMR (121.5 MHz, CD₂Cl₂, 25°C) δ (ppm): -6.73 (s, PPh₂), 7.92 (s, N₃P₃), 61.59 (s, P₃), 62.36 (s, P₂), 62.65(s, P₁);

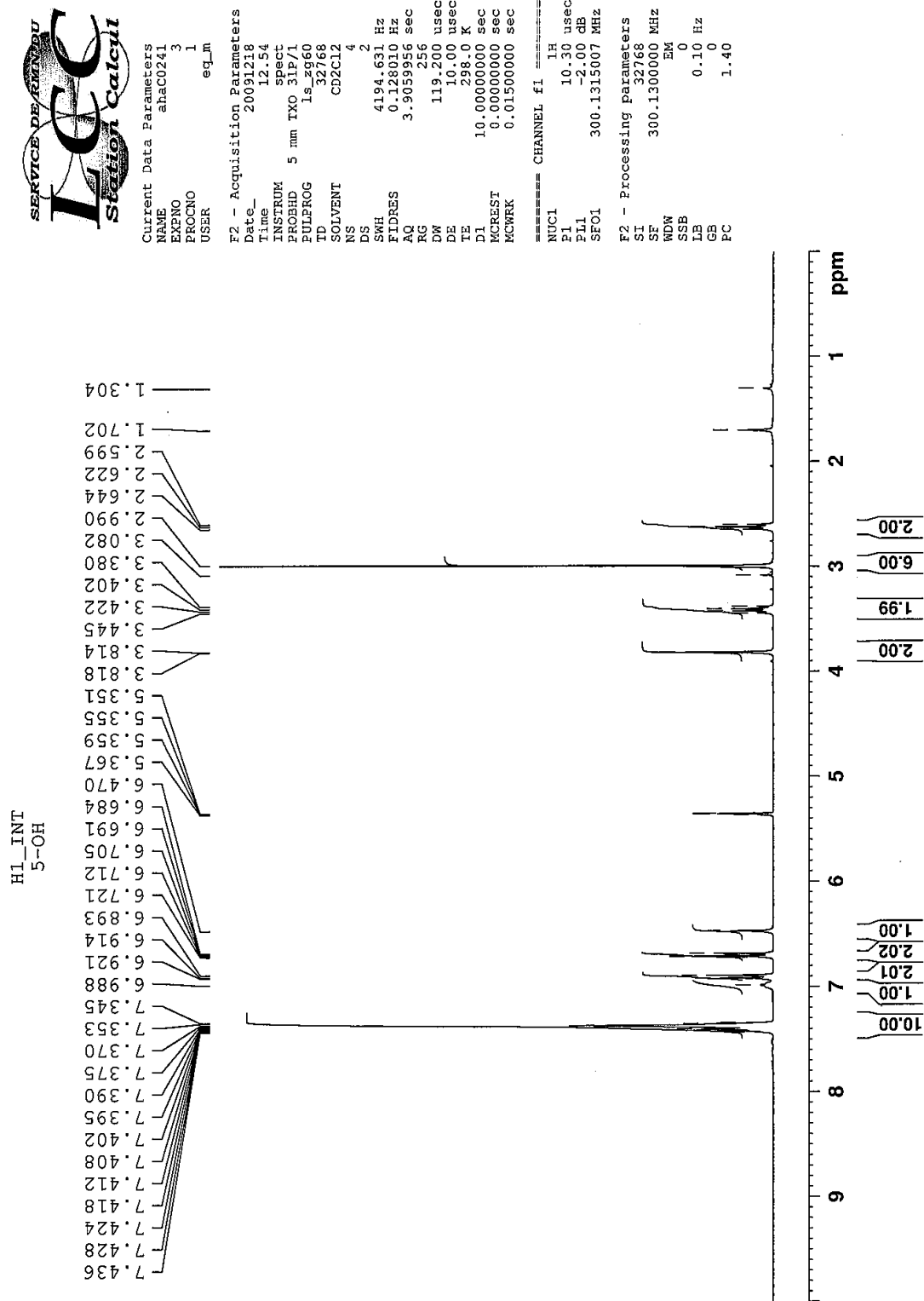
¹H NMR (300 MHz, CD₂Cl₂, 25°C) δ (ppm): 3.23-3.26 (m, 126H, Me₀, Me₁, Me₂), 7.02-7.05 (m, 12H, C₀²-H), 7.12-7.23 (m, 744H, C₁²-H, C₂²-H, C₃²-H, C₃³-H, PPh₂), 7.53-7.62 (m, 126H, C₀³-H, C₁³-H, C₂³-H, C₀⁵-H, C₁⁵-H, C₂⁵-H);

¹³C-{¹H} NMR (75 MHz, CD₂Cl₂, 25°C) δ (ppm): 32.85 (d, ²J_{C-P} = 13 Hz, Me₀, Me₁, Me₂), 121.39-121.73 (m, C₀², C₁², C₂², C₃²), 128.18 (s, C₂³), 128.28 (s, C₀³, C₁³), 128.54 (d, ³J_{C-P} = 7.1 Hz, C_m), 128.81 (s, C_p), 132.14-132.37 (m, C₀⁴, C₁⁴, C₂⁴),

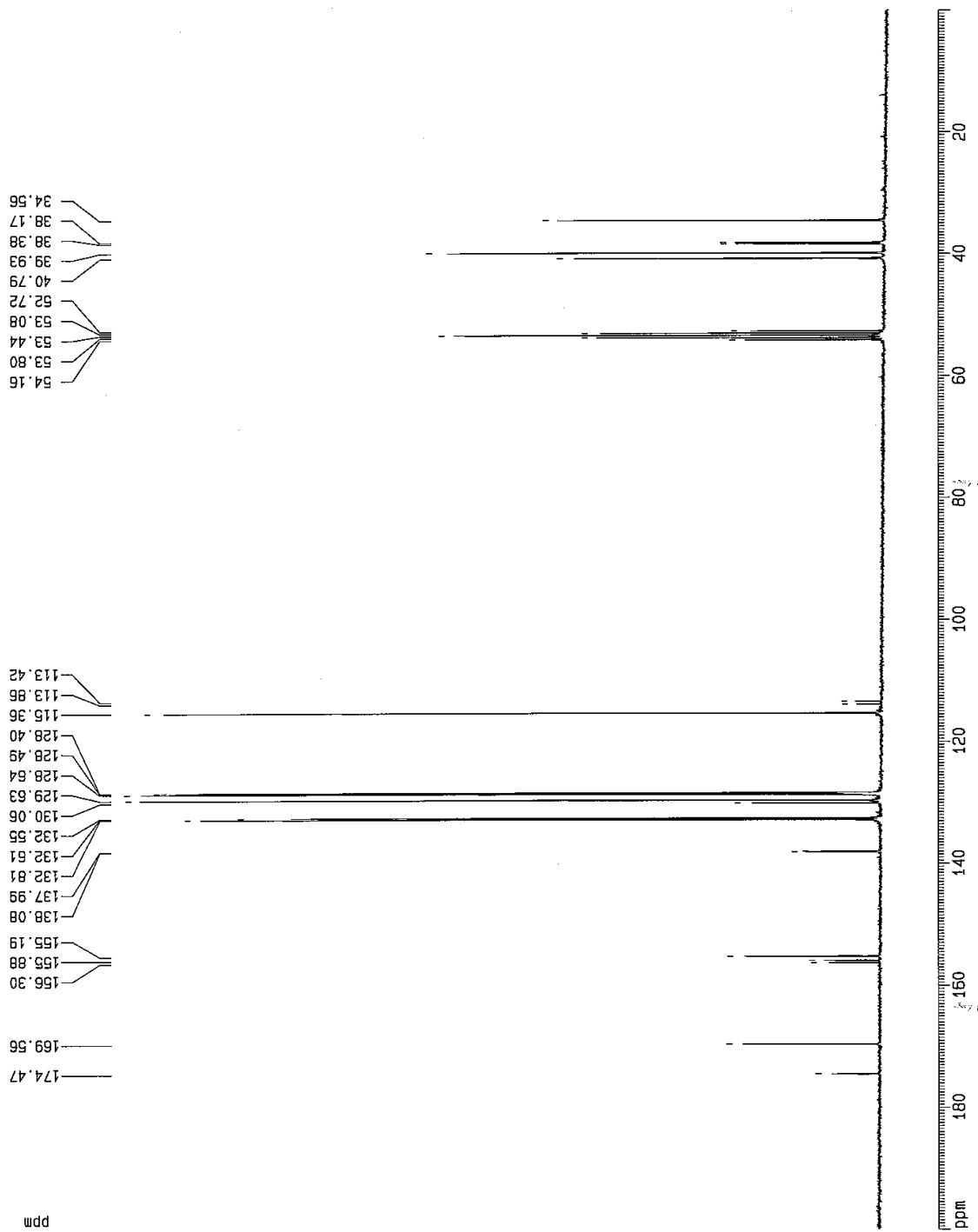
133.54 (d, $^2J_{C-P} = 19.9$ Hz, C_0), 134.33 (d, $^1J_{C-P} = 12$ Hz, C_3^4), 134.98 (d, $^2J_{C-P} = 21.3$ Hz, C_3^3), 137.02 (d, $^1J_{C-P} = 11.6$ Hz, C_i), 139.02 (s br, C_0^5 , C_1^5 , C_2^5), 151.18-151.29 (m, C_0^1 , C_1^1 , C_2^1 , C_3^1).
Note: Coupling constants J_{C-P} were calculated according to the ^{13}C Jmod spectrum (see section 2).

2. ¹H, ¹³C and ³¹P NMR spectra of ligands 5-OH, 5-OMe, 5-G₁, 5-G₃, 6-G₁ and 6-G₃ and of isolated coupling products (some selected examples).
Compound 5-OH





C13_DECOUPLE_H1
5-OH



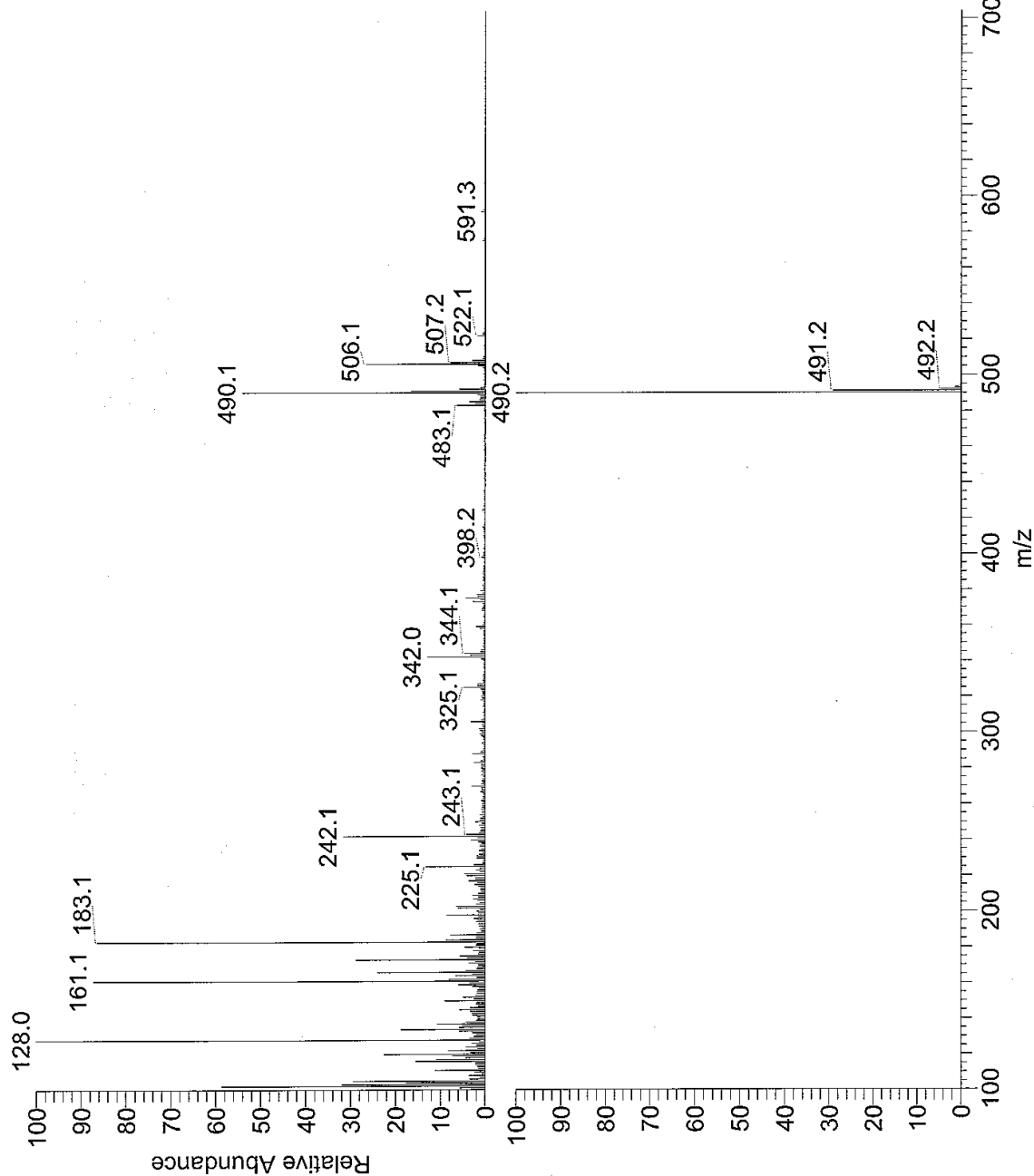
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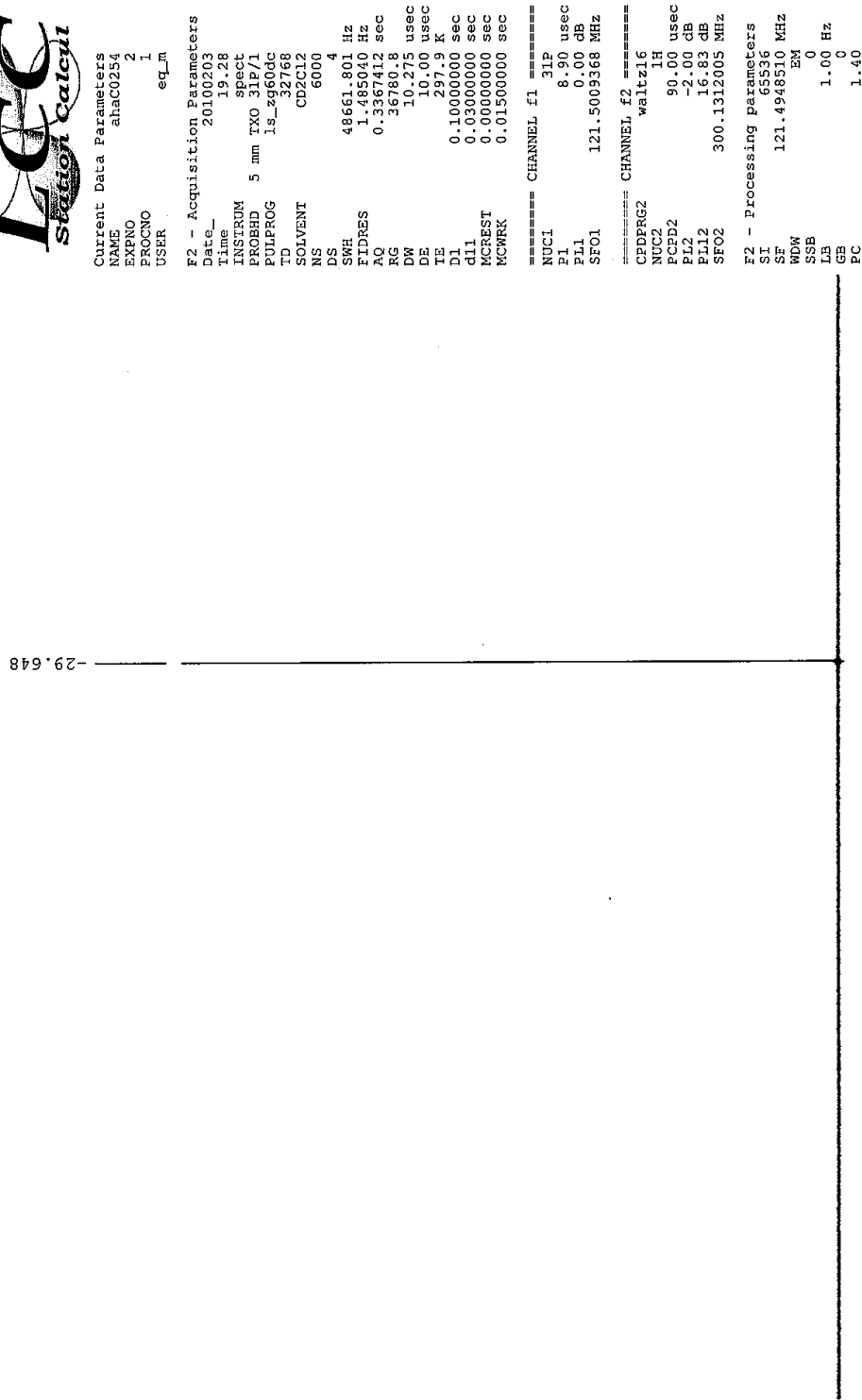
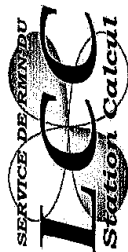
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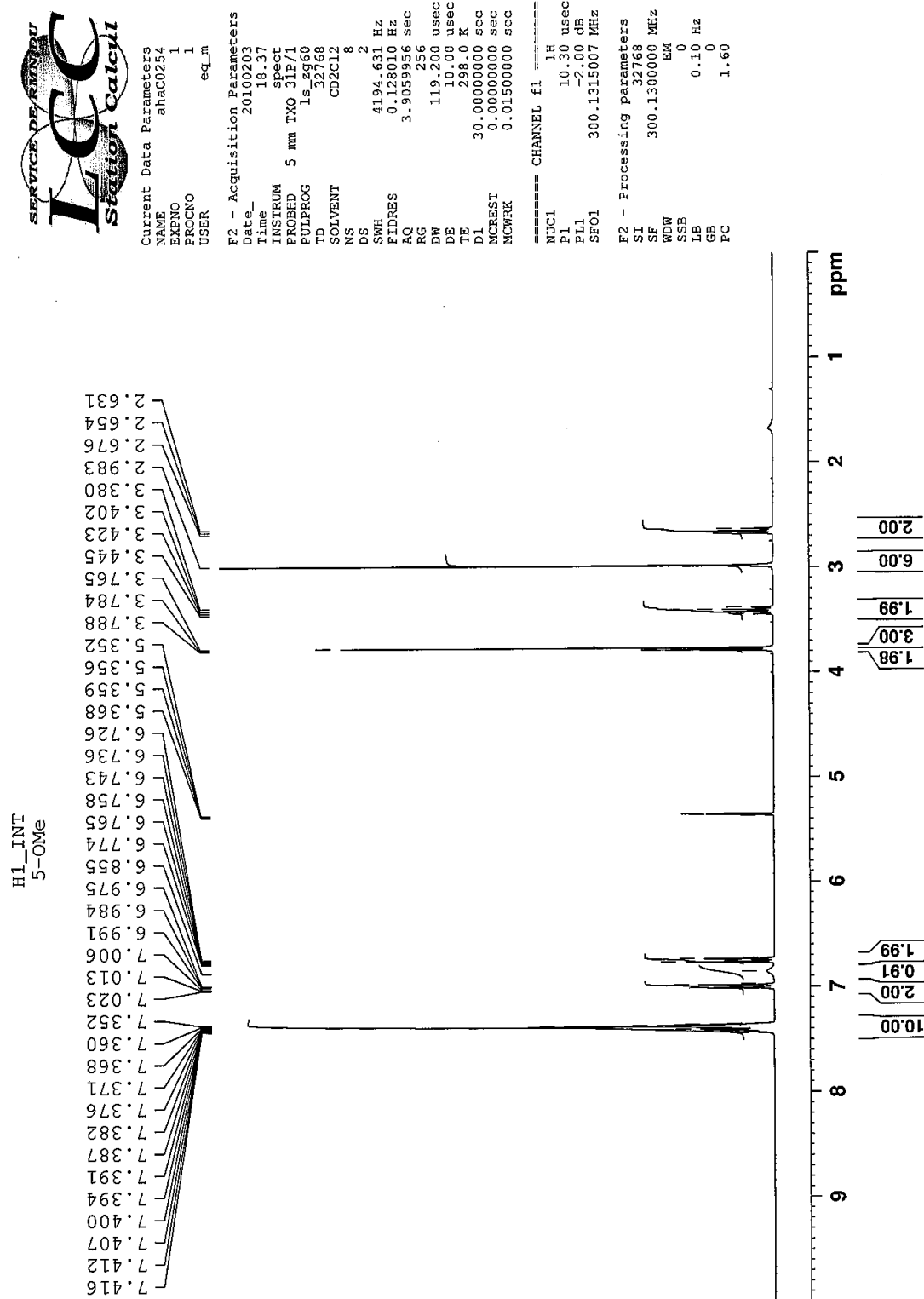


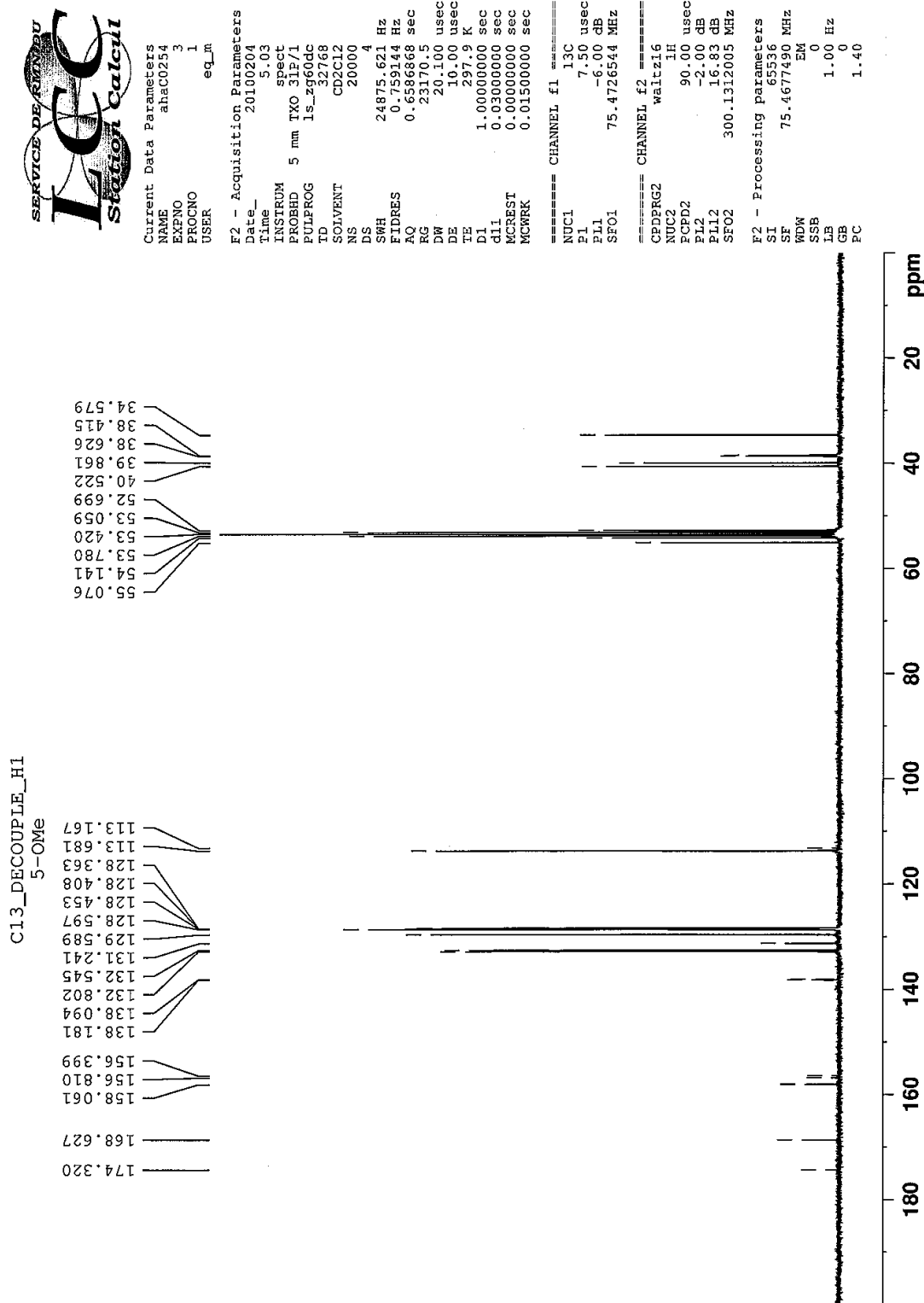
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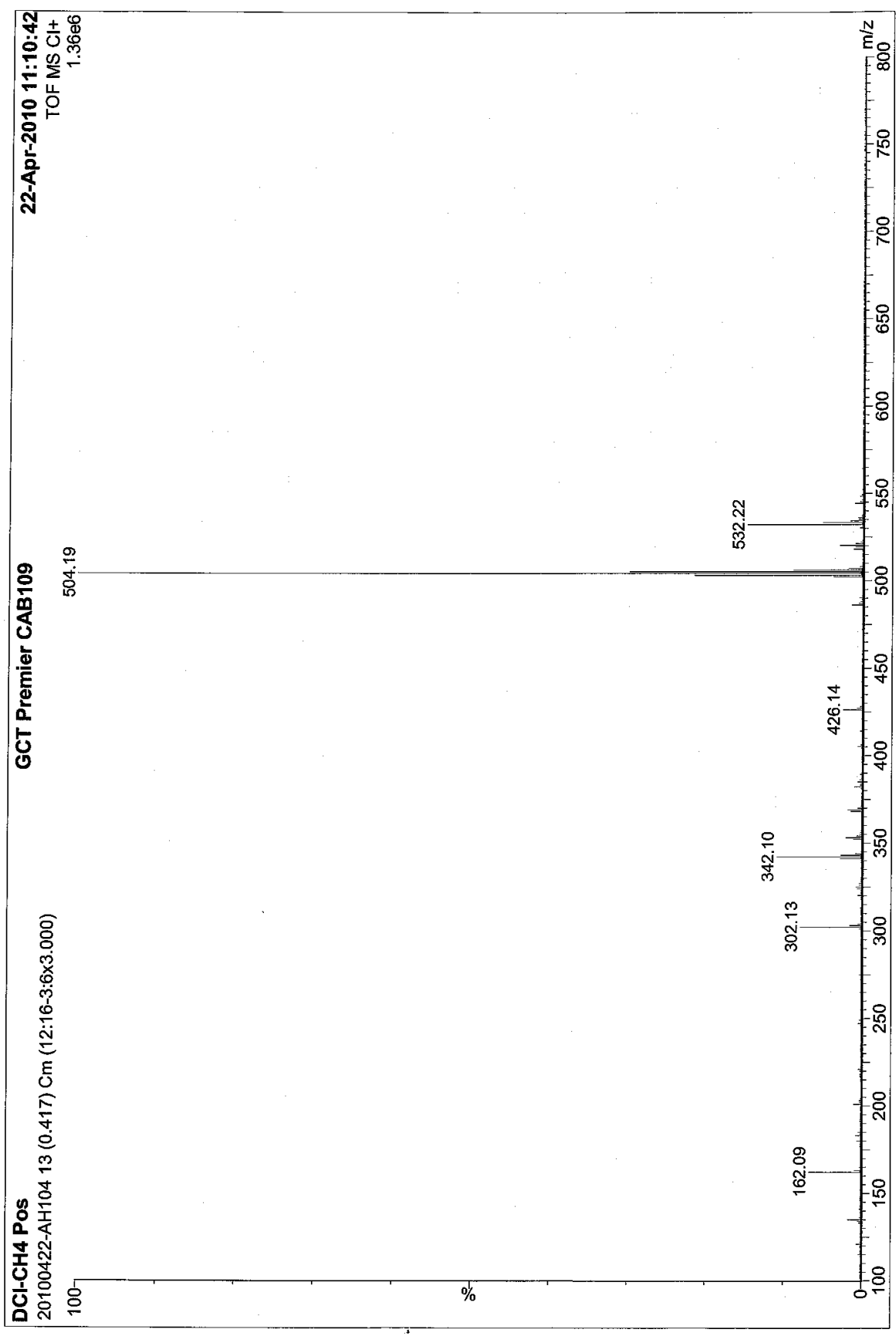
Compound 5-OMe

F31_DECOUPLE_H1
5-OMe

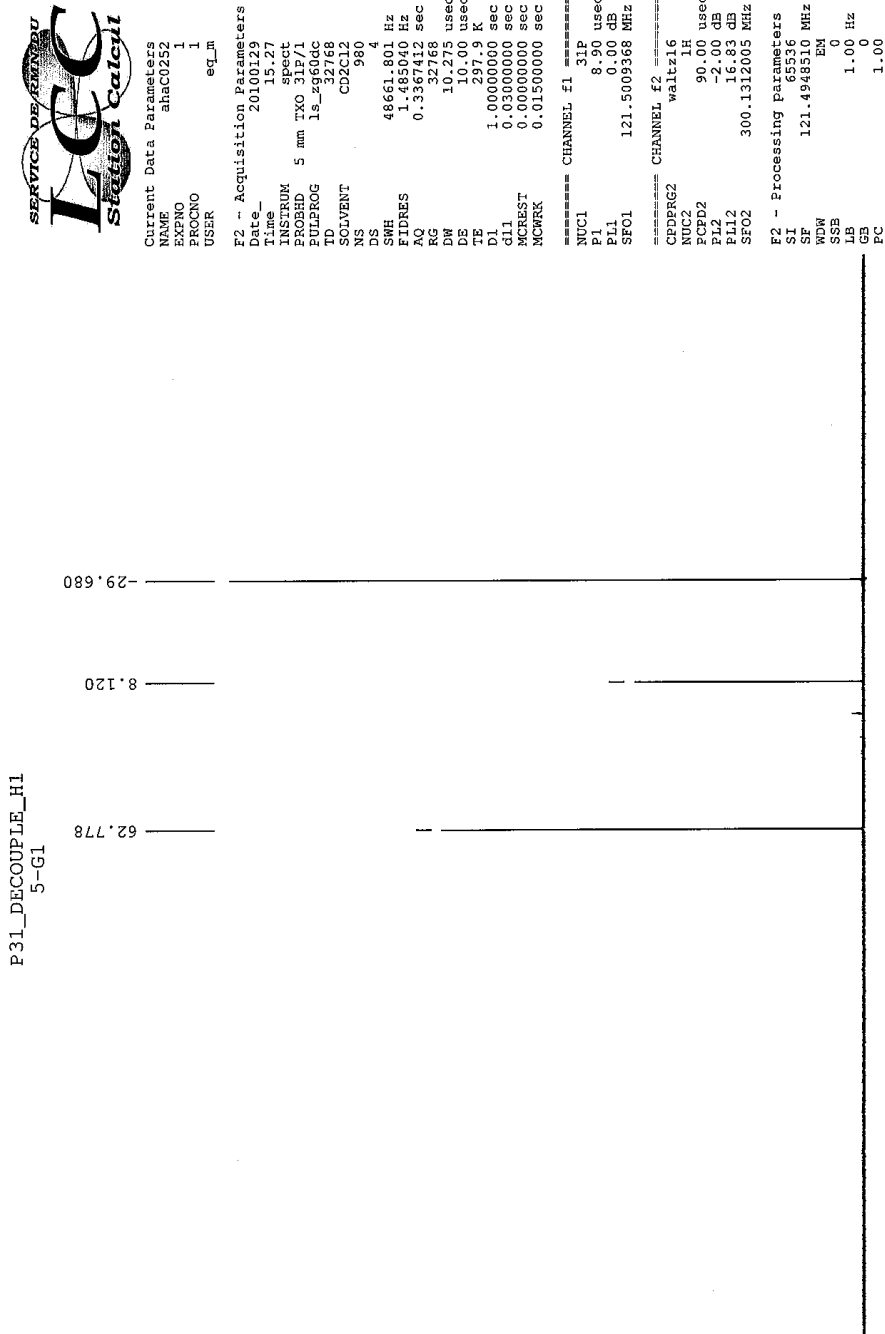
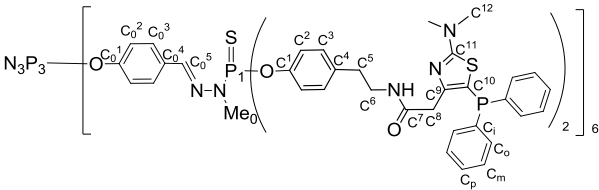


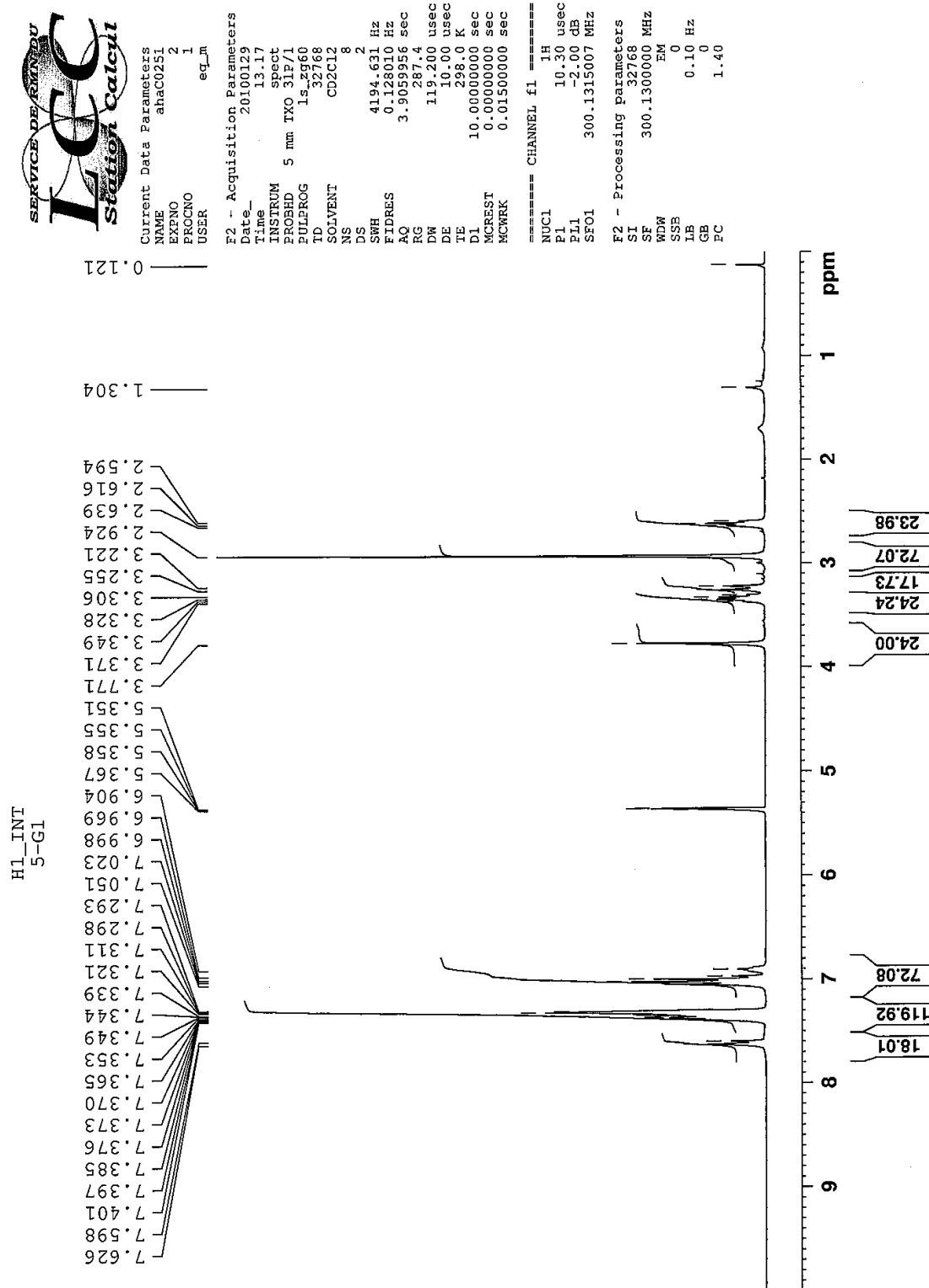




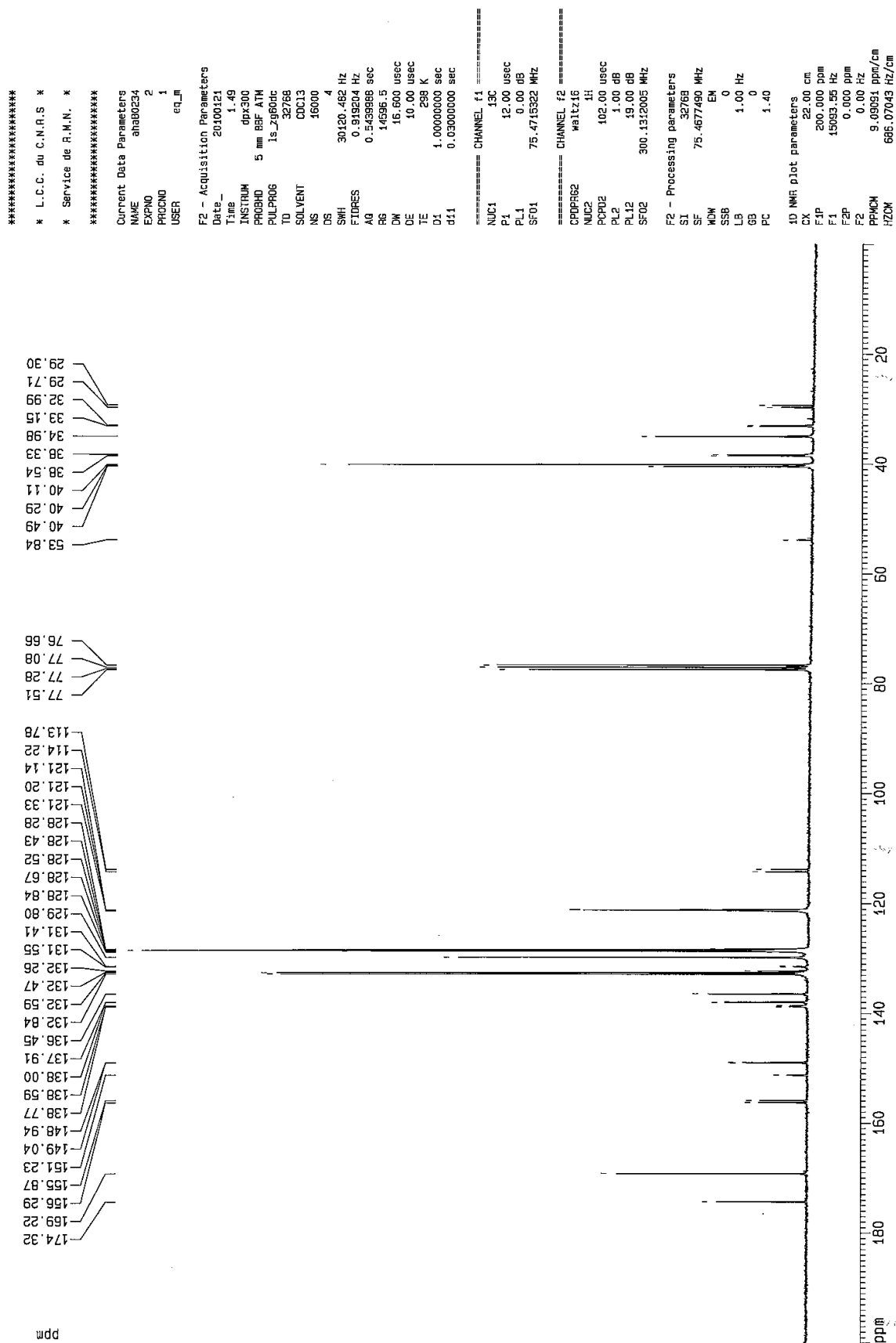


Compound 5-G₁ :

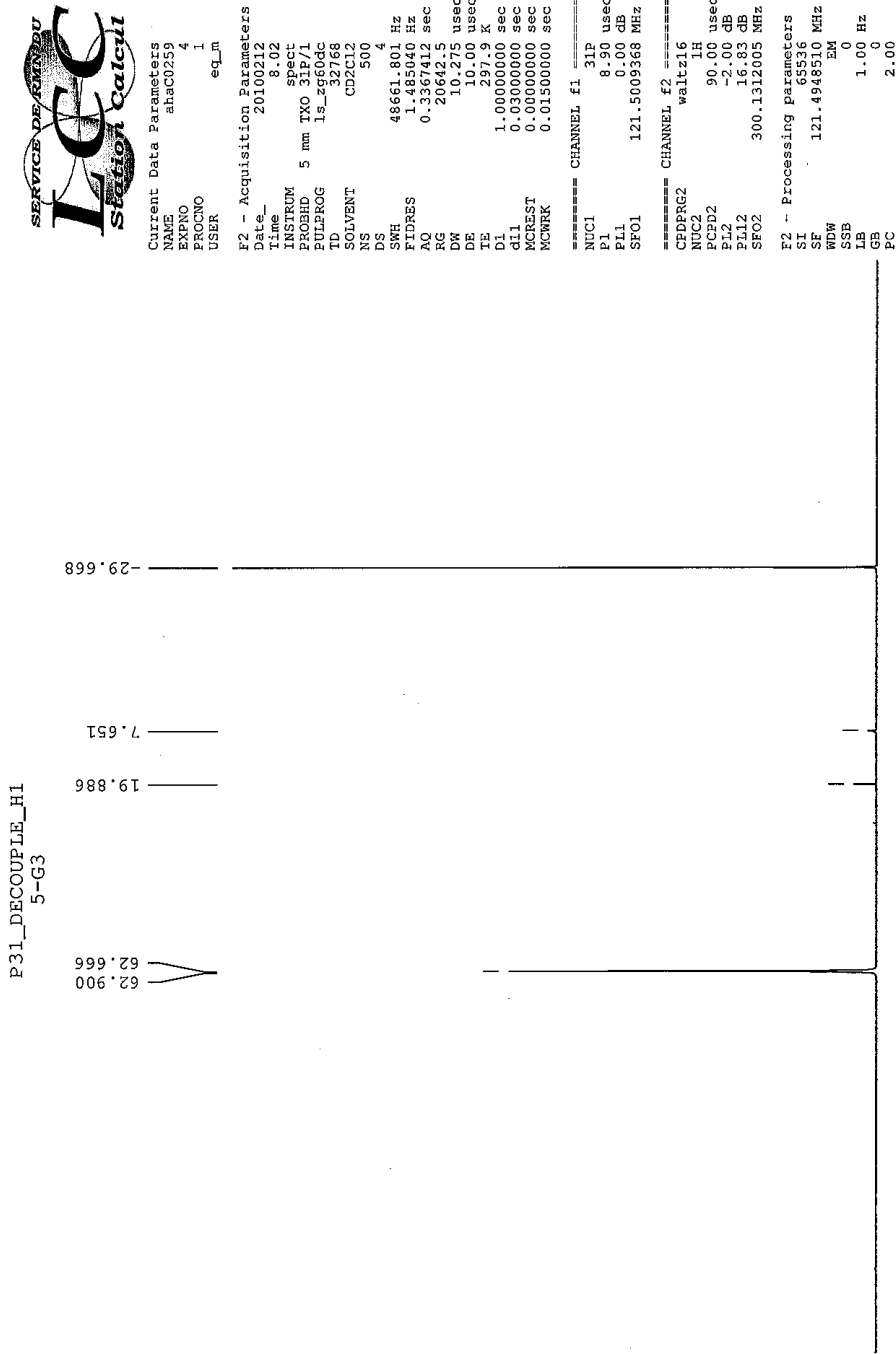
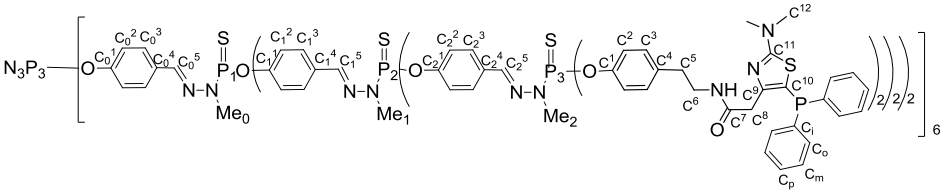


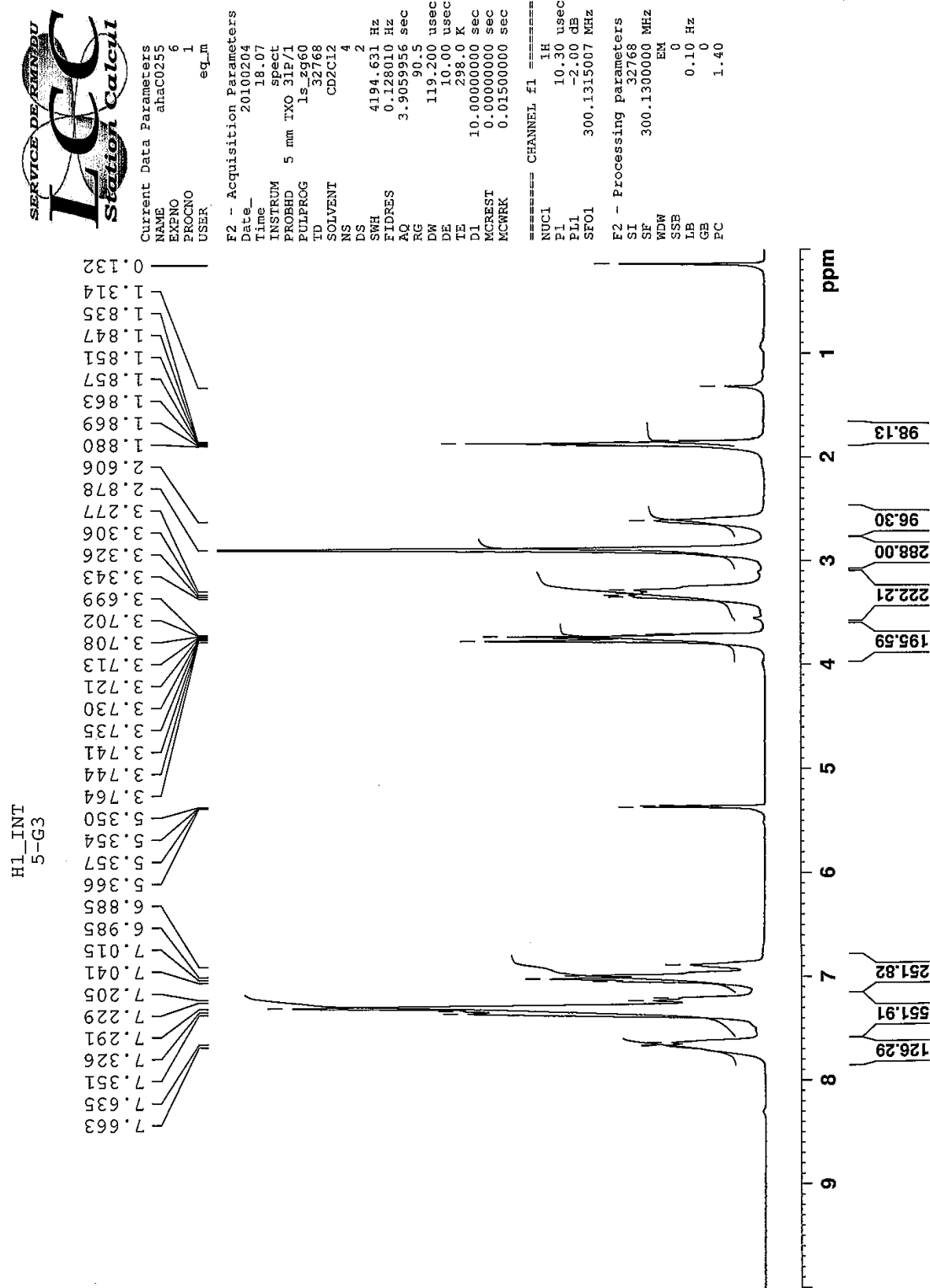


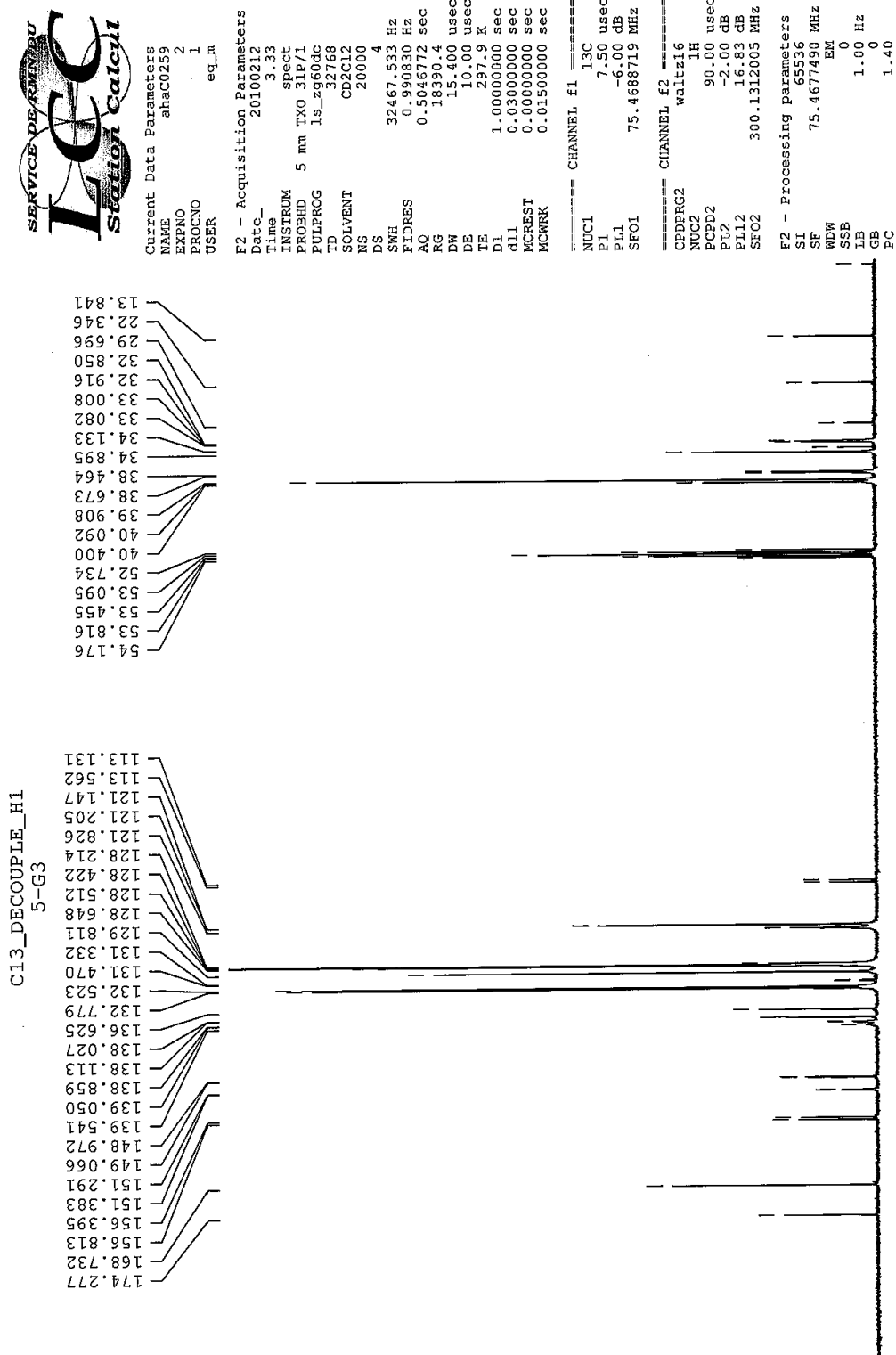
C13_DECOUPLE_H1
5-61

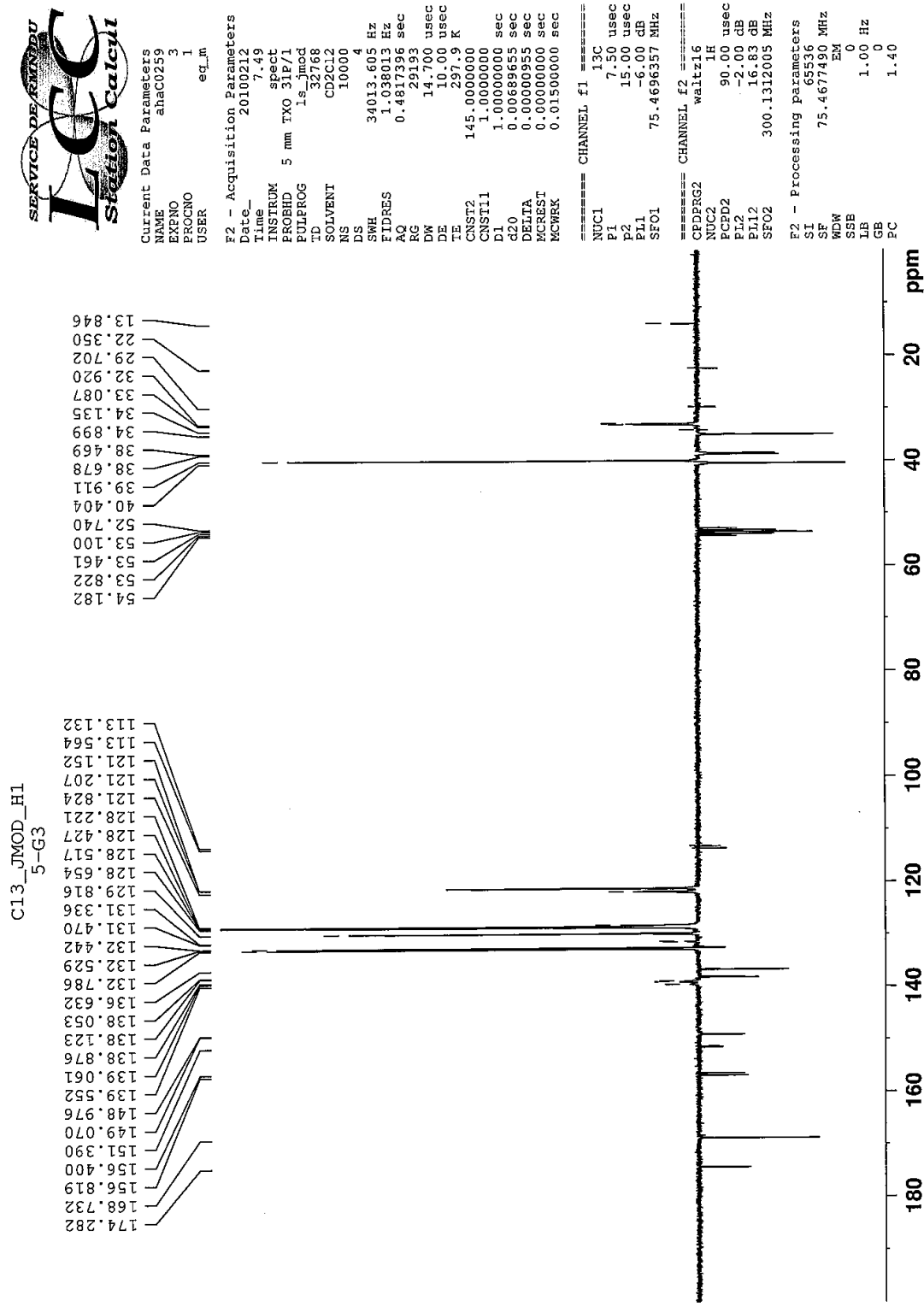


Compound 5-G₃ :







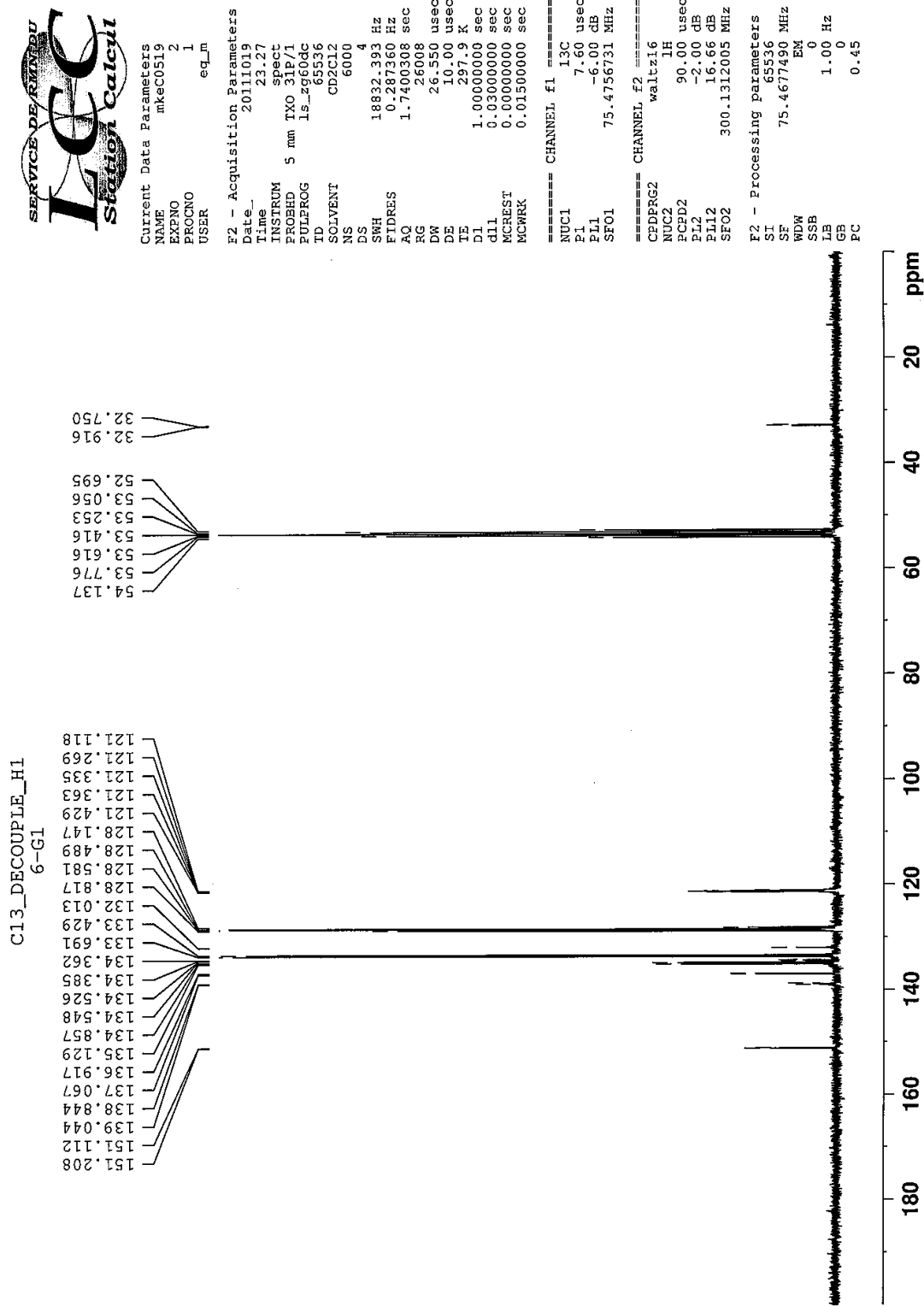


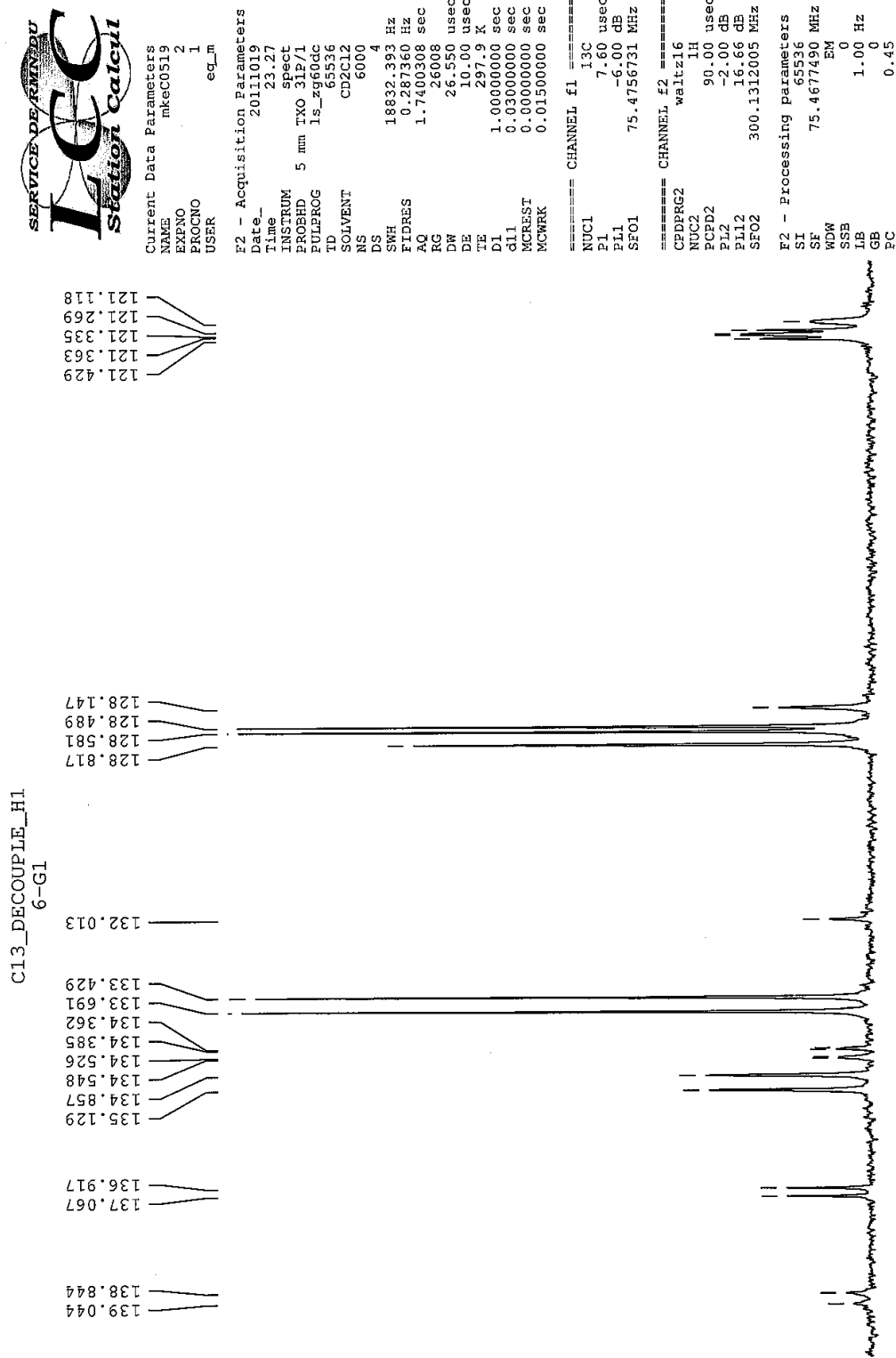
F2 - Acquisition Parameters

=====	CHANNEL f1	=====
NUC1	1H	
PI	10.60 usec	
PL1	-2.00 dB	
SFO1	300.1315007 MHz	

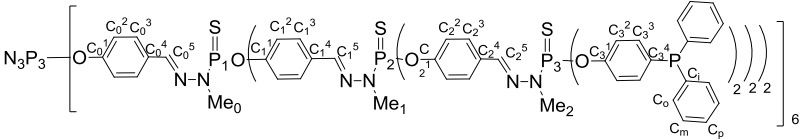
F2 - Processing parameters	
SI	32768
SF	300.130000 MHz
WDW	EM
SSB	0
LB	0.10 Hz
GB	0
PC	1.40



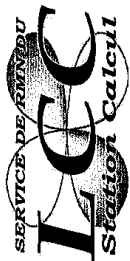




Compound 6-G₃ :



P31_DECOUPLE_H1
6-G3



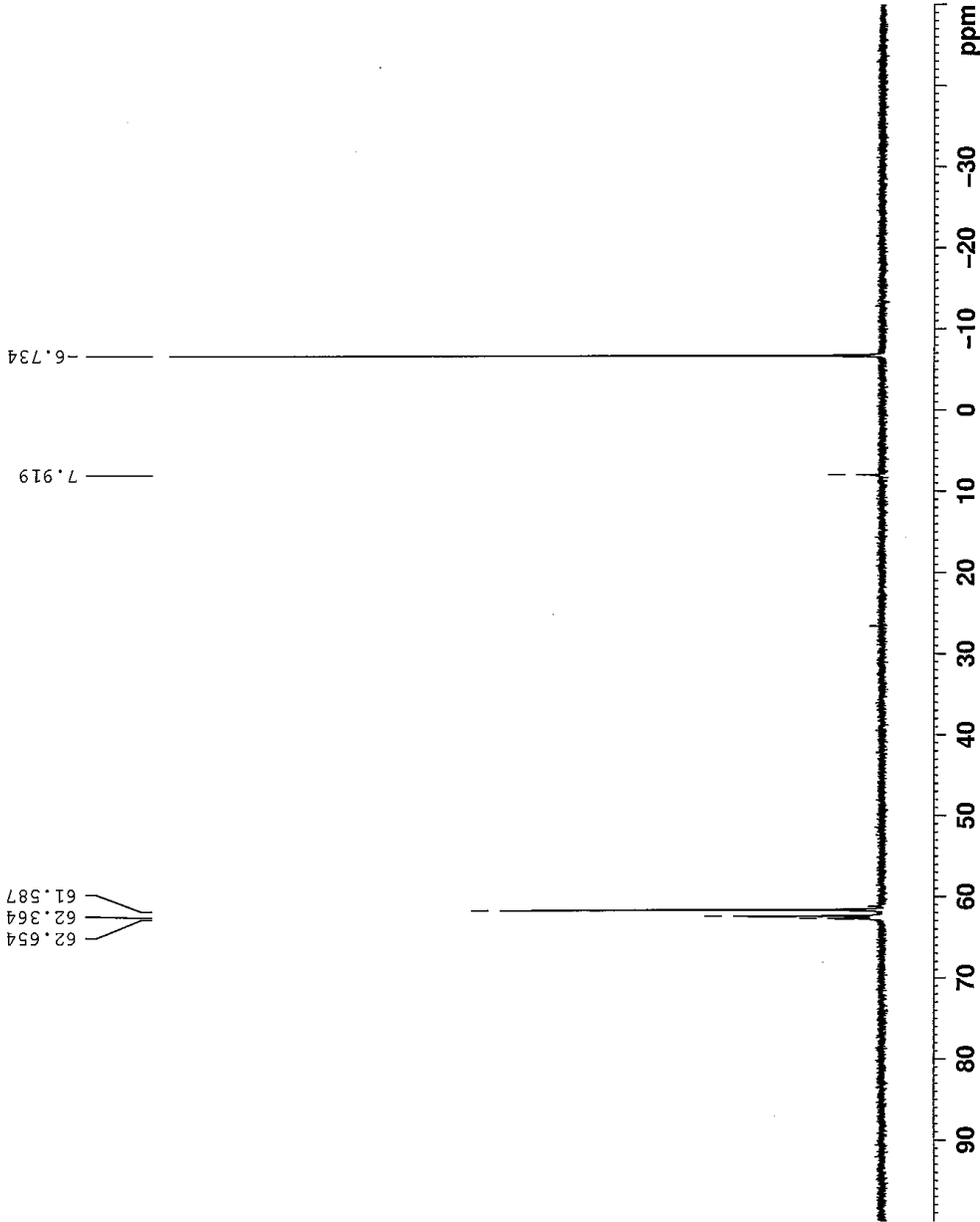
Current Data Parameters
NAME mkeC0535
EXPNO 2
PROCNO 1
USER eq_m

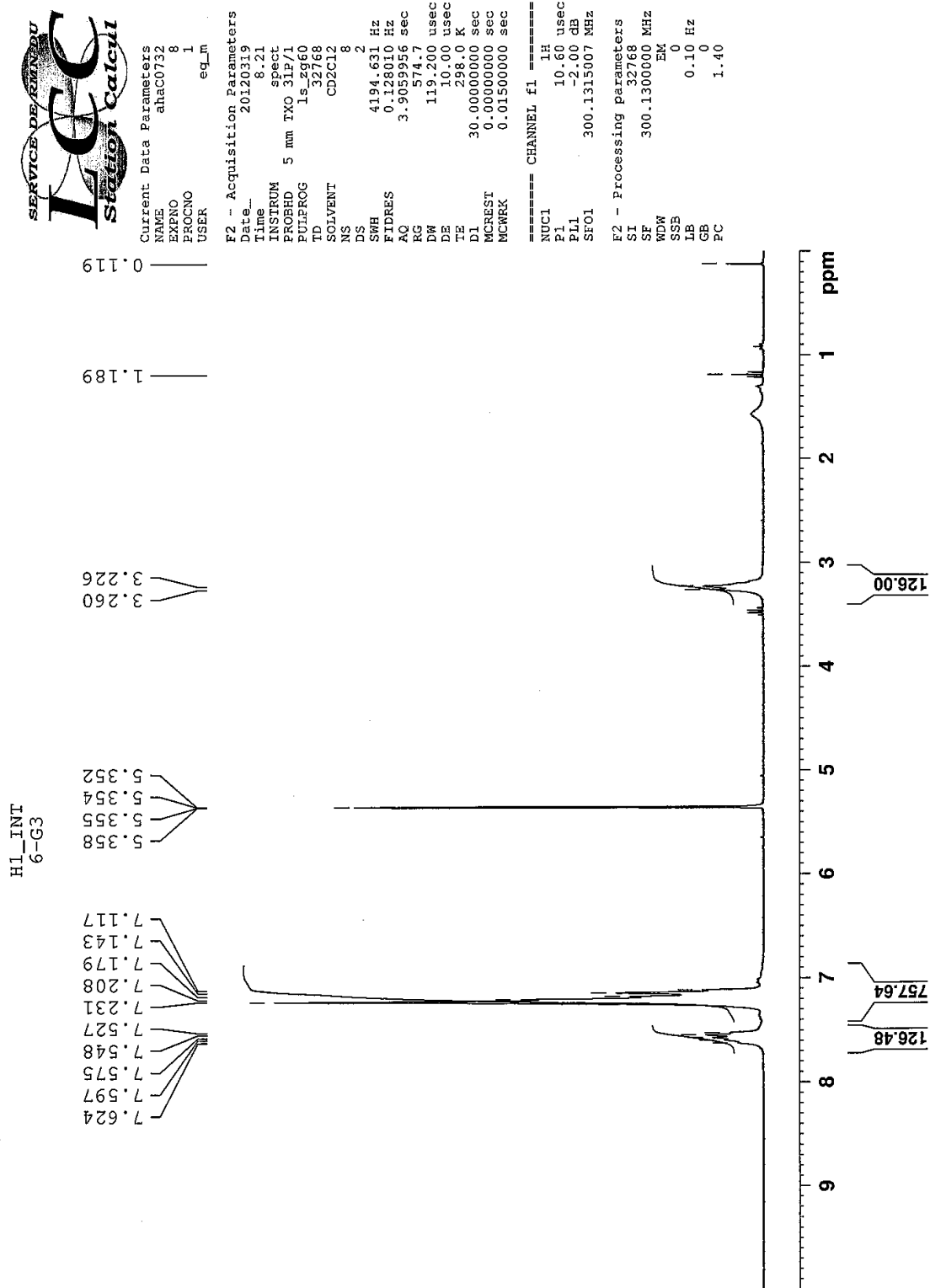
F2 - Acquisition Parameters
Date_ 20111108
Time 19.08
INSTRUM spect
PROBHD 5 mm TXO 31p/1
PULPROG ls_zgpg0dc
TD 65536
SOLVENT CD2Cl2
NS 32
DS 4
SWH 36496.352 Hz
FIDRES 0.556890 Hz
AQ 0.8978932 sec
RG 36780.8
DW 13.700 usec
DE 10.00 usec
TE 297.9 K
D1 1.00000000 sec
d11 0.03000000 sec
MCREST 0.00000000 sec
MCWEX 0.01500000 sec

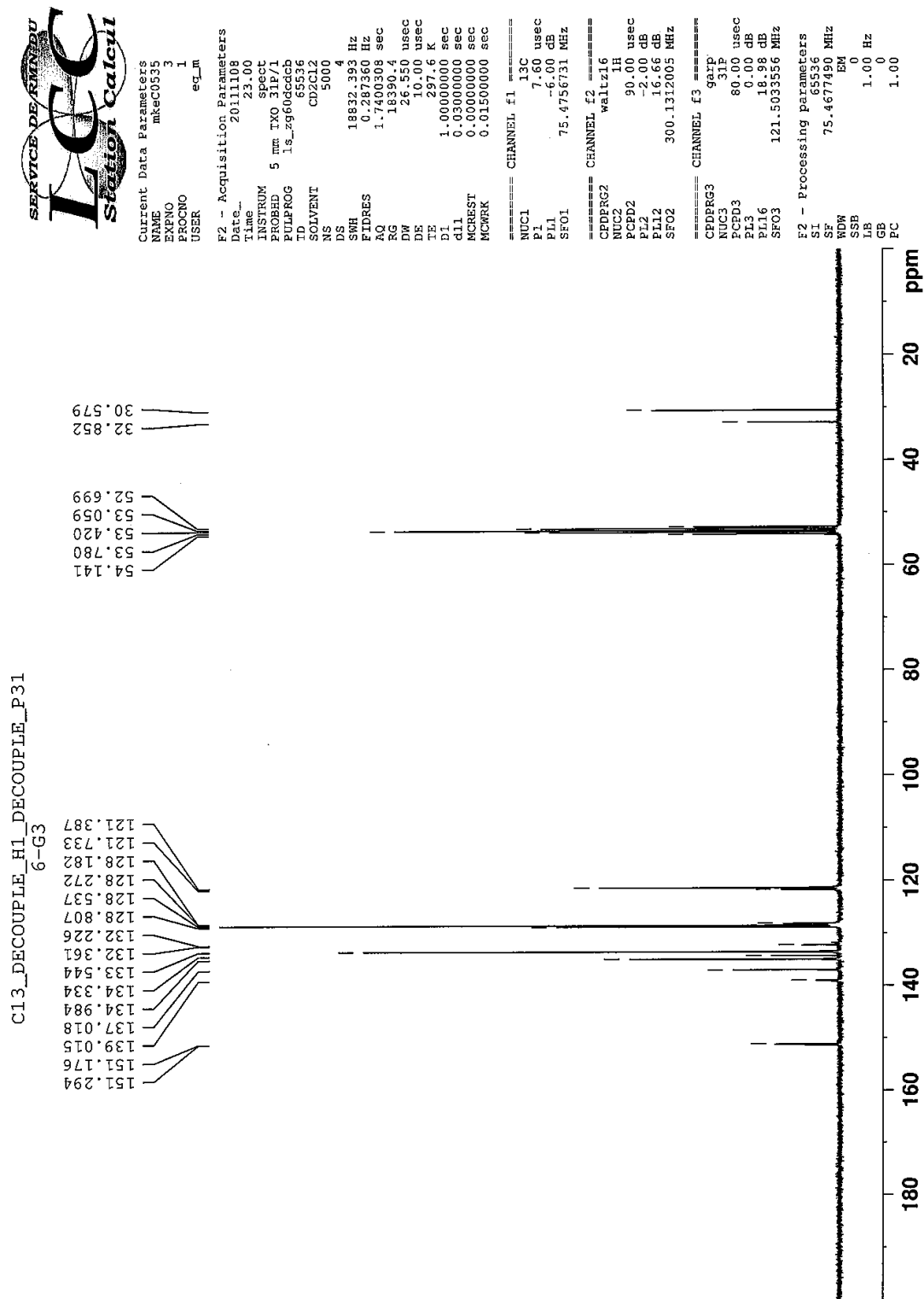
===== CHANNEL f1 =====
NUC1 31P
P1 9.00 usec
PL1 0.00 dB
SFO1 121.5070005 MHz

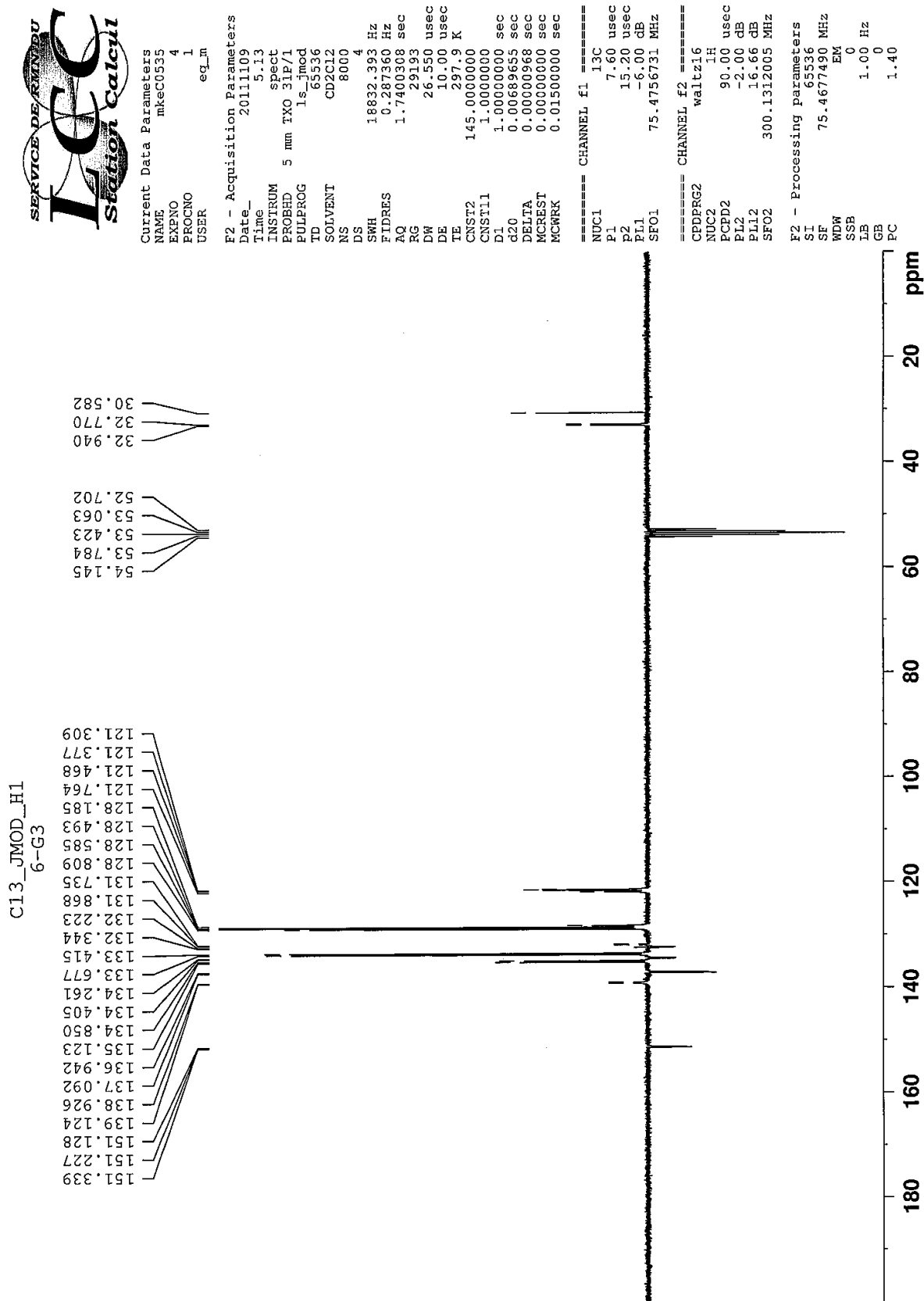
===== CHANNEL f2 =====
CPDPRG2 waltz16
NUC2 1H
PCPD2 90.00 usec
PL2 -2.00 dB
PL12 16.66 dB
SFO2 300.1312005 MHz

F2 - Processing parameters
SI 65536
SF 121.4948510 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
FC 1.40

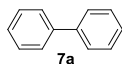








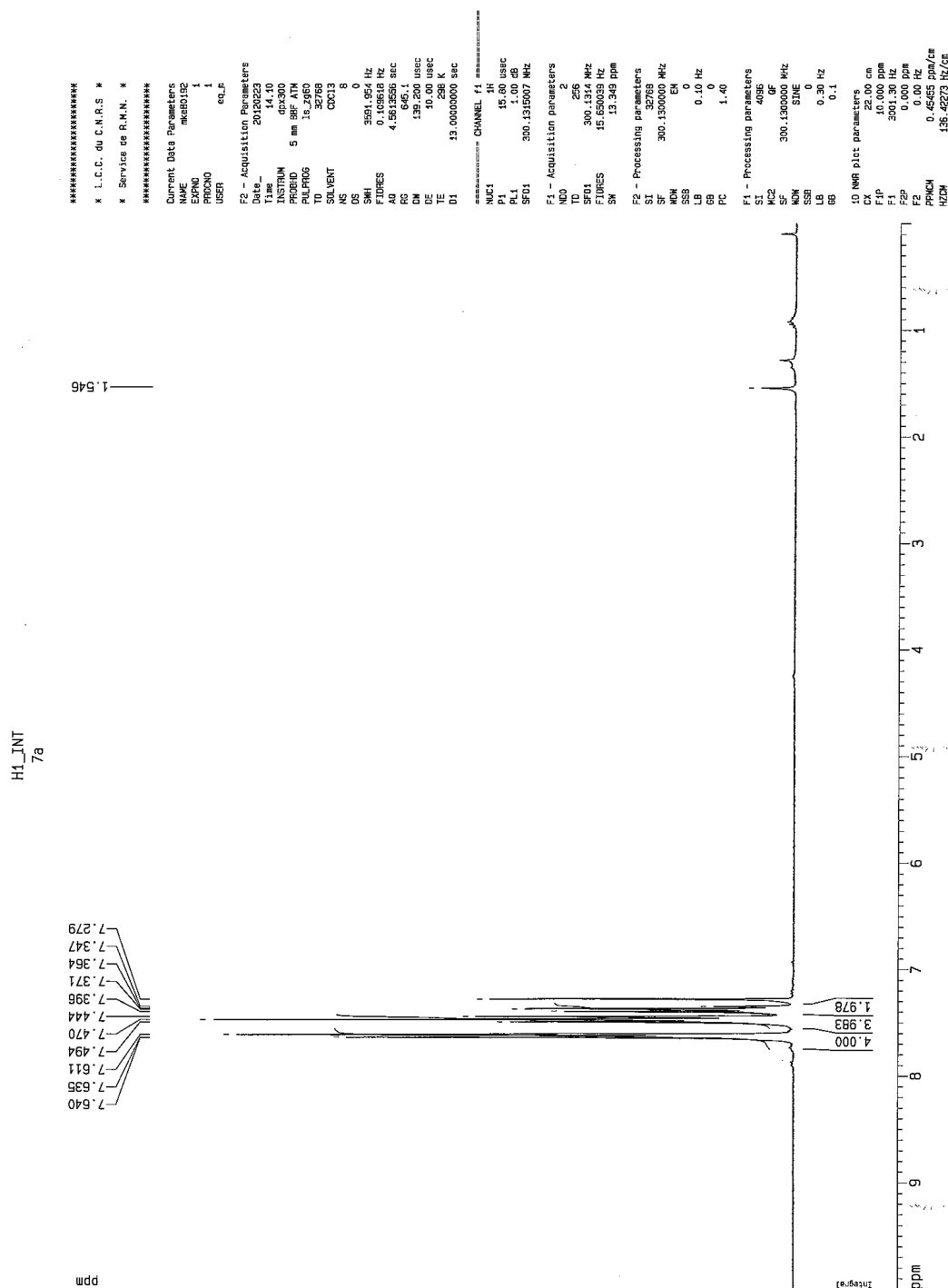
Compound 7a prepared by using **6-G₁** as the ligand:



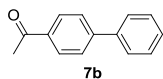
The standard procedure described above was applied by using Pd(OAc) (1,12 mg, 0.005 mmol), **6-G₁** (2.0 mg, 0.00042 mmol), THF (5 mL), water (2 mL), bromobenzene (105 µL, 1 mmol), phenyl boronic acid (139 mg, 1.14 mmol) and Na₂CO₃ (318 mg, 3 mmol). The filtrate obtained was purified by silica flash chromatography (Pentane) and biphenyl **7a** was obtained as a white powder in 82 % yield (127 mg). For purity: see spectra above.

¹H NMR (300 MHz, CDCl₃, 25°C) δ (ppm): 7.35-7.40 (m, 2H), 7.44-7.49 (m, 4H), 7.61-7.64 (m, 4H).

GC: rt = 15.82 min.

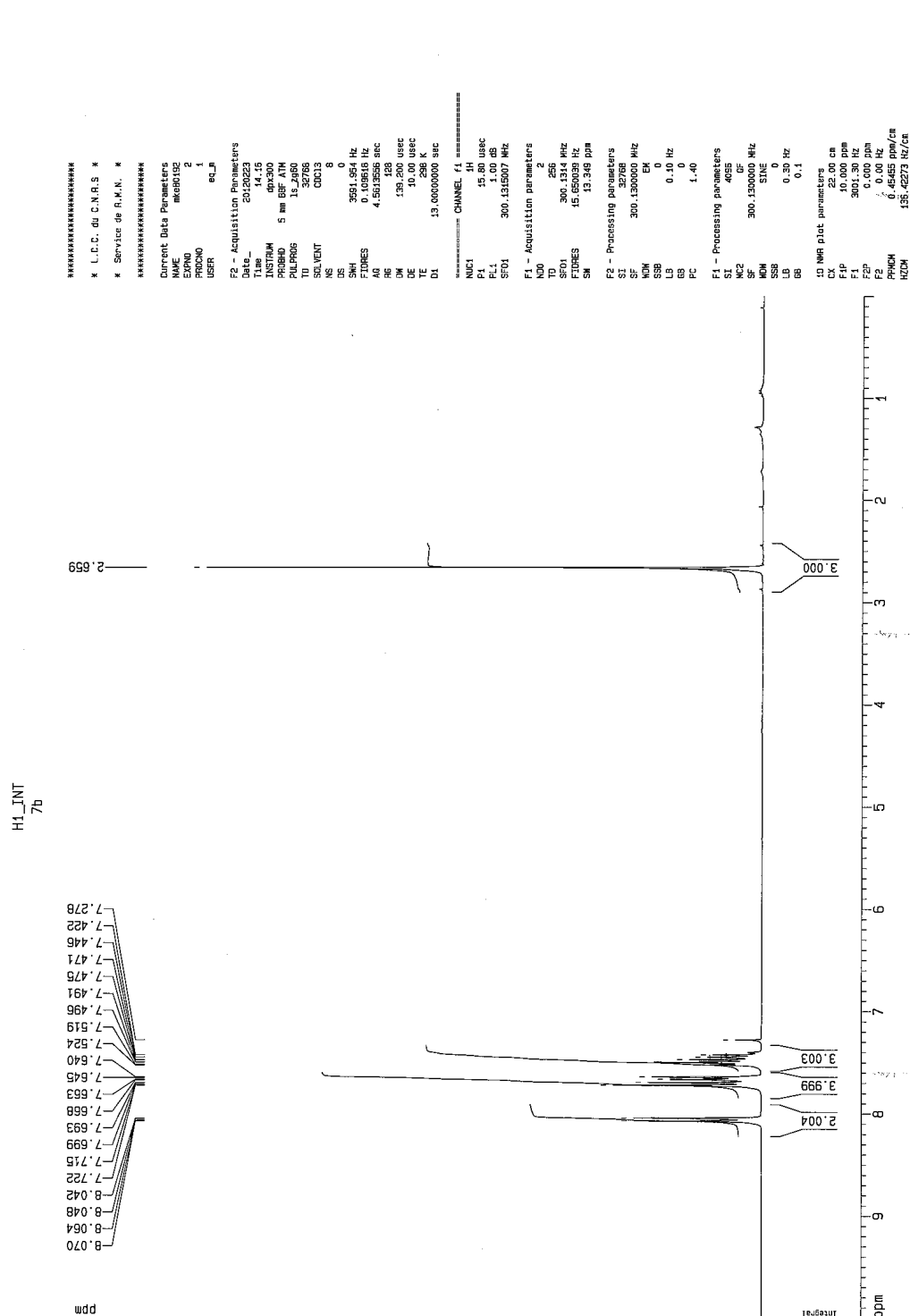


Compound 7b prepared by using **5-OMe** as the ligand:

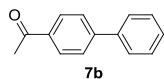


The standard procedure described above was applied by using Pd(OAc) (1.12 mg, 0.005 mmol), **5-OMe** (2.5 mg, 0.005 mmol), THF (5 mL), water (2 mL), 4-bromoacetophenone (199 mg, 1 mmol), phenyl boronic acid (139 mg, 1.14 mmol) and Na₂CO₃ (318 mg, 3 mmol). The filtrate obtained was purified by silica flash chromatography (Pentane/AcOEt) and **7b** was obtained as a white powder in 92 % yield (181 mg). For purity: see spectra above.

¹H NMR (300 MHz, CDCl₃, 25°C) δ (ppm): 2.66 (s, 3H, Me), 7.42-7.52 (m, 3H), 7.64-7.72 (m, 4H), 8.04-8.07 (m, 2H).
GC: rt = 21.50 min.

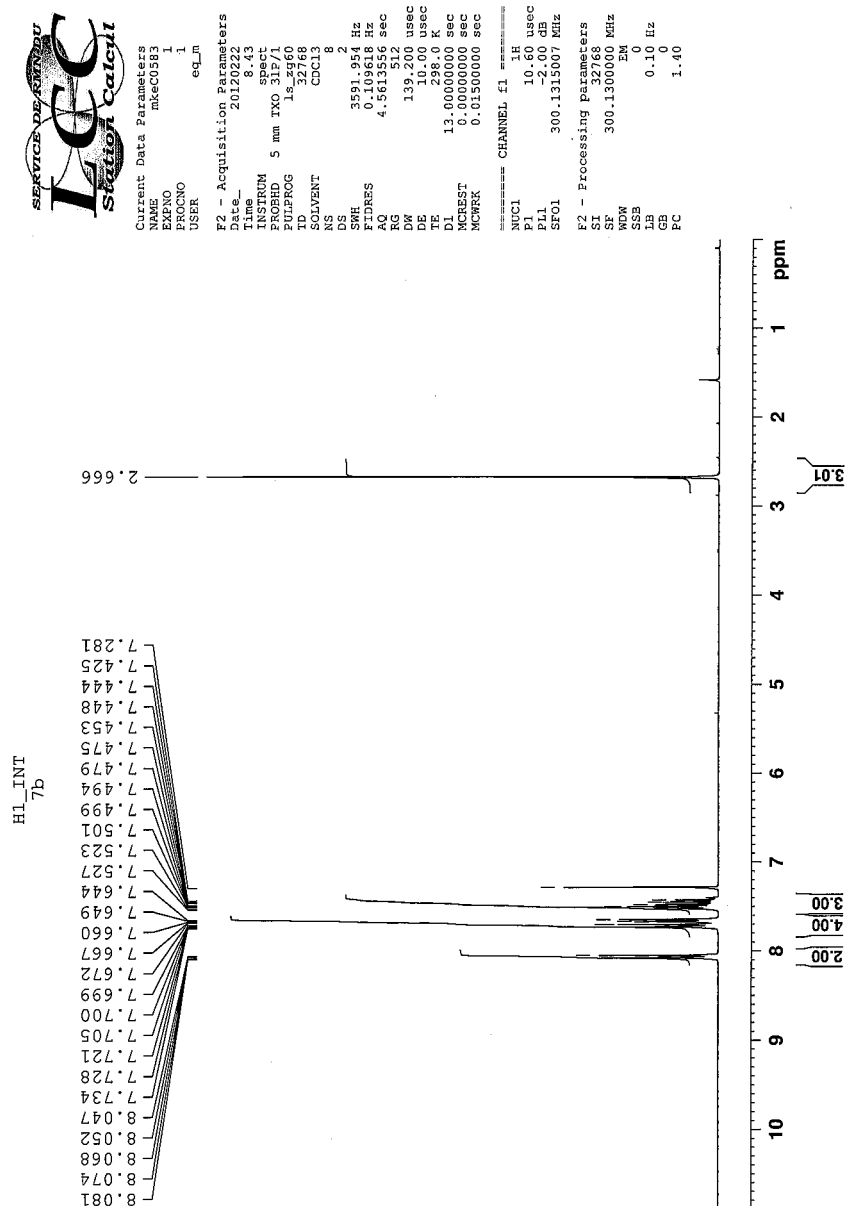


Compound 7b prepared by using **5-G₁** as the ligand:



The standard procedure described above was applied by using Pd(OAc) (1,12 mg, 0.005 mmol), **5-G₁** (3.0 mg, 0.00042 mmol), THF (5 mL), water (2 mL), 4-bromoacetophenone (199 mg, 1 mmol), phenyl boronic acid (139 mg, 1.14 mmol) and Na₂CO₃ (318 mg, 3 mmol). The filtrate obtained was purified by silica flash chromatography (Pentane/AcOEt) and **7b** was obtained as a white powder in 93 % yield (183 mg). For purity: see spectra above.

¹H NMR (300 MHz, CD₂Cl₂, 25°C) δ (ppm): 2.67 (s, 3H, Me), 7.43-7.53 (m, 3H), 7.64-7.73 (m, 4H), 8.05-8.08 (m, 2H).
GC: rt = 21.50 min.



3. RX data

Crystallographic data Diffraction data were collected at low temperature (180 K) on an Gemini Oxford Diffraction diffractometer for **5-OMe** and on a Bruker Kappa Apex II for **5-OH** and **3**, using a graphite-monochromated Mo K α radiation ($\lambda = 0.71073$ Å). Both diffractometers are equipped with an Oxford Cryosystems Cryostream cooler device. The structures were solved by direct methods with SIR92³ and all non-hydrogen atoms were refined anisotropically by means of least-squares procedures on F² with the aid of the program SHELXL-97.⁴

Compound 3

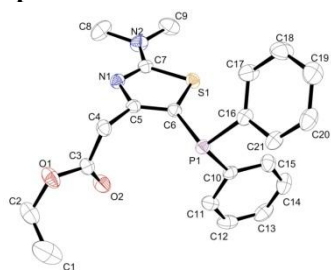


Table 1. Crystal data and structure refinement for **3**.

Identification code	3
Empirical formula	C ₂₁ H ₂₃ N ₂ O ₂ P S
Formula weight	398.44
Temperature	180 K
Wavelength	0.71073 Å
Crystal system, space group	triclinic, P -1
Unit cell dimensions	a = 8.7642(9) Å α = 94.396(5) deg. b = 10.8437(11) Å β = 106.808(5) deg. c = 12.1558(12) Å γ = 109.396(5) deg.
Volume	1023.87(19) Å ³
Z, Calculated density	2, 1.292 Mg/m ³
Absorption coefficient	0.254 mm ⁻¹
F(000)	420
Crystal size	0.45 x 0.3 x 0.125 mm
Theta range for data collection	2.03 to 25.35 deg.
Limiting indices	-10 ≤ h ≤ 10, -13 ≤ k ≤ 13, -14 ≤ l ≤ 14
Reflections collected / unique	20382 / 3739 [R(int) = 0.0186]
Completeness to theta = 25.35	99.9 %
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3739 / 0 / 247
Goodness-of-fit on F ²	1.047
Final R indices [I > 2 σ (I)]	R ₁ = 0.0324, wR ₂ = 0.0815
R indices (all data)	R ₁ = 0.0346, wR ₂ = 0.0834
Largest diff. peak and hole	0.714 and -0.305 e.Å ⁻³

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters (Å² $\times 10^3$) for **3**.
U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
S(1)	4481(1)	463(1)	1953(1)	26(1)
P(1)	5221(1)	3227(1)	3499(1)	23(1)
O(2)	8684(2)	4544(1)	2010(1)	38(1)
O(1)	11211(2)	5231(1)	3487(1)	41(1)
N(1)	7743(2)	1206(1)	2322(1)	26(1)
N(2)	6109(2)	-857(1)	1046(1)	35(1)
C(10)	3564(2)	3335(2)	2221(1)	24(1)
C(5)	7482(2)	2187(2)	2939(1)	23(1)
C(4)	9030(2)	3385(2)	3634(1)	27(1)

C(16)	3983(2)	2404(2)	4396(1)	26(1)
C(11)	4133(2)	4201(2)	1506(2)	31(1)
C(7)	6272(2)	233(2)	1751(1)	25(1)
C(15)	1815(2)	2638(2)	1936(1)	32(1)
C(6)	5841(2)	1993(1)	2877(1)	22(1)
C(3)	9566(2)	4430(2)	2928(1)	28(1)
C(21)	3366(2)	3181(2)	4986(1)	31(1)
C(9)	4479(3)	-1957(2)	548(2)	46(1)
C(20)	2517(2)	2680(2)	5749(2)	38(1)
C(12)	2975(3)	4350(2)	522(2)	38(1)
C(13)	1247(3)	3661(2)	256(2)	41(1)
C(19)	2266(3)	1398(2)	5939(2)	43(1)
C(18)	2901(3)	633(2)	5378(2)	45(1)
C(17)	3755(2)	1129(2)	4613(2)	36(1)
C(14)	665(2)	2805(2)	960(2)	41(1)
C(8)	7637(3)	-1063(2)	972(2)	47(1)
C(2)	11970(2)	6288(2)	2907(2)	45(1)
C(1)	11502(4)	7428(2)	3142(2)	74(1)

Table 3. Bond lengths [Å] and angles [deg] for **3**.

S(1)-C(7)	1.7488(16)
S(1)-C(6)	1.7503(15)
P(1)-C(6)	1.7943(15)
P(1)-C(16)	1.8308(16)
P(1)-C(10)	1.8362(16)
O(2)-C(3)	1.201(2)
O(1)-C(3)	1.3441(19)
O(1)-C(2)	1.464(2)
N(1)-C(7)	1.313(2)
N(1)-C(5)	1.3738(19)
N(2)-C(7)	1.347(2)
N(2)-C(9)	1.445(2)
N(2)-C(8)	1.455(2)
C(10)-C(15)	1.387(2)
C(10)-C(11)	1.396(2)
C(5)-C(6)	1.361(2)
C(5)-C(4)	1.497(2)
C(4)-C(3)	1.507(2)
C(4)-H(4A)	0.9700
C(4)-H(4B)	0.9700
C(16)-C(17)	1.387(2)
C(16)-C(21)	1.397(2)
C(11)-C(12)	1.387(3)
C(11)-H(11)	0.9300
C(15)-C(14)	1.385(2)
C(15)-H(15)	0.9300
C(21)-C(20)	1.384(2)
C(21)-H(21)	0.9300
C(9)-H(9A)	0.9600
C(9)-H(9B)	0.9600
C(9)-H(9C)	0.9600
C(20)-C(19)	1.382(3)
C(20)-H(20)	0.9300
C(12)-C(13)	1.373(3)
C(12)-H(12)	0.9300
C(13)-C(14)	1.383(3)
C(13)-H(13)	0.9300
C(19)-C(18)	1.378(3)
C(19)-H(19)	0.9300
C(18)-C(17)	1.388(3)
C(18)-H(18)	0.9300
C(17)-H(17)	0.9300
C(14)-H(14)	0.9300
C(8)-H(8A)	0.9600
C(8)-H(8B)	0.9600
C(8)-H(8C)	0.9600

C(2)-C(1)	1.460(3)
C(2)-H(2A)	0.9700
C(2)-H(2B)	0.9700
C(1)-H(1A)	0.9600
C(1)-H(1B)	0.9600
C(1)-H(1C)	0.9600
C(7)-S(1)-C(6)	89.19(7)
C(6)-P(1)-C(16)	105.74(7)
C(6)-P(1)-C(10)	101.62(7)
C(16)-P(1)-C(10)	103.00(7)
C(3)-O(1)-C(2)	116.76(13)
C(7)-N(1)-C(5)	110.12(13)
C(7)-N(2)-C(9)	121.04(15)
C(7)-N(2)-C(8)	119.69(15)
C(9)-N(2)-C(8)	118.06(15)
C(15)-C(10)-C(11)	118.70(15)
C(15)-C(10)-P(1)	124.68(12)
C(11)-C(10)-P(1)	116.62(12)
C(6)-C(5)-N(1)	117.63(13)
C(6)-C(5)-C(4)	125.08(14)
N(1)-C(5)-C(4)	117.29(13)
C(5)-C(4)-C(3)	113.67(13)
C(5)-C(4)-H(4A)	108.8
C(3)-C(4)-H(4A)	108.8
C(5)-C(4)-H(4B)	108.8
C(3)-C(4)-H(4B)	108.8
H(4A)-C(4)-H(4B)	107.7
C(17)-C(16)-C(21)	118.26(15)
C(17)-C(16)-P(1)	125.08(12)
C(21)-C(16)-P(1)	116.37(12)
C(12)-C(11)-C(10)	120.74(16)
C(12)-C(11)-H(11)	119.6
C(10)-C(11)-H(11)	119.6
N(1)-C(7)-N(2)	123.99(15)
N(1)-C(7)-S(1)	114.68(11)
N(2)-C(7)-S(1)	121.33(13)
C(14)-C(15)-C(10)	120.24(16)
C(14)-C(15)-H(15)	119.9
C(10)-C(15)-H(15)	119.9
C(5)-C(6)-S(1)	108.38(11)
C(5)-C(6)-P(1)	124.03(11)
S(1)-C(6)-P(1)	127.13(9)
O(2)-C(3)-O(1)	124.54(15)
O(2)-C(3)-C(4)	126.29(14)
O(1)-C(3)-C(4)	109.16(13)
C(20)-C(21)-C(16)	120.83(16)
C(20)-C(21)-H(21)	119.6
C(16)-C(21)-H(21)	119.6
N(2)-C(9)-H(9A)	109.5
N(2)-C(9)-H(9B)	109.5
H(9A)-C(9)-H(9B)	109.5
N(2)-C(9)-H(9C)	109.5
H(9A)-C(9)-H(9C)	109.5
H(9B)-C(9)-H(9C)	109.5
C(19)-C(20)-C(21)	120.33(16)
C(19)-C(20)-H(20)	119.8
C(21)-C(20)-H(20)	119.8
C(13)-C(12)-C(11)	119.87(16)
C(13)-C(12)-H(12)	120.1
C(11)-C(12)-H(12)	120.1
C(12)-C(13)-C(14)	120.01(16)
C(12)-C(13)-H(13)	120.0
C(14)-C(13)-H(13)	120.0
C(18)-C(19)-C(20)	119.28(17)
C(18)-C(19)-H(19)	120.4
C(20)-C(19)-H(19)	120.4
C(19)-C(18)-C(17)	120.70(18)
C(19)-C(18)-H(18)	119.6

C(17)-C(18)-H(18)	119.6
C(16)-C(17)-C(18)	120.58(16)
C(16)-C(17)-H(17)	119.7
C(18)-C(17)-H(17)	119.7
C(13)-C(14)-C(15)	120.44(17)
C(13)-C(14)-H(14)	119.8
C(15)-C(14)-H(14)	119.8
N(2)-C(8)-H(8A)	109.5
N(2)-C(8)-H(8B)	109.5
H(8A)-C(8)-H(8B)	109.5
N(2)-C(8)-H(8C)	109.5
H(8A)-C(8)-H(8C)	109.5
H(8B)-C(8)-H(8C)	109.5
C(1)-C(2)-O(1)	109.92(18)
C(1)-C(2)-H(2A)	109.7
O(1)-C(2)-H(2A)	109.7
C(1)-C(2)-H(2B)	109.7
O(1)-C(2)-H(2B)	109.7
H(2A)-C(2)-H(2B)	108.2
C(2)-C(1)-H(1A)	109.5
C(2)-C(1)-H(1B)	109.5
H(1A)-C(1)-H(1B)	109.5
C(2)-C(1)-H(1C)	109.5
H(1A)-C(1)-H(1C)	109.5
H(1B)-C(1)-H(1C)	109.5

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3**.
The anisotropic displacement factor exponent takes the form:
 $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
S(1)	25(1)	22(1)	28(1)	1(1)	11(1)	5(1)
P(1)	25(1)	22(1)	24(1)	2(1)	8(1)	10(1)
O(2)	30(1)	39(1)	34(1)	15(1)	3(1)	6(1)
O(1)	29(1)	40(1)	34(1)	9(1)	2(1)	-3(1)
N(1)	28(1)	29(1)	26(1)	7(1)	12(1)	14(1)
N(2)	45(1)	30(1)	34(1)	-1(1)	16(1)	18(1)
C(10)	30(1)	23(1)	24(1)	4(1)	12(1)	14(1)
C(5)	26(1)	23(1)	21(1)	7(1)	9(1)	10(1)
C(4)	23(1)	30(1)	25(1)	5(1)	5(1)	7(1)
C(16)	27(1)	32(1)	19(1)	4(1)	6(1)	14(1)
C(11)	40(1)	25(1)	31(1)	7(1)	15(1)	11(1)
C(7)	33(1)	26(1)	23(1)	8(1)	14(1)	14(1)
C(15)	31(1)	40(1)	28(1)	10(1)	14(1)	14(1)
C(6)	24(1)	21(1)	22(1)	4(1)	8(1)	8(1)
C(3)	24(1)	28(1)	27(1)	2(1)	7(1)	6(1)
C(21)	33(1)	37(1)	28(1)	5(1)	10(1)	19(1)
C(9)	62(1)	28(1)	43(1)	-2(1)	21(1)	10(1)
C(20)	41(1)	56(1)	29(1)	9(1)	16(1)	29(1)
C(12)	62(1)	31(1)	31(1)	12(1)	21(1)	23(1)
C(13)	53(1)	55(1)	26(1)	10(1)	10(1)	38(1)
C(19)	51(1)	62(1)	31(1)	20(1)	24(1)	28(1)
C(18)	67(1)	44(1)	40(1)	22(1)	30(1)	27(1)
C(17)	54(1)	37(1)	31(1)	12(1)	22(1)	25(1)
C(14)	31(1)	63(1)	33(1)	8(1)	10(1)	22(1)
C(8)	62(1)	58(1)	38(1)	2(1)	19(1)	41(1)
C(2)	36(1)	46(1)	40(1)	7(1)	10(1)	0(1)
C(1)	100(2)	47(1)	74(2)	7(1)	48(2)	9(1)

Compound 5-OH

Table 1. Crystal data and structure refinement for **5-OH**.

Identification code	5-OH
Empirical formula	C ₂₇ H ₂₈ N ₃ O ₂ P S
Formula weight	489.55
Temperature	180 K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, P 2 ₁ /c
Unit cell dimensions	a = 9.7499(4) Å alpha = 90 deg. b = 30.5632(13) Å beta = 101.967(2) deg. c = 8.5329(3) Å gamma = 90 deg.
Volume	2487.44(17) Å ³
Z, Calculated density	4, 1.307 Mg/m ³
Absorption coefficient	0.224 mm ⁻¹
F(000)	1032
Crystal size	0.375 x 0.05 x 0.05 mm
Theta range for data collection	1.33 to 28.80 deg.
Limiting indices	-12 ≤ h ≤ 13, -41 ≤ k ≤ 38, -11 ≤ l ≤ 11
Reflections collected / unique	33236 / 6476 [R(int) = 0.0421]
Completeness to theta = 28.80	99.7 %
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	6476 / 0 / 313
Goodness-of-fit on F ²	1.019
Final R indices [I > 2σ(I)]	R ₁ = 0.0413, wR ₂ = 0.0927
R indices (all data)	R ₁ = 0.0649, wR ₂ = 0.1032
Largest diff. peak and hole	0.341 and -0.231 e.Å ⁻³

Table 2. Atomic coordinates (× 10⁴) and equivalent isotropic displacement parameters (Å² × 10³) for **5-OH**.

U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
S(1)	10709(1)	822(1)	6131(1)	26(1)
P(1)	7587(1)	741(1)	6532(1)	21(1)
O(1)	15915(1)	2888(1)	8562(2)	44(1)
O(2)	7371(1)	2127(1)	9066(2)	37(1)
N(2)	11121(1)	1442(1)	8259(2)	23(1)
N(3)	13095(2)	1271(1)	7178(2)	31(1)
N(1)	9481(2)	2245(1)	8416(2)	25(1)
C(12)	9320(2)	970(1)	7034(2)	22(1)
C(9)	8506(2)	1992(1)	8845(2)	23(1)
C(22)	7347(2)	697(1)	4351(2)	23(1)
C(8)	9310(2)	2712(1)	8095(2)	29(1)
C(11)	9757(2)	1301(1)	8098(2)	21(1)
C(13)	11755(2)	1216(1)	7293(2)	23(1)
C(7)	10277(2)	2992(1)	9337(2)	32(1)
C(3)	12311(2)	3240(1)	8169(2)	32(1)
C(2)	13682(2)	3213(1)	7975(2)	35(1)
C(16)	7764(2)	173(1)	7223(2)	23(1)
C(5)	12719(2)	2661(1)	10061(2)	33(1)
C(1)	14579(2)	2903(1)	8822(2)	31(1)
C(23)	7055(2)	1087(1)	3491(2)	27(1)
C(10)	8859(2)	1510(1)	9124(2)	25(1)
C(27)	7415(2)	309(1)	3516(2)	29(1)

C(19)	7790(2)	-663(1)	8558(2)	33(1)
C(14)	13984(2)	1575(1)	8240(3)	42(1)
C(25)	6903(2)	700(1)	1017(2)	35(1)
C(4)	11796(2)	2963(1)	9208(2)	26(1)
C(6)	14098(2)	2630(1)	9885(2)	34(1)
C(17)	9011(2)	-31(1)	7891(3)	43(1)
C(18)	9018(2)	-448(1)	8551(3)	49(1)
C(26)	7189(2)	312(1)	1855(2)	35(1)
C(24)	6849(2)	1088(1)	1837(2)	32(1)
C(21)	6526(2)	-53(1)	7213(2)	33(1)
C(15)	13721(2)	986(1)	6165(3)	42(1)
C(20)	6538(2)	-469(1)	7866(2)	37(1)

Table 3. Bond lengths [Å] and angles [deg] for **5-OH**.

S(1)-C(13)	1.7470(16)
S(1)-C(12)	1.7497(16)
P(1)-C(12)	1.7962(16)
P(1)-C(16)	1.8308(16)
P(1)-C(22)	1.8326(16)
O(1)-C(1)	1.368(2)
O(1)-H(1)	0.8400
O(2)-C(9)	1.231(2)
N(2)-C(13)	1.324(2)
N(2)-C(11)	1.378(2)
N(3)-C(13)	1.341(2)
N(3)-C(15)	1.447(2)
N(3)-C(14)	1.452(2)
N(1)-C(9)	1.334(2)
N(1)-C(8)	1.457(2)
N(1)-H(100)	0.82(2)
C(12)-C(11)	1.367(2)
C(9)-C(10)	1.522(2)
C(22)-C(27)	1.391(2)
C(22)-C(23)	1.398(2)
C(8)-C(7)	1.527(2)
C(8)-H(8A)	0.9900
C(8)-H(8B)	0.9900
C(11)-C(10)	1.503(2)
C(7)-C(4)	1.509(2)
C(7)-H(7A)	0.9900
C(7)-H(7B)	0.9900
C(3)-C(2)	1.383(3)
C(3)-C(4)	1.393(2)
C(3)-H(3)	0.9500
C(2)-C(1)	1.387(3)
C(2)-H(2)	0.9500
C(16)-C(17)	1.380(2)
C(16)-C(21)	1.390(2)
C(5)-C(4)	1.386(2)
C(5)-C(6)	1.387(3)
C(5)-H(5)	0.9500
C(1)-C(6)	1.384(3)
C(23)-C(24)	1.384(2)
C(23)-H(23)	0.9500
C(10)-H(10A)	0.9900
C(10)-H(10B)	0.9900
C(27)-C(26)	1.389(2)
C(27)-H(27)	0.9500
C(19)-C(18)	1.367(3)

C(19)-C(20)	1.376(3)
C(19)-H(19)	0.9500
C(14)-H(14A)	0.9800
C(14)-H(14B)	0.9800
C(14)-H(14C)	0.9800
C(25)-C(24)	1.383(3)
C(25)-C(26)	1.384(3)
C(25)-H(25)	0.9500
C(6)-H(6)	0.9500
C(17)-C(18)	1.393(3)
C(17)-H(17)	0.9500
C(18)-H(18)	0.9500
C(26)-H(26)	0.9500
C(24)-H(24)	0.9500
C(21)-C(20)	1.387(2)
C(21)-H(21)	0.9500
C(15)-H(15A)	0.9800
C(15)-H(15B)	0.9800
C(15)-H(15C)	0.9800
C(20)-H(20)	0.9500
C(13)-S(1)-C(12)	89.29(8)
C(12)-P(1)-C(16)	105.69(7)
C(12)-P(1)-C(22)	100.74(7)
C(16)-P(1)-C(22)	103.91(7)
C(1)-O(1)-H(1)	109.5
C(13)-N(2)-C(11)	109.84(13)
C(13)-N(3)-C(15)	120.45(15)
C(13)-N(3)-C(14)	119.93(15)
C(15)-N(3)-C(14)	119.05(15)
C(9)-N(1)-C(8)	124.02(15)
C(9)-N(1)-H(100)	115.7(13)
C(8)-N(1)-H(100)	120.2(13)
C(11)-C(12)-S(1)	108.60(12)
C(11)-C(12)-P(1)	126.49(12)
S(1)-C(12)-P(1)	124.81(9)
O(2)-C(9)-N(1)	124.22(15)
O(2)-C(9)-C(10)	118.82(14)
N(1)-C(9)-C(10)	116.93(14)
C(27)-C(22)-C(23)	118.96(15)
C(27)-C(22)-P(1)	125.01(12)
C(23)-C(22)-P(1)	116.02(12)
N(1)-C(8)-C(7)	112.85(14)
N(1)-C(8)-H(8A)	109.0
C(7)-C(8)-H(8A)	109.0
N(1)-C(8)-H(8B)	109.0
C(7)-C(8)-H(8B)	109.0
H(8A)-C(8)-H(8B)	107.8
C(12)-C(11)-N(2)	117.46(14)
C(12)-C(11)-C(10)	124.20(14)
N(2)-C(11)-C(10)	118.32(14)
N(2)-C(13)-N(3)	124.67(15)
N(2)-C(13)-S(1)	114.80(12)
N(3)-C(13)-S(1)	120.53(12)
C(4)-C(7)-C(8)	113.00(14)
C(4)-C(7)-H(7A)	109.0
C(8)-C(7)-H(7A)	109.0
C(4)-C(7)-H(7B)	109.0
C(8)-C(7)-H(7B)	109.0
H(7A)-C(7)-H(7B)	107.8
C(2)-C(3)-C(4)	121.70(17)
C(2)-C(3)-H(3)	119.1

C(4)-C(3)-H(3)	119.1
C(3)-C(2)-C(1)	119.90(16)
C(3)-C(2)-H(2)	120.0
C(1)-C(2)-H(2)	120.0
C(17)-C(16)-C(21)	117.89(15)
C(17)-C(16)-P(1)	125.31(13)
C(21)-C(16)-P(1)	116.48(12)
C(4)-C(5)-C(6)	121.81(16)
C(4)-C(5)-H(5)	119.1
C(6)-C(5)-H(5)	119.1
O(1)-C(1)-C(6)	123.29(16)
O(1)-C(1)-C(2)	117.26(16)
C(6)-C(1)-C(2)	119.44(17)
C(24)-C(23)-C(22)	120.55(16)
C(24)-C(23)-H(23)	119.7
C(22)-C(23)-H(23)	119.7
C(11)-C(10)-C(9)	117.49(13)
C(11)-C(10)-H(10A)	107.9
C(9)-C(10)-H(10A)	107.9
C(11)-C(10)-H(10B)	107.9
C(9)-C(10)-H(10B)	107.9
H(10A)-C(10)-H(10B)	107.2
C(26)-C(27)-C(22)	120.20(16)
C(26)-C(27)-H(27)	119.9
C(22)-C(27)-H(27)	119.9
C(18)-C(19)-C(20)	119.43(16)
C(18)-C(19)-H(19)	120.3
C(20)-C(19)-H(19)	120.3
N(3)-C(14)-H(14A)	109.5
N(3)-C(14)-H(14B)	109.5
H(14A)-C(14)-H(14B)	109.5
N(3)-C(14)-H(14C)	109.5
H(14A)-C(14)-H(14C)	109.5
H(14B)-C(14)-H(14C)	109.5
C(24)-C(25)-C(26)	119.87(17)
C(24)-C(25)-H(25)	120.1
C(26)-C(25)-H(25)	120.1
C(5)-C(4)-C(3)	117.29(16)
C(5)-C(4)-C(7)	122.75(16)
C(3)-C(4)-C(7)	119.94(16)
C(1)-C(6)-C(5)	119.82(17)
C(1)-C(6)-H(6)	120.1
C(5)-C(6)-H(6)	120.1
C(16)-C(17)-C(18)	120.72(18)
C(16)-C(17)-H(17)	119.6
C(18)-C(17)-H(17)	119.6
C(19)-C(18)-C(17)	120.68(18)
C(19)-C(18)-H(18)	119.7
C(17)-C(18)-H(18)	119.7
C(25)-C(26)-C(27)	120.35(17)
C(25)-C(26)-H(26)	119.8
C(27)-C(26)-H(26)	119.8
C(25)-C(24)-C(23)	120.05(16)
C(25)-C(24)-H(24)	120.0
C(23)-C(24)-H(24)	120.0
C(20)-C(21)-C(16)	121.22(17)
C(20)-C(21)-H(21)	119.4
C(16)-C(21)-H(21)	119.4
N(3)-C(15)-H(15A)	109.5
N(3)-C(15)-H(15B)	109.5
H(15A)-C(15)-H(15B)	109.5
N(3)-C(15)-H(15C)	109.5

H(15A)-C(15)-H(15C)	109.5
H(15B)-C(15)-H(15C)	109.5
C(19)-C(20)-C(21)	120.03(17)
C(19)-C(20)-H(20)	120.0
C(21)-C(20)-H(20)	120.0

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **5-OH**
The anisotropic displacement factor exponent takes the form:
 $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
S(1)	21(1)	20(1)	38(1)	-6(1)	9(1)	0(1)
P(1)	19(1)	17(1)	29(1)	0(1)	5(1)	0(1)
O(1)	29(1)	26(1)	80(1)	2(1)	18(1)	0(1)
O(2)	28(1)	28(1)	60(1)	-2(1)	21(1)	1(1)
N(2)	21(1)	20(1)	28(1)	1(1)	4(1)	-2(1)
N(3)	18(1)	26(1)	47(1)	-2(1)	6(1)	-2(1)
N(1)	20(1)	19(1)	39(1)	2(1)	10(1)	1(1)
C(12)	19(1)	18(1)	29(1)	0(1)	7(1)	1(1)
C(9)	23(1)	22(1)	24(1)	-4(1)	6(1)	-2(1)
C(22)	16(1)	22(1)	29(1)	1(1)	5(1)	0(1)
C(8)	25(1)	20(1)	43(1)	5(1)	11(1)	2(1)
C(11)	23(1)	16(1)	25(1)	3(1)	6(1)	0(1)
C(13)	20(1)	18(1)	32(1)	4(1)	3(1)	1(1)
C(7)	36(1)	20(1)	44(1)	-5(1)	19(1)	-1(1)
C(3)	30(1)	22(1)	44(1)	6(1)	9(1)	1(1)
C(2)	34(1)	24(1)	49(1)	7(1)	16(1)	-3(1)
C(16)	24(1)	19(1)	25(1)	-1(1)	5(1)	-2(1)
C(5)	39(1)	26(1)	35(1)	3(1)	9(1)	-3(1)
C(1)	24(1)	19(1)	49(1)	-5(1)	10(1)	-2(1)
C(23)	23(1)	21(1)	36(1)	2(1)	6(1)	0(1)
C(10)	30(1)	19(1)	29(1)	-2(1)	11(1)	-4(1)
C(27)	29(1)	23(1)	35(1)	0(1)	8(1)	3(1)
C(19)	45(1)	20(1)	32(1)	3(1)	9(1)	-3(1)
C(14)	26(1)	46(1)	52(1)	0(1)	1(1)	-12(1)
C(25)	34(1)	45(1)	29(1)	3(1)	12(1)	-2(1)
C(4)	30(1)	17(1)	33(1)	-6(1)	9(1)	-4(1)
C(6)	33(1)	23(1)	43(1)	2(1)	3(1)	1(1)
C(17)	26(1)	30(1)	69(1)	17(1)	1(1)	-3(1)
C(18)	36(1)	32(1)	72(2)	20(1)	-5(1)	3(1)
C(26)	38(1)	32(1)	38(1)	-9(1)	14(1)	0(1)
C(24)	27(1)	32(1)	37(1)	10(1)	9(1)	-1(1)
C(21)	23(1)	24(1)	53(1)	5(1)	8(1)	0(1)
C(15)	26(1)	37(1)	68(1)	-1(1)	20(1)	5(1)
C(20)	32(1)	25(1)	56(1)	2(1)	16(1)	-6(1)

Compound 5-OMe

Table 1. Crystal data and structure refinement for **5-OMe**.

Identification code	5-OMe
Empirical formula	C ₂₈ H ₃₀ N ₃ O ₂ P S
Formula weight	503.59
Temperature	180(2) K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, P 2 ₁ /c
Unit cell dimensions	a = 16.0609(10) Å alpha = 90 deg. b = 18.7305(13) Å beta = 98.481(6) deg. c = 8.9676(6) Å gamma = 90 deg.
Volume	2668.2(3) Å ³
Z, Calculated density	4, 1.254 Mg/m ³
Absorption coefficient	0.211 mm ⁻¹
F(000)	1064
Crystal size	0.18 x 0.12 x 0.04 mm
Theta range for data collection	3.36 to 25.35 deg.
Limiting indices	-19 ≤ h ≤ 19, -22 ≤ k ≤ 22, -10 ≤ l ≤ 10
Reflections collected / unique	26130 / 4883 [R(int) = 0.1133]
Completeness to theta = 25.35	99.8 %
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	4883 / 1 / 322
Goodness-of-fit on F ²	0.664
Final R indices [I > 2σ(I)]	R ₁ = 0.0393, wR ₂ = 0.0432
R indices (all data)	R ₁ = 0.1210, wR ₂ = 0.0497
Largest diff. peak and hole	0.220 and -0.199 e.Å ⁻³

Table 2. Atomic coordinates (× 10⁴) and equivalent isotropic displacement parameters (Å² × 10³) for **5-OMe**.

U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
C(1)	-5795(2)	2206(2)	6983(3)	61(1)
C(2)	-4401(2)	1753(2)	7673(3)	36(1)
C(3)	-3796(2)	1262(1)	7415(2)	40(1)
C(4)	-2982(2)	1327(1)	8103(3)	39(1)
C(5)	-2729(2)	1877(2)	9109(3)	38(1)
C(6)	-3345(2)	2348(2)	9393(3)	50(1)
C(7)	-4173(2)	2301(2)	8682(3)	49(1)
C(8)	-1833(2)	1945(2)	9847(3)	46(1)
C(9)	-1398(2)	2594(2)	9363(3)	57(1)
C(10)	95(2)	2357(1)	9307(3)	37(1)
C(11)	966(2)	2315(1)	10244(3)	36(1)
C(12)	1101(2)	1550(1)	10717(2)	29(1)
C(13)	644(1)	600(1)	11800(2)	28(1)
C(14)	259(2)	-581(1)	12579(2)	39(1)
C(15)	-549(2)	507(1)	13111(2)	40(1)
C(16)	1678(2)	1087(1)	10309(2)	29(1)
C(17)	3458(2)	1050(1)	10766(3)	31(1)
C(18)	3421(2)	1061(1)	12314(3)	42(1)
C(19)	4132(2)	908(2)	13355(3)	54(1)
C(20)	4885(2)	753(1)	12860(3)	58(1)
C(21)	4937(2)	754(1)	11345(3)	51(1)

C(22)	4234(2)	901(1)	10318(3)	39(1)
C(23)	2569(2)	586(2)	7990(3)	32(1)
C(24)	2806(1)	-113(2)	8371(3)	34(1)
C(25)	2808(1)	-640(2)	7283(3)	44(1)
C(26)	2585(2)	-464(2)	5787(4)	57(1)
C(27)	2342(2)	219(2)	5375(3)	56(1)
C(28)	2336(2)	753(2)	6470(3)	47(1)
N(1)	-513(1)	2640(1)	9994(2)	35(1)
N(2)	518(1)	1280(1)	11554(2)	30(1)
N(3)	194(1)	190(1)	12622(2)	32(1)
O(1)	-5191(1)	1648(1)	6911(2)	50(1)
O(2)	-34(1)	2114(1)	8026(2)	62(1)
P(1)	2562(1)	1310(1)	9364(1)	35(1)
S(1)	1485(1)	239(1)	10992(1)	31(1)

Table 3. Bond lengths [Å] and angles [deg] for **5-OMe**.

C(1)-O(1)	1.434(3)
C(1)-H(1A)	0.9800
C(1)-H(1B)	0.9800
C(1)-H(1C)	0.9800
C(2)-O(1)	1.364(3)
C(2)-C(3)	1.381(3)
C(2)-C(7)	1.382(3)
C(3)-C(4)	1.367(3)
C(3)-H(3)	0.9500
C(4)-C(5)	1.390(3)
C(4)-H(4)	0.9500
C(5)-C(6)	1.376(3)
C(5)-C(8)	1.498(3)
C(6)-C(7)	1.391(3)
C(6)-H(6)	0.9500
C(7)-H(7)	0.9500
C(8)-C(9)	1.499(3)
C(8)-H(8A)	0.9900
C(8)-H(8B)	0.9900
C(9)-N(1)	1.452(3)
C(9)-H(9A)	0.9900
C(9)-H(9B)	0.9900
C(10)-O(2)	1.224(2)
C(10)-N(1)	1.340(3)
C(10)-C(11)	1.524(3)
C(11)-C(12)	1.502(3)
C(11)-H(11A)	0.9900
C(11)-H(11B)	0.9900
C(12)-C(16)	1.358(3)
C(12)-N(2)	1.379(3)
C(13)-N(2)	1.304(3)
C(13)-N(3)	1.346(3)
C(13)-S(1)	1.759(2)
C(14)-N(3)	1.450(2)
C(14)-H(14A)	0.9800
C(14)-H(14B)	0.9800
C(14)-H(14C)	0.9800
C(15)-N(3)	1.457(3)
C(15)-H(15A)	0.9800
C(15)-H(15B)	0.9800
C(15)-H(15C)	0.9800
C(16)-S(1)	1.748(2)
C(16)-P(1)	1.806(2)
C(17)-C(22)	1.392(3)
C(17)-C(18)	1.398(3)

C(17)-P(1)	1.833(2)
C(18)-C(19)	1.394(3)
C(18)-H(18)	0.9500
C(19)-C(20)	1.378(3)
C(19)-H(19)	0.9500
C(20)-C(21)	1.373(3)
C(20)-H(20)	0.9500
C(21)-C(22)	1.375(3)
C(21)-H(21)	0.9500
C(22)-H(22)	0.9500
C(23)-C(24)	1.391(3)
C(23)-C(28)	1.396(3)
C(23)-P(1)	1.834(3)
C(24)-C(25)	1.390(3)
C(24)-H(24)	0.9500
C(25)-C(26)	1.377(3)
C(25)-H(25)	0.9500
C(26)-C(27)	1.372(3)
C(26)-H(26)	0.9500
C(27)-C(28)	1.403(3)
C(27)-H(27)	0.9500
C(28)-H(28)	0.9500
N(1)-H(101)	0.863(9)

O(1)-C(1)-H(1A)	109.5
O(1)-C(1)-H(1B)	109.5
H(1A)-C(1)-H(1B)	109.5
O(1)-C(1)-H(1C)	109.5
H(1A)-C(1)-H(1C)	109.5
H(1B)-C(1)-H(1C)	109.5
O(1)-C(2)-C(3)	116.5(2)
O(1)-C(2)-C(7)	124.9(3)
C(3)-C(2)-C(7)	118.7(2)
C(4)-C(3)-C(2)	121.0(2)
C(4)-C(3)-H(3)	119.5
C(2)-C(3)-H(3)	119.5
C(3)-C(4)-C(5)	121.8(2)
C(3)-C(4)-H(4)	119.1
C(5)-C(4)-H(4)	119.1
C(6)-C(5)-C(4)	116.5(2)
C(6)-C(5)-C(8)	122.3(3)
C(4)-C(5)-C(8)	121.2(3)
C(5)-C(6)-C(7)	122.6(2)
C(5)-C(6)-H(6)	118.7
C(7)-C(6)-H(6)	118.7
C(2)-C(7)-C(6)	119.3(2)
C(2)-C(7)-H(7)	120.3
C(6)-C(7)-H(7)	120.3
C(5)-C(8)-C(9)	113.5(2)
C(5)-C(8)-H(8A)	108.9
C(9)-C(8)-H(8A)	108.9
C(5)-C(8)-H(8B)	108.9
C(9)-C(8)-H(8B)	108.9
H(8A)-C(8)-H(8B)	107.7
N(1)-C(9)-C(8)	114.0(2)
N(1)-C(9)-H(9A)	108.7
C(8)-C(9)-H(9A)	108.7
N(1)-C(9)-H(9B)	108.7
C(8)-C(9)-H(9B)	108.7
H(9A)-C(9)-H(9B)	107.6
O(2)-C(10)-N(1)	123.0(3)
O(2)-C(10)-C(11)	120.9(3)

N(1)-C(10)-C(11)	116.0(2)
C(12)-C(11)-C(10)	106.7(2)
C(12)-C(11)-H(11A)	110.4
C(10)-C(11)-H(11A)	110.4
C(12)-C(11)-H(11B)	110.4
C(10)-C(11)-H(11B)	110.4
H(11A)-C(11)-H(11B)	108.6
C(16)-C(12)-N(2)	117.3(2)
C(16)-C(12)-C(11)	127.5(2)
N(2)-C(12)-C(11)	114.9(2)
N(2)-C(13)-N(3)	124.5(2)
N(2)-C(13)-S(1)	114.60(18)
N(3)-C(13)-S(1)	120.86(19)
N(3)-C(14)-H(14A)	109.5
N(3)-C(14)-H(14B)	109.5
H(14A)-C(14)-H(14B)	109.5
N(3)-C(14)-H(14C)	109.5
H(14A)-C(14)-H(14C)	109.5
H(14B)-C(14)-H(14C)	109.5
N(3)-C(15)-H(15A)	109.5
N(3)-C(15)-H(15B)	109.5
H(15A)-C(15)-H(15B)	109.5
N(3)-C(15)-H(15C)	109.5
H(15A)-C(15)-H(15C)	109.5
H(15B)-C(15)-H(15C)	109.5
C(12)-C(16)-S(1)	108.81(17)
C(12)-C(16)-P(1)	126.45(18)
S(1)-C(16)-P(1)	124.54(14)
C(22)-C(17)-C(18)	117.4(2)
C(22)-C(17)-P(1)	120.09(19)
C(18)-C(17)-P(1)	122.16(19)
C(19)-C(18)-C(17)	120.7(2)
C(19)-C(18)-H(18)	119.6
C(17)-C(18)-H(18)	119.6
C(20)-C(19)-C(18)	119.9(3)
C(20)-C(19)-H(19)	120.1
C(18)-C(19)-H(19)	120.1
C(21)-C(20)-C(19)	120.2(3)
C(21)-C(20)-H(20)	119.9
C(19)-C(20)-H(20)	119.9
C(20)-C(21)-C(22)	120.0(3)
C(20)-C(21)-H(21)	120.0
C(22)-C(21)-H(21)	120.0
C(21)-C(22)-C(17)	121.8(2)
C(21)-C(22)-H(22)	119.1
C(17)-C(22)-H(22)	119.1
C(24)-C(23)-C(28)	118.4(2)
C(24)-C(23)-P(1)	124.04(19)
C(28)-C(23)-P(1)	117.6(2)
C(25)-C(24)-C(23)	121.7(2)
C(25)-C(24)-H(24)	119.1
C(23)-C(24)-H(24)	119.1
C(26)-C(25)-C(24)	119.1(3)
C(26)-C(25)-H(25)	120.5
C(24)-C(25)-H(25)	120.5
C(27)-C(26)-C(25)	120.6(3)
C(27)-C(26)-H(26)	119.7
C(25)-C(26)-H(26)	119.7
C(26)-C(27)-C(28)	120.5(3)
C(26)-C(27)-H(27)	119.7
C(28)-C(27)-H(27)	119.7
C(23)-C(28)-C(27)	119.7(3)

C(23)-C(28)-H(28)	120.2
C(27)-C(28)-H(28)	120.2
C(10)-N(1)-C(9)	122.3(2)
C(10)-N(1)-H(101)	117.2(15)
C(9)-N(1)-H(101)	117.3(15)
C(13)-N(2)-C(12)	110.4(2)
C(13)-N(3)-C(14)	120.5(2)
C(13)-N(3)-C(15)	117.5(2)
C(14)-N(3)-C(15)	118.6(2)
C(2)-O(1)-C(1)	117.2(2)
C(16)-P(1)-C(17)	102.08(11)
C(16)-P(1)-C(23)	103.14(11)
C(17)-P(1)-C(23)	100.44(12)
C(16)-S(1)-C(13)	88.90(12)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **5-OMe**.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2 [h^2 a^{*2} U_{11} + \dots + 2hk a^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
C(1)	35(2)	59(2)	88(2)	12(2)	1(2)	3(2)
C(2)	35(2)	40(2)	32(2)	6(2)	2(2)	1(2)
C(3)	50(2)	30(2)	37(2)	-4(1)	-1(2)	7(2)
C(4)	42(2)	39(2)	37(2)	1(2)	4(1)	16(2)
C(5)	40(2)	46(2)	26(2)	5(1)	2(1)	5(2)
C(6)	49(2)	55(2)	45(2)	-22(2)	5(2)	-2(2)
C(7)	39(2)	52(2)	57(2)	-15(2)	14(2)	9(2)
C(8)	44(2)	53(2)	36(2)	6(1)	-5(1)	4(2)
C(9)	45(2)	56(2)	60(2)	8(2)	-23(2)	-1(2)
C(10)	68(2)	13(2)	30(2)	7(1)	10(2)	0(2)
C(11)	42(2)	27(2)	44(2)	-2(1)	23(2)	-6(2)
C(12)	26(2)	25(2)	36(2)	2(1)	7(1)	-6(1)
C(13)	26(2)	26(2)	32(2)	4(1)	6(1)	-3(1)
C(14)	49(2)	30(2)	39(2)	6(1)	10(1)	-4(2)
C(15)	43(2)	37(2)	45(2)	4(1)	20(1)	2(2)
C(16)	29(2)	25(2)	36(2)	-1(1)	12(1)	-4(1)
C(17)	27(2)	30(2)	37(2)	-7(1)	6(1)	-7(1)
C(18)	32(2)	54(2)	42(2)	-8(1)	9(2)	-9(2)
C(19)	55(2)	69(2)	37(2)	-8(2)	3(2)	-19(2)
C(20)	42(2)	62(2)	62(2)	-7(2)	-17(2)	-4(2)
C(21)	27(2)	64(2)	59(2)	-19(2)	0(2)	-2(2)
C(22)	34(2)	44(2)	39(2)	-12(1)	5(2)	-4(2)
C(23)	26(2)	44(2)	27(2)	2(1)	9(1)	-6(2)
C(24)	24(2)	48(2)	31(2)	-6(2)	6(1)	1(2)
C(25)	32(2)	52(2)	49(2)	-13(2)	8(2)	-1(2)
C(26)	43(2)	82(3)	51(2)	-29(2)	23(2)	-23(2)
C(27)	49(2)	94(3)	25(2)	0(2)	5(1)	-32(2)
C(28)	36(2)	62(2)	41(2)	9(2)	5(2)	-13(2)
N(1)	41(2)	36(2)	26(1)	-2(1)	-7(1)	5(1)
N(2)	29(1)	23(1)	41(1)	6(1)	14(1)	0(1)
N(3)	32(1)	25(1)	43(1)	3(1)	15(1)	0(1)
O(1)	35(1)	52(1)	59(1)	1(1)	-1(1)	3(1)
O(2)	134(2)	31(1)	20(1)	-6(1)	11(1)	1(1)
P(1)	31(1)	35(1)	42(1)	2(1)	14(1)	-2(1)
S(1)	28(1)	27(1)	40(1)	3(1)	10(1)	5(1)

4. ICP-MS measurements

4.1 - Preparations of the samples:

A mixture of 1.5 mL HNO₃ (67-69%) and 0.5 mL of HCL (36%-37%) was added to the sample and the mixture was stirred at 90°C for 24 h then diluted.

4.2 - Conditions of ICP-MS:

Instrument:	Serie X 2 from Thermo Electron.
Nebuliser type:	Meinhard nebuliser.
Plasma power:	1400 W
Coolant gaz flow:	13 l/min
Auxiliary gaz flow:	0.7 l/min
Nebuliser flow:	0.88 l/min

Data acquisition:	
Detector:	ETP simulscan
Scan mode:	Peak hopping
Points per peak:	1
Dwell time per peak:	10 ms
Sweeps:	50
Isotopes used for metal determination:	105Pd, 106Pd, 108Pd
Isotopes used for internal standardisation:	115In, 185Re.

Please find above the corresponding reports.

Report R1204-271-V2

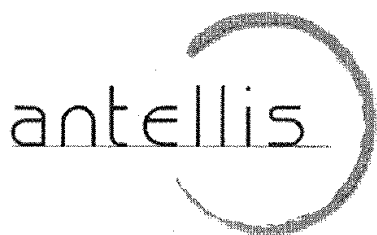
- Sample 893av corresponds to the crude product **7b** when using dendrimer **5-G₁** as the ligand (< 0.55 ppm Pd).
- Sample 915av corresponds to the crude product **7b** when using monomer **5-OMe** as the ligand (~ 1400 ppm Pd).
- Sample 915ap corresponds to the product **7b** after two purifications by column chromatography when using **5-OMe** as the ligand. Noteworthy, even after these two purifications, the coupling product does not meet the requirements of pharmaceutical industry in terms of Pd contaminants (~ 16 ppm Pd on **7b**).

Report R1204-271-V3

- Sample 915B corresponds to the crude product **7b** when using PPh₃ as the ligand (~ 2200 ppm Pd).

Report R1205-293-V2

- Sample 952 corresponds to the crude product **7b** when using dendrimer **6-G₁** as the ligand (~ 173 ppm Pd).



Toulouse, le 22 Mai 2012

Rapport d'essai

Mme OUALI

LCC
205, route de Narbonne
31077 Toulouse Cedex 04

Rapport n° : R1204-271-V2

Madame,

Nous avons le plaisir de vous communiquer les résultats des analyses réalisées dans le cadre de la prestation décrite ci-dessous.

Afin d'améliorer continuellement nos services, nous vous invitons à nous communiquer toute remarque relative à la prestation réalisée : satisfactions, insatisfactions, services supplémentaires souhaités. Veuillez donc adresser vos commentaires par fax, courriel ou directement par téléphone. L'équipe d'Antellis est à votre écoute de 9h à 18h, du lundi au vendredi.

1. Description de la prestation :

nom du projet : Polymères

devis / bon de commande : /

date de réception des échantillons : 07/03/2012

<i>Méthode(s) d'analyse</i>	<i>préparation</i>	<i>traitement</i>	<i>technique d'analyse</i>
A	aucune	Digestion HNO ₃	ICP-MS basé sur ISO 17294
B	aucune	Digestion HNO ₃ , HCl	ICP-MS basé sur ISO 17294

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Rapport n°R1204-271-V2

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2. Résultats d'analyse :

id. échantillon : 893av; 267 mg

id. Antellis : 12C016-6

Elément	Méthode d'analyse	conc. (mg.kg ⁻¹)	incertitude (2s)
Pd	B	< 0,55	-

id. échantillon : 915ap; 100 mg

id. Antellis : 12C016-8

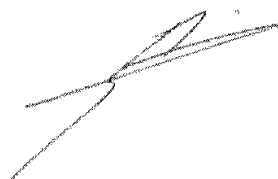
Elément	Méthode d'analyse	conc. (mg.kg ⁻¹)	incertitude (2s)
Pd	B	16,09	1,02

id. échantillon : 915av; 80 mg

id. Antellis : 12C016-10

Elément	Méthode d'analyse	conc. (mg.kg ⁻¹)	incertitude (2s)
Pd	B	1 432,71	46,08

Sébastien Aries



Dr. Sébastien Aries
Responsable Scientifique

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Toulouse, le 16 Juillet 2012

Rapport d'essai

Mme OUALI

LCC
205, route de Narbonne
31077 Toulouse Cedex 04

Rapport n° : R1204-271-v3

Madame,

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Afin d'améliorer continuellement nos services, nous vous invitons à nous communiquer toute remarque relative à la prestation réalisée : satisfactions, insatisfactions, services supplémentaires souhaités. Veuillez donc adresser vos commentaires par fax, courriel ou directement par téléphone. L'équipe d'Antellis est à votre écoute de 9h à 18h, du lundi au vendredi.

1. Description de la prestation :

nom du projet : Polymères
devis / bon de commande : /
date de réception des échantillons : 07/03/2012

<i>Méthode(s) d'analyse</i>	<i>préparation</i>	<i>traitement</i>	<i>technique d'analyse</i>
A	aucune	Digestion HNO ₃	ICP-MS basé sur ISO 17294
B	aucune	Digestion HNO ₃ , HCl	ICP-MS basé sur ISO 17294

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Rapport n°R1204-271-v3

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2. Résultats d'analyse :

id. échantillon : 915B; 100 mg
id. Antellis : 12C016-9

Elément	Méthode d'analyse	conc. (mg.kg ⁻¹)	incertitude (2s)
Pd	B	2 227,82	62,73

Sébastien Aries



Dr. Sébastien Aries
Responsable Scientifique

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2/2



Toulouse, le 12 Juillet 2012

Rapport d'essai

Mme OUALI

LCC
205, route de Narbonne
31077 Toulouse Cedex 04

Rapport n° : R1205-293-v2

Madame,

Nous avons le plaisir de vous communiquer les résultats des analyses réalisées dans le cadre de la prestation décrite ci-dessous.

Afin d'améliorer continuellement nos services, nous vous invitons à nous communiquer toute remarque relative à la prestation réalisée : satisfactions, insatisfactions, services supplémentaires souhaités. Veuillez donc adresser vos commentaires par fax, courriel ou directement par téléphone. L'équipe d'Antellis est à votre écoute de 9h à 18h, du lundi au vendredi.

1. Description de la prestation :

nom du projet : Polymères
devis / bon de commande : /
date de réception des échantillons : 24/04/2012

<i>Méthode(s) d'analyse</i>	<i>préparation</i>	<i>traitement</i>	<i>technique d'analyse</i>
A	aucune	Digestion HNO ₃ , HCl	ICP-MS basé sur ISO 17294

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Rapport n°R1205-293-v2

1/2

2. Résultats d'analyse :

id. échantillon : MK952, 300 mg

id. Antellis : 12D039-4

Elément	Méthode d'analyse	conc. (mg.kg ⁻¹)	incertitude (2s)
Pd	A	173,21	3,09

Sébastien Aries



Dr. Sébastien Aries
Responsable Scientifique

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Rapport n°**R1205-293-v2**

2/2

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