

**Solar Photochemistry: Optimisation of the Photo Friedel-Crafts Acylation
of Naphthoquinones**

Lorna J. Mitchell, William Lewis and Christopher J. Moody*

Supplementary Information

Table of Contents

Experimental Details for the Preparation of Starting Materials	S2
X-Ray Crystallographic Data	S11
Copies of ^1H and ^{13}C NMR Spectra	S12
Selected HMBC Spectra for Regiochemical Assignments	S90
Wavelength Emission Spectra	S94
References	S97

* School of Chemistry, University of Nottingham, University Park, Nottingham NG7 2RD,
U.K; Fax: (+44) 115 951 3564; E-mail: c.j.moody@nottingham.ac.uk

Experimental Details for the Preparation of Starting Materials

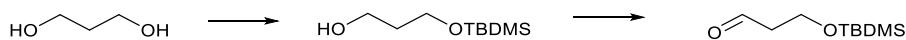
Preparation of Functionalised Aldehydes

4-Chlorobutanal



The reaction was carried out according to the procedure in Snyder *et al.*¹ with minor modifications. To 4-chloro-1-butanol (2.80 mL, 28.2 mmol) in dichloromethane (100 mL) was added PCC (7.30 g, 33.9 mmol) and silica gel (7.30 g) and the reaction mixture was stirred at ambient temperature for 16 h. The resulting black slurry was filtered through a short column of florasil, washed with ether (3 × 100 mL) and concentrated *in vacuo*. The *title compound* (2.17 g, 73%) was isolated as a colourless oil which was used in the next step without further purification; δ_{H} (400 MHz; CDCl₃) 9.79 (1 H, s, CH), 3.57 (2 H, t, *J* 6.50, CH₂Cl), 2.65 (2 H, td, *J* 7.0, 0.8, CH₂CO), 2.08 (2 H, pent, *J* 6.5, CH₂); δ_{C} (100 MHz; CDCl₃) 200.9 (CHO), 44.0 (CH₂), 40.8 (CH₂), 24.8 (CH₂). Data are consistent with those reported in the literature.²

3-(*t*-Butyldimethylsiloxy)propanal



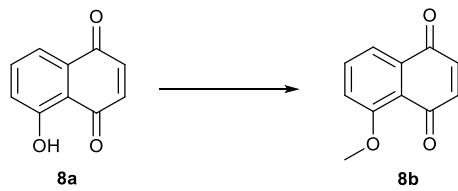
The reaction was carried out according to the procedure in McDougal *et al.*³ with minor modifications. Sodium hydride (2.63 g, 65.7 mmol) was washed with pentane (3 × 10 mL) and dried under argon. THF (125 mL) was added followed by subsequent slow addition of the propane-1,3-diol (5.00 g, 65.7 mmol) and the reaction mixture was stirred at ambient temperature. After 45 min, TBDMS-Cl (9.90 g, 65.7 mmol) was added and the reaction mixture was stirred for a further 45 min. The reaction was diluted with ether (200 mL) and washed with sodium carbonate solution (200 mL), brine (200 mL),

dried (MgSO_4), filtered and concentrated *in vacuo*. The residue was purified by flash chromatography on silica gel (light petroleum \rightarrow 70:30 light petroleum/ethyl acetate) to afford product (10.6 g, 85%) as a colourless oil; δ_{H} (400 MHz; CDCl_3) 3.73 (1 H, td, J 6.1, 1.5, CH_2), 3.68 (1 H, td, J 5.8, 1.8, CH_2), 3.14 (1 H, s, OH), 1.69 (1 H, quint., J 5.6, CH_2), 0.83 (9 H, d, J 0.6, 3 \times CH_3), 0.00 (6 H, d, J 1.6, 2 \times CH_3); δ_{C} (100 MHz; CDCl_3) 62.0 (CH_2), 61.2 (CH_2), 34.4 (CH_2), 25.7 (CH_3), 18.0 (C), -5.67 (CH_3). Data are consistent with those reported in the literature.³

The next step was carried out according to the procedure by Grünanger *et al.*⁴ Oxalyl chloride (1.04 mL, 12.2 mmol) in dichloromethane (27 mL) was cooled to -60 °C and DMSO (1.8 mL, 25.2 mmol) in dichloromethane (5 mL) was added and the reaction mixture stirred. After 5 min, 3-(*t*-butyldimethylsiloxy)propan-1-ol (2.00 g, 10.5 mmol) was added dropwise, followed by further dropwise addition of triethylamine (7.38 mL, 53.2 mmol). The reaction mixture was stirred whilst being gradually warmed to ambient temperature. After 1 h, water (50 mL) was added and the phases separated. The aqueous phase was back-extracted with further dichloromethane (50 mL) and the combined organic layers were washed with brine (100 mL), water (50 mL) and further brine (50 mL), dried (MgSO_4), filtered and concentrated *in vacuo*. The residue was dissolved in ether, filtered through Celite and concentrated *in vacuo* to afford the *title compound* (1.69 g, 86%) as a colourless oil; δ_{H} (400 MHz; CDCl_3) 9.79 (1 H, t, J 2.3, CH), 3.98 (2 H, t, J 6.0, CH_2), 2.59 (2 H, td, J 6.0, 2.2, CH_2), 0.87 (9 H, s, 3 \times CH_3), 0.05 (6 H, s, 2 \times CH_3); δ_{C} (100 MHz; CDCl_3) 201.9 (CH), 57.4 (CH_2), 46.5 (CH_2), 25.8 (CH_3), 18.2 (C), -5.5 (CH_3). Data are consistent with those previously reported.⁵

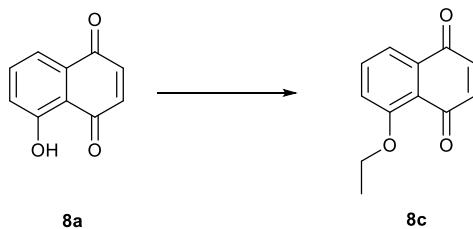
Preparation of Functionalised Naphthoquinones

5-Methoxy-1,4-naphthoquinone 8b



The reaction was carried out according to the procedure in Tietze *et al.*,⁶ with minor modifications. To 5-hydroxy-1,4-naphthoquinone **8a** (100 mg, 0.57 mmol) and silver(I) oxide (100 mg, 0.46 mmol) in dichloromethane (2 mL) was added iodomethane (0.07 mL, 1.12 mmol) and the mixture stirred at ambient temperature. After 20 h further iodomethane (0.03 mL, 0.46 mmol) and silver(I) oxide (100 mg, 0.46 mmol) were added and the reaction mixture stirred for a further 3 h. The reaction mixture was filtered through Celite, washed (dichloromethane) and concentrated *in vacuo*. The residue was purified by flash chromatography on silica gel (light petroleum → 8:2 light petroleum: ethyl acetate) to give the *title compound* **8b** (96 mg, 90%) as fine orange crystals; mp 180–183 °C (lit.,⁷ mp 180–185 °C); δ_H (400 MHz; CDCl₃) 7.72 (2 H, m, ArH), 7.32 (1 H, dd, J 8.2, 1.0, ArH), 6.87 (2 H, app. br s, CH), 4.02 (3 H, s, OMe); δ_C (100 MHz; CDCl₃) 185.1 (C), 184.3 (C), 159.6 (C), 140.8 (CH), 136.1 (CH), 134.9 (CH), 134.0 (C), 119.6 (C), 119.1 (CH), 117.9 (CH), 56.4 (CH₃). Data are consistent with those reported in the literature.^{6,7}

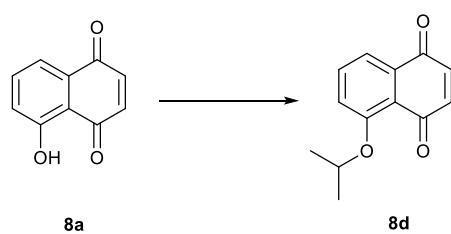
5-Ethoxy-1,4-naphthoquinone **8c**



To 5-hydroxy-1,4-naphthoquinone **8a** (100 mg, 0.57 mmol) and silver(I) oxide (100 mg, 0.46 mmol) in dichloromethane (2 mL) was added iodoethane (0.09 mL,

1.12 mmol) and the reaction mixture was stirred at ambient temperature. After 16 h further silver(I) oxide (100 mg, 0.46 mmol) and iodoethane (0.04 mL, 0.46 mmol) were added and the reaction mixture was stirred for a further 4 h. The reaction mixture was concentrated *in vacuo* and the product was crystallised from the minimum volume of hot toluene/light petroleum to yield the *title compound* **8c** (114 mg, 99%) as a brown solid; mp 78–79 °C; (Found: C, 71.28; H, 4.98; $C_{12}H_{10}O_3$ requires C, 71.32; H, 5.18 %); (Found: M+Na⁺, 225.0521. $C_{12}H_{10}O_3Na^+$ requires 225.0522); λ_{max} (CH₂Cl₂)/nm 274 (log ε 5.71), 400 (4.37); ν_{max} (CHCl₃)/cm⁻¹ 3686, 3012, 2415, 1521, 1239; δ_{H} (400 MHz; CDCl₃) 7.69 (1 H, dd, *J* 7.6, 1.4, ArH), 7.63 (1 H, app. br t, *J* 8.3, ArH), 7.27 (1 H, dd, *J* 8.3, 1.4, ArH), 6.83 (2 H, app. br s, CH), 4.20 (2 H, q, *J* 7.0, CH₂), 1.53 (3 H, t, *J* 7.0, CH₃); δ_{C} (100 MHz; CDCl₃) 185.3 (C), 184.2 (C), 159.0 (C), 140.9 (CH), 136.1 (CH), 134.8 (CH), 134.0 (C), 119.8 (C), 119.1 (CH), 119.0 (CH), 65.1 (CH₂), 14.6 (CH₃); *m/z* (ESI) 225 (M+Na⁺, 100%), 103 (M+H⁺, 5).

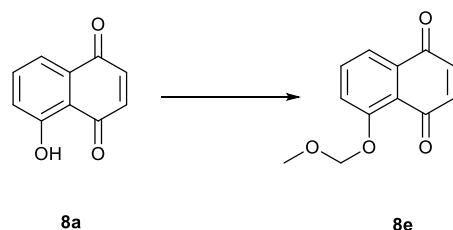
5-Isopropoxy-1,4-Naphthoquinone **8d**



To 5-hydroxy-1,4-naphthoquinone **8a** (200 mg, 1.15 mmol) and silver(I) oxide (220 mg, 0.96 mmol) in dichloromethane (2 mL) was added 2-iodopropane (0.12 mL, 2.29 mmol) and the reaction mixture was stirred at ambient temperature. After 16 h further silver(I) oxide (440 mg, 1.92 mmol) and 2-iodopropane (0.4 mL, 9.20 mmol) were added and the reaction mixture was stirred for a further 4 h. The reaction mixture was concentrated *in vacuo* to yield the *title compound* **8d** (248 mg, 100%) as a dark green solid; mp 89–90 °C; (Found: M+Na⁺, 239.0669. $C_{13}H_{12}O_3Na^+$ requires 239.0679);

λ_{max} (CH_2Cl_2)/nm 275 (log ϵ 4.40), 322 (2.91), 403 (3.46); ν_{max} (CH_2Cl_2)/ cm^{-1} 2983, 1662, 1585, 1464, 1298, 1251; δ_{H} (400 MHz; CDCl_3) 7.70 (1 H, dd, J 7.6, 1.2, ArH), 7.63 (1 H, app. br t, J 8.3, ArH), 7.30 (1 H, app. br d, J 8.3, ArH), 6.85 (2 H, app. br s, CH), 4.69 (1 H, sept, J 6.2, CH), 1.45 (6 H, d, J 6.2, CH_3); δ_{C} (100 MHz; CDCl_3) 185.3 (C), 184.0 (C), 158.3 (C), 141.0 (CH), 136.0 (CH), 134.5 (CH), 134.2 (C), 121.3 (CH), 120.8 (C), 119.1 (CH), 72.3 (CH), 22.0 (CH_3); m/z (ESI) 239 ($\text{M}+\text{Na}^+$, 100%).

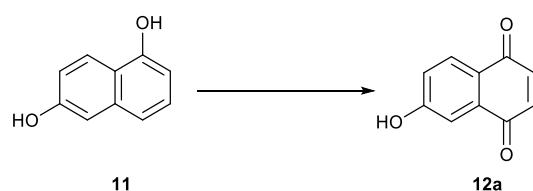
5-Methoxymethoxy-1,4-naphthoquinone **8e**



The reaction was carried out according to the procedure in Nandaluru *et al.*⁸ with minor modifications. To 5-hydroxy-1,4-naphthoquinone **8a** (200 mg, 1.15 mmol) and MOM-Cl (0.23 mL, 2.87 mmol) in dichloromethane (3 mL) at 0 °C was added DIPEA (0.40 mL, 2.30 mmol) dropwise over 5 min and the reaction mixture was stirred at ambient temperature for 16 h. To the reaction mixture was added saturated ammonium chloride solution (50 mL) and the phases separated. The aqueous layer was washed with dichloromethane (50 mL), and the organic layers combined, dried (MgSO_4), filtered and concentrated *in vacuo*. The residue was purified by flash chromatography on silica gel (4:1 light petroleum: ethyl acetate) to afford the *title compound* **8e** (179 mg, 79%) as an orange solid; mp 98-100 °C (lit.,⁸ mp 98-101 °C); (Found: C, 66.00; H, 4.67; $\text{C}_{12}\text{H}_{10}\text{O}_4$ requires C, 66.05; H, 4.62 %); (Found: $\text{M}+\text{Na}^+$, 241.0467. $\text{C}_{12}\text{H}_{10}\text{O}_4\text{Na}^+$ requires 241.0471); λ_{max} (CH_2Cl_2)/nm 246 (log ϵ 5.16), 380 (4.28); ν_{max} (CHCl_3)/ cm^{-1} 3012, 1663, 1587, 1335, 994; δ_{H} (400 MHz; CDCl_3) 7.75 (1 H, dd, J 7.5, 1.0, ArH), 7.63

(1 H, app. b. t, J 8.3, ArH), 7.51 (1 H, dd, J 8.3, 1.0, ArH), 6.85 (2 H, app. br. s, CH), 5.33 (2 H, s, CH₂), 3.52 (3 H, s, OCH₃); δ _C (100 MHz; CDCl₃) 184.8 (C), 184.0 (C), 157.0 (C), 140.6 (CH), 136.2 (CH), 134.6 (CH), 133.8 (C), 122.2 (CH), 120.5 (C), 120.3 (CH), 95.0 (CH₂), 56.5 (CH₃); *m/z* (ESI) 241 (M+Na⁺, 100%).

6-Hydroxy-1,4-naphthoquinone **12a**



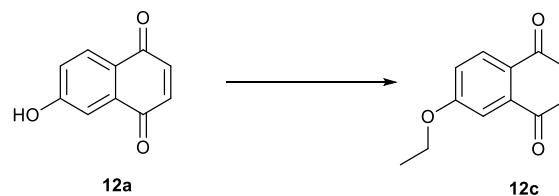
A stirred solution of 1,6-dihydroxynaphthalene **11** (5.00 g, 31.2 mmol) and salcomine (507 mg, 1.56 mmol) in DMF (33 mL) was bubbled rapidly with oxygen at ambient temperature. After 16 h, ether (200 mL) and brine (200 mL) were added and phases separated. The aqueous layer was back-extracted with further ether (10 × 200 mL) and the organic layers combined, dried (Na₂SO₄), filtered and concentrated *in vacuo*. The crude product was purified by flash chromatography on silica gel (7:3 light petroleum: ethyl acetate) to give the *title compound* **12a** (2.50 g, 46%) as a bright orange solid; mp 170–171 °C (decomp) (lit.,⁹ mp 170 °C); (Found: C, 68.67; H, 3.42; C₁₀H₆O₃ requires C, 68.97; H, 3.47 %); (Found: M-H⁺, 173.0238. C₁₀H₅O₃⁻ requires 173.0244); λ_{max} (CH₂Cl₂)/nm 274 (log ϵ 5.26), 385 (3.26); ν_{max} (CHCl₃)/cm⁻¹ 3689, 3607, 3045, 1602, 1523, 1240; δ _H (400 MHz; MeOD) 7.92 (1 H, d, J 8.5, ArH), 7.34 (1 H, d, J 2.6, ArH), 7.13 (1 H, dd, J 8.5, 2.6, ArH), 6.90 (2 H, d, J 2.1, CH); δ _C (100 MHz; MeOD) 186.7 (C), 185.6 (C), 164.6 (C), 140.3 (CH), 139.5 (CH), 135.6 (C), 130.2 (CH), 125.8 (C), 121.9 (CH), 113.2 (CH); *m/z* (ESI) 173 (M-H⁺, 100%).

6-Methoxy-1,4-naphthoquinone **12b**



To 6-hydroxy-1,4-naphthoquinone **12a** (100 mg, 0.57 mmol) and silver(I) oxide (200 mg, 0.92 mmol) in CH_2Cl_2 (2 mL) was added iodomethane (0.1 mL, 1.60 mmol) and the mixture stirred at ambient temperature for 3 d. The reaction mixture was filtered through Celite, washed with dichloromethane (50 mL) and concentrated *in vacuo* to afford the *title compound* **12b** (107 mg, 100%) as a fine yellow solid; mp 132–133 °C (lit.,¹⁰ mp 133–135 °C); (Found: C, 69.82; H, 4.22; $\text{C}_{11}\text{H}_8\text{O}_3$ requires C, 70.21; H, 4.29 %); (Found: $\text{M}+\text{H}^+$, 189.0546. $\text{C}_{11}\text{H}_9\text{O}_3^+$ requires 189.0546); λ_{max} (CH_2Cl_2)/nm 261 (log ε 4.91), 382 (3.89); ν_{max} (CHCl_3)/cm⁻¹; 3012, 1667, 1593, 131, 1292; δ_{H} (400 MHz; CDCl_3) 8.02 (1 H, d, *J* 8.6, ArH), 7.50 (1 H, d, *J* 2.7, ArH), 7.21 (1 H, dd, *J* 8.6, 2.7, ArH), 6.92 (2 H, app br. s, CH), 3.95 (3 H, s, OCH₃); δ_{C} (100 MHz; CDCl_3) 185.2 (C), 184.1 (C), 164.1 (C), 139.0 (CH), 138.2 (CH), 133.9 (C), 128.9 (CH), 125.5 (C), 120.5 (CH), 109.6 (CH), 55.9 (CH₃); *m/z* (ESI) 399 (2M+Na⁺, 100%), 211 (M+Na⁺, 98), 189 (M+H⁺, 47).

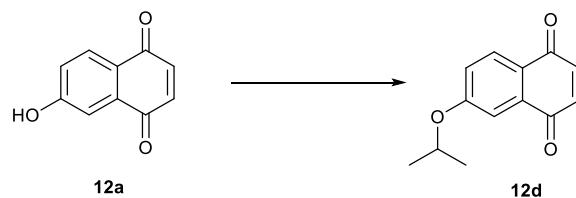
6-Ethoxy-1,4-naphthoquinone **12c**



To 6-hydroxy-1,4-naphthoquinone **12a** (50 mg, 0.29 mmol) and silver(I) oxide (106 mg, 0.46 mmol) in CH_2Cl_2 (1 mL) was added iodoethane (0.06 mL, 0.81 mmol) and the

mixture stirred at ambient temperature for 16 h. The reaction mixture was filtered through Celite, washed with dichloromethane (50 mL) and concentrated *in vacuo* to afford the *title compound* **12c** (44 mg, 76%) as a fine yellow solid; mp 116–117 °C; (Found: C, 71.07; H, 4.99; $C_{13}H_{12}O_3$ requires C, 71.28; H, 4.98%); (Found: M+Na⁺, 225.0522. $C_{12}H_{10}O_3Na^+$ requires 225.0522); λ_{max} (CH_2Cl_2)/nm 255 (log ε 4.37), 263 (4.39), 383 (3.27), 386 (3.20); ν_{max} (CHCl_3)/cm⁻¹; 3012, 2415, 1666, 1595, 1333, 1239, 965; δ_{H} (400 MHz; CDCl_3) 8.02 (1 H, d, *J* 8.6, ArH), 7.50 (1 H, d, *J* 2.7, ArH), 7.20 (1 H, dd, *J* 8.6, 2.7, ArH), 6.92 (2 H, app br. s, CH), 4.19 (2 H, q, *J* 6.9, CH_2), 1.48 (3 H, t, *J* 7.0, CH_3); δ_{C} (100 MHz; CDCl_3) 185.2 (C), 184.1 (C), 163.6 (C), 139.0 (CH), 138.2 (CH), 133.9 (C), 128.9 (CH), 125.3 (C), 120.8 (CH), 110.1 (CH), 64.4 (CH_2), 14.6 (CH_3); *m/z* (ESI) 225 (M+Na⁺, 100%), 203 (M+H⁺, 35).

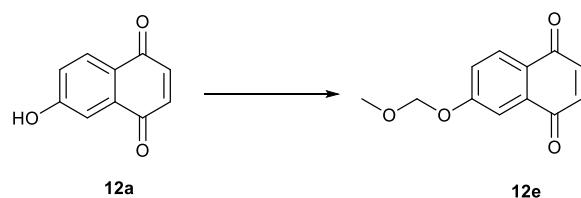
6-Isopropoxy-1,4-naphthoquinone **12d**



To 6-hydroxy-1,4-naphthoquinone **12a** (100 mg, 0.57 mmol) and silver(I) oxide (110 mg, 0.48 mmol) in dichloromethane (1 mL) was added 2-iodopropane (0.1 mL, 1.12 mmol) and the mixture stirred at ambient temperature. After 16 h further silver(I) oxide (220 mg, 0.96 mmol) and 2-iodopropane (0.2 mL, 2.2 mmol) was added and the reaction mixture was stirred for a further 4 h. The reaction mixture was concentrated *in vacuo* to yield the *title compound* **12d** (123 mg, 100%) as a pale green solid; mp 111–113 °C; (Found: M+Na⁺, 239.0671. $C_{13}H_{12}O_3Na^+$ requires 239.0679); λ_{max} (CH_2Cl_2)/nm 257 (log ε 4.54), 264 (4.56), 330 (2.96), 390 (3.54); ν_{max} (CHCl_3)/cm⁻¹; 2983, 1665, 1592, 1491, 1316, 1110, 1045, 961; δ_{H} (400 MHz; CDCl_3) 8.02 (1 H, d, *J* 8.6, ArH), 7.48

(1 H, d, *J* 2.6, ArH), 7.17 (1 H, dd, *J* 8.6, 2.7, ArH), 6.92 (2 H, app. br. s, CH), 4.76 (1 H, sept, *J* 6.0, CH), 1.40 (6 H, d, *J* 6.0, CH₃); δ_C (100 MHz; CDCl₃) 185.3 (C), 184.1 (C), 162.7 (C), 139.0 (CH), 138.2 (CH), 133.9 (C), 129.0 (CH), 125.0 (C), 121.7 (CH), 111.0 (CH), 70.8 (CH), 21.8 (CH₃); *m/z* (ESI) 239 (M+Na⁺, 100%), 217 (M+H⁺, 11).

6-Methoxymethoxy-1,4-naphthoquinone 12e



To 6-hydroxy-1,4-naphthoquinone **12a** (100 mg, 0.57 mmol) and MOM-Cl (0.12 mL, 1.44 mmol) in dichloromethane (2 mL) at 0 °C was added DIPEA (0.2 mL, 1.15 mmol) dropwise over 5 min and the reaction mixture was stirred at ambient temperature for 16 h. To the reaction mixture was added saturated ammonium chloride solution (50 mL) and the phases separated. The aqueous layer was washed with dichloromethane (50 mL) and the organic layers combined, dried (MgSO₄), filtered and concentrated *in vacuo*. The residue was purified by flash chromatography on silica gel (4:1 light petroleum: ethyl acetate) to afford the *title compound* **12e** (82 mg, 66%) as an orange solid; mp 110–112 °C; (Found: M+Na⁺, 241.0468. C₁₂H₁₀O₄Na⁺ requires 241.0471); λ_{max} (CH₂Cl₂)/nm 254 (log ε 4.69), 260 (4.69), 372 (3.69); ν_{max} (CHCl₃)/cm⁻¹; 3012, 1668, 1595, 1312, 988; δ_H (400 MHz; CDCl₃) 8.00 (1 H, d, *J* 8.6, ArH), 7.62 (1 H, d, *J* 2.6, ArH), 7.32 (1 H, dd, *J* 8.6, 2.6, ArH), 6.91 (2 H, app br. s, CH), 5.29 (2 H, s, CH₂), 3.48 (3 H, s, CH₃); δ_C (100 MHz; CDCl₃) 184.8 (C), 184.0 (C), 161.6 (C), 138.8 (CH), 138.3 (CH), 133.8 (C), 128.8 (CH), 126.1 (C), 121.5 (CH), 112.5 (CH), 94.1 (CH₂), 56.4 (CH₃); *m/z* (ESI) 241 (M+Na⁺, 100%), 219 (M+H⁺, 17).

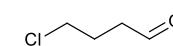
X-Ray Crystallographic Studies

All data sets were collected either with a SuperNova Cu diffractometer or a GV1000 diffractometer. The crystal was kept at 120(2)K during data collection. Using Olex2, the structure was solved with the olex2.solve structure solution program using Charge Flipping and refined with the ShelXL refinement package using Least Squares minimisation. Crystal structure data and details are listed in Table 1.

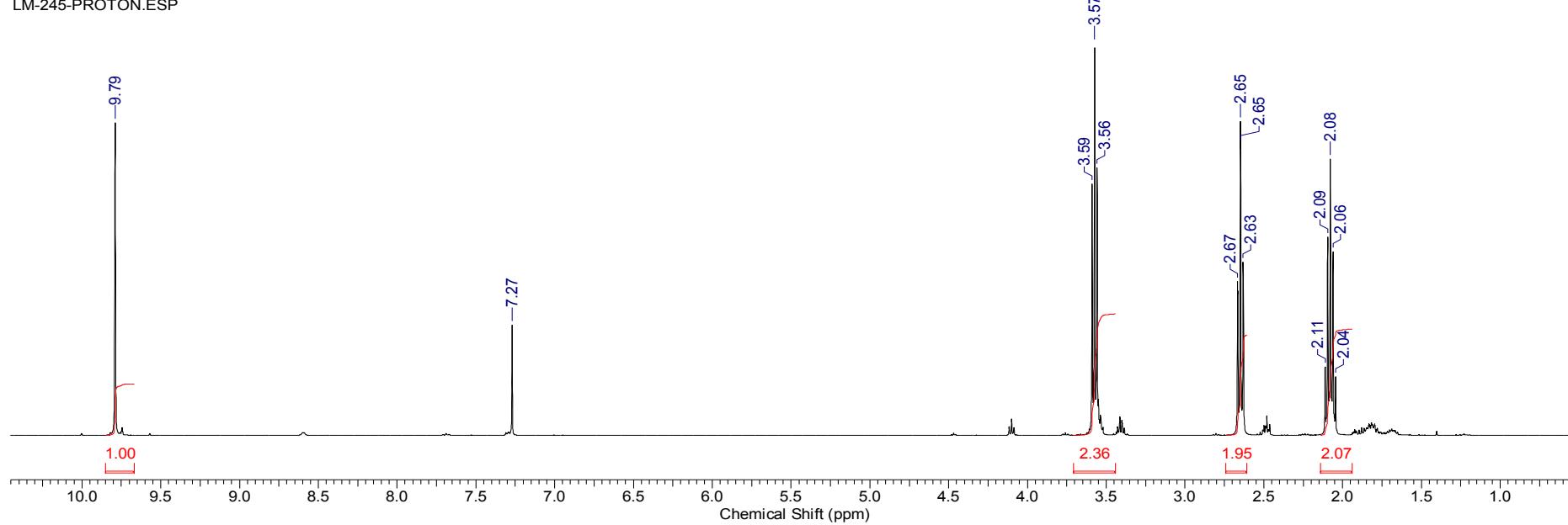
Table 1. Crystal structure data

Compound	9e	10c	13b	
CCDC	952123	952121	952122	
Empirical formula	C ₁₆ H ₁₈ O ₅	C ₁₆ H ₁₈ O ₄	C ₁₅ H ₁₆ O ₄	
Formula weight	290.30	274.30	260.28	
Crystal size/mm ³	0.51 × 0.0864 × 0.0426	0.126 × 0.1016 × 0.0629	0.1243 × 0.0791 × 0.0316	
Crystal system	monoclinic	monoclinic	orthorhombic	
Space group	P2 ₁ /c	P2 ₁ /n	Pbca	
Unit cell dimensions	a/Å	15.7243(11)	7.7968(2)	12.8287(3)
	b/Å	5.0123(3)	21.1512(5)	7.9789(3)
	c/Å	18.5815(12)	9.1046(3)	24.2643(9)
	α/°	90	90	90
	β/°	107.285	114.189	90
	γ/°	90	90	90
Volume/Å ³		1398.35(17)	1369.64(7)	2483.67(15)
Z		4	4	8
Reflections collected		5181	12604	13106
Independent reflections		2731 [R(int) = 0.0207]	2766 [R(int) = 0.0207]	2494 [R(int) = 0.0371]
Final R indexes R [I>=2σ (I)]		R ₁ = 0.0362, wR ₂ = 0.0987	R ₁ = 0.0329, wR ₂ = 0.0900	R ₁ = 0.0438, wR ₂ = 0.1128
Final R indexes (all data)		R ₁ = 0.0362, wR ₂ = 0.0987	R ₁ = 0.0378, wR ₂ = 0.0941	R ₁ = 0.0525 wR ₂ = 0.1191

4-Chlorobutanal

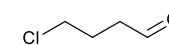


LM-245-PROTON.ESP

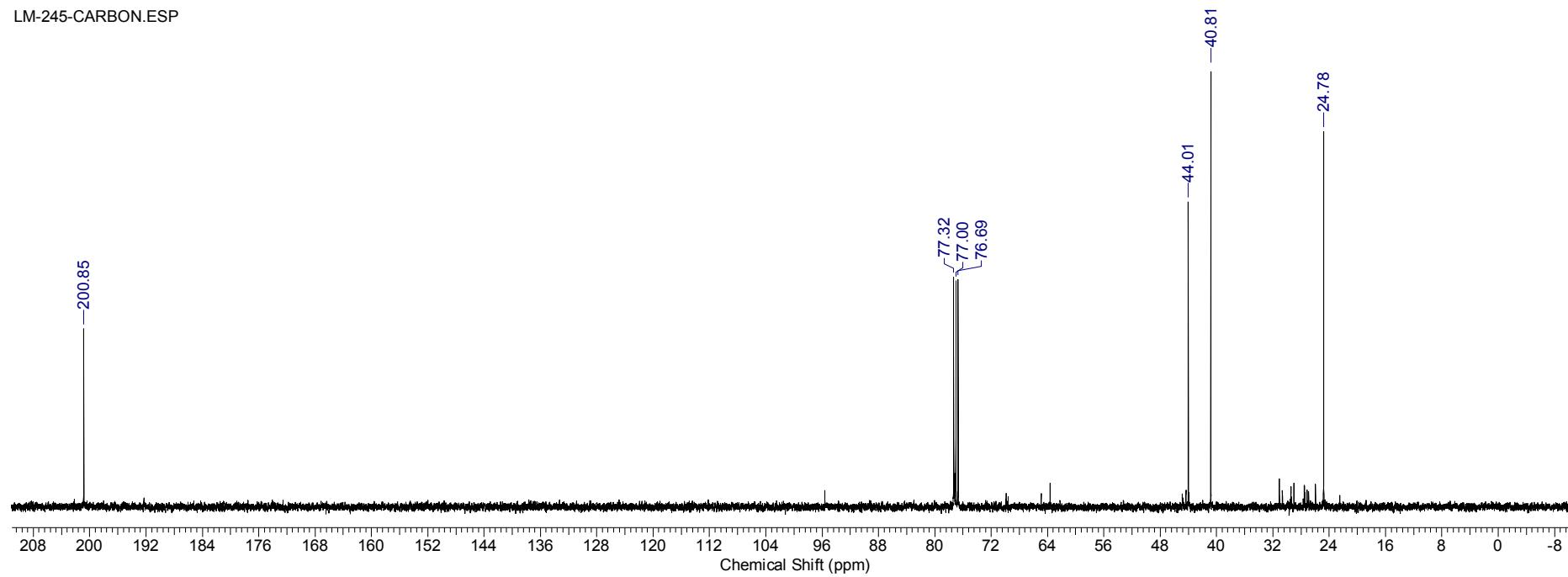


^1H NMR at 400 MHz in CDCl_3

4-Chlorobutanal

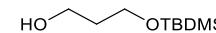


LM-245-CARBON.ESP

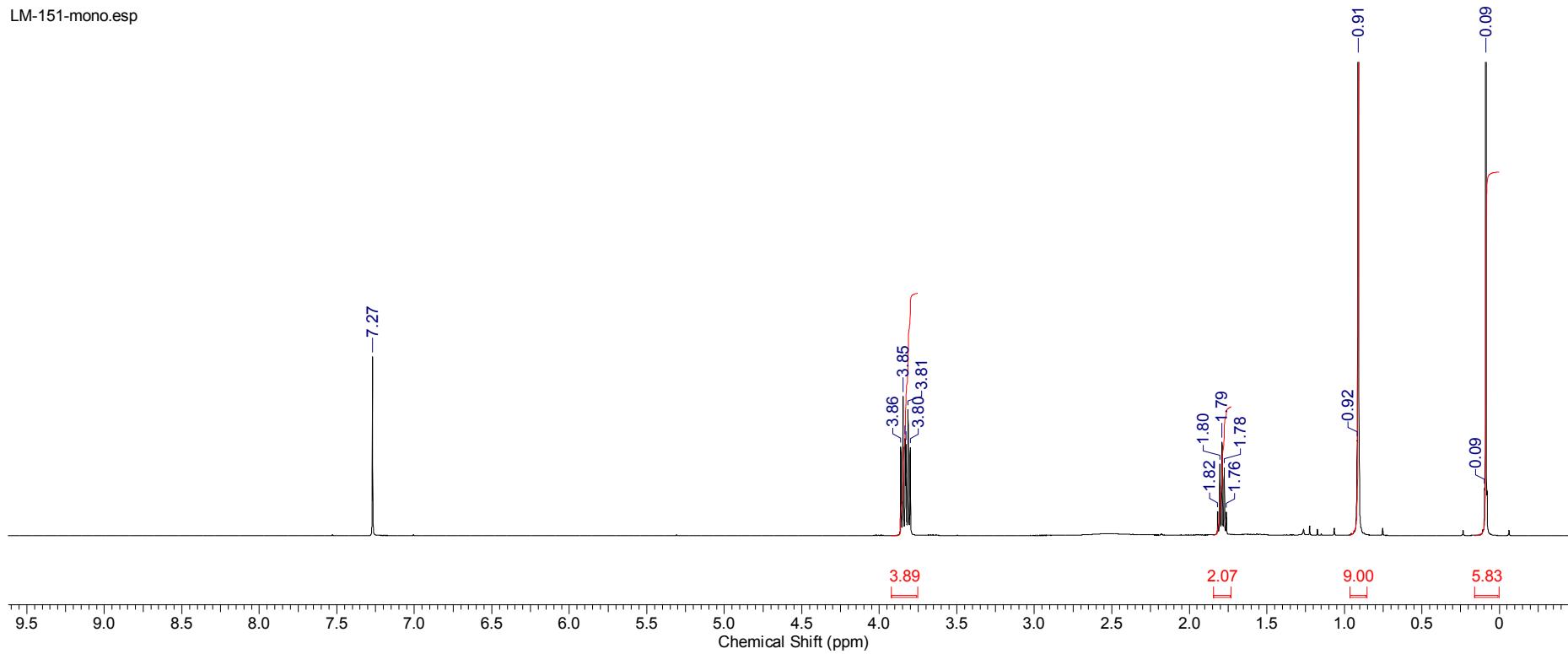


¹³C NMR at 100 MHz in CDCl₃

3-((*t*-butyldimethylsilyl)oxy)propan-1-ol



LM-151-mono.esp

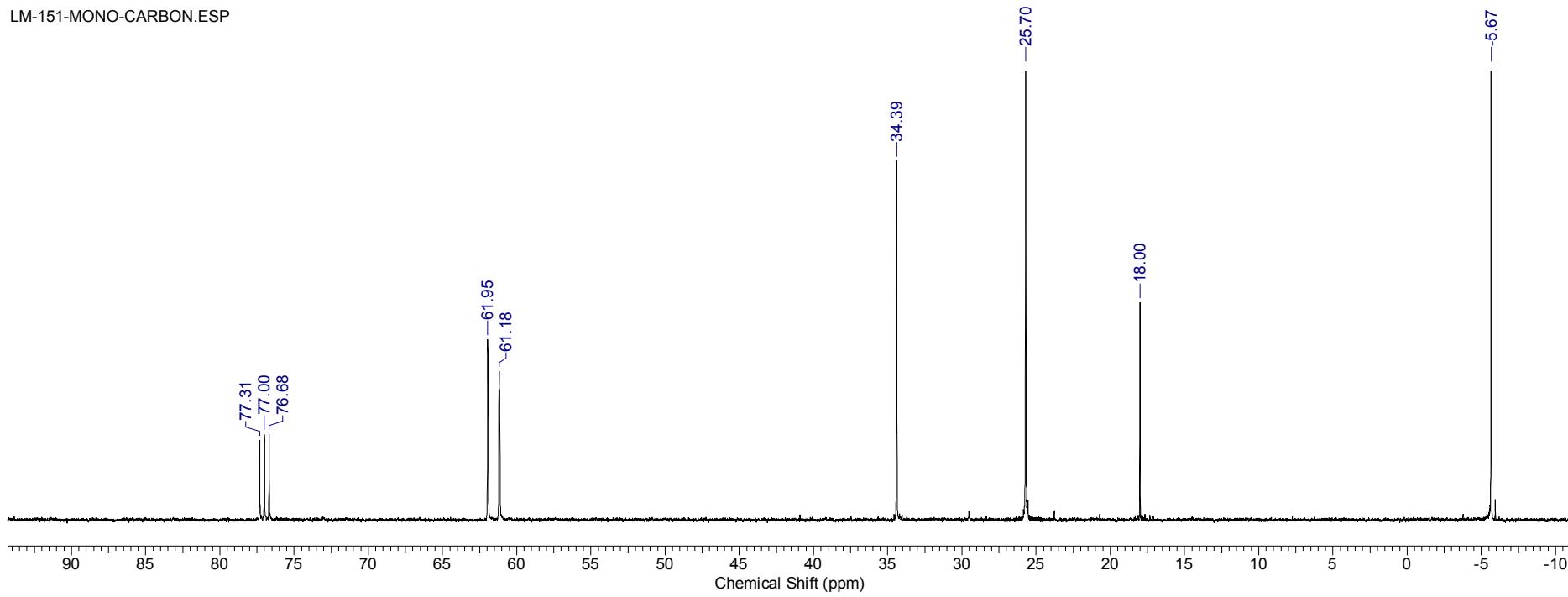


^1H NMR at 400 MHz in CDCl_3

3-((*t*-butyldimethylsilyl)oxy)propan-1-ol

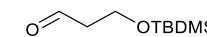


LM-151-MONO-CARBON.ESP

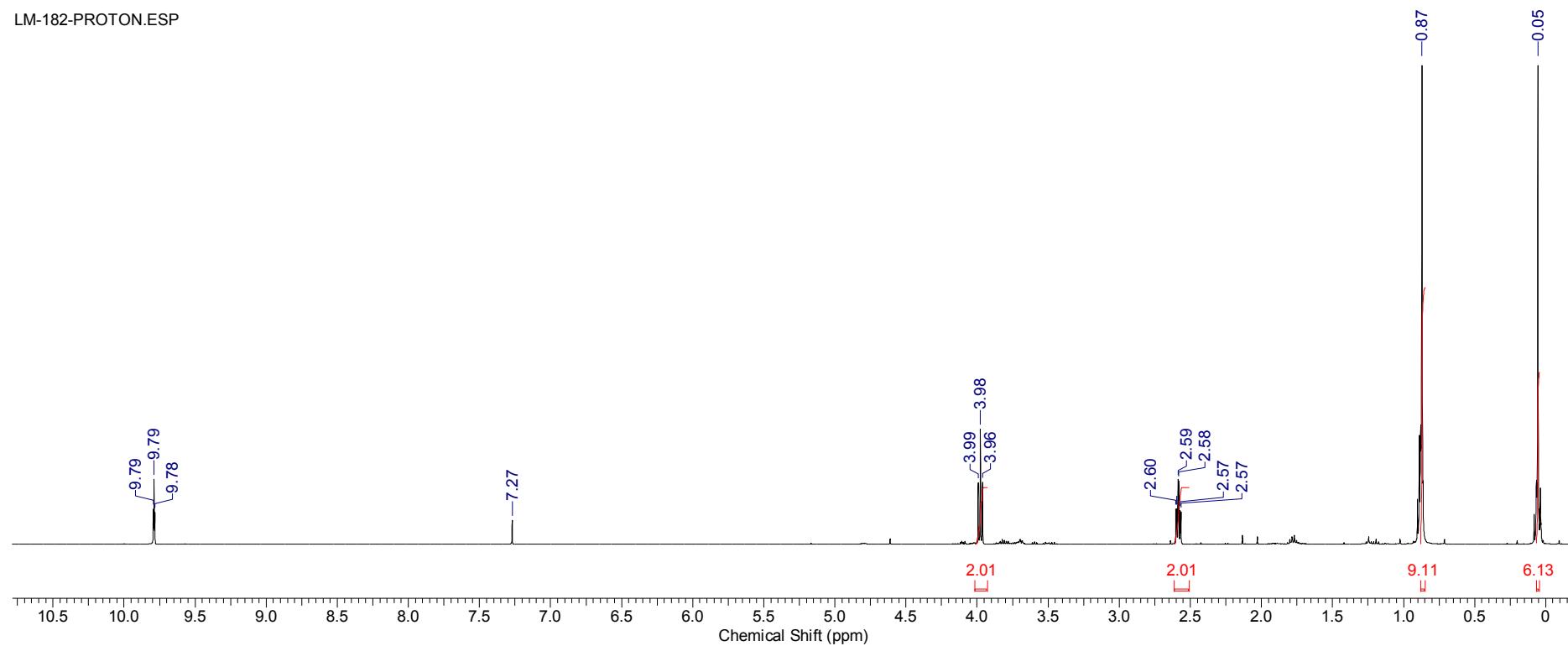


^{13}C NMR at 100 MHz in CDCl_3

3-(*t*-Butyldimethylsiloxy)propanal

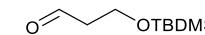


LM-182-PROTON.ESP

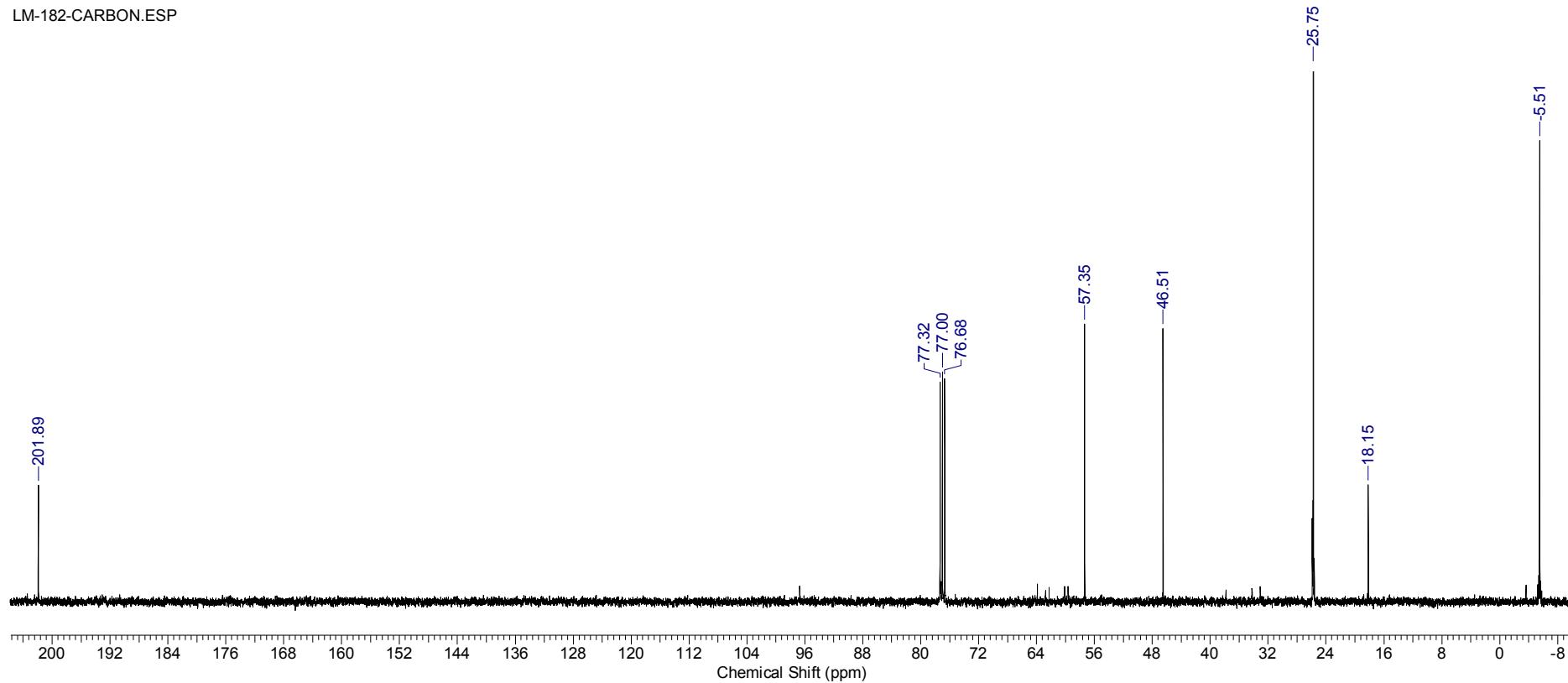


^1H NMR at 400 MHz in CDCl_3

3-(t-Butyldimethylsiloxy)propanal

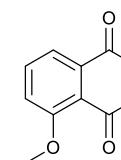


LM-182-CARBON.ESP

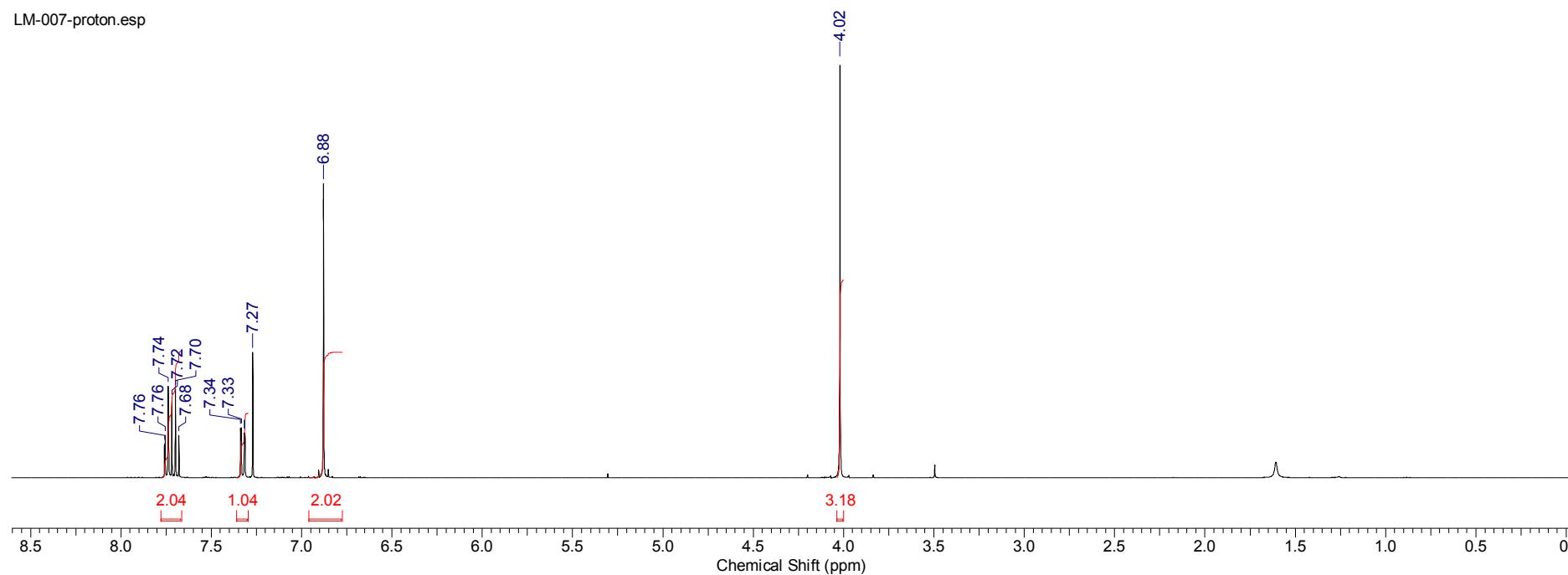


^{13}C NMR at 100 MHz in CDCl_3

5-Methoxy-1,4-naphthoquinone 8b

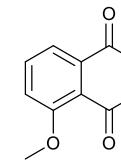


LM-007-proton.esp

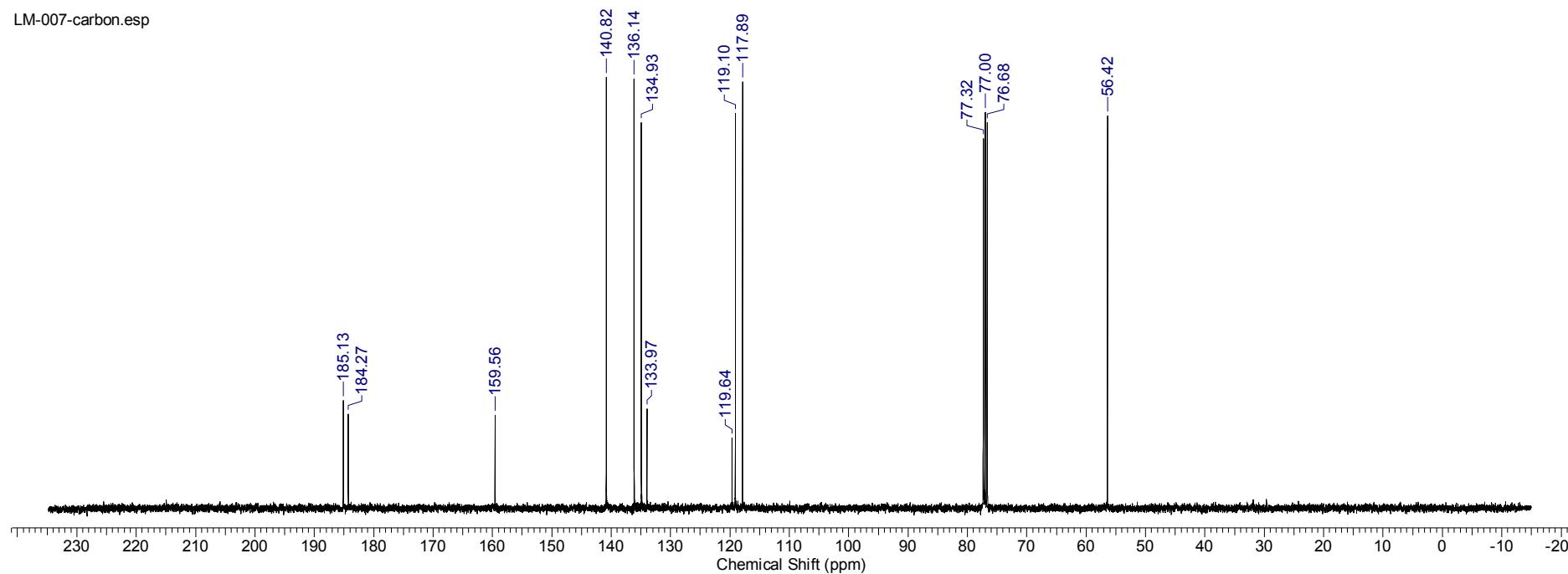


^1H NMR at 400 MHz in CDCl_3

5-Methoxy-1,4-naphthoquinone 8b

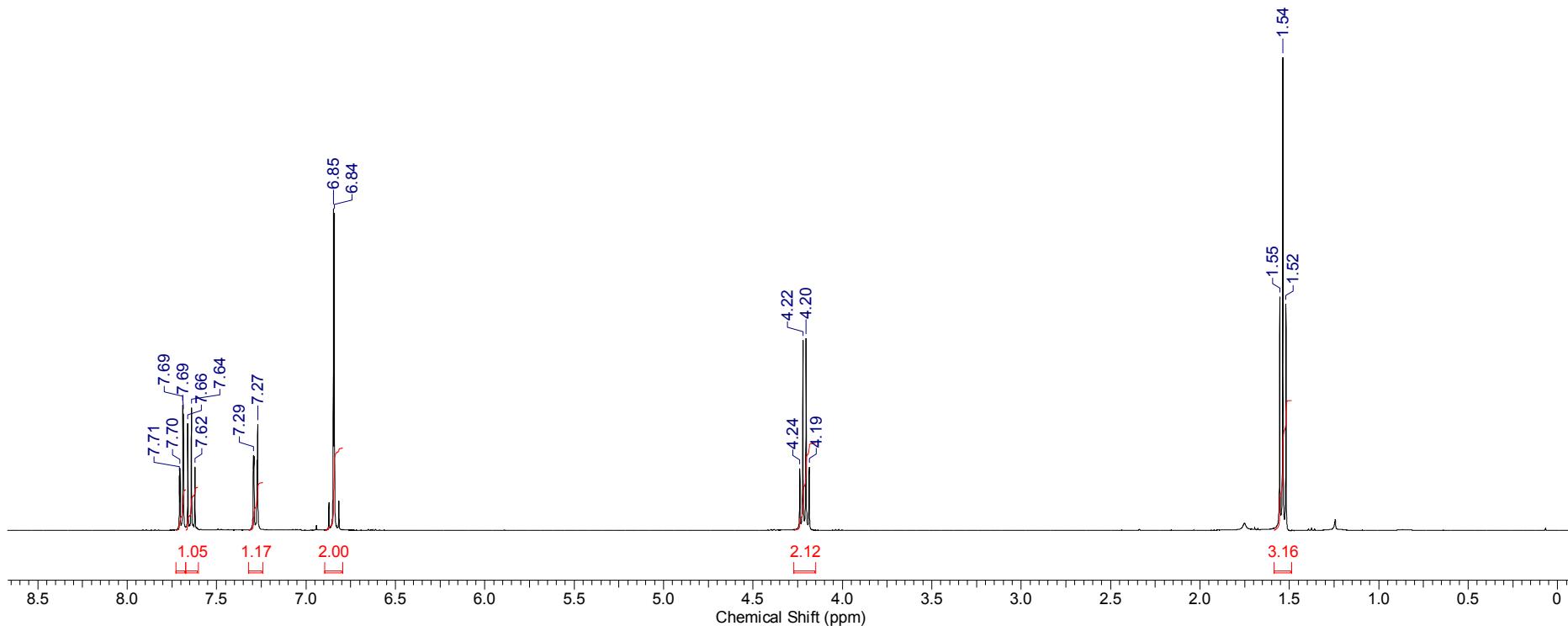
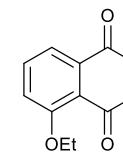


LM-007-carbon.esp



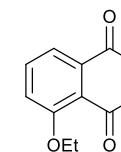
^{13}C NMR at 100 MHz in CDCl_3

5-Ethoxy-1,4-naphthoquinone 8c

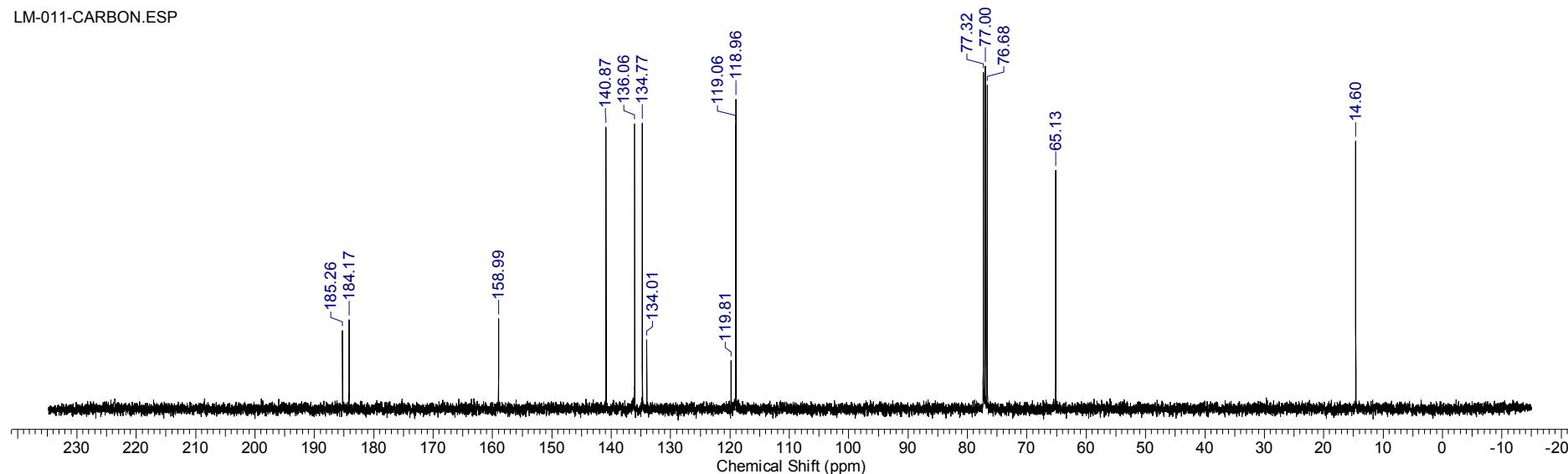


¹H NMR at 400 MHz in CDCl₃

5-Ethoxy-1,4-naphthoquinone 8c

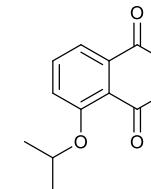


LM-011-CARBON.ESP

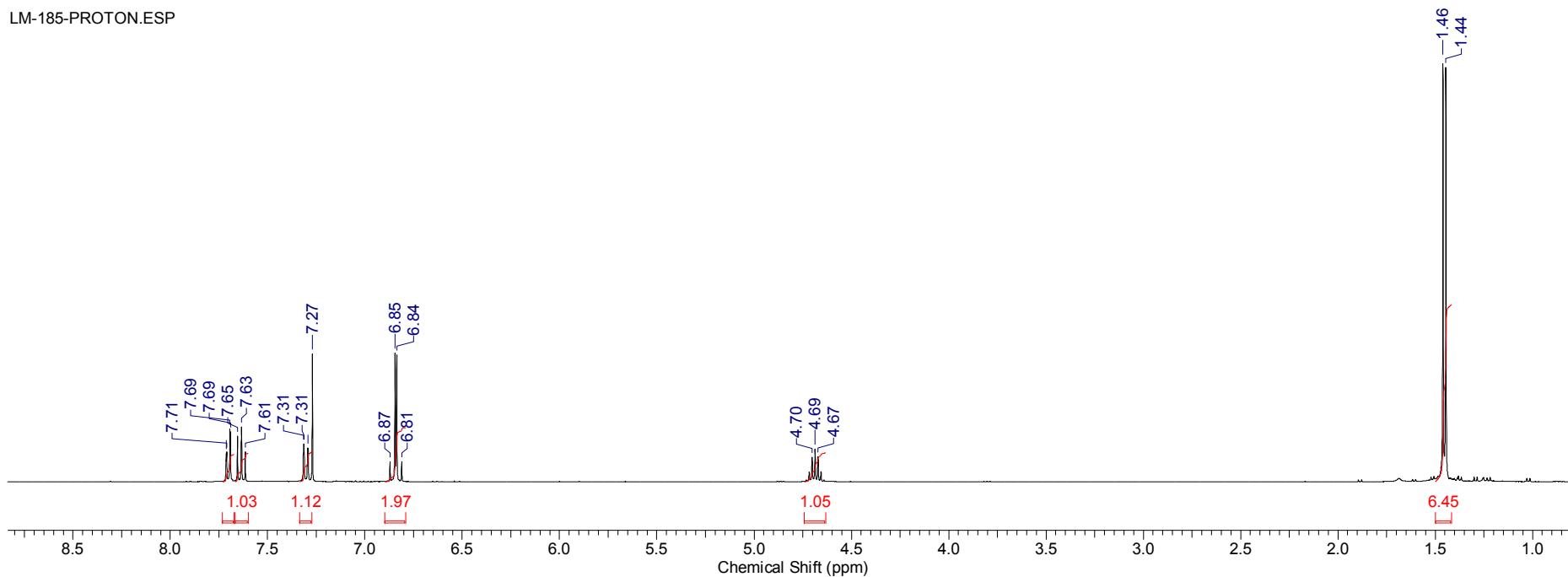


^{13}C NMR at 100 MHz in CDCl_3

5-Isopropoxy-1,4-Naphthoquinone 8d

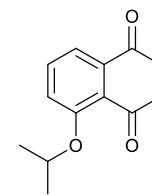


LM-185-PROTON.ESP

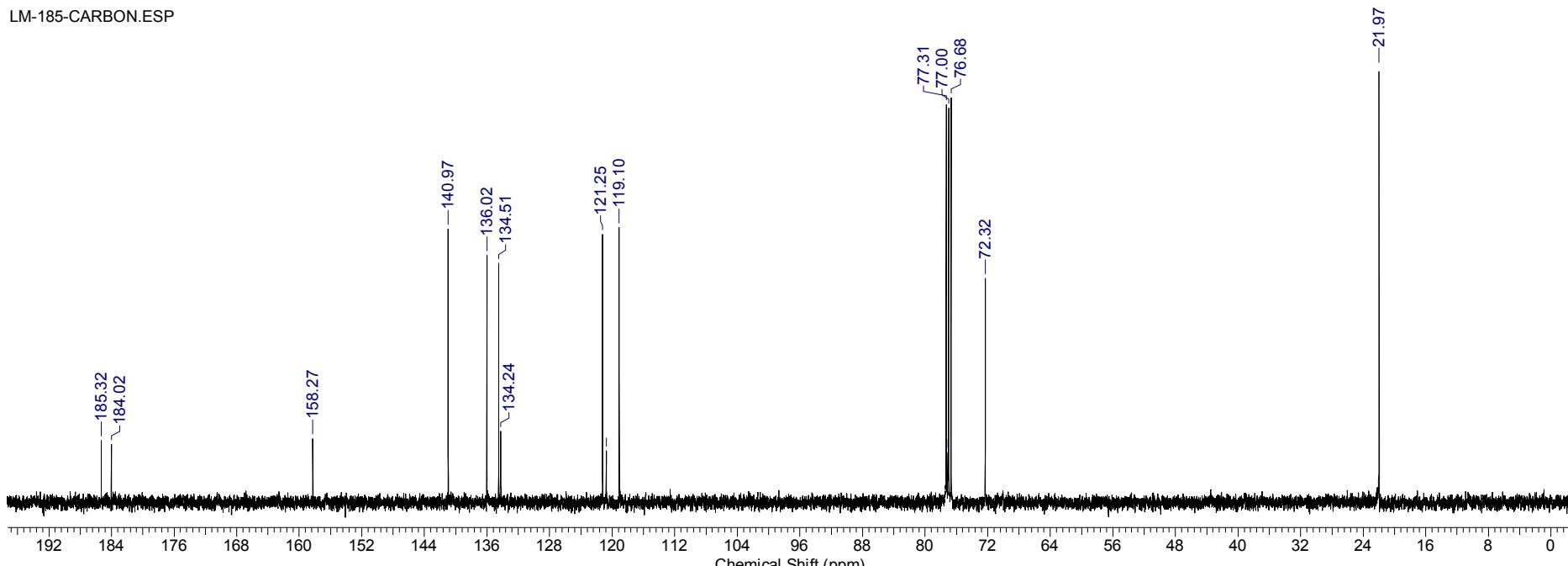


¹H NMR at 400 MHz in CDCl₃

5-Isopropoxy-1,4-Naphthoquinone 8d

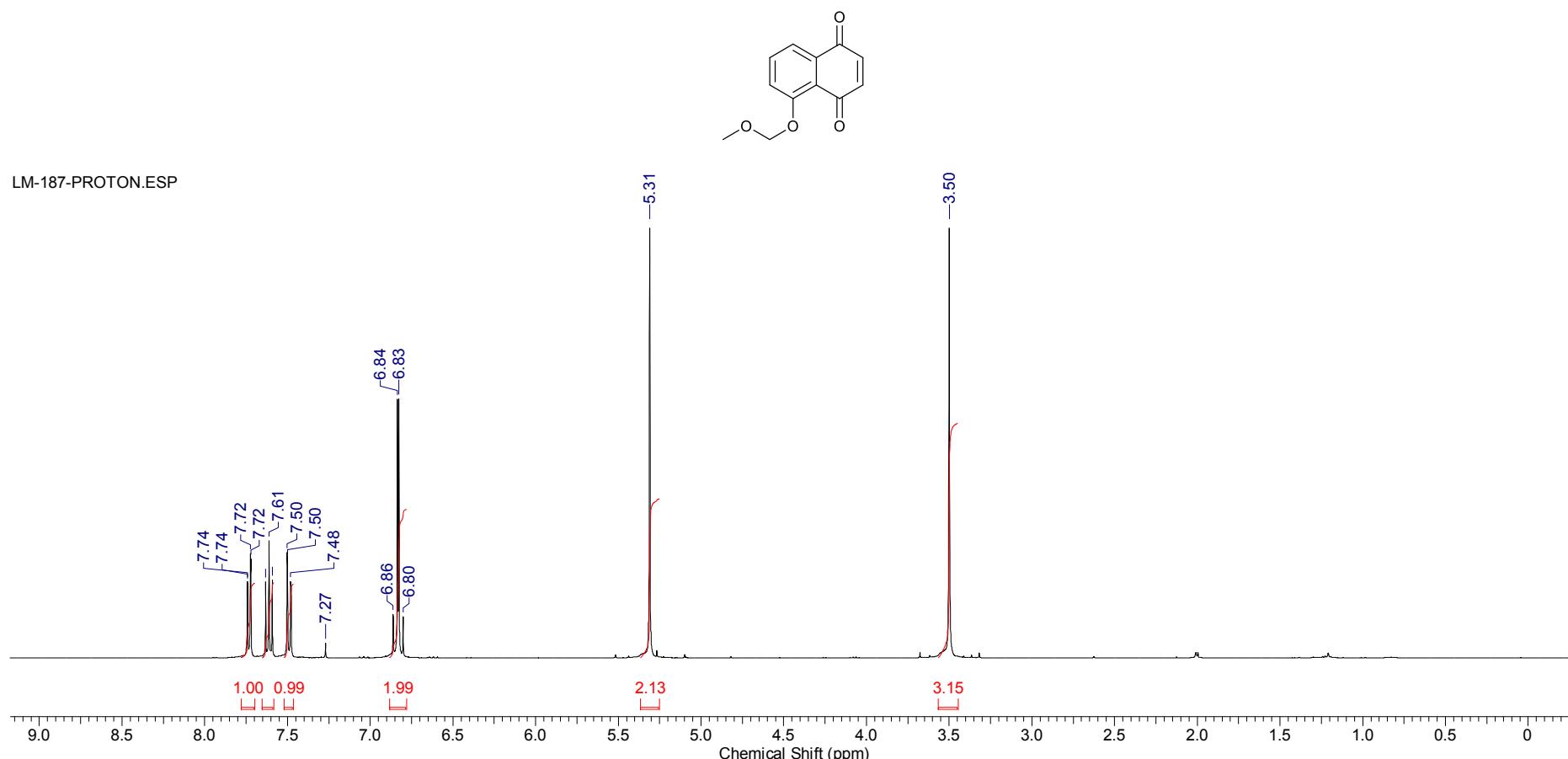


LM-185-CARBON.ESP



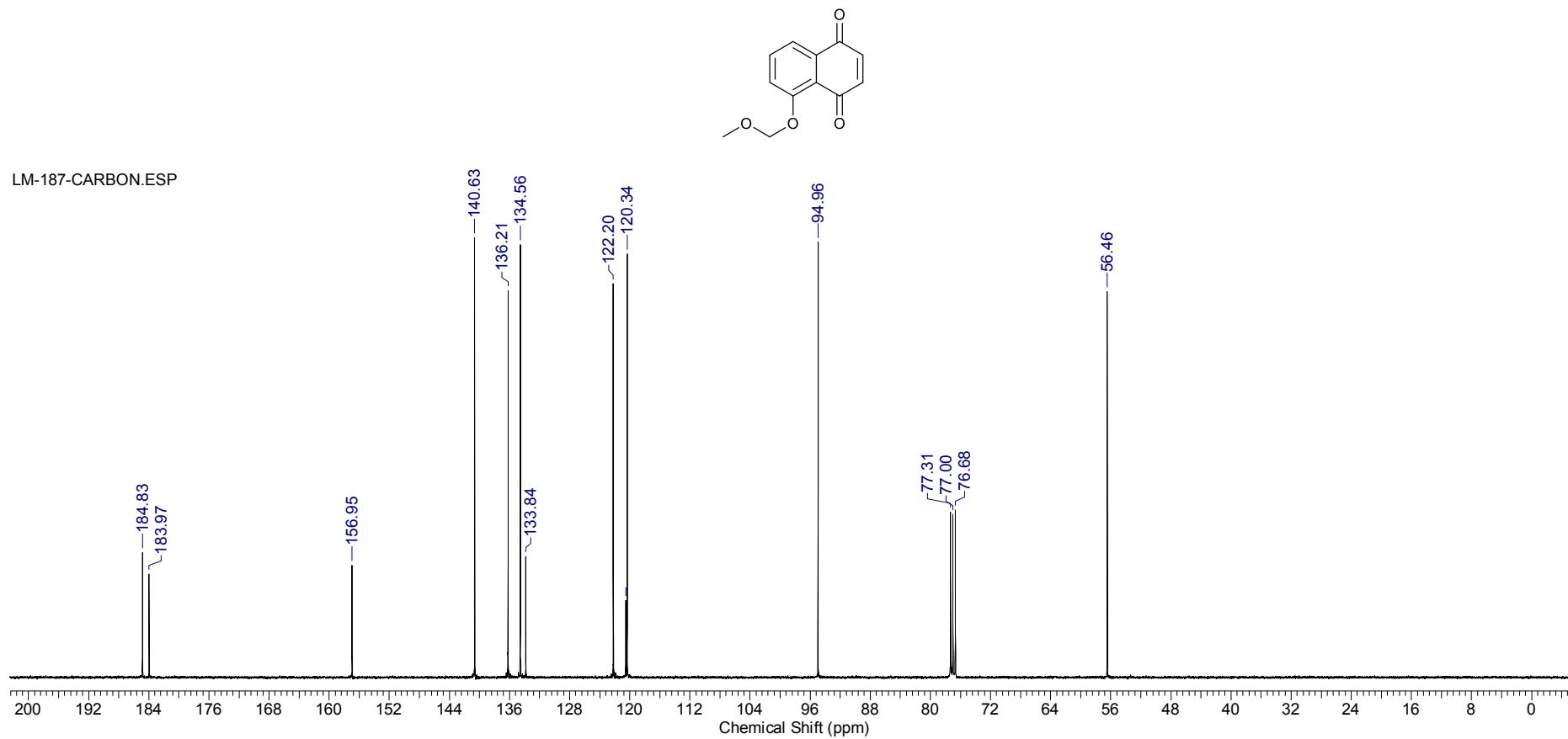
^{13}C NMR at 100 MHz in CDCl_3

5-Methoxymethoxy-1,4-naphthoquinone 8e



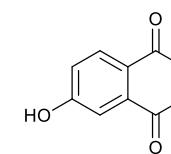
^1H NMR at 400 MHz in CDCl_3

5-Methoxymethoxy-1,4-naphthoquinone 8e

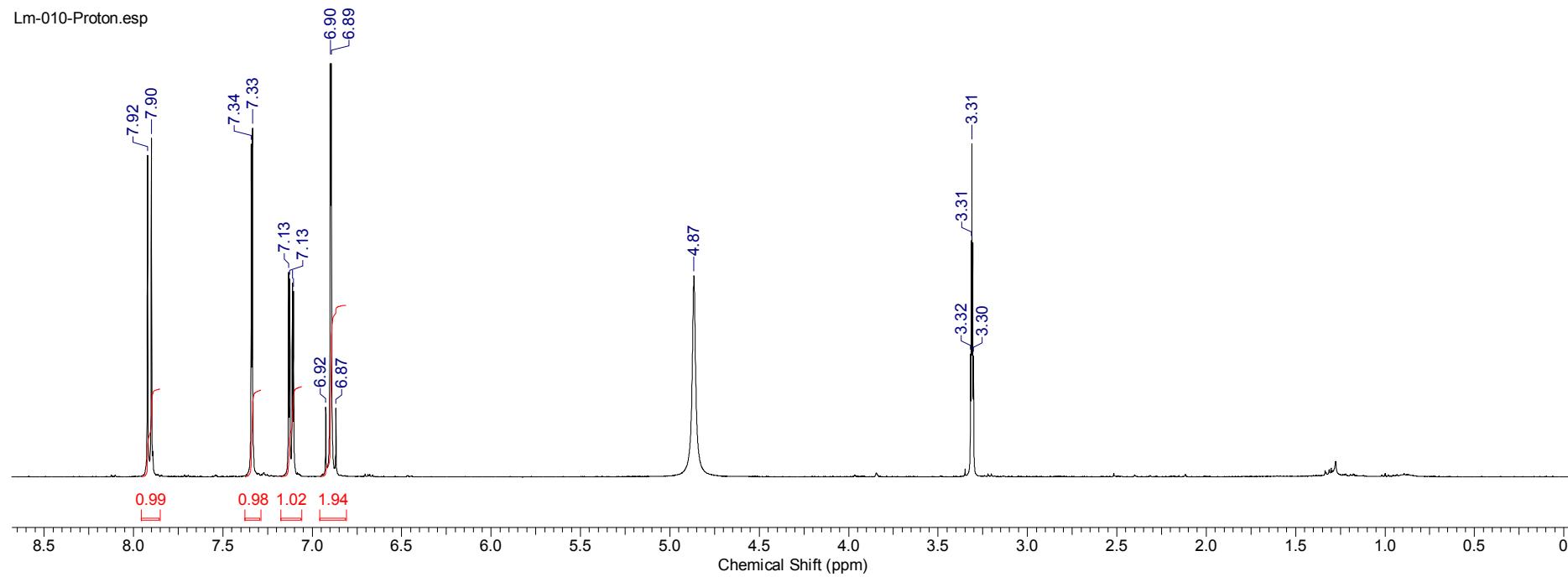


^{13}C NMR at 100 MHz in CDCl_3

6-Hydroxy-1,4-naphthoquinone 12a

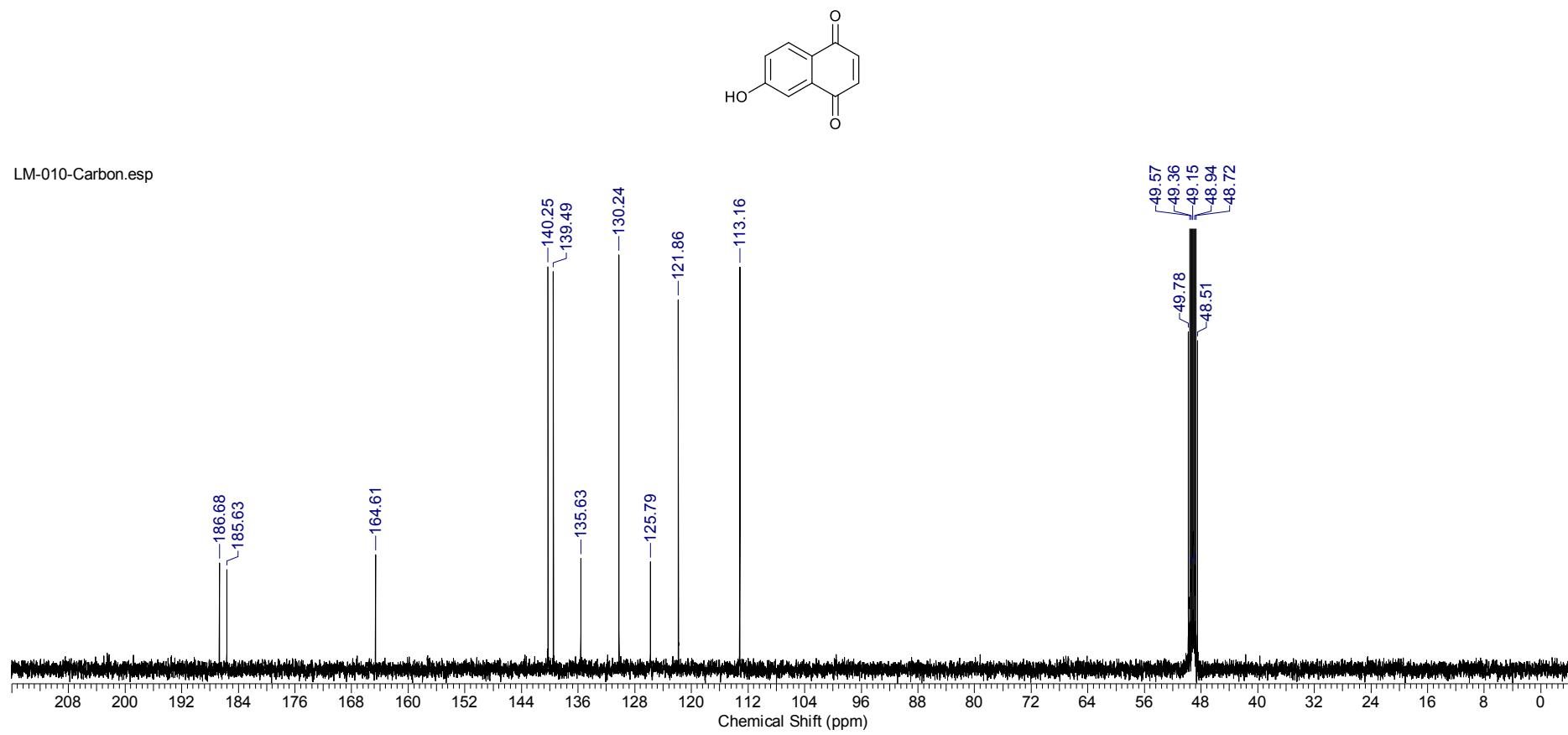


Lm-010-Proton.esp



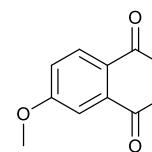
^1H NMR at 400 MHz in MeOD

6-Hydroxy-1,4-naphthoquinone 12a

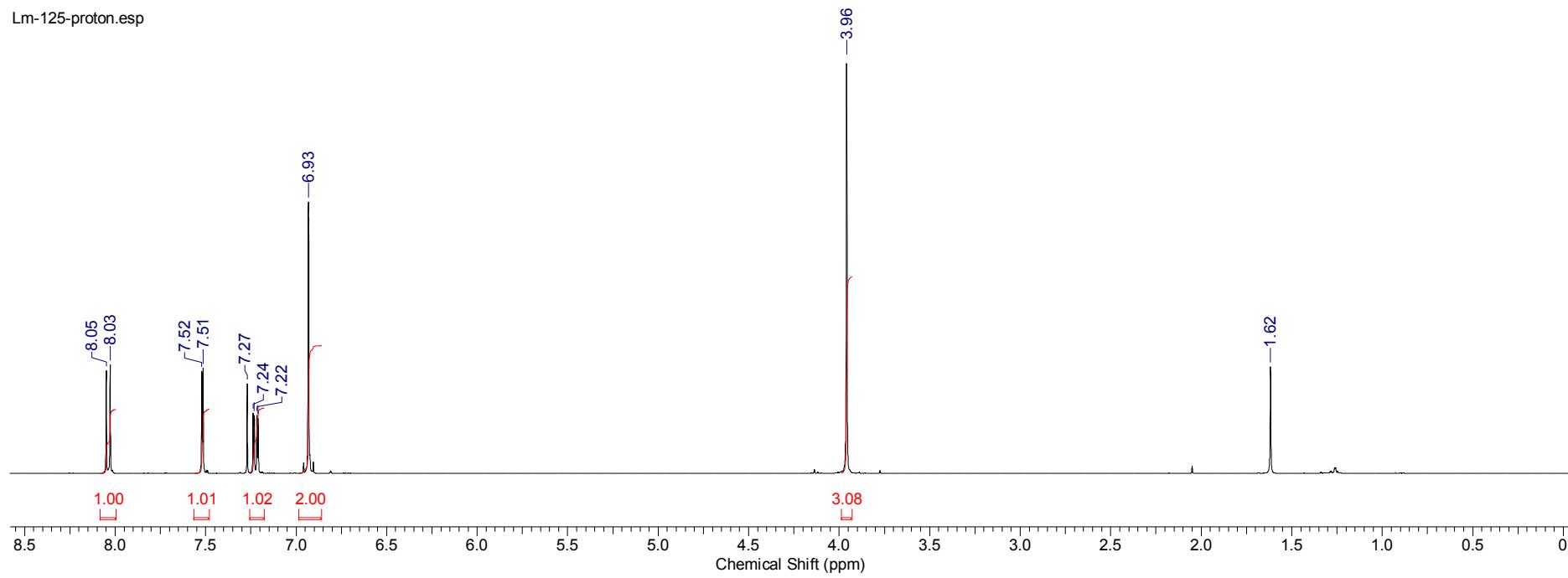


^{13}C NMR at 100 MHz in MeOD

6-Methoxy-1,4-naphthoquinone 12b

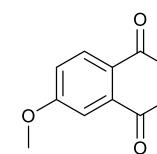


Lm-125-proton.esp

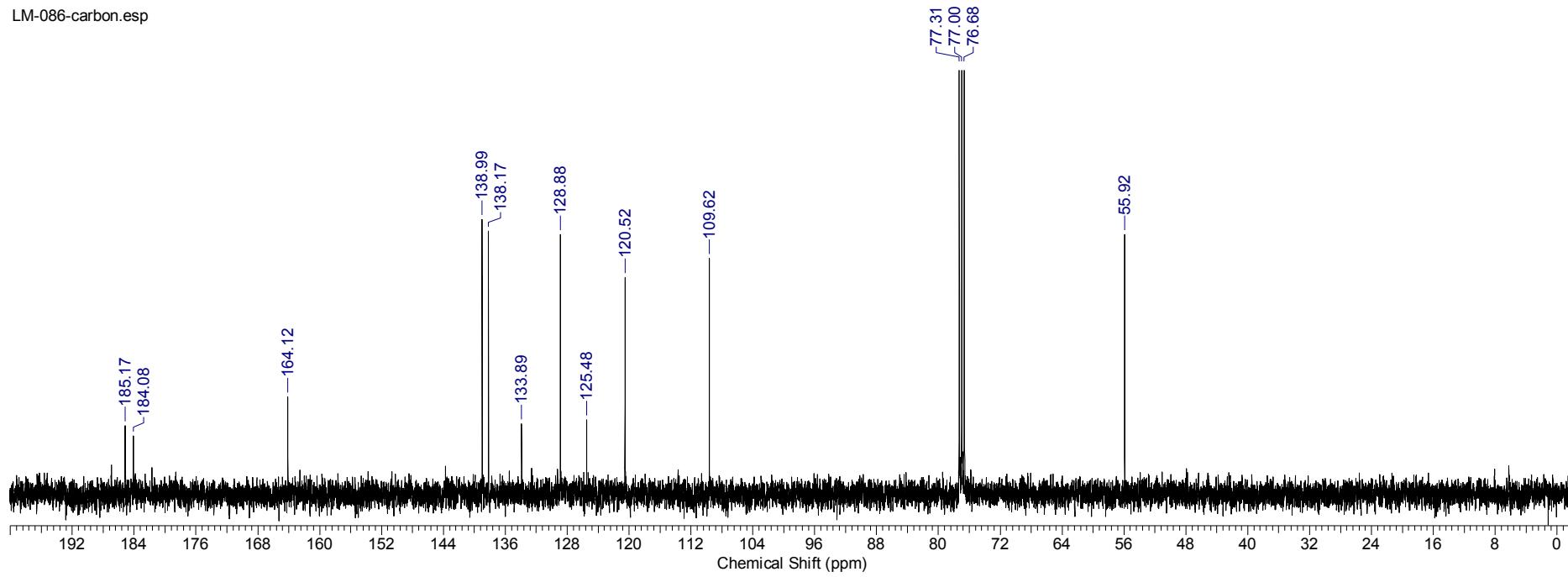


^1H NMR at 400 MHz in CDCl_3

6-Methoxy-1,4-naphthoquinone 12b

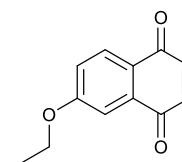


LM-086-carbon.esp

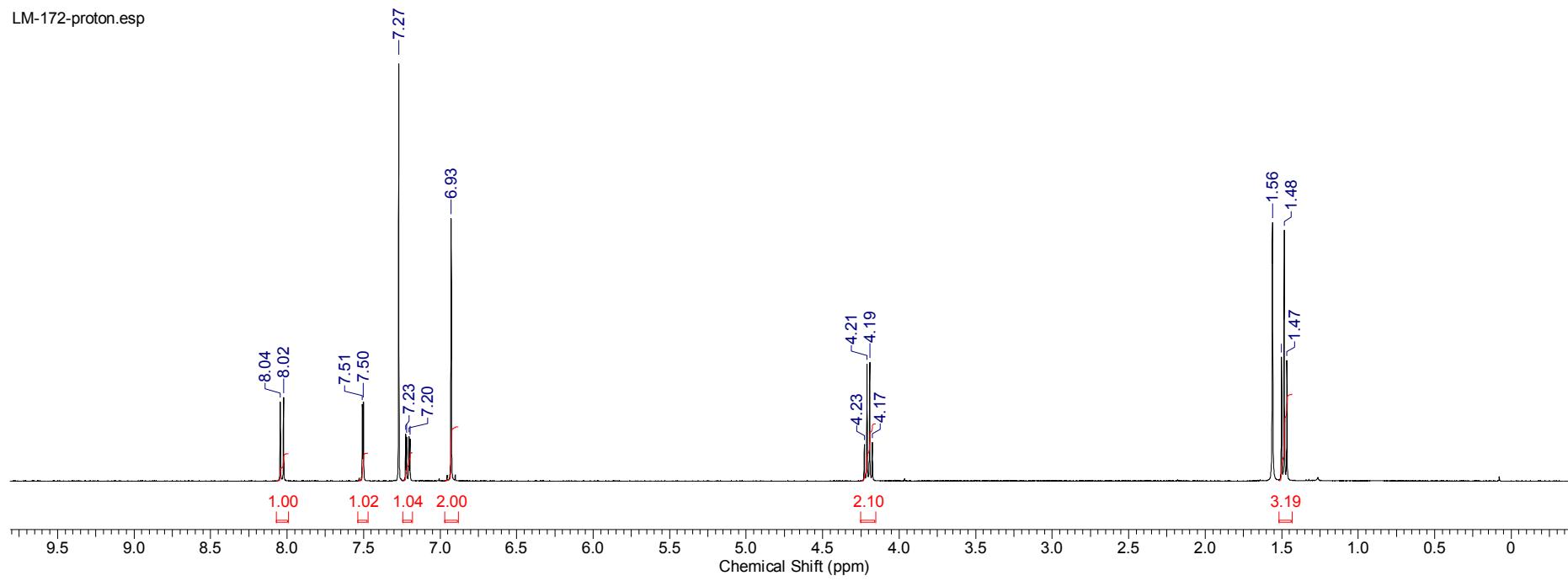


^{13}C NMR at 100 MHz in CDCl_3

6-Ethoxy-1,4-naphthoquinone 12c

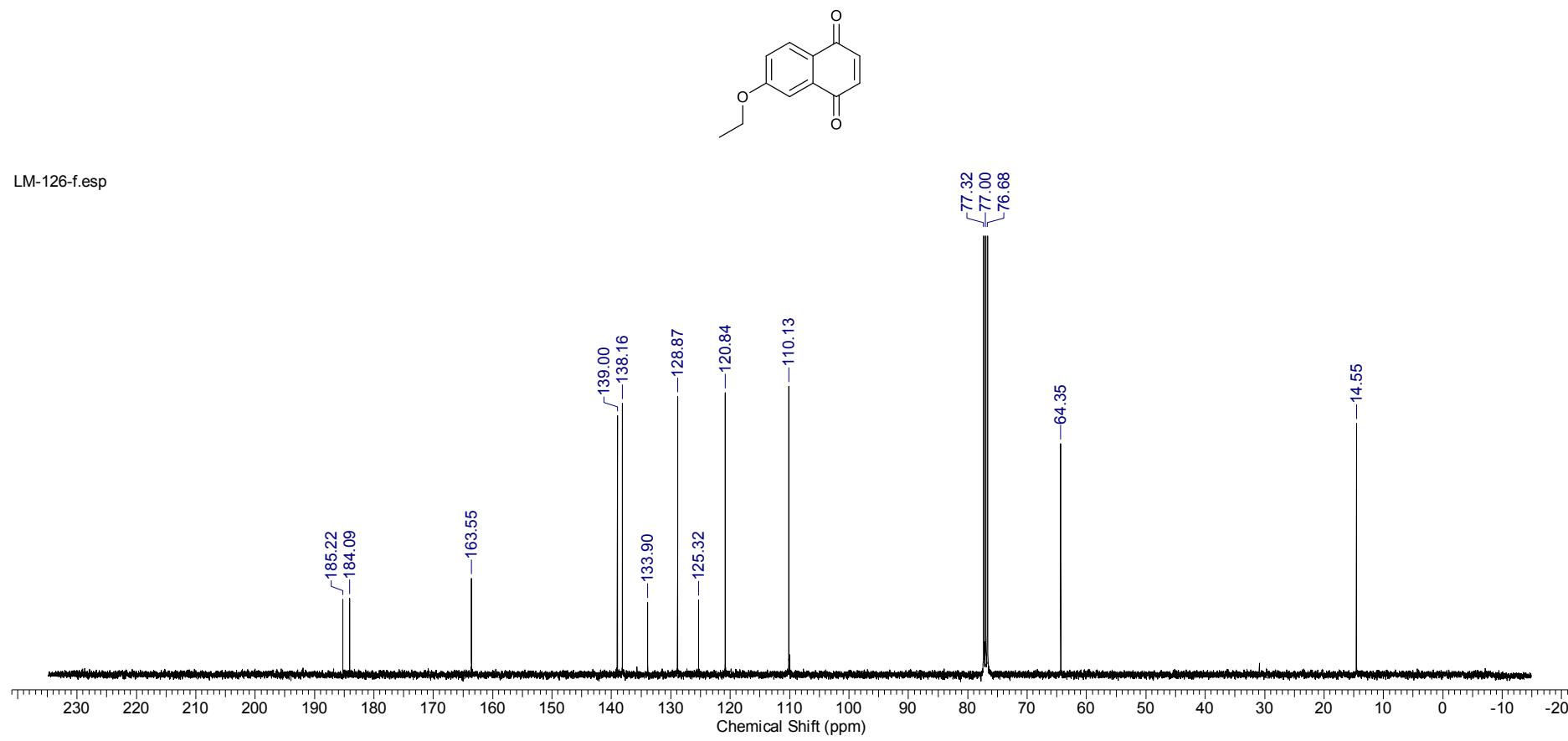


LM-172-proton.esp



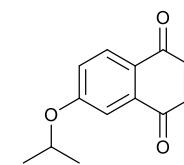
¹H NMR at 400 MHz in CDCl₃

6-Ethoxy-1,4-naphthoquinone 12c

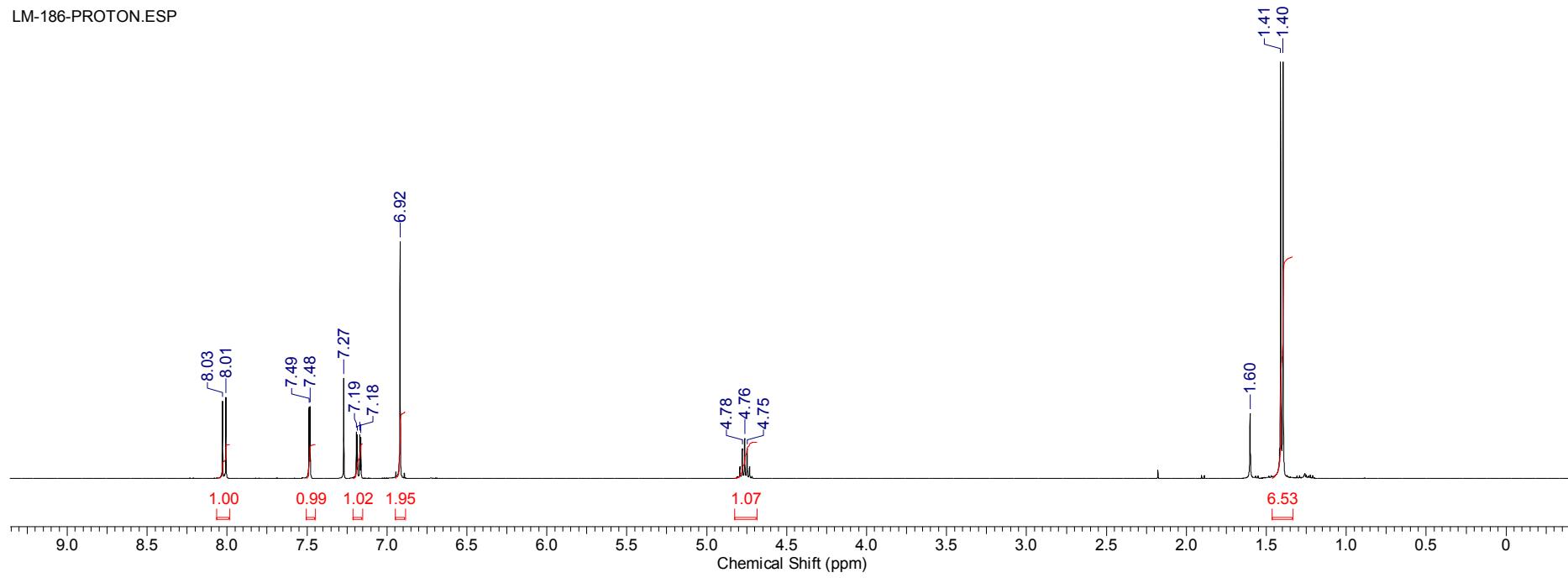


¹³C NMR at 100 MHz in CDCl₃

6-Isopropoxy-1,4-naphthoquinone 12d

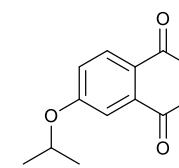


LM-186-PROTON.ESP

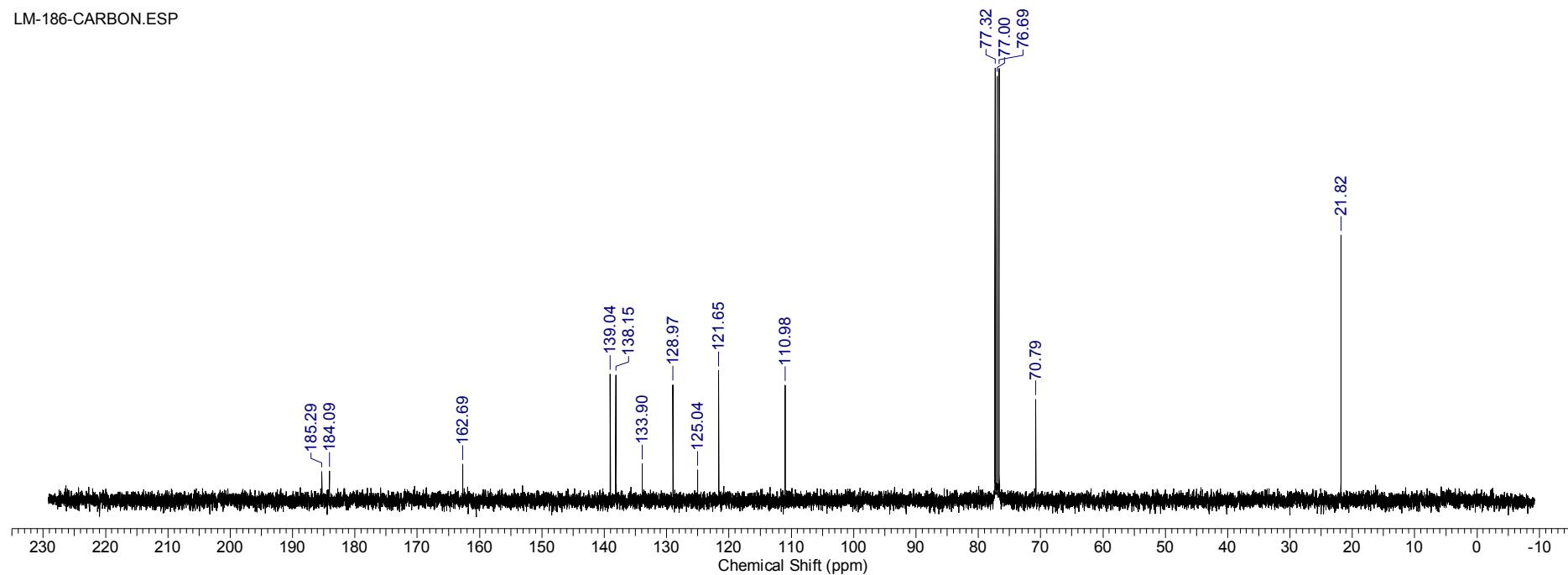


^1H NMR at 400 MHz in CDCl_3

6-Isopropoxy-1,4-naphthoquinone 12d

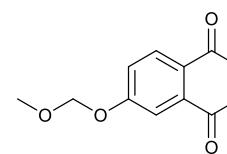


LM-186-CARBON.ESP

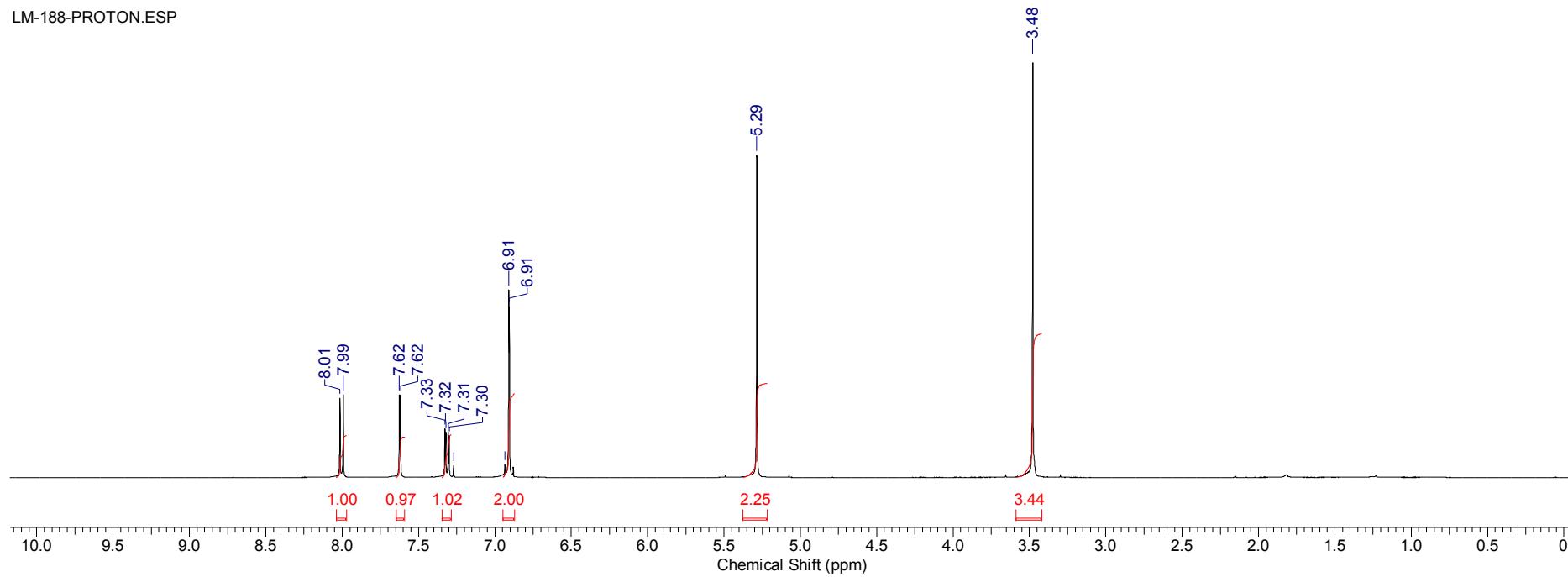


^{13}C NMR at 100 MHz in CDCl_3

6-(Methoxymethoxy)-1,4-naphthoquinone 12e

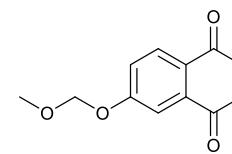


LM-188-PROTON.ESP

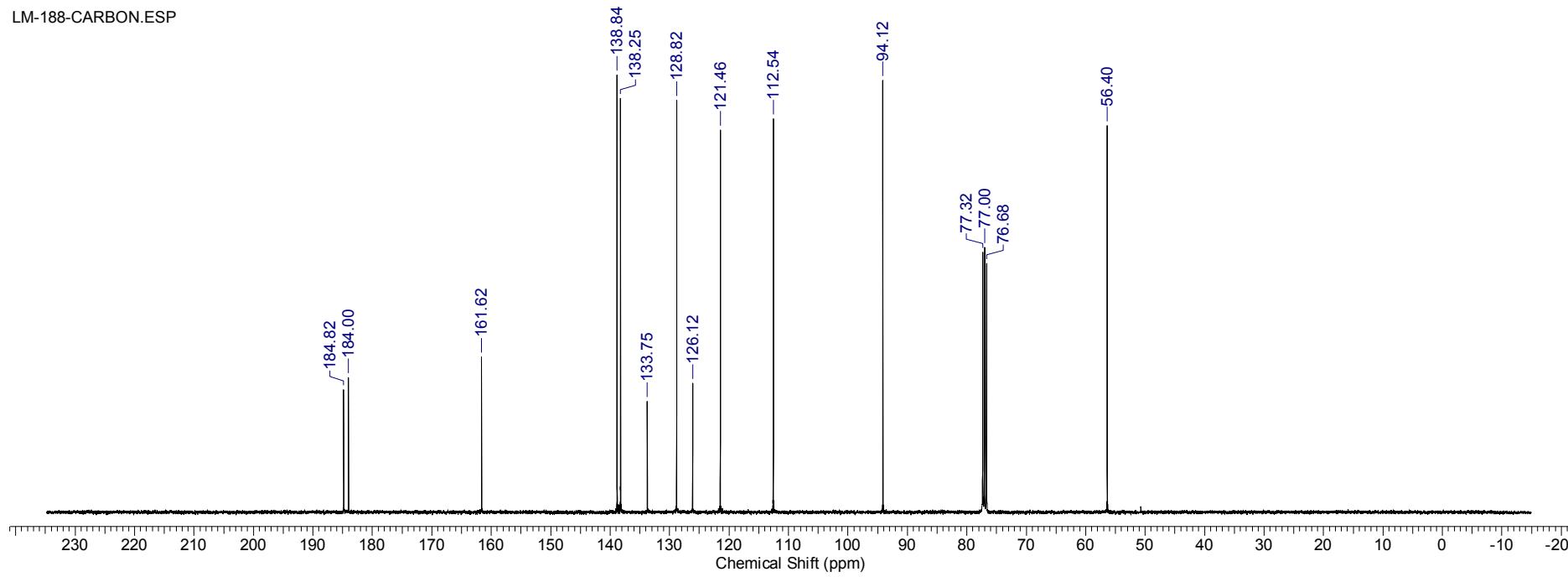


^1H NMR at 400 MHz in CDCl_3

6-(Methoxymethoxy)-1,4-naphthoquinone 12e

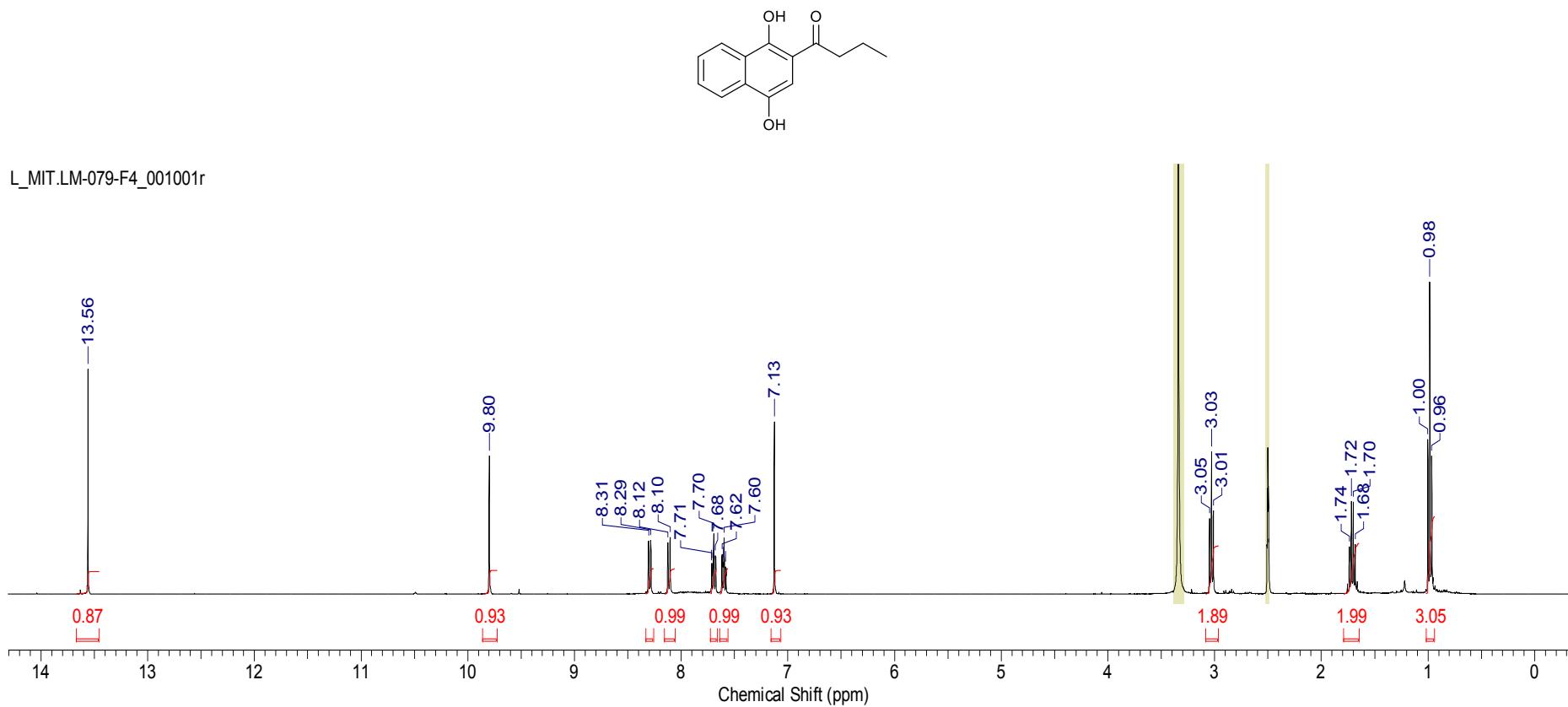


LM-188-CARBON.ESP



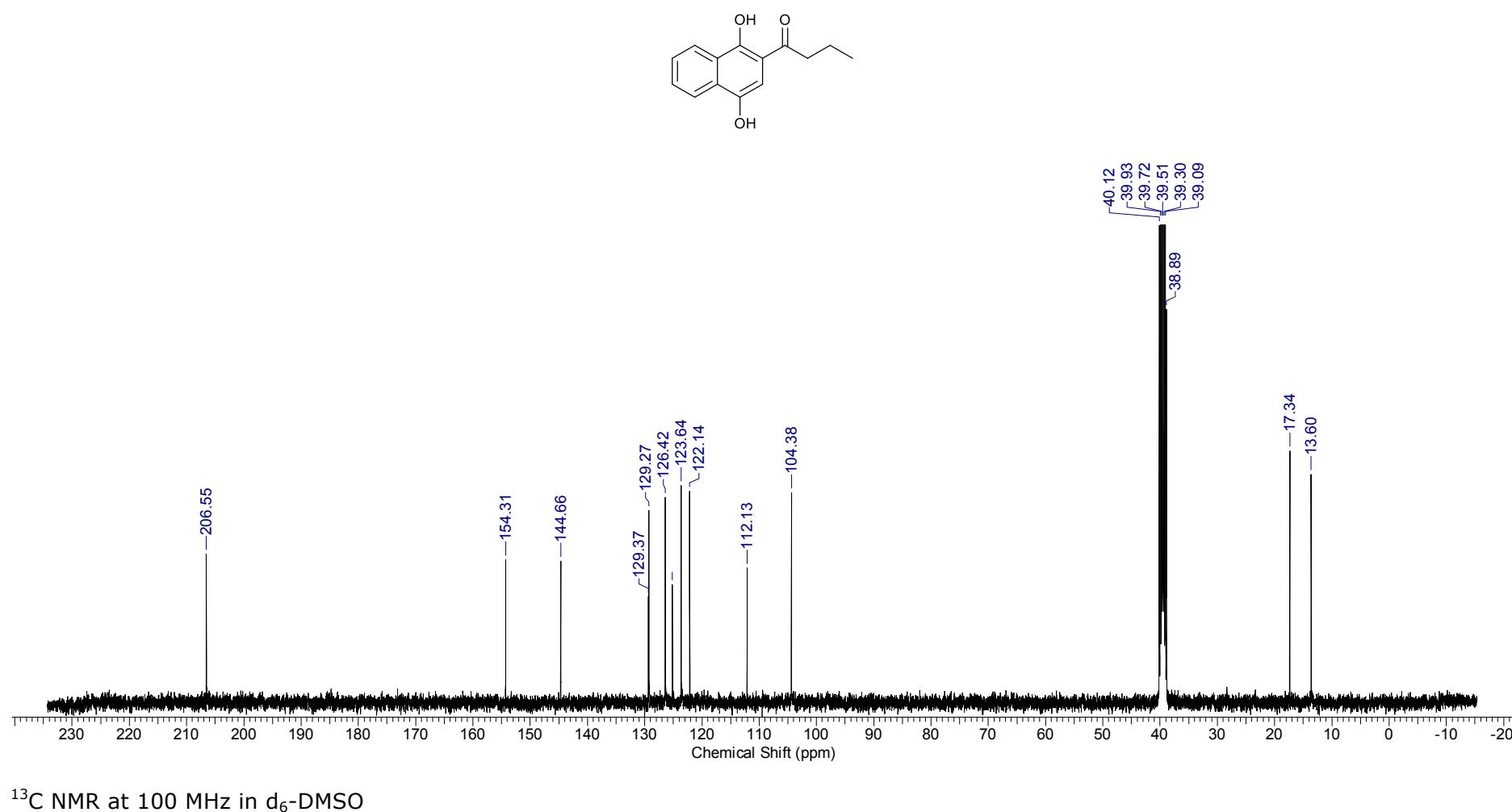
^{13}C NMR at 100 MHz in CDCl_3

1-(1,4-Dihydroxy-2-naphthyl)-1-butanone 6a

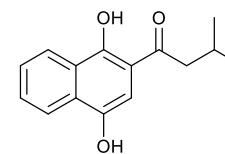


^1H NMR at 400 MHz in $\text{d}_6\text{-DMSO}$

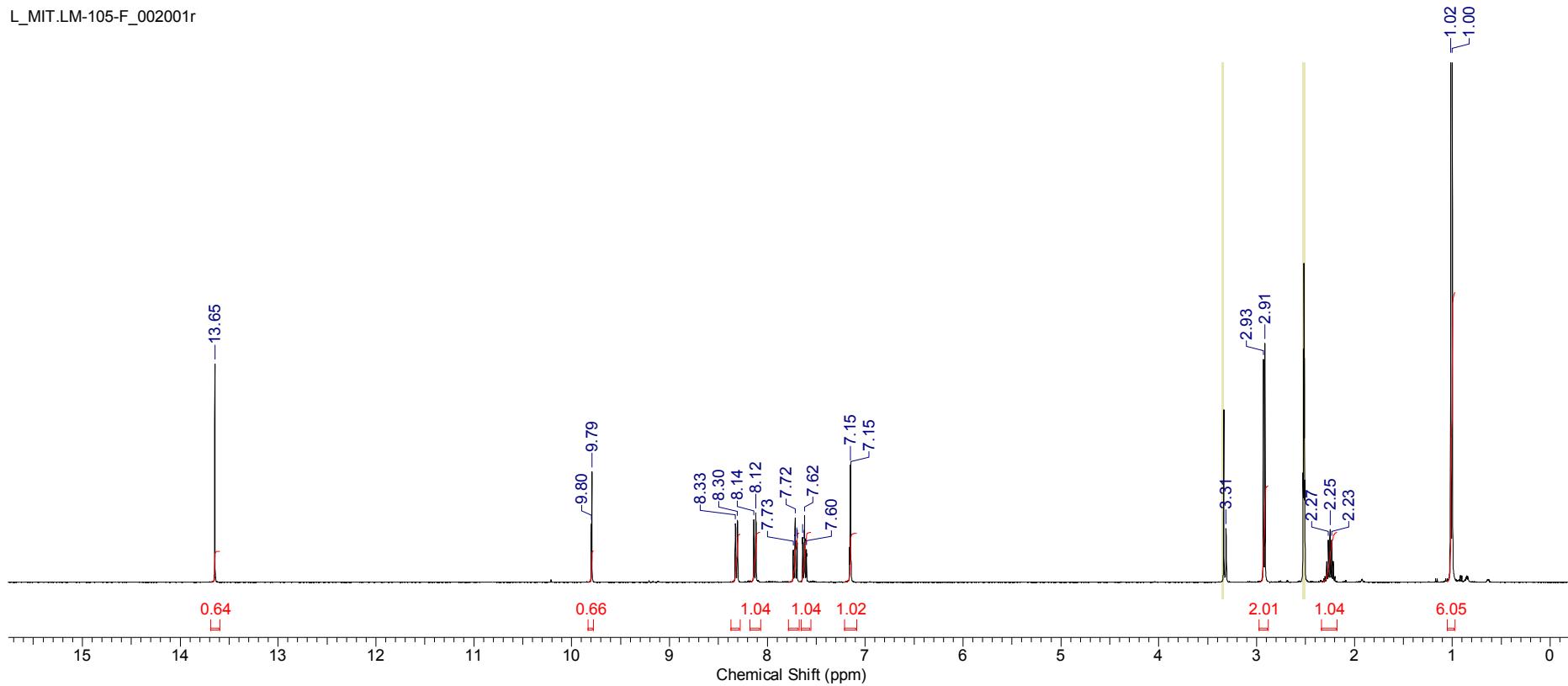
1-(1,4-Dihydroxy-2-naphthyl)-1-butanone 6a



1-(1,4-Dihydroxy-2-naphthyl)-2-methyl-1-propanone 6b

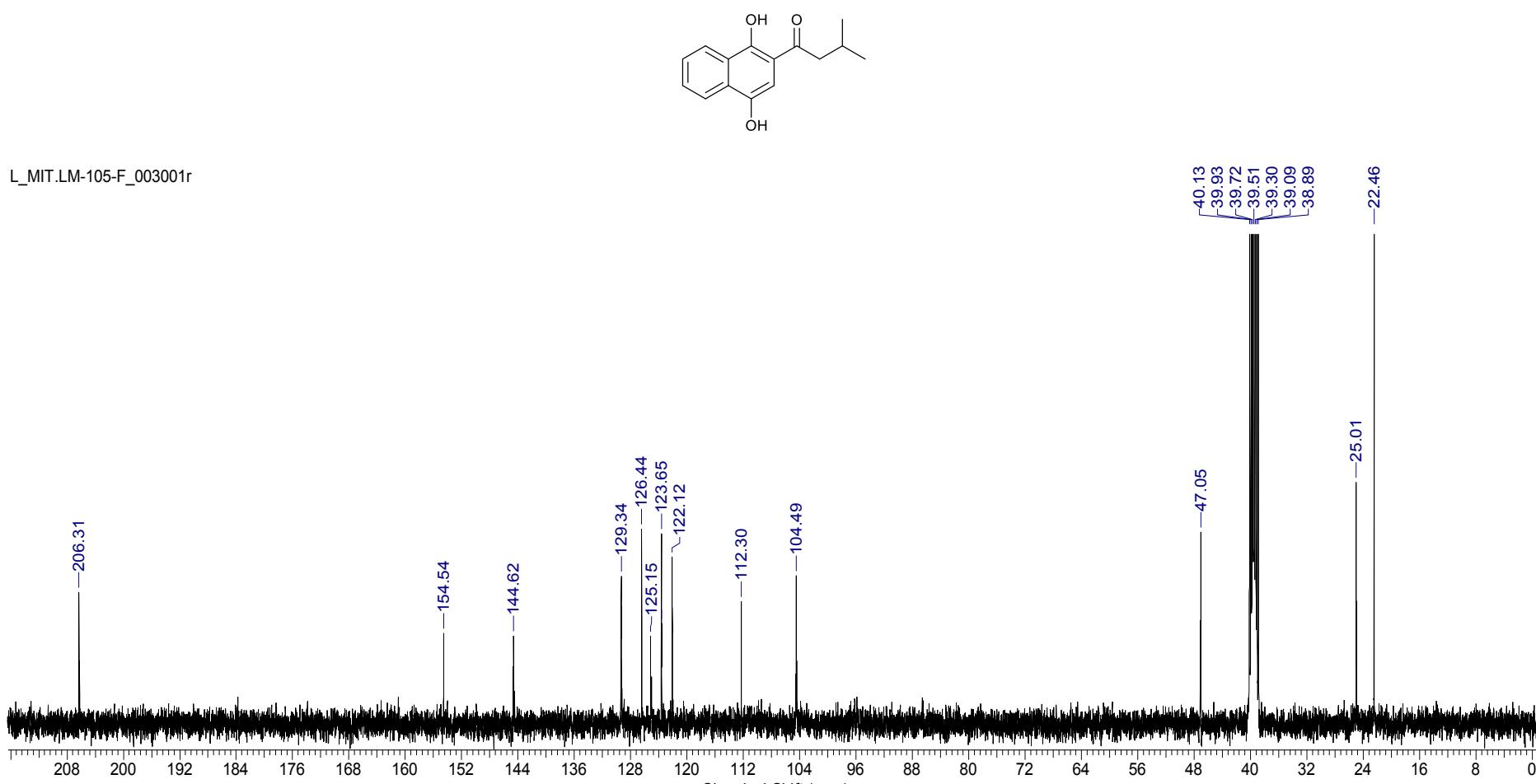


L/MIT.LM-105-F_002001r



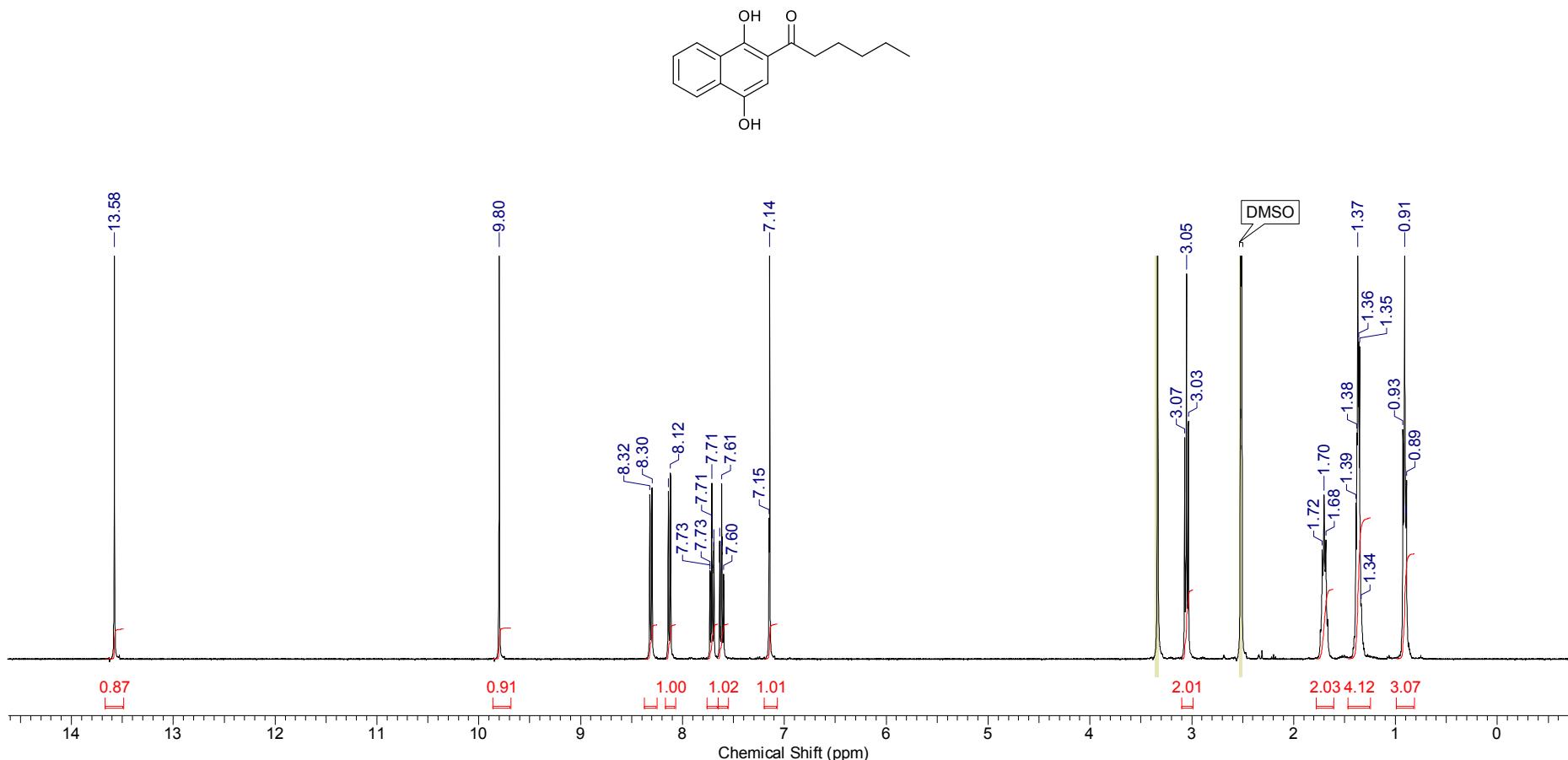
^1H NMR at 400 MHz in $\text{d}_6\text{-DMSO}$

1-(1,4-Dihydroxy-2-naphthyl)-2-methyl-1-propanone 6b



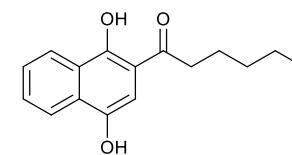
¹³C NMR at 100 MHz in d₆-DMSO

1-(1,4-Dihydroxy-2-naphthyl)-1-hexanone 6c

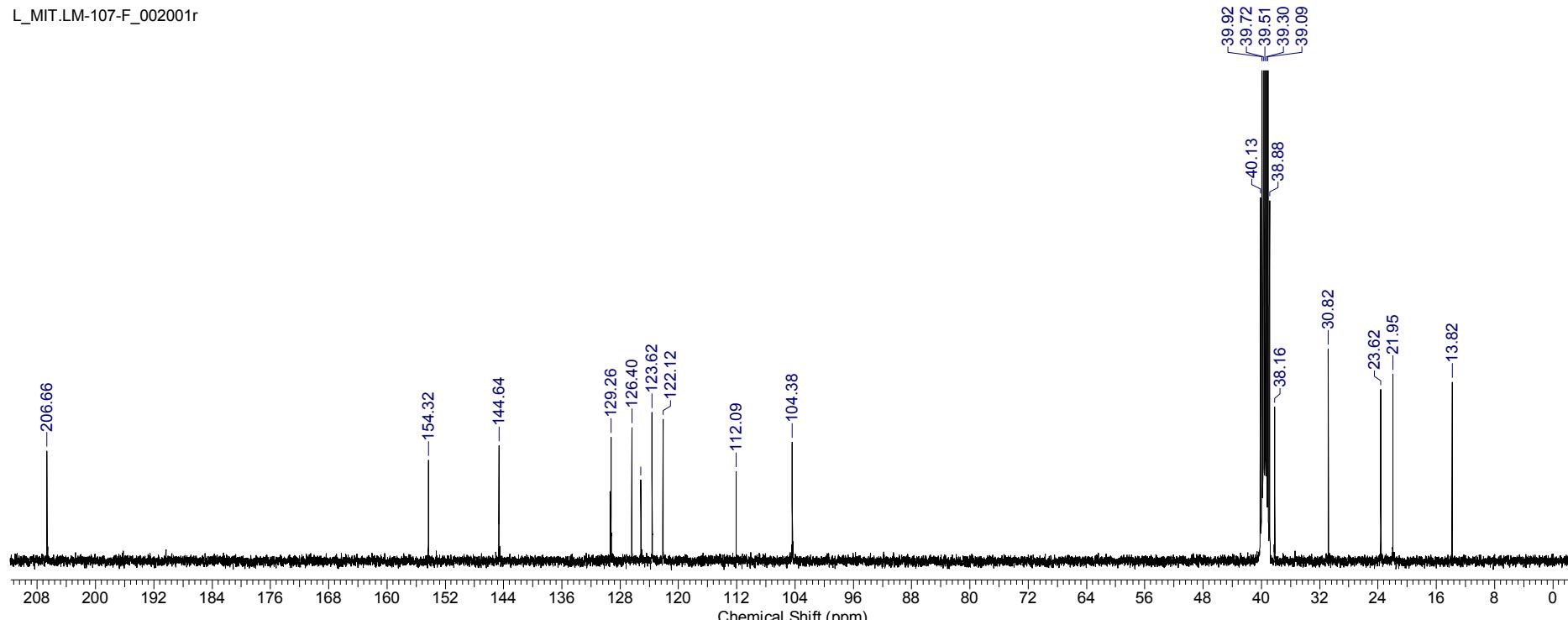


^1H NMR at 400 MHz in $\text{d}_6\text{-DMSO}$

1-(1,4-Dihydroxy-2-naphthyl)-1-hexanone **6c**

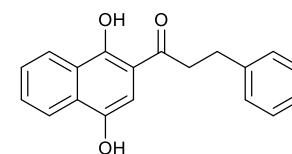


L/MIT.LM-107-F_002001r

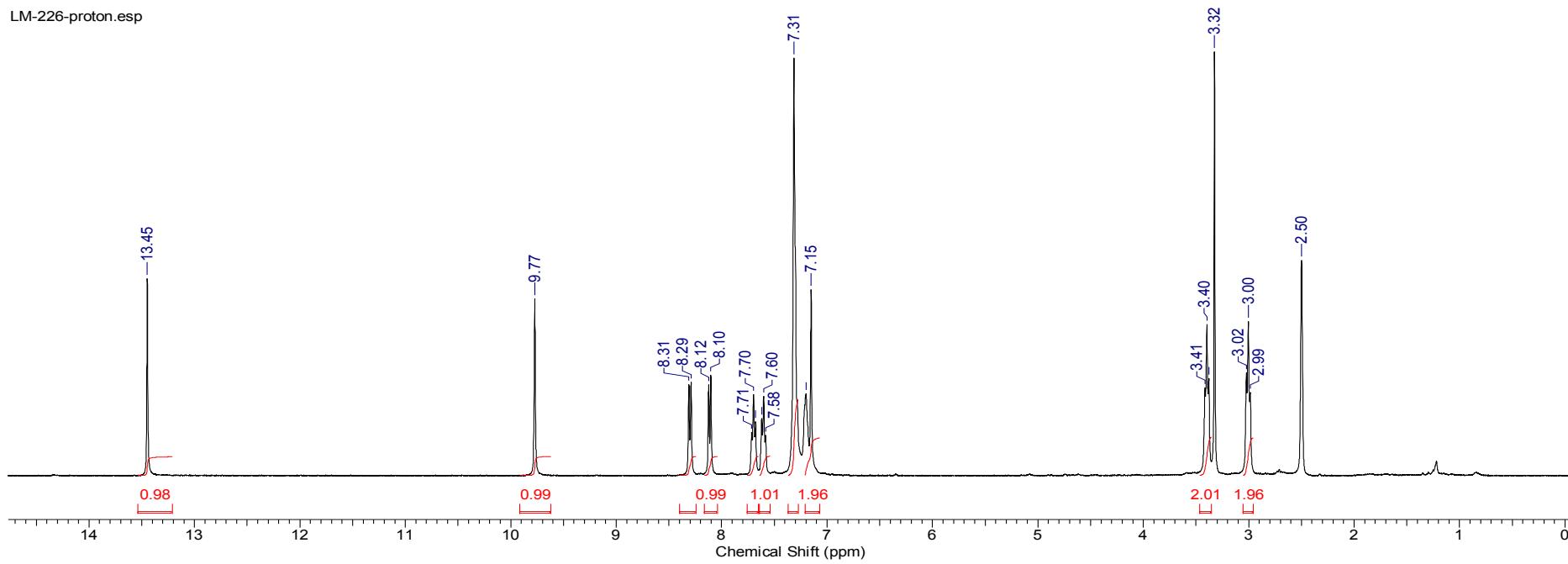


^{13}C NMR at 100 MHz in $\text{d}_6\text{-DMSO}$

1-(1,4-dihydroxynaphthalen-2-yl)-3-phenylpropan-1-one 6d

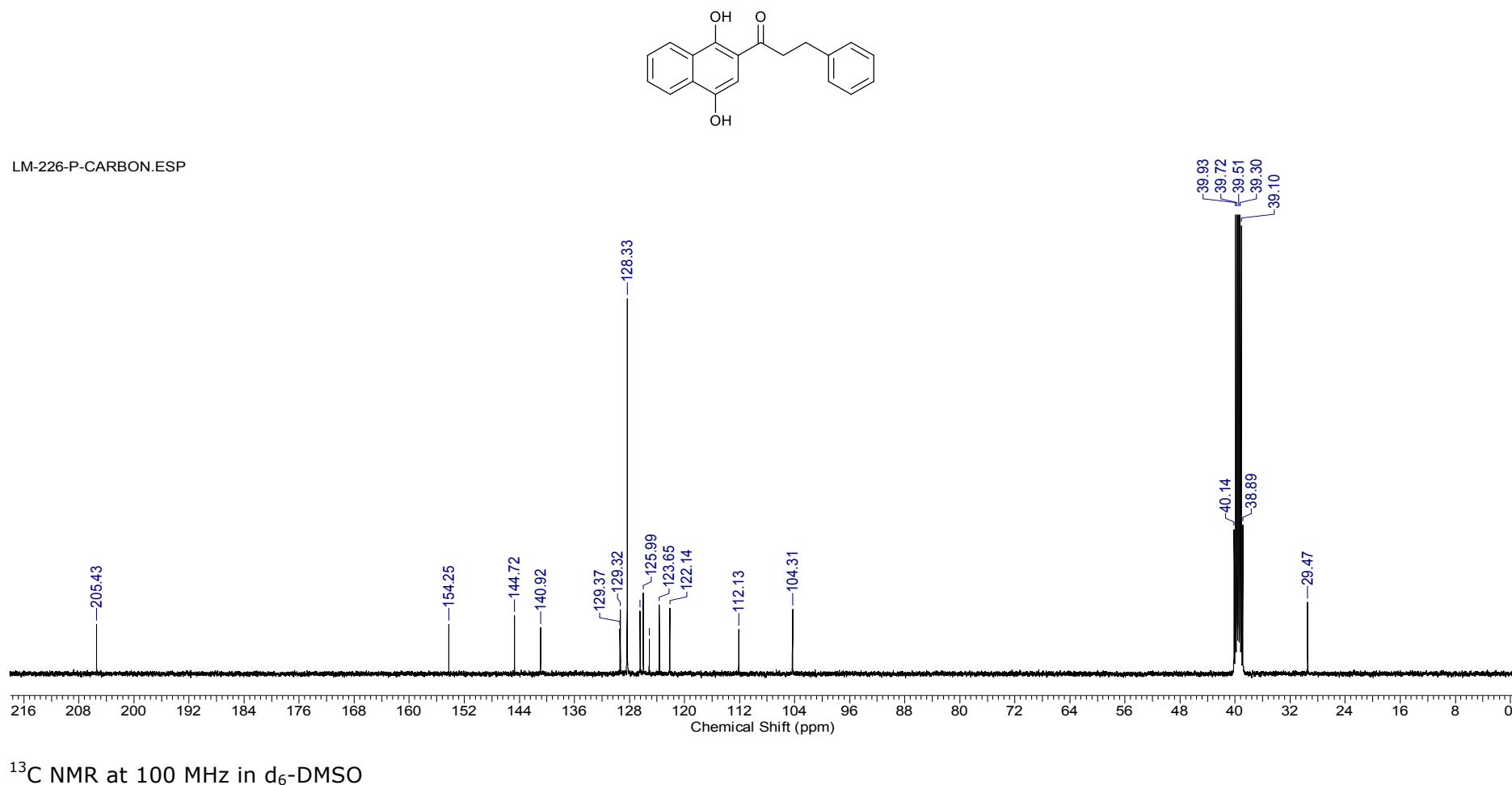


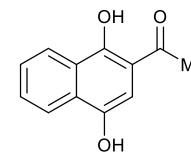
LM-226-proton.esp



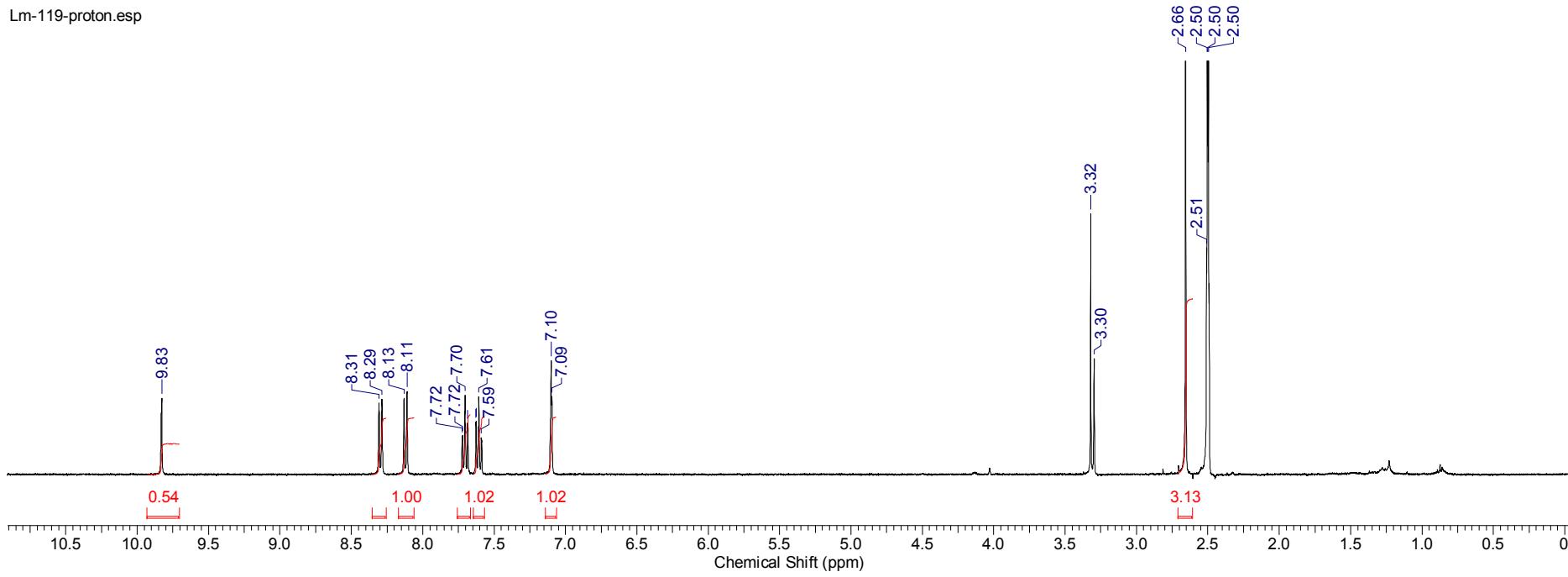
¹H NMR at 400 MHz in d₆-DMSO

1-(1,4-dihydroxynaphthalen-2-yl)-3-phenylpropan-1-one 6d



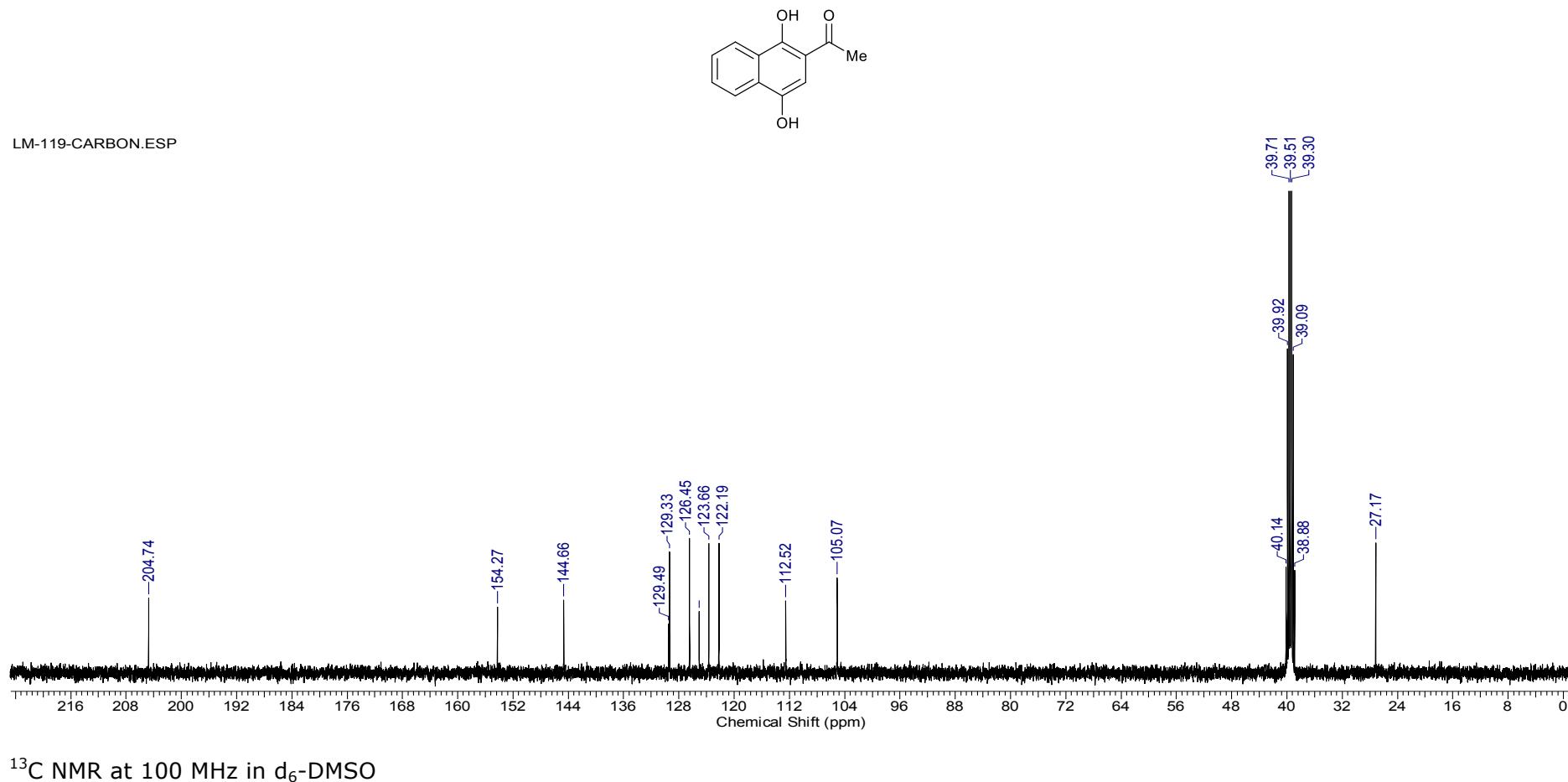


Lm-119-proton.esp

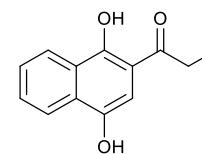


^1H NMR at 400 MHz in $\text{d}_6\text{-DMSO}$

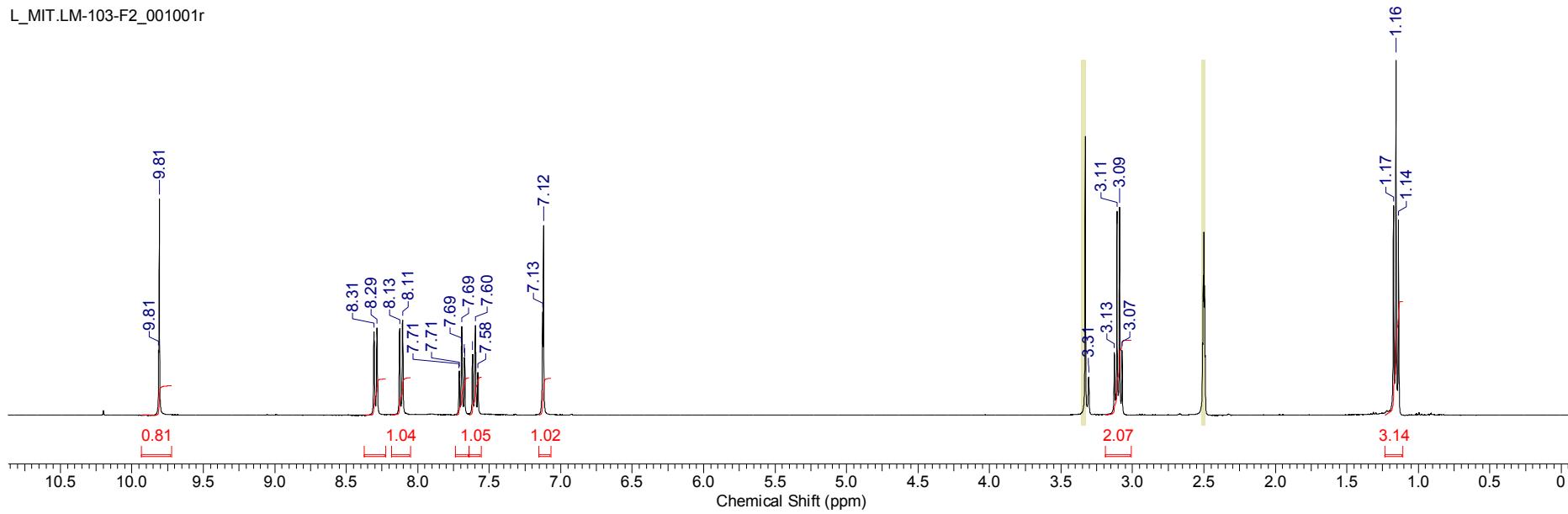
1-(1,4-Dihydroxy-2-naphthyl)-1-ethanone 6e



1-(1,4-Dihydroxy-2-naphthyl)-1-propanone 6f

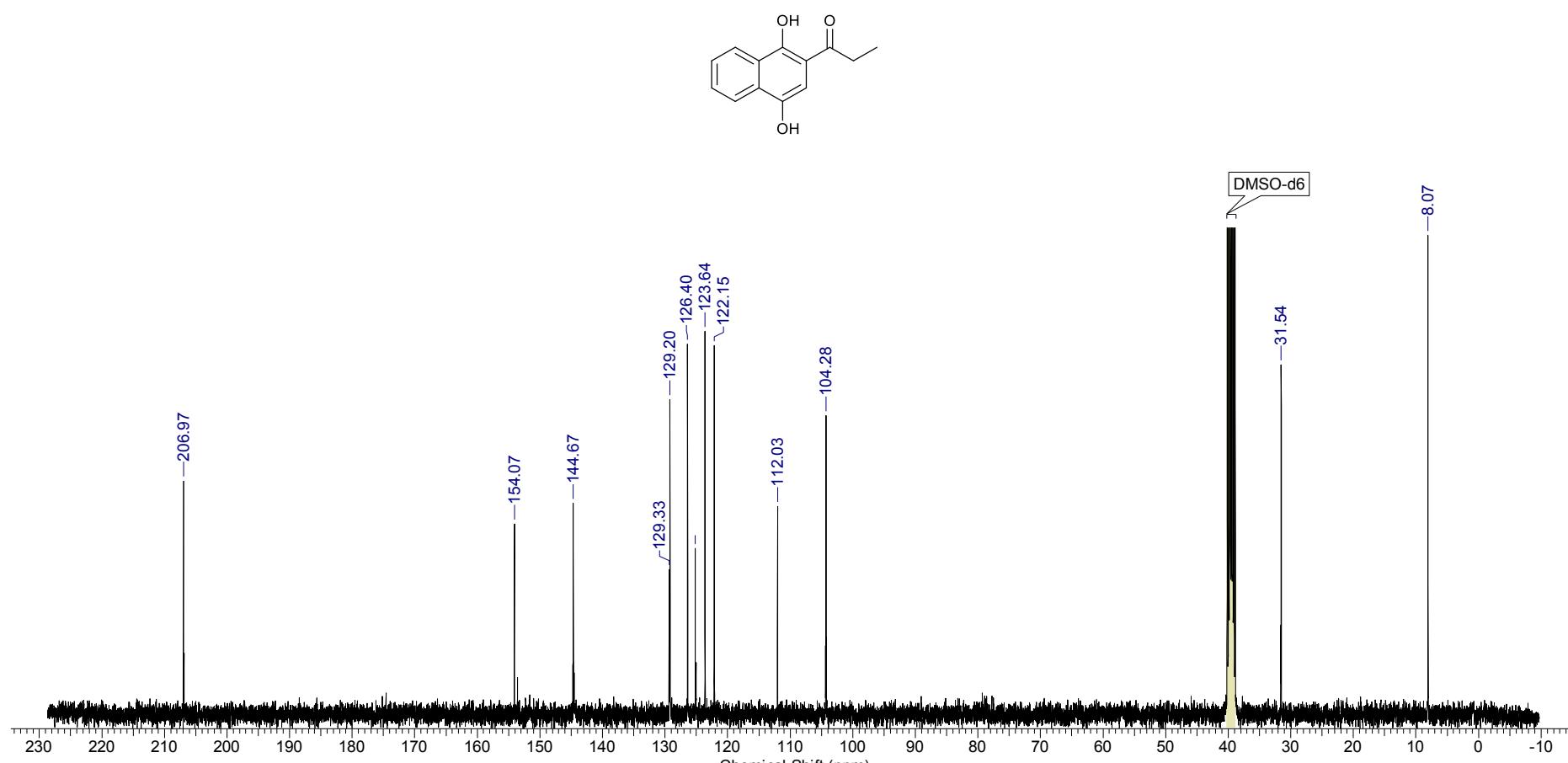


L/MIT.LM-103-F2_001001r



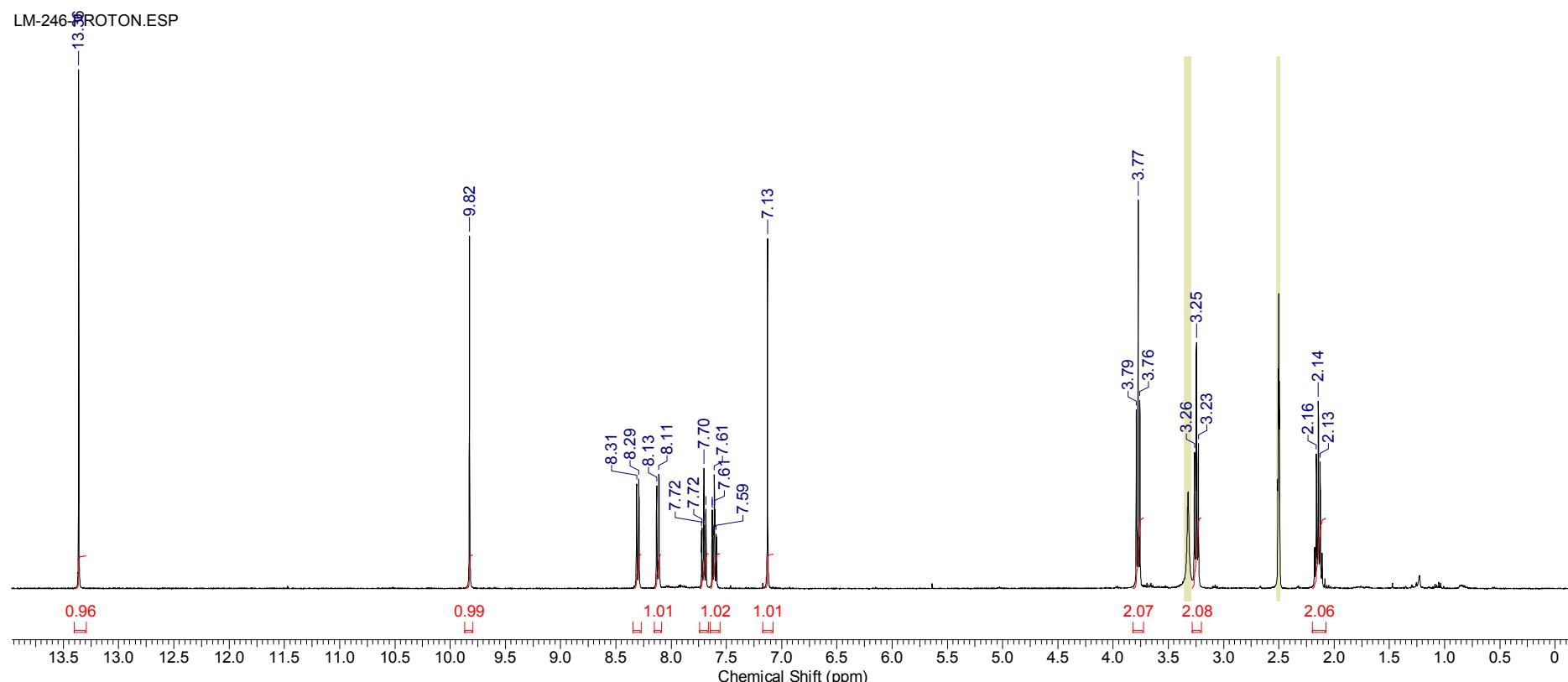
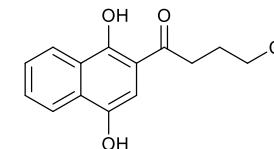
^1H NMR at 400 MHz in $\text{d}_6\text{-DMSO}$

1-(1,4-Dihydroxy-2-naphthyl)-1-propanone 6f



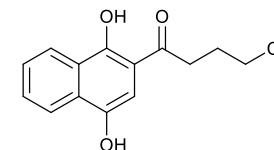
^{13}C NMR at 100 MHz in $\text{d}_6\text{-DMSO}$

4-Chloro-1-(1,4-dihydroxynaphthalen-2-yl)butan-1-one 6g

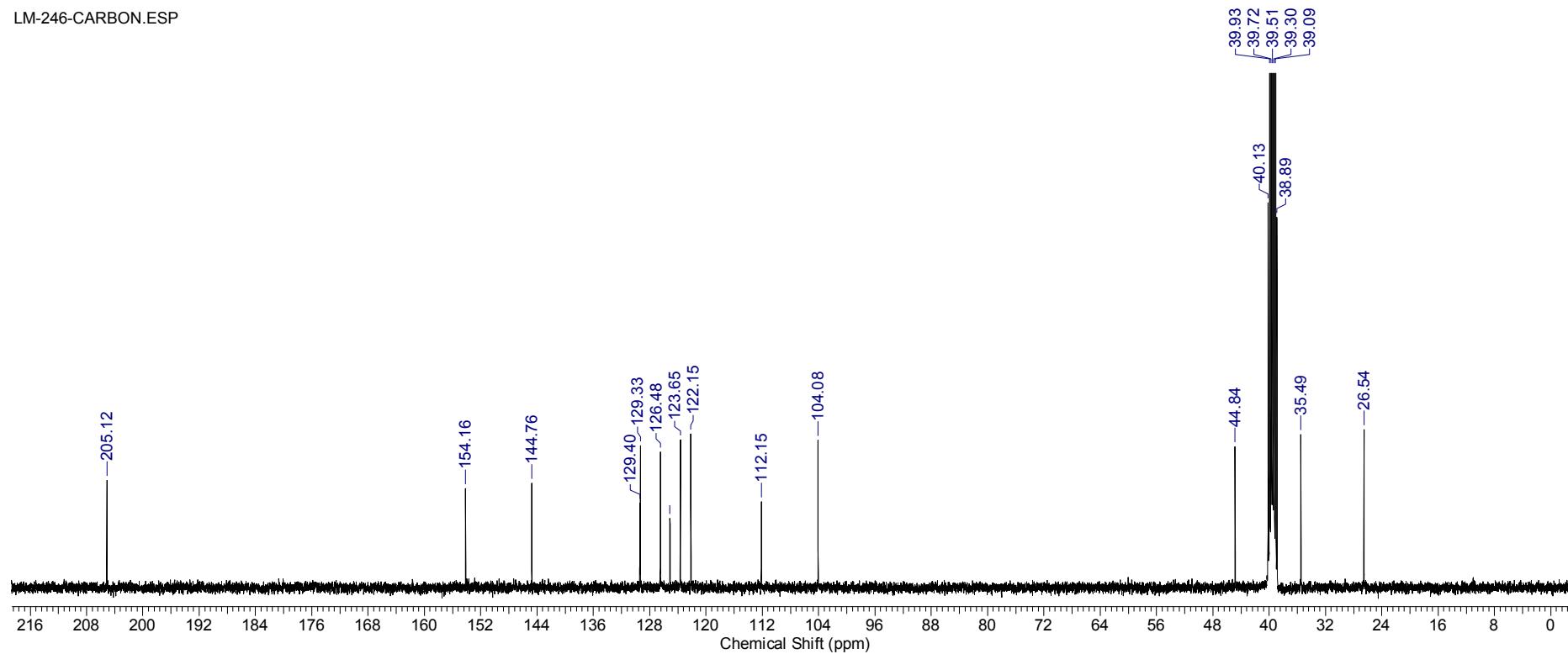


¹H NMR at 400 MHz in d₆-DMSO

4-Chloro-1-(1,4-dihydroxynaphthalen-2-yl)butan-1-one 6g

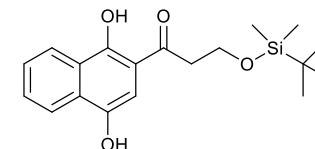


LM-246-CARBON.ESP

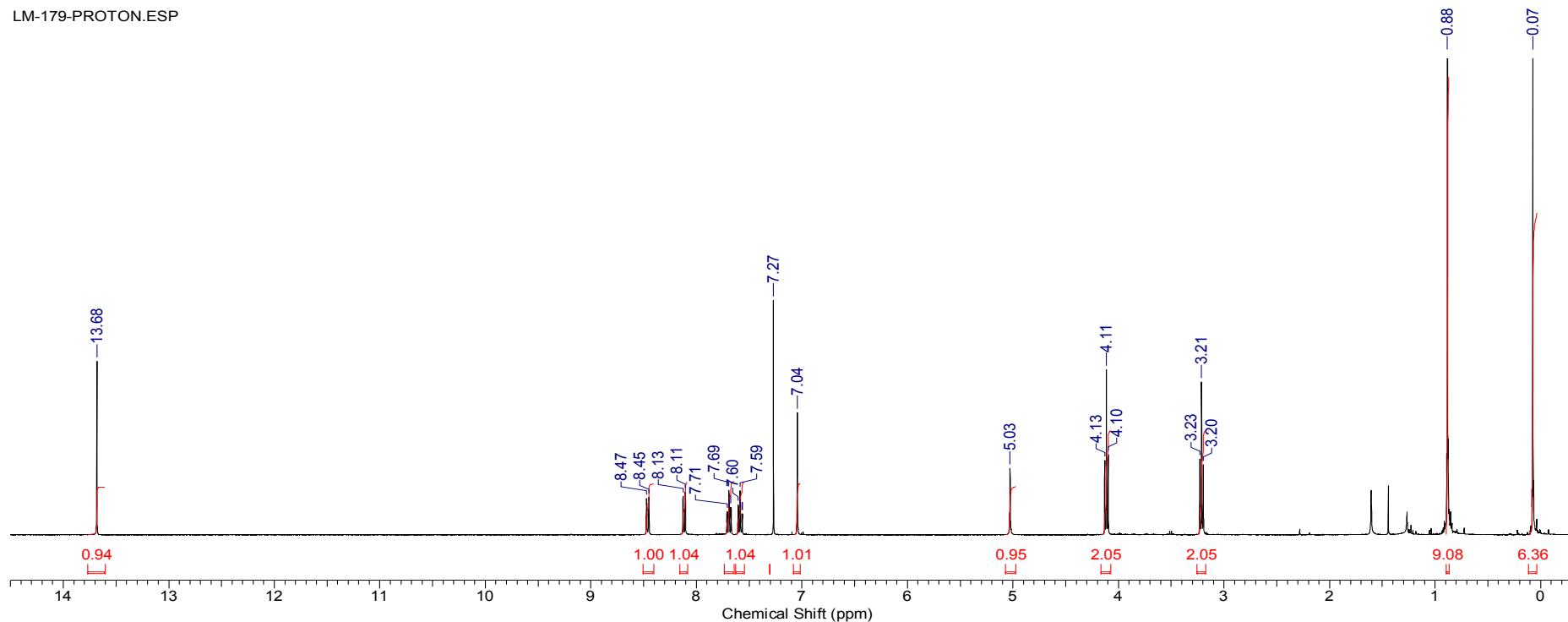


¹³C NMR at 100 MHz in d₆-DMSO

3-((*t*-Butyldimethylsilyl)oxy)-1-(1,4-dihydroxynaphthalen-2-yl)1-propanone 6h

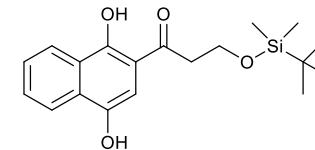


LM-179-PROTON.ESP

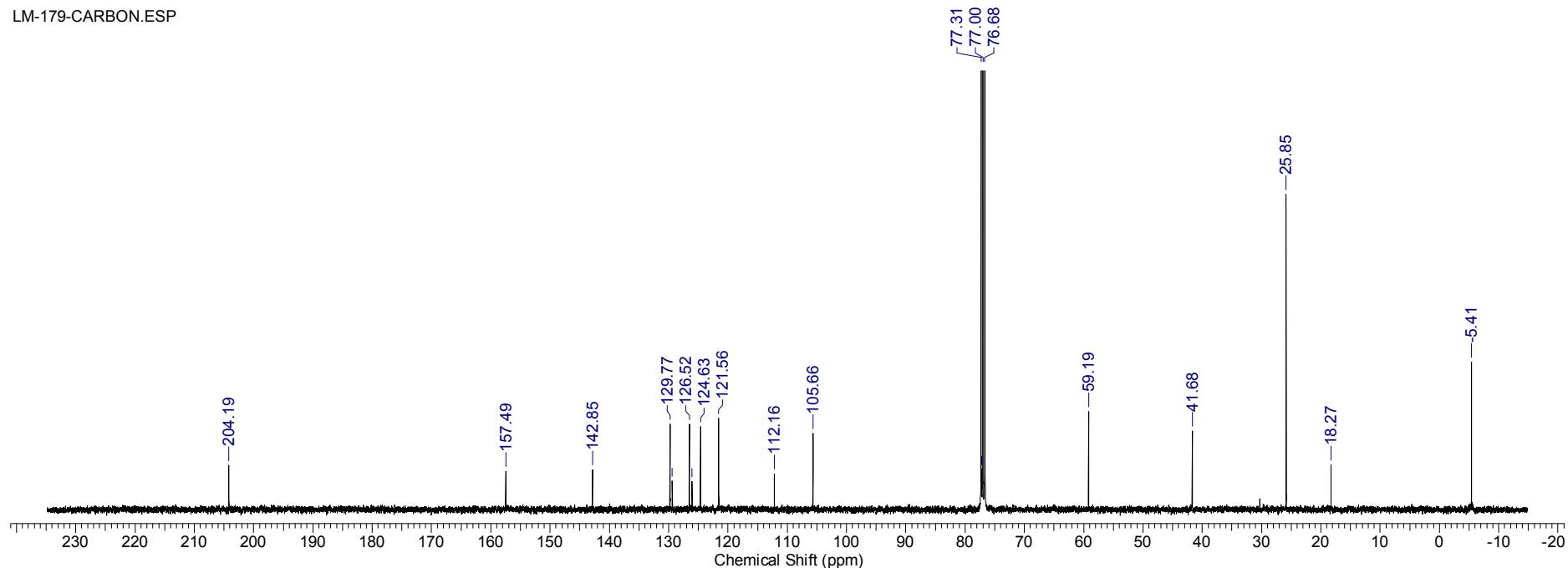


¹H NMR at 400 MHz in CDCl₃

3-((*t*-Butyldimethylsilyl)oxy)-1-(1,4-dihydroxynaphthalen-2-yl)1-propanone 6h

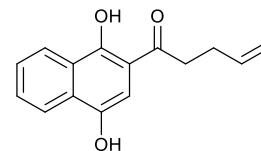


LM-179-CARBON.ESP

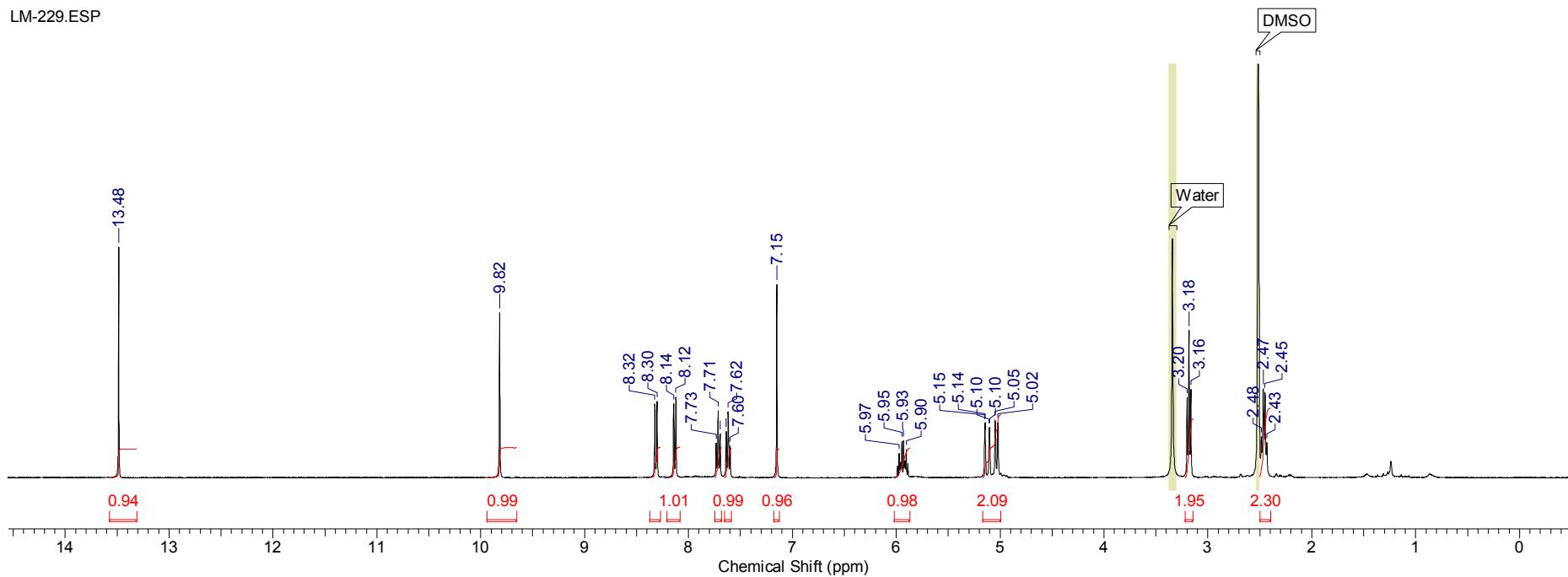


¹³C NMR at 100 MHz in CDCl₃

1-(1,4-Dihydroxynaphthalen-2-yl)pent-4-en-1-one 6i

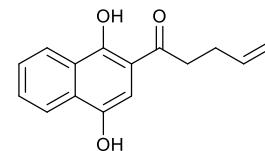


LM-229.ESP

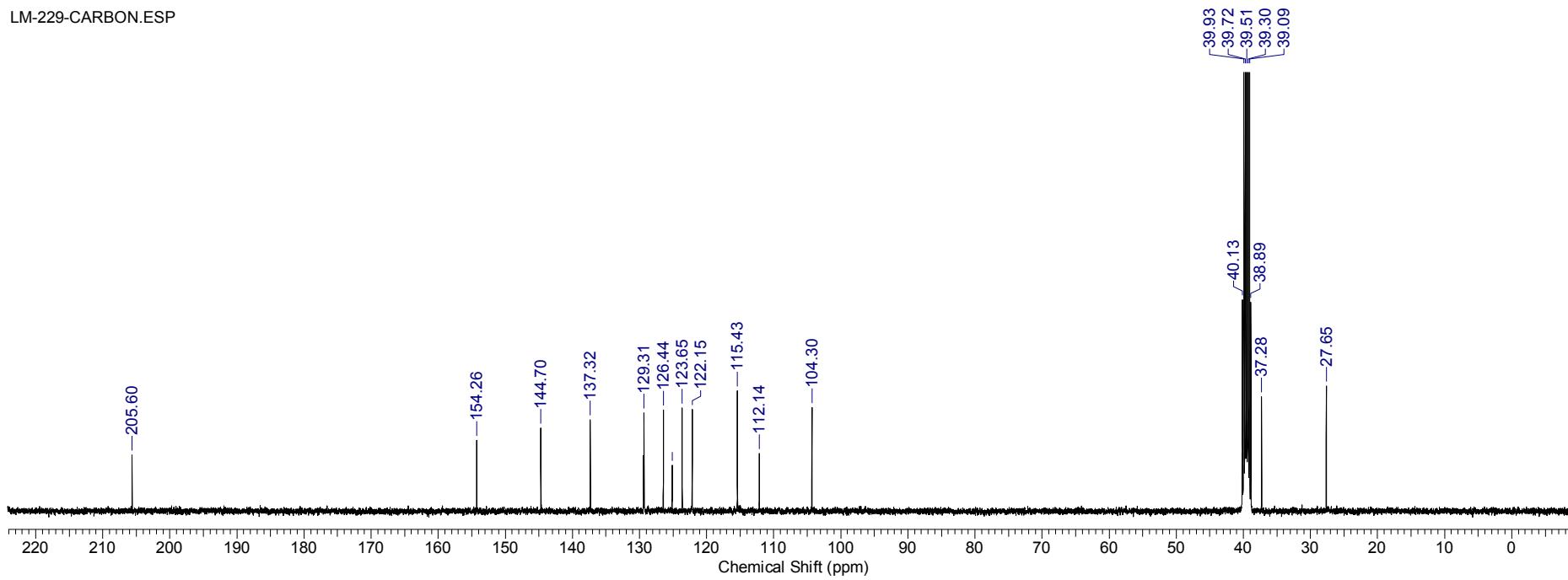


¹H NMR at 400 MHz in d₆-DMSO

1-(1,4-Dihydroxynaphthalen-2-yl)pent-4-en-1-one 6i

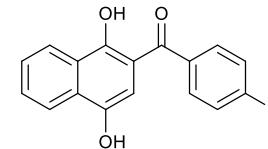


LM-229-CARBON.ESP

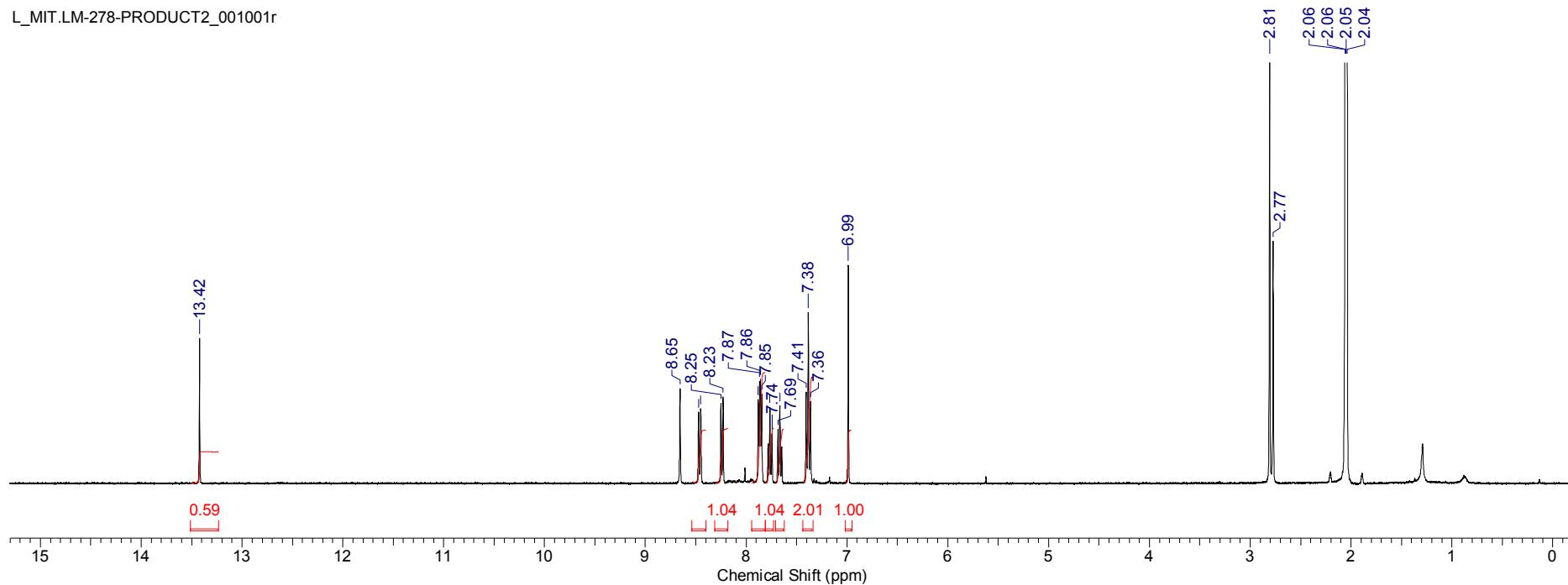


^{13}C NMR at 100 MHz in d_6 -DMSO

(1,4-dihydroxynaphthalen-2-yl)(4-fluorophenyl)methanone 6j

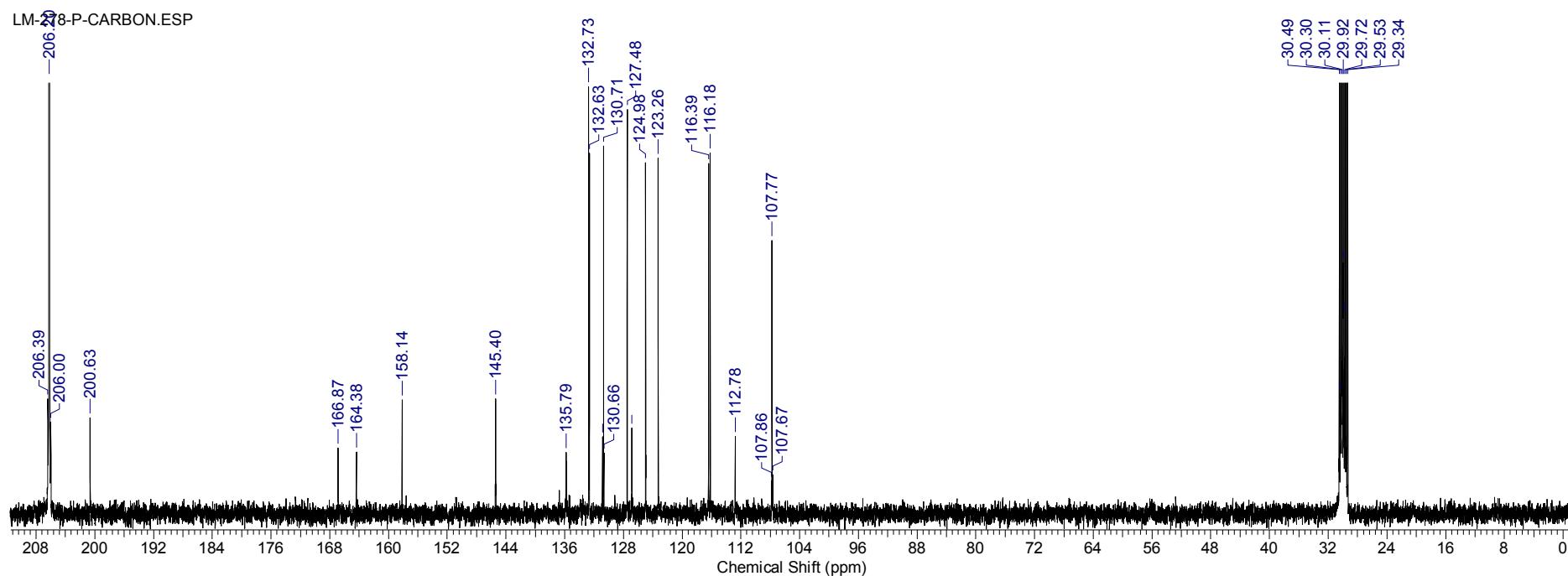
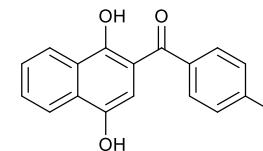


L/MIT.LM-278-PRODUCT2_001001r



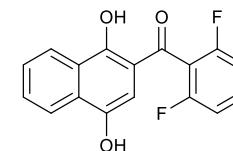
¹H NMR at 400 MHz in d₆-acetone

(1,4-dihydroxynaphthalen-2-yl)(4-fluorophenyl)methanone 6j

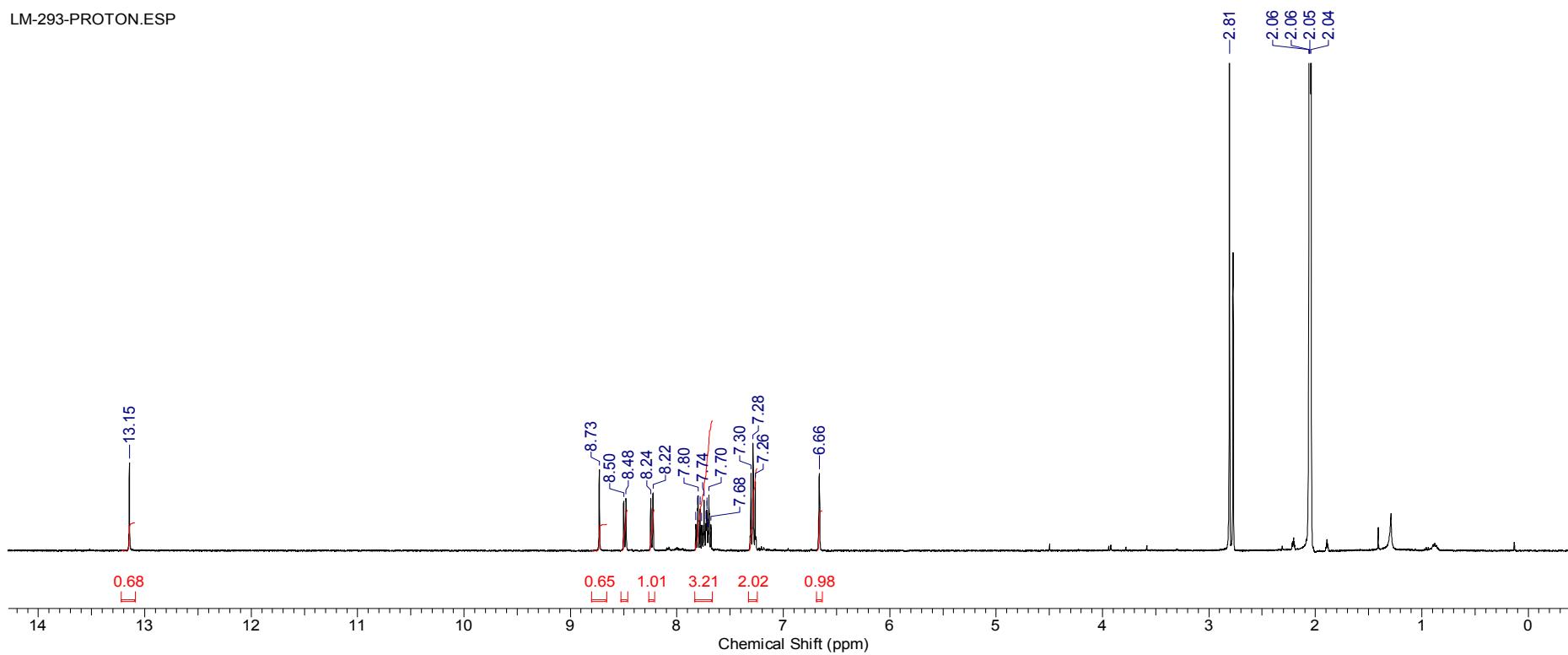


^{13}C NMR at 100 MHz in d_6 -acetone

(2,6-difluorophenyl)(1,4-dihydroxynaphthalen-2-yl)methanone 6k

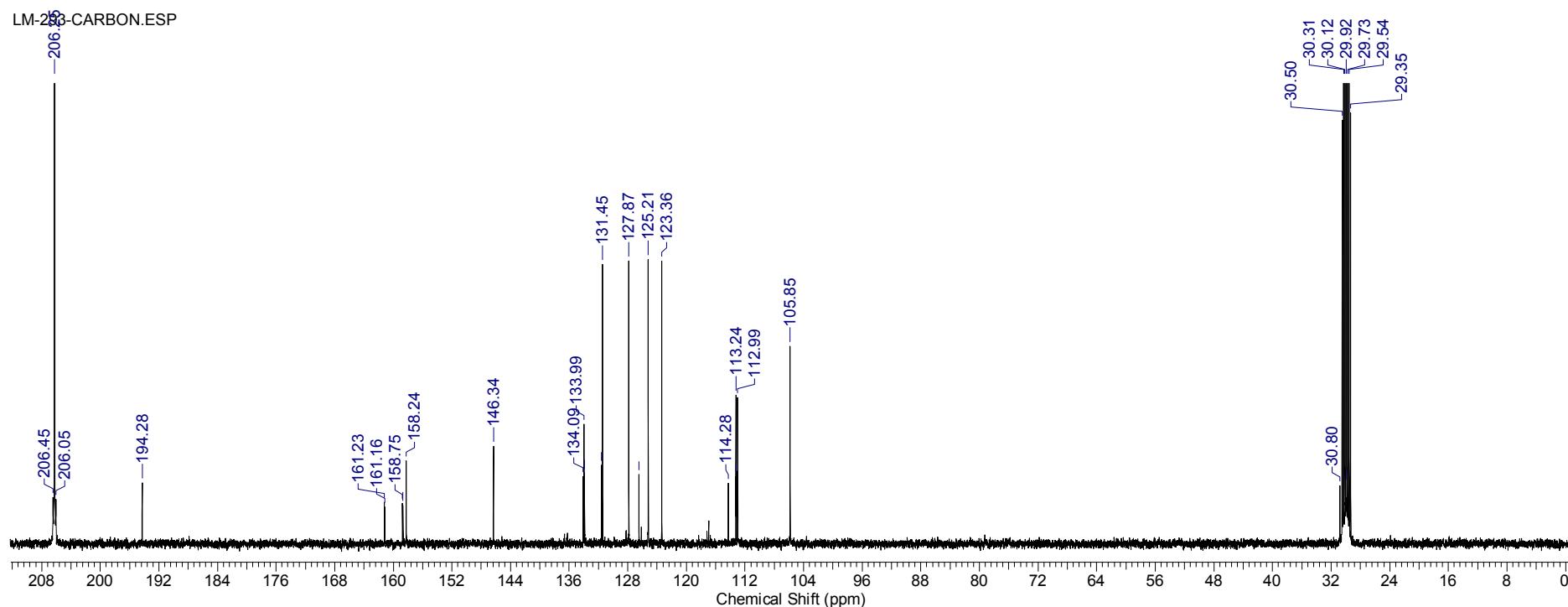
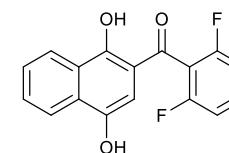


LM-293-PROTON.ESP



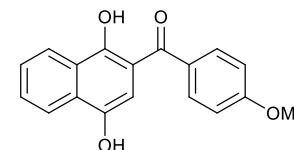
¹H NMR at 400 MHz in d₆-acetone

(2,6-difluorophenyl)(1,4-dihydroxynaphthalen-2-yl)methanone 6k

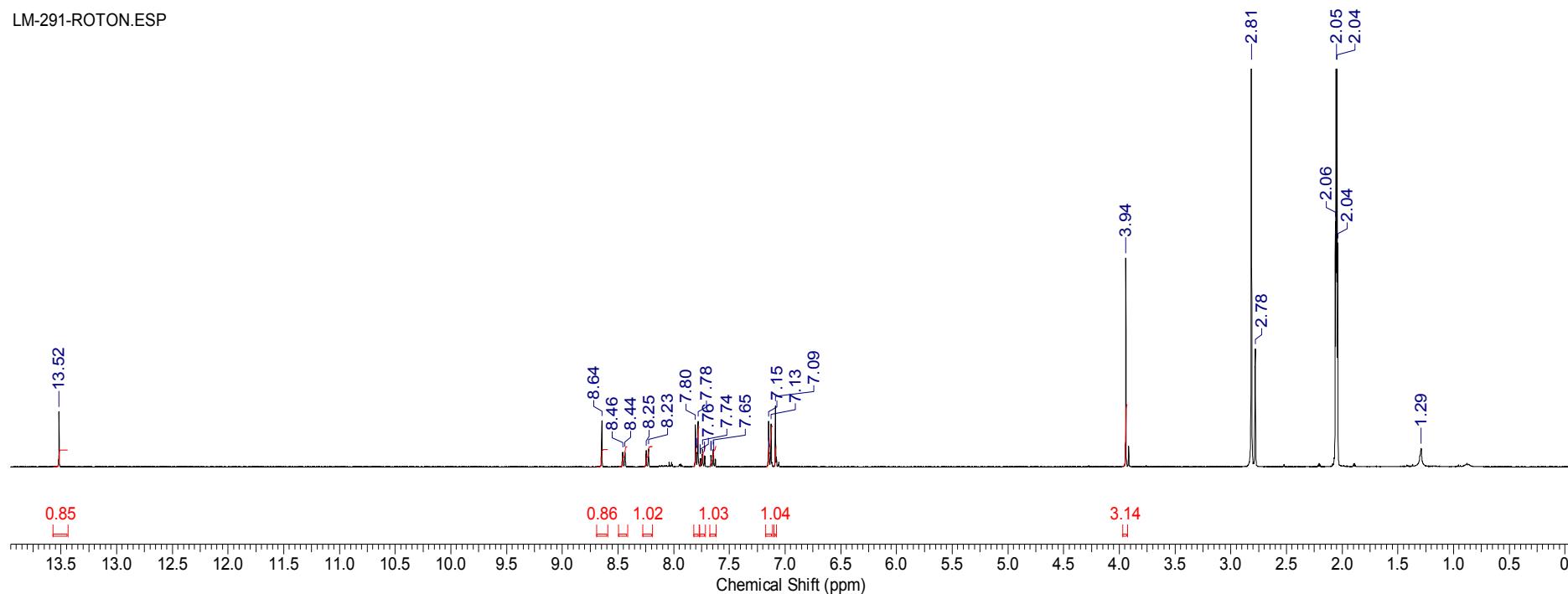


^{13}C NMR at 100 MHz in d_6 -acetone

(1,4-dihydroxynaphthalen-2-yl)(4-methoxyphenyl)methanone 6l

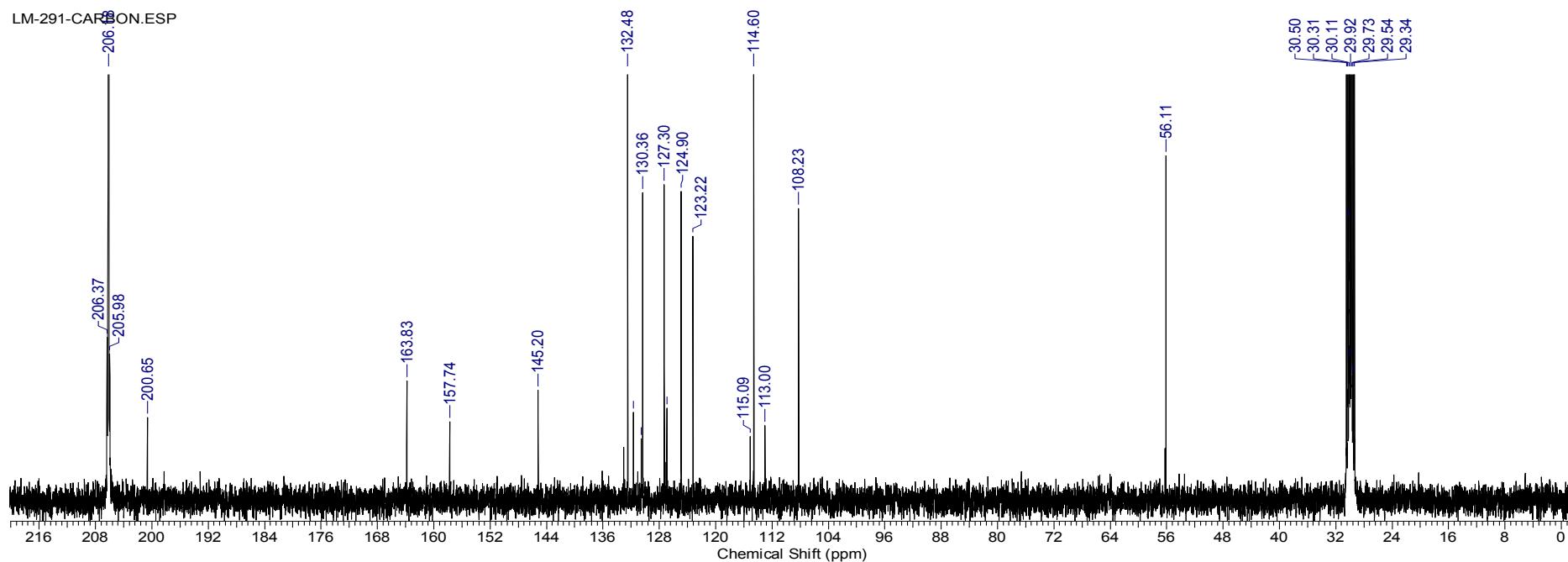
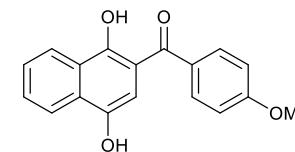


LM-291-ROTON.ESP



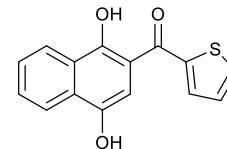
¹H NMR at 400 MHz in d₆-acetone

(1,4-dihydroxynaphthalen-2-yl)(4-methoxyphenyl)methanone 6l

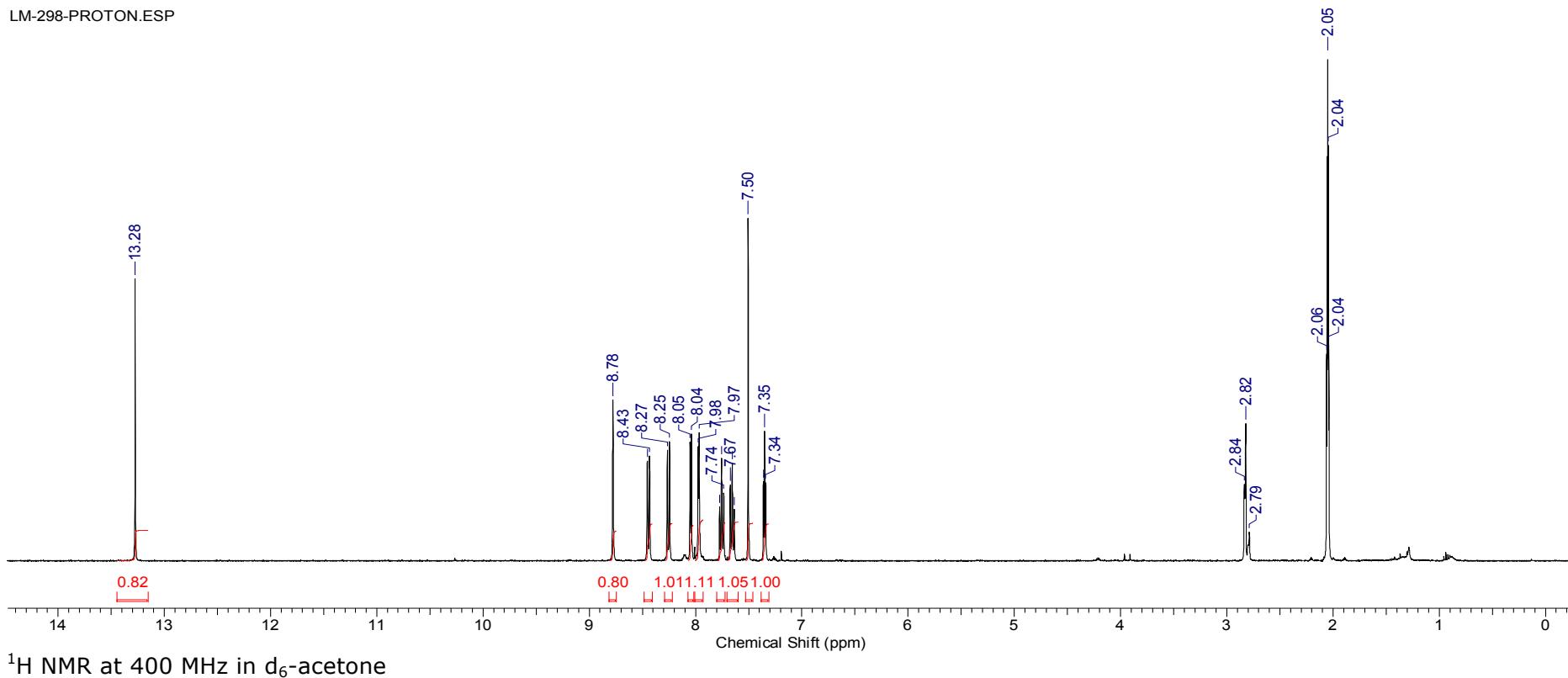


¹³C NMR at 100 MHz in d₆-acetone

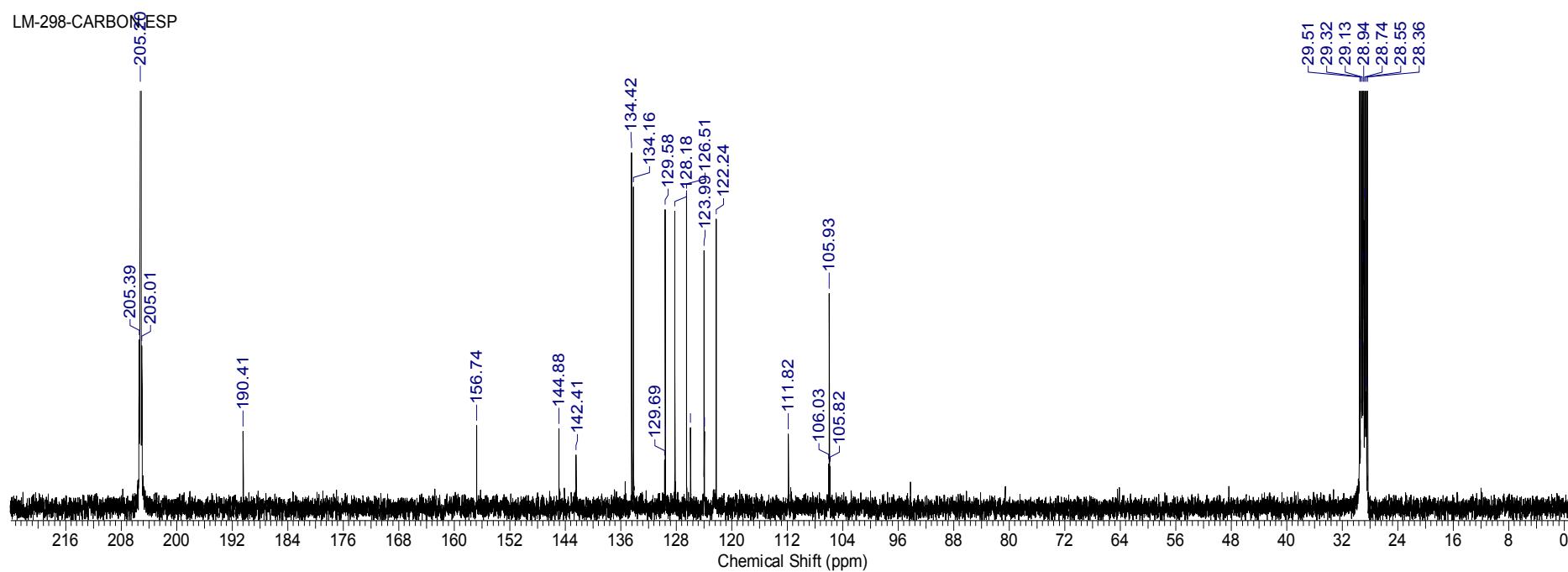
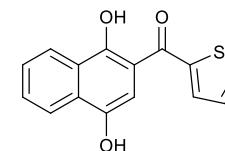
(1,4-dihydroxynaphthalen-2-yl)(thiophen-2-yl)methanone 6m



LM-298-PROTON.ESP

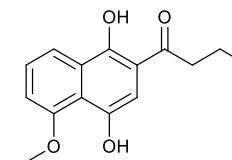


(1,4-dihydroxynaphthalen-2-yl)(thiophen-2-yl)methanone 6m

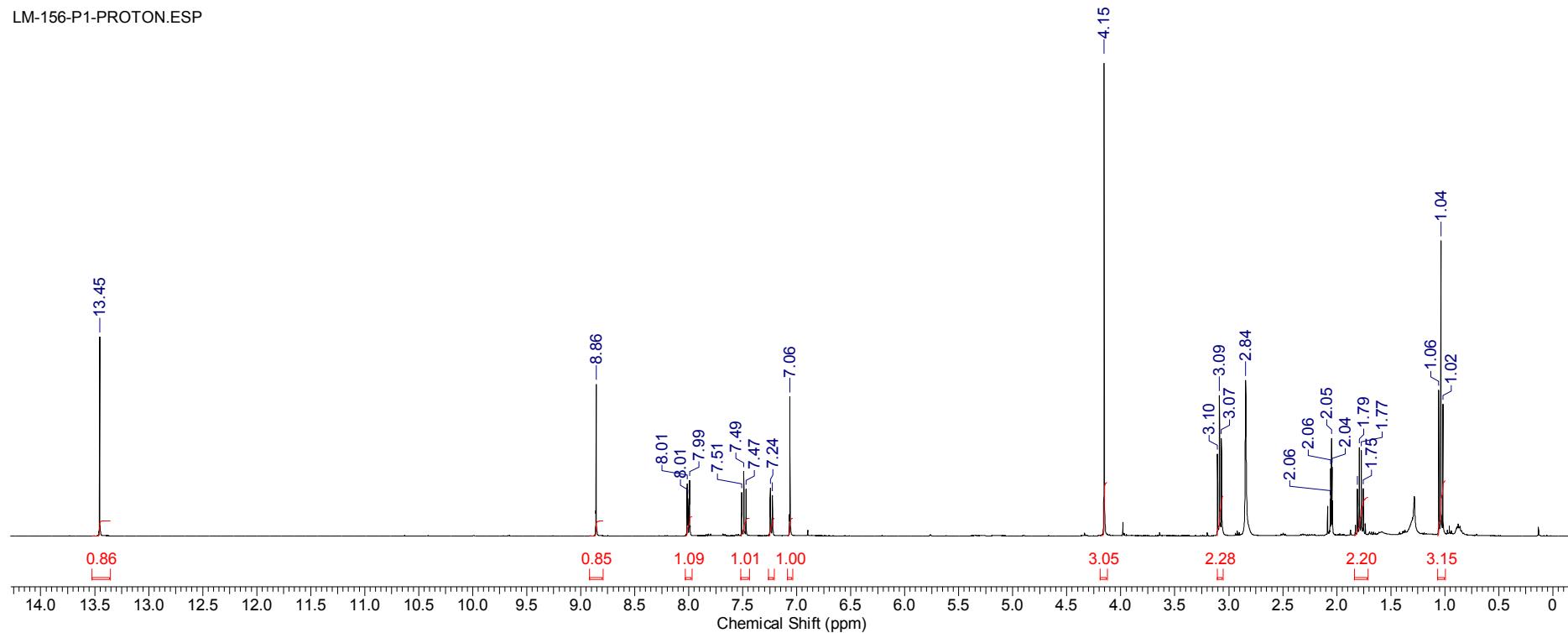


¹³C NMR at 100 MHz in d₆-acetone

1-(1,4-Dihydroxy-5-methoxynaphthalen-2-yl)butan-1-one 9b

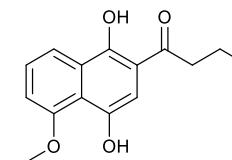


LM-156-P1-PROTON.ESP

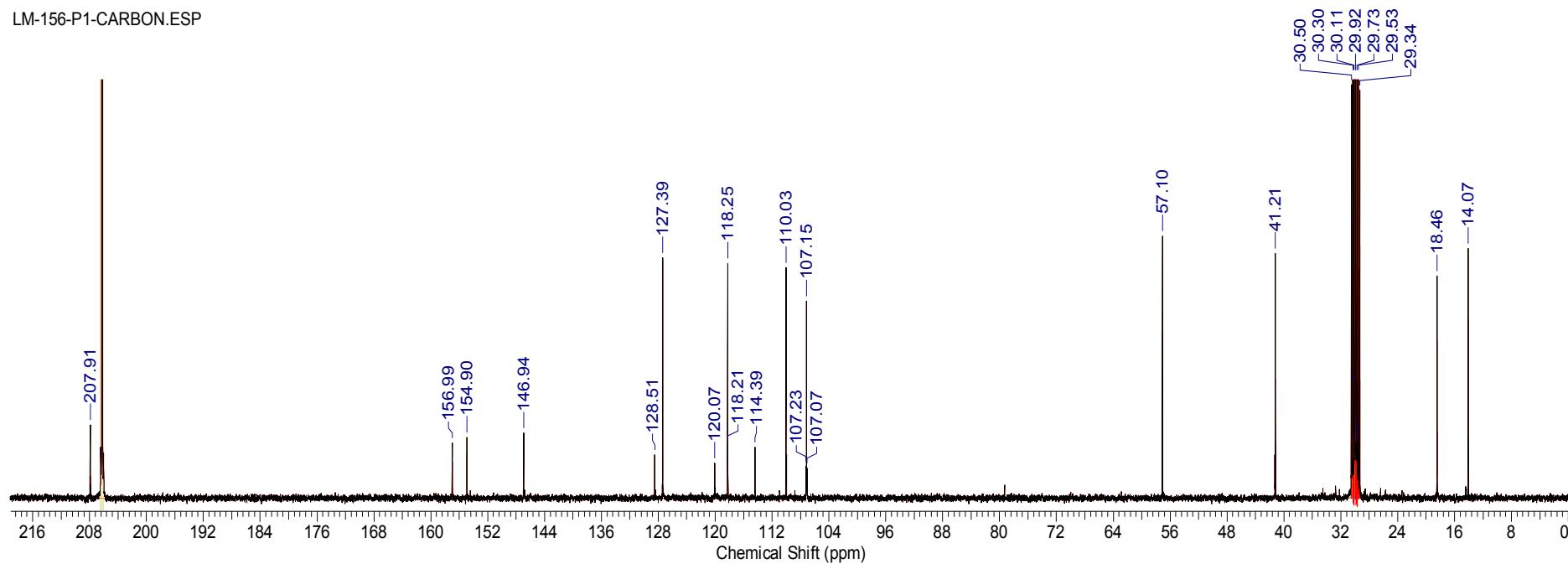


¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-5-methoxynaphthalen-2-yl)butan-1-one 9b

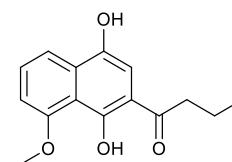


LM-156-P1-CARBON.ESP

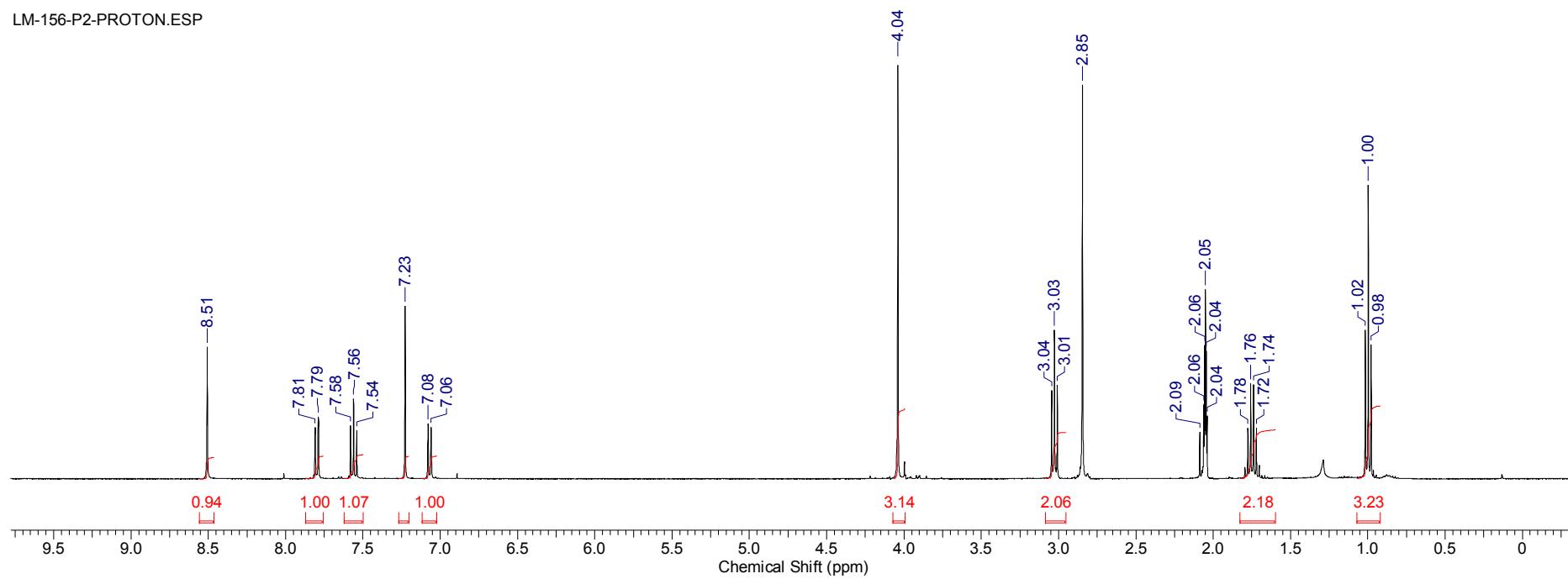


¹³C NMR at 100 MHz in d₆-acetone

1-(1,4-Dihydroxy-8-methoxynaphthalen-2-yl)butan-1-one 10b

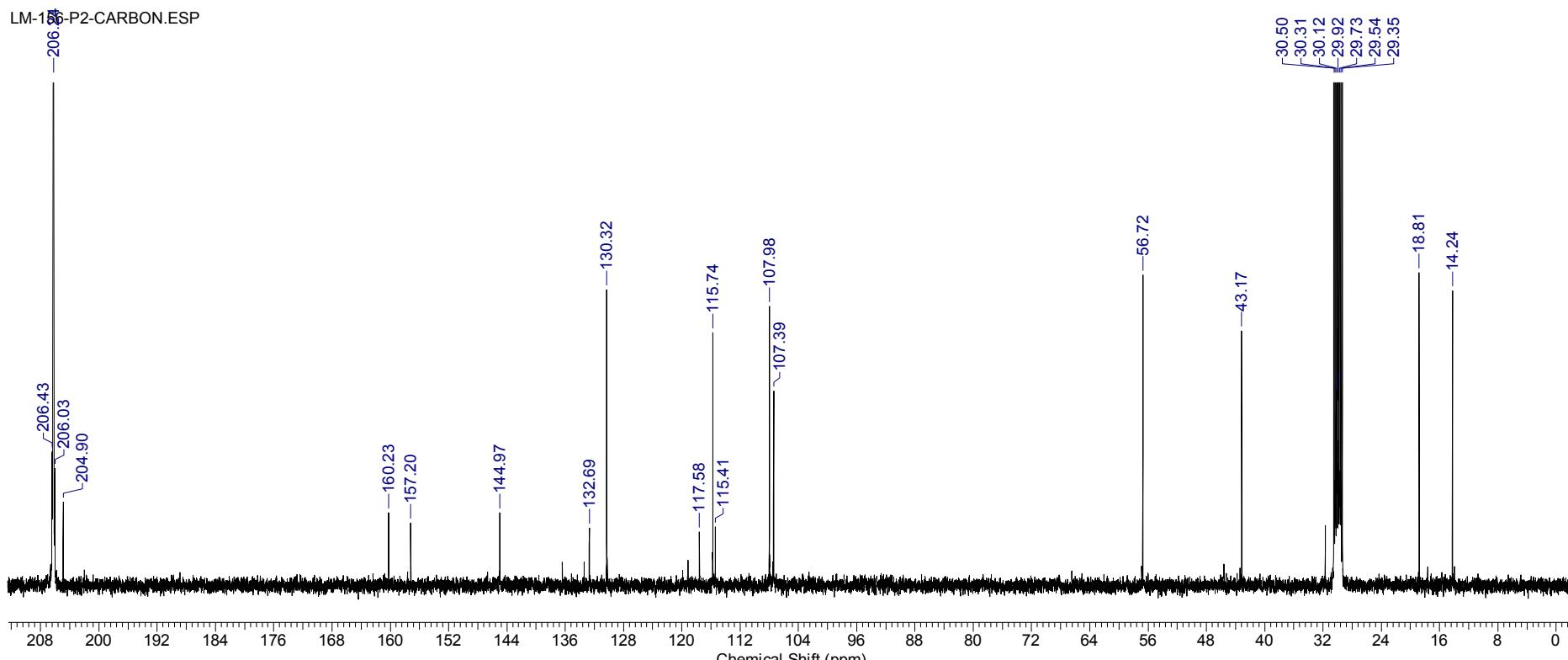
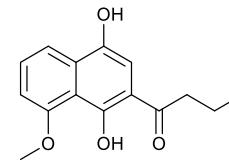


LM-156-P2-PROTON.ESP



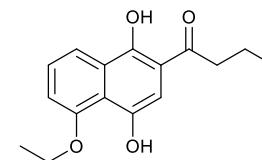
¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-8-methoxynaphthalen-2-yl)butan-1-one 10b

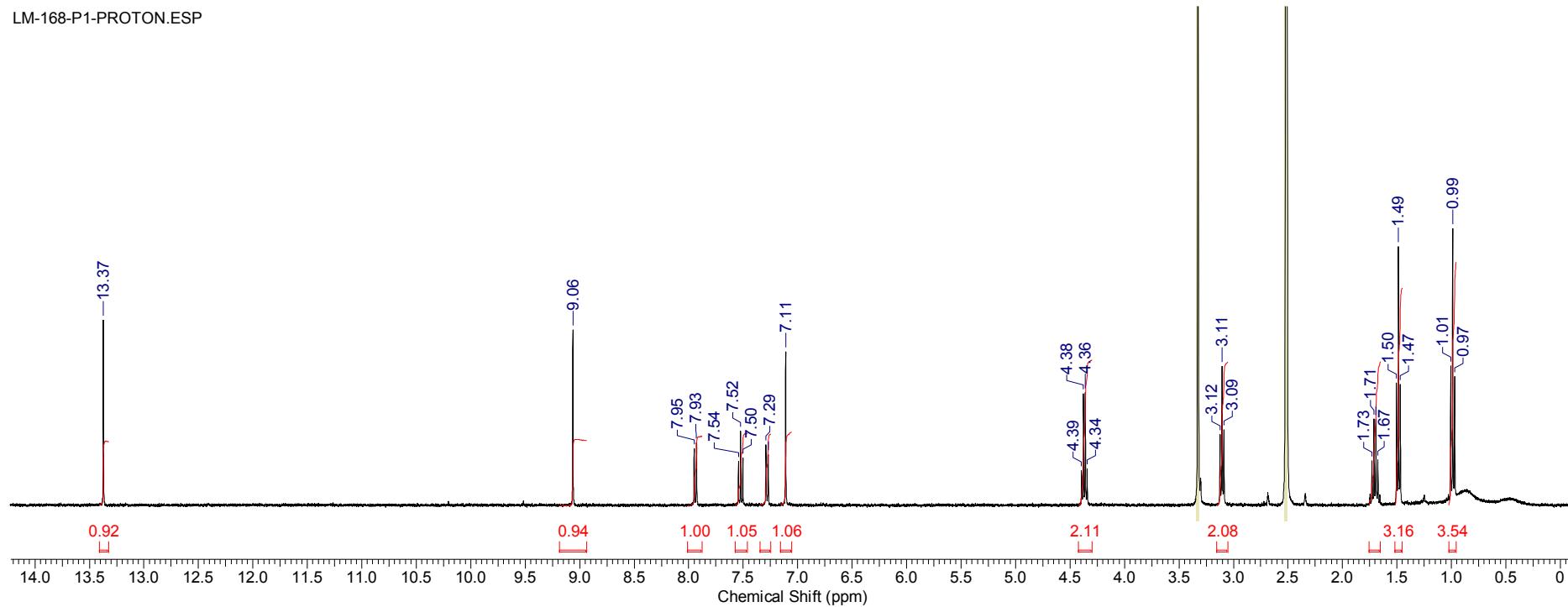


¹³C NMR at 100 MHz in d₆-acetone

1-(5-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 9c

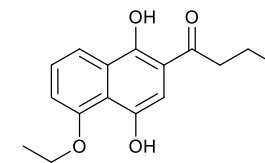


LM-168-P1-PROTON.ESP

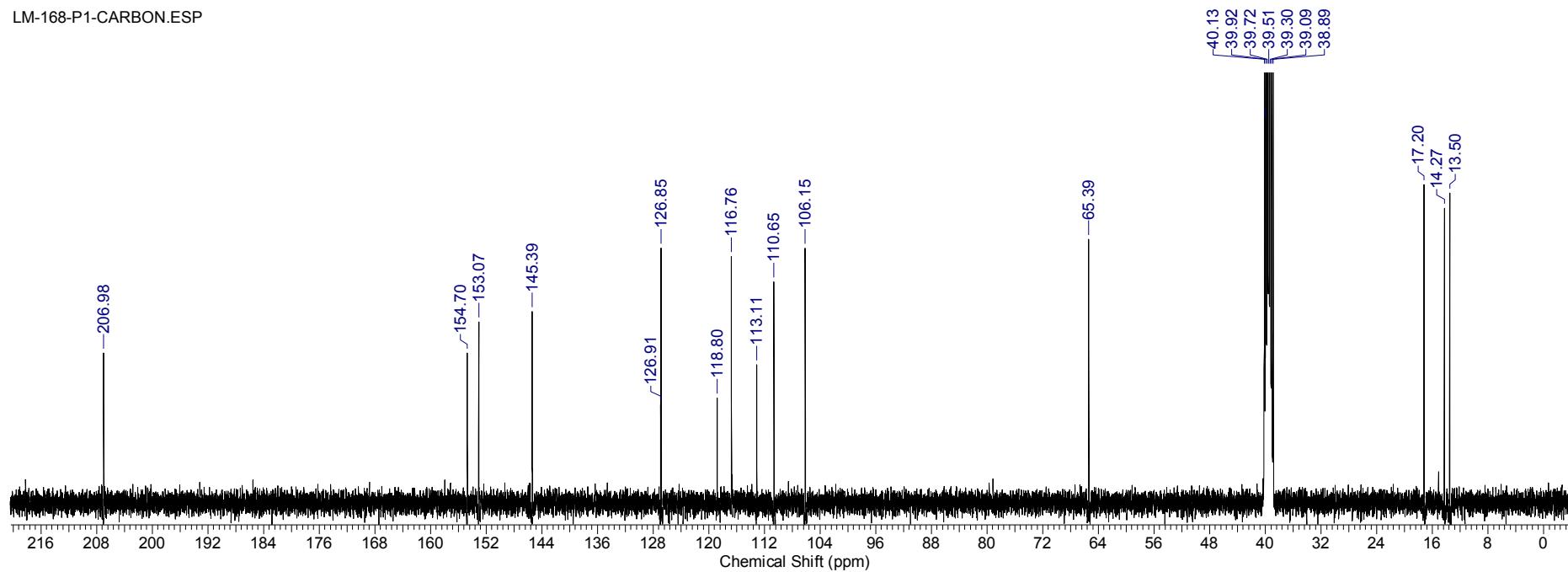


¹H NMR at 400 MHz in d₆-DMSO

1-(5-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 9c

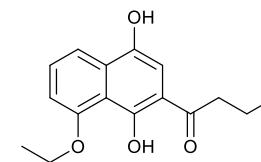


LM-168-P1-CARBON.ESP

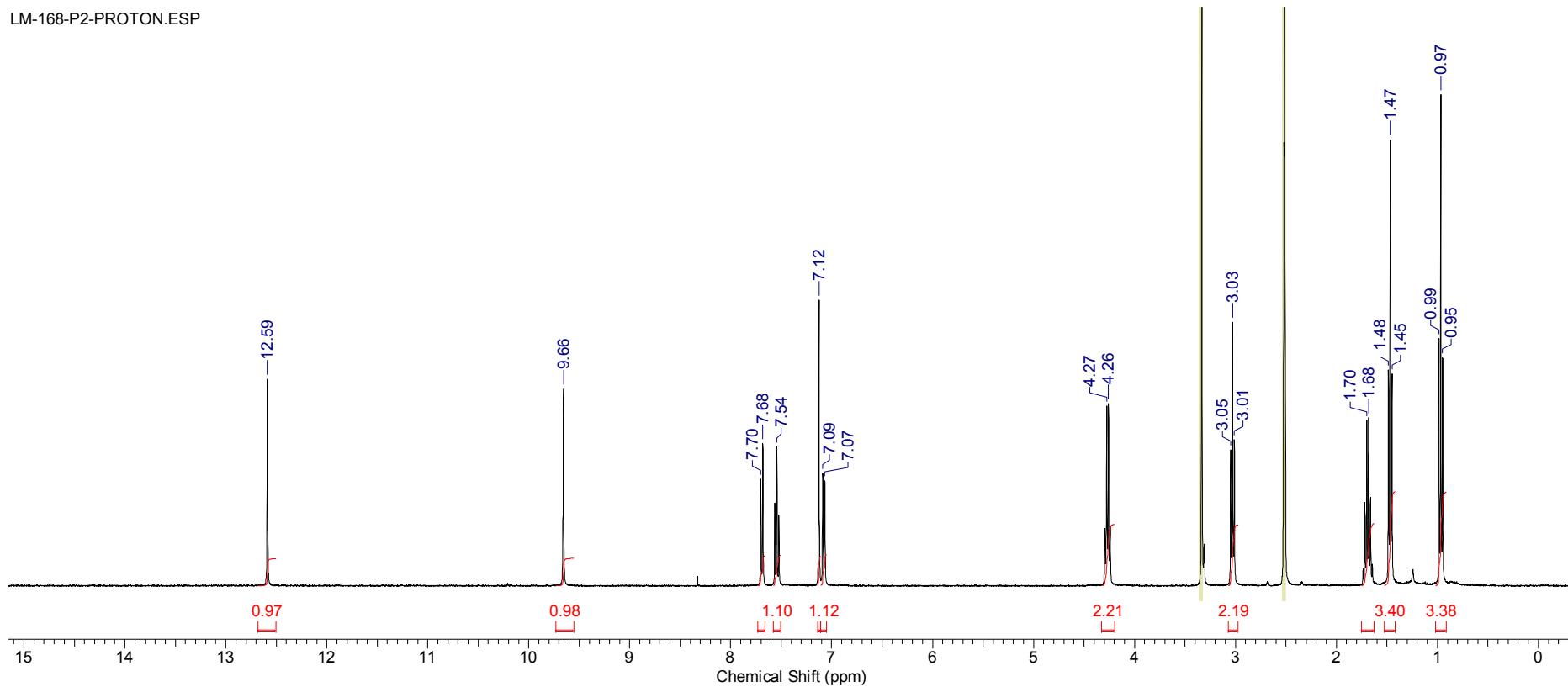


¹³C NMR at 100 MHz in d₆-DMSO

1-(8-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 10c

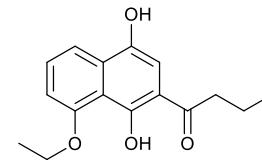


LM-168-P2-PROTON.ESP

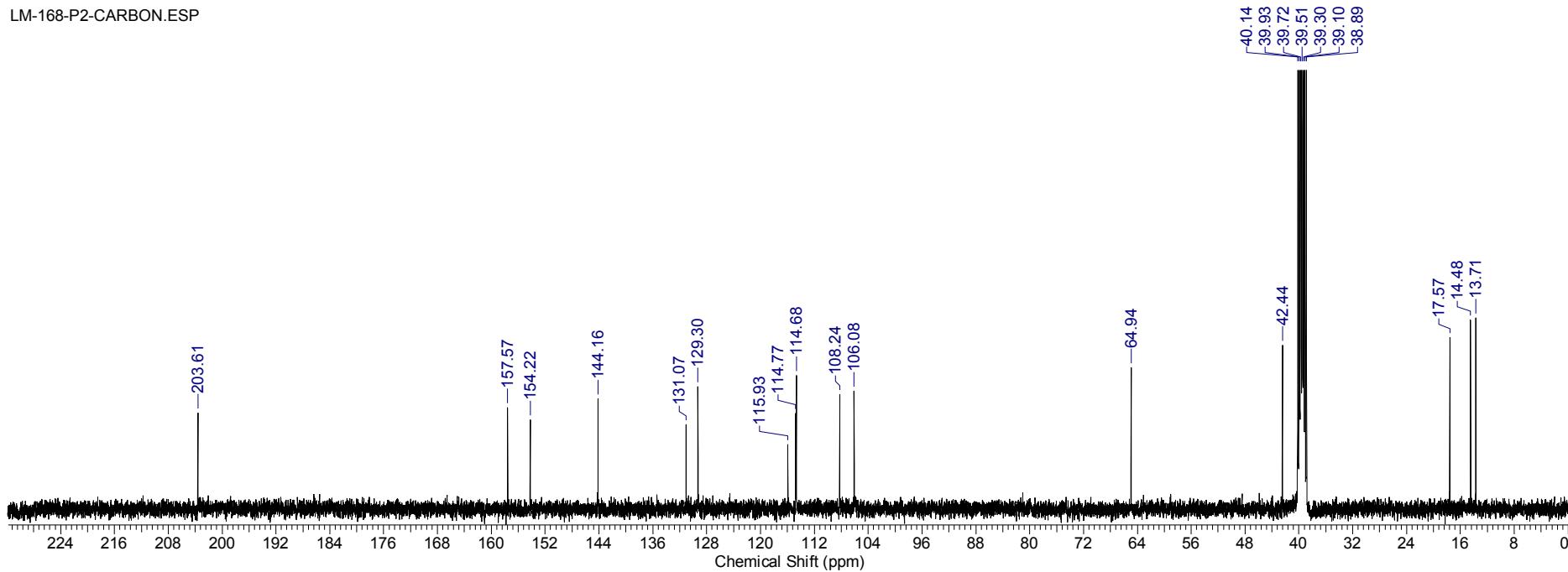


¹H NMR at 400 MHz in d₆-DMSO

1-(8-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 10c

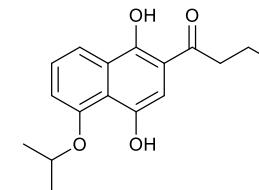


LM-168-P2-CARBON.ESP

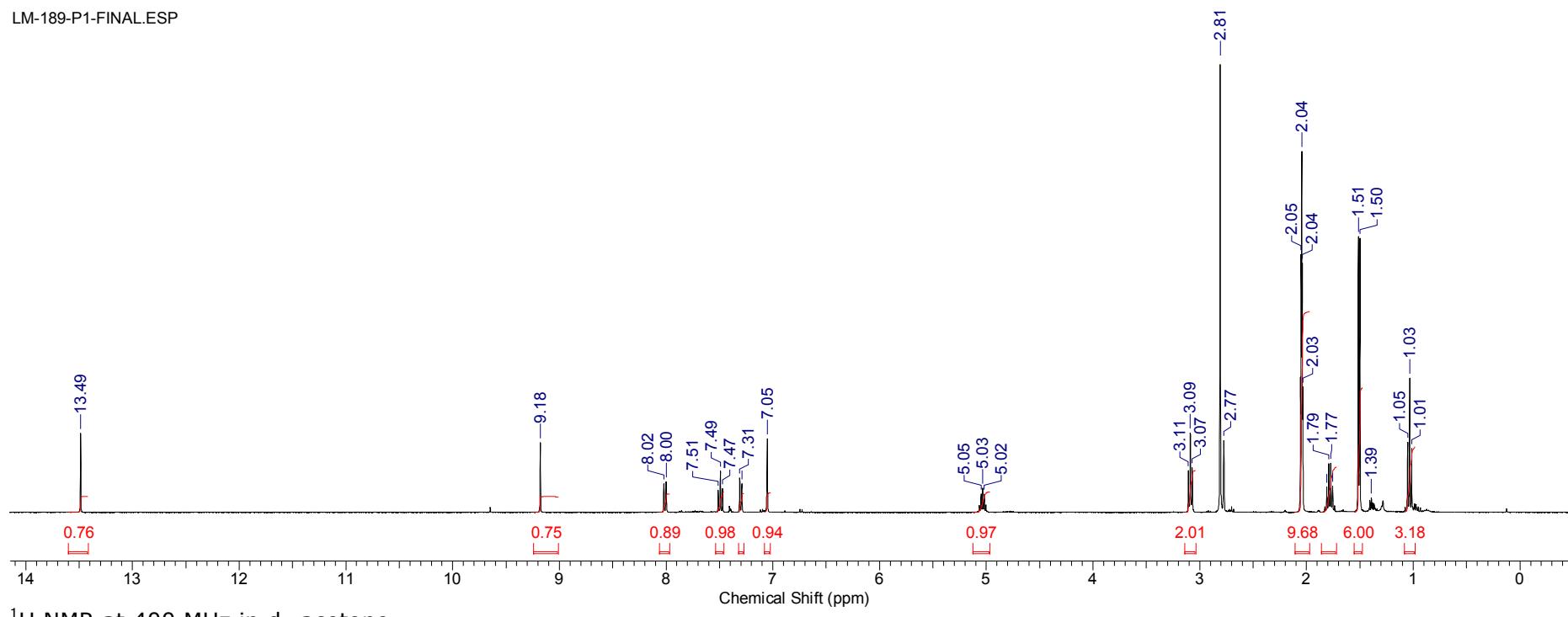


¹³C NMR at 100 MHz in d₆-DMSO

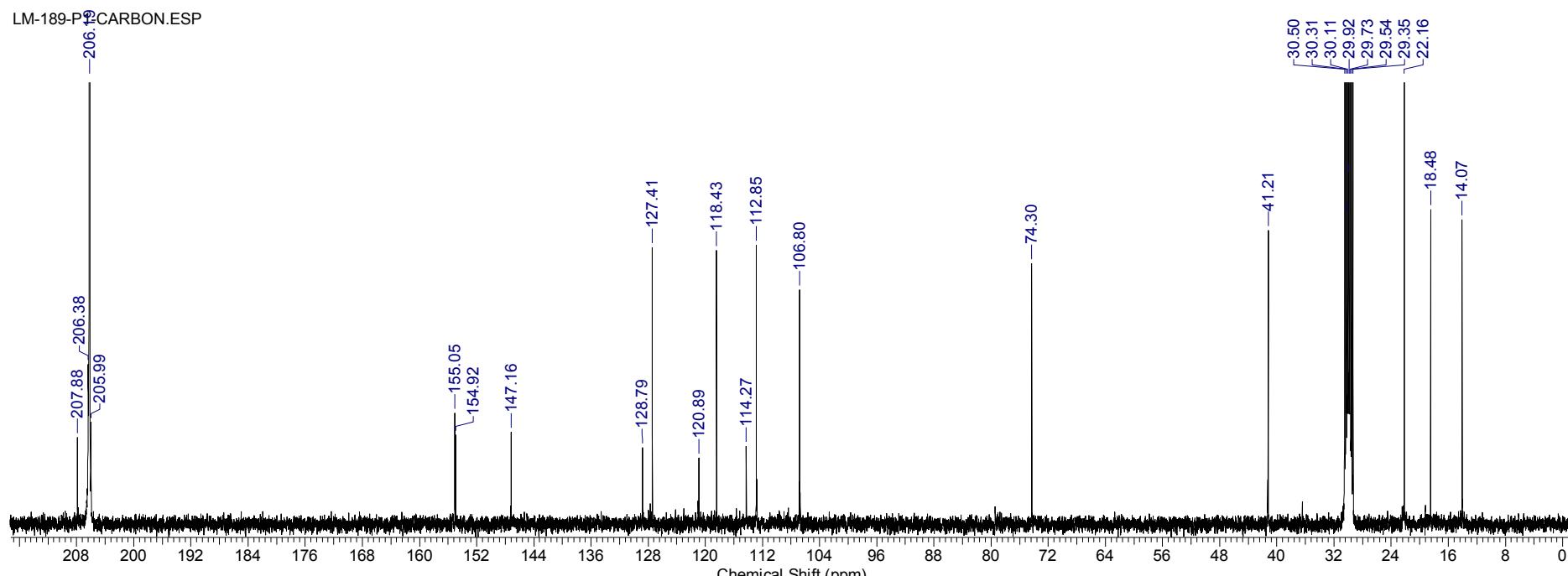
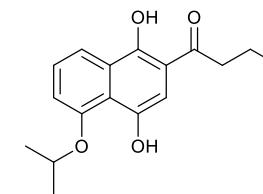
1-(1,4-Dihydroxy-5-isopropoxynaphthalen-2-yl)butan-1-one 9d



LM-189-P1-FINAL.ESP

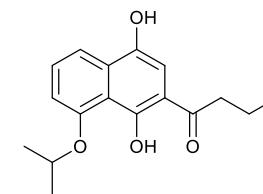


1-(1,4-Dihydroxy-5-isopropoxynaphthalen-2-yl)butan-1-one 9d

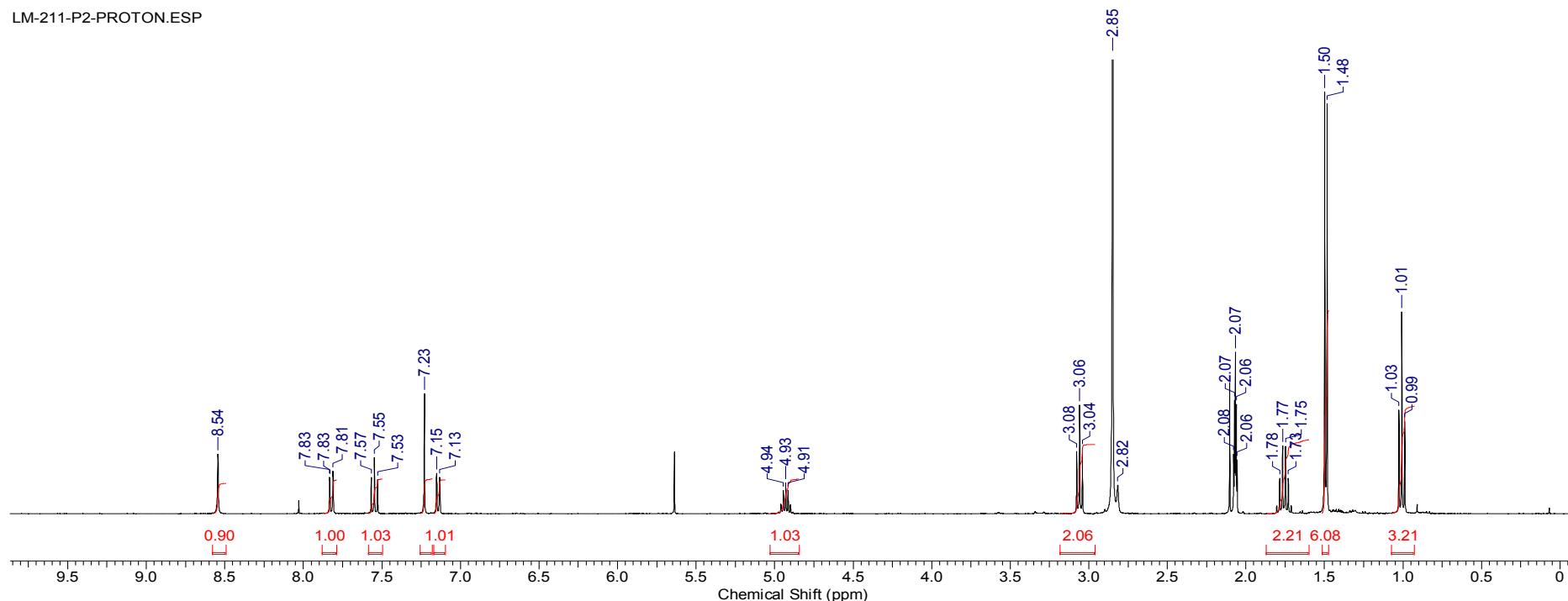


¹³C NMR at 100 MHz in d₆-acetone

1-(1,4-Dihydroxy-8-isopropoxynaphthalen-2-yl)butan-1-one 10d

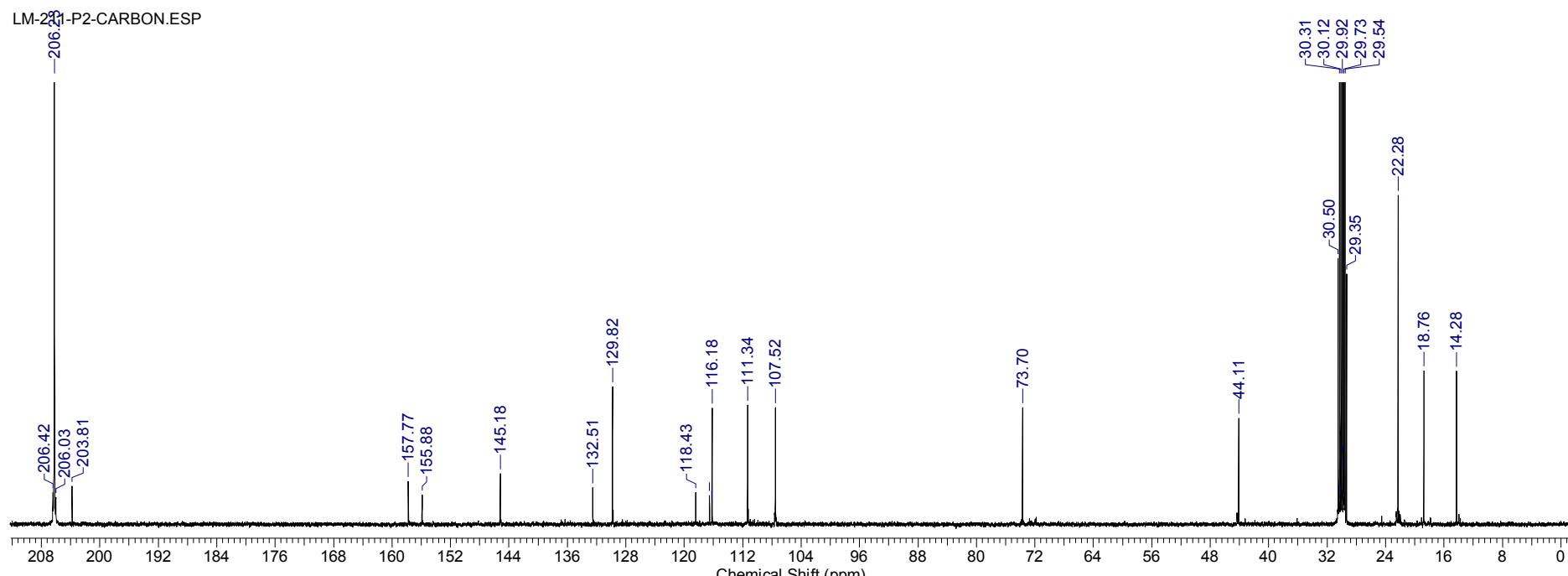
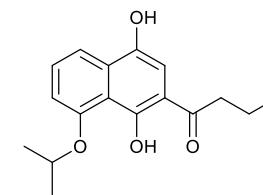


LM-211-P2-PROTON.ESP

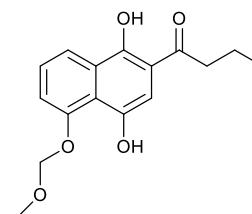


¹H NMR at 400 MHz in d₆-acetone

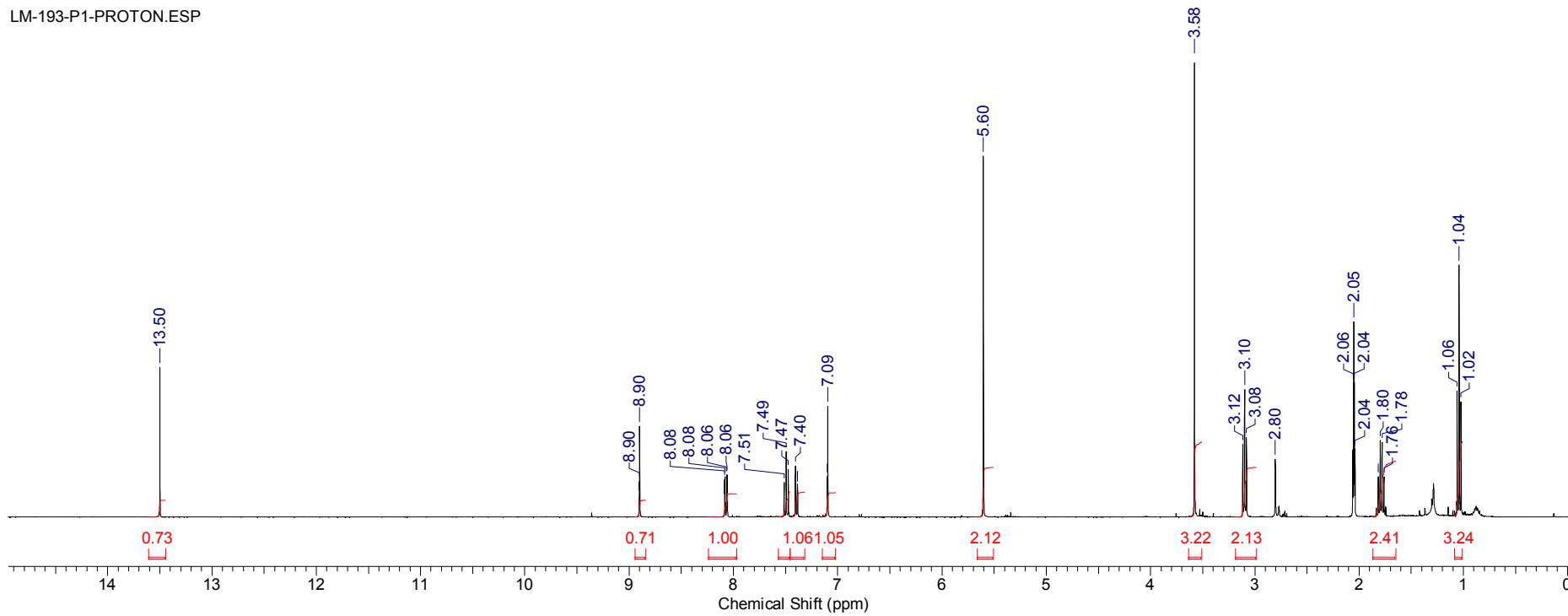
1-(1,4-Dihydroxy-8-isopropoxynaphthalen-2-yl)butan-1-one 10d



1-(1,4-Dihydroxy-5-(methoxymethoxy)naphthalen-2-yl)butan-1-one 9e

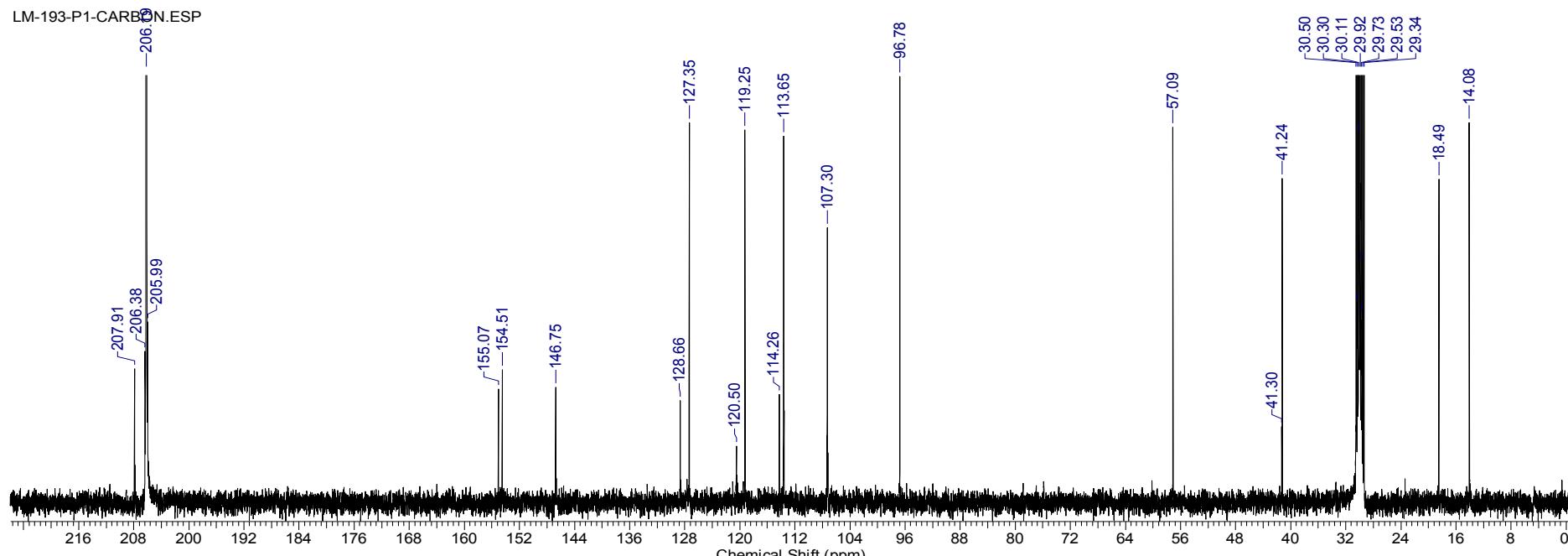
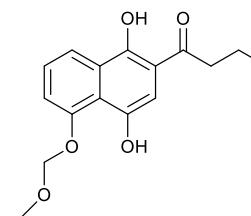


LM-193-P1-PROTON.ESP



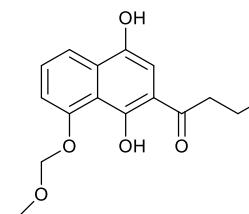
¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-5-(methoxymethoxy)naphthalen-2-yl)butan-1-one 9e

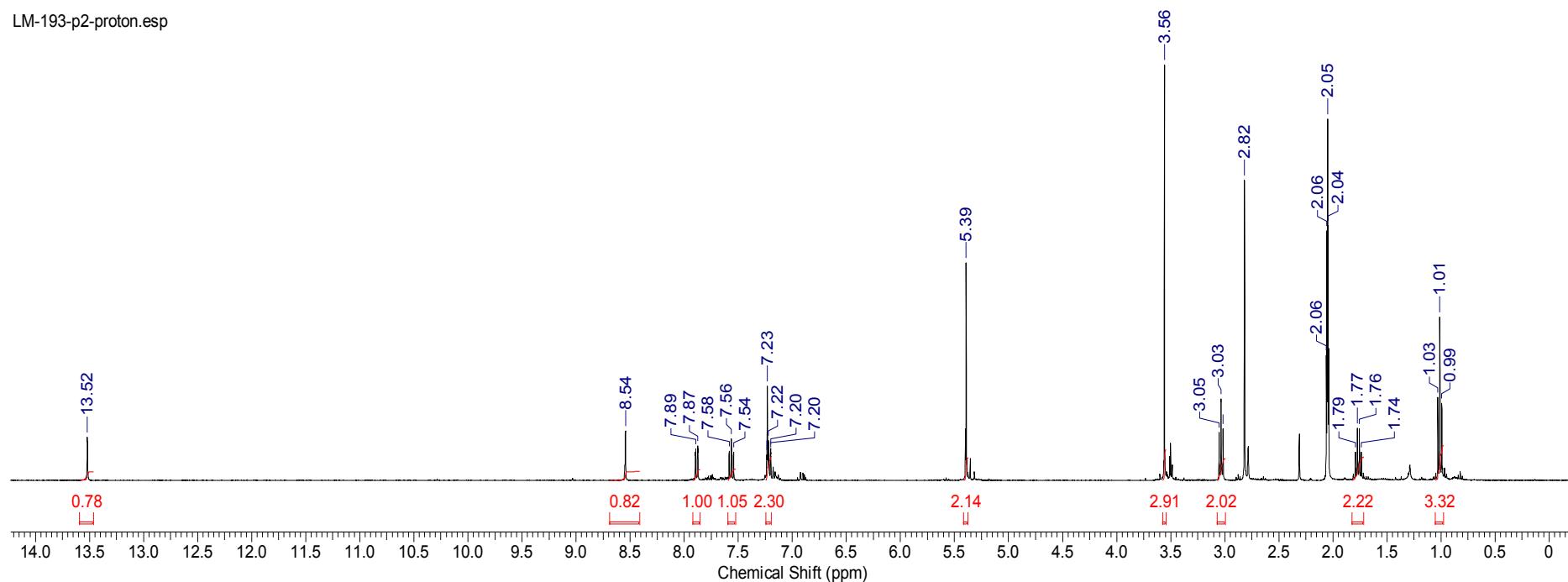


^{13}C NMR at 100 MHz in d_6 -acetone

1-(1,4-Dihydroxy-8-(methoxymethoxy)naphthalen-2-yl)butan-1-one 10e

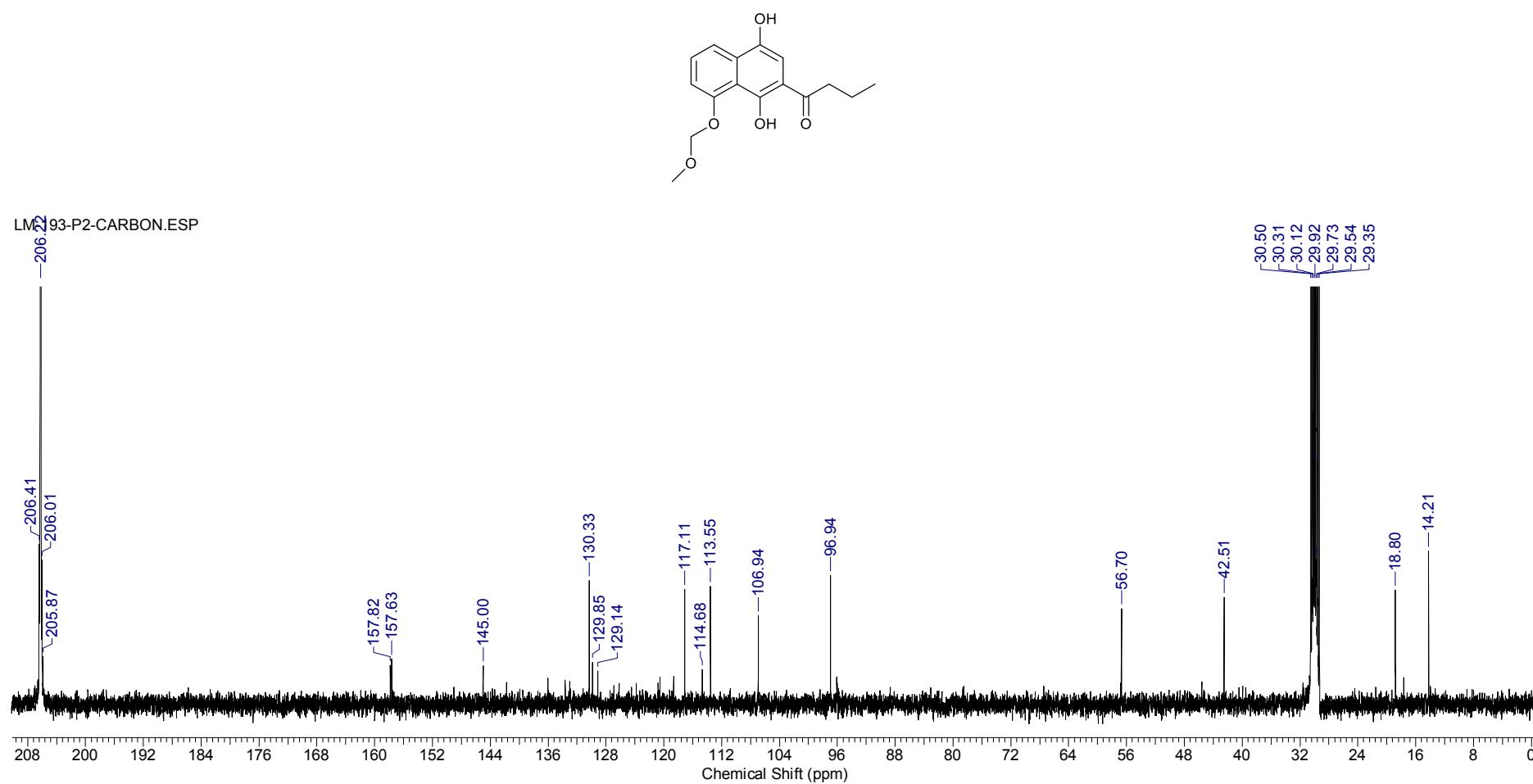


LM-193-p2-proton.esp



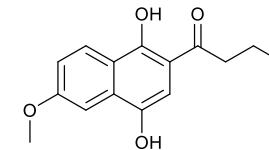
¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-8-(methoxymethoxy)naphthalen-2-yl)butan-1-one 10e

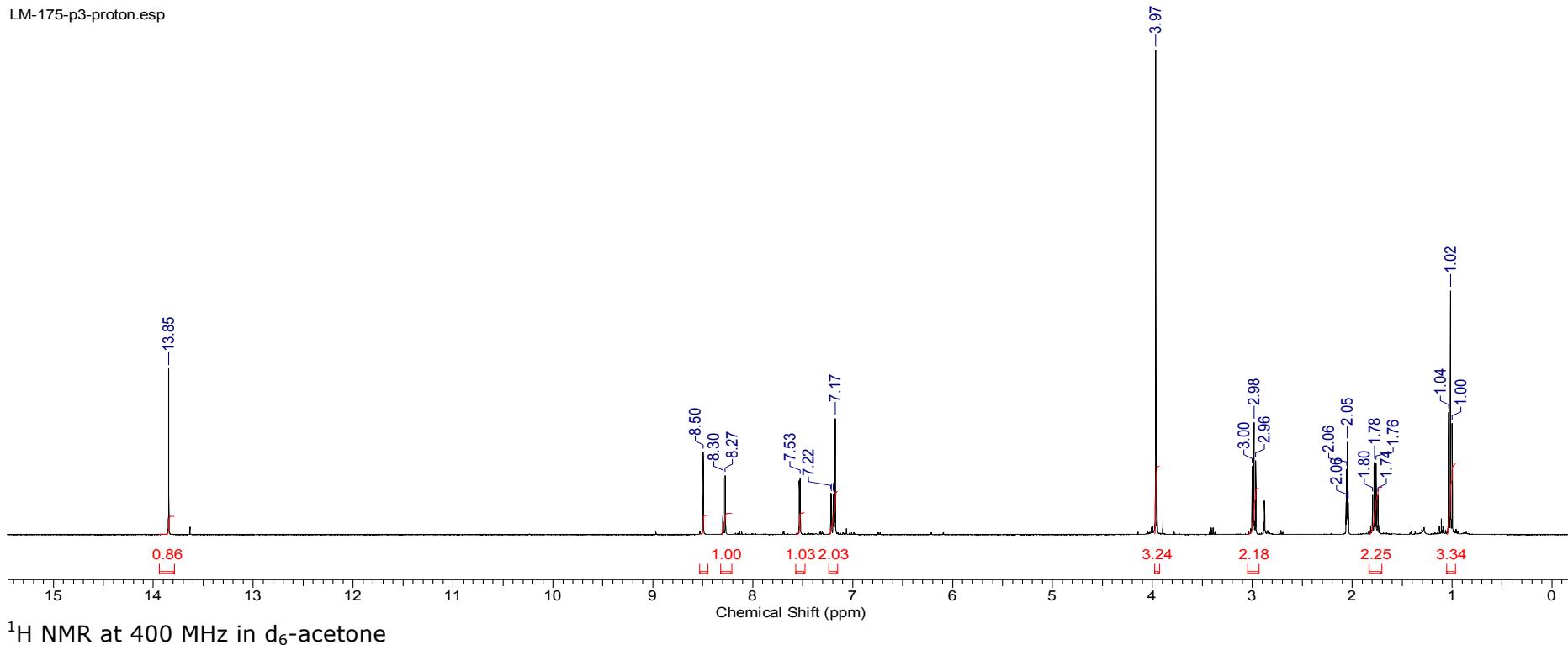


¹³C NMR at 100 MHz in d₆-acetone

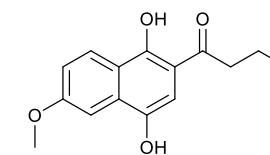
1-(1,4-Dihydroxy-6-methoxynaphthalen-2-yl)butan-1-one 13b



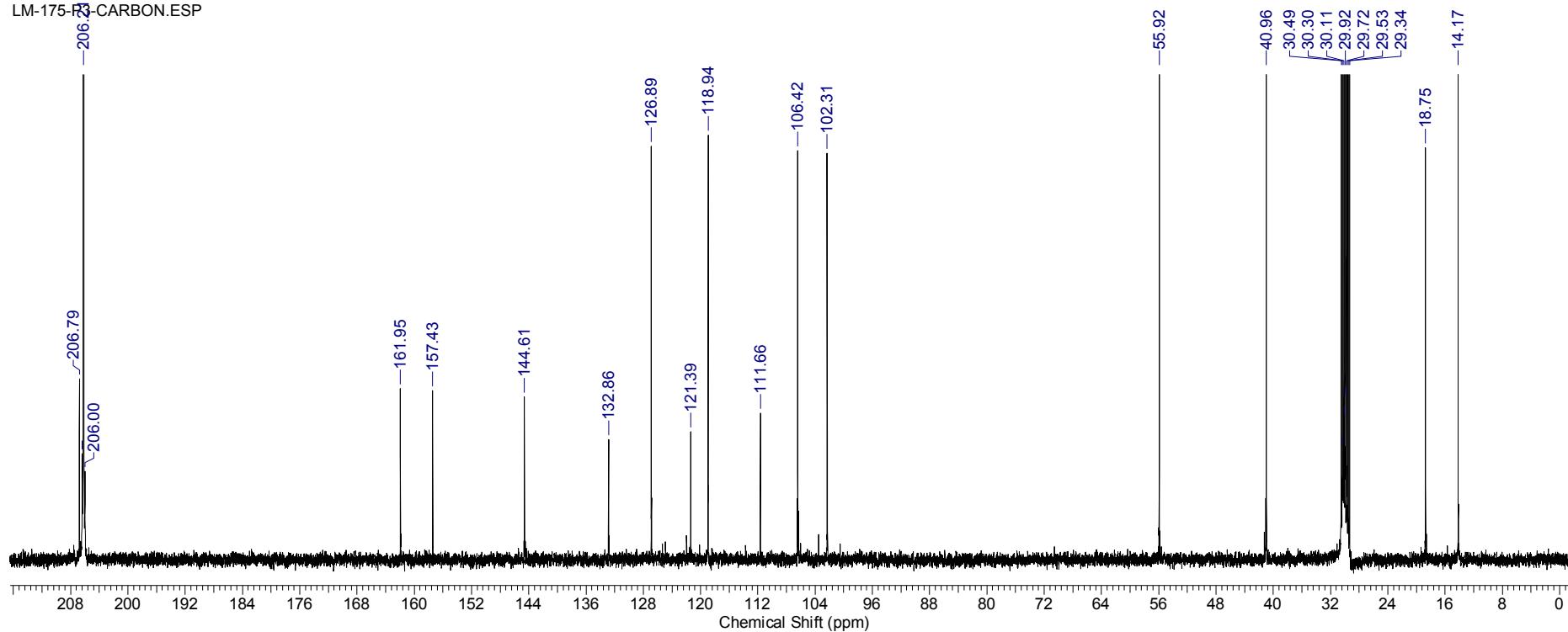
LM-175-p3-proton.esp



1-(1,4-Dihydroxy-6-methoxynaphthalen-2-yl)butan-1-one 13b

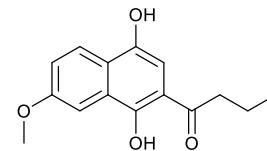


LM-175-R3-CARBON.ESP

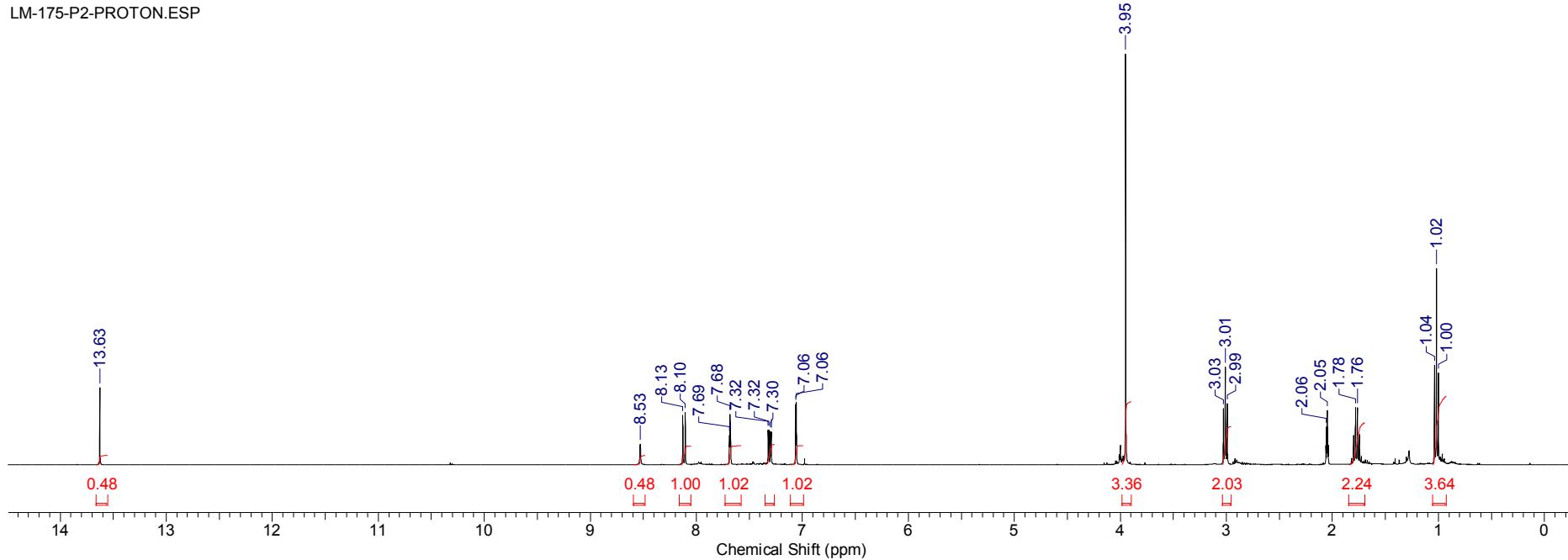


^{13}C NMR at 100 MHz in d_6 -acetone

1-(1,4-Dihydroxy-7-methoxynaphthalen-2-yl)butan-1-one 14b

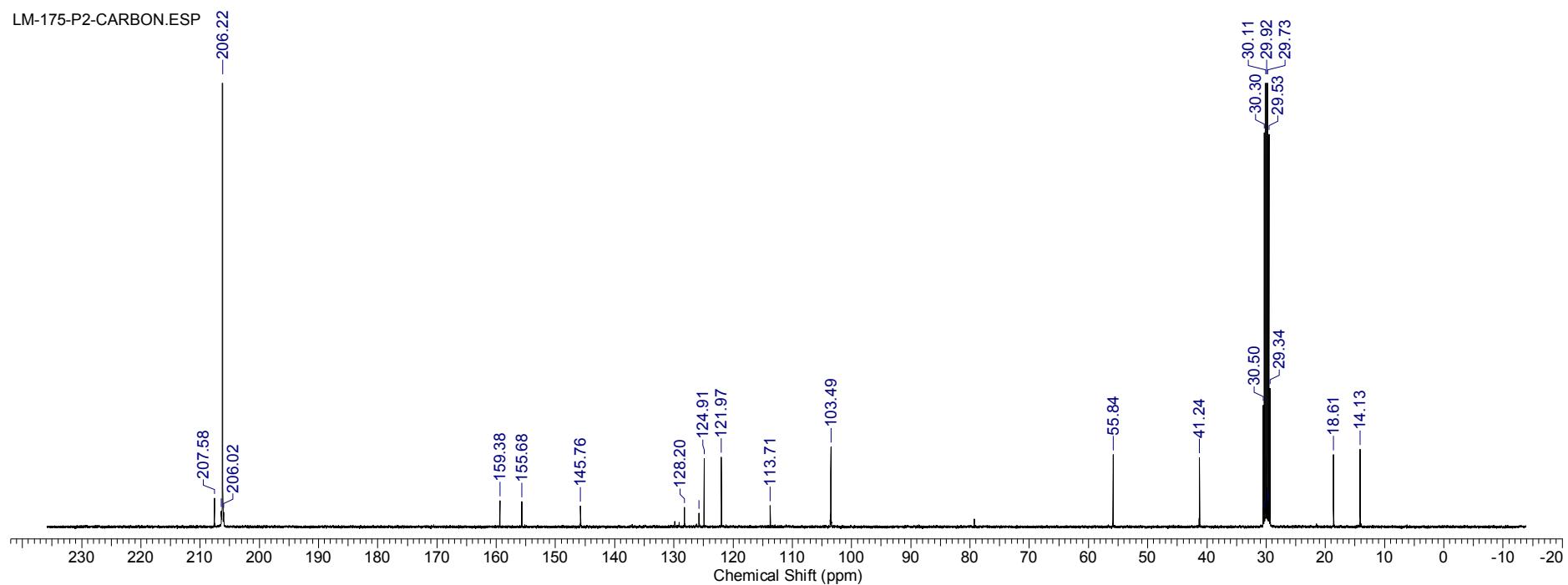
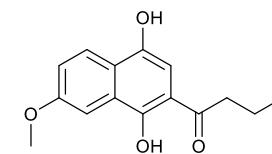


LM-175-P2-PROTON.ESP

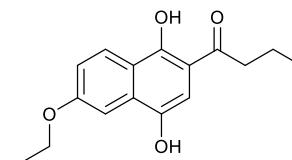


¹H NMR at 400 MHz in d₆-acetone

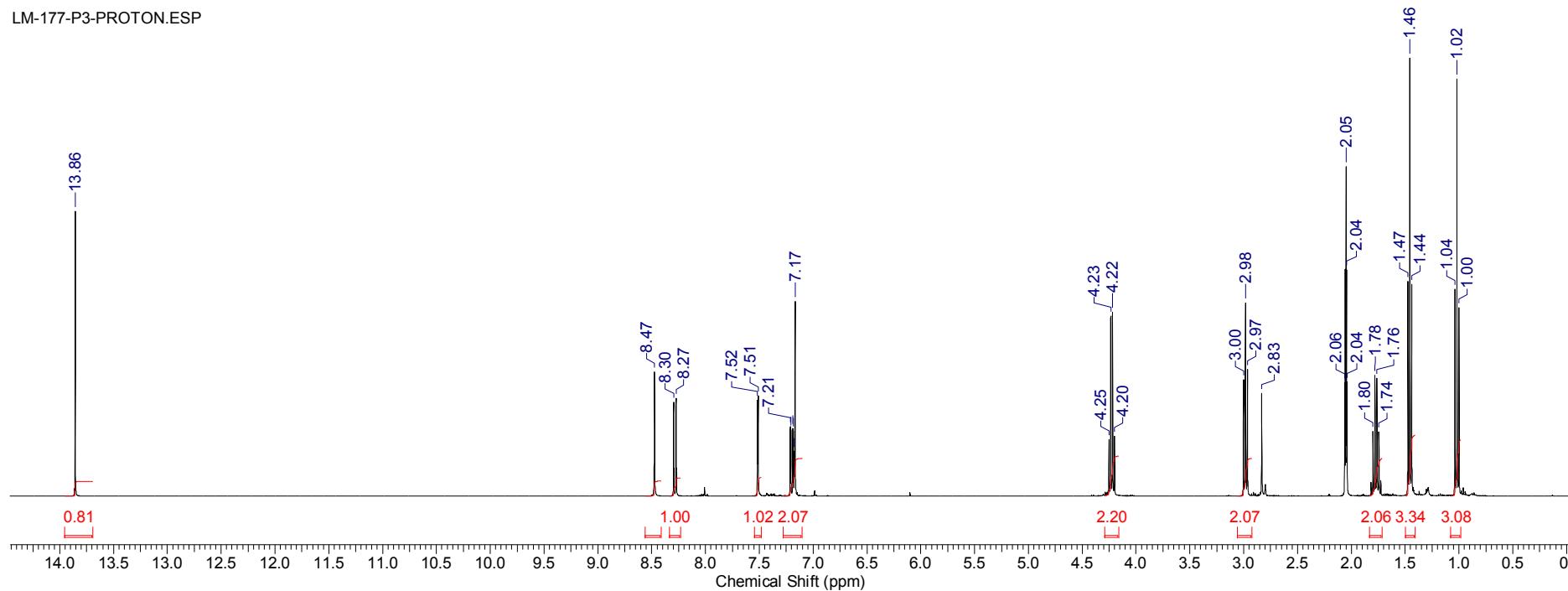
1-(1,4-Dihydroxy-7-methoxynaphthalen-2-yl)butan-1-one 14b



1-(6-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 13c

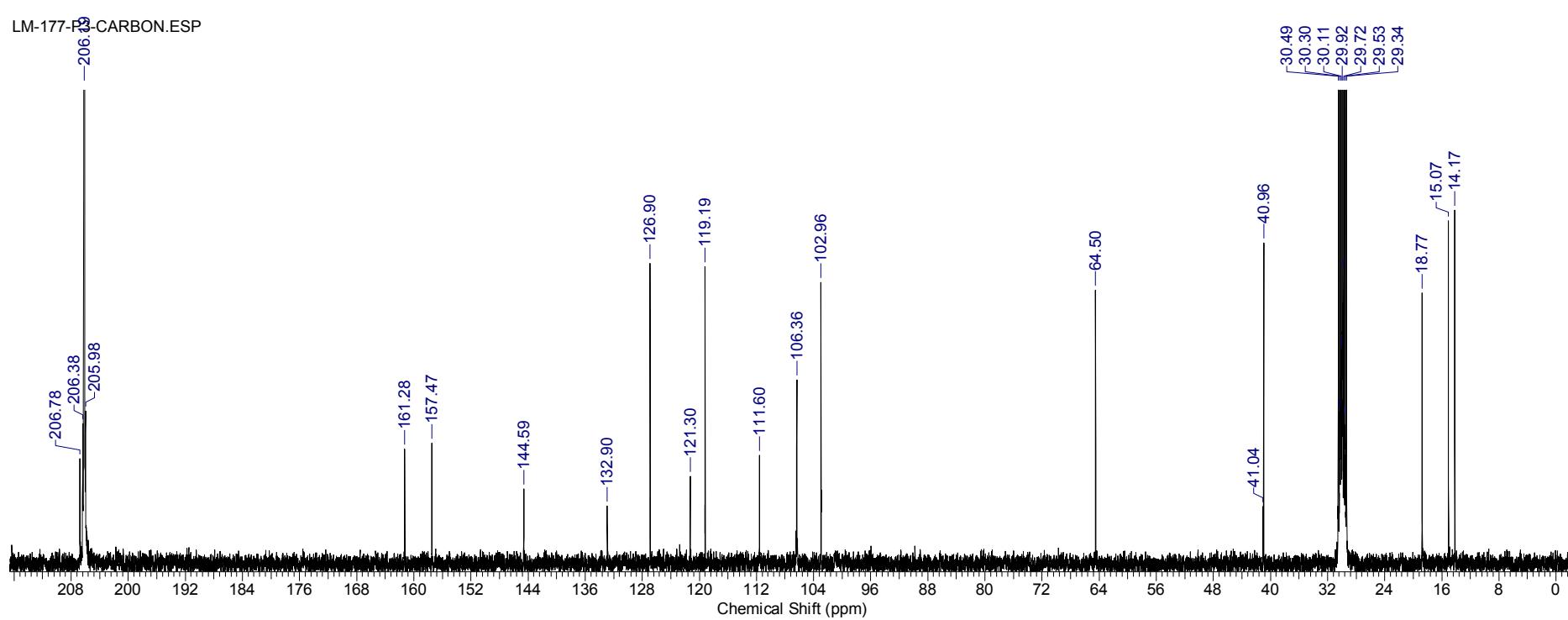
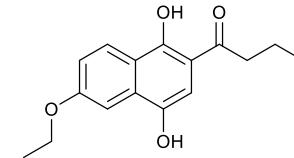


LM-177-P3-PROTON.ESP



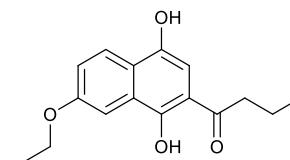
¹H NMR at 400 MHz in d₆-acetone

1-(6-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 13c

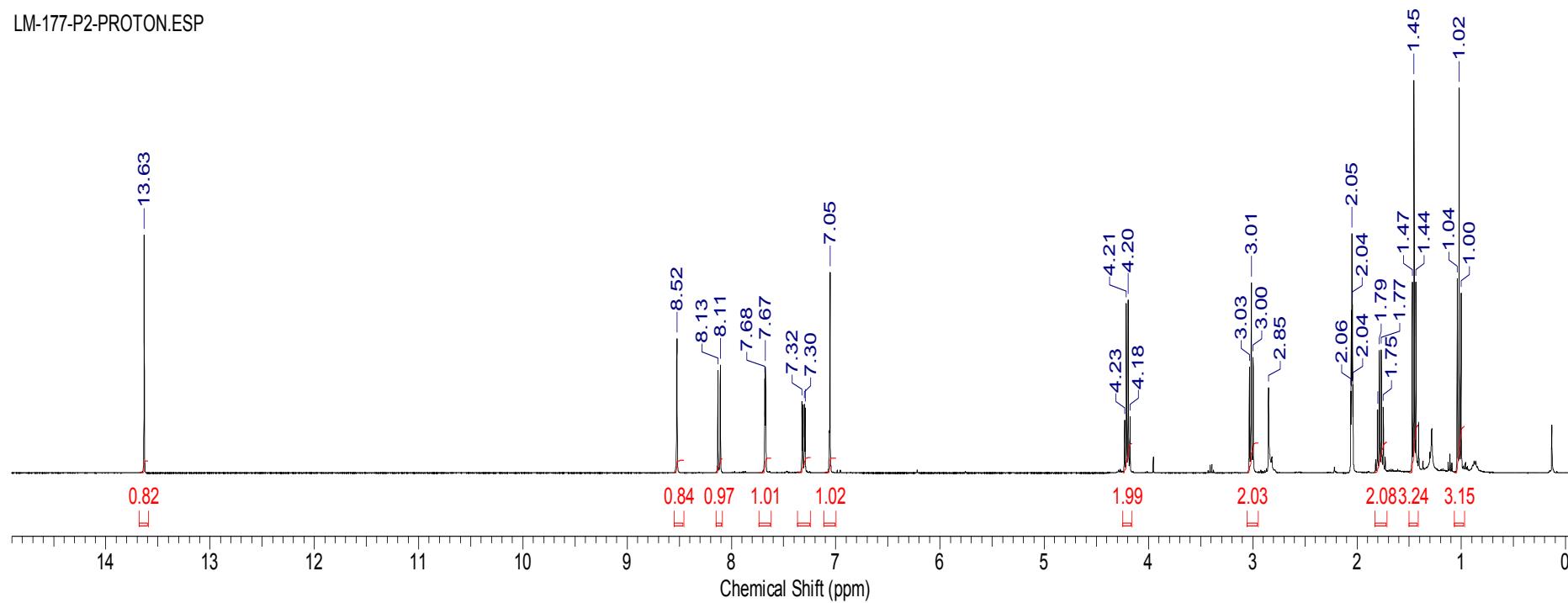


¹³C NMR at 100 MHz in d₆-acetone

1-(7-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 14c

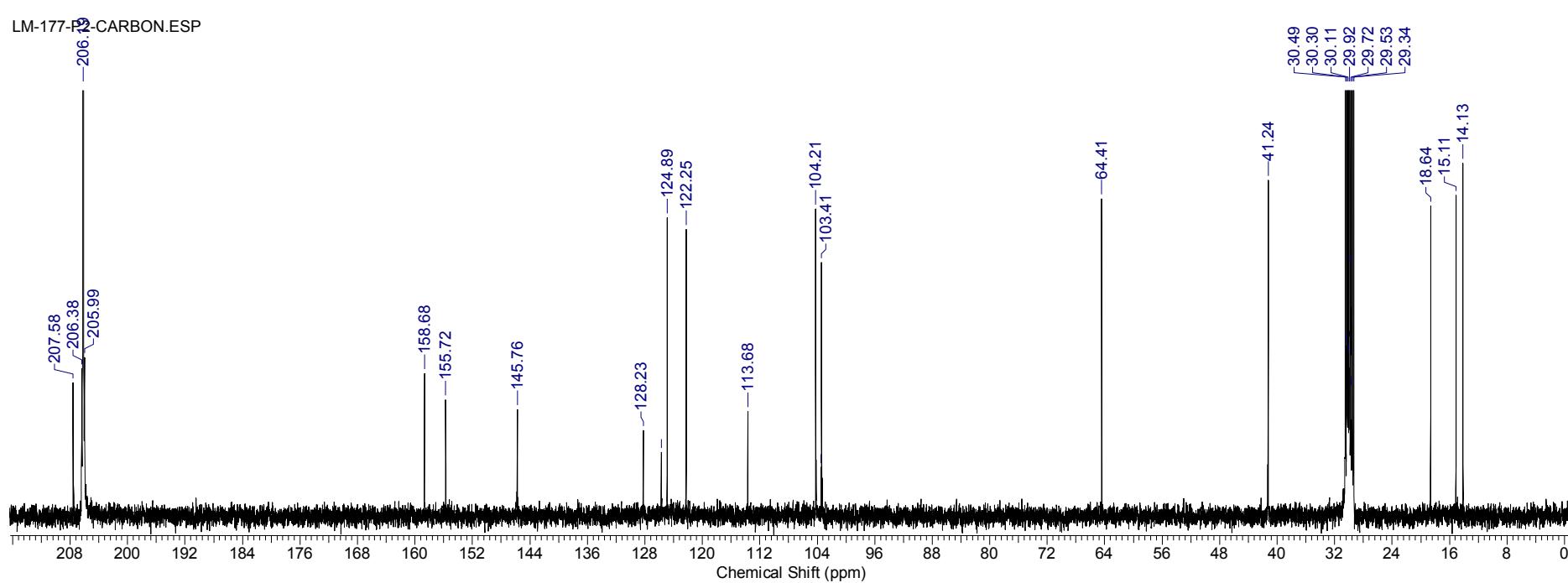
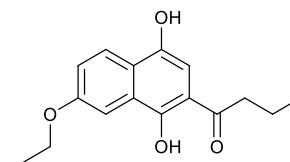


LM-177-P2-PROTON.ESP



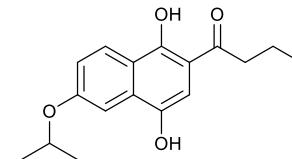
¹H NMR at 400 MHz in d₆-acetone

1-(7-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 14c

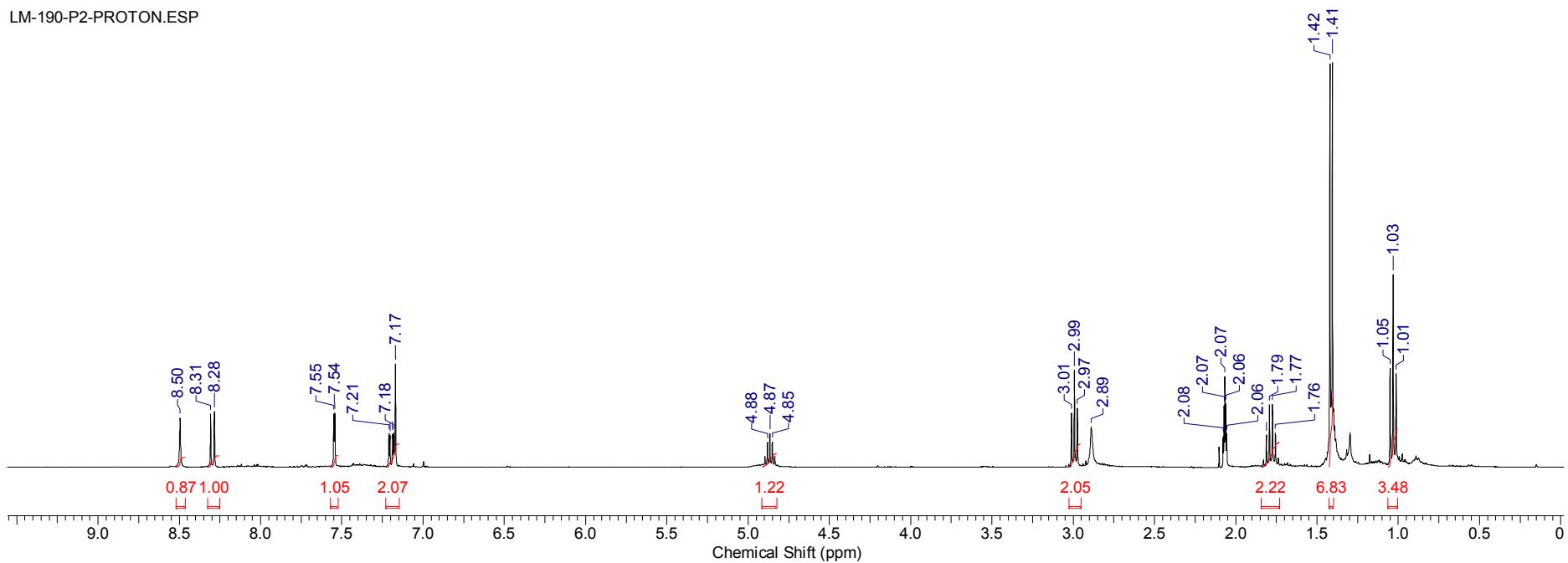


¹³C NMR at 100 MHz in d₆-acetone

1-(1,4-Dihydroxy-6-isopropoxynaphthalen-2-yl)butan-1-one 13d

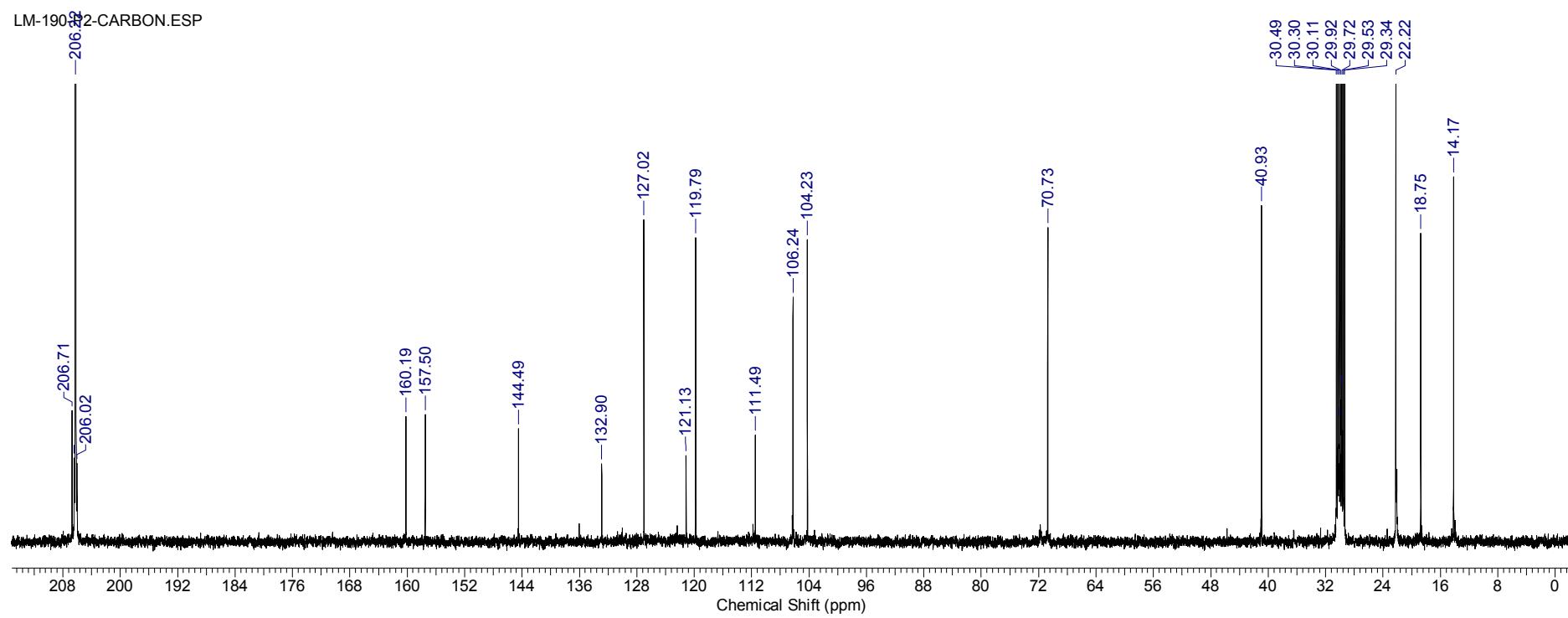
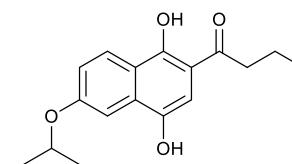


LM-190-P2-PROTON.ESP



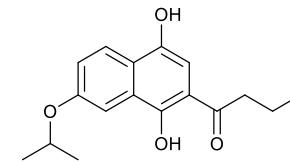
¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-6-isopropoxynaphthalen-2-yl)butan-1-one 13d

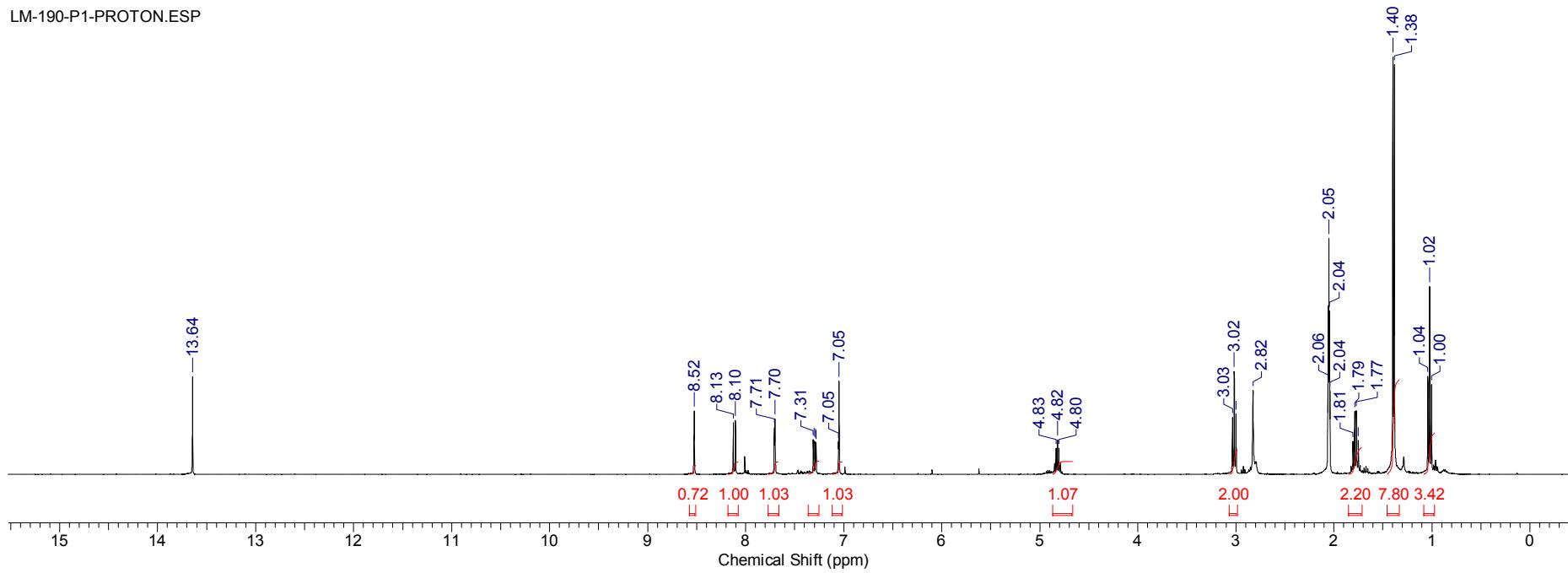


¹³C NMR at 100 MHz in d₆-acetone

1-(1,4-Dihydroxy-7-isopropoxynaphthalen-2-yl)butan-1-one 14d

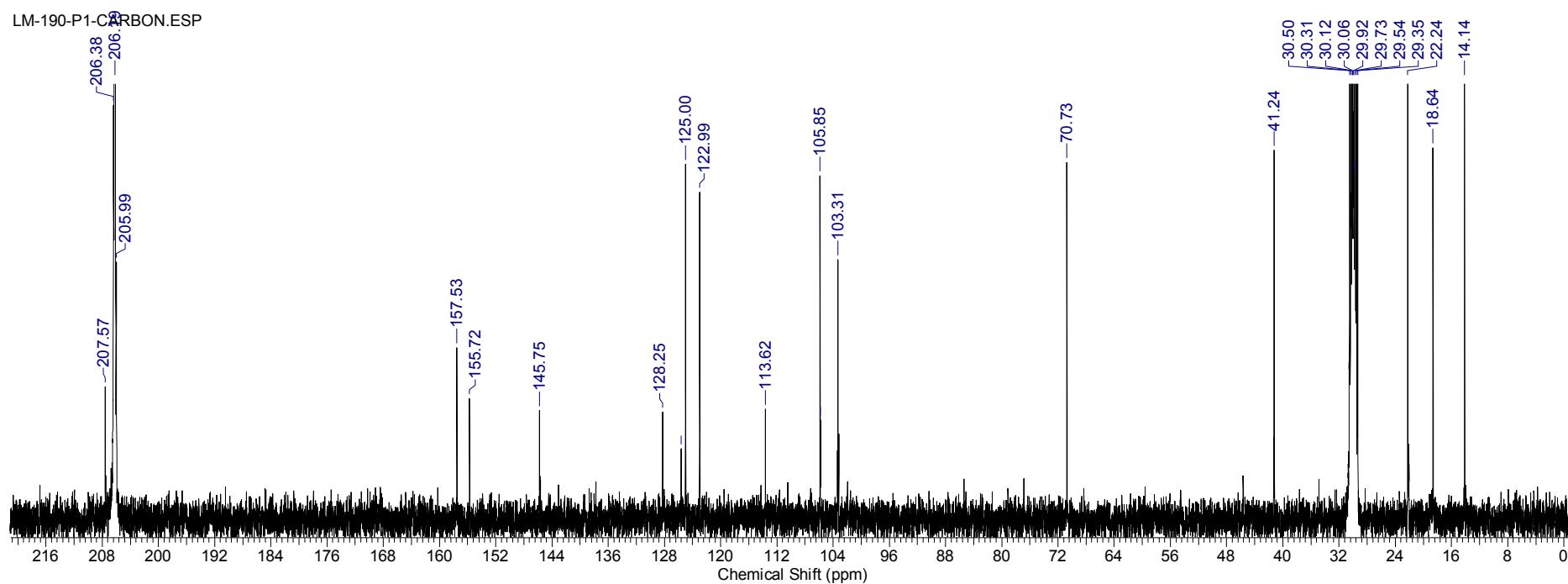
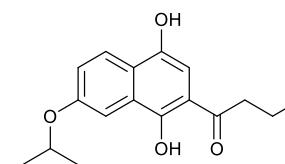


LM-190-P1-PROTON.ESP



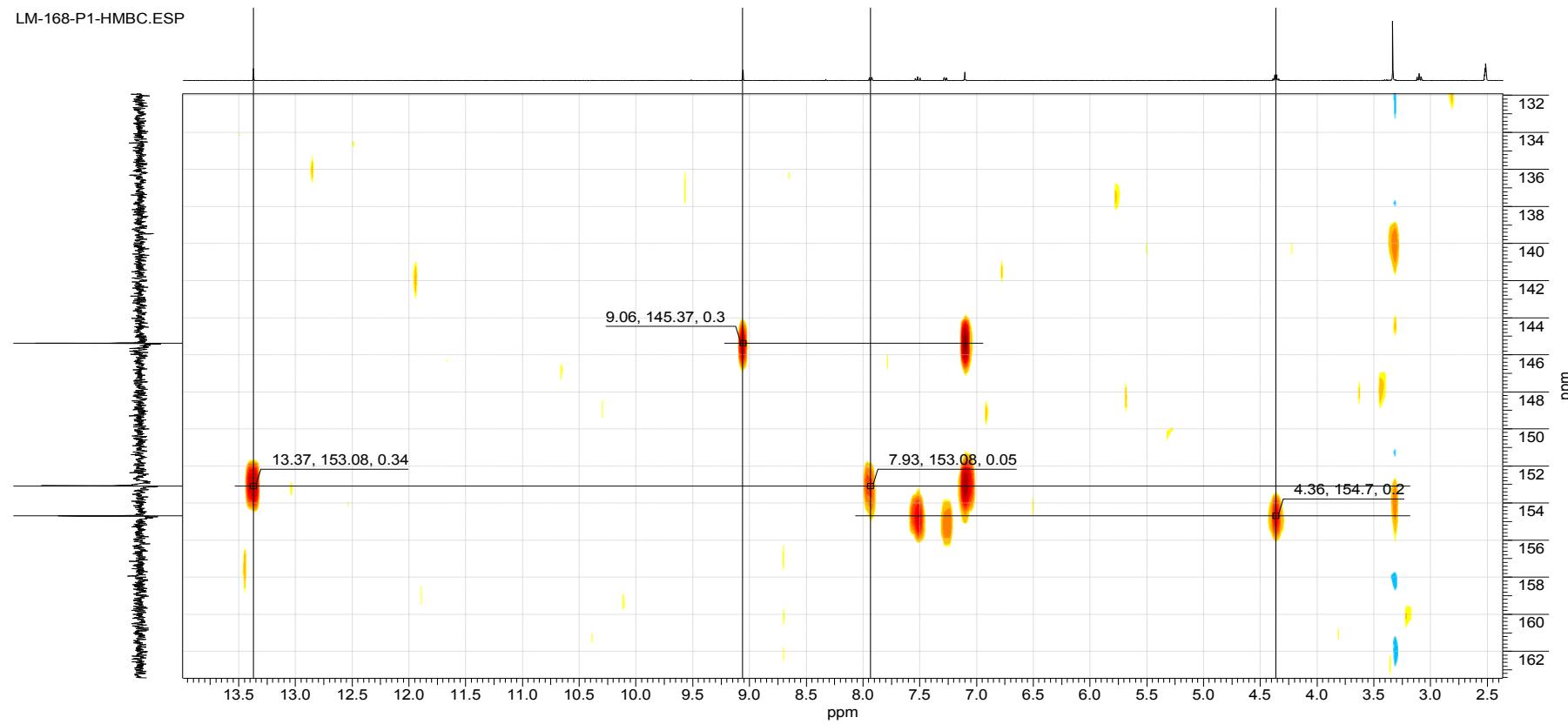
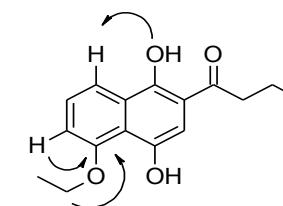
¹H NMR at 400 MHz in d₆-acetone

1-(1,4-Dihydroxy-7-isopropoxynaphthalen-2-yl)butan-1-one 14d

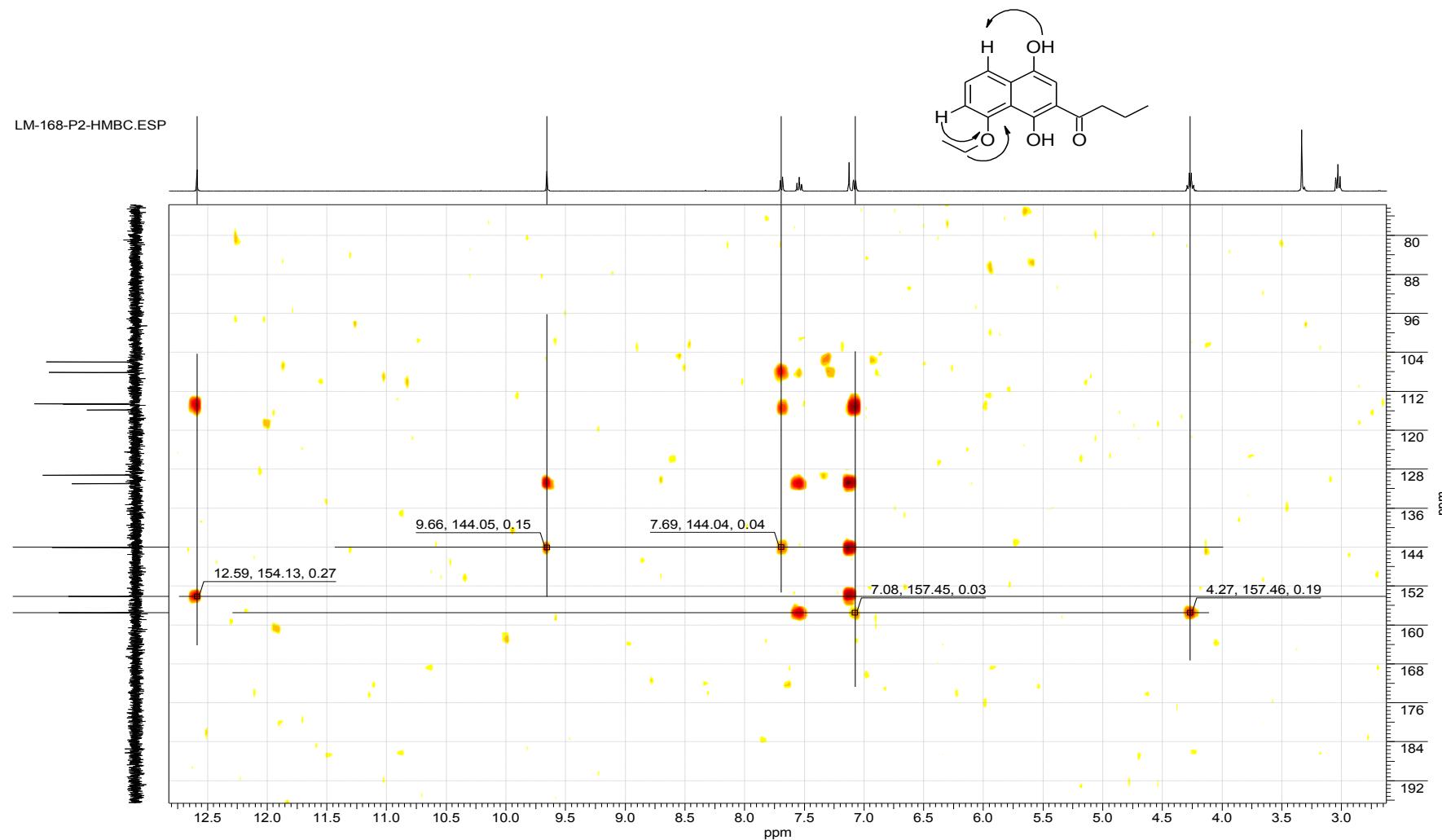


¹³C NMR at 100 MHz in d₆-acetone

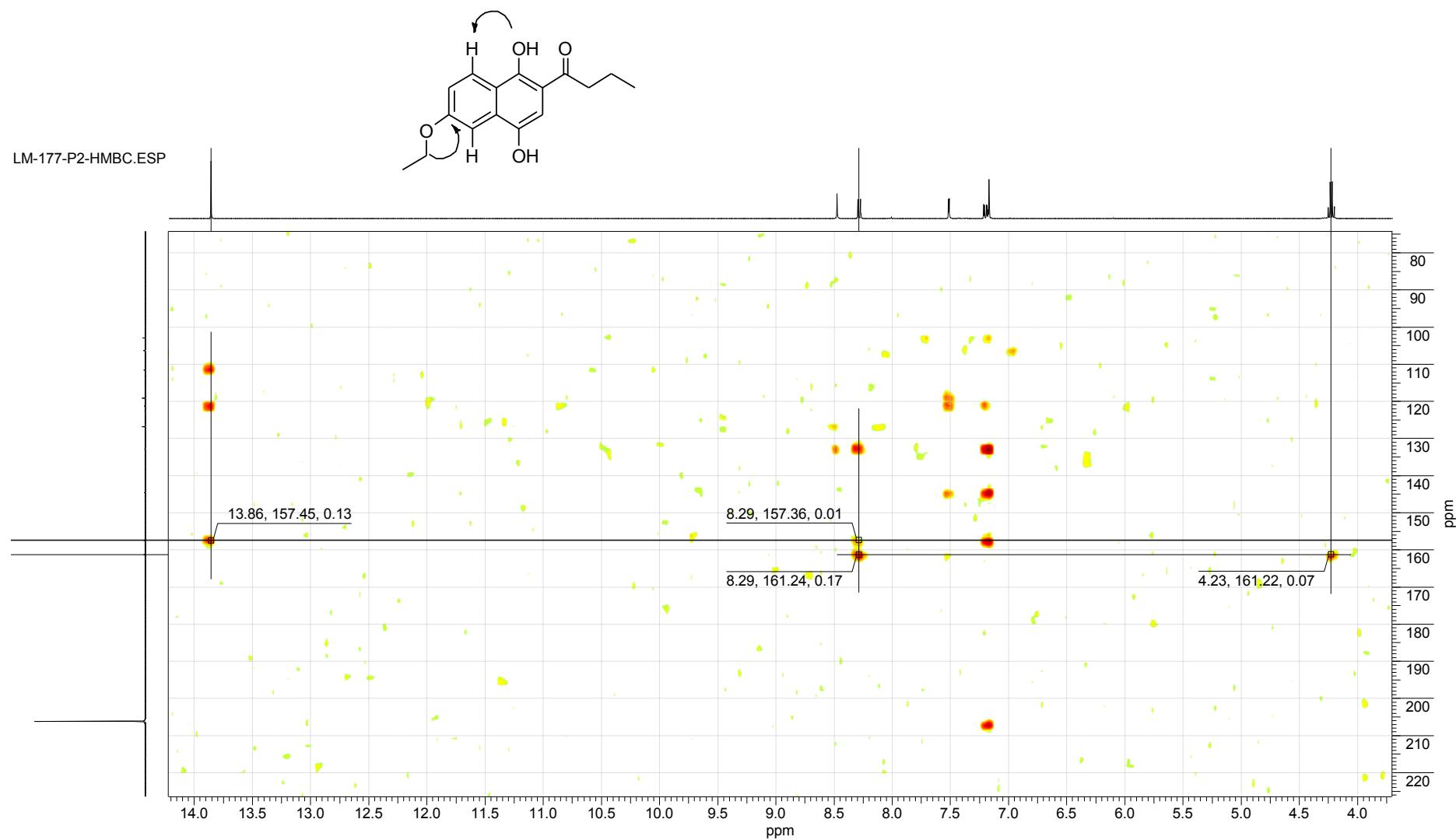
Key HMBC Interactions: 1-(5-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 9c



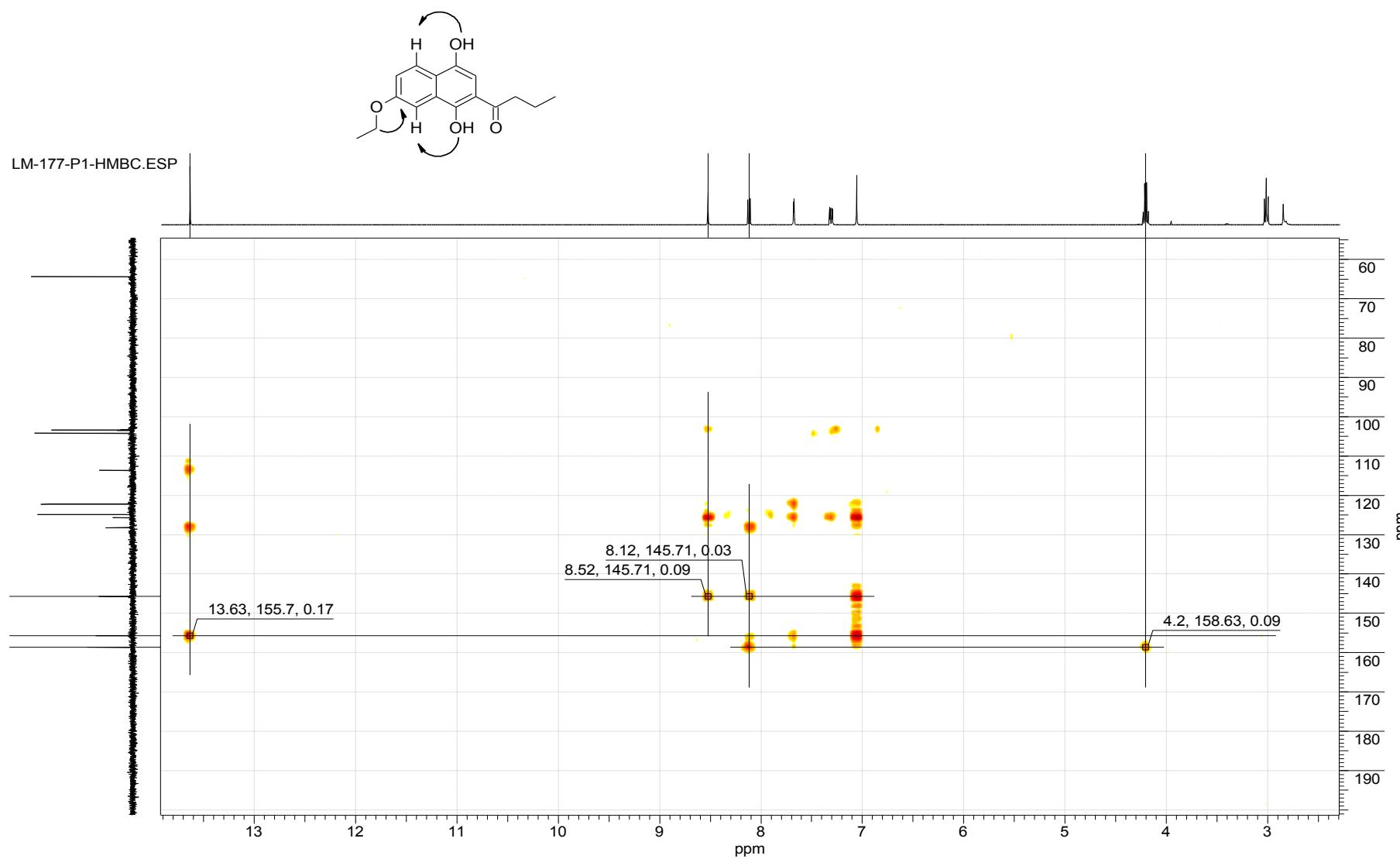
Key HMBC Interactions: 1-(8-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 10c



Key HMBC Interactions: 1-(6-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 13c

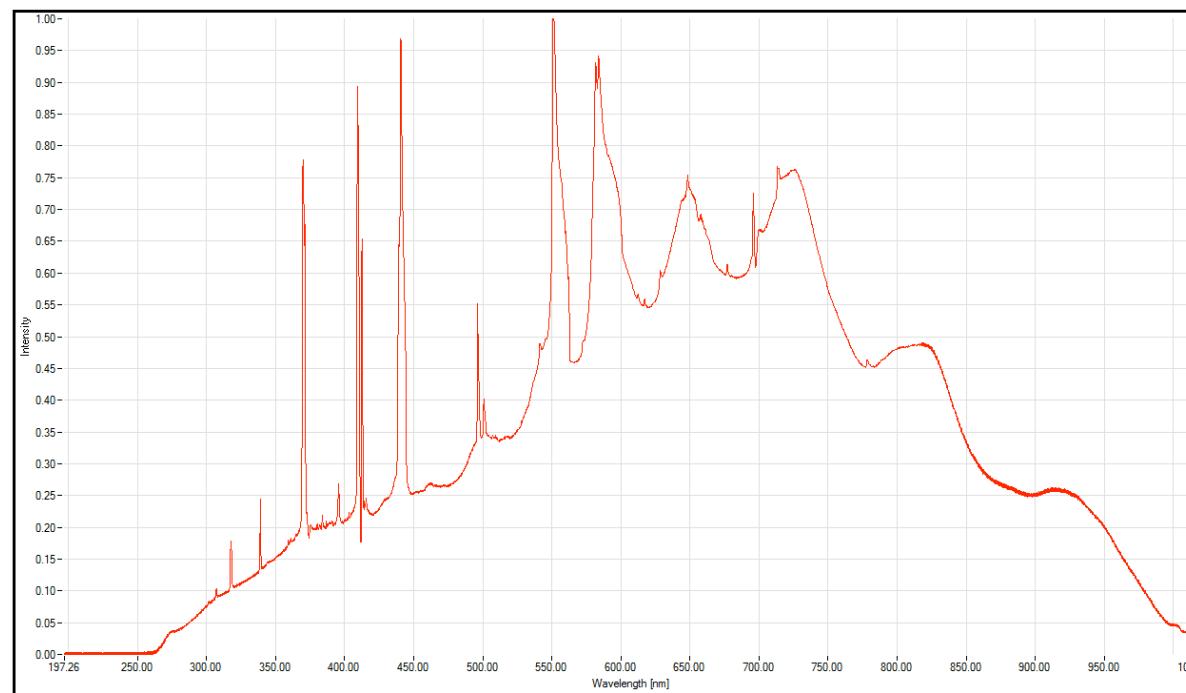


Key HMBC Interactions: 1-(7-Ethoxy-1,4-dihydroxynaphthalen-2-yl)butan-1-one 14c



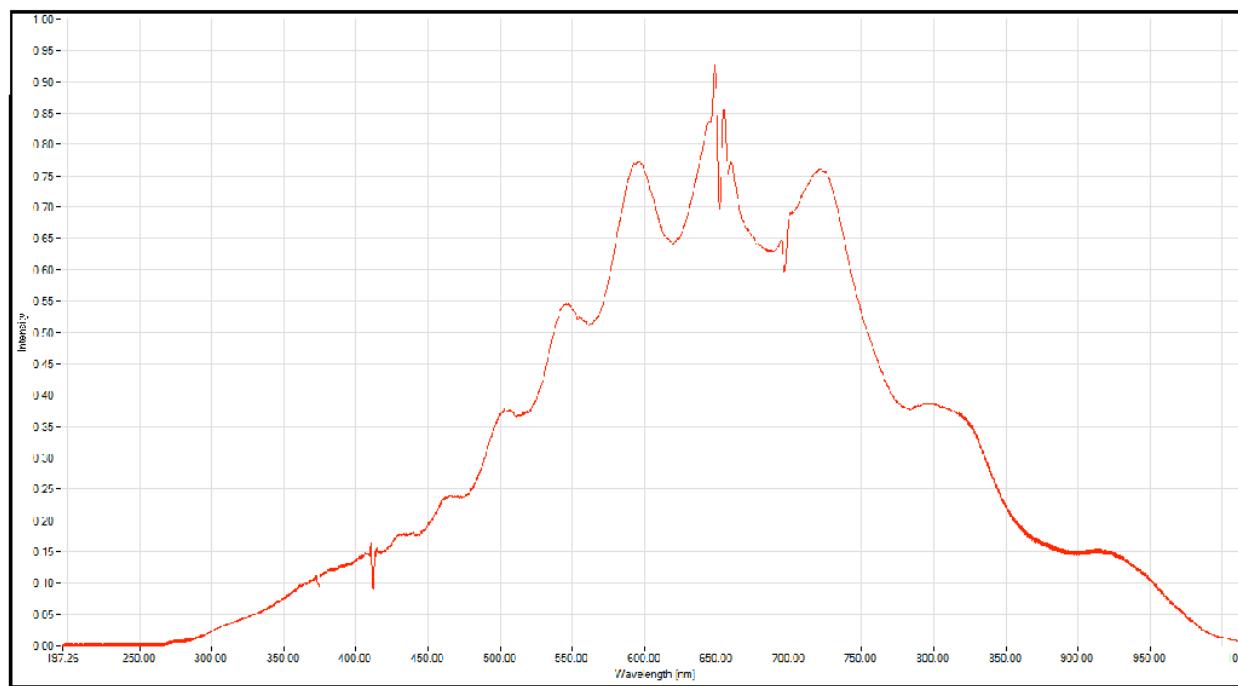
Wavelength Emission Spectra

Wavelength Emission Spectra of the UV Sunlamp (Table 2, Entry 3)



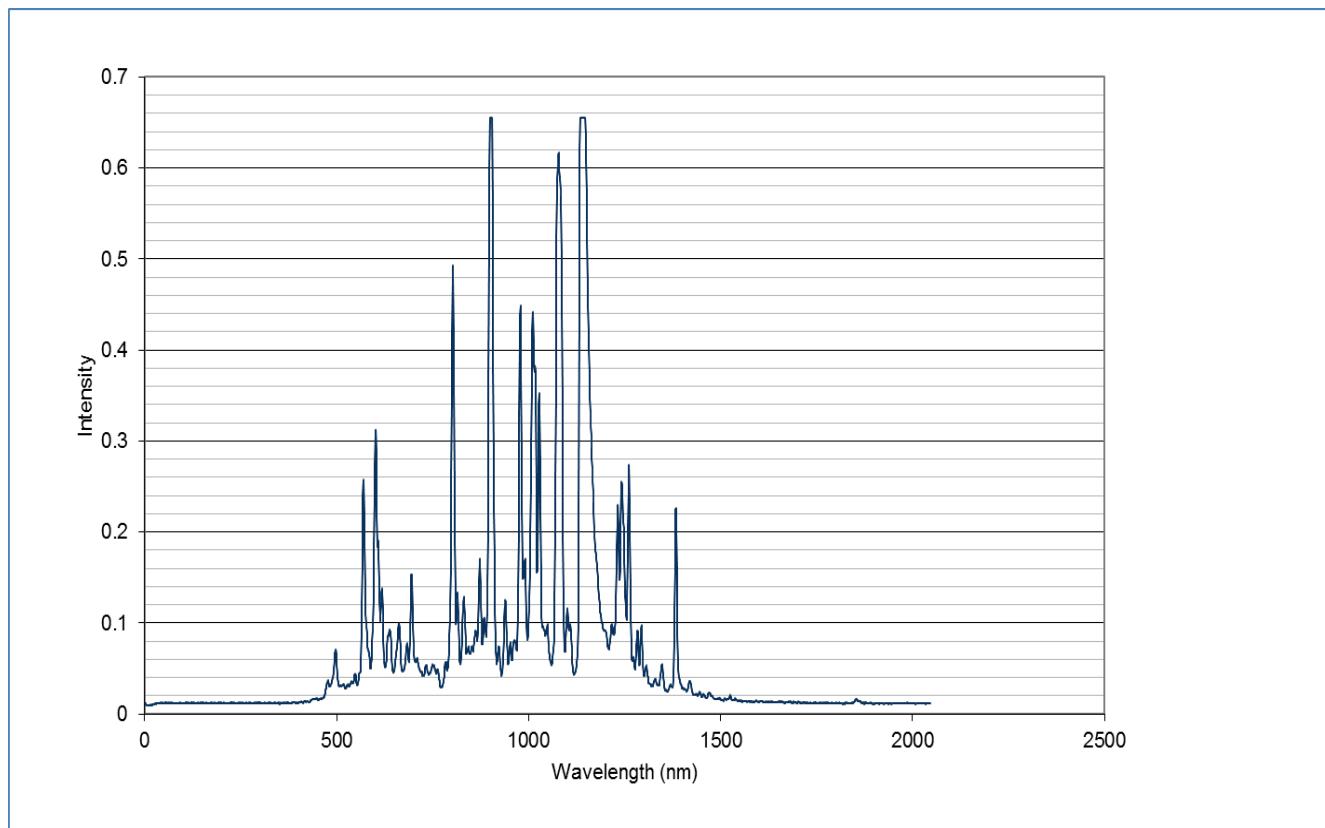
Measured using a compact CCD (Model: CCS150) spectrometer, produced by Thorlabs

Wavelength Emission Spectra of the Halogen Floodlight (Table 2, Entry 4)



Measured using a compact CCD (Model: CCS150) spectrometer, produced by Thorlabs

Wavelength Emission Spectra of the HQI-T HID Light (Table 2, Entry 5)



Measured using a USB Miniature Fiber Optic Spectrometer, produced by Ocean Optics.

References

1. S. A. Snyder, Z.-Y. Tang and R. Gupta, *J. Am. Chem. Soc.*, 2009, 131, 5744-5745.
2. M. Vuagnoux-d'Augustin and A. Alexakis, *Chem. Eur. J.*, 2007, 13, 9647-9662.
3. P. G. McDougal, J. G. Rico, Y. I. Oh and B. D. Condon, *J. Org. Chem.*, 1986, 51, 3388-3390.
4. C. U. Grünanger and B. Breit, *Angew. Chem. Int. Ed.*, 2010, 49, 967-970.
5. M. T. Gieseler and M. Kalesse, *Org. Lett.*, 2011, 13, 2430-2432.
6. L. F. Tietze, C. Güntner, K. M. Gericke, I. Schuberth and G. Bunkoczi, *Eur. J. Org. Chem.*, 2005, 2005, 2459-2467.
7. R. L. Hannan, R. B. Barber and H. Rapoport, *J. Org. Chem.*, 1979, 44, 2153-2158.
8. P. R. Nandaluru and G. J. Bodwell, *J. Org. Chem.*, 2012, 77, 8028-8037.
9. M. De Min, S. Croux, C. Tournaire, M. Hocquaux, B. Jackquet, E. Oliveros and M.-T. Maurette, *Tetrahedron*, 1992, 48, 1869-1882.
10. D. Cameron and M. Sidell, *Aust. J. Chem.*, 1978, 31, 1323-1333.