

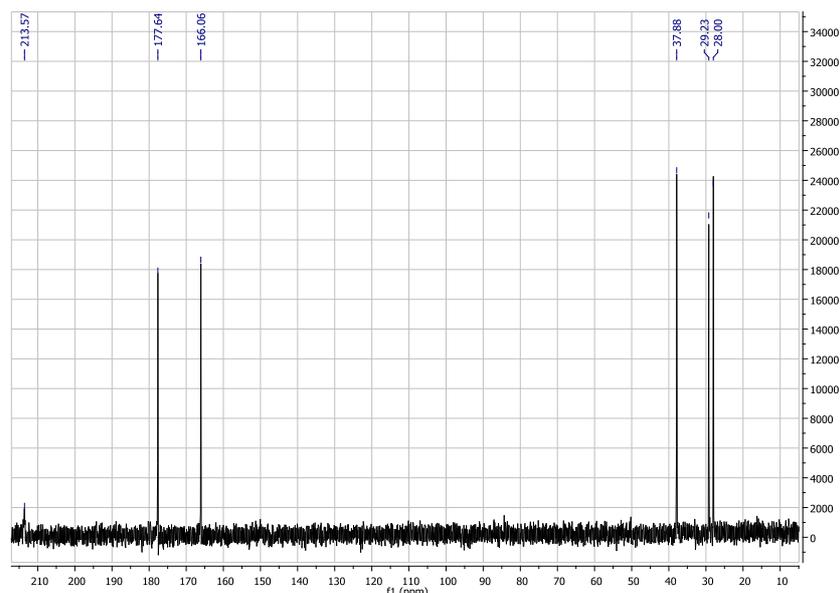
# Microwave-assisted conversion of carbohydrates to levulinic acid: an essential step in biomass conversion

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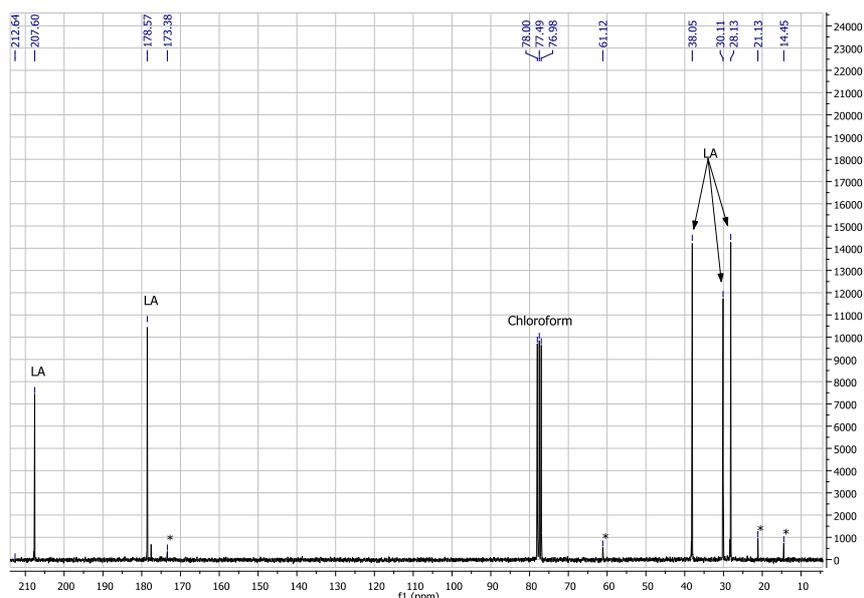
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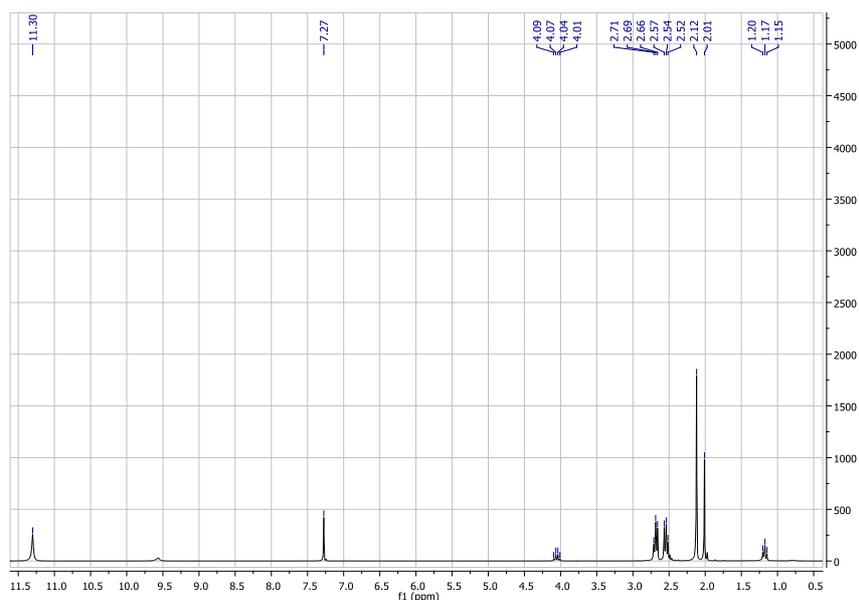
## Electronic Supplementary Information (ESI)



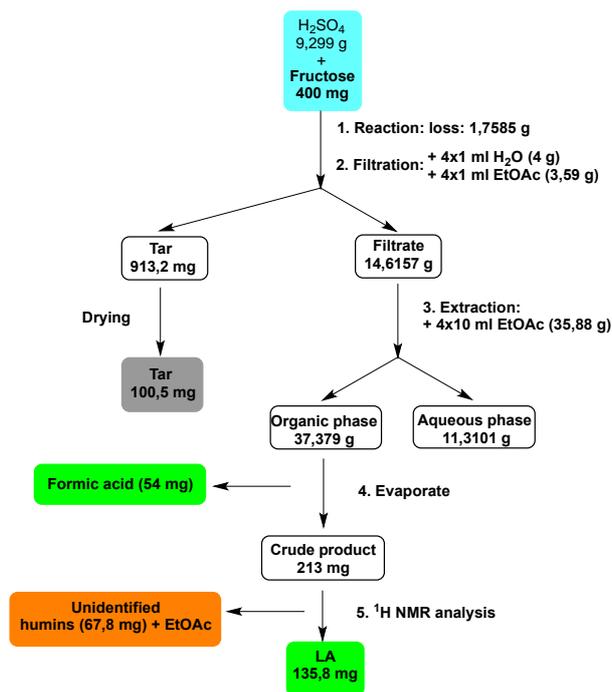
ESI-FIG 1. <sup>13</sup>C-NMR of the neat reaction mixture of sulfuric acid dehydrated fructose (400 mg in 8 mL 2M H<sub>2</sub>SO<sub>4</sub>).



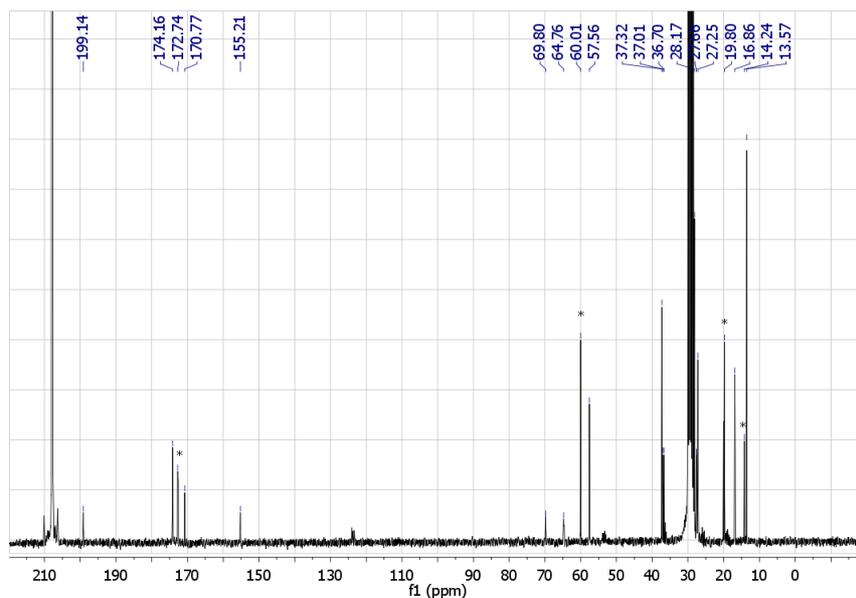
ESI-FIG 2. <sup>13</sup>C-NMR of the product of the reaction mixture of sulfuric acid dehydrated fructose after workup procedure. \*: Ethyl-acetate residue from extraction



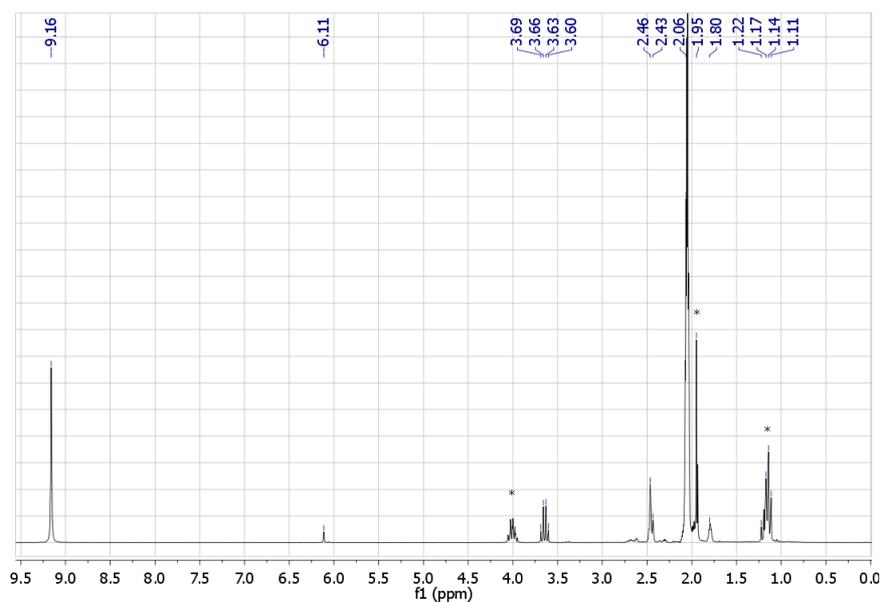
**ESI-FIG 3.** Typical  $^1\text{H}$ -NMR spectrum of the product mixture of sulfuric acid dehydrated fructose after addition of benzene (10  $\mu\text{L}$ ) as internal standard.



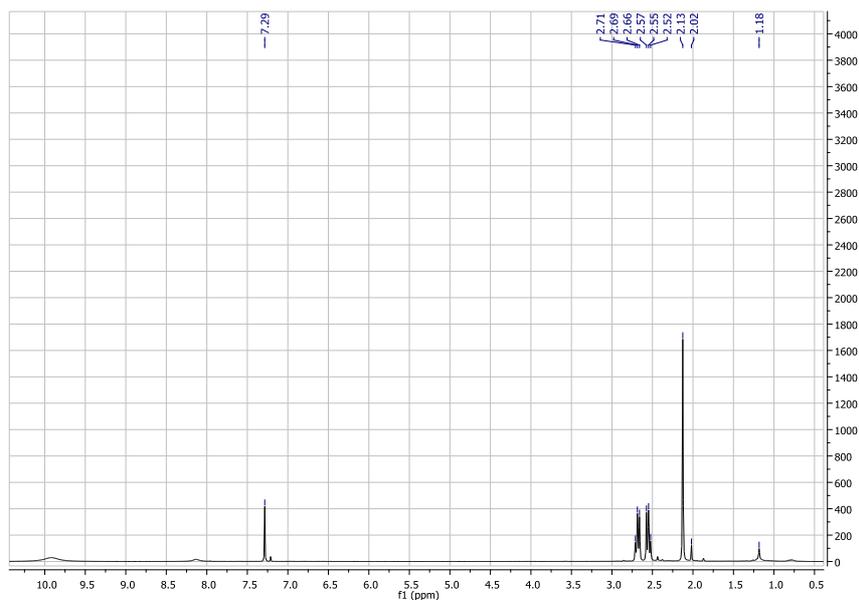
**ESI-FIG 4.** Mass balance of the process of the sulfuric acid catalyzed dehydration of 400 mg fructose.



**ESI-FIG 5.**  $^{13}\text{C}$ -NMR spectrum of the humins obtained from ethyl-acetate extraction. The dark brown residue (67.8 mg) was dissolved in acetone- $d_6$  after EtOAc removal by vacuum.



**ESI-FIG 6.**  $^1\text{H}$ -NMR spectrum of the humins obtained from ethyl-acetate extraction. The dark brown residue (67.8 mg) was dissolved in acetone- $d_6$  after EtOAc removal by vacuum.



**ESI-FIG 7.** Typical  $^1\text{H-NMR}$  spectrum of the product mixture of sulfuric acid dehydrated glucosamine after addition of benzene (10  $\mu\text{L}$ ) as internal standard.

### Method of calculation for LA

Benzene was used as internal standard for quantitative analysis of levulinic acid (LA) formed.

Int. stand.	M (g/mol)	m (g)	d (g/ml)	V (ml)	n (mol)
benzene	78	0,008786	0,8786	0,01	0,000113

The formula used to calculate amount of LA:

$$n_{LA} = \frac{2 \cdot n_B \cdot I_{LA}}{I_B} = 2 \cdot n_B \cdot I_{LA} = 0.000226 \cdot I_{LA}$$

$n_B$  = mol of benzene added as internal standard

$n_{LA}$  = mol of levulinic acid formed

$I_B$  =  $^1\text{H-NMR}$  integral of benzene (6H). Value was set to 1.

$I_{LA}$  =  $^1\text{H-NMR}$  integral of peak of levulinic acid at 2.1 ppm ( $-\text{CH}_3$ ; 3H)

**The table of yields and the corresponding integrals:**

substrate	$c_{\text{substrate}}/\text{mol/dm}^3$	catalyst	t/min	Yield/%	m/m%	Integral(I <sub>LA</sub> )
cellulose	0,6175	H <sub>2</sub> SO <sub>4</sub>	30	21,4	15,4	2,35
cellulose	0,6175	HCl	50	46,0	31,0	4,74
cellulose	0,6175	H <sub>2</sub> SO <sub>4</sub>	50	34,2	23,0	3,53
glucose	0,5550	HCl	30	48,6	31,4	4,80
glucose	0,5550	H <sub>2</sub> SO <sub>4</sub>	30	40,5	26,1	4,00
fructose	0,5550	HCl	30	49,4	31,8	4,87
fructose	0,5550	H <sub>2</sub> SO <sub>4</sub>	30	42,7	27,5	4,21
cellobiose	0,2925	HCl	30	44,0	29,9	4,57
cellobiose	0,2925	H <sub>2</sub> SO <sub>4</sub>	30	41,3	28,0	4,29
chitin	0,4925	HCl	30	32,7	18,7	2,86
chitin	0,3700	H <sub>2</sub> SO <sub>4</sub>	30	37,8	21,6	2,48
l-chitosane	0,4125	HCl	20	31,9	22,7	2,60
l-chitosane	0,4175	H <sub>2</sub> SO <sub>4</sub>	20	19,3	13,7	1,57
M-chitosane	0,4175	HCl	20	37,0	26,3	3,02
M-chitosane	0,4125	H <sub>2</sub> SO <sub>4</sub>	20	32,1	22,8	2,62
glucosamine	0,4175	HCl	10	36,4	19,6	2,25
glucosamine	0,4175	H <sub>2</sub> SO <sub>4</sub>	10	25,6	13,8	1,58
N-acetyl-glucosamine	0,3400	HCl	10	22,4	11,8	1,35
N-acetyl-glucosamine	0,3400	H <sub>2</sub> SO <sub>4</sub>	10	20,6	10,8	1,24