

Supplementary Information

**Selective aqueous phase oxidation of 5-hydroxymethylfurfural to  
2,5-furandicarboxylic acid over Pt/C catalysts:  
Influence of the base and effect of bismuth promotion.**

**Hicham AIT RASS, Nadine ESSAYEM, Michèle BESSON\***

*IRCELYON, Institut de recherches sur la catalyse et l'environnement de Lyon, UMR5256  
CNRS-Université Lyon 1, 2 Avenue Albert Einstein, 69626 Villeurbanne Cedex, France*

\*Corresponding author:

[michele.besson@ircelyon.univ-lyon1.fr](mailto:michele.besson@ircelyon.univ-lyon1.fr), phone +33(0)472445358, fax +33(0)472445399

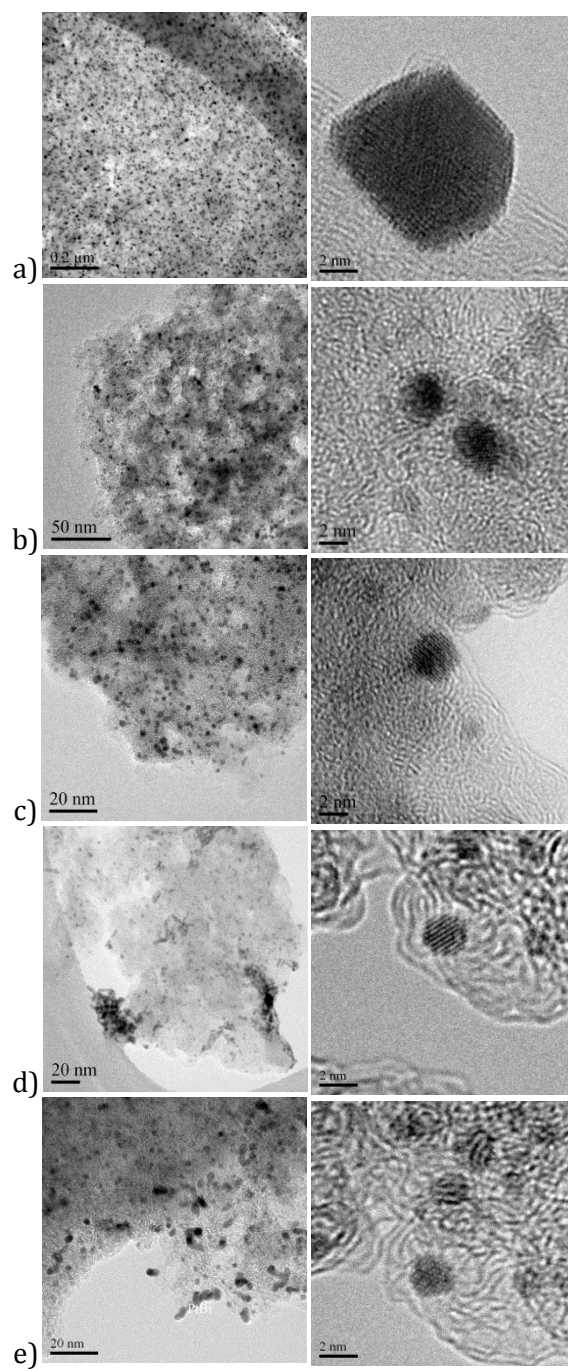


Fig. S1 TEM images of (a) 3.3%Pt<sub>ex</sub>/C<sub>L3S</sub>, (b) 3.6%Pt<sub>imp</sub>/C<sub>L3S</sub>, (c) 3.6%Pt<sub>imp</sub>0.9%Bi/C<sub>L3S</sub>, (d), 5.1%Pt<sub>imp</sub>/C<sub>3SW</sub> and (e) 5.1%Pt<sub>imp</sub>1%Bi/C<sub>3SW</sub>.

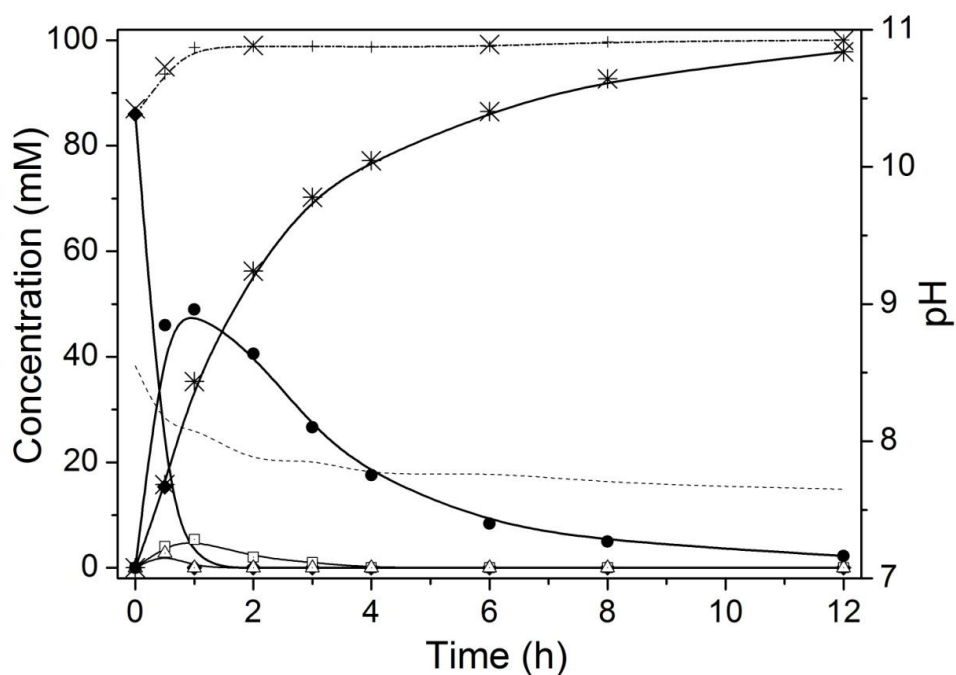


Fig. S2 Evolution of concentrations of HMF and products DFF, HMFCFA, FFCA, and FDCA as a function of time over (a) 3.6%Pt<sub>imp</sub>/C<sub>L3S</sub> after further reduction at 300°C under H<sub>2</sub> in the presence of NaHCO<sub>3</sub> (NaHCO<sub>3</sub>/HMF = 4).

◆ HMF, △ DFF, □ HMFCFA, ● FFCA, \* FDCA, + molar balance, × TOC and --- pH

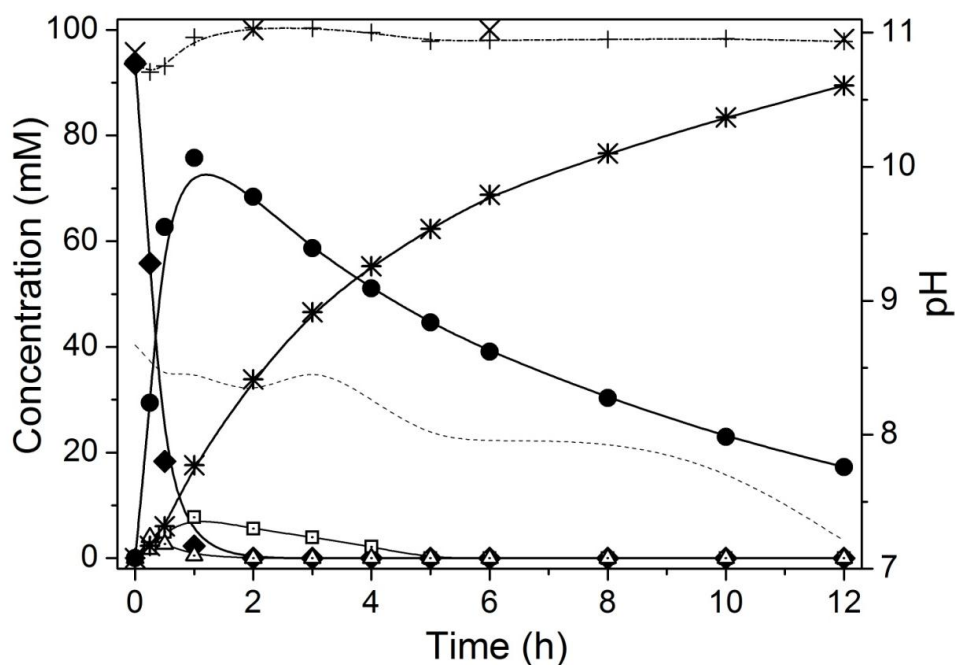


Fig. S3 Evolution of concentrations of HMF and products DFF, HMFCFA, FFCA, and FDCA as a function of time over 5.1%Pt<sub>imp</sub>/C<sub>3SW</sub>, in the presence of NaHCO<sub>3</sub> (NaHCO<sub>3</sub>/HMF = 2)

◆ HMF, △ DFF, □ HMFCFA, ● FFCA, \* FDCA, + molar balance, × TOC and --- pH

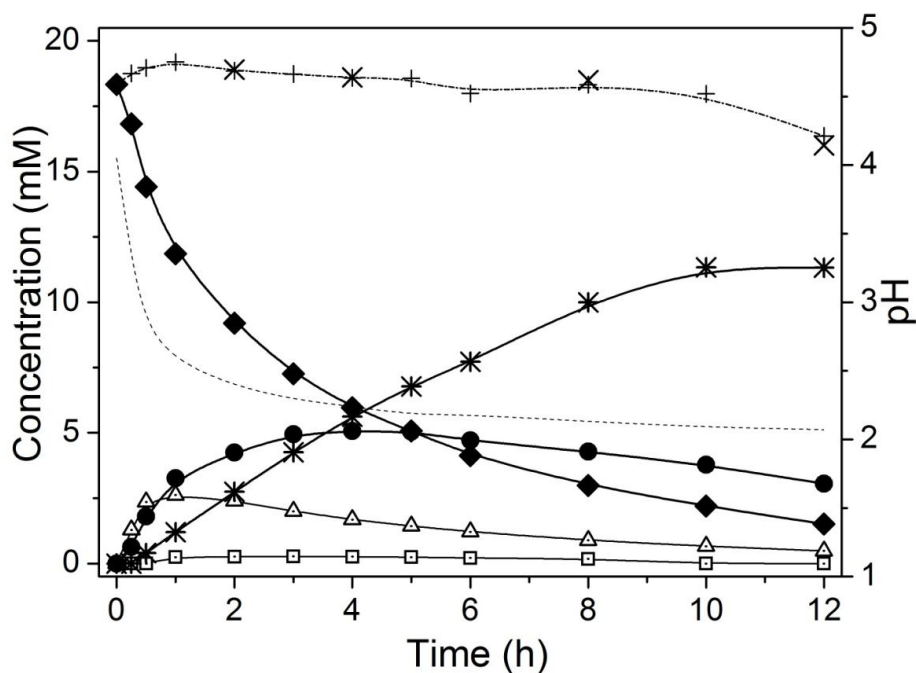


Fig. S4 Evolution of concentrations of HMF and products DFF, HMFCA, FFCA, and FDCA as a function of time over 5.1%Pt<sub>imp</sub>/C<sub>3SW</sub> under base free conditions ( $C_0(\text{HMF}) = 0.02 \text{ mol.L}^{-1}$ , HMF/Pt=100);  $\blacklozenge$  HMF,  $\triangle$  DFF,  $\square$  HMFCA,  $\bullet$  FFCA,  $*$  FDCA,  $+$  molar balance,  $\times$  TOC and --- pH

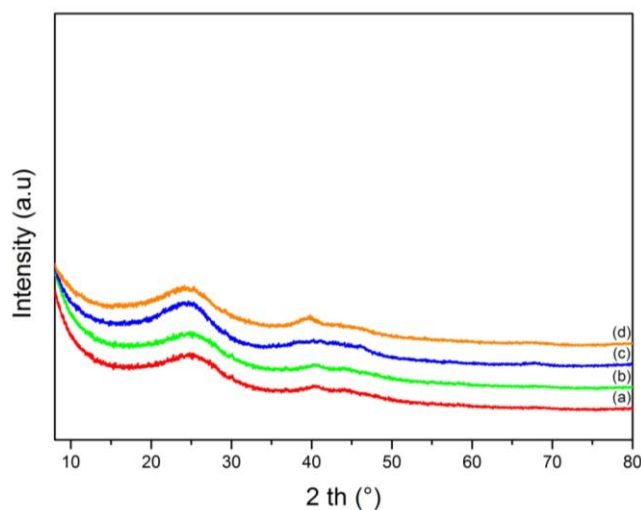


Fig. S5 XRD patterns of Pt and PtBi catalysts prepared by liquid impregnation method. (a) 3.6%Pt<sub>imp</sub>/C<sub>L3S</sub>, (b) 3.6%Pt<sub>imp</sub>0.9%Bi/C<sub>L3S</sub> reduced under liquid phase conditions, (c) 3.6%Pt<sub>imp</sub>/C<sub>L3S</sub>, (d) 3.6%Pt<sub>imp</sub>0.9%Bi/C<sub>L3S</sub> after further pre-treatment under hydrogen at 300°C