

α,ω -Di(glycerol carbonate) Telechelic Polyesters and Polyolefins

as Precursors to PolyHydroxyUrethanes: an Isocyanate-free approach

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Electronic Supplementary Information

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Figure S1. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs) (* stands for residual solvent resonances)

Figure S2. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs) (* stands for residual solvent resonances)

Figure S3. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectra of PPG_{400,1600,2800}-GC₂ prepared from the reaction of the corresponding PPG_{400,1600,2800}-OH₂ with GC-OTS.

Figure S4. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PPG₄₀₀-OH₂ (* stands for residual solvent resonances, and x stands for an unidentified impurity).

Figure S5. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PPG₄₀₀-GC₂.

Figure S6. ^1H - ^{13}C (DEPT) HMQC NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PPG₄₀₀-GC₂.

Figure S7. (a) ^1H (500 MHz, CDCl_3 , 25 °C) and (b) $^{13}\text{C}\{\text{H}\}$ (125 MHz, CDCl_3 , 25 °C) NMR spectra of PEG₄₀₀-OH₂.

Figure S8. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of the PEG₄₀₀-GC₂.

Figure S9. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of the PEG₄₀₀-GC₂.

Figure S10. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of the PEG₄₀₀-GC₂.

Figure S11. FTIR spectra of PEG₄₀₀-OH₂ (black trace) and the resulting PEG₄₀₀-GC₂ (red trace).

Figure S12. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of the PEE-OH₂.

Figure S13. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of the PEE-OH₂.

Figure S14. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PEE-GC₂ (* marker stands for residual toluene and ** for the starting reagent).

Figure S15. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of PEE-GC₂ (* marker stands for residual toluene).

Figure S16. ^1H - ^1H COSY NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PEE-GC₂.

Figure S17. ^1H - ^{13}C (DEPT) HMQC NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PEE-GC₂.

Figure S18. FTIR spectra of PEE-OH₂ (black trace) and the resulting PEE-GC₂ (red trace).

Figure S19. MALDI-ToF MS spectrum of PEE-GC₂.

Figure S20. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PBD-OH₂.

Figure S21. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PBD-GC₂ (* marker stands for residual toluene).

Figure S22. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of PBD-GC₂.

Figure S23. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PBD-GC₂.

Figure S24. FTIR spectra of PBD-OH₂ (black trace) and the resulting PBD-GC₂ (red trace).

Scheme S1. Synthesis of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs).

Table S1. α,ω -Dihydroxy and dicyclocarbonate telechelic PEGs characteristics.

Table S2. α,ω -Dihydroxy and dicyclocarbonate telechelic PEE characteristics.

Table S3. α,ω -Dihydroxy and dicyclocarbonate telechelic PBD characteristics.

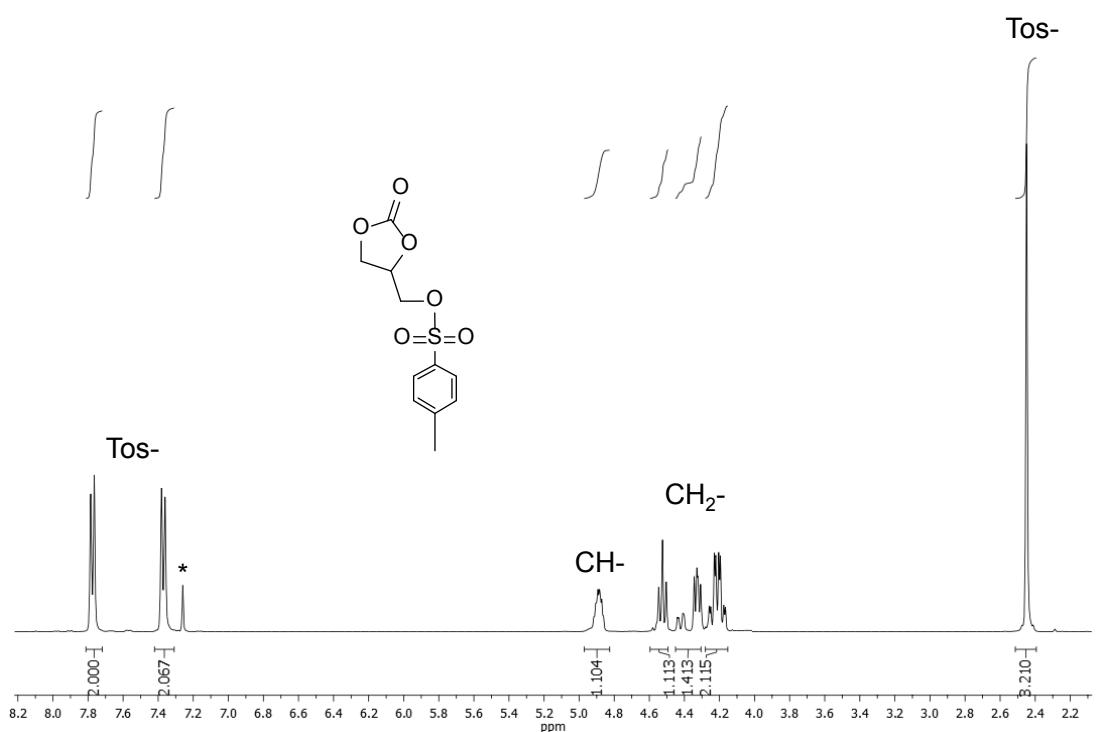


Figure S1. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs) (* stands for residual solvent resonances).

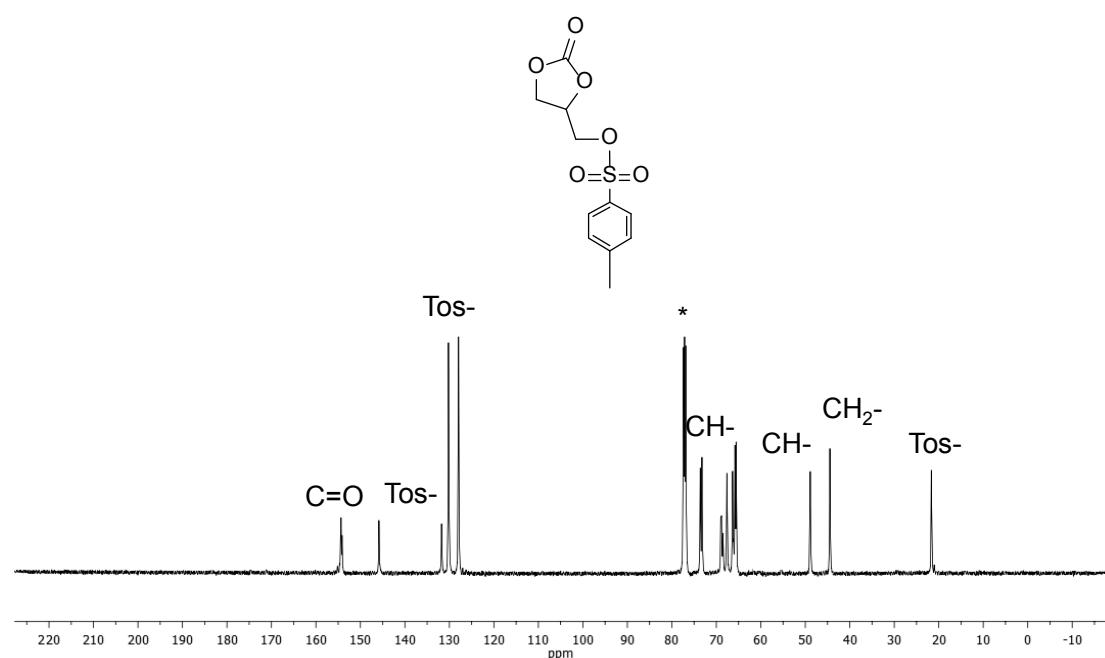


Figure S2. $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs) (* stands for residual solvent resonances).

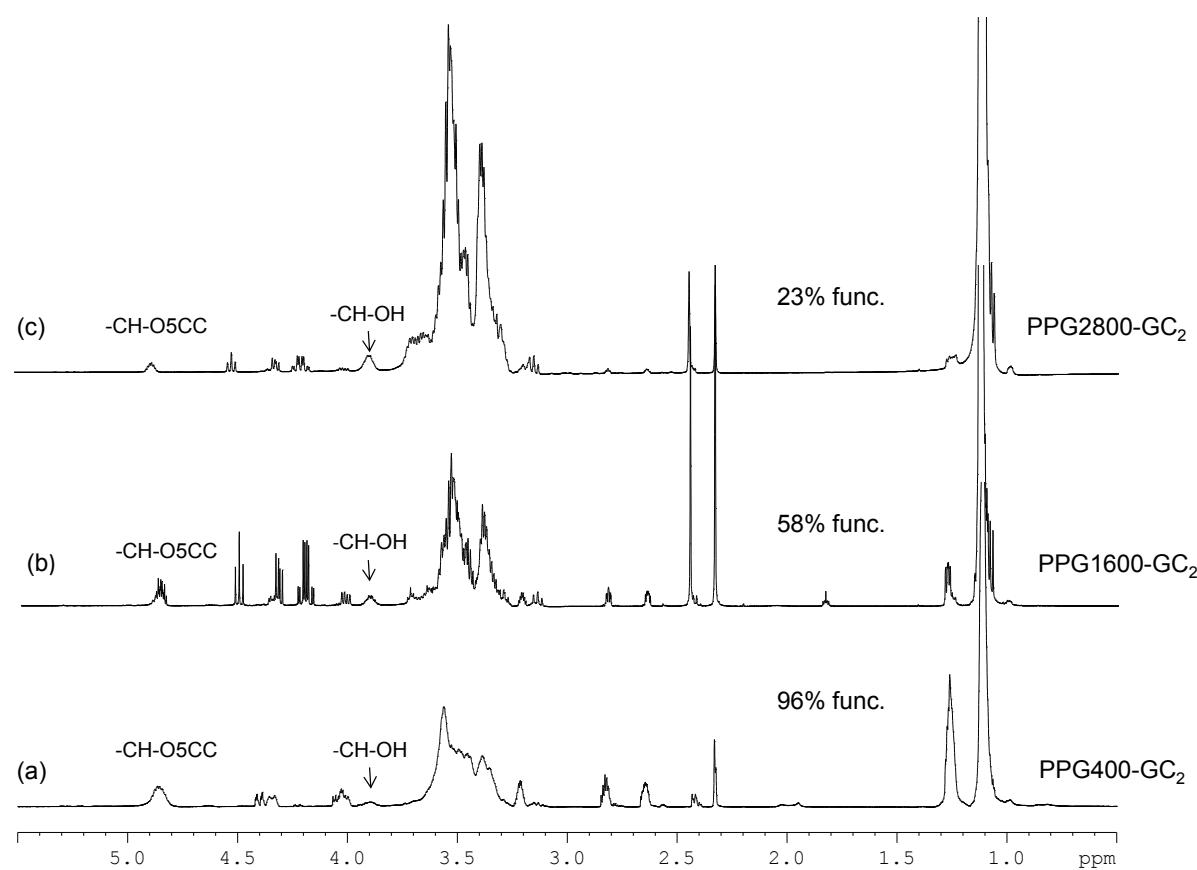


Figure S3. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectra of PPG_{400,1600,2800}-GC₂ prepared from the reaction of the corresponding PPG_{400,1600,2800}-OH₂ with GC-OTS.

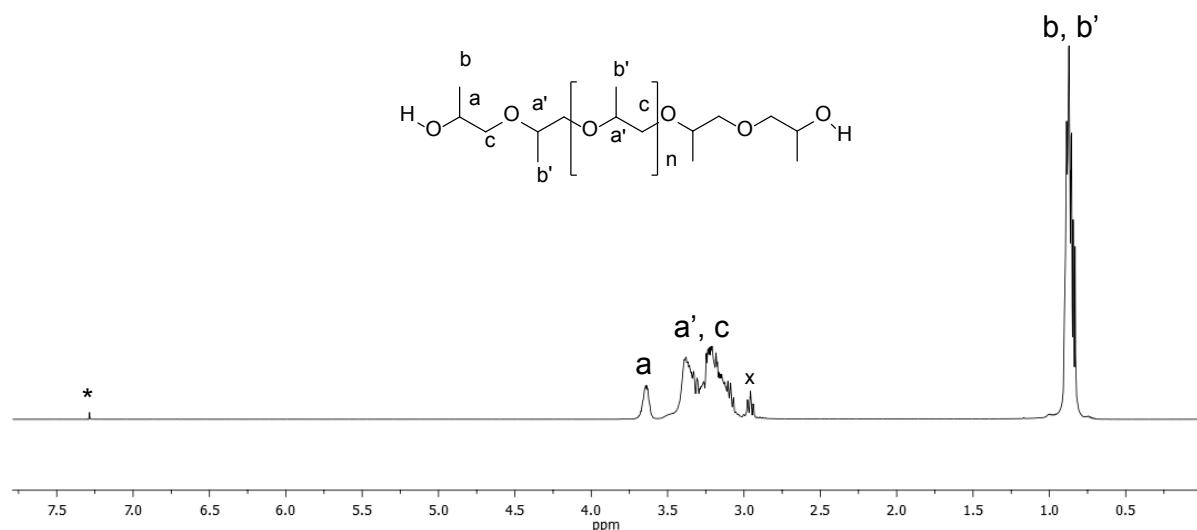


Figure S4. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PPG₄₀₀-OH₂ (* stands for residual solvent resonances, and x stands for an unidentified impurity).

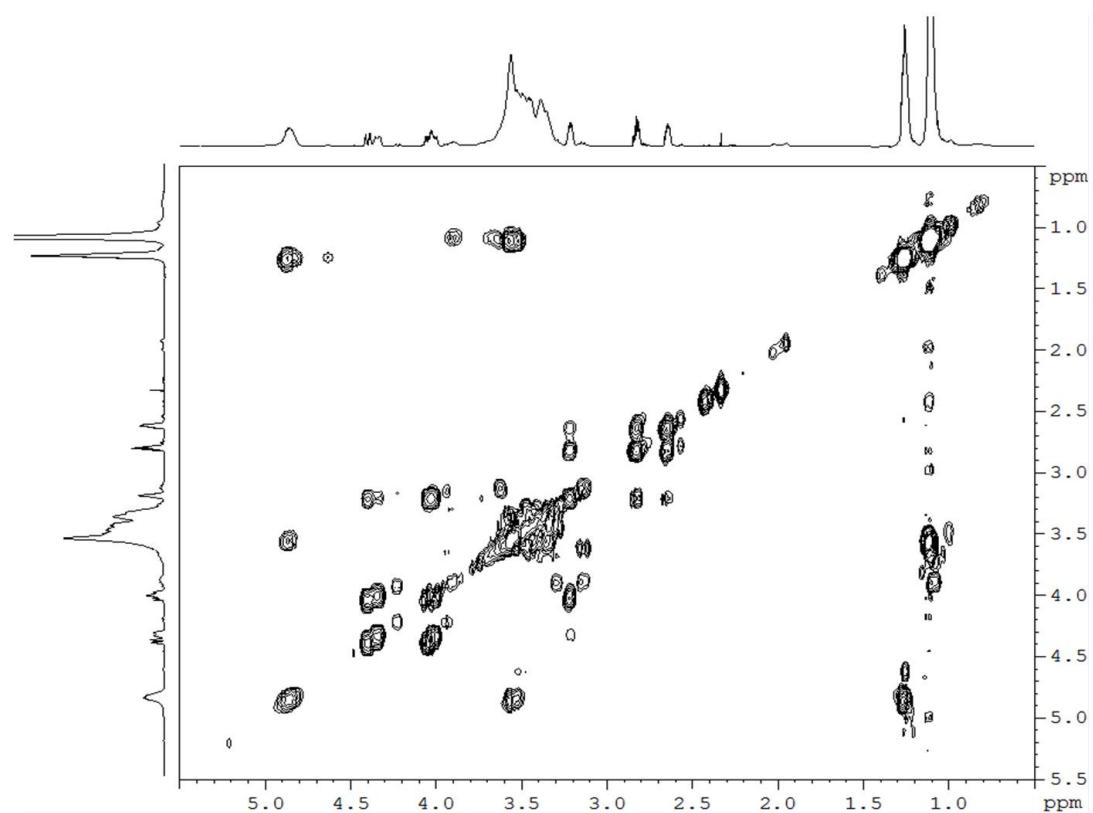


Figure S5. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PPG₄₀₀-GC₂.

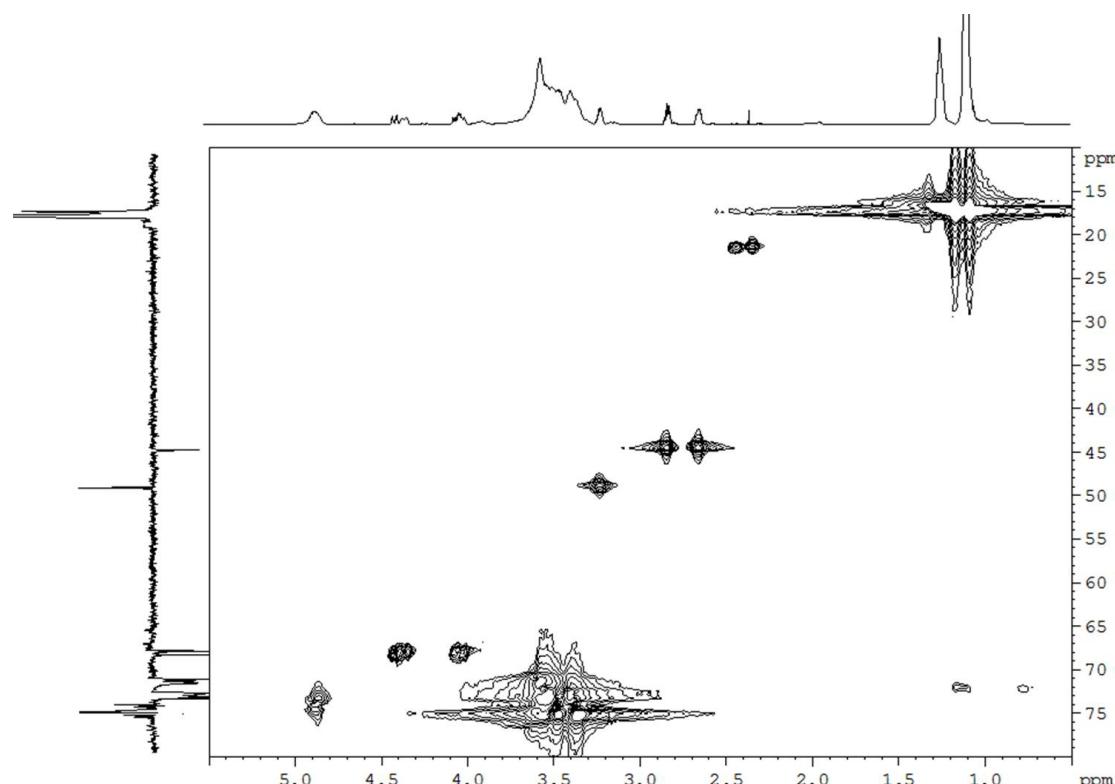


Figure S6. ^1H - ^{13}C (DEPT) HMQC NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PPG₄₀₀-GC₂.

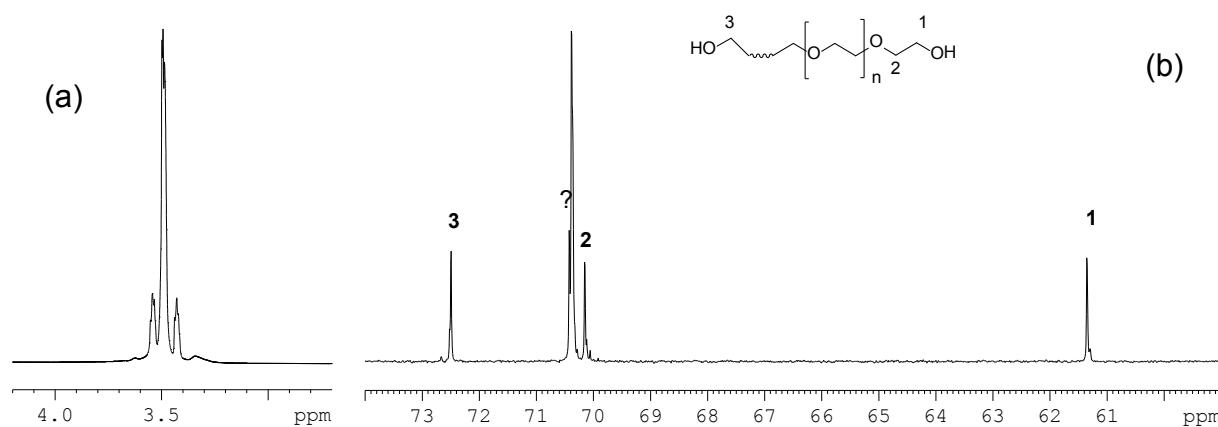


Figure S7. (a) ¹H (500 MHz, CDCl₃, 25 °C) and (b) ¹³C {¹H} (125 MHz, CDCl₃, 25 °C) NMR spectra of PEG₄₀₀-OH₂.

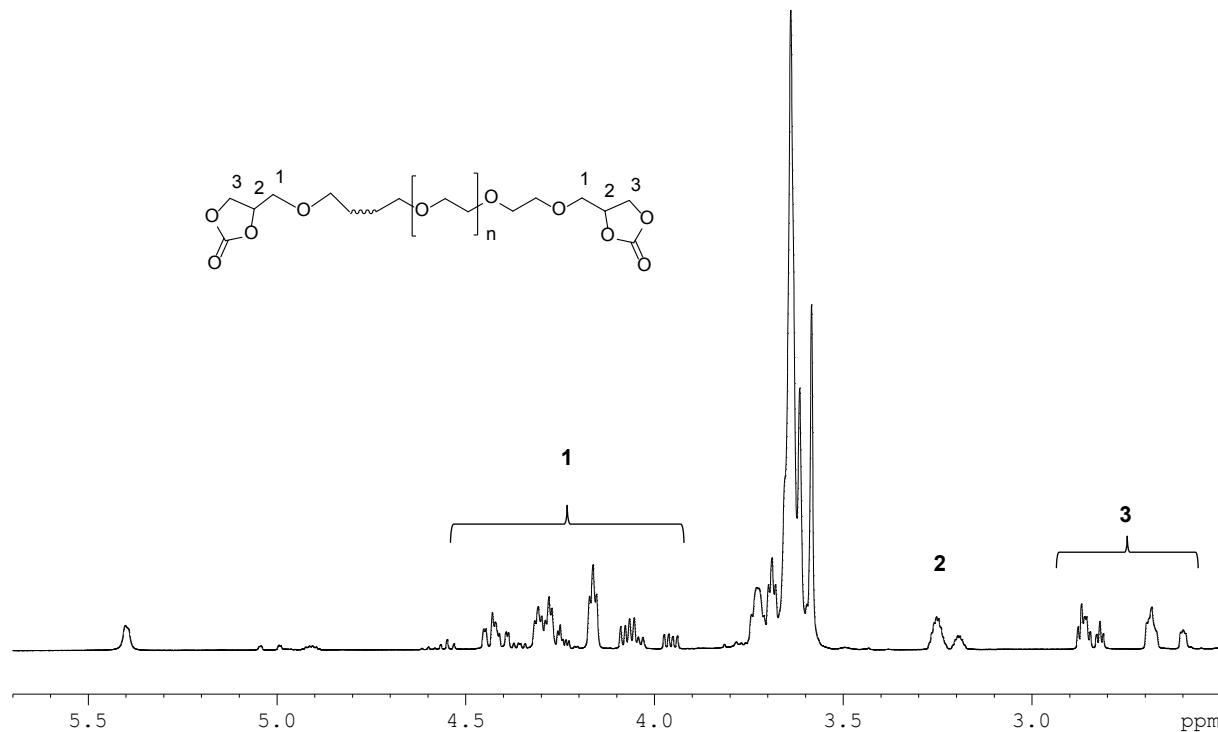


Figure S8. ¹H NMR (500 MHz, CDCl₃, 25 °C) spectrum of the PEG₄₀₀-GC₂.

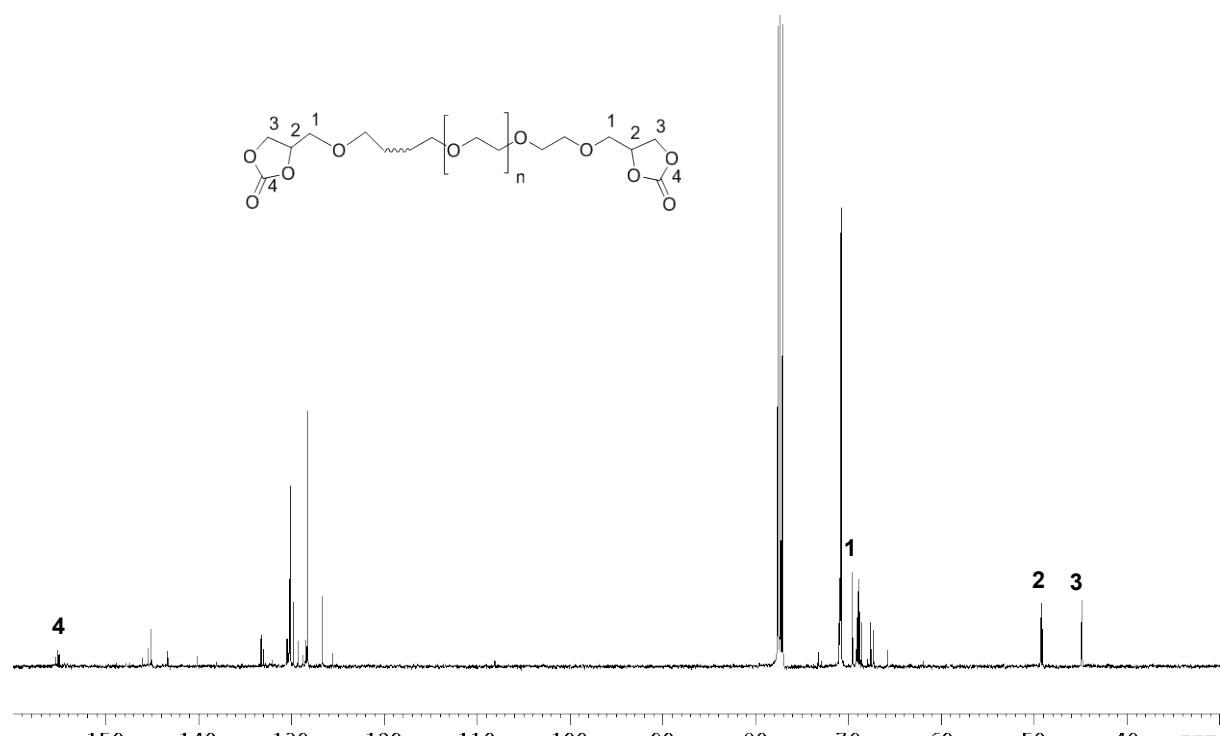


Figure S9. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of the PEG₄₀₀-GC₂.

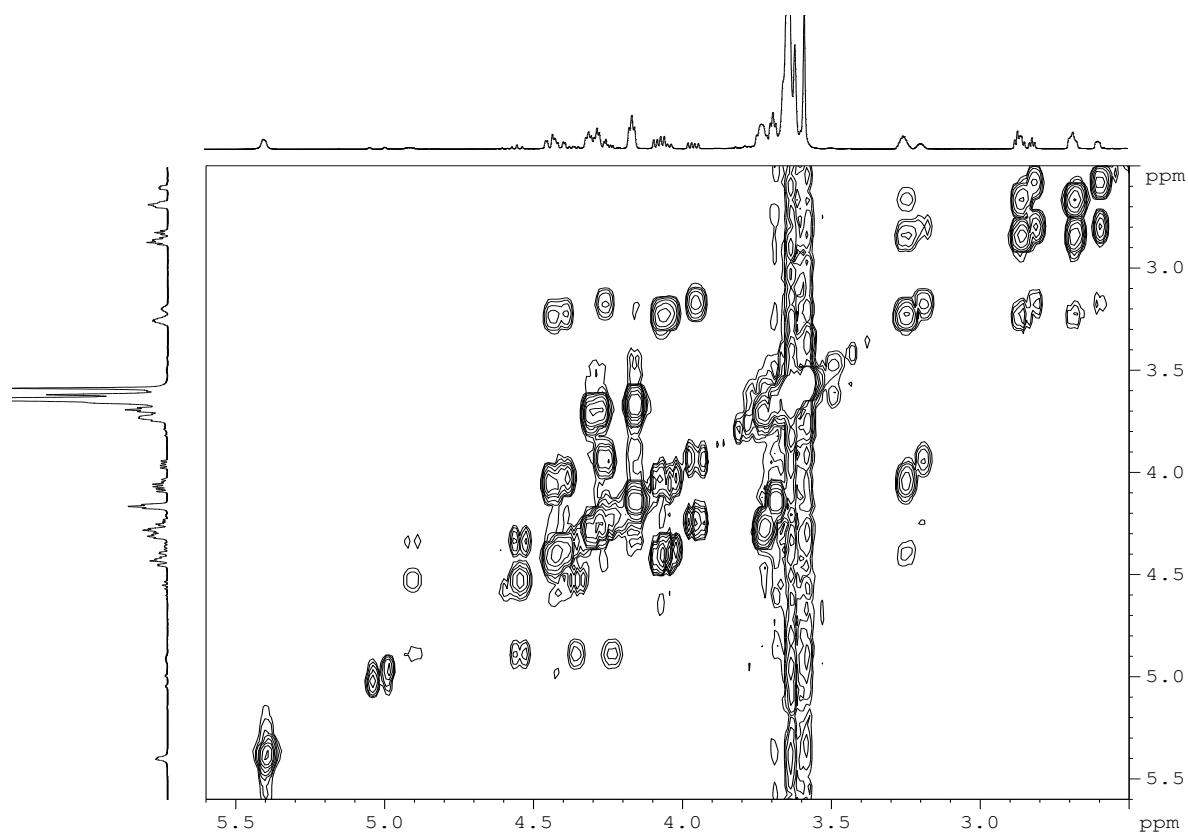


Figure S10. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of the PEG₄₀₀-GC₂.

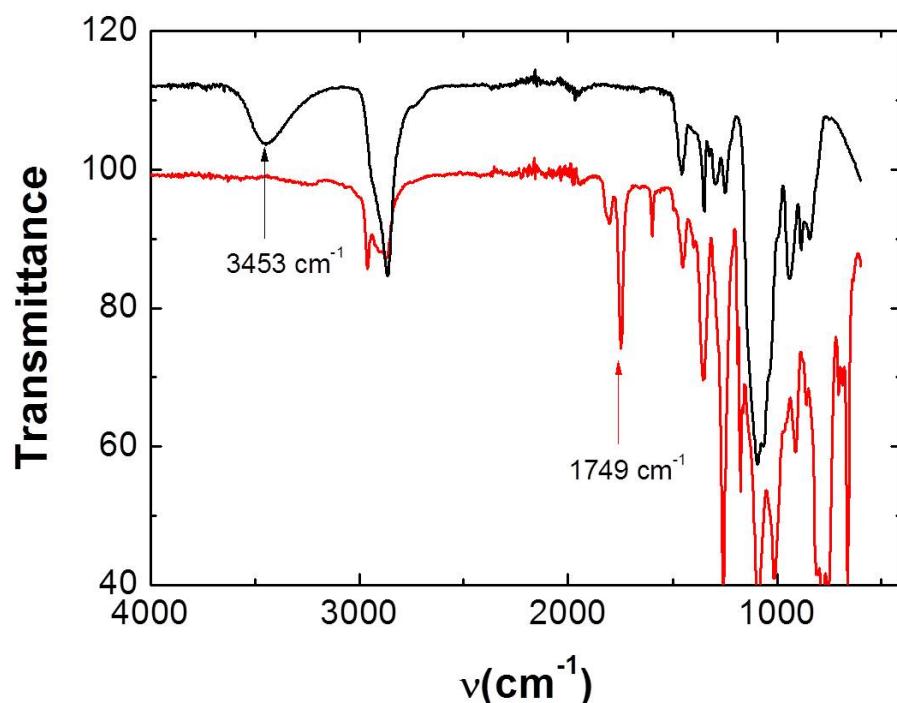


Figure S11. FTIR spectra of PEG₄₀₀-OH₂ (black trace) and the resulting PEG₄₀₀-GC₂ (red trace).

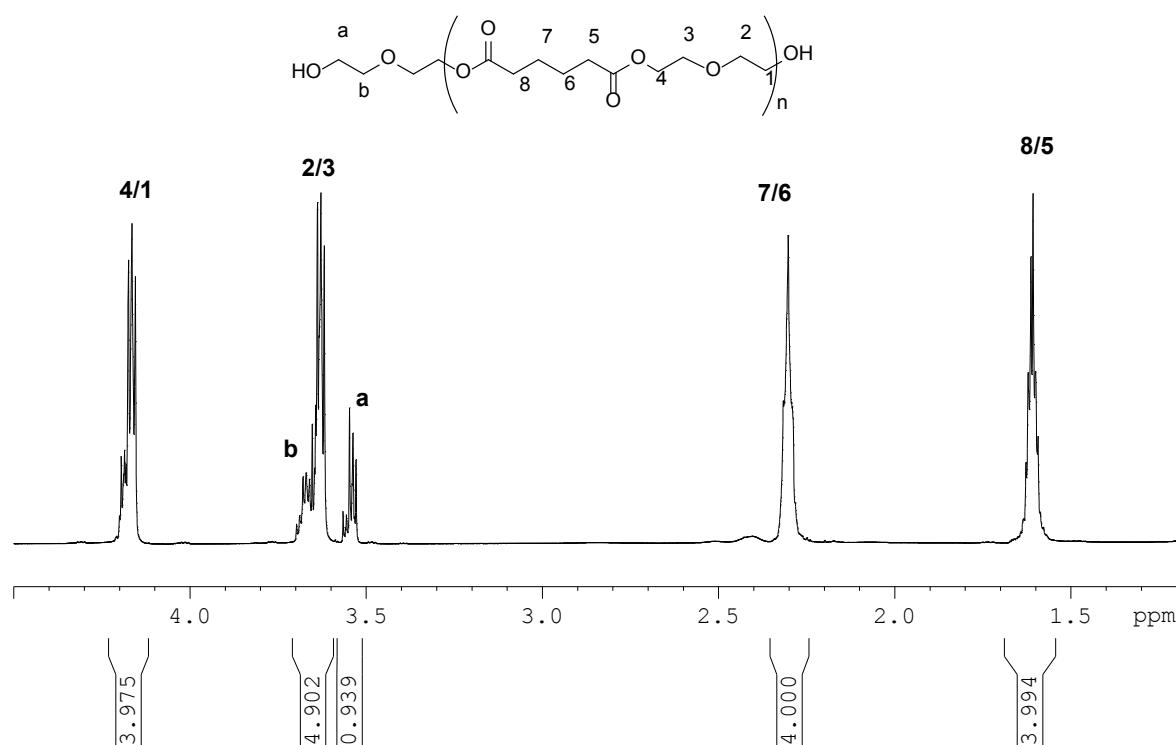


Figure S12. ¹H NMR (500 MHz, CDCl₃, 25 °C) spectrum of the PEE-OH₂.

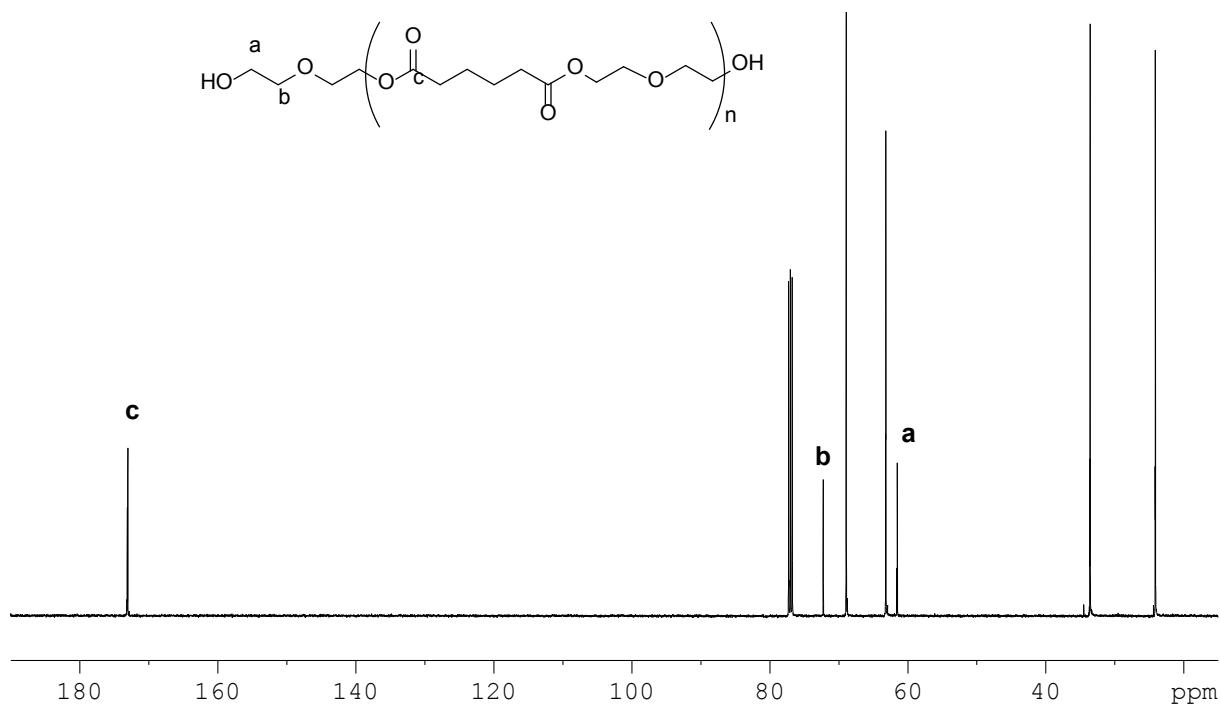


Figure S13. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of the PEE-OH₂.

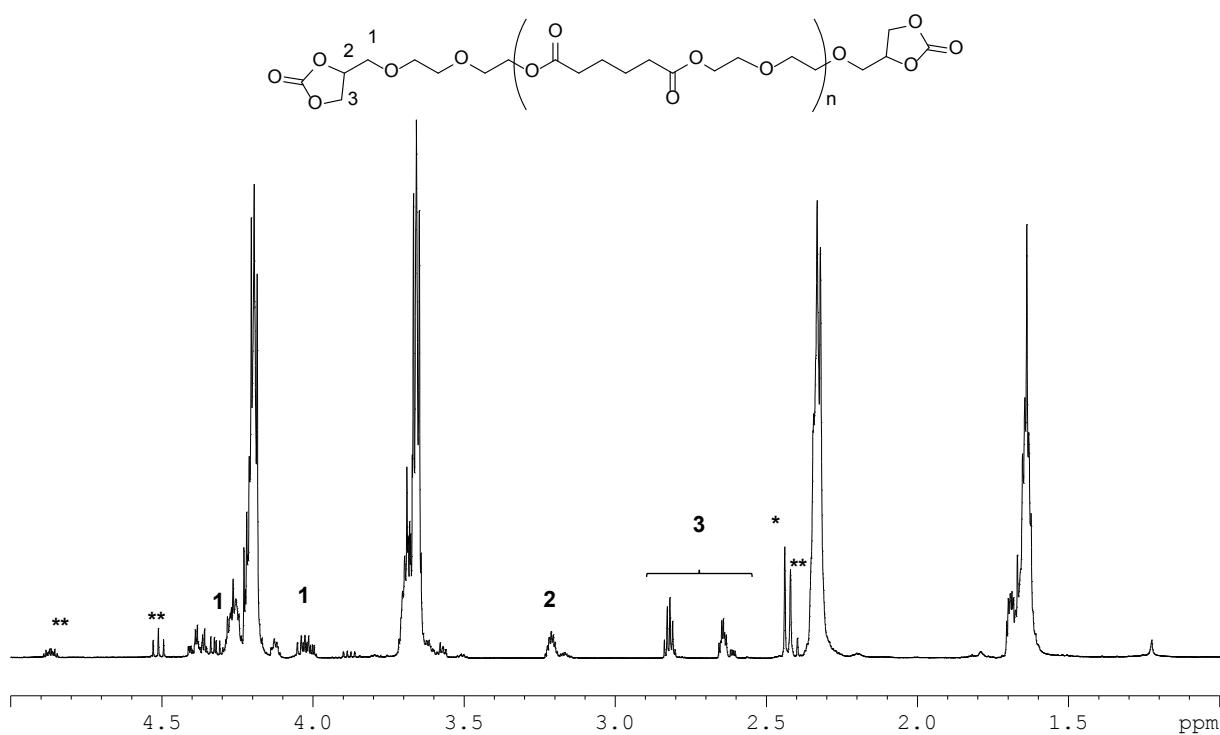


Figure S14. ^1H NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PEE-GC₂ (* marker stands for residual toluene and ** for the starting reagent).

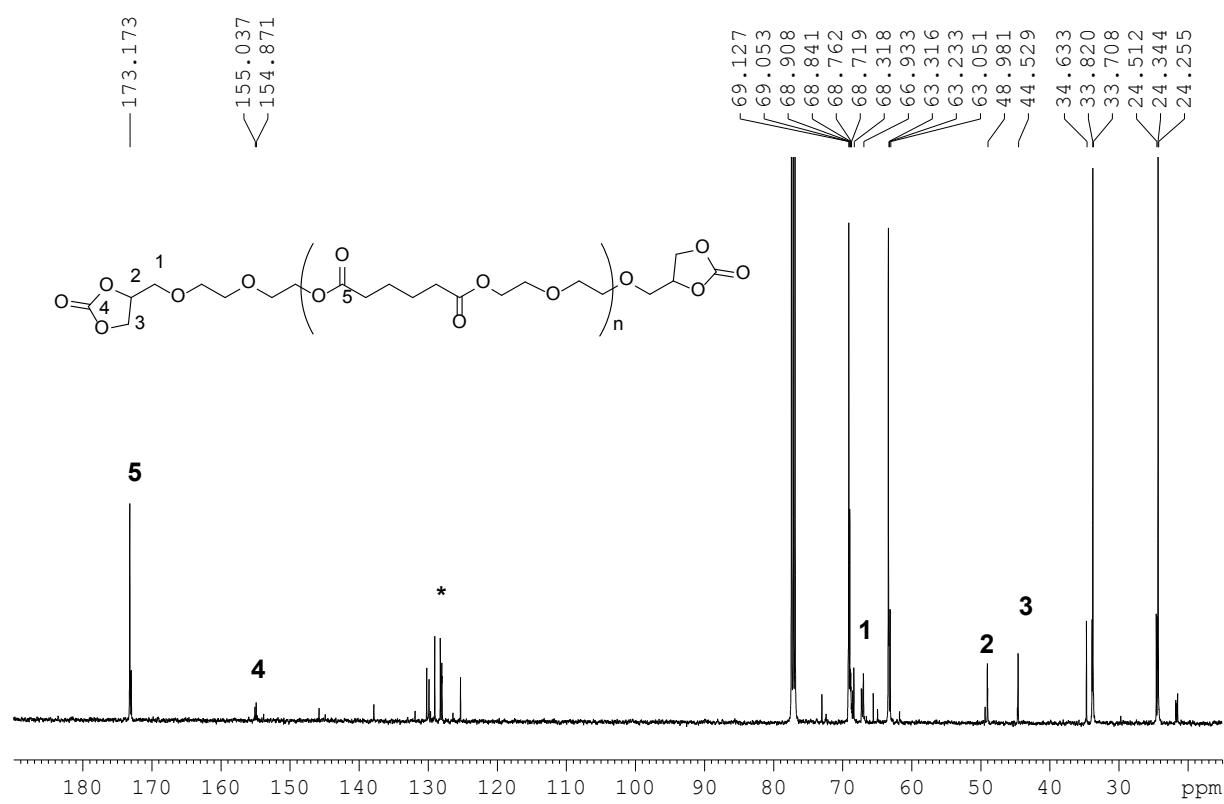


Figure S15. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of PEE-GC₂ (* marker stands for residual toluene).

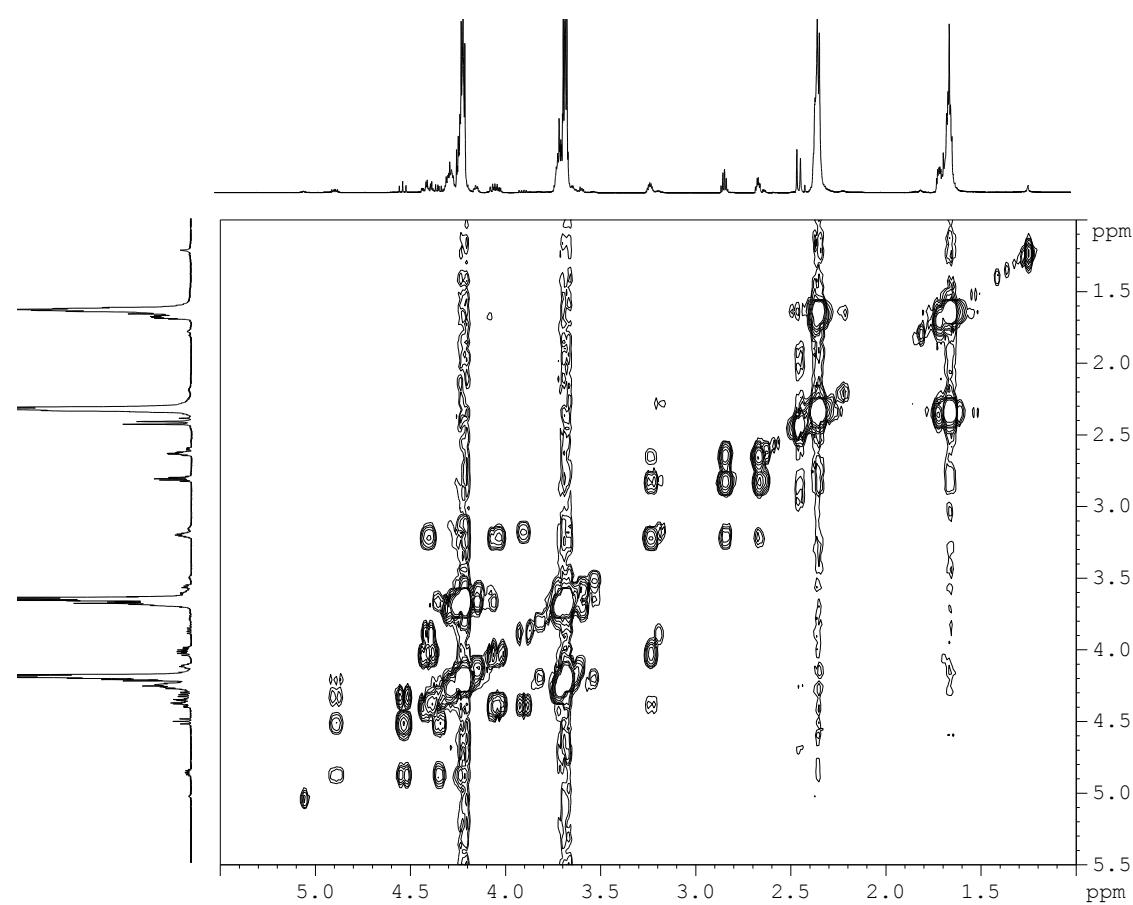


Figure S16. ^1H - ^1H COSY NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PEE-GC₂.

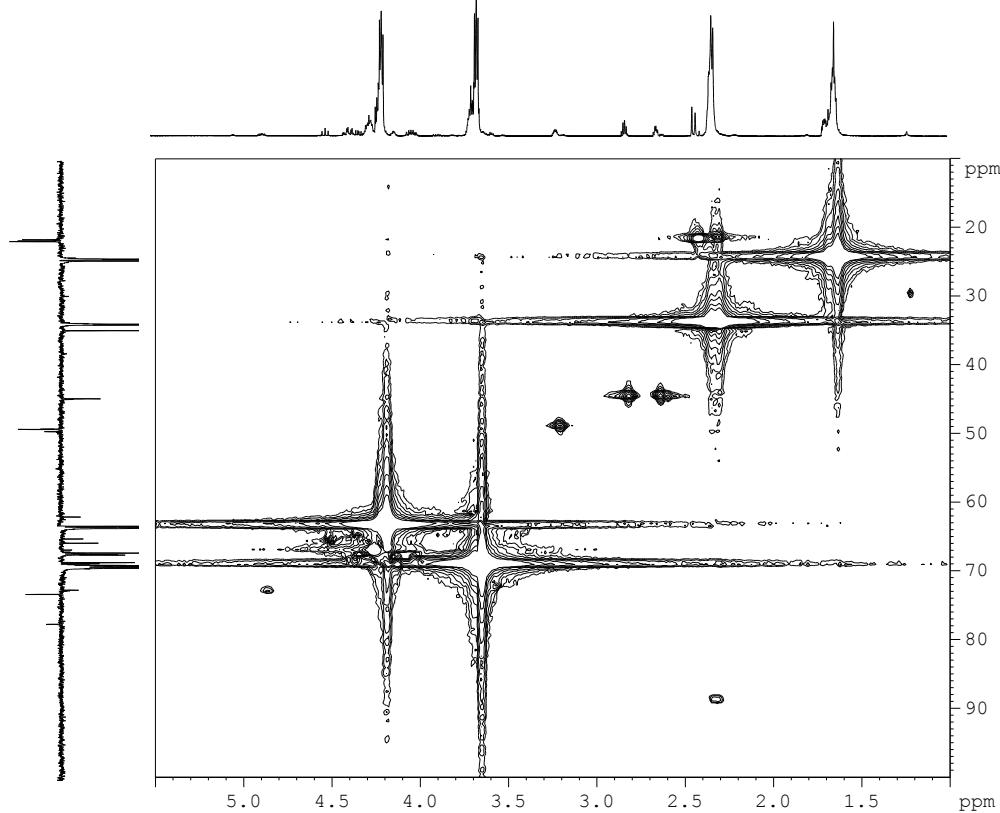


Figure S17. ^1H - ^{13}C (DEPT) HMQC NMR spectrum (500 MHz, CDCl_3 , 25 °C) of PEE-GC₂.

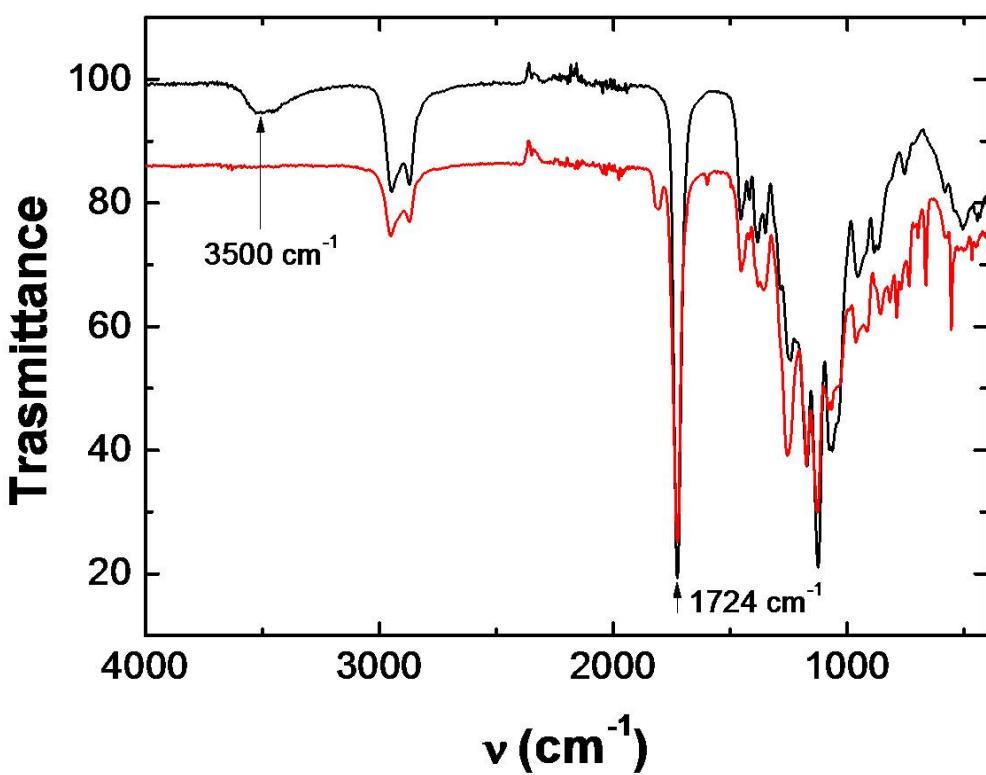


Figure S18. FTIR spectra of PEE-OH₂ (black trace) and the resulting PEE-GC₂ (red trace).

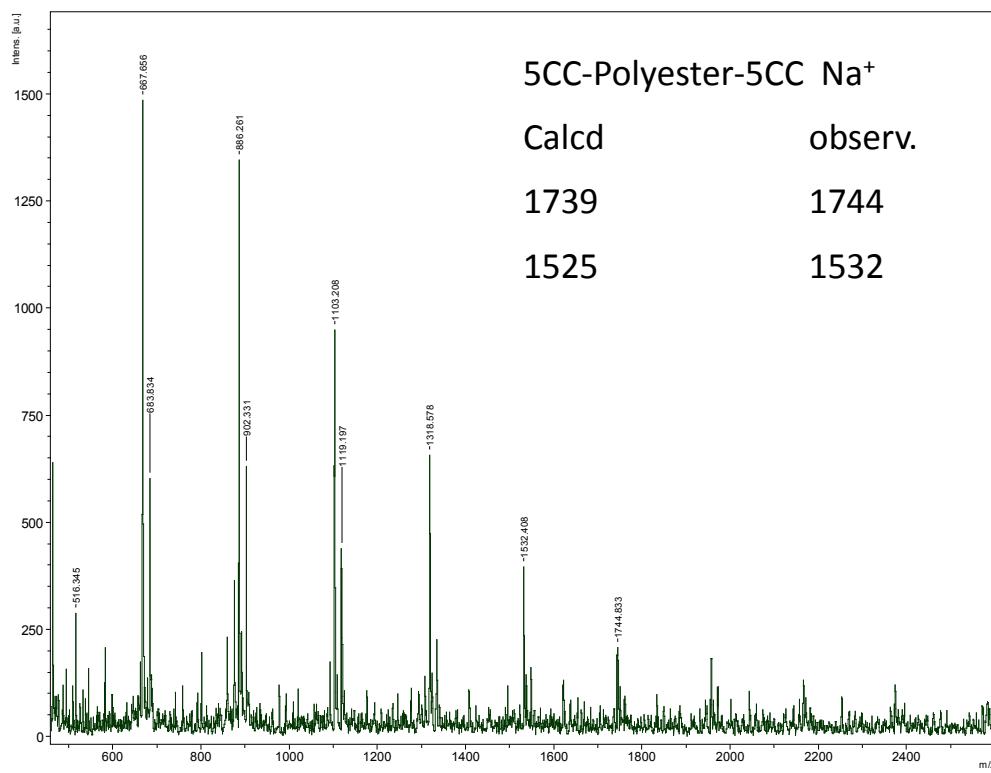


Figure S19. MALDI-ToF MS spectrum of PEE-GC₂.

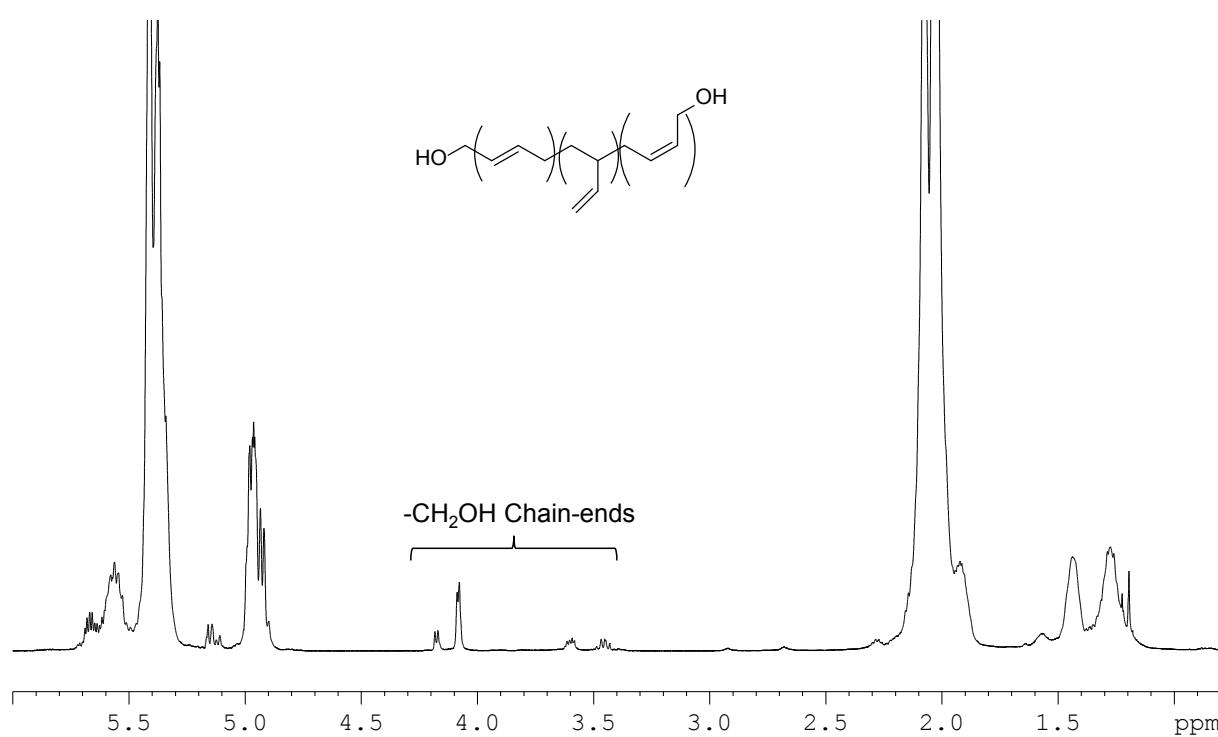


Figure S20. ¹H NMR (500 MHz, CDCl₃, 25 °C) spectrum of PBD-OH₂.

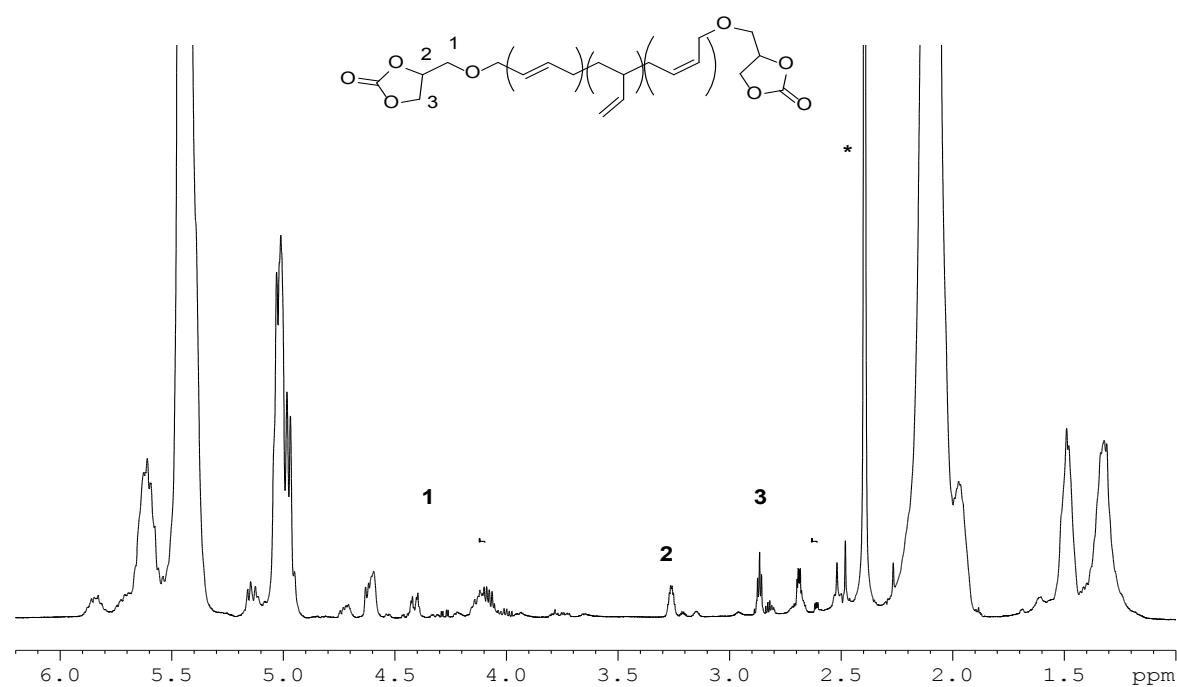


Figure S21. ¹H NMR (500 MHz, CDCl₃, 25 °C) spectrum of PBD-GC₂ (* marker stands for residual toluene).

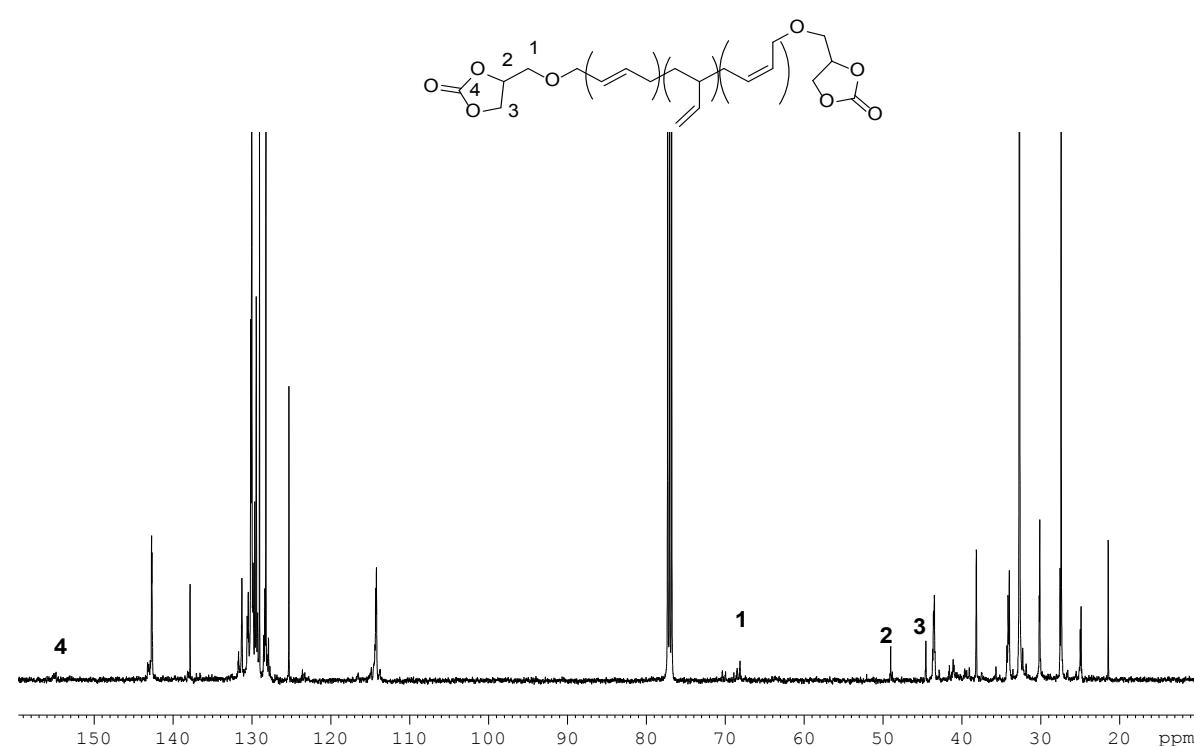


Figure S22. $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, CDCl_3 , 25 °C) spectrum of PBD-GC₂.

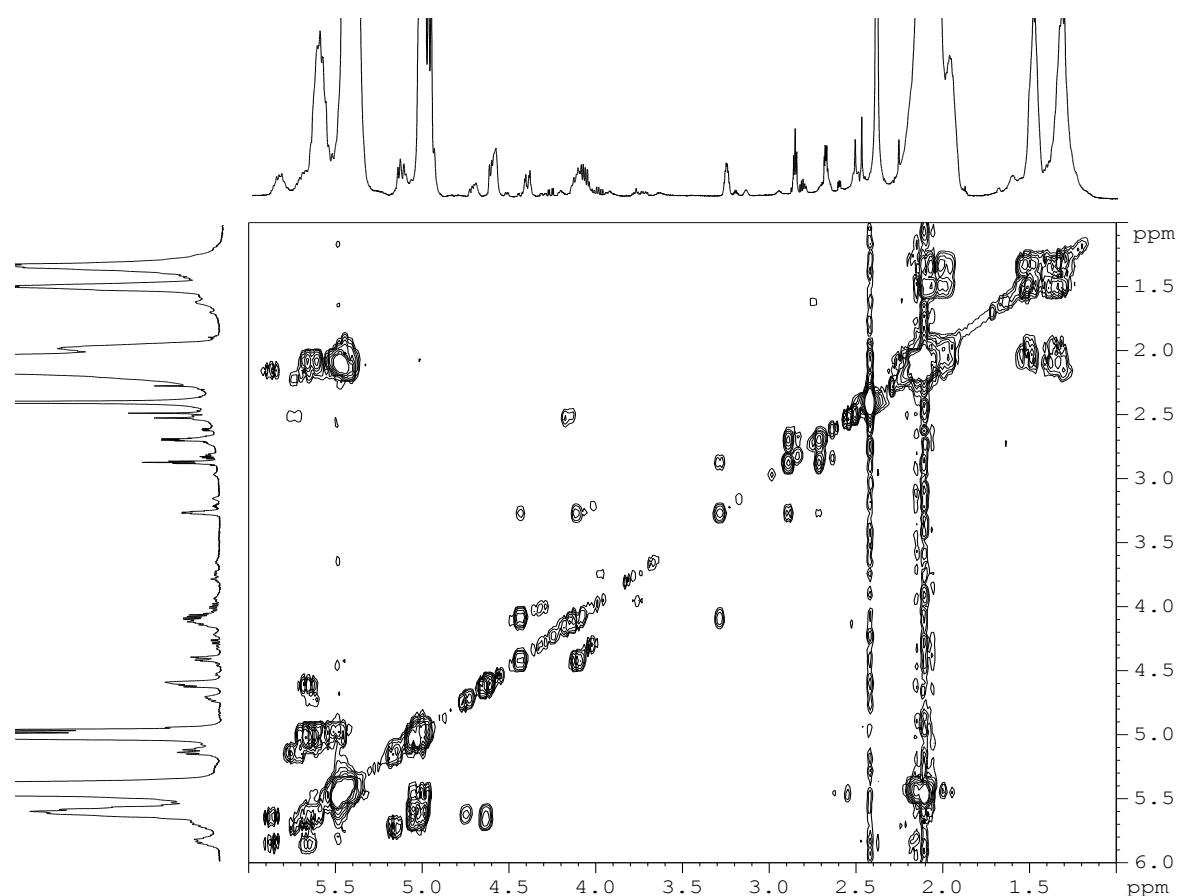


Figure S23. ^1H - ^1H COSY NMR (500 MHz, CDCl_3 , 25 °C) spectrum of PBD-GC₂.

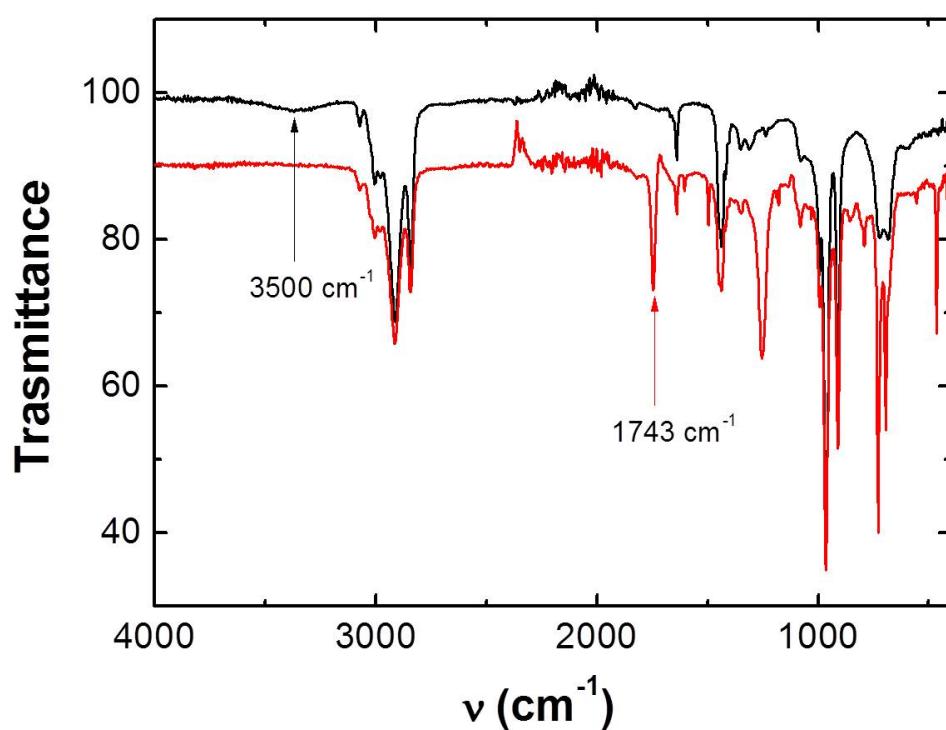


Figure S24. FTIR spectra of PBD-OH₂ (black trace) and the resulting PBD-GC₂ (red trace).

Scheme S1. Synthesis of 4-tosylmethyl-1,3-dioxolan-2-one (GC-OTs).

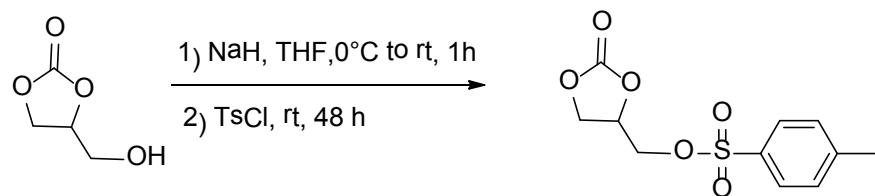


Table S1. α,ω -Dihydroxy and dicyclocarbonate telechelic PEGs characteristics.

	$M_{n,\text{SEC}}^{\text{b}}$	D_M^{b}
PEG ₄₀₀ -OH ₂	-	-
PEG ₄₀₀ -GC ₂	-	-
PEG ₄₀₀₀ -OH ₂	3950	1.10
PEG ₄₀₀₀ -GC ₂	4400	1.18

^a Determined by SEC in THF at 30 °C vs. polystyrene standards (uncorrected M_n values).

Table S2. α,ω -Dihydroxy and dicyclocarbonate telechelic PEE characteristics.

	$M_{n,NMR}^a$	$M_{n,SEC}^b$	D_M^b
PEE-OH ₂	1000	1040	2.24
PEE-GC ₂	1200	1090	2.14

^a Determined by NMR analysis of the isolated polymer, from ¹H resonances of both terminal groups

^b Determined by SEC in THF at 30 °C vs. polystyrene standards (uncorrected M_n values).

Table S3. α,ω -Dihydroxy and dicyclocarbonate telechelic PBD characteristics.

	$M_{n,SEC}^a$	D_M^a	% 1,4-cis units	% 1,4-trans units	% 1,2 units
PBD-OH ₂	3450	2.4	20.0	60.0	20.0
PBD-GC ₂	3800	2.29	20.0	60.0	20.0

^a Determined by SEC in THF at 30 °C vs. polystyrene standards (uncorrected M_n values).