

Supporting information

Superbase/cellulose: An environmentally benign catalyst for chemical fixation of carbon dioxide into cyclic carbonates

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i) Catalyst recycle experiments

Next, a series of catalytic cycles were run to investigate the constancy of the catalyst for the cycloaddition reaction of PO (30 mL each time). In each cycle, the catalyst was recovered via distillation under vacuum, and then used for the next cycle. The results (Fig. S1) showed that the catalyst could be reusable for at least up to four times with high activity, while the selectivity of the product remains the same. The decrease in conversion is associated with the inevitable loss of catalyst during the catalyst recovery operation. As a rough calculation, a near 90% recover yield of catalyst was obtained after the 4th -reuse.

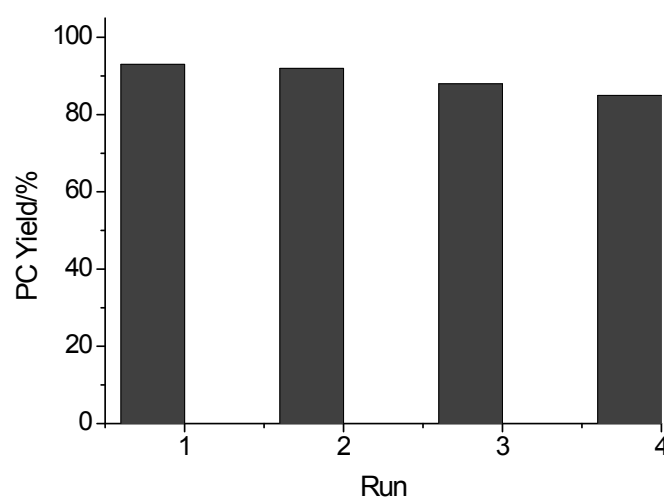


Fig. S1. Catalyst recycling for cycloaddition reaction.

Reaction condition: PO (30 mL, 25 °C), DBU 6.75 g, cellulose 0.45 g, 120 °C, 2.0 MPa, 2 h.

ii) Characterization of catalyst before and after use

XRD

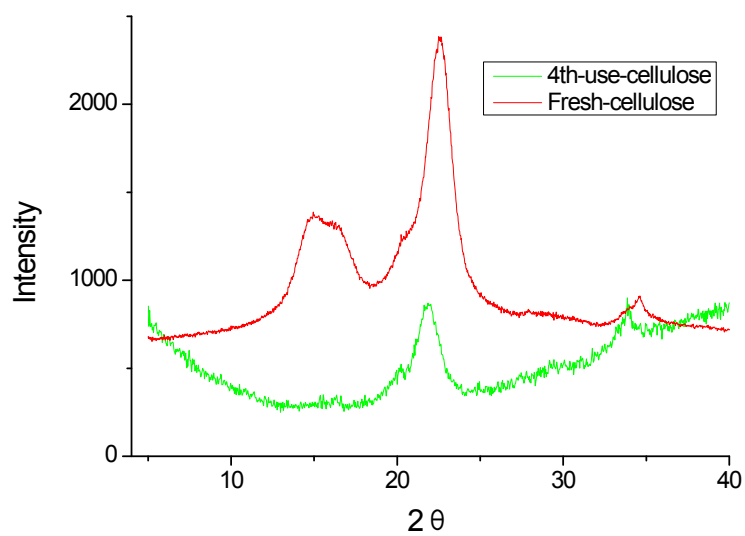


Fig. S2. XRD of fresh and recycled cellulose

TGA

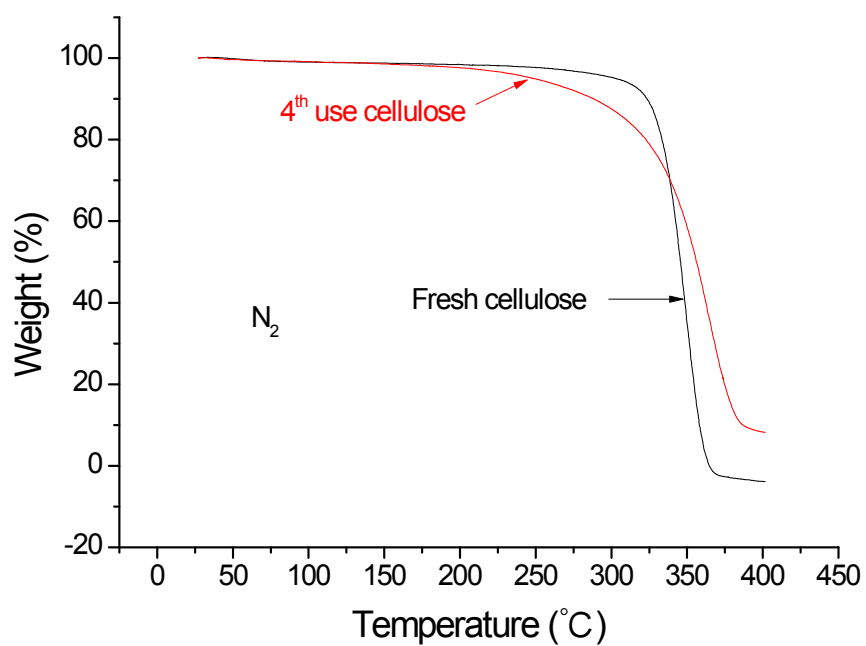


Fig. S3 TGA curves of fresh and recycled cellulose in the atmosphere of N₂.

SEM

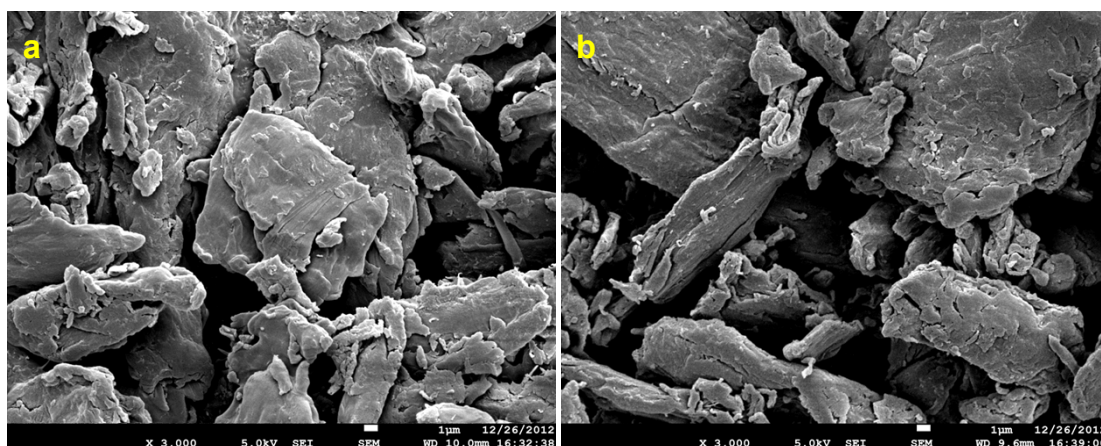


Fig. S4. SEM spectra of fresh cellulose (a) and 4th-reuse cellulose (b)

LC

Qualitative analysis procedure: After the reaction, un-reacted propylene oxide and the remaining CO₂ were evaporated from the obtained liquid samples. Remaining portion was then treated with dilute H₂SO₄ to reach a pH value of 2, and analyzed by liquid chromatography with a Waters e2695 separations module under the following conditions: column, bio-rad 87H, temperature, 65 °C; flow-rate, 0.4 mL/min; eluent, dilute H₂SO₄ (pH=2); detector, Waters refractive index detector (RID), temperature, 50 °C. Figure S5 showed the liquid chromatography detection result of product after 12 hours reaction at 120 °C, 2 MPa.

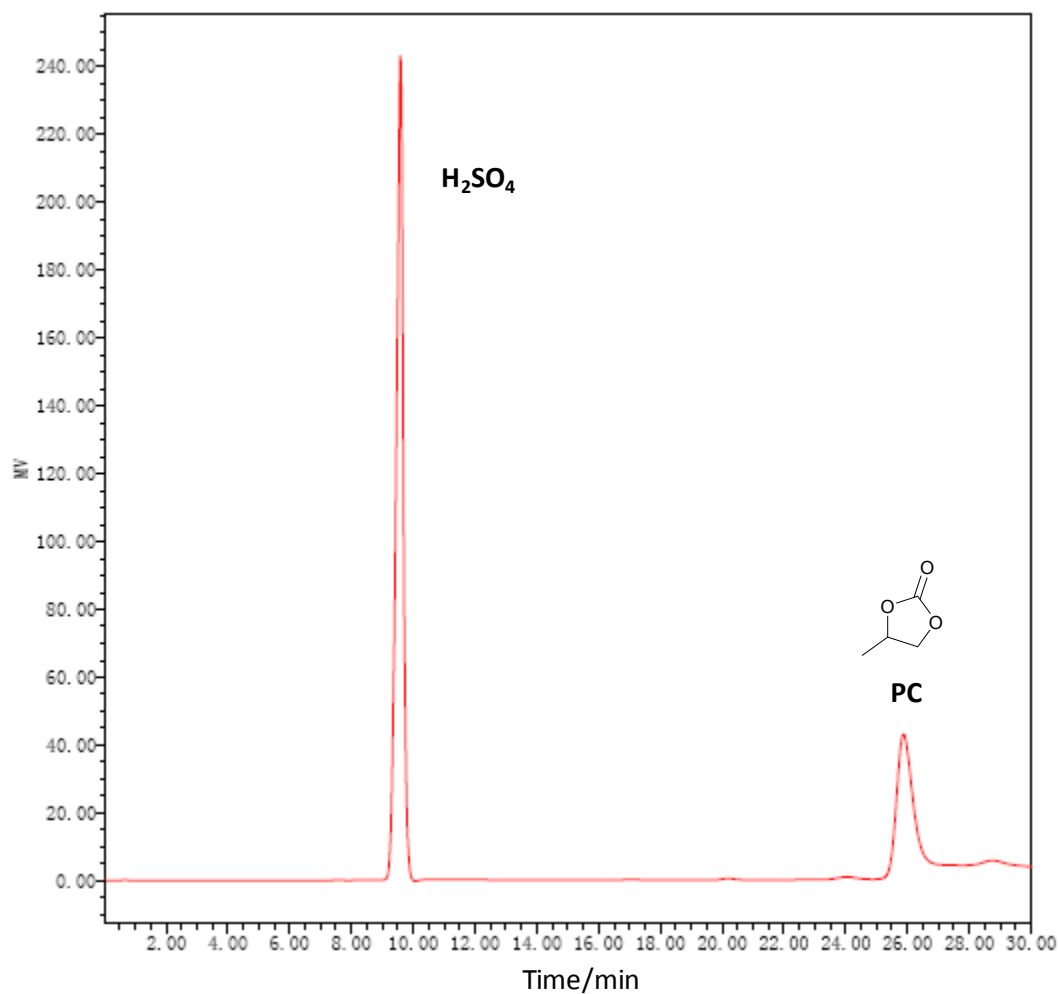


Fig. S5 Liquid chromatography detection result of liquid product after 12 hours reaction.

NMR

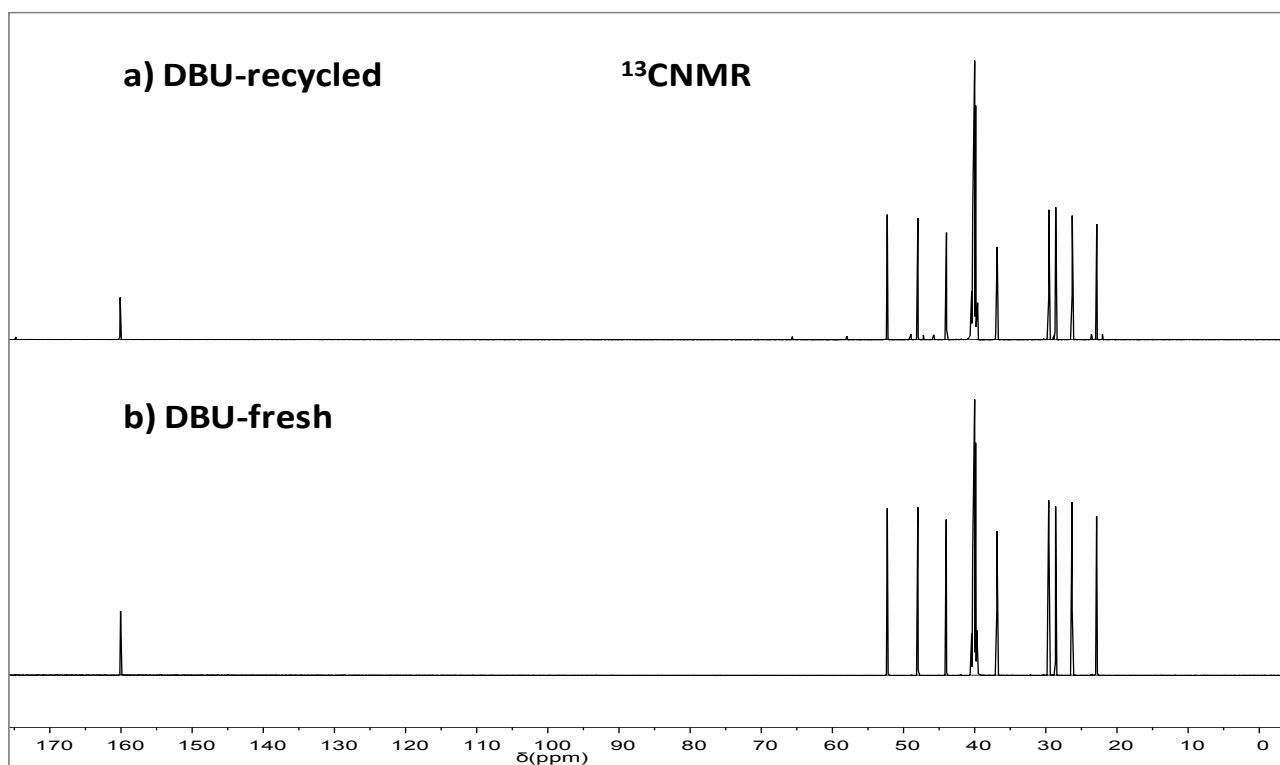
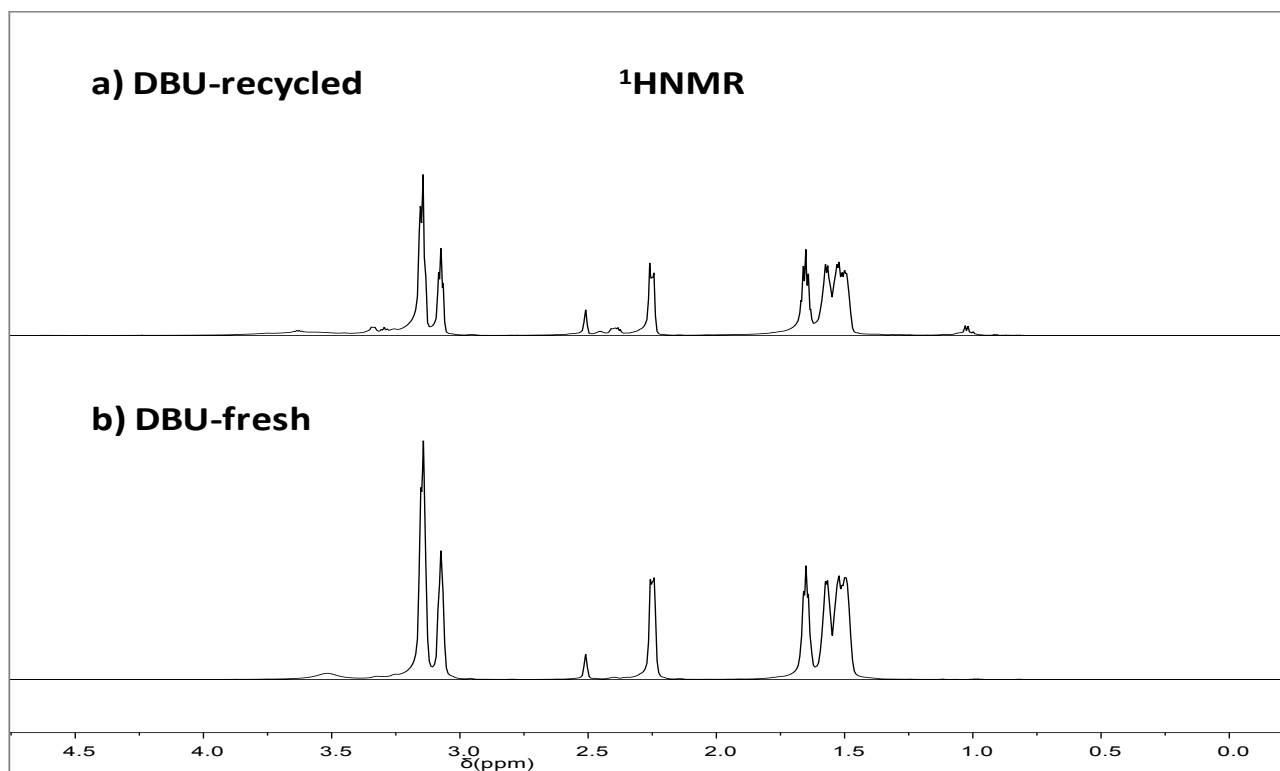


Fig. S6 NMR spectrum of recycled (after 4th -reuse) (a) and fresh (b) DBU.

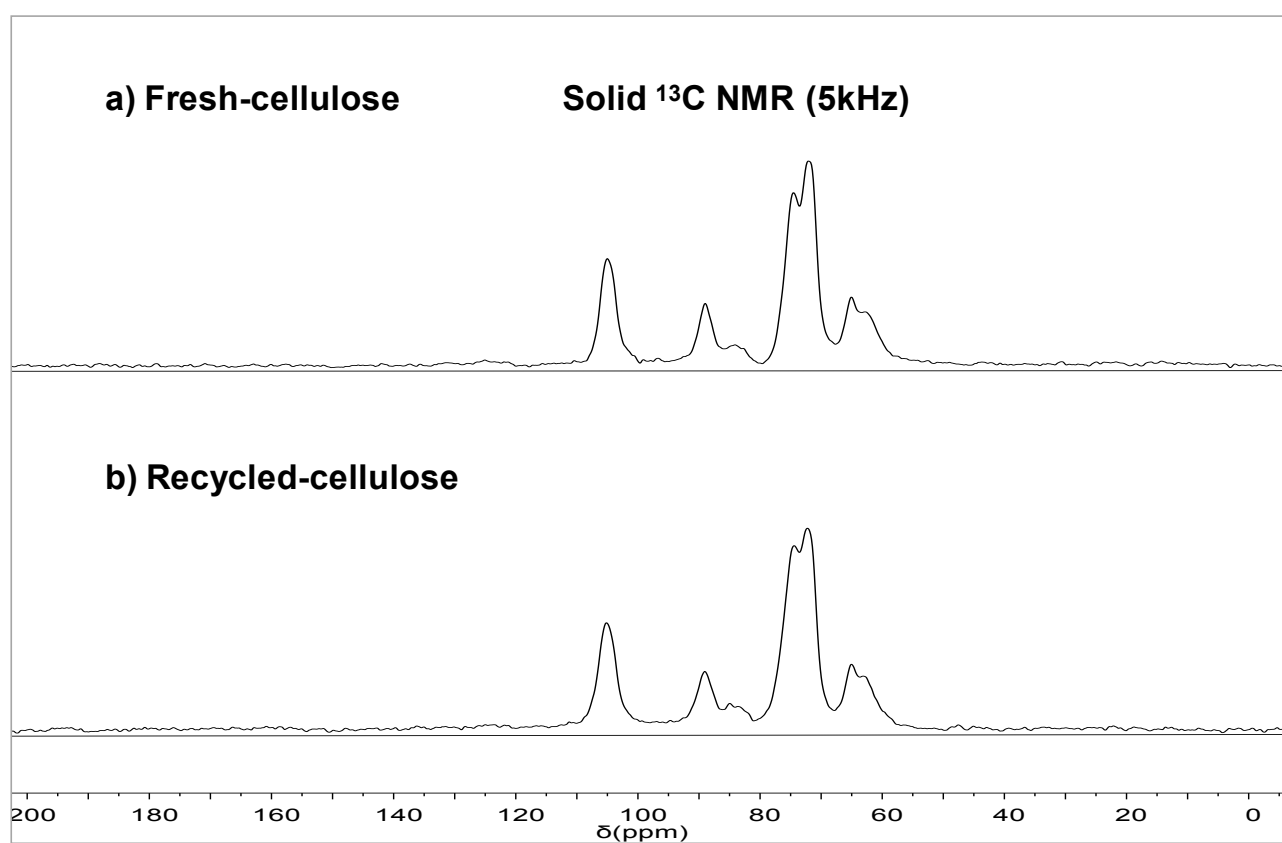
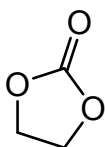


Fig. S7 NMR spectrum of cellulose before (a) and after (b) it had been used four times.

iii) Experimental characterization data for compounds 2a-2h

Spectral characteristics of the products (cyclic carbonates **2a–2h**) in Table 5 were provided as follows:

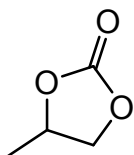
1, 3-dioxolan-2-one (**2a**):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 4.53 (t, $J=10$ Hz, 4H);

^{13}C NMR (100.4 MHz, CDCl_3), δ (ppm): 64.55, 156.36 (C=O).

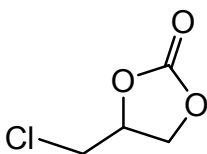
4-methyl-1, 3-dioxolan-2-one (**2b**):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 1.49 (d, $J=6.0$ Hz, 3H), 4.05 (t, $J=8.8$ Hz, 1H), 4.60 (t, $J=8.0$ Hz, 1H), 4.86-4.94 (m, 1H);

^{13}C NMR (100.4 MHz, CDCl_3), δ (ppm): 19.15, 70.53, 73.48, 154.95 (C=O).

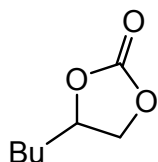
4-chloromethyl-1, 3-dioxolan-2-one (**2c**):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 3.72-3.85 (m, 2H), 4.43 (t, $J=7.2$ Hz, 1H), 4.62 (t, $J=8.6$ Hz, 1H), 4.99-5.05 (m, 1H);

^{13}C NMR (100.4 MHz, CDCl_3), δ (ppm): 43.84, 66.83, 75.49, 154.28 (C=O).

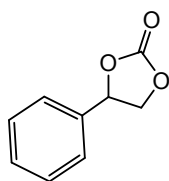
4-butyl-1, 3-dioxolan-2-one (**2d**):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 0.84 (t, $J=7.1$ Hz, 3H); 1.29-1.41 (m, 2H); 1.62-1.76 (m, 2H); 3.93-4.02 (m, 2H); 4.47 (d, $J=8.3$ Hz, 2H); 4.61-4.68 (m, 1H).

^{13}C NMR (75.5 MHz, CDCl_3), δ (ppm): 13.38, 21.85, 26.05, 33.09, 69.12, 76.89, 154.86 (C=O).

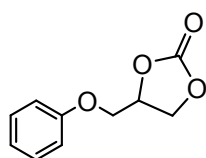
4-phenyl-1, 3-dioxolan-2-one (2e):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 4.34 (t, $J=8.4$ Hz, 1H), 4.80 (t, $J=8.4$ Hz, 1H), 5.68 (t, 1H, $J=8.0$ Hz), 7.35-7.44 (m, 5H);

^{13}C NMR (100.4 MHz, CDCl_3), δ (ppm): 71.13, 77.94, 125.83, 129.20, 129.71, 135.77, 154.78 (C=O).

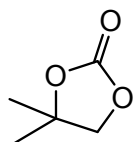
4-phenyloxymethyl-1, 3-dioxolan-2-one (2f):



^1H NMR (400 MHz, CDCl_3), δ (ppm): 4.15 (dd, $J=4.4$ Hz, $J=10.8$ Hz, 1H), 4.24 (dd, $J=3.6$ Hz, $J=10.8$ Hz, 1H), 4.55 (dd, $J=8.4$ Hz, $J=6$ Hz, 1H), 4.62 (t, $J=8.4$ Hz, 1H), 5.03 (m, 1H), 6.91 (d, $J=8.0$ Hz, 2H), 7.02 (t, $J=7.4$ Hz, 2H), 7.31 (t, $J=8.0$ Hz, 2H).

^{13}C NMR (100.4 MHz, CDCl_3), δ (ppm): 66.17, 68.84, 74.11, 114.57, 121.92, 129.62, 154.65 (C=O), 157.71 (C-O).

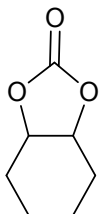
4,4-dimethyl-1,3-dioxolan-2-one (2g):



^1H NMR (600 MHz, DMSO), δ (ppm): 1.45 (s, 6H), 4.24 (s, 2H);

^{13}C NMR (150.9 MHz, DMSO), δ (ppm): 25.82, 40.27, 75.22, 82.32, 154.55 (C=O).

4, 5-tetramethylene-1, 3-dioxolan-2-one (2h):

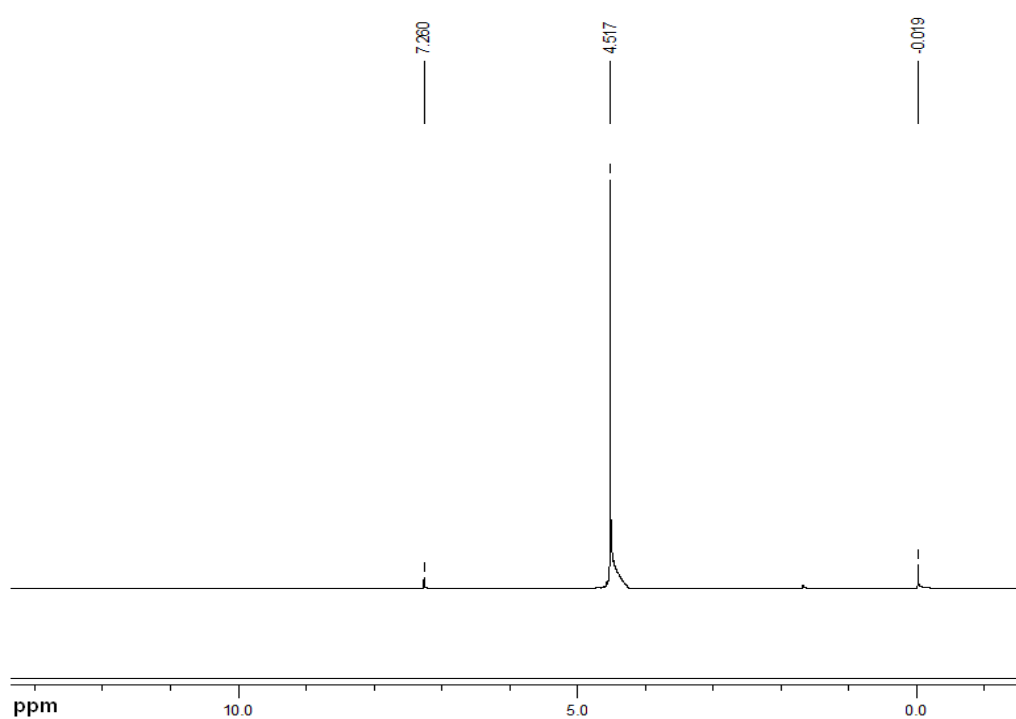


^1H NMR (600 MHz, CDCl_3), δ (ppm): 1.40-1.51 (m, 2H); 1.54-1.73 (m, 2H); 1.89-1.92 (m, 4H);

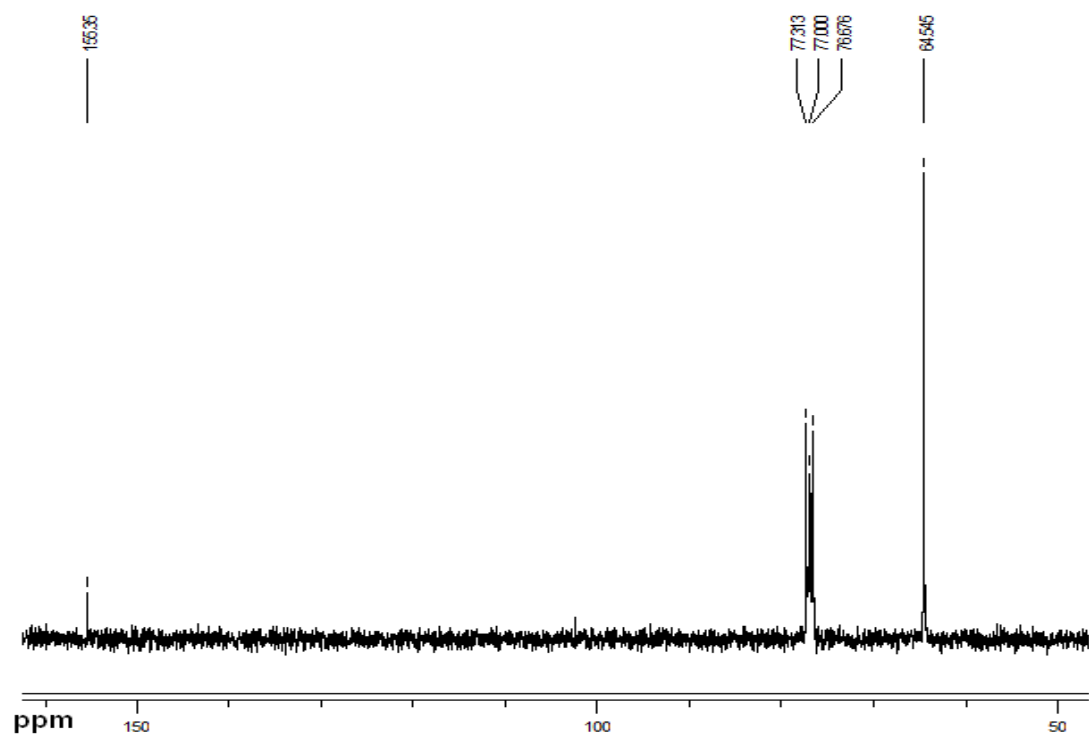
4.68-4.71 (m, 2H); ^{13}C NMR (150.9 MHz, CDCl_3), δ (ppm): 19.11, 26.78, 75.75, 155.36 (C=O).

^1H NMR and ^{13}C NMR copies of product 2a-2h

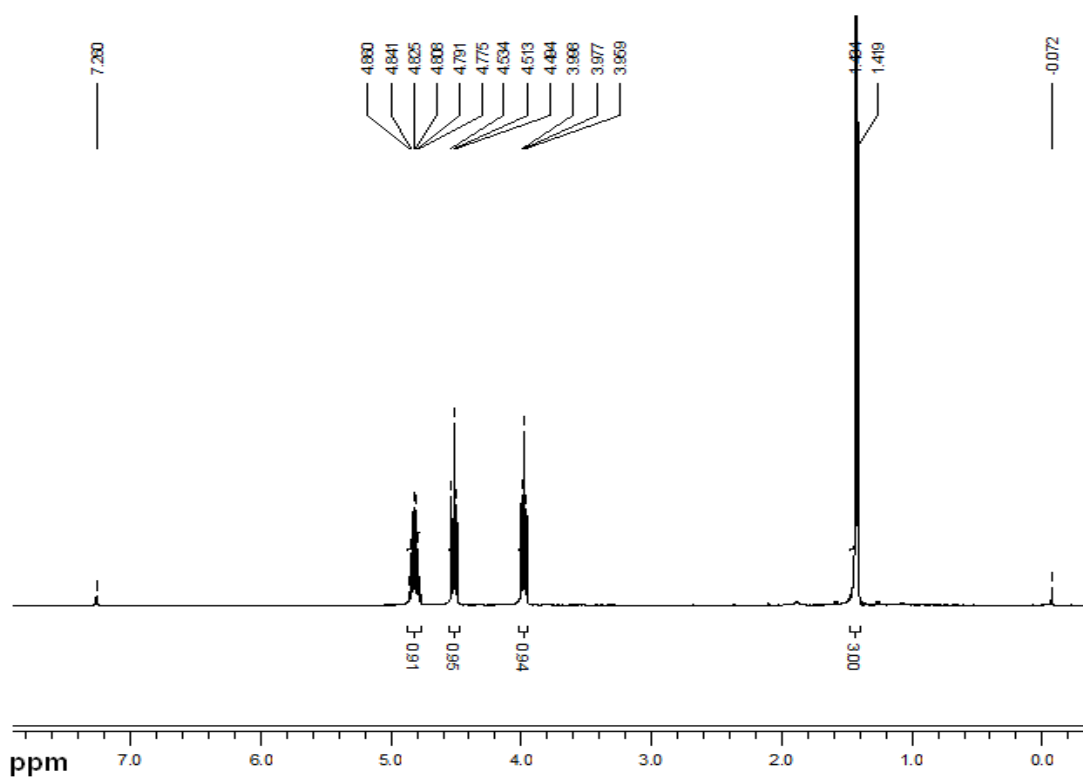
2a ^1H NMR



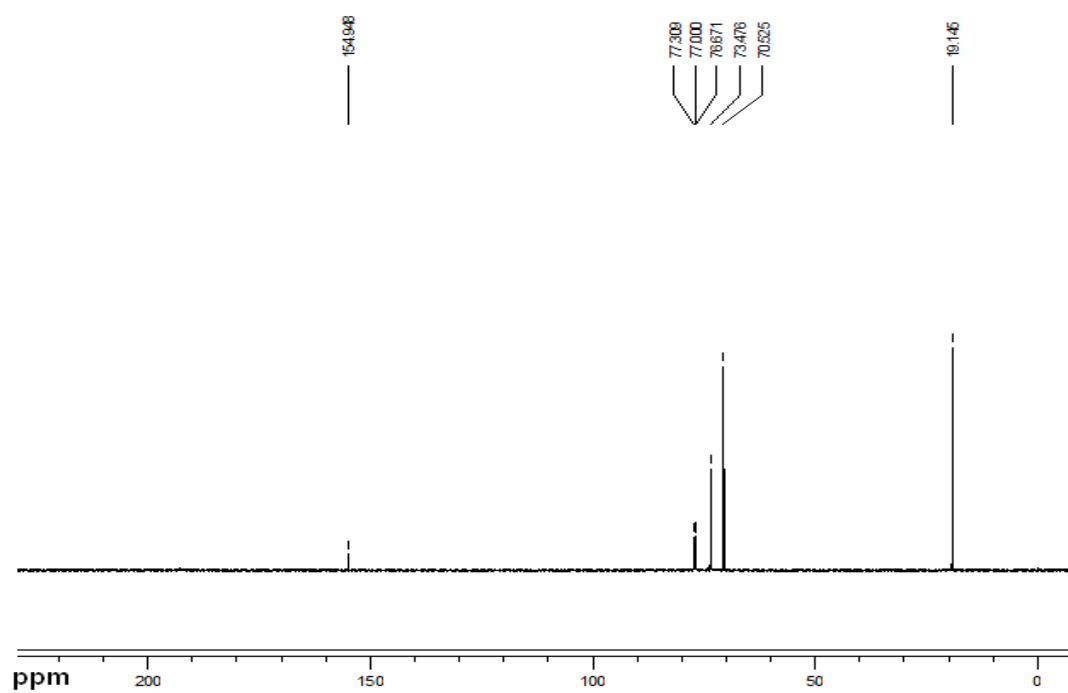
2a ^{13}C NMR



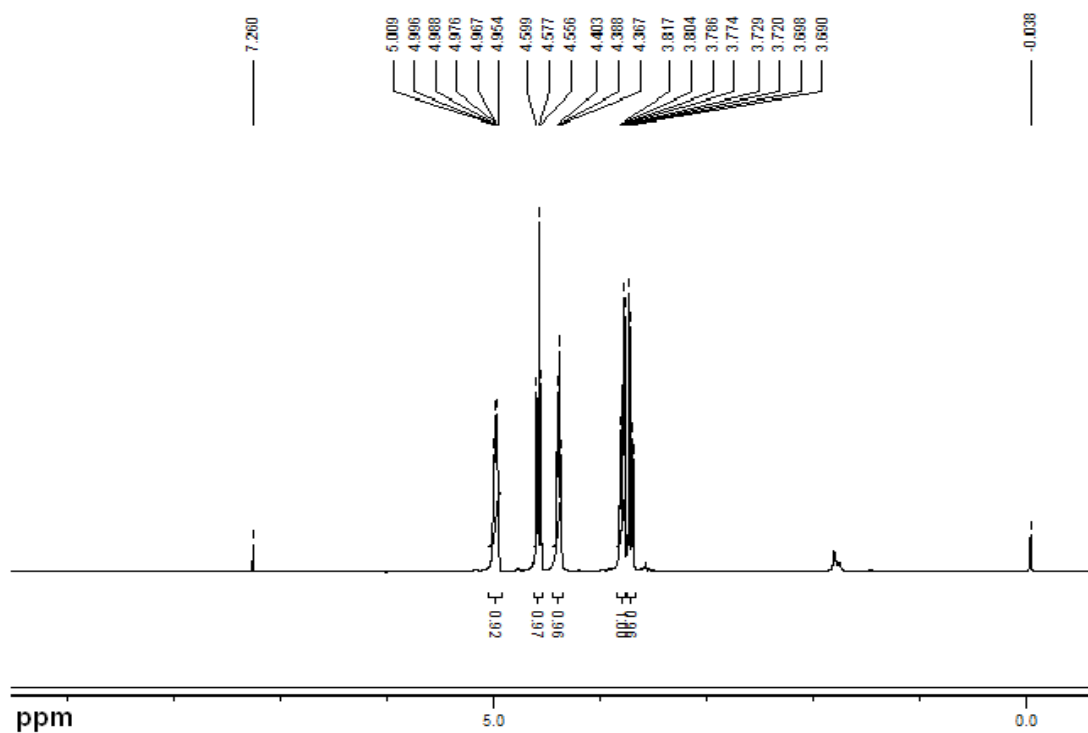
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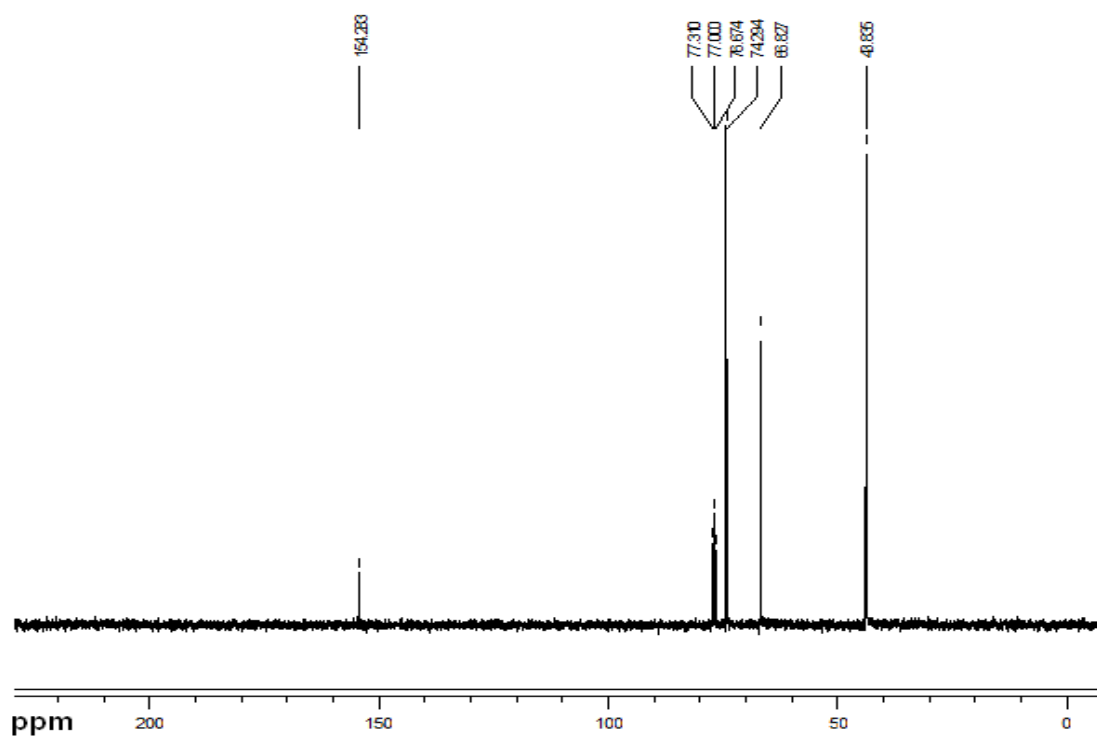
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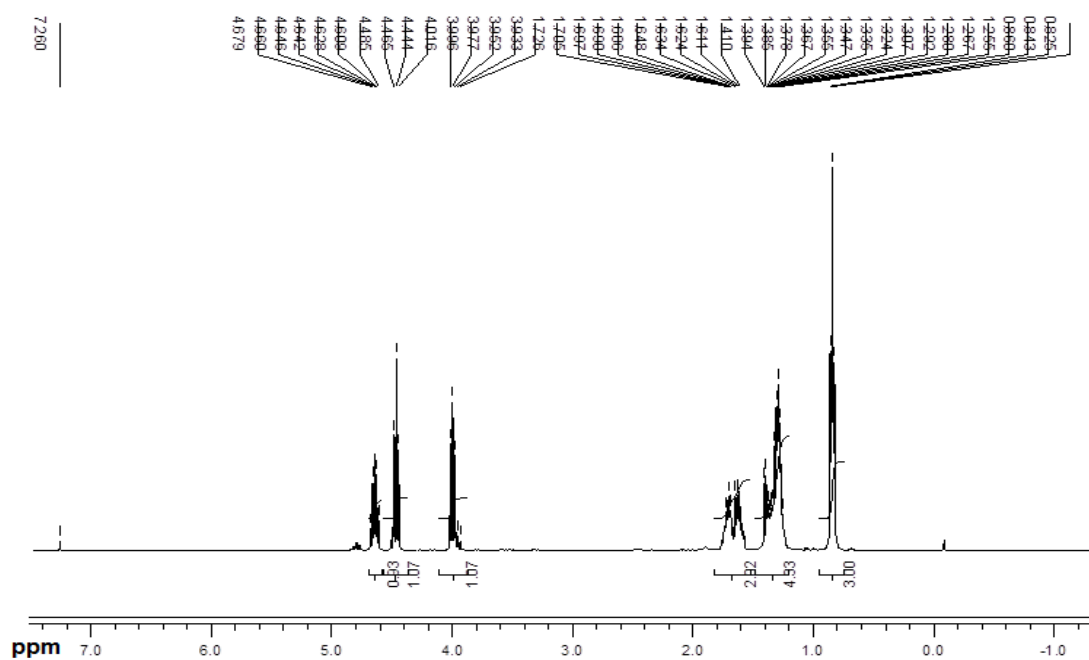
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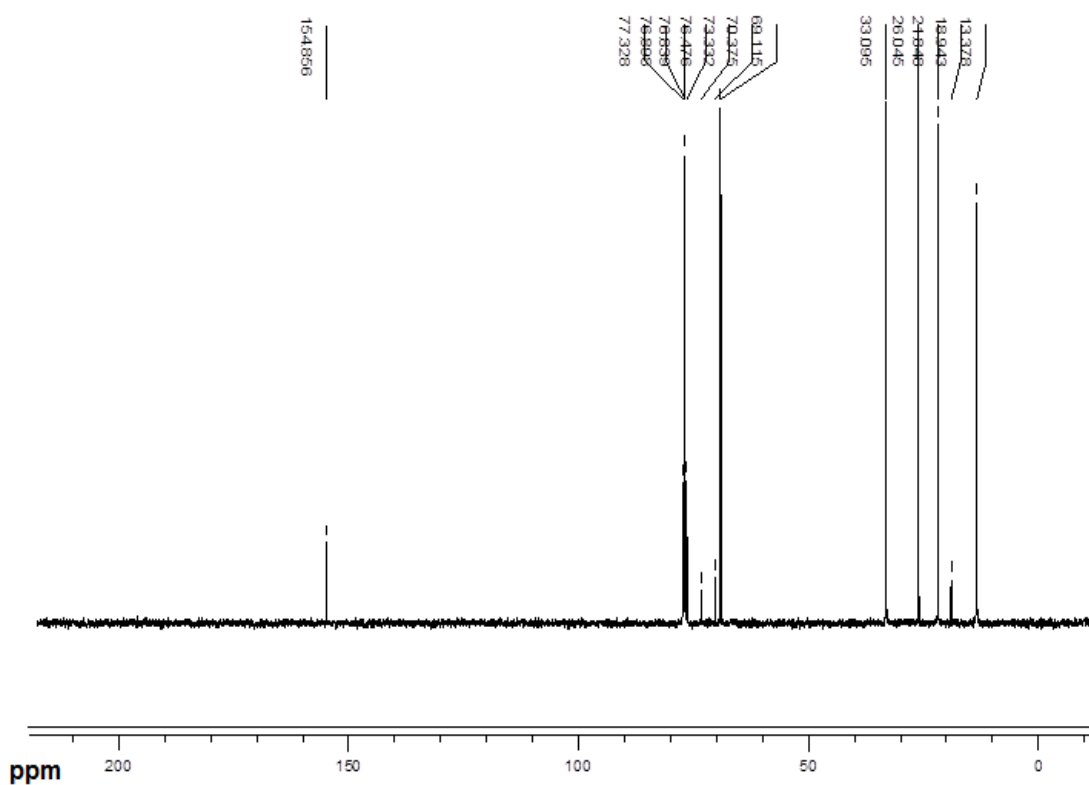
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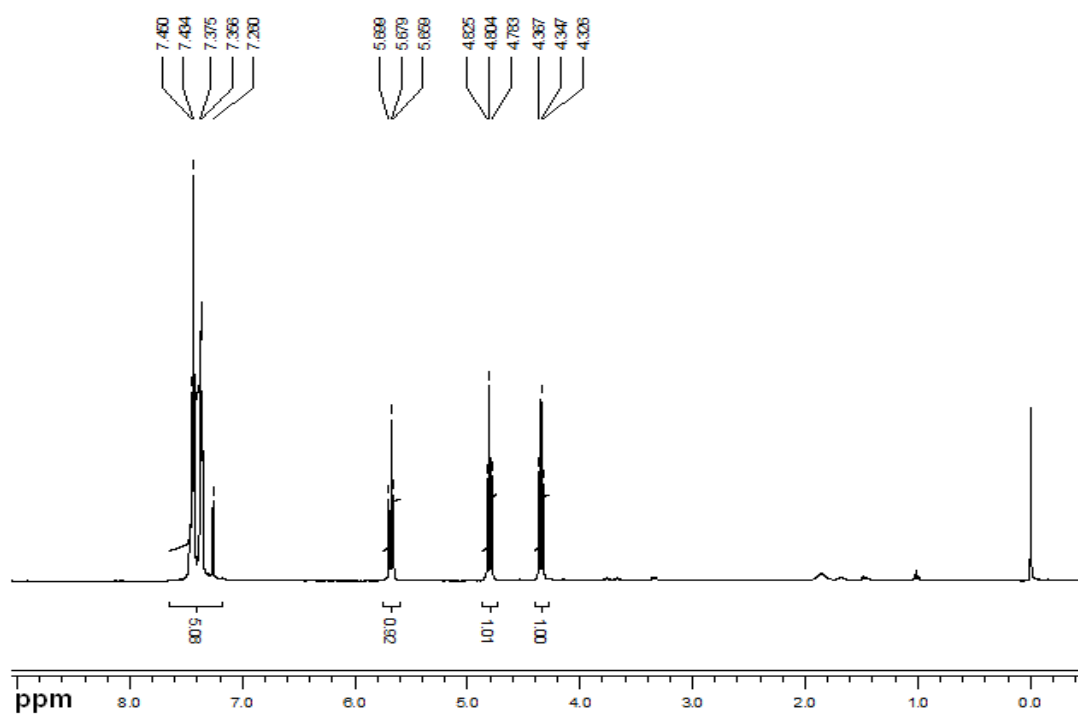
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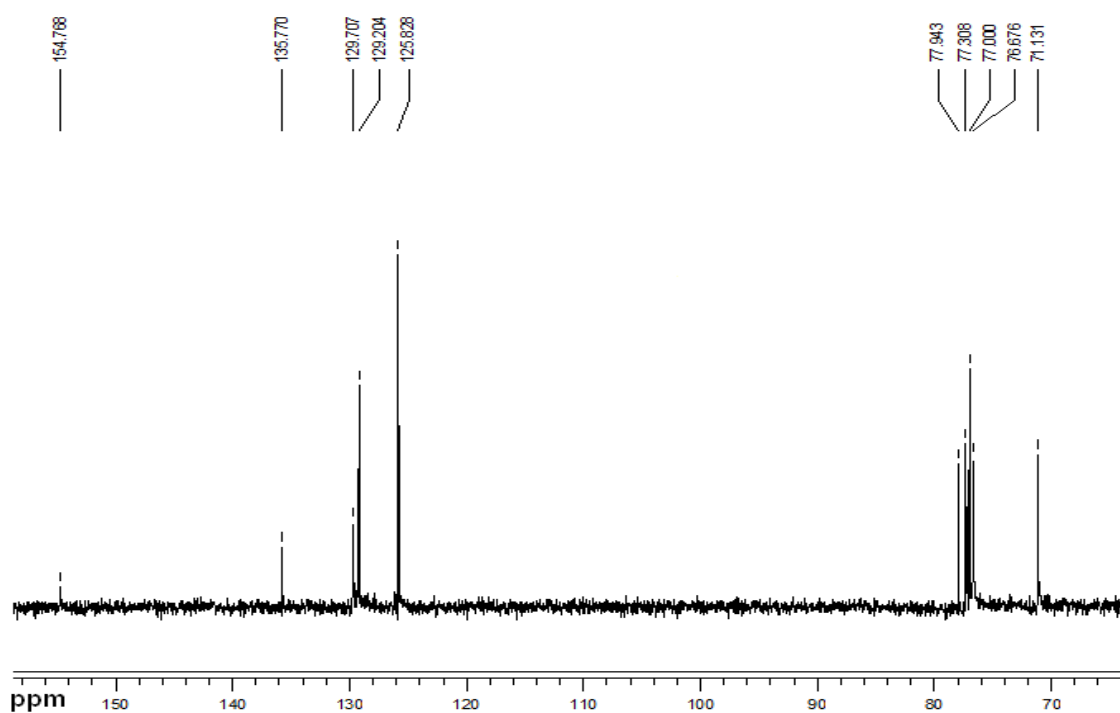
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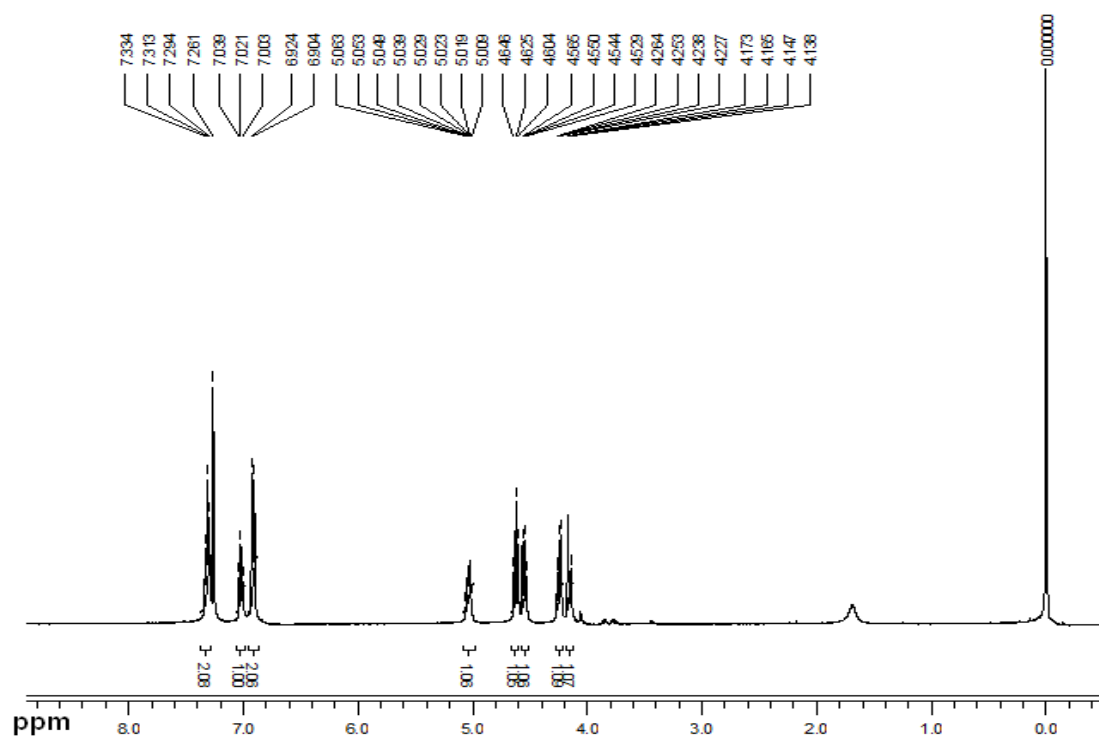
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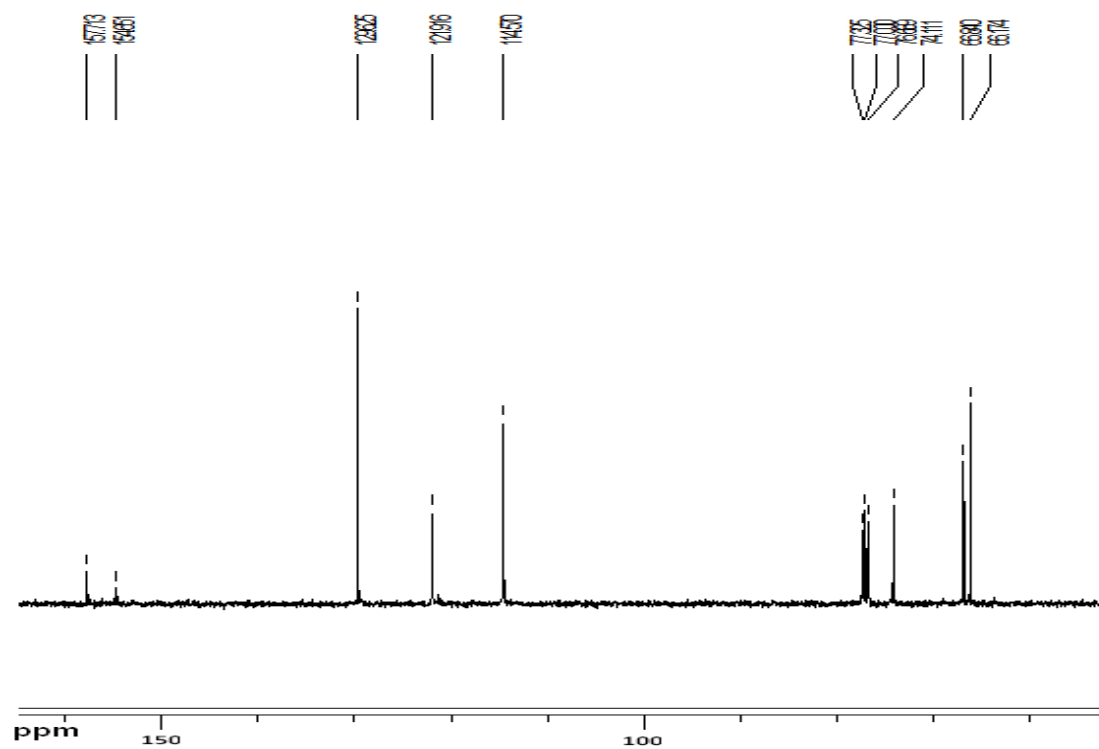
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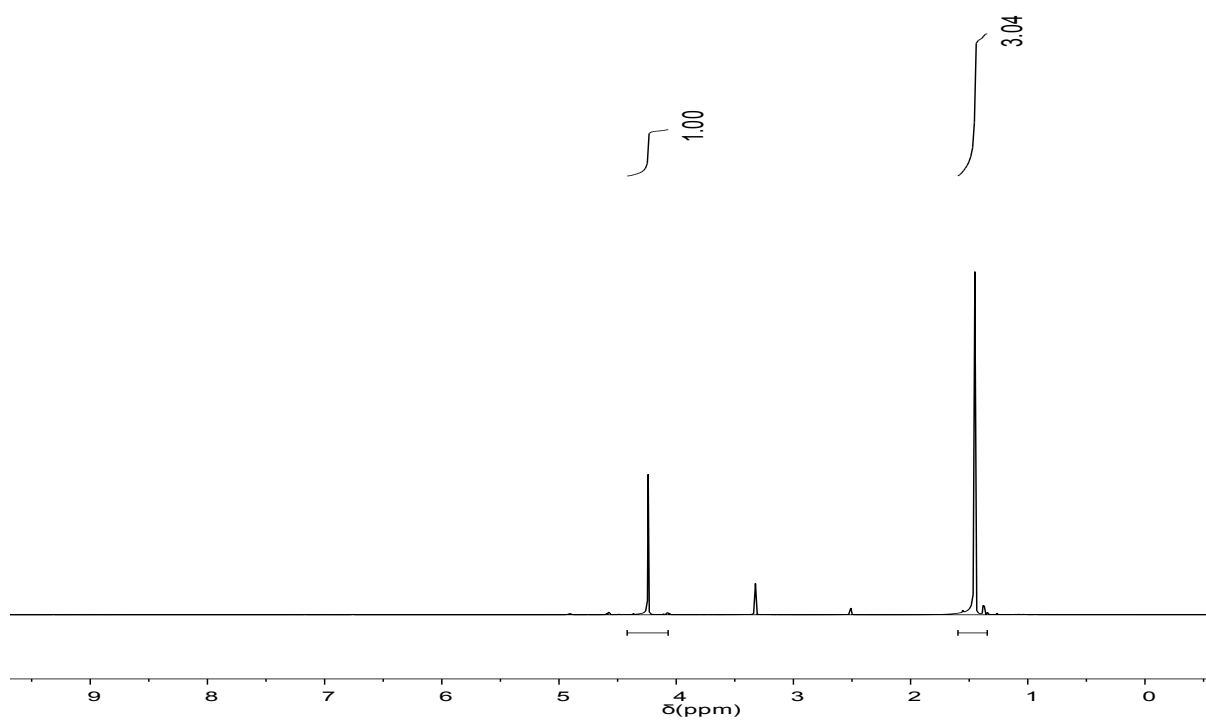
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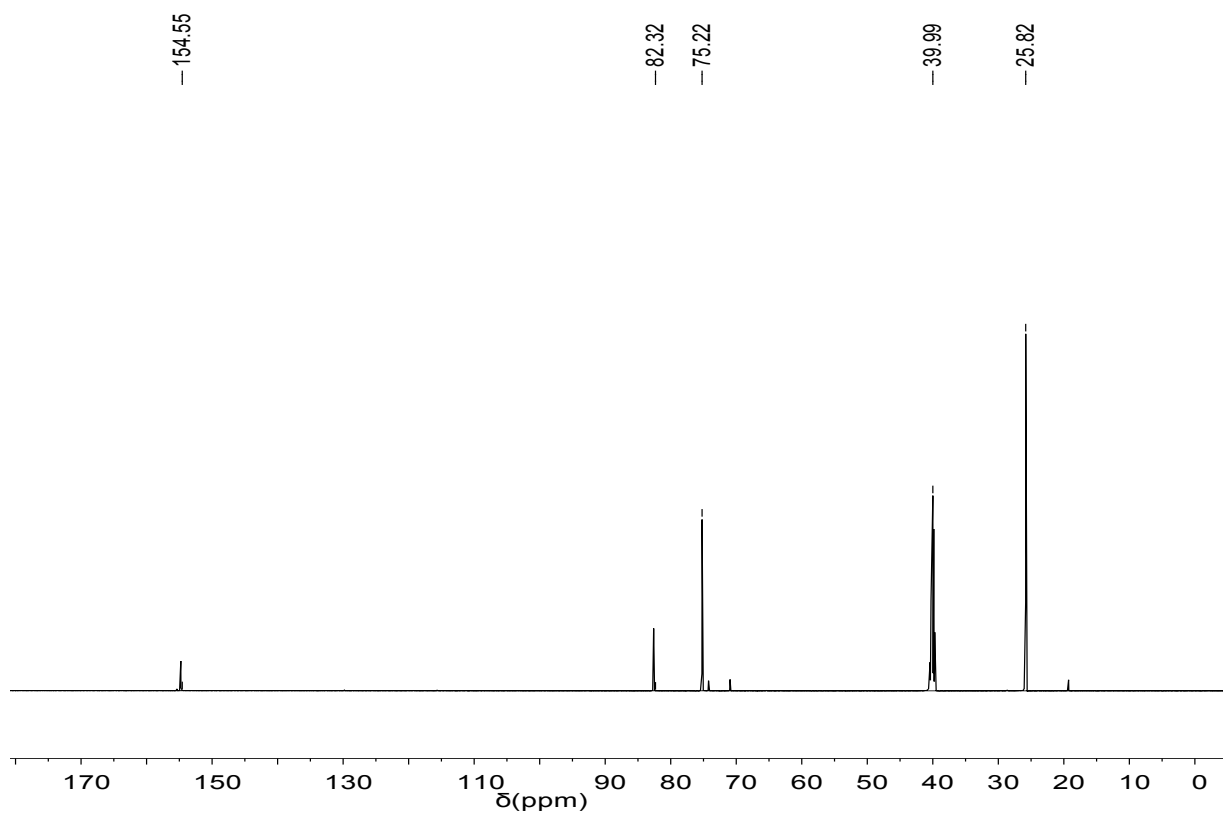
2f ¹³C NMR



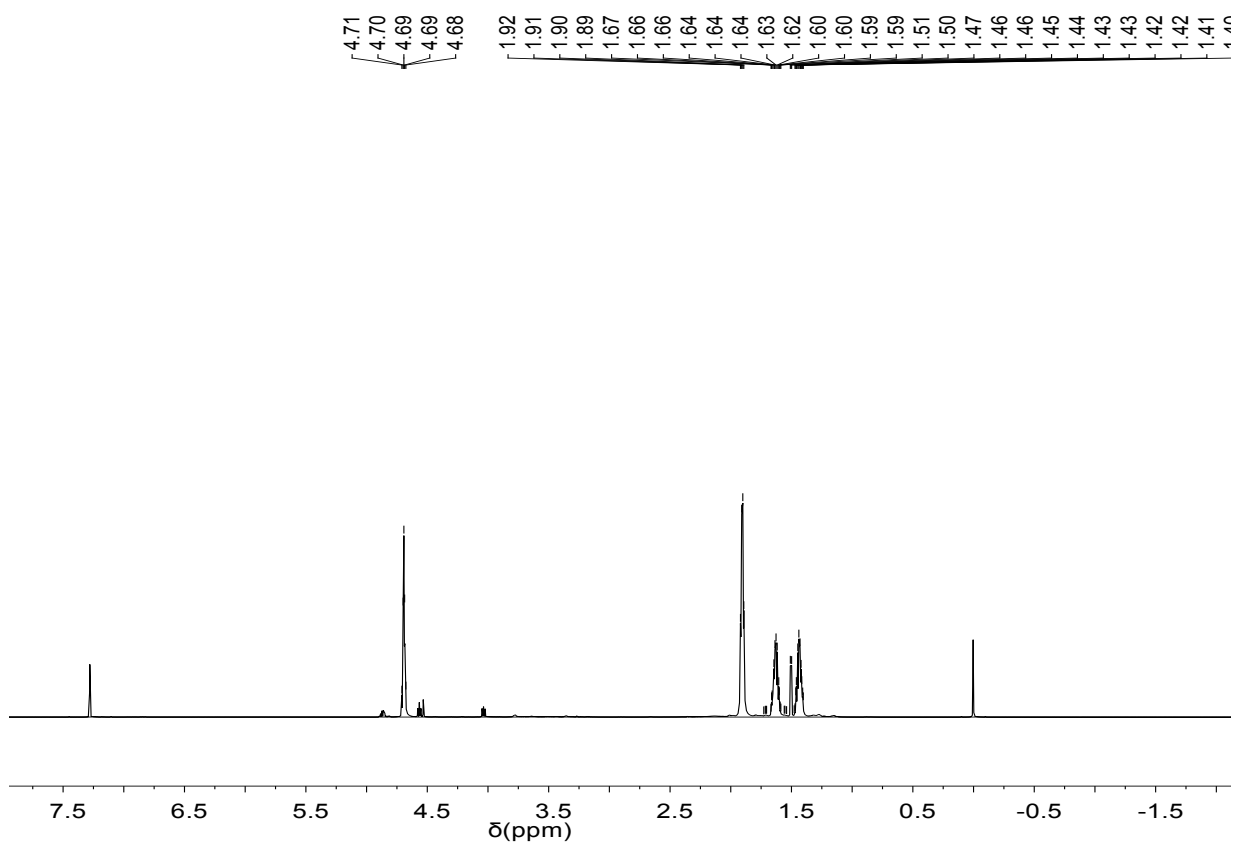
2g ^1H NMR



2g ^{13}C NMR



2h ¹H NMR



2h ¹³C NMR

