

Supplementary Information

A highly efficient approach to vanillin starting from 4-cresol

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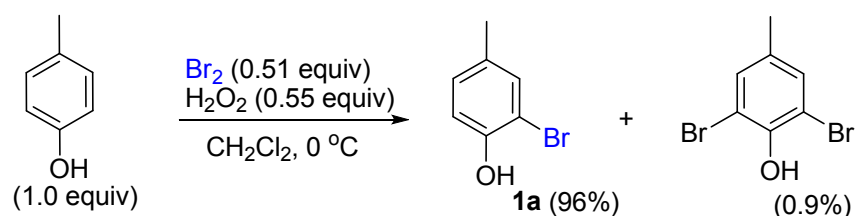
1. General Information

Unless otherwise indicated, all reagents were obtained from commercial sources and used as received without further purification. All reactions were carried out in oven-dried glassware and monitored by thin layer chromatography (TLC, pre-coated silica gel plates containing HF₂₅₄). Reaction products were purified *via* column chromatography on silica gel (300–400 mesh). Melting points were determined using an open capillaries and uncorrected. NMR spectra were determined on Bruker AV400 in CDCl₃ or DMSO-*d*₆ with TMS as internal standard for ¹H NMR (400 MHz) and ¹³C NMR (100 MHz), respectively. HRMS were carried out on a QSTAR Pulsar I LC/TOF MS mass spectrometer or Micromass GCTTM gas chromatograph-mass spectrometer. GC-MS were carried out on an Agilent 6890-5973N gas chromatograph-mass spectrometer.

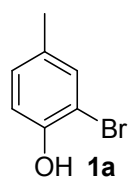
2. General preparation procedure and characterization data

2.1 Mild and eco-friendly oxybromination.

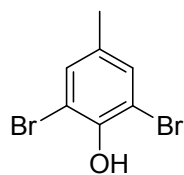
2.1.1 Oxybromination of 4-cresol into 1a.



General procedure: a three-necked flask was charged with 4-cresol (1.08 g, 10 mmol), CH₂Cl₂ (10 mL) and H₂O₂ (30%, 0.58 mL, *d* = 1.11 g/mL, 5.5 mmol), and then the solution was cooled to 0 °C. Under the temperature, to the solution was slowly added a solution of bromine (0.26 mL, *d* = 3.12 g/mL, 5.1 mmol) and CH₂Cl₂ (5 mL) over 4 h through a syringe pump. Afterwards, the mixture was further stirred for another 4 h. An aqueous NaHSO₃ (10 mL, 2%) was added to the mixture at 0 °C, and the reaction solution was allowed to stir for 1 h at room temperature. Furthermore, the solution was partitioned into two layers, and the aqueous phase was extracted with CH₂Cl₂ (5 mL × 3). Finally, the combined organic layers were dried over anhydrous Na₂SO₄, and concentrated to give a crude oil, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the desired product **1a** (chromatographic separation can provide a more credible yield).

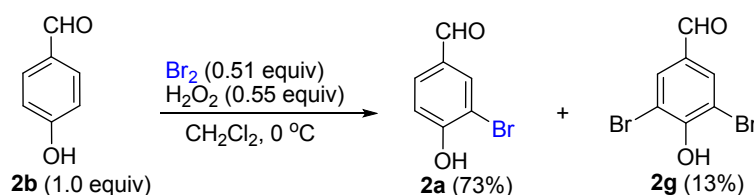


2-Bromo-4-methylphenol (1a): yellow oil, 1.80 g (96%); ¹H NMR (400 MHz, CDCl₃, ppm): δ 7.28 (br d, *J* = 1.6 Hz, 1H), 7.02 (br d, *J* = 8.0, 1H), 6.92 (br d, *J* = 8.0 Hz, 1H), 5.40 (br s, 1H), 2.27 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 150.0, 132.1, 131.4, 129.8, 115.8, 109.8, 20.2; HRMS (EI): *m/z* [M⁺] calcd. for C₇H₇OBr 185.9680, found 185.9682.

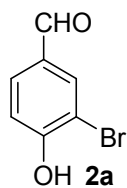


2,6-Dibromo-4-methylphenol:¹ yellow solid, 24 mg (0.9%), m.p. 44–46 °C (lit¹ m.p. 49–51 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 7.26 (br s, 2H), 5.71 (br s, 1H), 2.26 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 147.1, 132.4 (4C), 109.4, 20.0; HRMS (EI): m/z [M⁺] calcd. for C₇H₆OBr₂ 263.8785, found 263.8787.

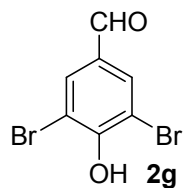
2.1.2 Oxybromination of 4-hydroxybenzaldehyde into 2a.



General procedure: a three-necked flask was charged with 4-hydroxybenzaldehyde (1.22 g, 10 mmol), CH₂Cl₂ (10 mL) and H₂O₂ (30%, 0.58 mL, $d = 1.11$ g/mL, 5.5 mmol), and then the solution was cooled to 0 °C. Under the temperature, to the solution was slowly added a solution of bromine (0.26 mL, $d = 3.12$ g/mL, 5.1 mmol) and CH₂Cl₂ (5 mL) over 4 h through a syringe pump. Afterwards, the mixture was further stirred for another 4 h. An aqueous NaHSO₃ (10 mL, 2%) was added to the mixture at 0 °C, and the reaction solution was allowed to stir for 1 h at room temperature. Furthermore, the solution was partitioned into two layers, and the aqueous phase was extracted with CH₂Cl₂ (5 mL $\times$ 3). Finally, the combined organic layers were dried over anhydrous Na₂SO₄, and concentrated to give a crude oil, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the desired product **2a** and dibromination product **2g**.



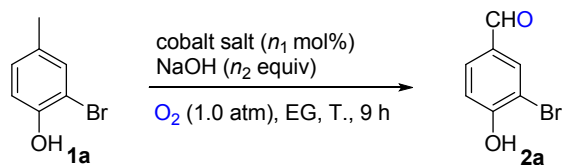
3-Bromo-4-hydroxybenzaldehyde (2a):² pale yellow solid, 1.46 g (73% yield), m.p. 130–132 °C (lit² m.p. 130–132 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.83 (br s, 1H), 8.04 (br s, 1H), 7.77 (br d, $J = 8.4$ Hz, 1H), 7.15 (br d, $J = 8.4$ Hz, 1H), 6.43 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 192.7, 151.8, 132.9, 130.3, 128.1, 127.5, 127.4; HRMS (EI): m/z [M⁺] calcd. for C₇H₅O₂Br 199.9473, found 199.9474.



3,5-Dibromo-4-hydroxybenzaldehyde (2g):³ white solid, 0.36 g (13%), m.p. 182–184 °C (lit³ m.p. 181–183 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.80 (br s, 1H), 8.00 (br s, 2H), 6.40 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 188.3, 154.5, 133.8 (2C), 131.5, 110.9 (2C); HRMS (ESI): m/z [M–H⁺] calcd. for C₇H₃Br₂O₂ 276.8500, found 276.8517.

2.2 Oxidation of 1a into 2a.

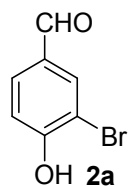
Table S1. Optimizations for the reaction conditions.^a



Entry	Co salt (n_1 mol%)	NaOH (n_2 equiv)	T. (°C)	Yield (%) ^b
1	CoCl ₂ (3.0)	2.0	50	trace
2	CoBr ₂ (3.0)	2.0	50	trace
3	CoF ₂ (3.0)	2.0	50	trace
4	Cobalt(II) acetylacetonate (3.0)	2.0	50	45
5	Cobalt tetramethoxyphenylporphyrin (3.0)	2.0	50	13
6	Co(C ₂ O ₄) · 2H ₂ O (3.0)	2.0	50	27
7	Co(OAc) ₂ · 4H ₂ O (3.0)	2.0	50	53
8	Co(OAc) ₂ · 4H ₂ O (3.0)	2.0	60	59
9	Co(OAc) ₂ · 4H ₂ O (3.0)	2.0	70	68
10	Co(OAc) ₂ · 4H ₂ O (3.0)	2.0	80	76
11	Co(OAc) ₂ · 4H ₂ O (3.0)	0	80	0
12	Co(OAc) ₂ · 4H ₂ O (3.0)	3.0	80	90
13	Co(OAc) ₂ · 4H ₂ O (3.0)	4.0	80	90
14	Co(OAc) ₂ · 4H ₂ O (2.0)	4.0	80	90
15	Co(OAc) ₂ · 4H ₂ O (1.0)	4.0	80	90
16	Co(OAc) ₂ · 4H ₂ O (0.5)	4.0	80	79
17	Co(OAc) ₂ · 4H ₂ O (1.0)	4.0	80	trace ^c

^aReaction conditions: **1a** (5.0 mmol), cobalt salt (n_1 mol%), NaOH (n_2 equiv), EG (10 mL), O_2 (1.0 atm), 9 h. ^bIsolated yield. ^cPerformed under argon atmosphere.

General procedure: a three-necked flask was charged with EG (10 mL), **1a** (0.94 g, 5.0 mmol), cobalt salt (n mol%), and solid NaOH (n_2 equiv), and then the solution was heated to 80 °C. The molecular oxygen was continuously supplied to the reaction through a top tube inlet for 9 h. Hydrochloric acid (10 mL, 10%) and methyl *tert*-butyl ether (MTBE, 15 mL) were successively added to the reaction mixture at room temperature. The MTBE phase was separated, and the aqueous phase was further extracted with MTBE (15 mL × 2). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated in vacuo to give a residue, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the desired product **2a**.

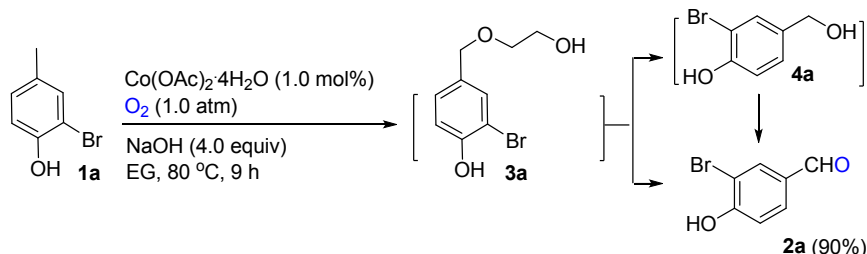


3-Bromo-4-hydroxybenzaldehyde (2a):² pale yellow solid, 0.90 g (as the best yield of 90%), m.p. 130–132 °C (lit² m.p. 130–132 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.83 (br s, 1H), 8.04 (br s, 1H), 7.77 (br d, J = 8.4 Hz, 1H), 7.15 (br d, J = 8.4 Hz, 1H), 6.43 (br s, 1H); ¹³C

NMR (100 MHz, CDCl₃, ppm): δ 192.7, 151.8, 132.9, 130.3, 128.1, 127.5, 127.4; HRMS (EI): m/z [M⁺] calcd. for C₇H₅O₂Br 199.9473, found 199.9474.

2.3 Reaction process analysis for the oxidation of 1a.

2.3.1 The oxidation process of 1a in EG (Fig. 1 in the text).



Oxidation process analysis: a three-necked flask was charged with EG (10 mL), 1a (0.94 g, 5.0 mmol), Co(OAc)₂·4H₂O (12.5 mg, 0.05 mmol) and solid NaOH (0.80 g, 20 mmol), and then the solution was heated to 80 °C. The molecular oxygen was continuously supplied to the reaction through a top tube inlet. About 0.2 mL of sample, withdrawn from the reaction mixture at the specified time (1.5, 3.0, 4.5, 6.0, 7.5 and 9.0 h), was acidified by hydrochloric acid (2.0 M) and was extracted with MTBE. The extracted sample was analyzed by GC-MS to evaluate the product distributions. It should be noted that the sample should be immediately analyzed after extraction.

Besides, another incomplete oxidation reaction (performed for 4 h) was worked up and purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the intermediates 3a and 4a for structure confirmation.

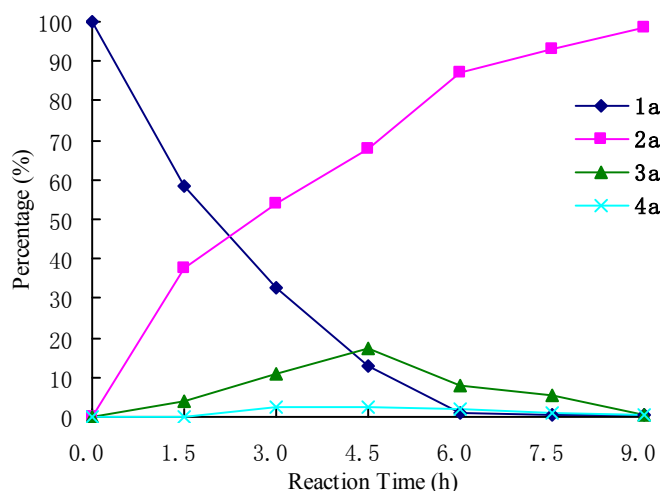
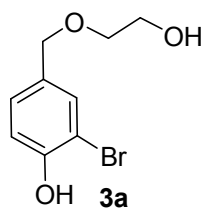
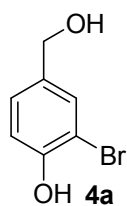


Fig. 1 Time-dependence curves for the oxidation 1a to 2a in EG (percentages as their respective ratios by area normalization method in the total ionization chromatography).



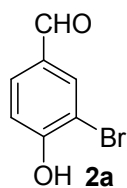
2-Bromo-4-((2-hydroxyethoxy)methyl)phenol (3a): yellow oil, ^1H NMR (400

MHz, CDCl_3 , ppm): δ 7.45 (br d, $J = 2.0$ Hz, 1H), 7.16 (dd, $J = 8.0, 2.0$ Hz, 1H), 6.95 (br d, $J = 8.0$ Hz, 1H), 4.44 (s, 2H), 3.76 (t, $J = 4.8$ Hz, 2H), 3.57 (t, $J = 4.8$ Hz, 2H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 152.4, 132.2, 131.2, 128.8, 116.2, 110.0, 72.2, 71.3, 61.7; HRMS (ESI): m/z [M^+] calcd. for $\text{C}_9\text{H}_{10}\text{O}_3\text{Br}$ 244.9813, found 244.9823.



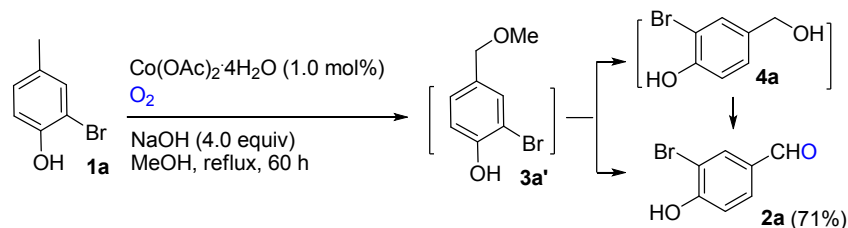
2-Bromo-4-(hydroxymethyl)phenol (4a):⁴ white solid, m.p. 126–128 °C (lit⁴ m.p.

127–129 °C); ^1H NMR (400 MHz, $\text{DMSO}-d_6$, ppm): δ 7.41 (br d, $J = 2.0$ Hz, 1H), 7.11 (dd, $J = 8.0, 2.0$ Hz, 1H), 6.90 (br d, $J = 8.4$ Hz, 1H), 4.36 (s, 2H); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$, ppm): δ 153.2, 135.3, 131.5, 127.5, 116.5, 109.3, 62.4; HRMS (EI): m/z [M^+] calcd. for $\text{C}_7\text{H}_7\text{O}_2\text{Br}$ 201.9629, found 201.9631.



3-Bromo-4-hydroxybenzaldehyde (2a): the spectral data see 2.1 section.

2.3.2 The oxidation process of 1a in methanol (Fig. S1).



Oxidation process analysis: a three-necked flask was charged with MeOH (10 mL), **1a** (0.94 g, 5.0 mmol), $\text{Co(OAc)}_2\cdot 4\text{H}_2\text{O}$ (12.5 mg, 0.05 mmol) and solid NaOH (0.8 g, 20 mmol). The mixture was heated to reflux (about 75 °C), and molecular oxygen was charged to the reaction solution. About 0.2 mL of sample, withdrawn from the reaction mixture at the specified time (12, 24, 36, 48 and 60 h), was acidified by hydrochloric acid (2.0 M) and was extracted with MTBE. The extracted sample was analyzed by GC-MS to evaluate the product distributions. It should be noted that the sample should be immediately analyzed after extraction.

Besides, an incomplete oxidation reaction (performed for 16 h) was worked up and purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the intermediates **3a'** and **4a** for structure confirmation.

Furthermore, a preparation reaction (performed for 60 h) was worked up and purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the desired product **2a**.

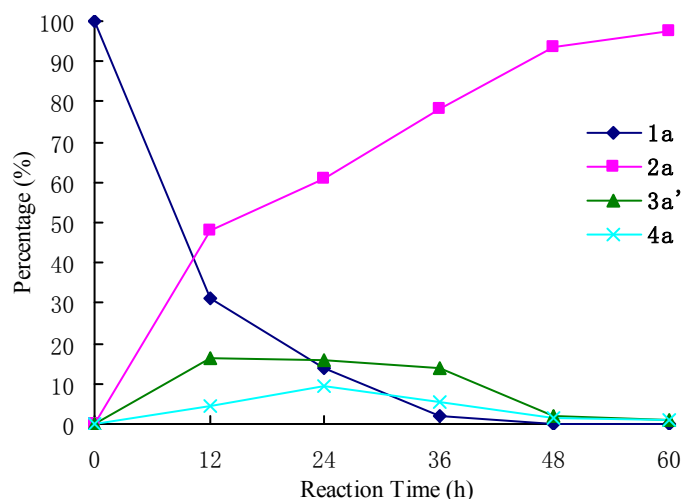
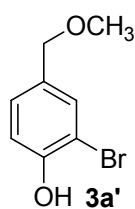
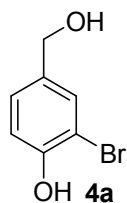


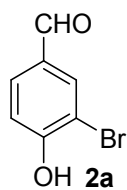
Fig. S1 Time-dependence curves for the oxidation of **1a** into **2a** in methanol (percentages as their respective ratios by area normalization method in the total ionization chromatography).



2-Bromo-4-(methoxymethyl)phenol (3a'): yellow oil, ^1H NMR (400 MHz, CDCl_3 , ppm): δ 7.39 (br s, 1H), 7.10 (br d, $J = 8.0$ Hz, 1H), 6.89 (br d, $J = 8.0$ Hz, 1H), 5.66 (br s, 1H), 4.29 (s, 2H), 3.29 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 150.8, 130.7, 130.6, 127.8, 114.9, 109.1, 72.5, 56.9; HRMS (EI): m/z [M^+] calcd. for $\text{C}_8\text{H}_9\text{O}_2\text{Br}$ 215.9786, found 215.9792.

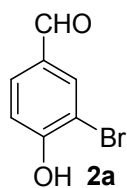


2-Bromo-4-(hydroxymethyl)phenol (4a): the spectral data see 2.3.2 section.

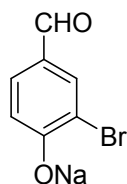


2a **3-Bromo-4-hydroxybenzaldehyde (2a):** pale yellow solid, 0.71 g (71% yield); the spectral data see 2.1 section.

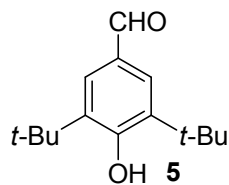
2.4 ^1H NMR spectra of compounds 2a and 5, in $\text{CD}_3\text{ONa}/\text{CD}_3\text{OD}$ and CD_3OD , respectively.



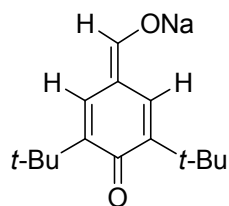
2a **3-Bromo-4-hydroxybenzaldehyde (2a):** ^1H NMR (400 MHz, CD_3OD , ppm): δ 9.75 (br s, 1H), 8.03 (br d, $J = 2.0$ Hz, 1H), 7.74 (dd, $J = 8.0, 2.0$ Hz, 1H), 7.02 (br d, $J = 8.0$ Hz, 1H).



phenolate of **2a** **Sodium 2-bromo-4-formylphenolate:** ^1H NMR (400 MHz, CD_3OD , **2a** (6 mg) and CD_3ONa (7 mg) dissolved in 0.5 mL CD_3OD , ppm): δ 9.39 (br s, 1H), 7.89 (br d, $J = 2.0$ Hz, 1H), 7.51 (dd, $J = 8.4, 2.0$ Hz, 1H), 6.63 (br d, $J = 8.4$ Hz, 1H).

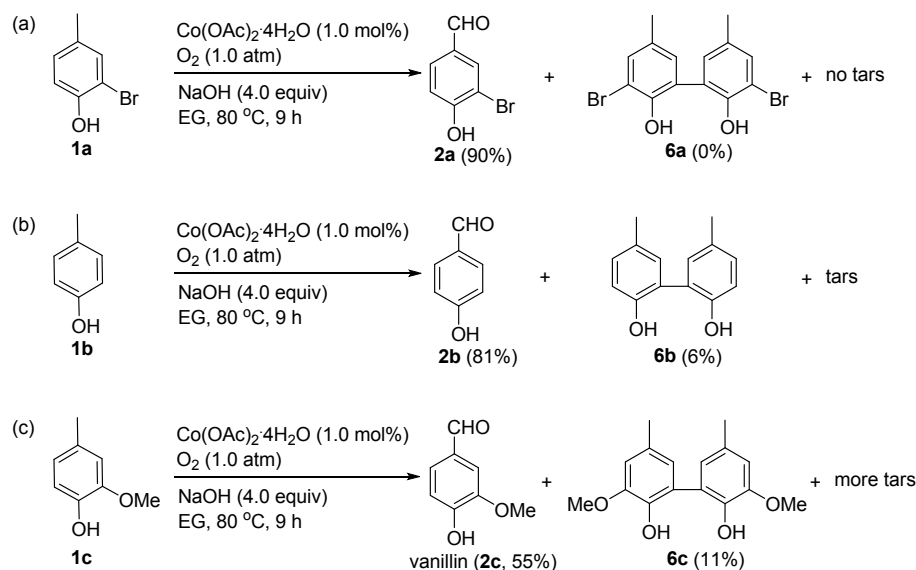


5 **3,5-Di-tert-butyl-4-hydroxybenzaldehyde (5):** ^1H NMR (400 MHz, CD_3OD , ppm): δ 9.77 (br s, 1H), 7.75 (br s, 2H), 1.46 (s, 18H).



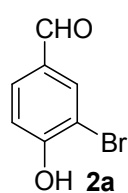
enolate of **5** **Sodium (3,5-di-tert-butyl-4-oxocyclohexa-2,5-dienylidene)methanolate:** ^1H NMR (400 MHz, CD_3OD , **5** (7 mg) and CD_3ONa (7 mg) dissolved in 0.5 mL CD_3OD , ppm): δ 9.02 (br s, 1H), 7.67 (br s, 1H), 7.25 (br s, 1H), 1.38 (s, 18H).

2.5 $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ -catalyzed oxidation of 2-bromo-4-cresol (1a), 4-cresol (1b) and 2-methoxy-4-cresol (1c, Scheme 5 in the text).

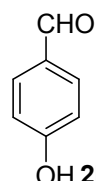


Scheme 5. Oxidations of **1a**, **1b** and **1c** under the standard conditions. Reaction performed with **1** (5.0 mmol), $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ (0.05 mmol), NaOH (20 mmol), EG (10 mL) and O_2 (1.0 atm) at 80 °C for 9 h.

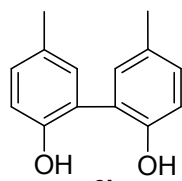
General procedure: a three-necked flask was charged with EG (10 mL), **1a**, **1b** or **1c** (5.0 mmol), $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ (12.5 mg, 0.05 mmol) and solid NaOH (0.80 g, 20 mmol), and then the solution was heated to 80 °C. The molecular oxygen was continuously supplied to the reaction through a top tube inlet for 9 h. Hydrochloric acid (10 mL, 10%) and methyl *tert*-butyl ether (MTBE, 15 mL) were successively added to the reaction mixture at room temperature. The MTBE phase was separated, and the aqueous phase was further extracted with MTBE (15 mL $\times$ 2). The combined organic layers were dried over Na_2SO_4 and concentrated in vacuo to give a residue, which was purified *via* column chromatography on silica gel (eluents: petroleum ether/ethyl acetate 20:1) to provide **2** and dimer **6**, respectively.



2a **3-Bromo-4-hydroxybenzaldehyde (2a):** pale yellow solid, 0.90 g (90%); the spectral data see 2.1 section.



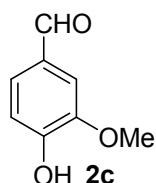
2b **4-Hydroxybenzaldehyde (2b):**⁵ yellow solid, 0.49 g (81% yield), m.p. 116–118 °C ((lit⁵ m.p. 115–118 °C); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 9.87 (br s, 1H), 7.83 (br d, J = 8.8 Hz, 2H), 7.98 (br d, J = 8.8 Hz, 2H), 6.29 (br s, 1H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 191.2, 161.5, 132.5 (2C), 129.9, 116.0 (2C); HRMS (EI): m/z [M^+] calcd. for $\text{C}_7\text{H}_6\text{O}_2$ 122.0368, found 122.0367.



6b

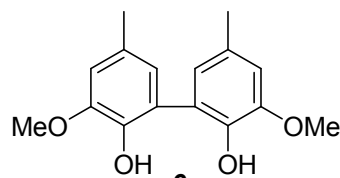
5,5'-Dimethylbiphenyl-2,2'-diol (6b):⁶ white solid, 32.1 mg (6% yield), m.p.

148–150 °C (lit⁶ m.p. 155 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 7.11 (dd, *J* = 8.0, 1.6 Hz, 2H), 7.06 (br d, *J* = 1.6 Hz, 2H), 6.92 (br d, *J* = 8.0 Hz, 2H), 5.43 (br s, 2H), 2.32 (s, 6H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 150.6 (2C), 131.6 (2C), 130.8 (2C), 130.3 (2C), 123.7 (2C), 116.5 (2C), 20.5 (2C); HRMS (EI): *m/z* [M⁺] calcd. for C₁₄H₁₄O₂ 214.0994, found 214.0992.



2c

Vanillin (2c):⁷ white solid, 0.42 g (55%), m.p. 82–83 °C (lit⁷ m.p. 80–81 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.82 (br s, 1H), 7.43–7.41 (m, 2H), 7.04 (br d, *J* = 8.8 Hz, 1H), 6.30 (br s, 1H), 3.96 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 191.1, 151.8, 147.2, 129.8, 127.6, 114.5, 108.9, 56.1; HRMS (ESI): *m/z* [M-H⁺] calcd. for C₈H₇O₃ 151.0395, found 151.0400.

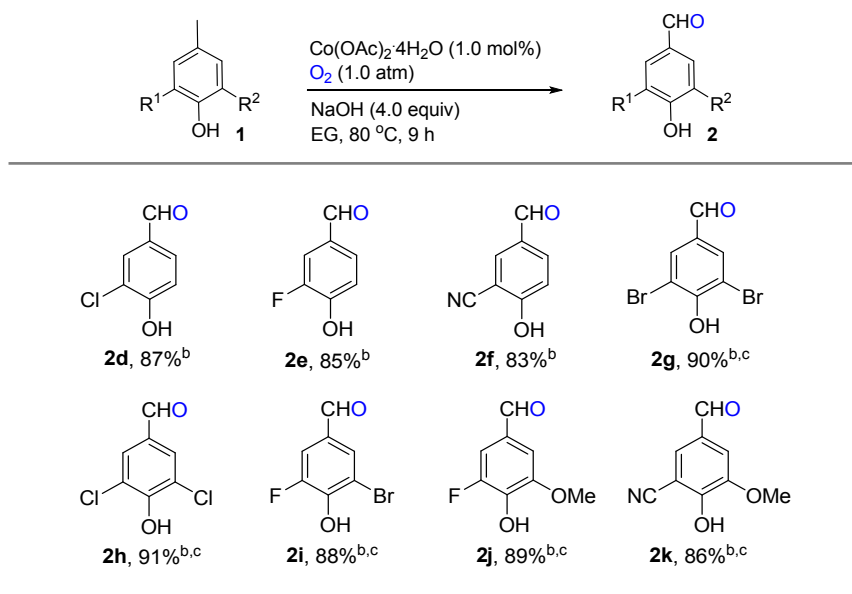


6c

3,3'-Dimethoxy-5,5'-dimethylbiphenyl-2,2'-diol (6c):⁸ brown solid, 75.4 mg (11% yield), m.p. 132–134 °C (lit⁸ m.p. 133–135 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 6.73 (br s, 2H), 6.72 (br s, 2H), 5.96 (br s, 2H), 3.91 (s, 6H), 2.33 (s, 6H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 147.1 (2C), 140.3 (2C), 129.6 (2C), 124.4 (2C), 123.4 (2C), 111.3 (2C), 56.0 (2C), 21.2 (2C); HRMS (ESI): *m/z* [M+H⁺] calcd. for C₁₆H₁₉O₄ 275.1283, found 275.1283.

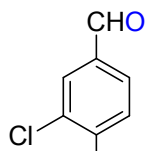
2.6 Co(OAc)₂·4H₂O-catalyzed oxidation of **1** (Table 2 in the text).

Table 2. Co(OAc)₂-catalyzed oxidation of **1**.^a

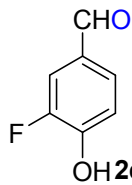


^aReaction conditions: substrates **1** (5.0 mmol), Co(OAc)₂·4H₂O (0.05 mmol), NaOH (20.0 mmol), EG (10 mL) and O₂ (1.0 atm) at 80 °C for 9 h. ^b Isolated yield *via* column chromatography. ^cPerformed with NaOH (10.0 mmol).

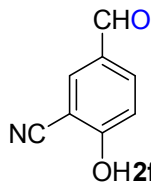
General procedure: a three-necked flask was charged with EG (10 mL), **1** (5.0 mmol), Co(OAc)₂·4H₂O (12.5 mg, 0.05 mmol) and solid NaOH (0.80 g, 20 mmol for **1d**, **1e**, **1f**; 0.40 g, 10 mmol for **2g**, **2h**, **2i**, **2j**, **2k**), and then the solution was heated to 80 °C. The molecular oxygen was continuously supplied to the reaction through a top tube inlet for 9 h. Hydrochloric acid (10 mL, 10%) and methyl *tert*-butyl ether (MTBE, 15 mL) were successively added to the reaction mixture at room temperature. The MTBE phase was separated, and the aqueous phase was further extracted with MTBE (15 mL × 2). The combined organic layers were dried over anhydrous Na₂SO₄ and concentrated in vacuo to give a residue, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 20:1) to provide the desired products **2**.



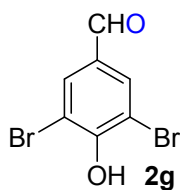
2d **3-Chloro-4-hydroxybenzaldehyde (2d):** yellow oil, 0.68 g (87%); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.85 (br s, 1H), 7.54 (br d, *J* = 2.4 Hz, 1H), 7.47 (dd, *J* = 8.8, 2.4 Hz, 1H), 6.96 (br d, *J* = 8.8 Hz, 1H), 5.98 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 195.6, 160.3, 137.1, 132.7, 124.8, 121.3, 119.6; HRMS (EI): *m/z* [*M*⁺] calcd. for C₇H₅ClO₂ 155.9978, found 155.9977.



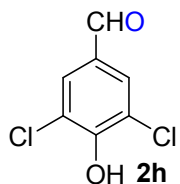
2e **3-Fluoro-4-hydroxybenzaldehyde (2e):** yellow oil, 0.60 g (85%); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.85 (br s, 1H), 7.67-7.61 (m, 2H), 7.15 (br d, *J* = 8.0 Hz, 1H), 5.98 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 190.2 (d, *J* = 2.2 Hz), 151.2 (d, *J* = 239.8 Hz), 149.6 (d, *J* = 14.7 Hz), 130.2 (d, *J* = 4.8 Hz), 128.6 (d, *J* = 2.8 Hz), 117.6 (d, *J* = 1.9 Hz), 115.8 (d, *J* = 18.3 Hz); HRMS (EI): *m/z* [*M*⁺] calcd. for C₇H₅FO₂ 140.0274, found 140.0270.



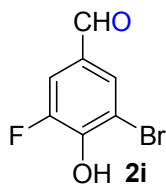
2f **5-Formyl-2-hydroxybenzonitrile (2f):** yellow oil, 0.61 g (83%); ¹H NMR (400 MHz, CDCl₃, ppm): δ 12.35 (br s, 1H), 9.83 (br s, 1H), 8.24 (br d, *J* = 2.4 Hz, 1H), 8.02 (dd, *J* = 8.8, 2.0 Hz, 1H), 7.18 (br d, *J* = 8.8 Hz, 1H); ¹³C NMR (100 MHz, DMSO-*d*₆, ppm): δ 190.5, 165.4, 137.4, 135.3, 128.9, 117.3, 116.4, 100.1; HRMS (ESI): *m/z* [*M*-H⁺] calcd. for C₈H₄NO₂ 146.0242, found 146.0258.



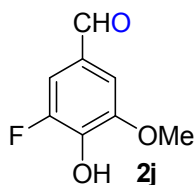
3,5-Dibromo-4-hydroxybenzaldehyde (2g): white solid, 1.26 g (90%); the spectra see 2.1 section.



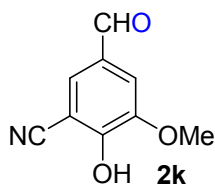
3,5-Dichloro-4-hydroxybenzaldehyde (2h):⁹ white solid, 0.87 g (91% yield), m.p. 158–160 °C (lit⁹ m.p. 157 °C); ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.81 (br s, 1H), 7.83 (br s, 2H), 6.43 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 188.6, 153.1, 130.3 (2C), 129.9 (2C), 122.4; HRMS (ESI): m/z [M–H⁺] calcd. for C₇H₃Cl₂O₂ 188.9510, found 188.9511.



3-Bromo-5-fluoro-4-hydroxybenzaldehyde (2i): white solid, 0.96 g (88% yield), m.p. 138–140 °C; ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.74 (d, J = 2.0 Hz, 1H), 7.78 (t, J = 1.6 Hz, 1H), 7.54 (dd, J = 9.6, 1.6 Hz, 1H), 6.33 (br s, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 188.8, 151.2 (d, J = 247.3 Hz), 147.3 (d, J = 15.0 Hz), 130.7 (d, J = 2.7 Hz), 130.1 (d, J = 5.4 Hz), 115.7 (d, J = 18.8 Hz), 111.6 (d, J = 1.4 Hz); HRMS (ESI): m/z [M–H⁺] calcd. for C₇H₃BrFO₂ 216.9300, found 216.9296.

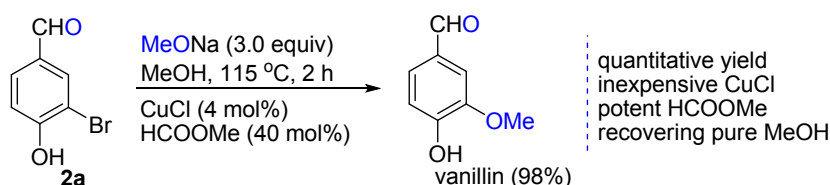


3-Fluoro-4-hydroxy-5-methoxybenzaldehyde (2j): white solid, 0.76 g (89% yield), m.p. 116–118 °C; ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.81 (br s, 1H), 7.32–7.26 (m, 2H), 5.98 (br s, 1H), 4.00 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 190.2 (d, J = 2.3 Hz), 150.6 (d, J = 243.8 Hz), 148.8 (d, J = 5.3 Hz), 140.1 (d, J = 13.5 Hz), 128.2 (d, J = 6.3 Hz), 113.1 (d, J = 18.5 Hz), 106.1 (d, J = 1.7 Hz), 56.8; HRMS (ESI): m/z [M–H⁺] calcd. for C₈H₆FO₃ 169.0301, found 169.0305.

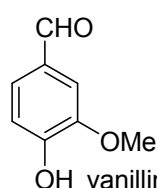


5-Formyl-2-hydroxy-3-methoxybenzonitrile (2k): white solid, 0.76 g (86% yield), m.p. 196–198 °C; ¹H NMR (400 MHz, CDCl₃, ppm): δ 9.83 (br s, 1H), 7.66 (br s, 1H), 7.58 (br s, 1H), 6.90 (br s, 1H), 4.03 (s, 3H); ¹³C NMR (100 MHz, DMSO-*d*₆, ppm): δ 190.2, 155.6, 148.5, 129.2, 128.7, 115.8, 113.1, 99.3, 56.4; HRMS (ESI): m/z [M–H⁺] calcd. for C₉H₆NO₃ 176.0348, found 176.0361.

2.7 Methoxylation of **2a** for preparing vanillin (Scheme 7 in the text).



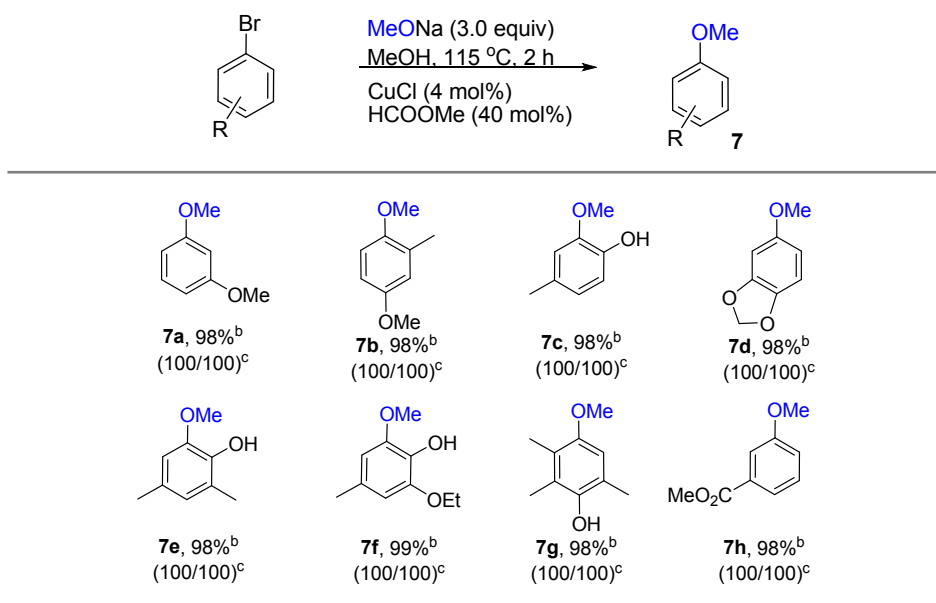
General procedure: the Teflon-lined autoclave (25 mL) was charged with **2a** (0.80 g, 4.0 mmol), MeOH (5 mL), MeONa (0.65 g, 12 mmol), CuCl (15.8 mg, 0.16 mmol) and HCOOMe (0.10 mL, $d = 0.97$ g/mL, 1.6 mmol). The autoclave was heated to 115 °C and stirred for 2 h. After the completion of reaction, the reactor was cooled to room temperature. The reaction mixture was stirred for 0.5 h in open system, and then concentrated to recover pure MeOH. To the residue was added MTBE (5 mL) and diluted hydrochloric acid (1.0 M, 8 mL) to adjust pH to 2.0–3.0. Furthermore, the solution was partitioned into two layers, and the aqueous phase was extracted with MTBE (5 mL $\times$ 3). The combined organic layers were dried over anhydrous Na₂SO₄, and concentrated in vacuo to give a solid, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 15:1) to provide the desired vanillin. Besides, the purity of the recovered MeOH was more than 99% measured by GC, and water content of the recovered MeOH was less than 0.12% measured by Karl Fischer method.



Vanillin: white solid, 596.4 mg (98% yield); the spectral data see 2.5 section.

2.8 CuCl/HCOOMe-catalyzed methoxylation (Table 3 in the text).

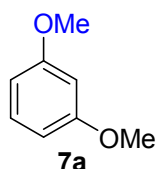
Table 3. CuCl/HCOOMe-catalyzed methoxylation.^a



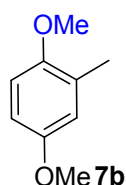
^a Reaction conditions: aryl bromides (4.0 mmol), MeOH (5 mL), MeONa (0.65 g, 12 mmol), CuCl (15.8 mg, 0.16 mmol) and HCOOMe (0.10 mL, $d = 0.97$ g/mL, 1.6 mmol). ^b Isolated yield *via*

column chromatography. ° Conv. (%) / Sele. (%) determined by GC-MS by area normalization method in the total ionization chromatography.

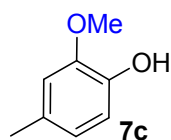
General procedure: the Teflon-lined (25 mL) was charged with the aryl bromides (4.0 mmol), MeOH (5 mL), freshly prepared MeONa (0.65 g, 12 mmol), CuCl (15.8 mg, 0.16 mmol) and HCOOMe (0.10 mL, $d = 0.97$ g/mL, 1.6 mmol). The autoclave was heated to 110 °C and stirred for 2 h. After the completion of reaction, the reactor was cooled to room temperature. The reaction mixture was stirred for 0.5 h in open system, and then concentrated to recover pure MeOH. To the residue was added MTBE (5 mL) and diluted hydrochloric acid (1.0 M, 8 mL) to adjust pH to 2.0–3.0. Furthermore, the solution was partitioned into two layers, and the aqueous phase was extracted with MTBE (5 mL $\times$ 3). The combined organic layers were dried over anhydrous Na₂SO₄, and concentrated to give a crude product, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 15:1) to provide the desired vanillin. Generally, the purity of the recovered MeOH was more than 99% measured by GC, and water content of the recovered MeOH was less than 0.12% measured by Karl Fischer method. In particular, the sample of the crude product was used to determine conversion and selectivity by GC-MS by area normalization method in the total ionization chromatography.



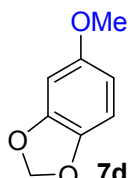
1,3-Dimethoxybenzene (7a): colorless oil, 541.6 mg (98% yield); ¹H NMR (400 MHz, CDCl₃, ppm): δ 7.19 (t, $J = 8.0$ Hz, 1H), 6.52 (dd, $J = 8.0, 2.0$ Hz, 2H), 6.48 (t, $J = 2.0$ Hz, 1H), 3.80 (s, 6H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 160.9, 129.9, 106.2 (2C), 100.5 (2C), 55.2 (2C); HRMS (EI): m/z [M^+] calcd. for C₈H₁₀O₂ 138.0681, found 138.0680.



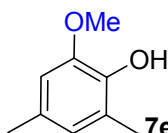
1,4-Dimethoxy-2-methylbenzene (7b): colorless oil, 596.6 mg (98% yield); ¹H NMR (400 MHz, CDCl₃, ppm): δ 6.76 (br d, $J = 8.8$ Hz, 2H), 6.69 (dd, $J = 8.8, 3.2$ Hz, 1H), 3.79 (s, 3H), 3.76 (s, 3H), 2.22 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 153.5, 152.2, 127.8, 117.1, 110.9, 110.8, 55.8, 55.6, 16.4; HRMS (ESI): m/z [$M+H^+$] calcd. for C₉H₁₃O₂ 153.0916, found 153.0924.



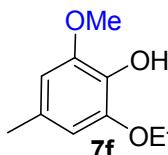
4-Methyl-2-methoxyphenol (7c): colorless oil, 541.6 mg (98% yield); ¹H NMR (400 MHz, CDCl₃, ppm): δ 6.82 (br d, $J = 8.0$ Hz, 1H), 6.69 (br s, 1H), 6.68 (br d, $J = 8.0$ Hz, 1H), 5.50 (br s, 1H), 3.87 (s, 3H), 2.30 (s, 3H); ¹³C NMR (100 MHz, CDCl₃, ppm): δ 146.3, 143.4, 129.7, 121.6, 111.8, 55.9, 21.1; HRMS (EI): m/z [M^+] calcd. for C₈H₁₀O₂ 138.0681, found 138.0682.



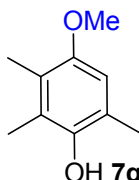
7d 5-Methoxybenzo[d][1,3]-dioxole (7d): pale yellow oil, 596.4 mg (98% yield); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 6.71 (br d, $J = 8.4$ Hz, 1H), 6.49 (br s, 1H), 6.32 (br d, $J = 8.4$ Hz, 1H), 5.91 (s, 2H), 3.75 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 155.2, 148.3, 141.6, 107.9, 104.7, 101.1, 97.5, 56.0; HRMS (ESI): m/z $[\text{M}+\text{H}^+]$ calcd. for $\text{C}_8\text{H}_9\text{O}_3$ 153.0552, found 153.0531.



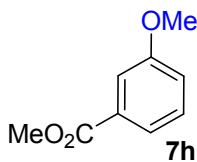
7e 2-Methoxy-4,6-dimethylphenol (7e):¹⁰ pale yellow solid, 596.6 mg (98% yield), m.p. 28–30 °C (lit¹⁰ m.p. 34–35 °C); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 6.57 (br s, 1H), 6.55 (br s, 1H), 5.53 (br s, 1H), 3.86 (s, 3H), 2.27 (s, 3H), 2.24 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 146.1, 141.5, 128.6 (2C), 123.6, 109.2, 56.0, 21.1, 15.4; HRMS (ESI): m/z $[\text{M}+\text{H}^+]$ calcd. for $\text{C}_9\text{H}_{13}\text{O}_2$ 153.0916, found 153.0893.



7f 2-Ethoxy-6-methoxy-4-methylphenol (7f): pale yellow oil, 721.6 mg (99% yield); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 6.38 (br s, 2H), 5.36 (br s, 1H), 4.09 (q, $J = 6.8$ Hz, 2H), 3.87 (s, 3H), 2.28 (s, 3H), 1.43 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 146.9, 146.1, 132.7, 128.7, 106.7, 105.7, 64.7, 56.2, 21.6, 15.0; HRMS (ESI): m/z $[\text{M}+\text{H}^+]$ calcd. for $\text{C}_{10}\text{H}_{15}\text{O}_3$ 183.1021, found 183.1023.



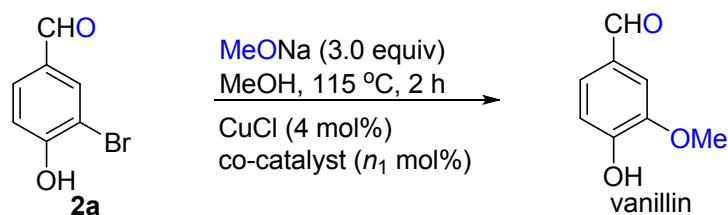
7g 4-Methoxy-2,3,6-trimethylphenol (7g):¹¹ white solid, 651.6 mg (98% yield), m.p. 104–106 °C (lit¹¹ m.p. 101–102 °C); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 6.52 (s, 1H), 4.26 (br s, 1H), 3.76 (s, 3H), 2.24 (s, 3H), 2.18 (s, 3H), 2.14 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 151.3, 145.9, 123.9, 123.6, 120.0, 110.9, 56.3, 16.3, 12.2, 11.9; HRMS (ESI): m/z $[\text{M}+\text{H}^+]$ calcd. for $\text{C}_{10}\text{H}_{15}\text{O}_2$ 167.1072, found 167.1071.



7h Methyl 3-methoxybenzoate (7h): colorless oil, 651.4 mg (98% yield); ^1H NMR (400 MHz, CDCl_3 , ppm): δ 7.64 (dt, $J = 8.0, 1.2$ Hz, 1H), 7.56 (dd, $J = 2.8, 1.6$ Hz, 1H), 7.34 (t, $J = 8.0$ Hz, 1H), 7.10 (ddd, $J = 8.0, 2.8$ Hz, 2.4 Hz, 1H), 3.92 (s, 3H), 3.85 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3 , ppm): δ 167.0, 159.5, 131.4, 129.4, 122.0, 119.5, 113.9, 55.4, 52.2; HRMS (ESI): m/z $[\text{M}+\text{H}^+]$ calcd. for $\text{C}_9\text{H}_{11}\text{O}_3$ 167.0708, found 167.0719.

2.9 CuCl-catalyzed methoxylation of **2a** with different co-catalysts (Table 4 in the text).

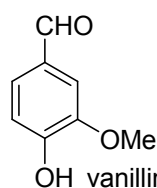
Table 4. CuCl-catalyzed methoxylation of **2a** with different co-catalysts.^a



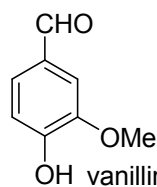
Entry	Co-catalyst	n_1 mol%	Conv./Sele. (%) ^b	Yield (%) ^c
1	HCOOMe	40	100/100	98
2	MeCOOMe	40	100/95	91
3	DMF	40	100/94	89

^a Reaction conditions: **2a** (4.0 mmol), MeOH (5 mL), MeONa (0.65 g, 12 mmol), CuCl (15.8 mg, 0.16 mmol) and co-catalyst (n_1 mol%). ^b Determined by GC-MS. ^c Isolated yield *via* column chromatography.

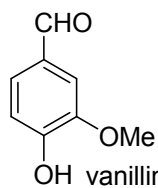
General procedure: the Teflon-lined autoclave (25 mL) was charged with **2a** (0.80 g, 4.0 mmol), MeOH (5 mL), MeONa (0.65 g, 12 mmol), CuCl (15.8 mg, 0.16 mmol) and co-catalyst (n_1 mol%). The autoclave was heated to 115 °C and stirred for 2 h. After the completion of reaction, the reactor was cooled to room temperature. The reaction mixture was stirred for 0.5 h in open system, and then concentrated to recover pure MeOH. To the residue was added MTBE (5 mL) and diluted hydrochloric acid (1.0 M, 8 mL) to adjust pH to 2.0–3.0. Furthermore, the solution was partitioned into two layers, and the aqueous phase was extracted with MTBE (5 mL $\times$ 3). The combined organic layers were dried over anhydrous Na₂SO₄, and concentrated in vacuo to give a crude product, which was purified *via* column chromatography on silica gel (eluent: petroleum ether/ethyl acetate 15:1) to provide the desired vanillin. In particular, the sample of the crude product was used to determine conversion and selectivity with GC-MS by area normalization method in the total ionization chromatography.



Vanillin: white solid, 596.4 mg (98% yield with HCOOMe as co-catalyst); the spectral data see 2.5 section.



Vanillin: white solid, 553.8 mg (91% yield with MeCOOMe as co-catalyst); the spectral data see 2.5 section.



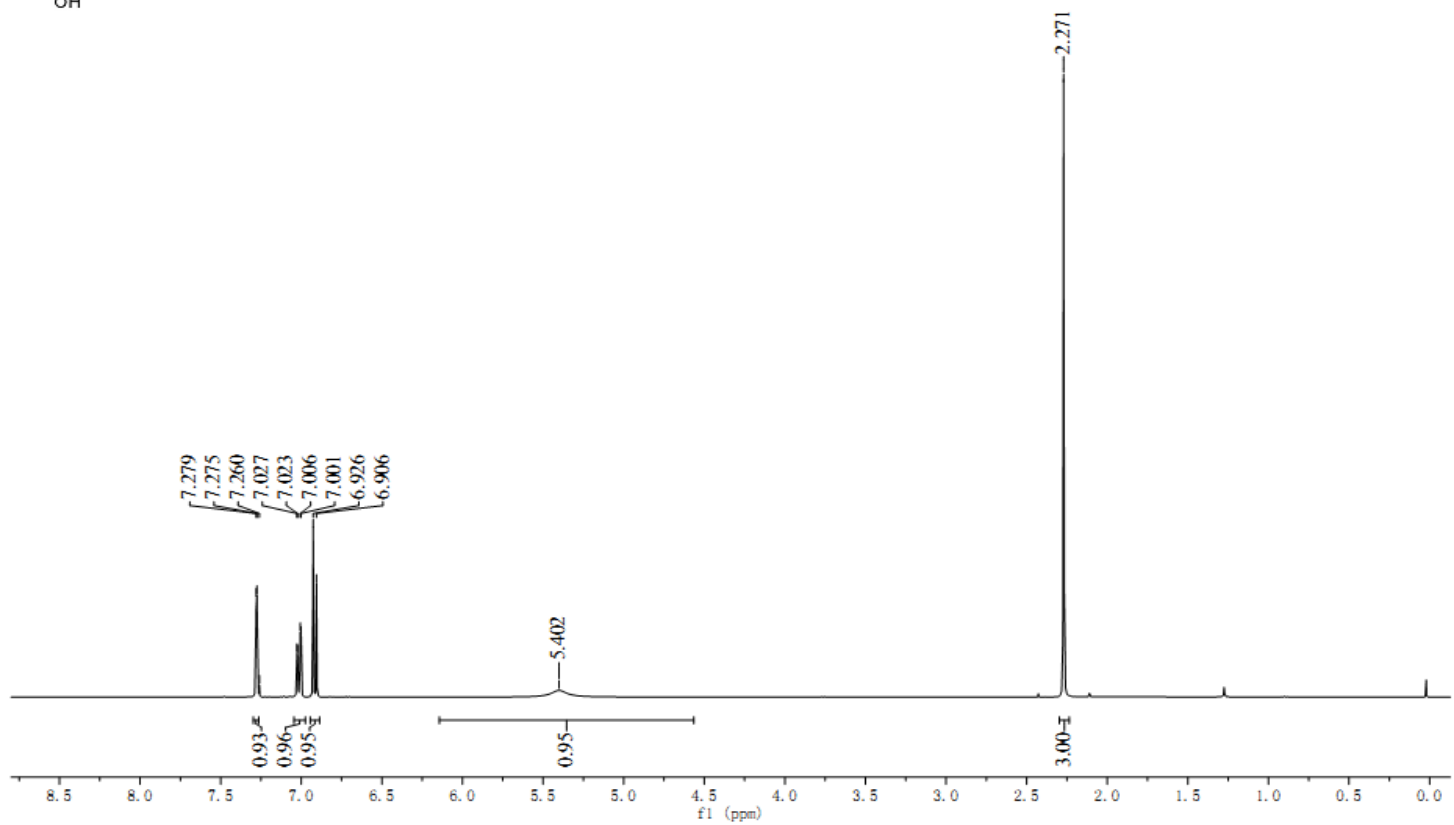
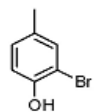
vanillin **vanillin**: white solid, 541.6 mg (89% yield with DMF as co-catalyst); the spectral data see 2.5 section.

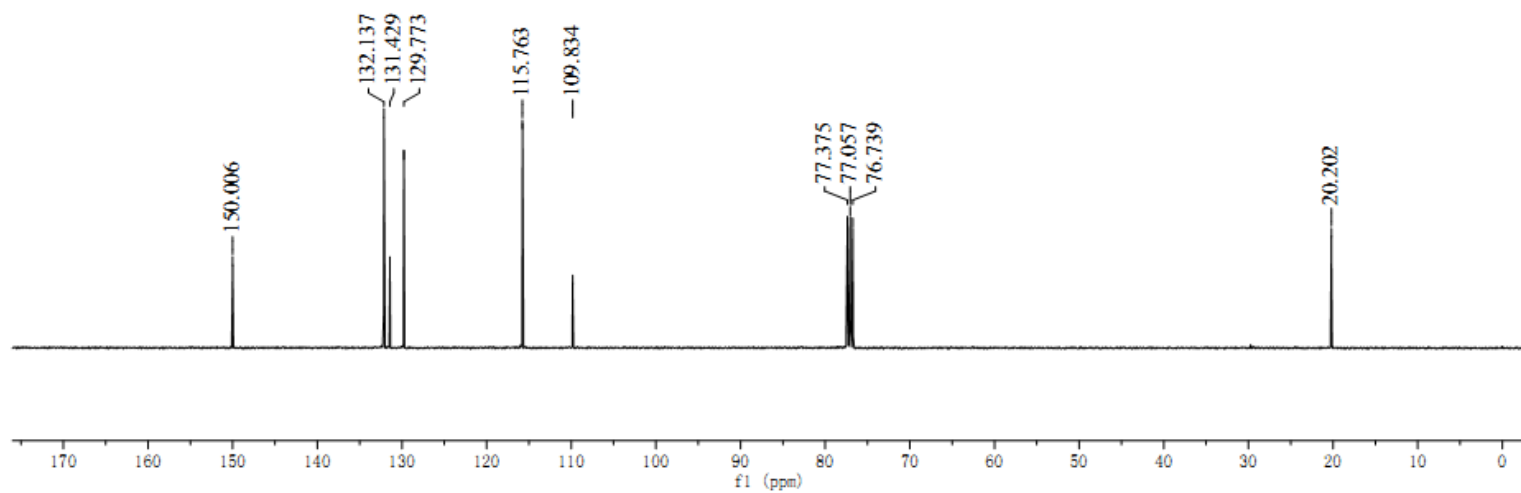
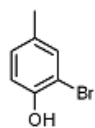
3. References

- [1] S. Adimurthy, G. Ramachandraiah, A. V. Bedekar, S. Ghosh, B. C. Ranu and P. K. Ghosh, *Green Chem.*, 2006, **8**, 916.
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- [11] K. A. Kun and H. G. Cassidy, *J. Org. Chem.*, 1962, **27**, 841.

4. Copies of Spectra

4.1 Copies of spectra for oxybromination of 4-cresol into 1a.





Elemental Composition Report

Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions

72 formula(e) evaluated with 11 results within limits (up to 50 closest results for each mass)

Elements Used:

C: 0-7 H: 0-7 O: 0-1 79Br: 0-1 81Br: 0-1

lyf-ja-1

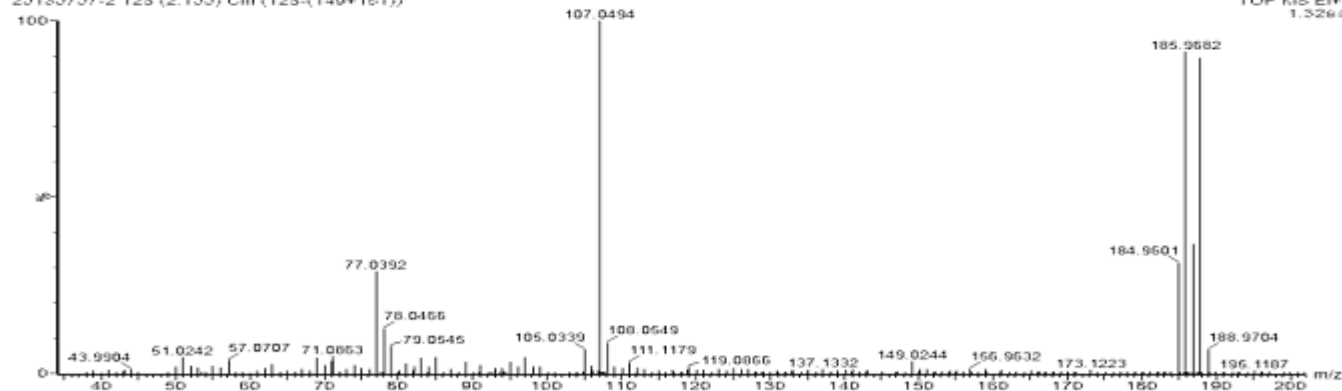
20130707-2 12S (2.100) Cm (12S-(149+161))

Waters GCT Premier

02-Jul-2013 16:24:06

TOF MS ESI+

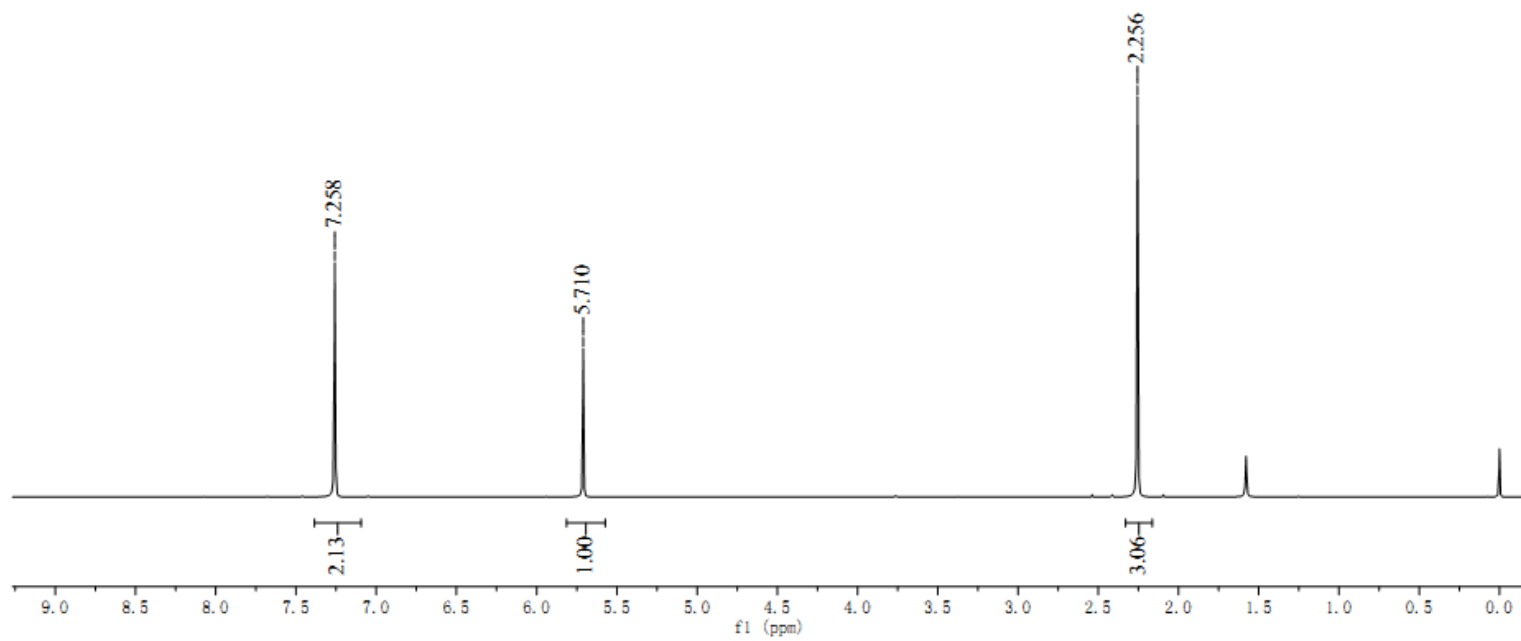
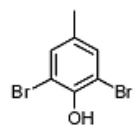
1.3264

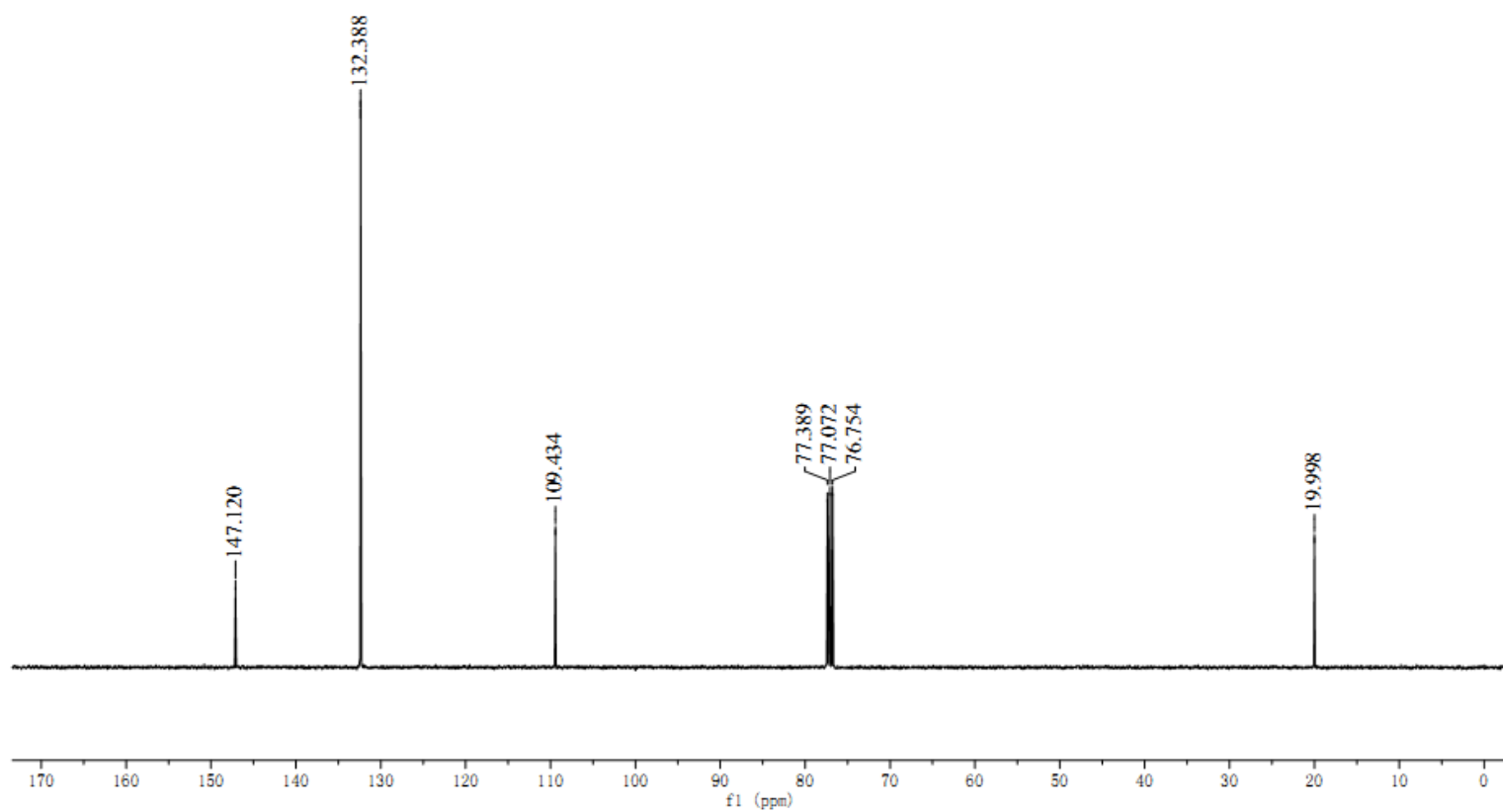
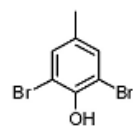


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
185.9682	91.31	185.9680	0.2	1.1	4.0	1594.1	C7 H7 O 79Br





Elemental Composition Report

Single Mass Analysis

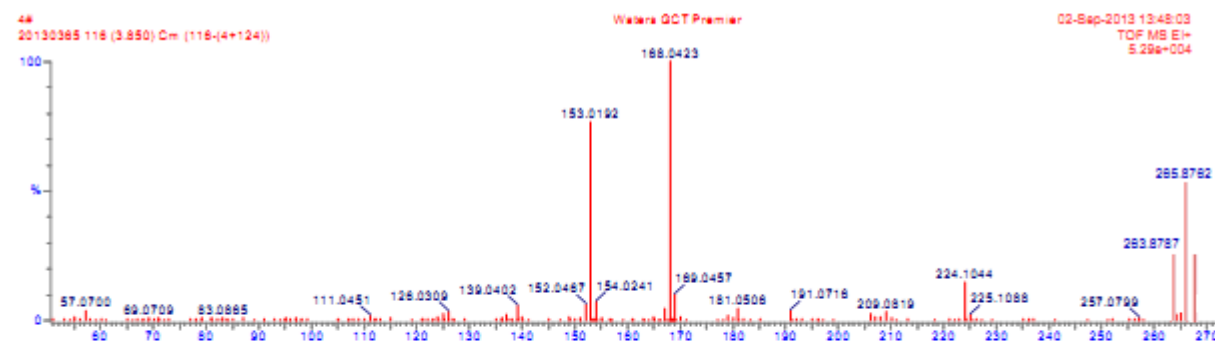
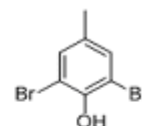
Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Monoisotopic Mass, Odd and Even Electron Ions

19 formula(e) evaluated with 5 results within limits (up to 50 closest results for each mass)

Elements Used:

C: 0-7 H: 0-6 Br: 0-2 O: 0-4

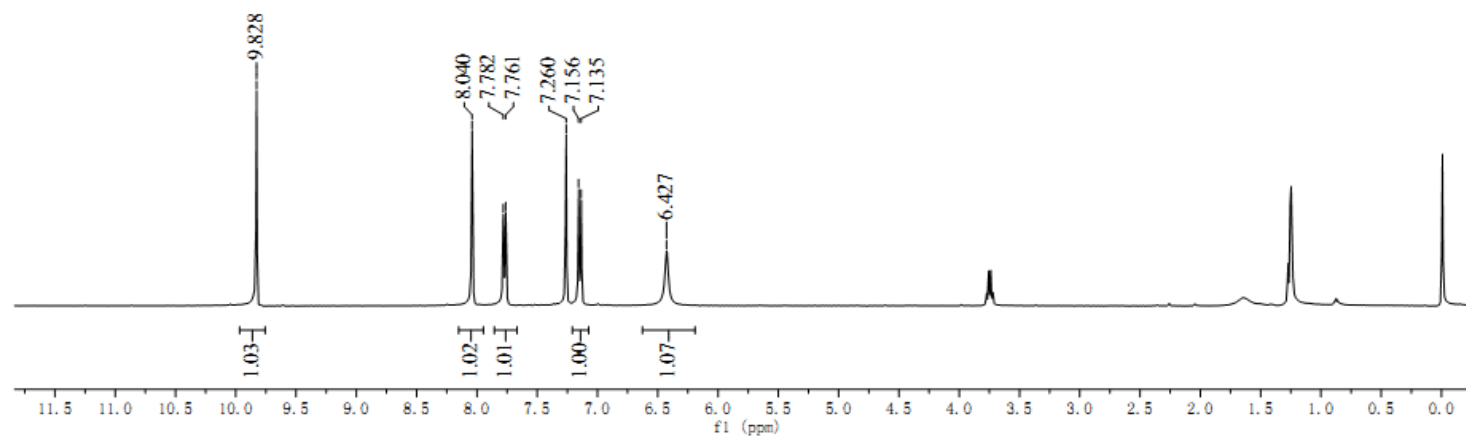
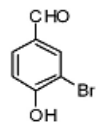


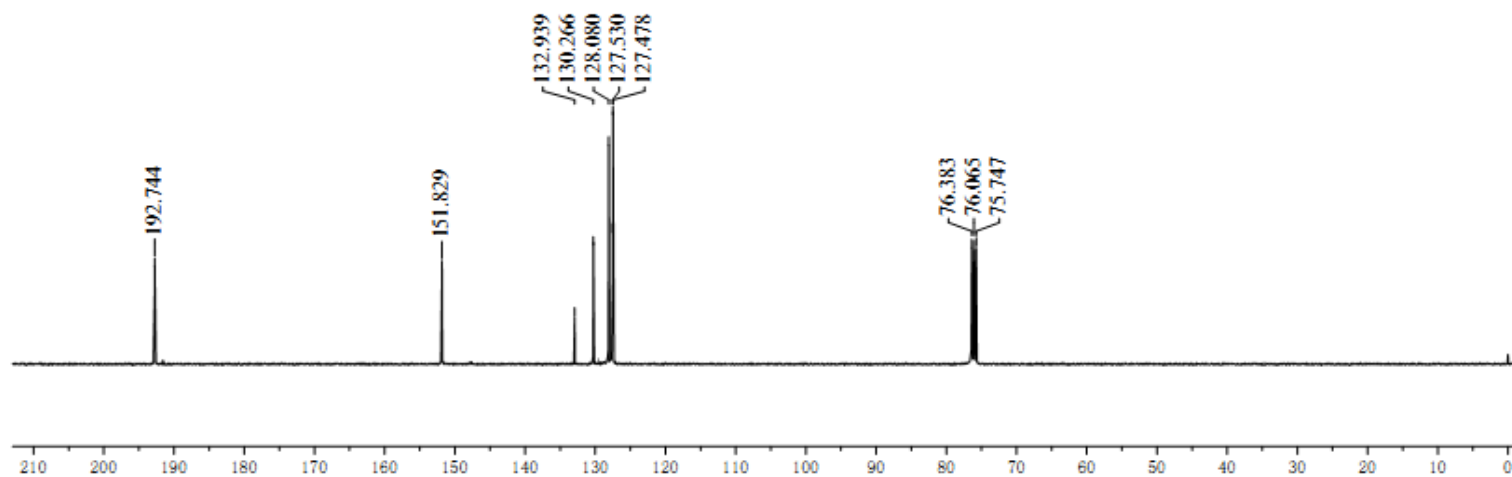
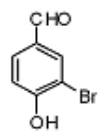
Minimum: -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	Score	Formula
263.8787	25.00	263.8785	0.2	0.4	5.0	1	C7 H6 Br2 O4

4.2 Copies of spectra for oxybromination of 4-hydroxybenzaldehyde into 2a.





Elemental Composition Report

Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions

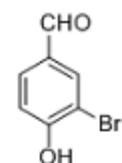
85 formula(e) evaluated with 12 results within limits (up to 50 closest results for each mass)

Elements Used:

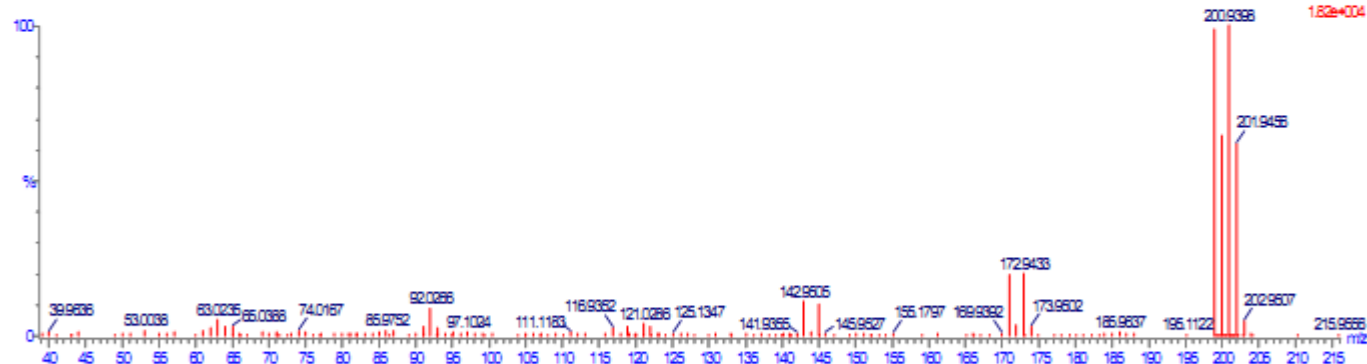
C: 0-7 H: 0-5 O: 0-2 79Br: 0-1 81Br: 0-1

jjf@t1
20130713 94(1.558) Cm(94@11)

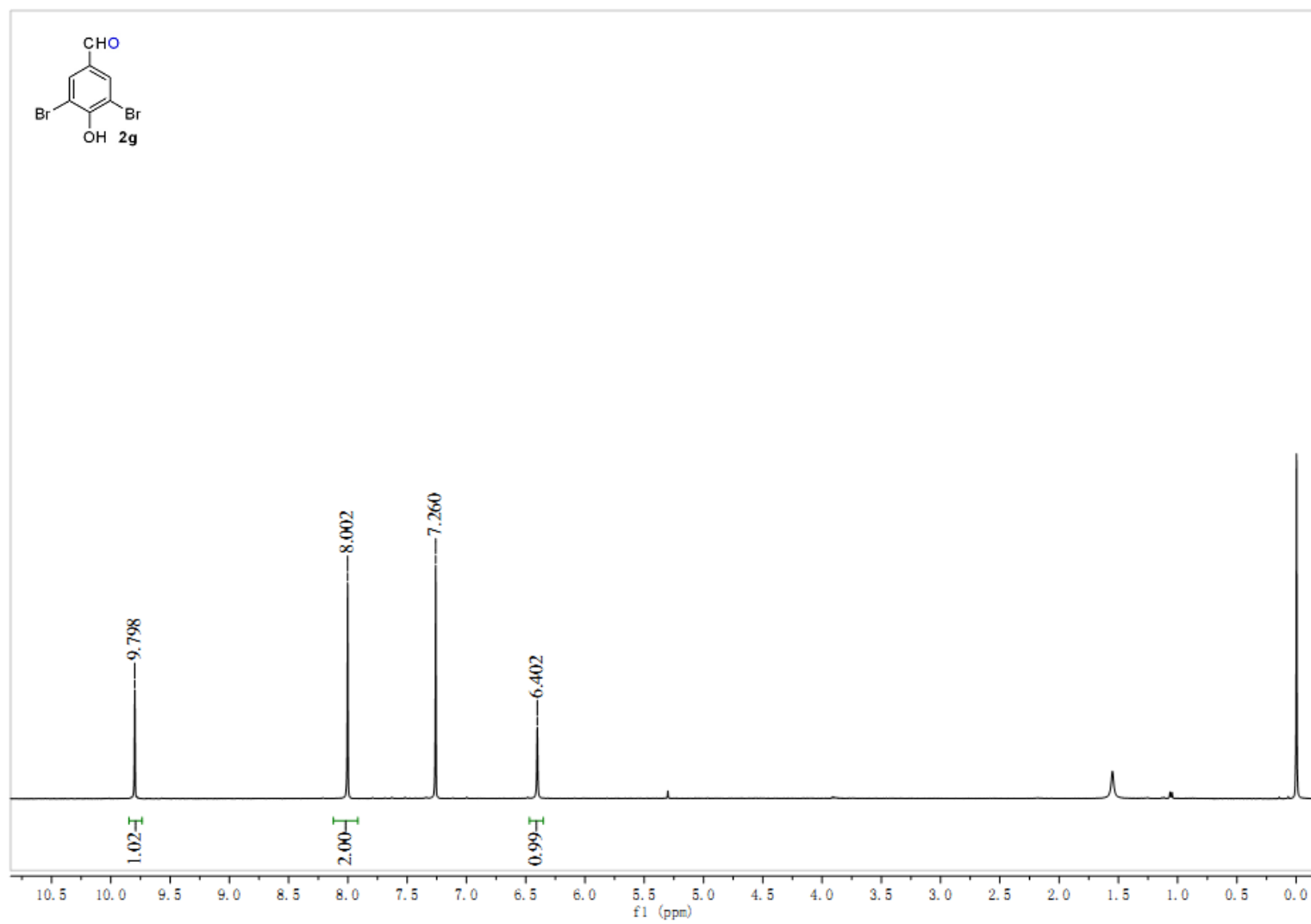
Waters GCT Premier

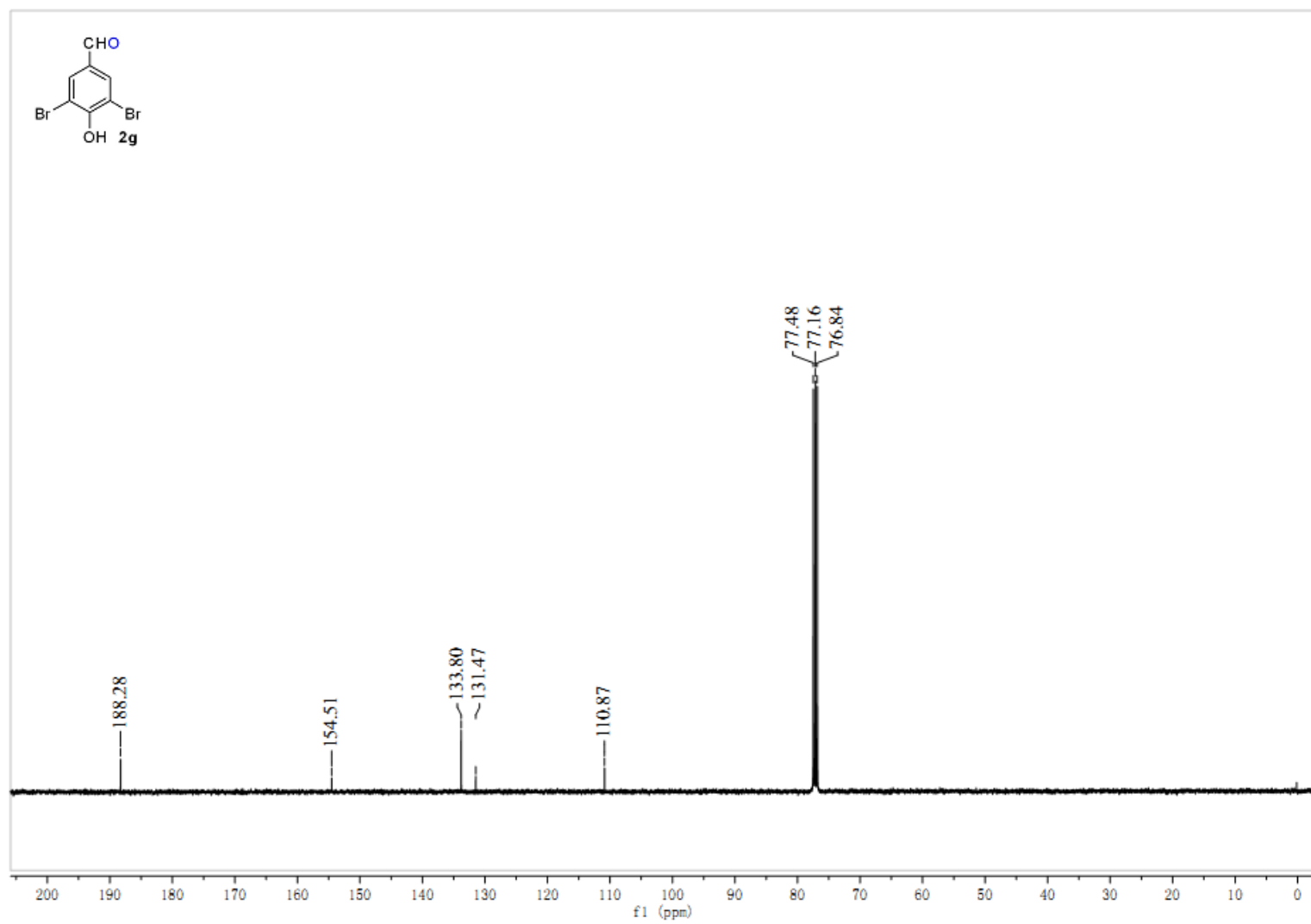


09-JU-2013 173420
TOF/MSB+
1.82e+004



Minimum:	3.00				-1.5				
Maximum:	100.00		5.0	10.0	50.0				
Mass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula		
199.9474	64.49	199.9473	0.1	0.5	5.0	8332.2	C7	H5	O2 79Br





Elemental Composition Report

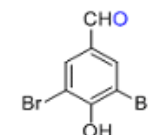
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

28 formula(e) evaluated with 1 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 O: 0-2 Br: 0-2

YF-JI

ECUST institute of Fine Chem

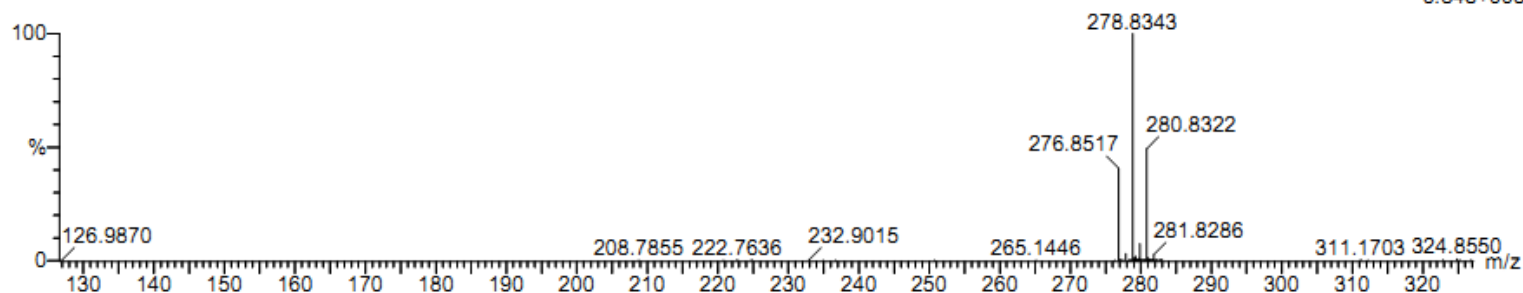
16-Dec-2013

21:30:33

JYF-JA-1 14 (0.541) Cm (14)

2: TOF MS ES-

6.84e+003



Minimum:

-1.5

Maximum:

30.0

50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
------	------------	-----	-----	-----	-------	--------------	---------

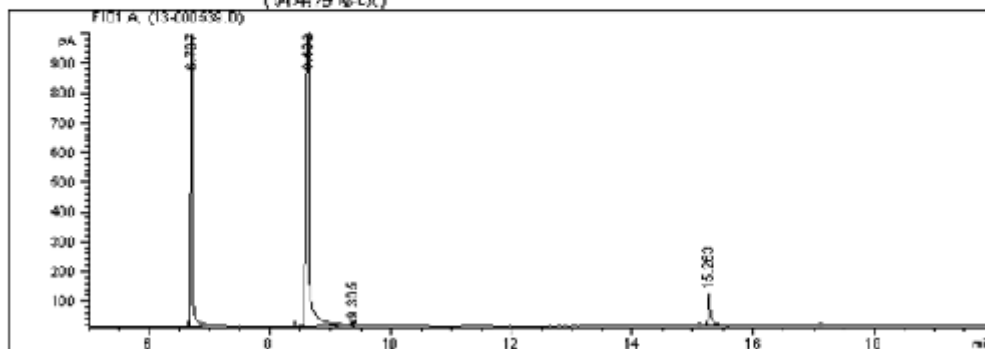
276.8517	276.8500	1.7	6.1	5.5	46.5	0.0	C7 H3 O2 Br2
----------	----------	-----	-----	-----	------	-----	--------------

4.3 Copies of spectra for the oxidation process of 1a in EG.

数据文件: C:\CHEM32\1\DATA\13-000539.D
样品名称: 5#

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操作者      :
仪器        : 仪器 1                      位置 : 样品瓶 101
进样日期    : 12-Nov-13, 10:54:06          进样次数 : 1
                                          进样量 : 手动

采集方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-12 10:50:36
分析方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-12 14:16:38
              (调用后修改)
=====
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面积百分比报告

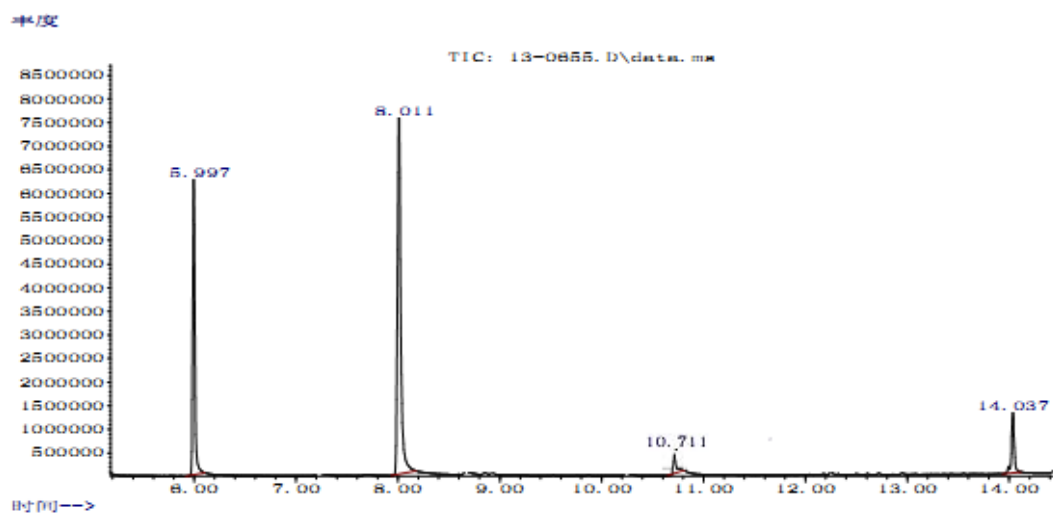
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=====
排序      :      信号
乘积因子:      :      1.0000
稀释因子:      :      1.0000
内标使用乘积因子和稀释因子
=====
```

信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	6.707	BV	0.0280	3055.84058	1715.18091	58.19582
2	8.632	BV	0.0525	4650.44141	1405.76208	37.58209
3	9.365	VV	0.0262	17.95808	10.39642	0.22228
4	15.263	VB	0.0437	324.76053	106.20811	4.01981

总量 : 8079.00059 3317.54752

*** 报告结束 ***



面积百分比报告

数据路径 : E:\2013-2\

数据文件 : 13-0655.D

采集 : 25 Oct 2013 14:54

样品 : 药学院

峰 #	R.T. 分钟	起始 扫描	TOP 扫描	截止 扫描	峰 峰类型	峰高	修正面积 总面积	% 比总数
1	5.997	744	758	778	BB	6086961	93257278	32.515%
2	8.011	1134	1149	1181	BB	7613886	170161516	53.855%
3	10.711	1660	1672	1692	BB	1414163	28206104	2.927%
4	14.037	2298	2318	2332	BB	1278196	24336551	10.702%

修正后的面积和: 315961448

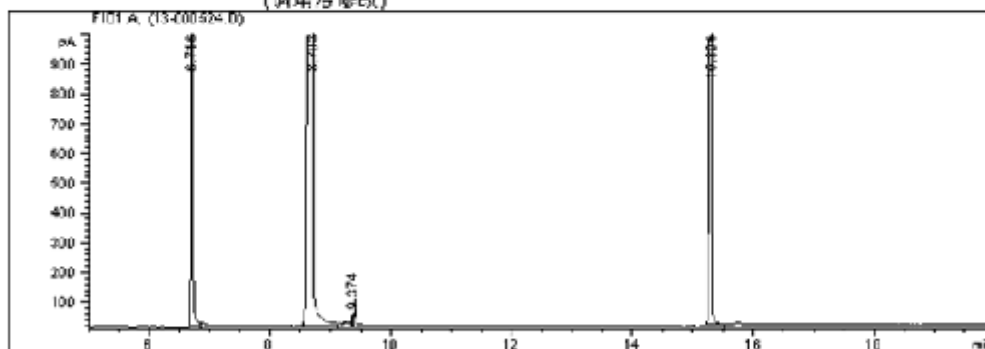
源文件: C:\CHEM32\1\DATA\13-000524.D

样品名称: 1#

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操作者	:		位置	:	样品瓶 101
仪器	:	仪器 1	进样次数	:	1
进样日期	:	08-Nov-13, 09:33:15	进样量	:	手动
采集方法	:	C:\CHEM32\1\METHODS\含溴化合物.M			
最后修改	:	2013-11-8 9:29:48			
分析方法	:	C:\CHEM32\1\METHODS\含溴化合物.M			
最后修改	:	2013-11-13 14:14:43			

(调用后修改)



=====
面积百分比报告
=====

排序 : 信号
乘积因子: : 1.0000
稀释因子: : 1.0000
内标使用乘积因子和稀释因子

信号 1: FID1 A,

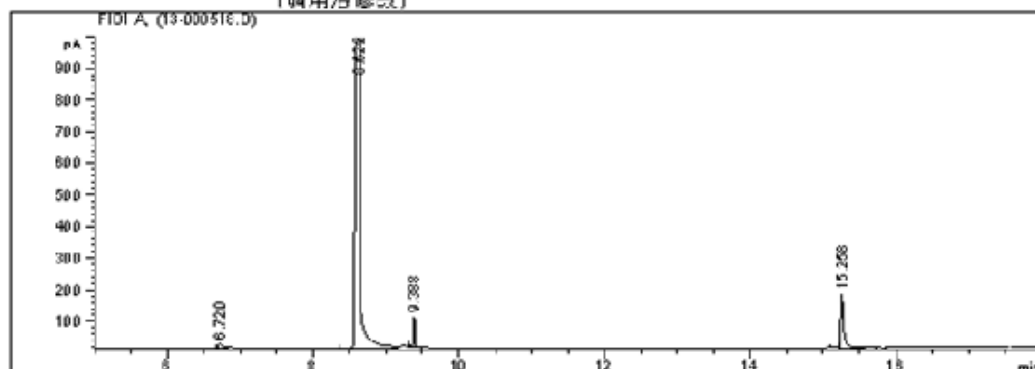
峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	6.716	BV	0.0274	3270.10156	1823.35535	12.87977
2	9.374	BV	0.0606	1.31134e4	2701.00024	67.67934
3	9.374	VV	0.0228	54.44823	42.31525	2.31296
4	15.306	BB	0.0355	4144.91748	1585.42542	17.12792

总量 : 2.05929e4 6152.09626

=====
*** 报告结束 ***
=====

数据文件: C:\CHEM32\1\DATA\13-000516.D

操作者 :
仪器 : 仪器 1 位置 : 样品瓶 101
进样日期 : 05-Nov-13, 14:06:02 进样次数 : 1
进样量 : 手动
采集方法 : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改 : 2013-11-5 10:49:22
分析方法 : C:\CHEM32\1\METHODS\甲酯化.M
最后修改 : 2013-11-5 15:23:01
(调用后修改)



=====
面积百分比报告
=====

排序 : 信号
乘积因子: : 1.0000
稀释因子: : 1.0000
内标使用乘积因子和稀释因子

信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	6.720	BV	0.0671	72.11262	15.27919	1.13594
2	8.622	BV	0.0507	5662.54443	1585.15845	89.19833
3	9.300	VD	0.0265	160.89392	96.40725	2.53445
4	15.258	VB	0.0394	452.71216	167.94901	7.13128

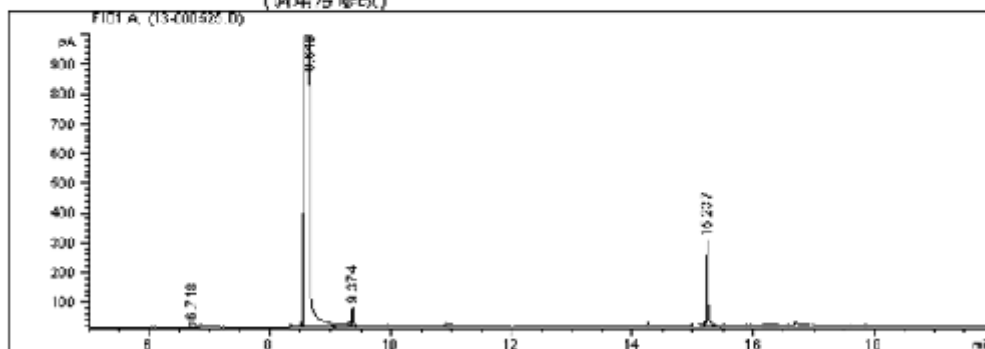
总量 : 6346.25313 1864.87390

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*** 报告结束 ***
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 样品名称: 4#

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操作者      :
仪器        : 仪器 1                      位置 : 样品瓶 101
进样日期    : 08-Nov-13, 10:13:02          进样次数 : 1
                                           进样量 : 手动

采集方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-8 10:08:57
分析方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-13 14:13:18
              (调用后修改)
=====
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=====
 面积百分比报告
 =====

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排序      :      信号
乘积因子:      :      1.0000
稀释因子:      :      1.0000
内标使用乘积因子和稀释因子
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信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	6.718	BV	0.0613	80.06580	13.40467	0.55385
2	8.640	BV	0.0573	1.01005e4	2290.01299	93.20000
3	9.374	VB	0.0248	96.72139	65.26538	0.89185
4	15.237	VB	0.0314	579.80878	288.46210	5.34630

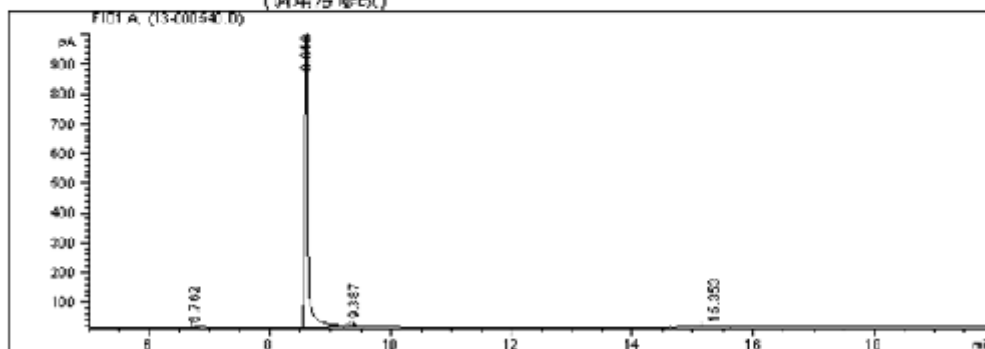
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*** 报告结束 ***

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 样品名称: 6#

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操作者      :
仪器        : 仪器 1                      位置 : 样品瓶 101
进样日期    : 12-Nov-13, 12:14:03        进样次数 : 1
                                           进样量 : 手动

采集方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-12 11:29:51
分析方法    : C:\CHEM32\1\METHODS\含溴化合物.M
最后修改    : 2013-11-13 14:09:05
              (调用后修改)
=====
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=====
 面积百分比报告
 =====

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排序      :      信号
乘积因子:      :      1.0000
稀释因子:      :      1.0000
内标使用乘积因子和稀释因子
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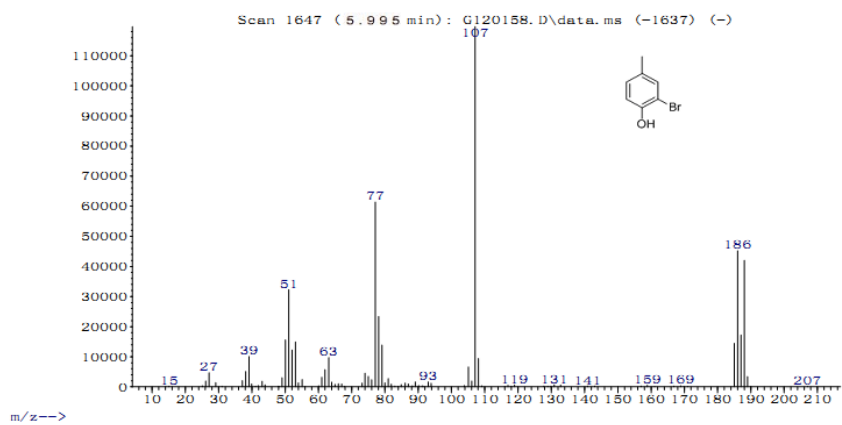
信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	6.762	BV	0.0607	19.70951	4.62469	0.41265
2	8.610	BV	0.0459	4702.00604	1411.73096	98.44574
3	9.387	VB	0.0390	34.87370	12.33783	0.73015
4	15.353	BB	0.1298	19.65178	1.88251	0.41145

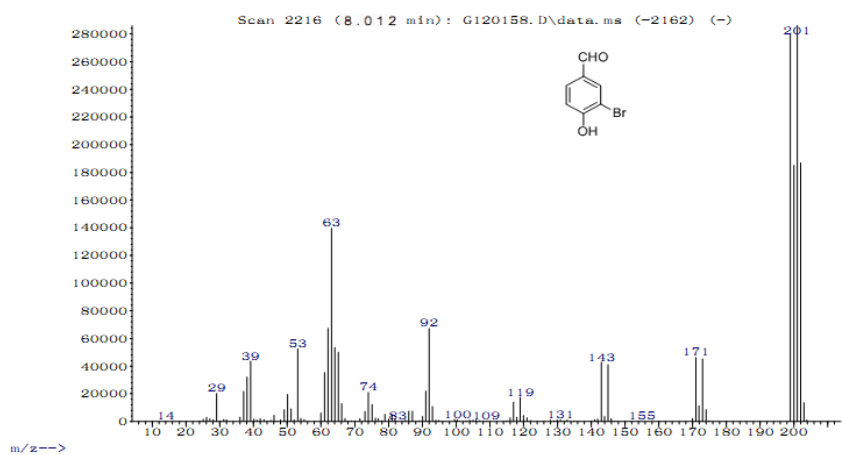
总量 : 4776.24183 1430.57898

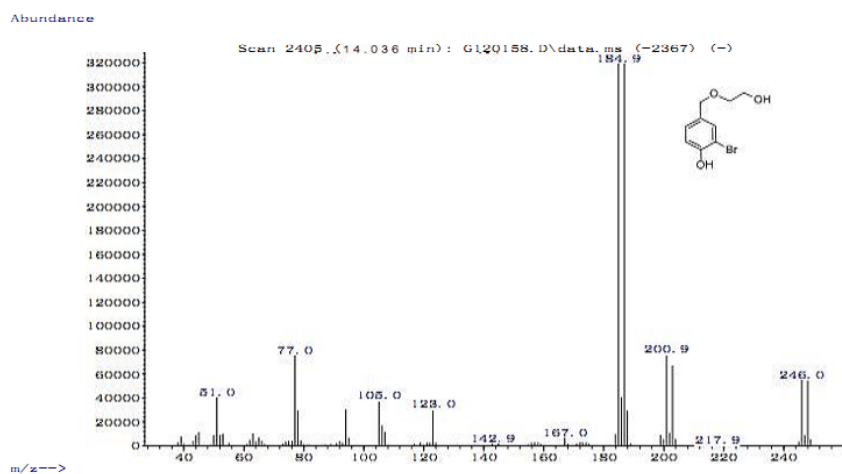
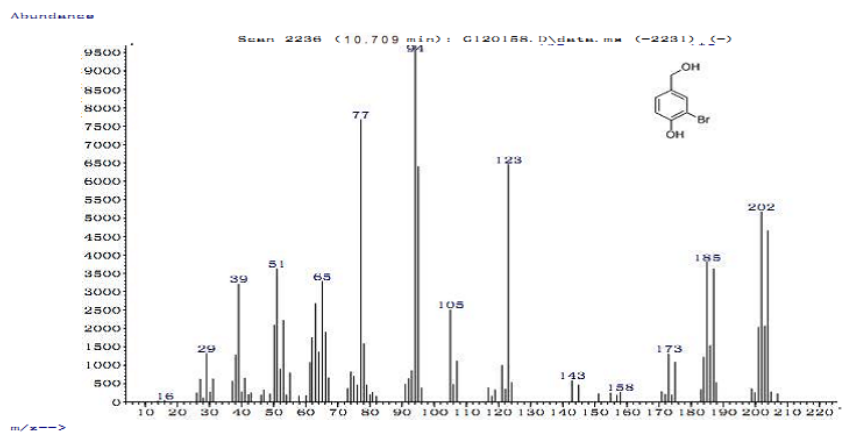
*** 报告结束 ***

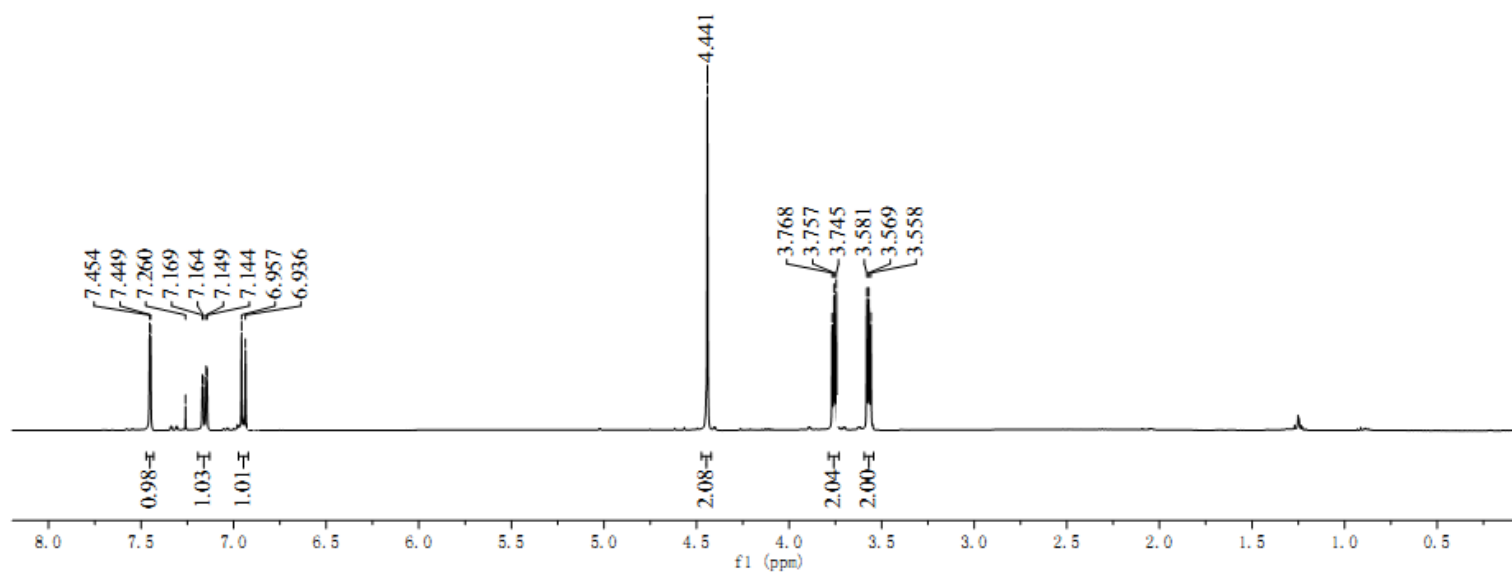
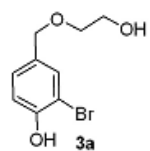
Abundance

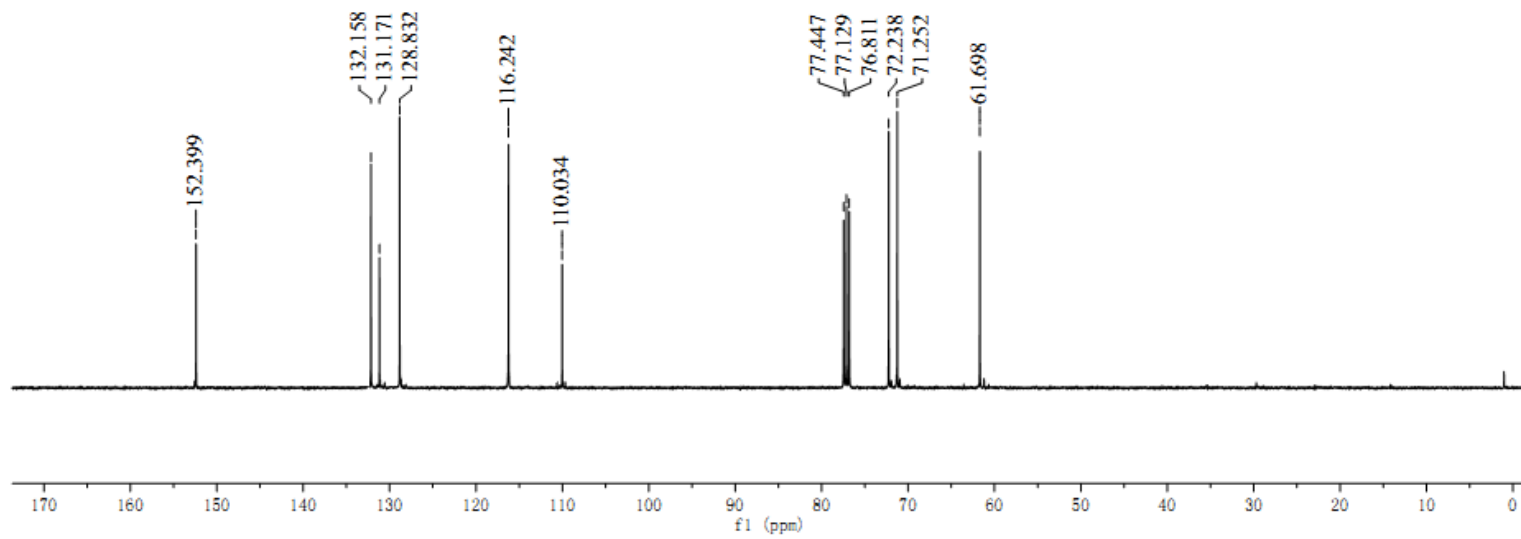
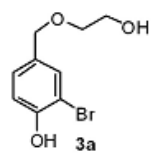


Abundance









Elemental Composition Report

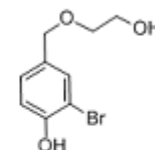
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

52 formula(e) evaluated with 11 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-35 H: 0-100 O: 0-10 Br: 0-1

YF-JI

ECUST institute of Fine Chem

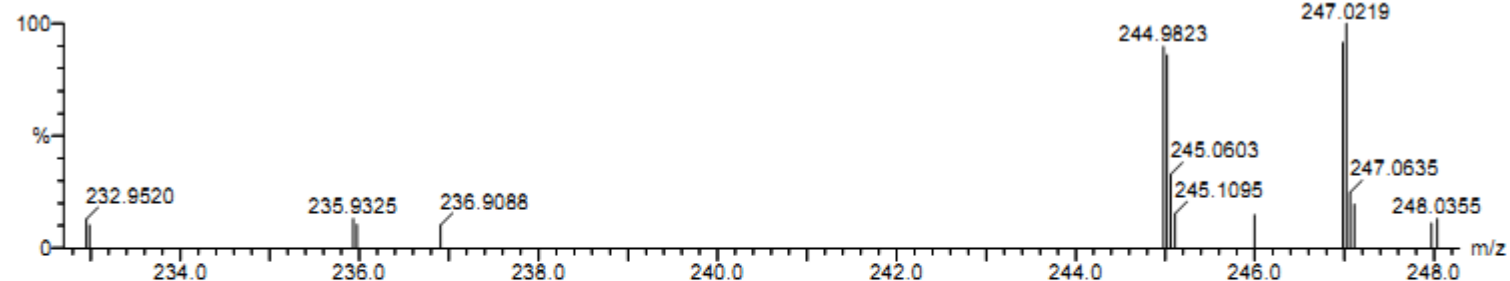
13-Nov-2013

08:21:36

1: TOF MS ES-

8.71e+002

JYF-DJL-1 169 (1.169) Cm (166:173)



Minimum:

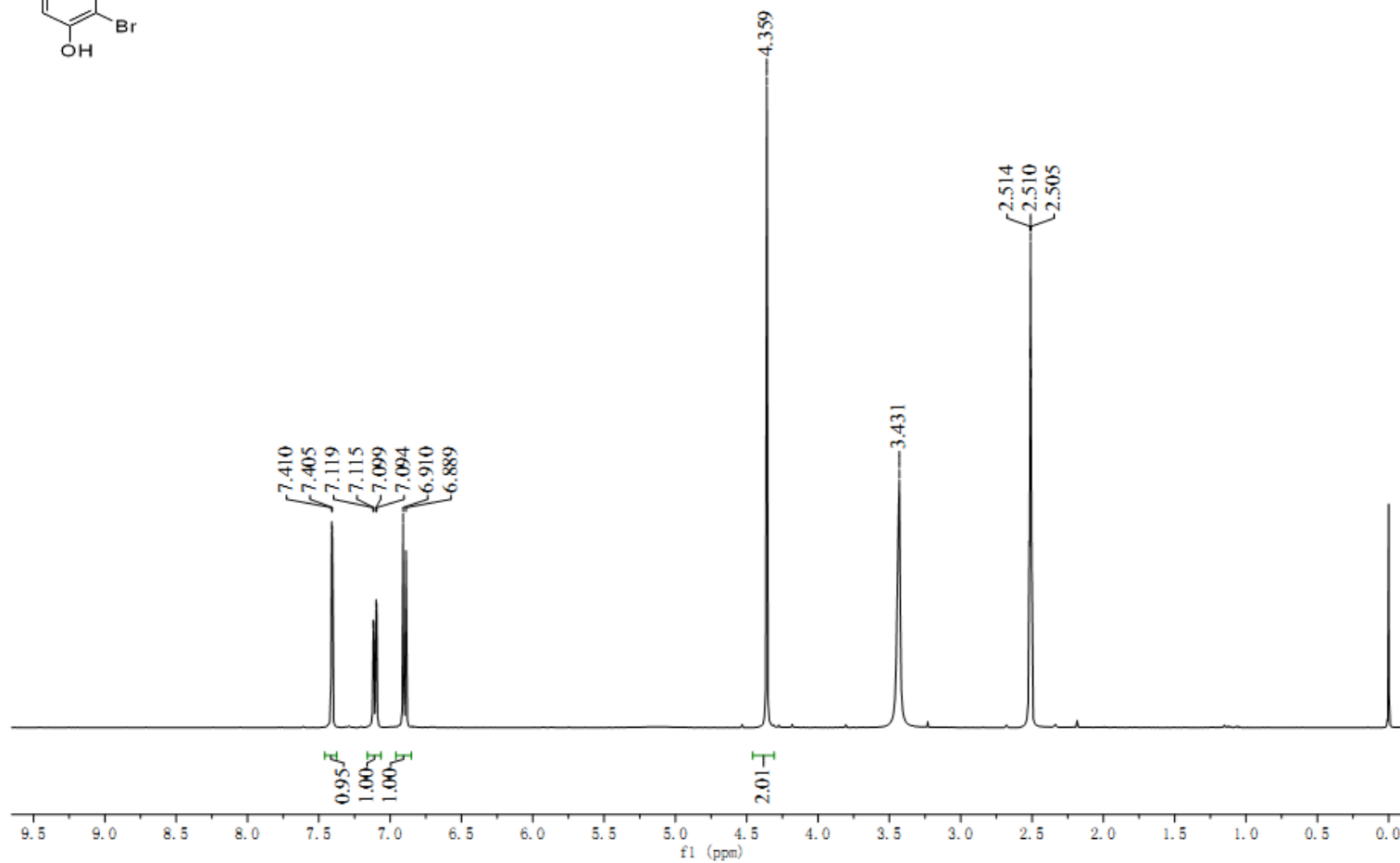
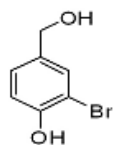
Maximum:

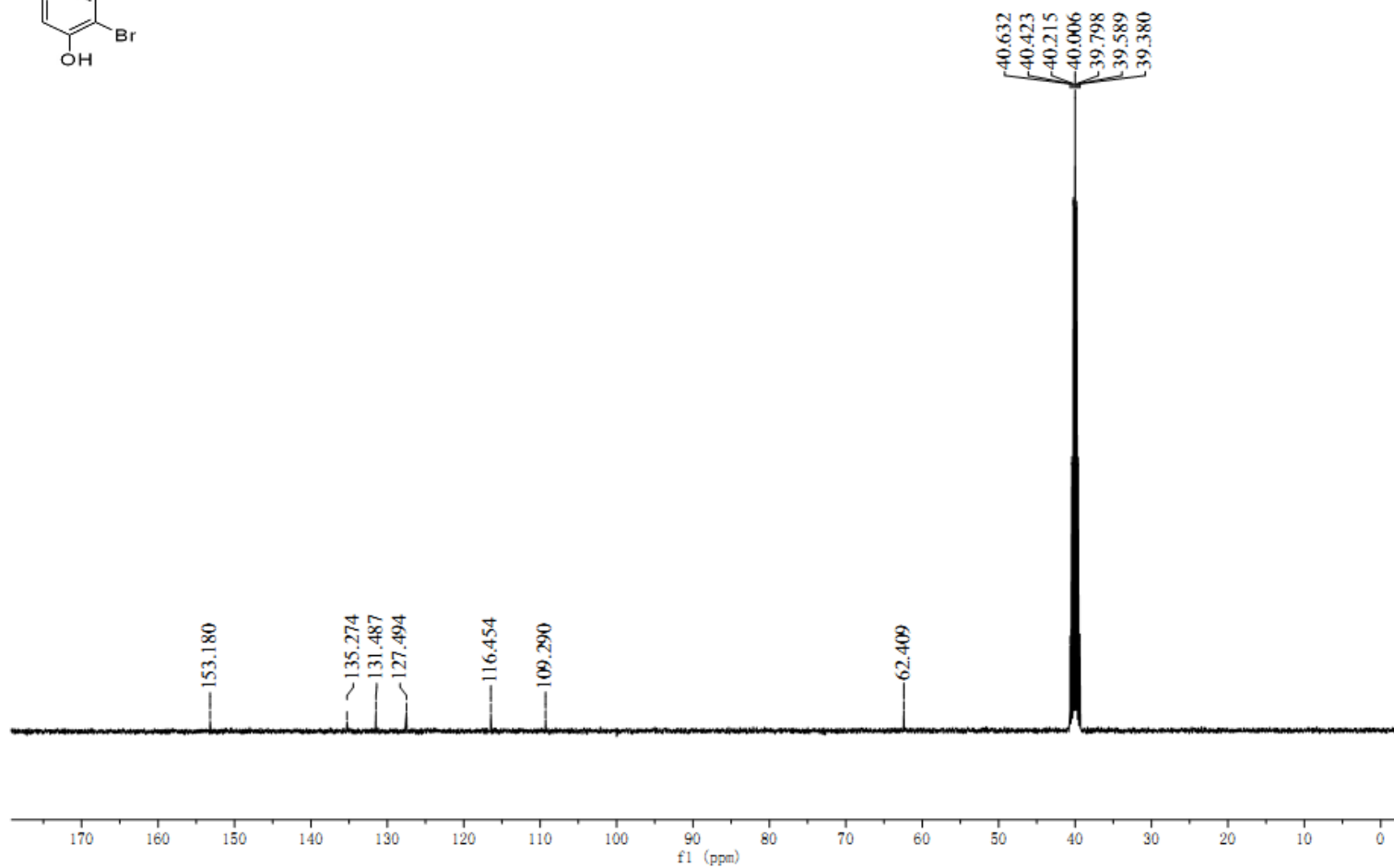
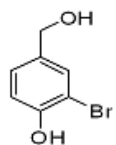
-1.5

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
------	------------	-----	-----	-----	-------	--------------	---------

244.9823	244.9813	1.0	4.1	4.5	49.6	0.0	C9 H10 O3 Br
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Elemental Composition Report

Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions

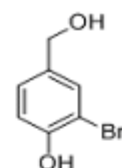
226 formula(e) evaluated with 35 results within limits (up to 50 closest results for each mass)

Elements Used:

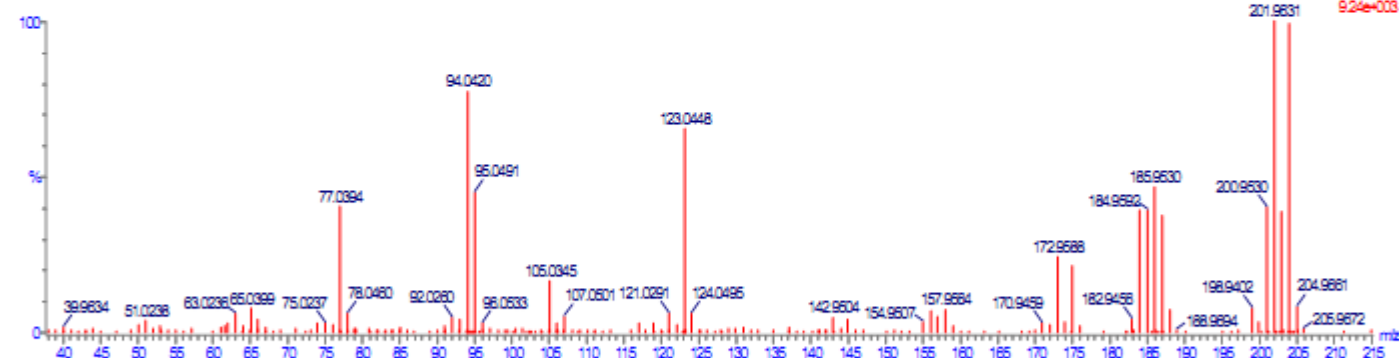
C: 0-7 H: 0-7 O: 0-2 79Br: 0-1 81Br: 0-1

jy4a5
20130711 119(1.983) Om(119(143+144))

Waters GCT Premier



03-JUL-2013 17:17:55
TOFMS.D+
9.24e+003

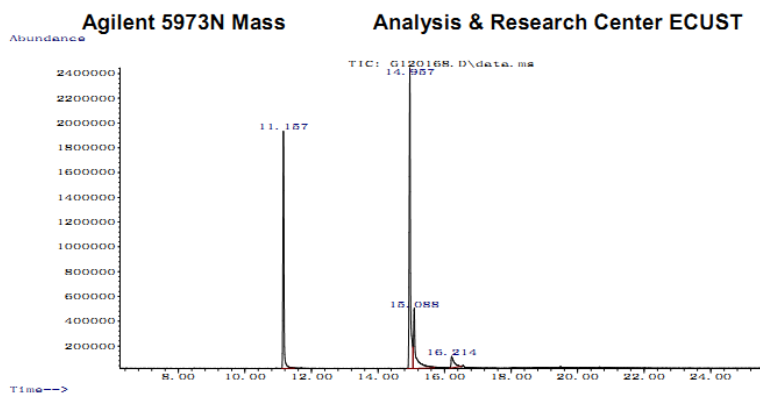


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
201.9631	100.00	201.9629	0.2	1.0	4.0	1133.7	C7 H7 O2 79Br

4.4 Copies of spectra for the oxidation process of 1a in methanol.



Data Path : E:\msdata\test\
 Data File : G120168.D
 Acq On : 23 Apr 2012 14:21
 Operator :
 Sample : 1#
 Misc :
 ALS Vial : 1 Sample Multiplier: 1

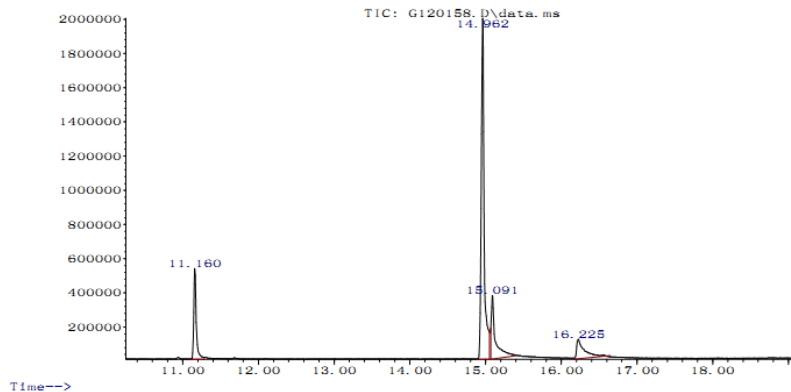
Integration Parameters : autointl.e
 Integrator : ChemStation

Method : C:\msdchem\1\METHODS\30 (3) - 8 - 280 . M
 Title :

Signal : TIC: G120168.D\data.ms

Peak #	R.T. min	first scan	max scan	last scan	PK TY	peak height	corr. area	corr. % max	% of total
1	11.154	1350	1358	1398	M	1976767	40192866	64.78%	31.054%
2	14.956	1916	1927	1940	M	2445817	62047946	100.00%	47.939%
3	15.090	1941	1947	2028	M3	492810	21289685	34.31%	16.449%
4	16.212	2106	2115	2161	M6	92910	5900144	9.51%	4.559%

Agilent 5973N Mass Analysis & Research Center ECUST



Area Percent Report

Data Path : E:\msdata\test\
 Data File : G120158.D
 Acq On : 19 Apr 2012 17:14
 Operator :
 Sample : 1#
 Misc :
 ALS Vial : 1 Sample Multiplier: 1

Integration Parameters : autointl.e
 Integrator : ChemStation

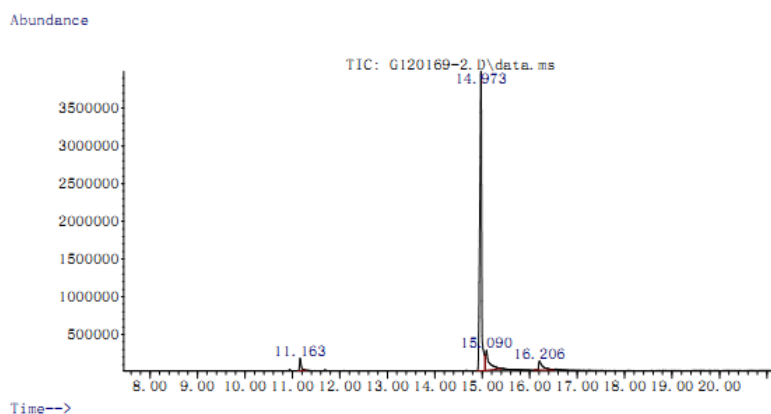
Method : C:\msdchem\1\METHODS\30 (3) - 8 - 280.M
 Title :

Signal : TIC: G120158.D\data.ms

Peak #	R.T. min	first scan	max scan	last scan	PK TY	peak height	corr. area	corr. % max	% of total
1	11.158	1637	1647	1666	M	533930	12111190	22.67%	13.808%
2	14.961	2204	2216	2230	M	2025628	53433483	100.00%	60.921%
3	15.087	2232	2235	2283	M2	378561	14038890	26.27%	16.006%
4	16.223	2395	2405	2467	M4	114203	8126268	15.21%	9.265%

Agilent 5973N Mass

Analysis & Research Center ECUST



Area Percent Report

Data Path : E:\msdata\test\
 Data File : G120169.D
 Acq On : 23 Apr 2012 16:16
 Operator :
 Sample : 2#
 Misc :
 ALS Vial : 1 Sample Multiplier: 1

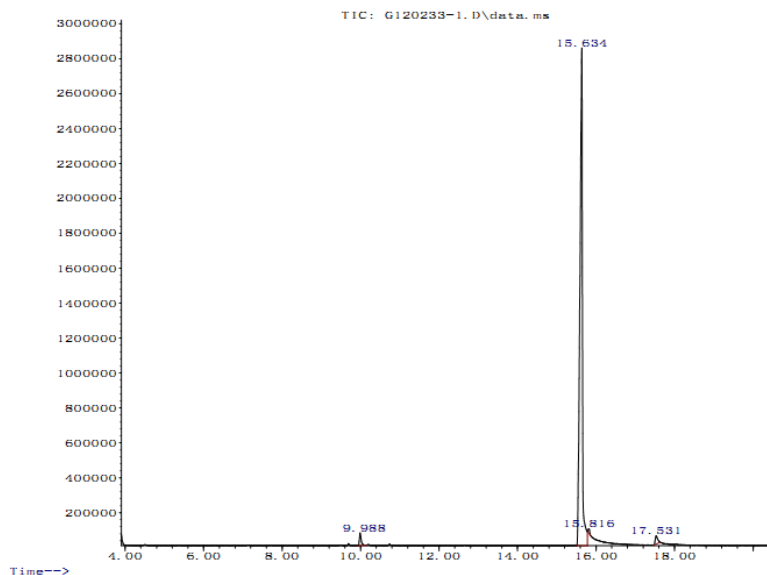
Integration Parameters : autoint1.e
 Integrator : ChemStation

Method : C:\msdchem\1\METHODS\30 (3) - 8 - 280.M
 Title :

Signal : TIC: G120169.D\data.ms

Peak #	R.T. min	first scan	max scan	last scan	PK TY	peak height	corr. area	corr. % max	% of total
1	11.161	1352	13589	1377	M	166598	4051834	3.76%	2.937%
2	14.976	1917	1930	1943	M	4165701	107672228	100.00%	78.039%
3	15.090	1943	1947	2029	M5	273280	19040969	17.68%	13.801%
4	16.206	2104	2114	2158	M4	124498	7207699	6.69%	5.224%

Agilent 5973N Mass Analysis & Research Center ECUST



Area Percent Report

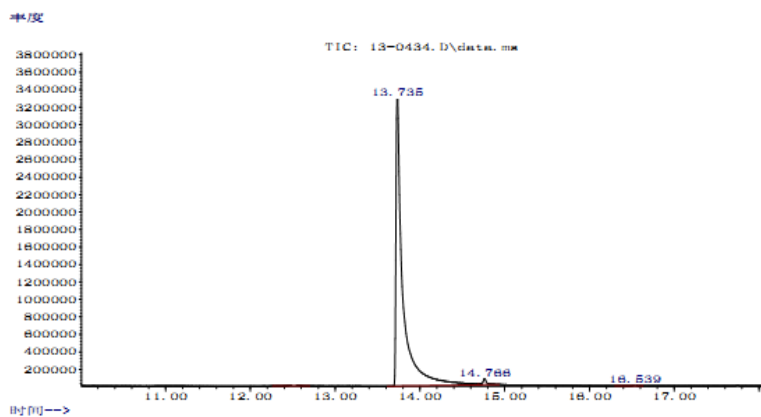
Data Path : E:\msdata\test\
 Data File : G120233-1.D
 Acq On : 11 May 2012 14:35
 Operator :
 Sample : 1#
 Misc :
 ALS Vial : 1 Sample Multiplier: 1

Integration Parameters : autoint1.e
 Integrator : ChemStation

Method : C:\msdchem\1\METHODS\30 (3) - 8 - 280 . M
 Title :

Signal : TIC: G120168 . D\data.ms

Peak #	R.T. min	first scan	max scan	last scan	PK TY	peak height	corr. area	corr. % max	% of total
1	9.989	1461	1470	1398	M	73048	2302379	1.85%	1.782%
2	15.636	2295	2315	2336	M	2866806	124190818	100.00%	96.118%
3	15.816	2590	2599	2612	M5	48006	1941923	1.56%	1.503%
4	17.533	2337	2342	2353	M8	27555	771401	0.62%	0.597%



Area Percent Report

Data Path : E:\msdata\test\
 Data File : 12-0445.D
 Acq On : 12 May 2012 14:35
 Operator :
 Sample : 1#
 Misc :
 ALS Vial : 1 Sample Multiplier: 1

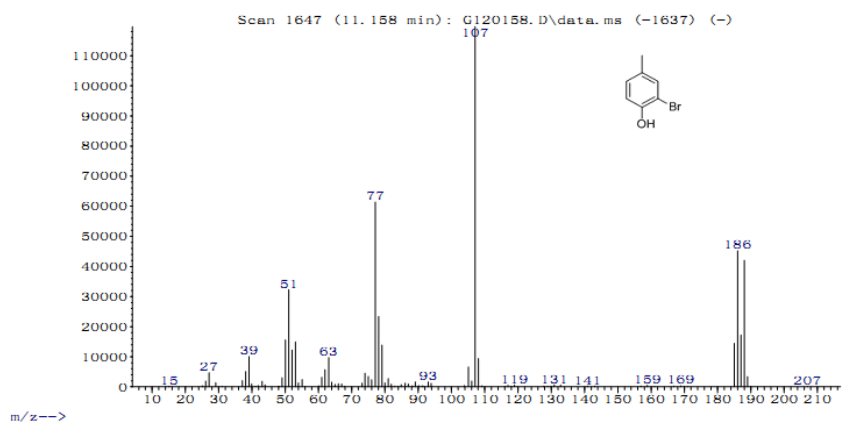
Integration Parameters : autoint1.e
 Integrator : ChemStation

Method : C:\msdchem\1\METHODS\30 (3) - 8 - 280.M
 Title :

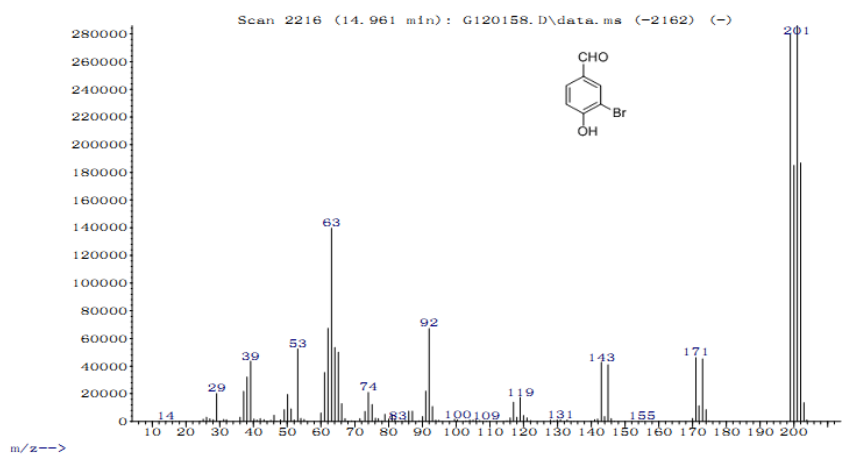
Signal : TIC: G120168.D\data.ms

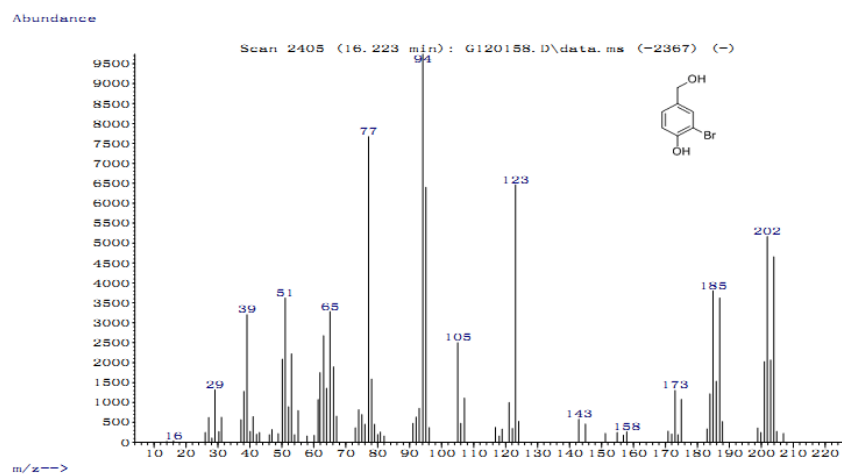
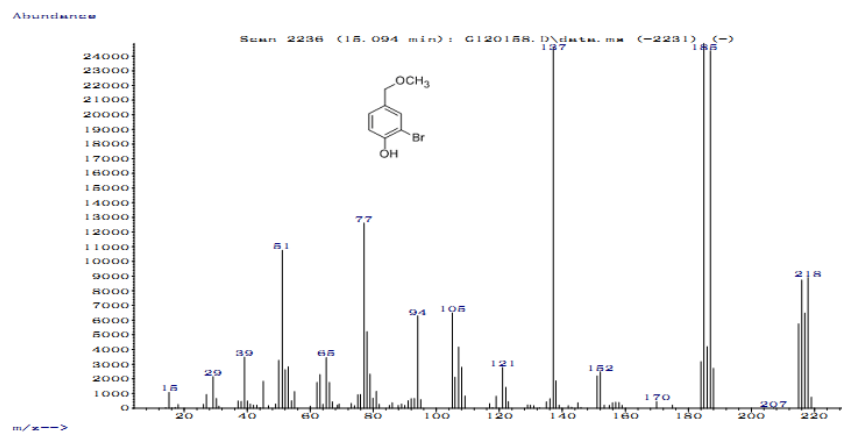
Peak #	R.T. min	first scan	max scan	last scan	PK TY	peak height	corr. area	corr. % max	% of total
1	13.735	1964	1983	2154	BB	3292515	181419490	100.00%	98.963%
2	16.539	2154	2175	2189	BB 2	65489	1525770	0.85%	0.832%
3	14.766	2485	2505	2514	BV	25086	375587	0.21%	0.205%

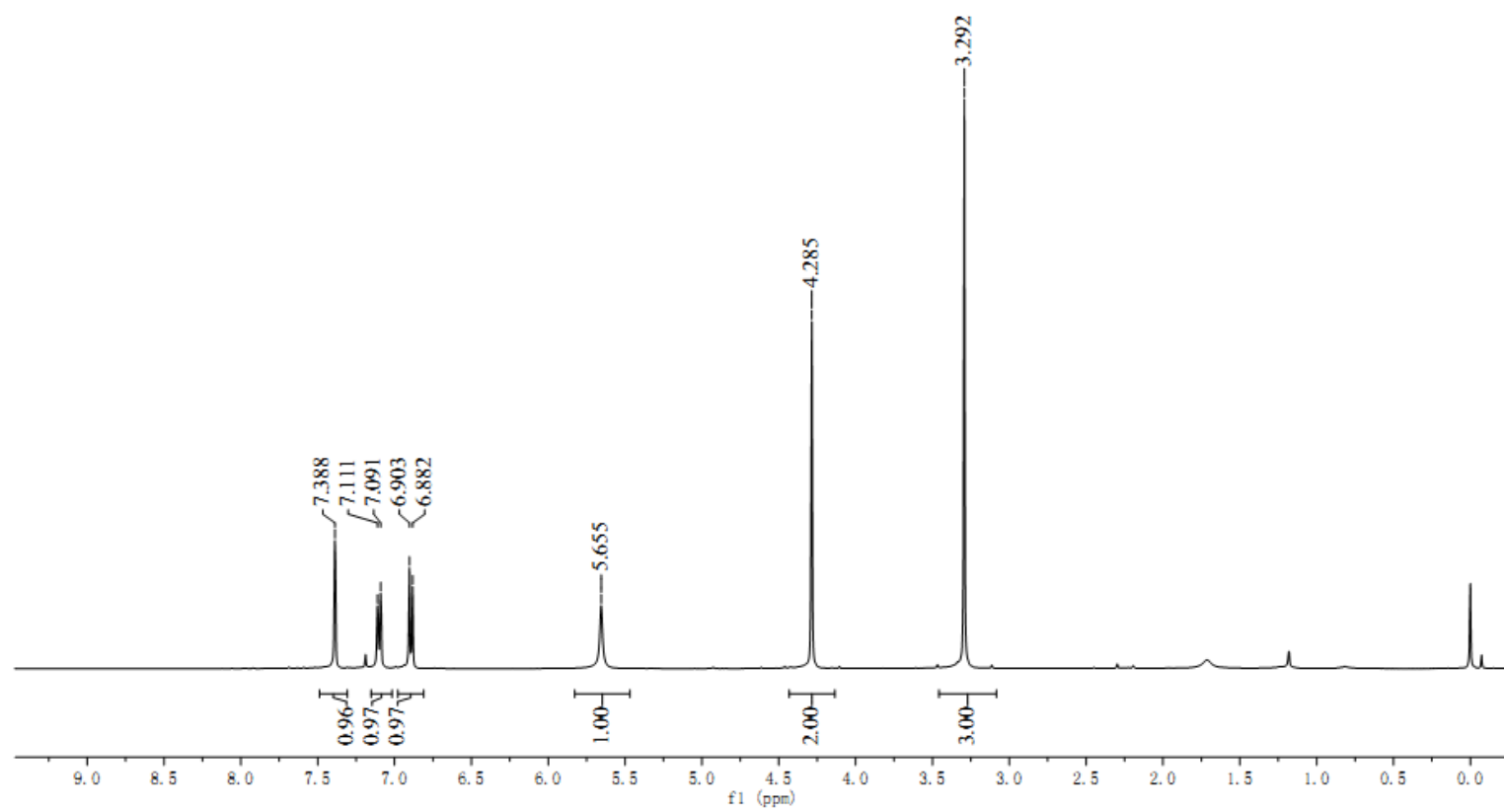
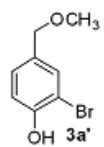
Abundance

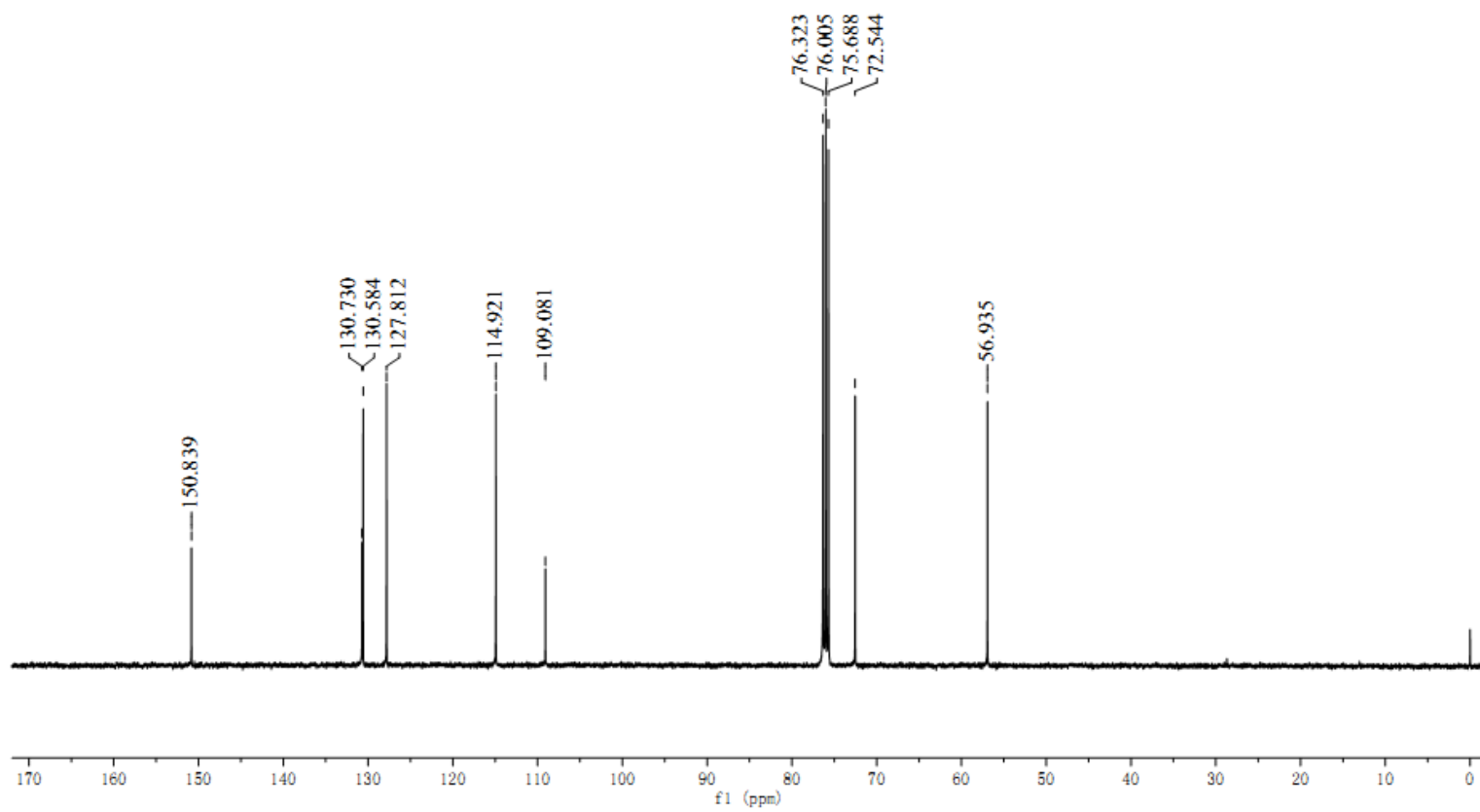
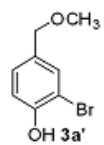


Abundance









Elemental Composition Report

Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Monoisotopic Mass, Odd and Even Electron Ions

382 formula(e) evaluated with 25 results within limits (up to 50 closest results for each mass)

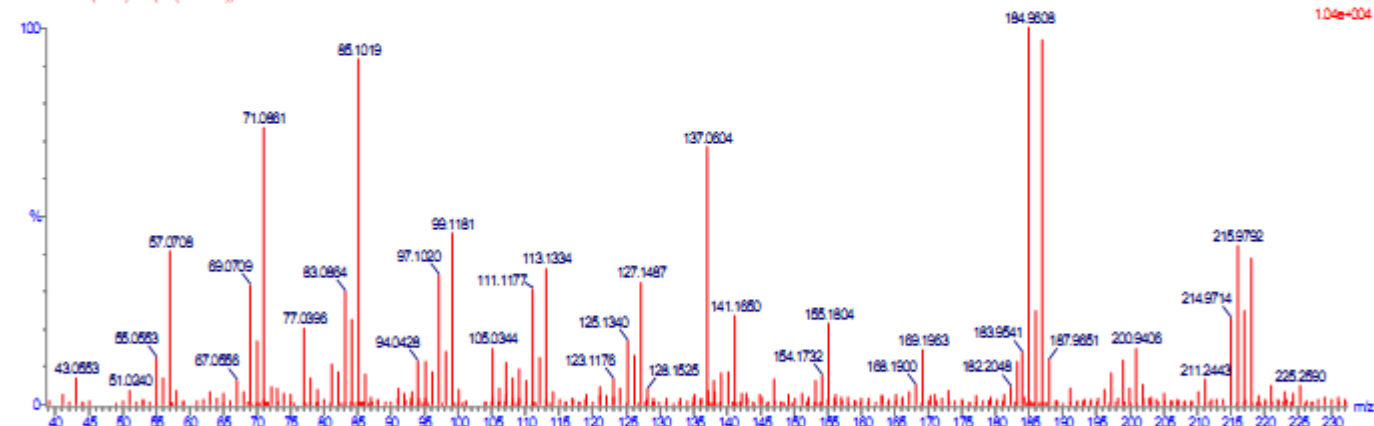
Elements Used:

C: 0-8 H: 0-9 O: 0-2 79Br: 0-1 81Br: 0-1

jjfj23
20130709 21 (0.350) Qm(214(122+123))

Vibrans GCT Premier

03-JUL-2013 18:51:19
TOF/MS E+
1.04e-004



Minimum:

-1.5

Maximum:

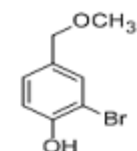
100.00

5.0

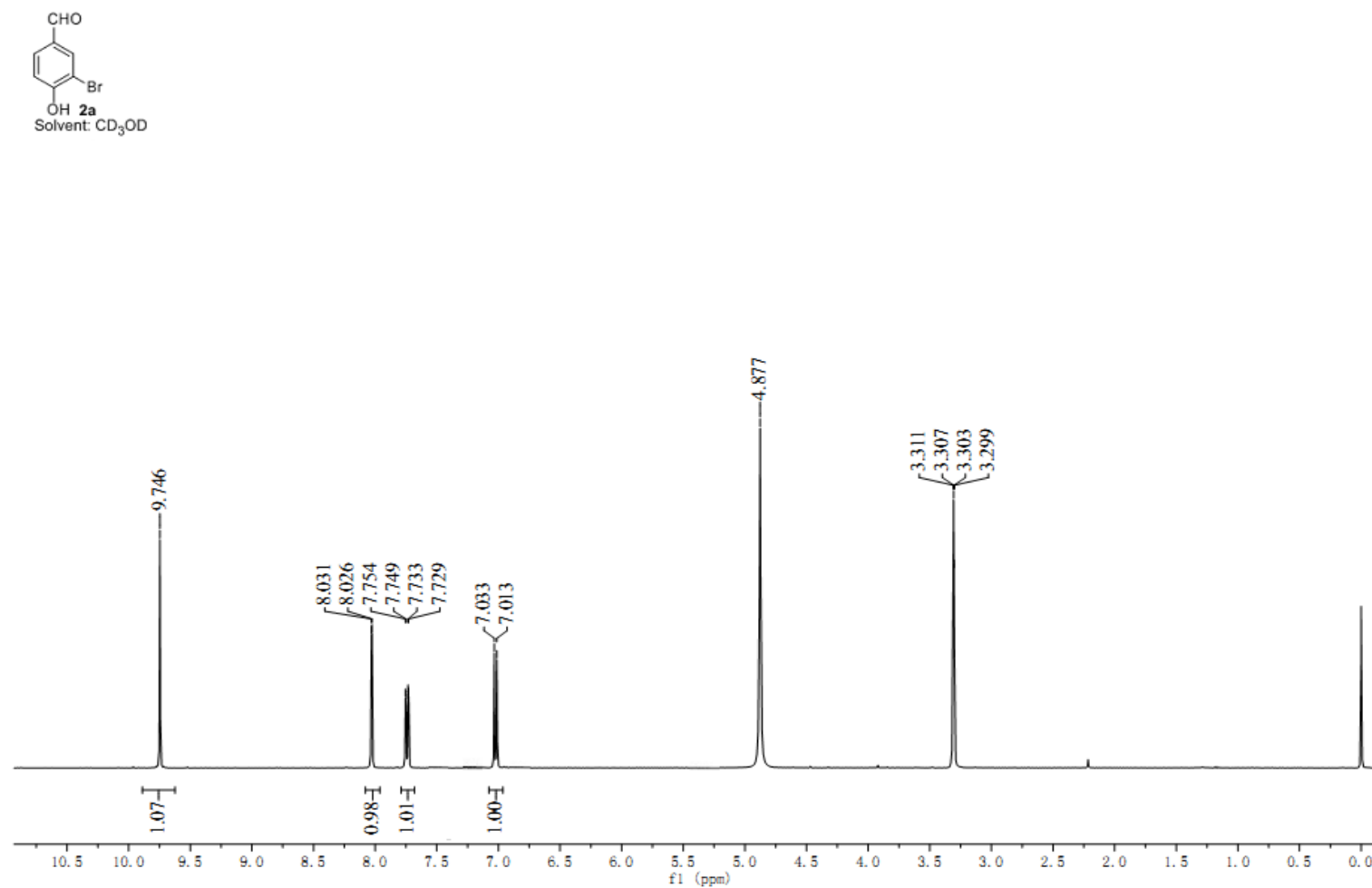
10.0

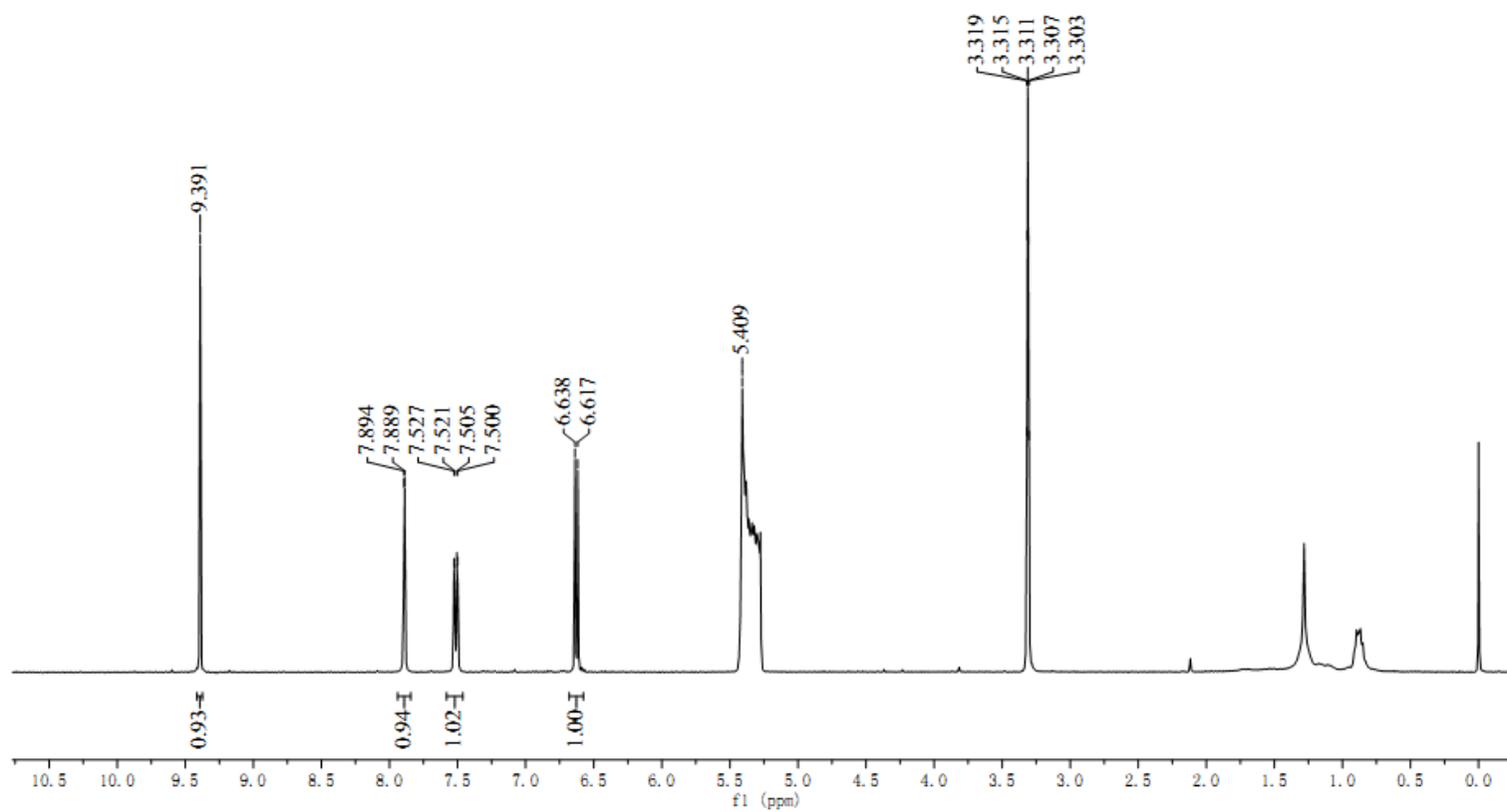
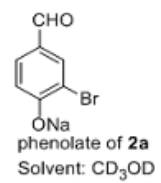
50.0

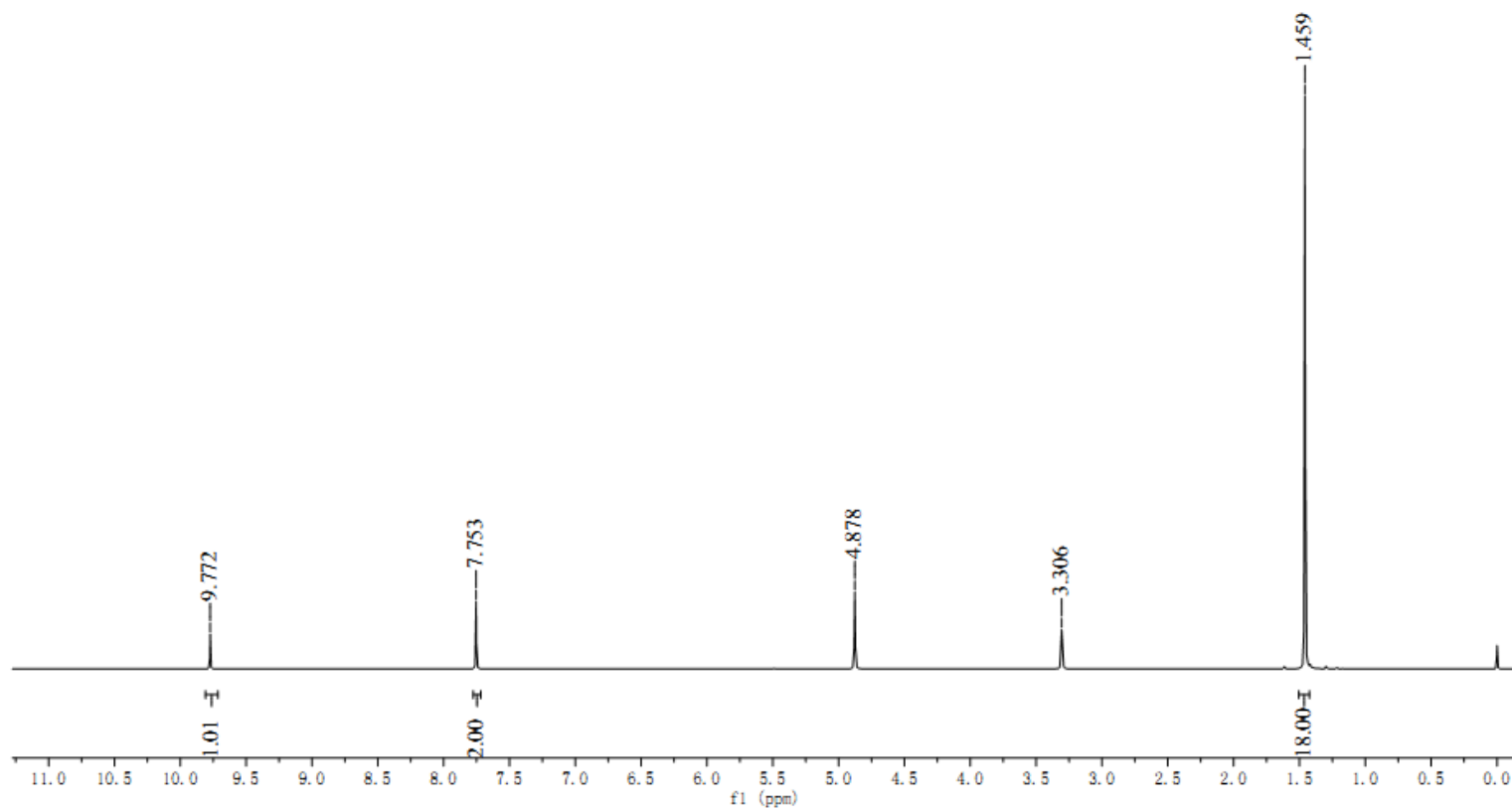
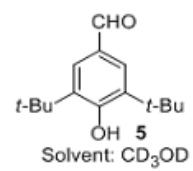
Mass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
215.9792	41.98	215.9786	0.6	2.8	4.0	936.5	C8 H9 O2 79Br

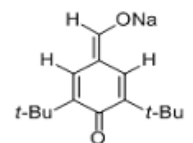


4.5 Copies of ^1H NMR spectra for compounds 2a and 5, in $\text{CD}_3\text{ONa}/\text{CD}_3\text{OD}$ and CD_3OD , respectively.

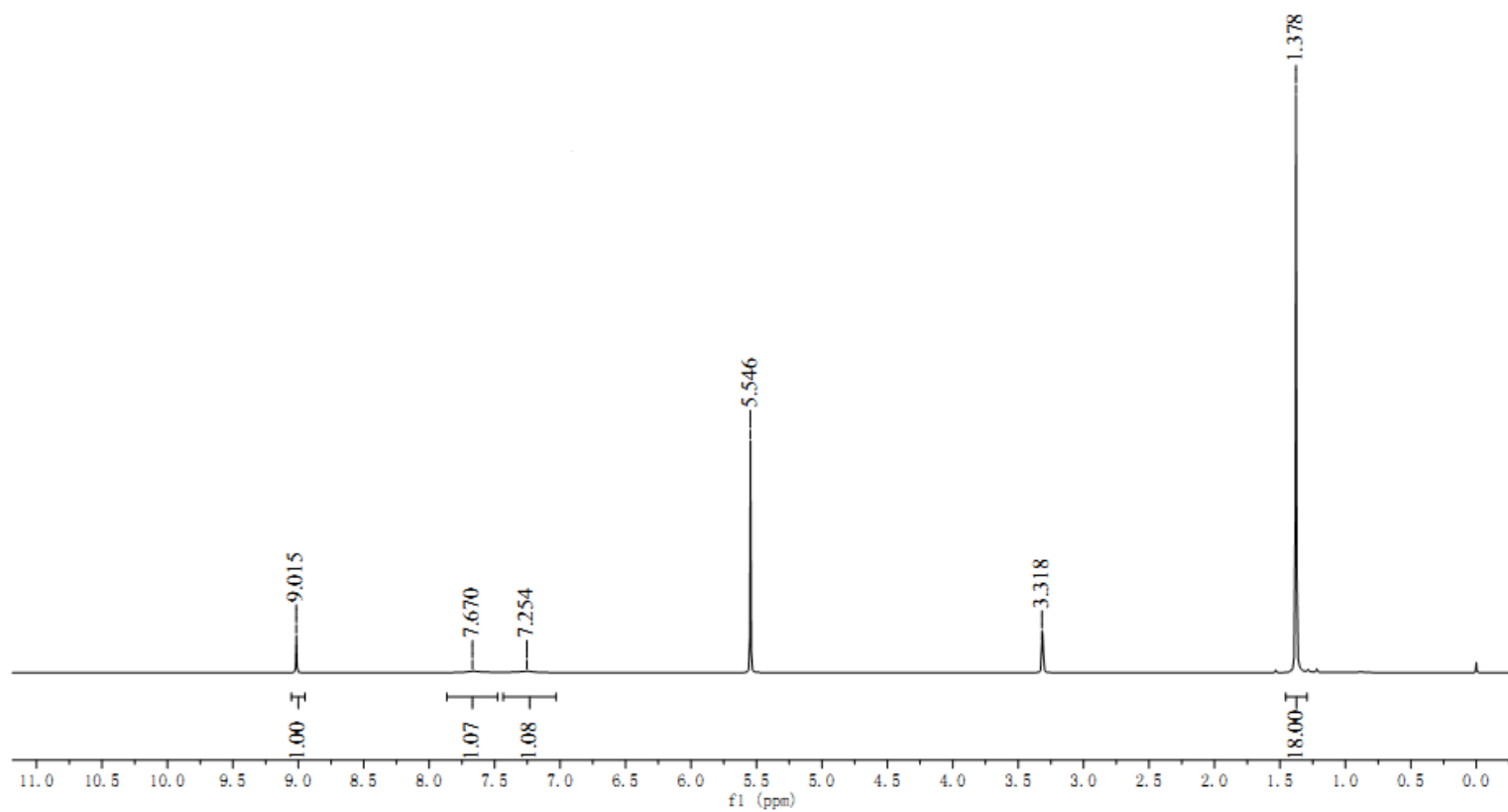




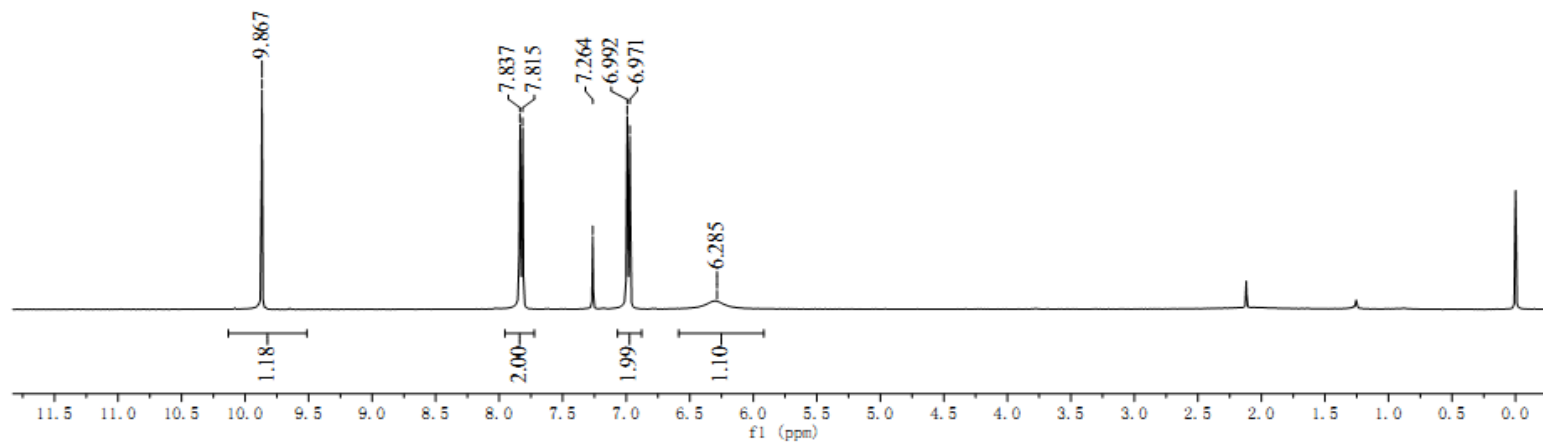
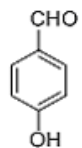


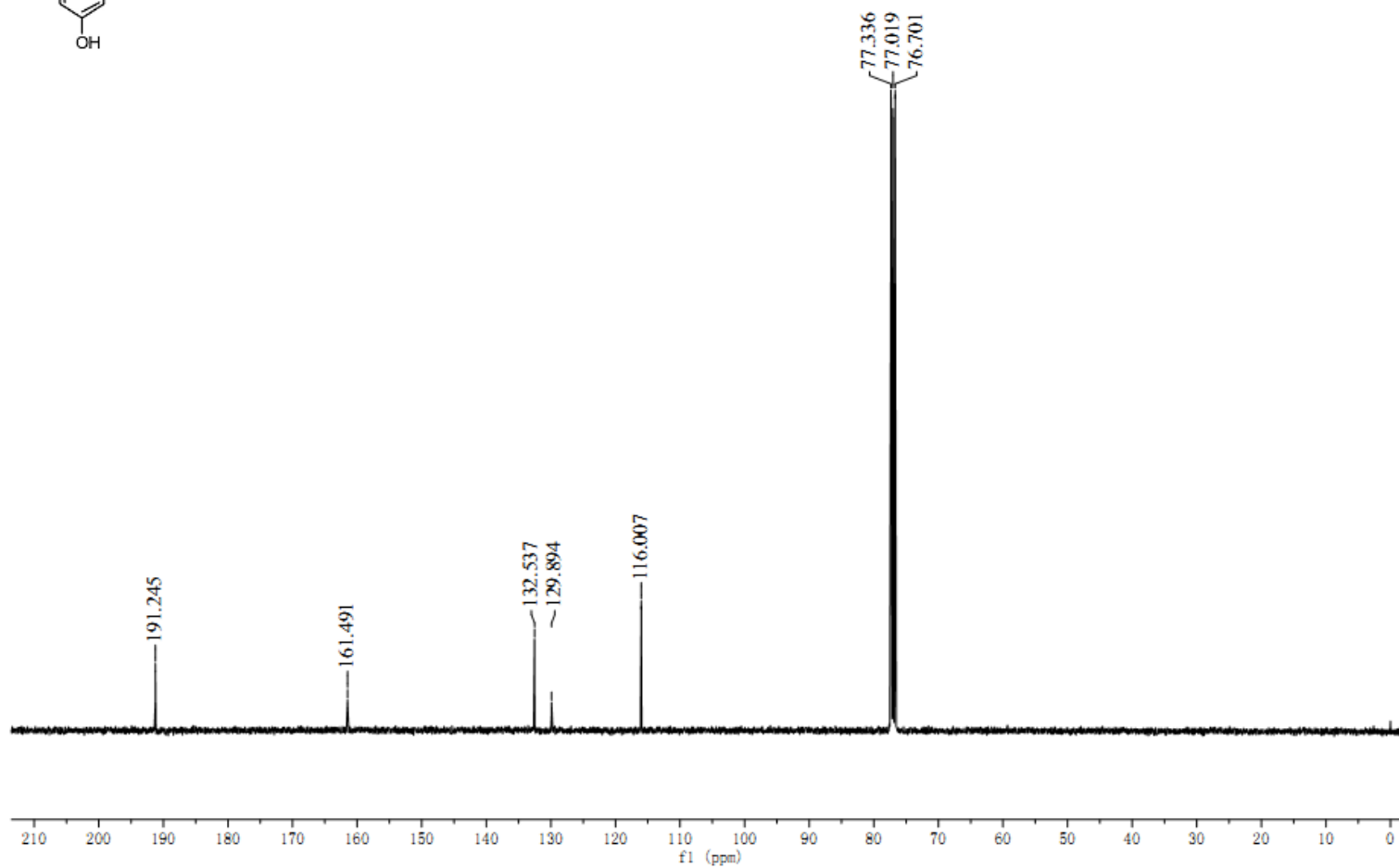
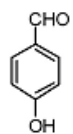


enolate of **5**
Solvent: CD₃OD



4.6 Copies of spectra for $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ -catalyzed oxidation of 1a, 1b and 1c.





Elemental Composition Report

Single Mass Analysis

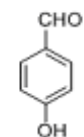
Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Monoisotopic Mass, Odd and Even Electron Ions

19 formula(e) evaluated with 9 results within limits (up to 50 closest results for each mass)

Elements Used:

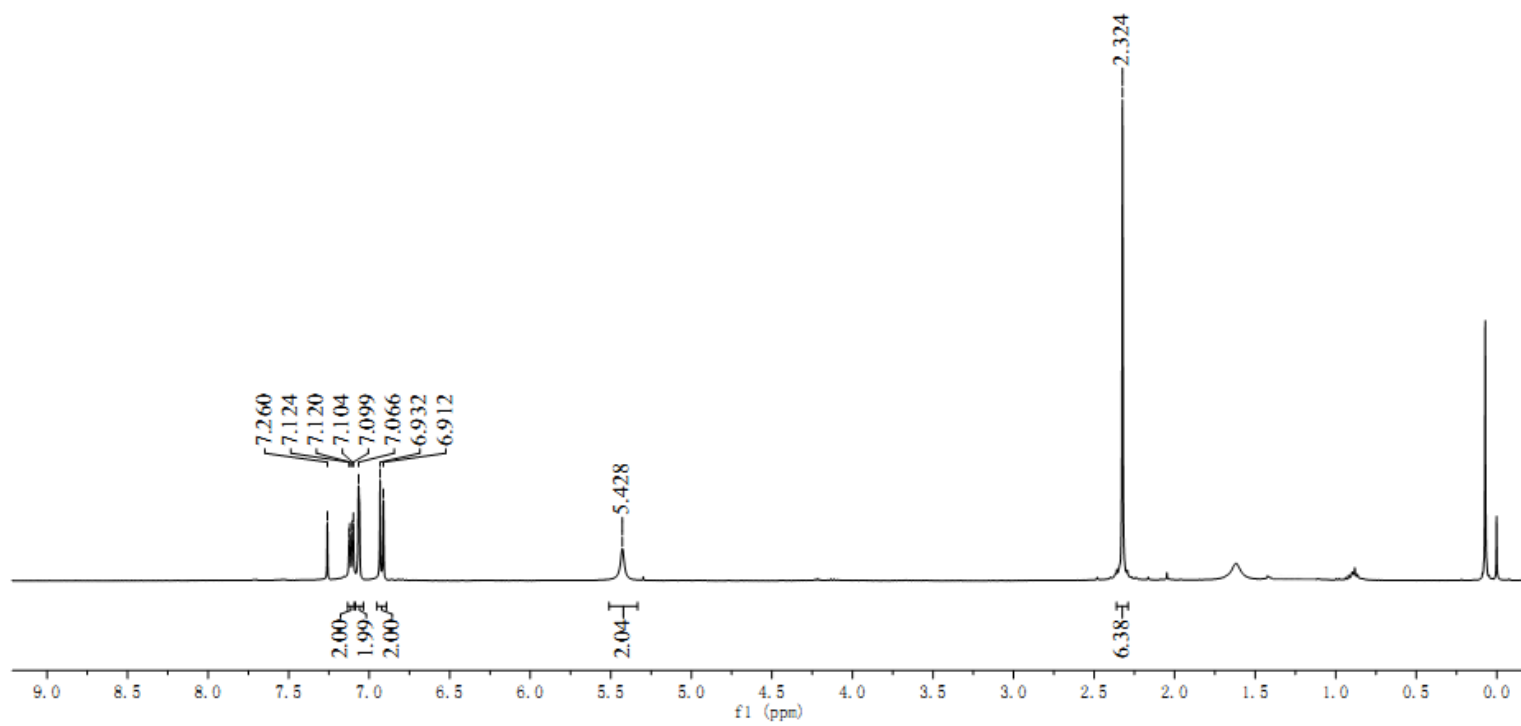
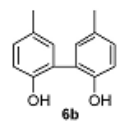
C: 0-7 H: 0-6 O: 0-2

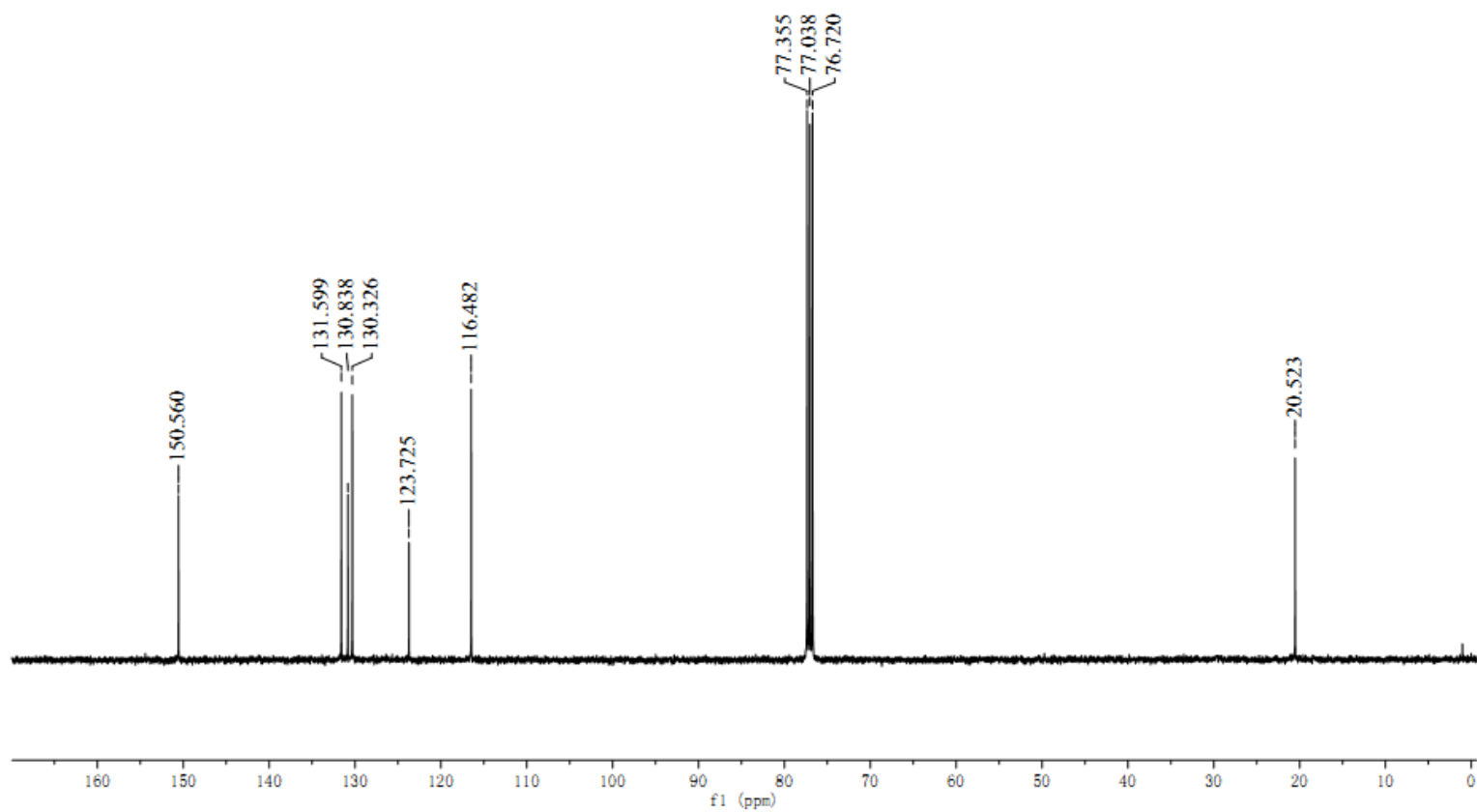
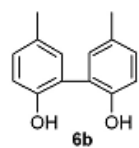


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	I-FIT	Formula
122.0367	84.79	122.0368	-0.1	-0.8	5.0	3.5	C7 H6 O2





Elemental Composition Report

Single Mass Analysis

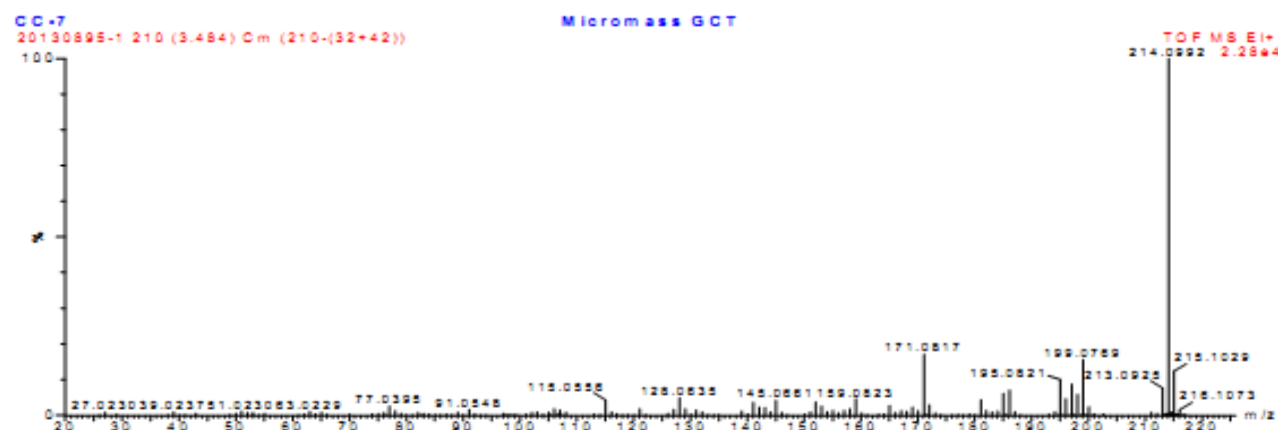
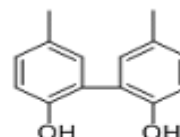
Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Monoisotopic Mass, Odd and Even Electron Ions

68 formula(e) evaluated with 20 results within limits (up to 50 closest results for each mass)

Elements Used:

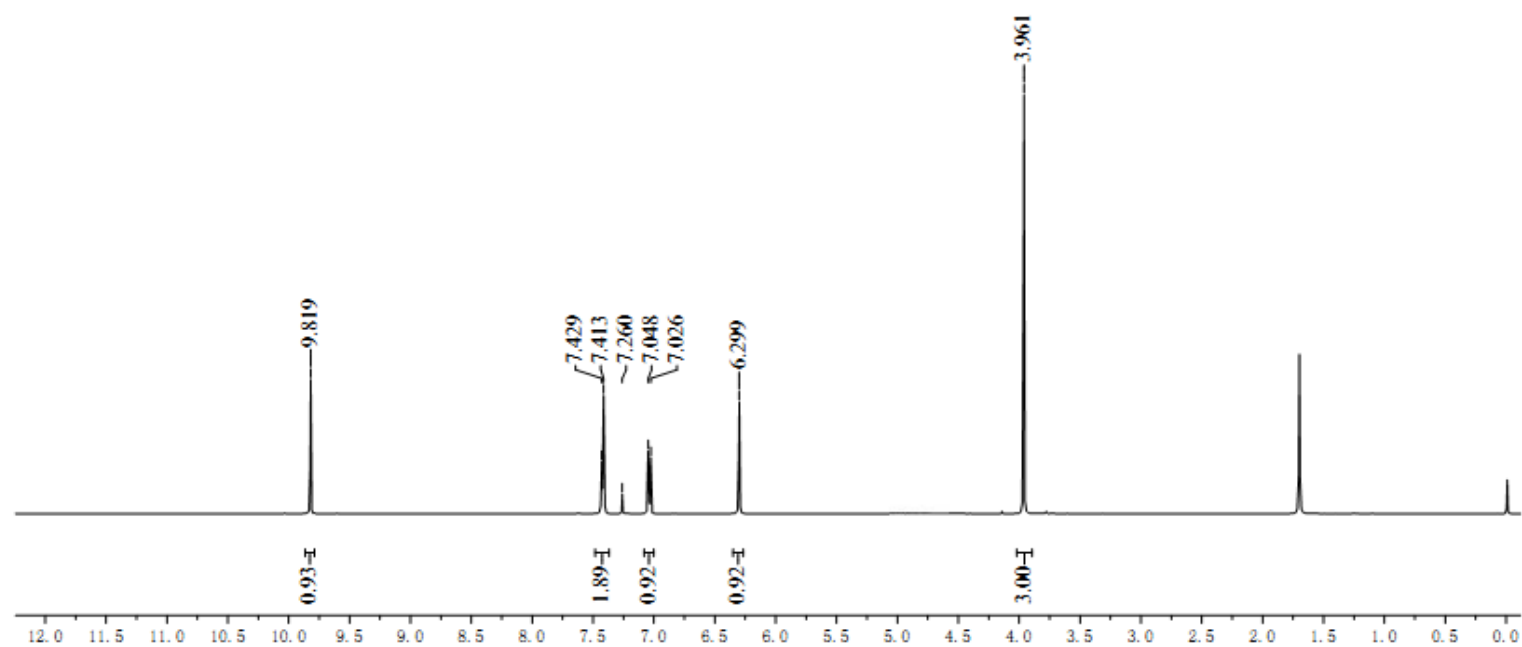
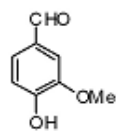
C: 0-14 H: 0-14 O: 0-2

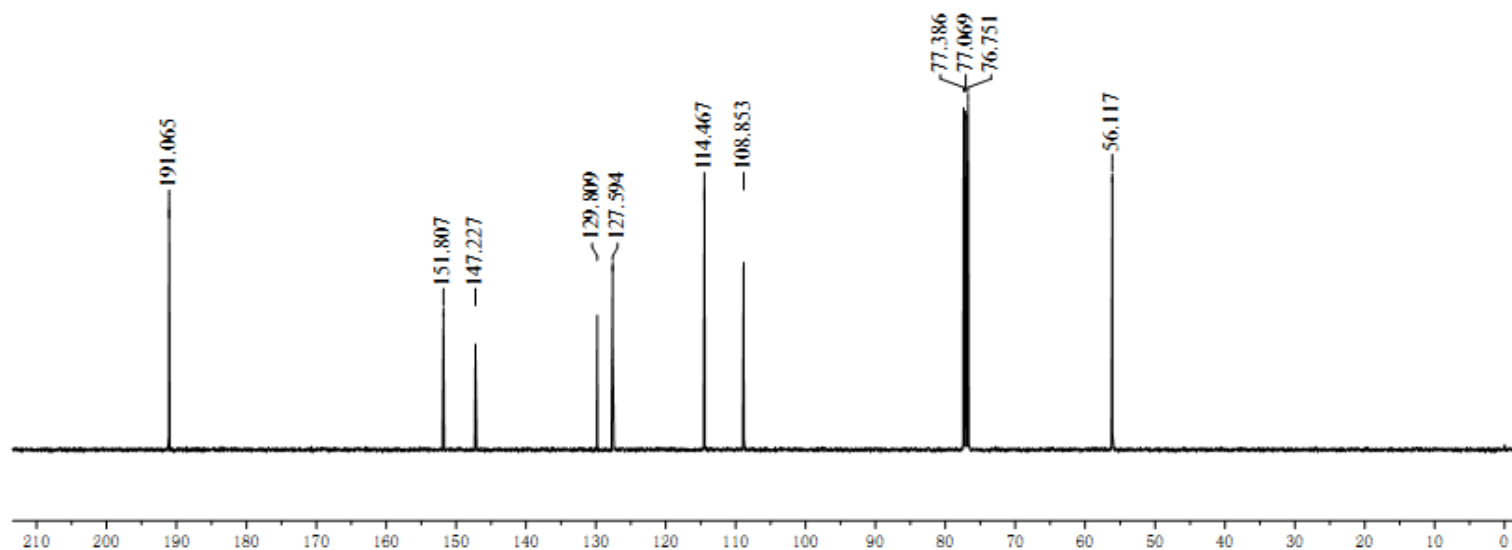
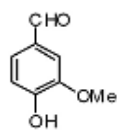


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 5.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	Score	Formula
214.0992	100.00	214.0994	-0.2	-0.8	8.0	1	C14 H14 O2





Elemental Composition Report

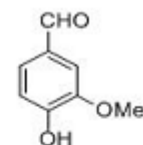
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

5 formula(e) evaluated with 1 results within limits (up to 1 best isotopic matches for each mass)

Elements Used:

C: 0-8 H: 0-30 O: 0-3

YF-JI

ECUST institute of Fine Chem

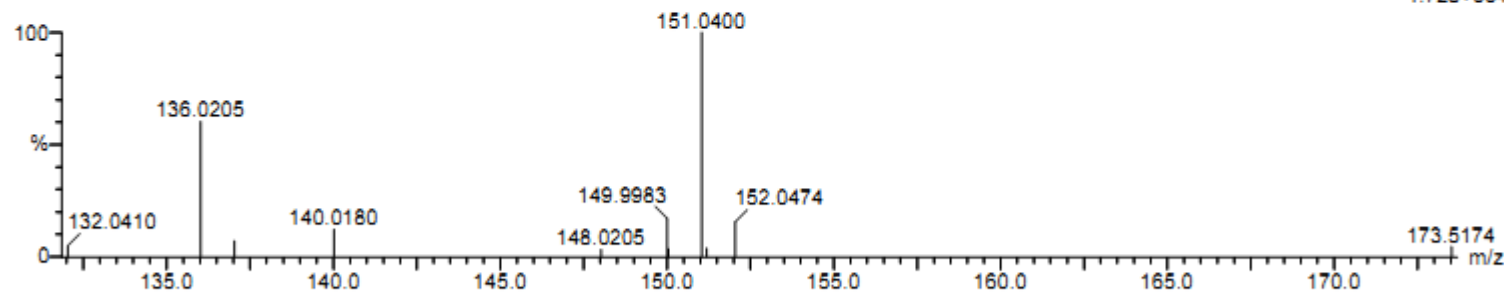
16-May-2013

08:23:46

1: TOF MS ES-

1.72e+004

JYF-JA-6 29 (0.292) Cm (6:33)



Minimum:

Maximum:

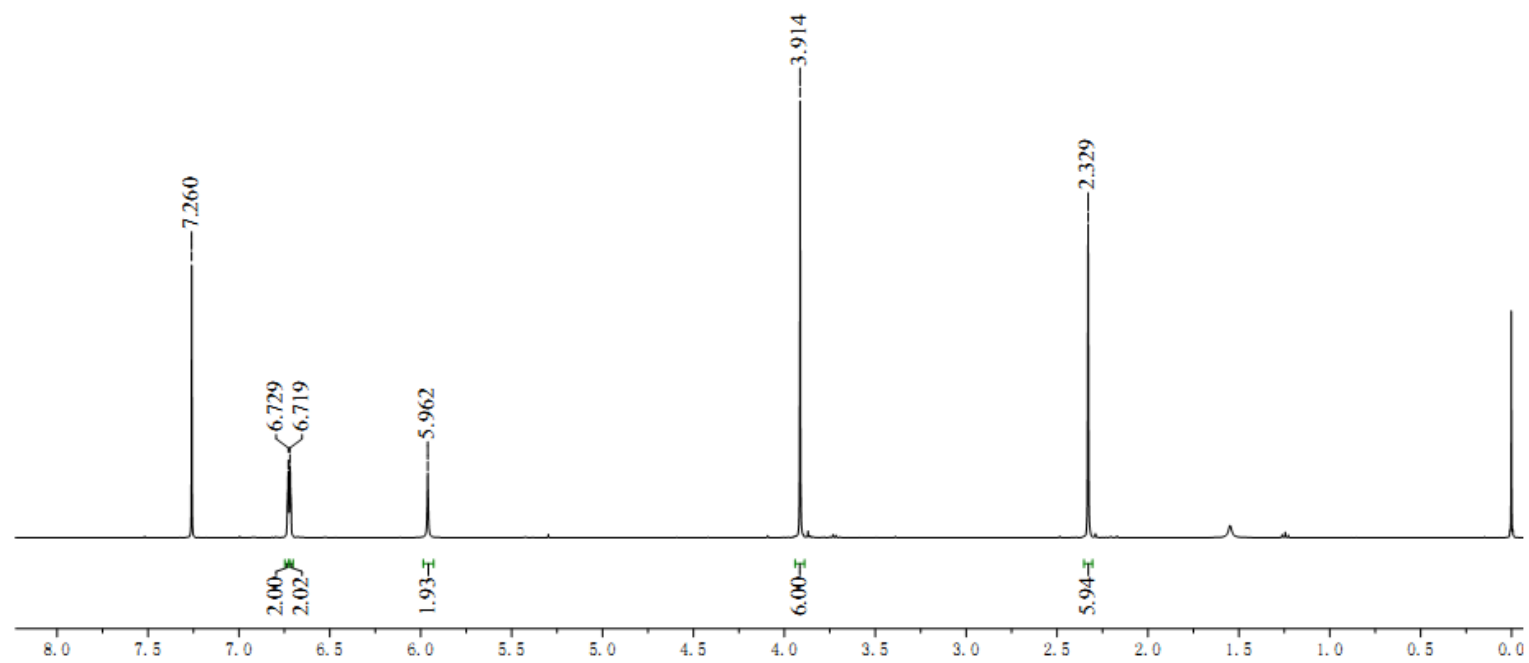
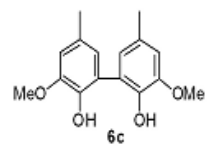
30.0

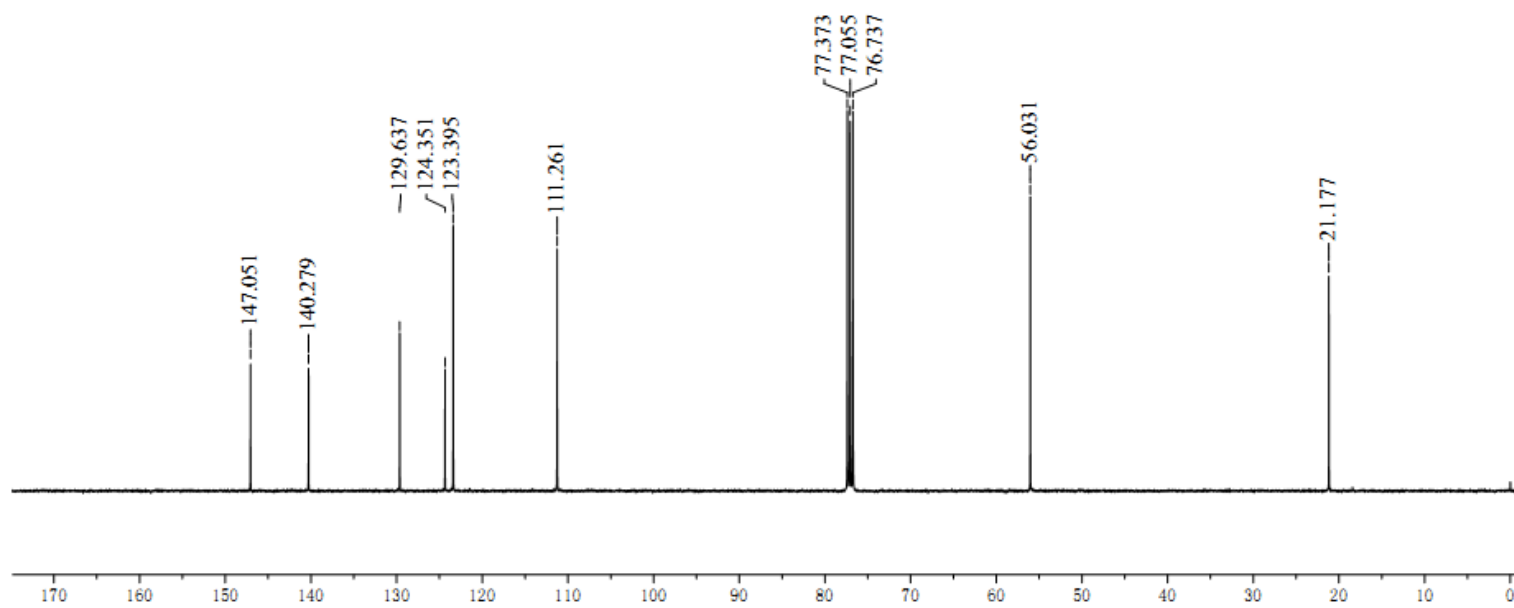
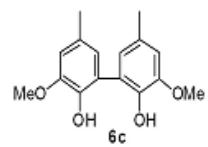
50.0

-1.5

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
151.0400	151.0395	0.5	3.3	5.5	33.3	0.0	C8 H7 O3





Elemental Composition Report

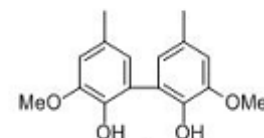
Page 1

Single Mass Analysis

Tolerance = 100.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

33 formula(e) evaluated with 17 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-39 H: 0-60 O: 0-8

YF-JI

ECUST institute of Fine Chem

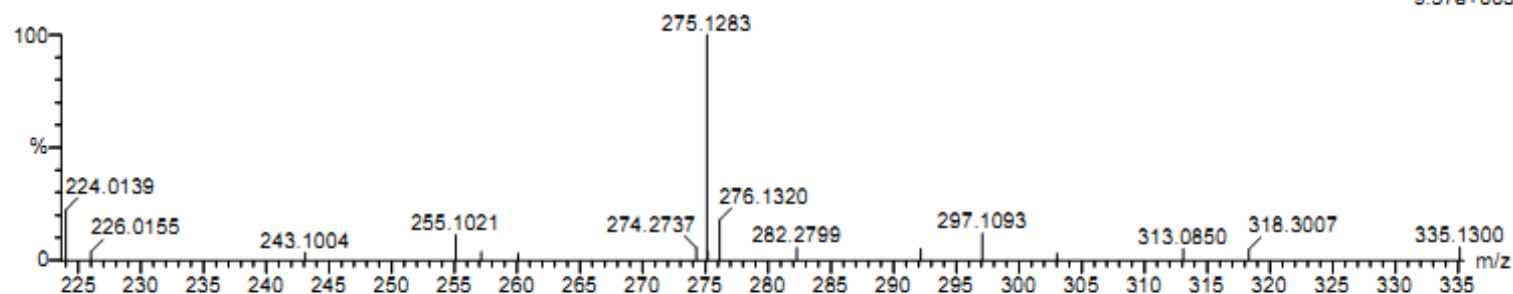
02-Jan-2013

21:59:41

1: TOF MS ES+

9.57e+003

JYF-JA-105 16 (0.569) Cm (1:16)

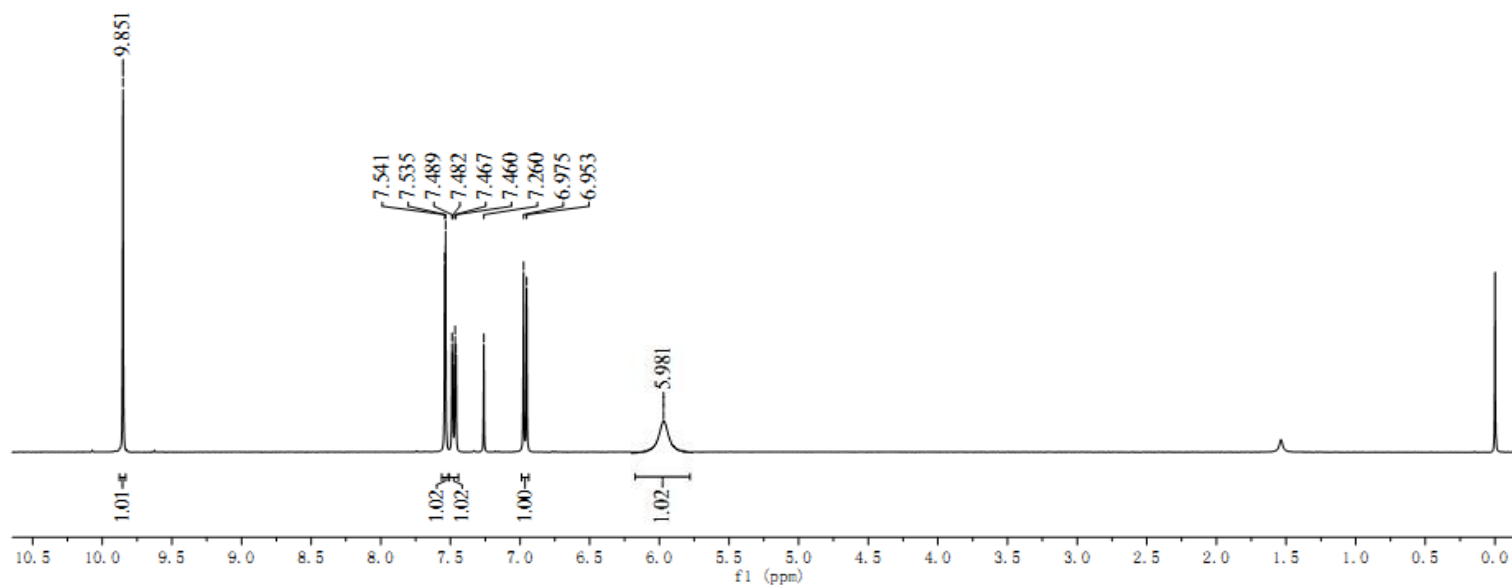
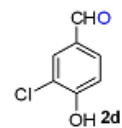


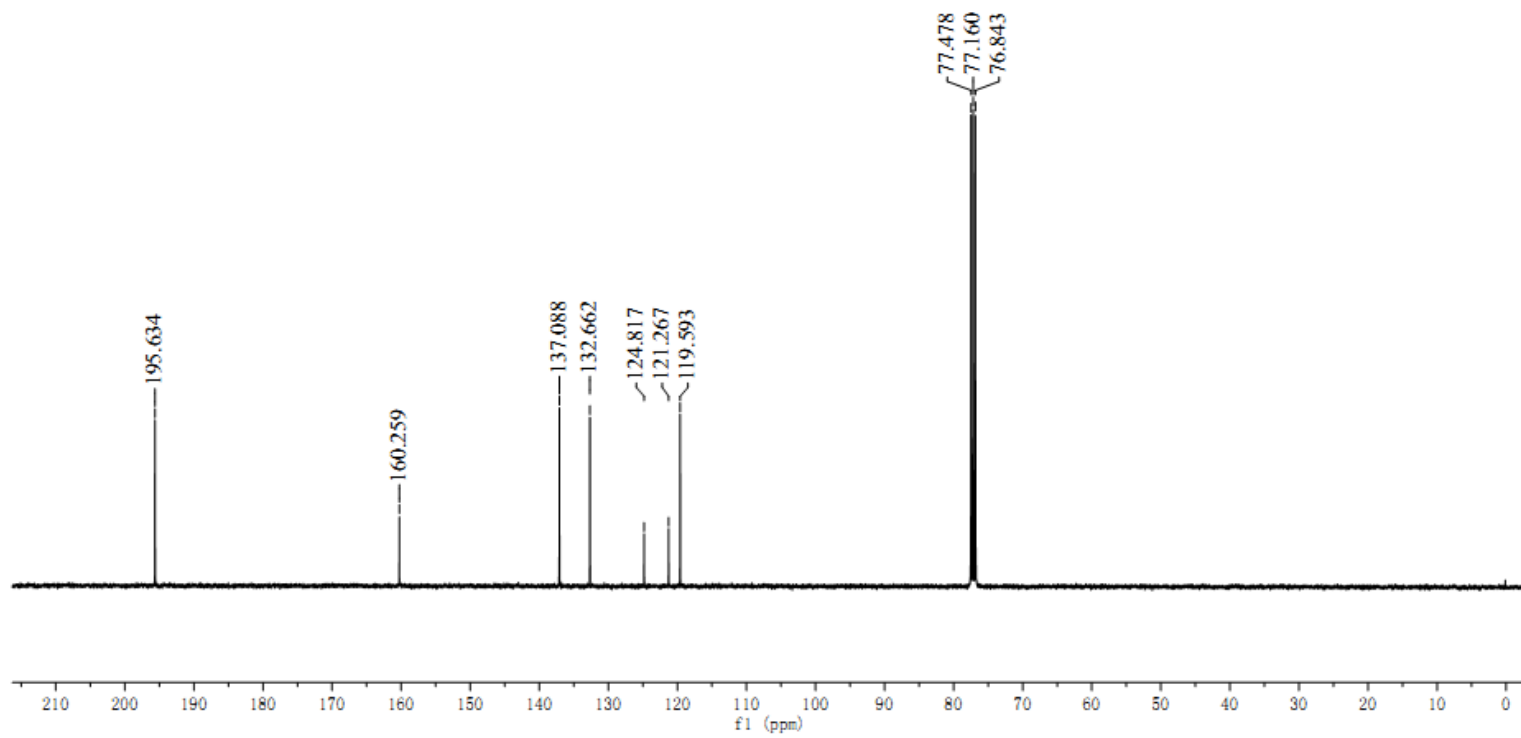
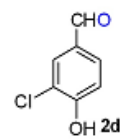
Minimum:

Maximum: 100.0 50.0 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
275.1283	275.1283	0.0	0.0	7.5	18.4	0.0	C16 H19 O4

4.7 Copies of spectra for $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ -catalyzed oxidation of 1.





Elemental Composition Report

Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50

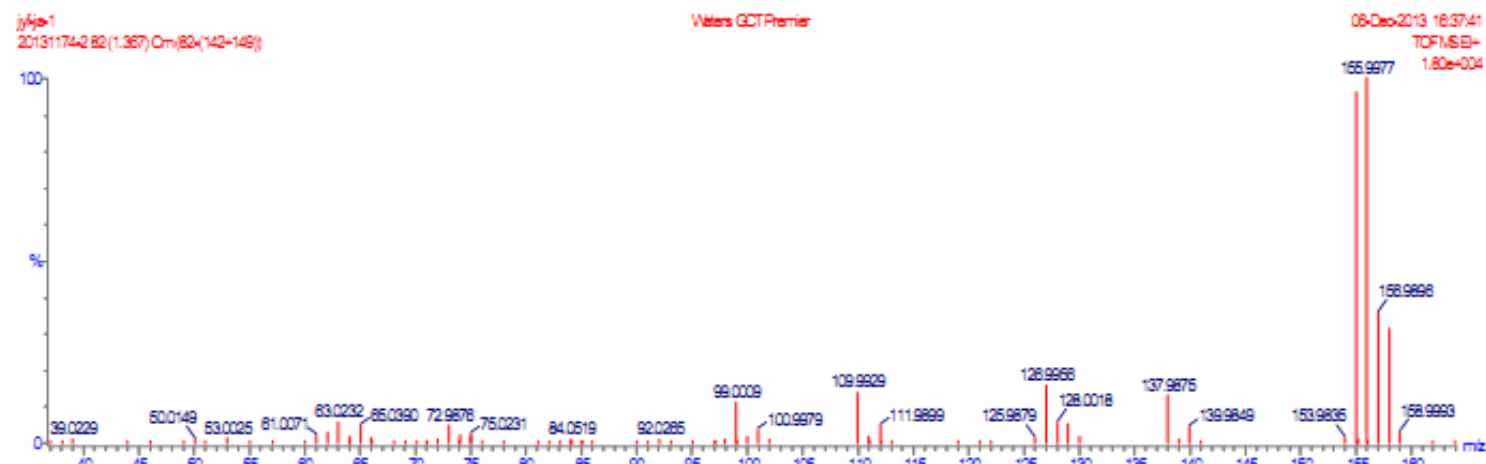
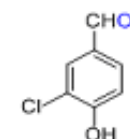
Monoisotopic Mass, Odd and Even Electron Ions

109 formula(e) evaluated with 19 results within limits (up to 50 closest results for each mass)

Elements Used: C: 0-7 H: 0-5 O: 0-2 ³⁵Cl: 0-1 ³⁷Cl: 0-1

jyAja-1
201311742 82 (1.367) Cm(82+142+148)

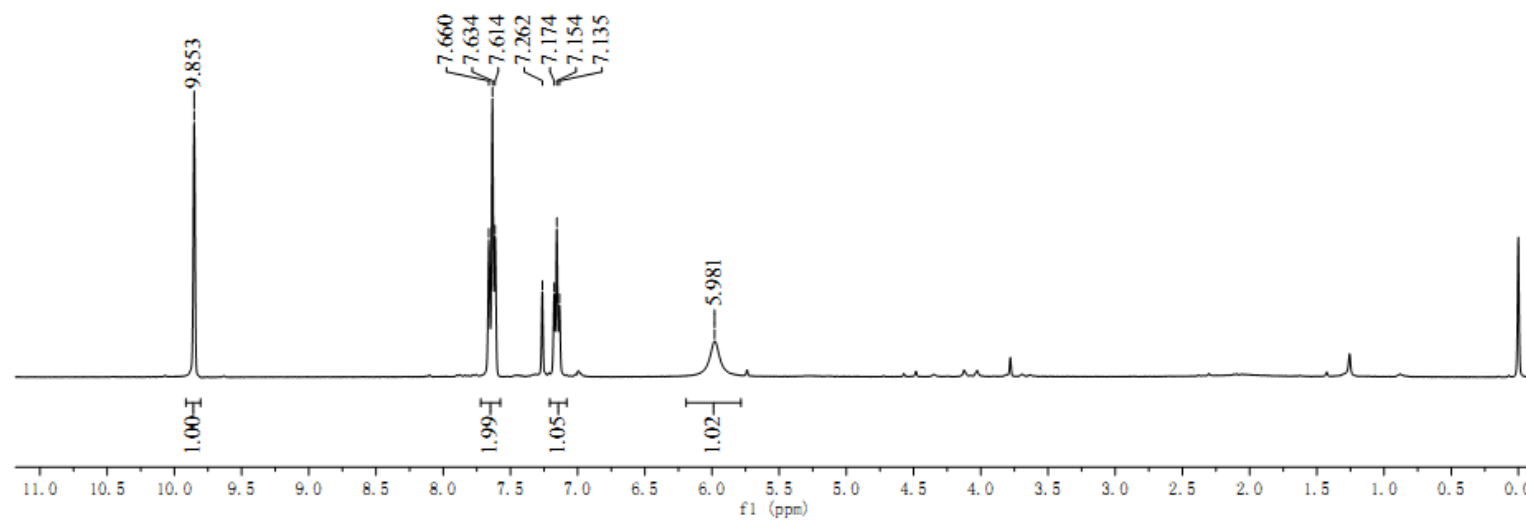
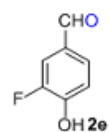
Waters GCT Premier

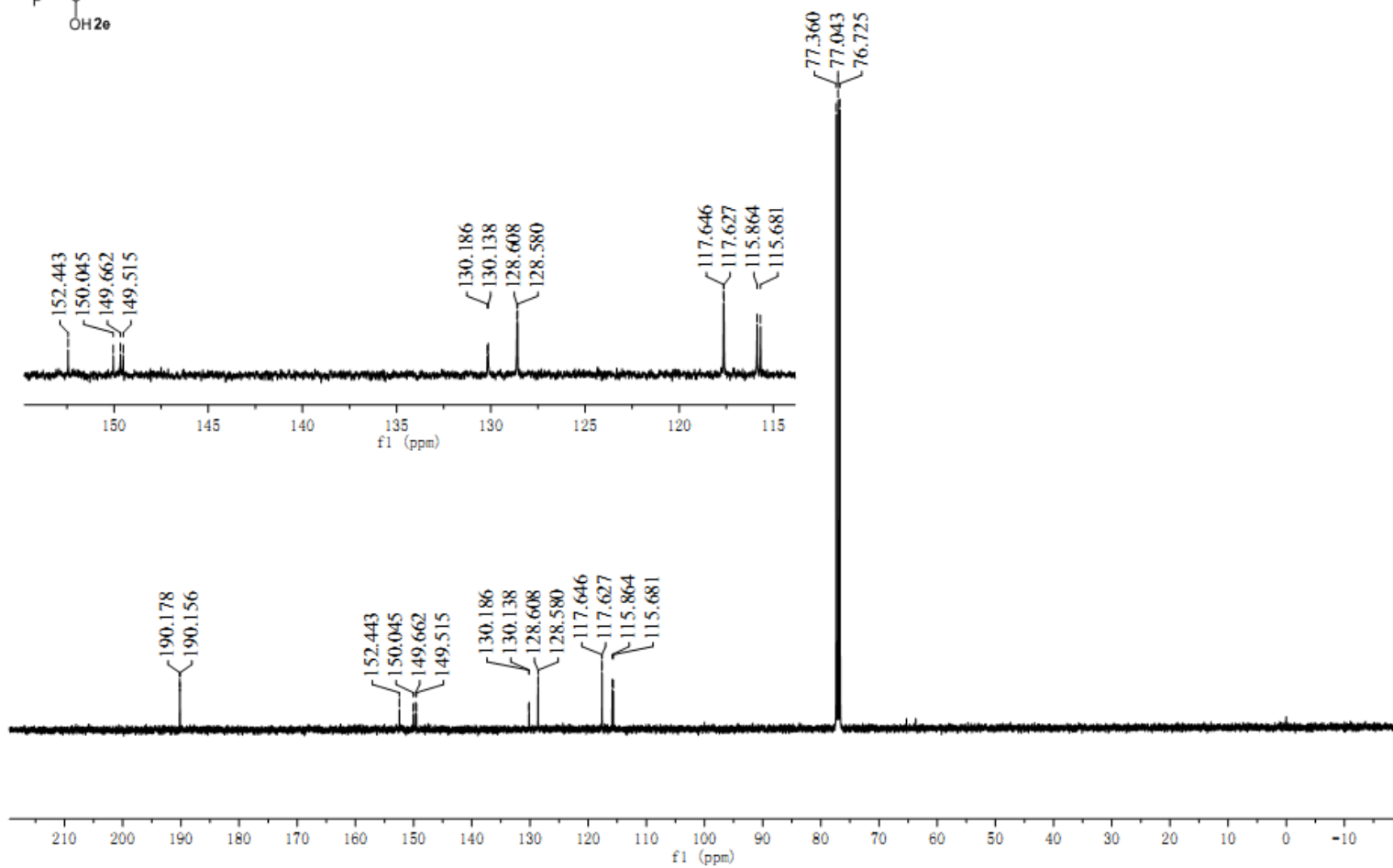
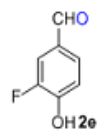


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	I-FIT	Formula
155.9977	100.00	155.9978	-0.1	-0.6	5.0	2009.1	C7 H5 O2 ³⁵ Cl





Elemental Composition Report

Single Mass Analysis

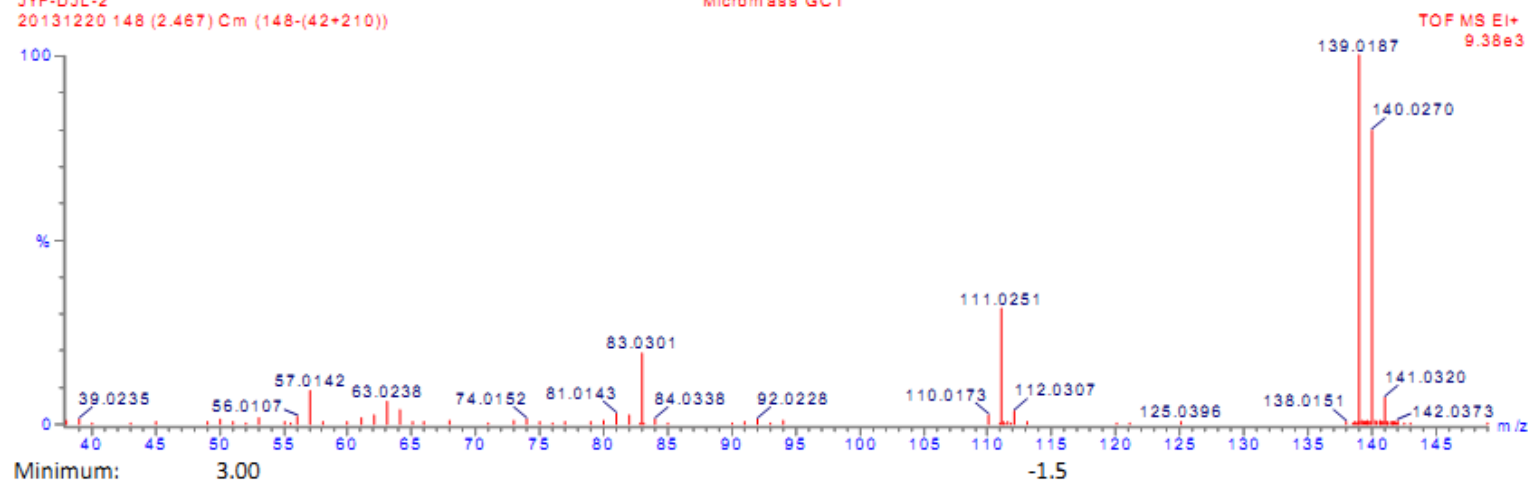
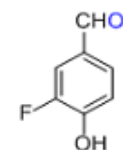
Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50

Monoisotopic Mass, Odd and Even Electron Ions

32 formula(e) evaluated with 10 results within limits (up to 50 closest results for each mass)

JYF-DJL-2
20131220 148 (2.467) Cm (148-(42+210))

Micromass GCT



Minimum: 3.00

-1.5

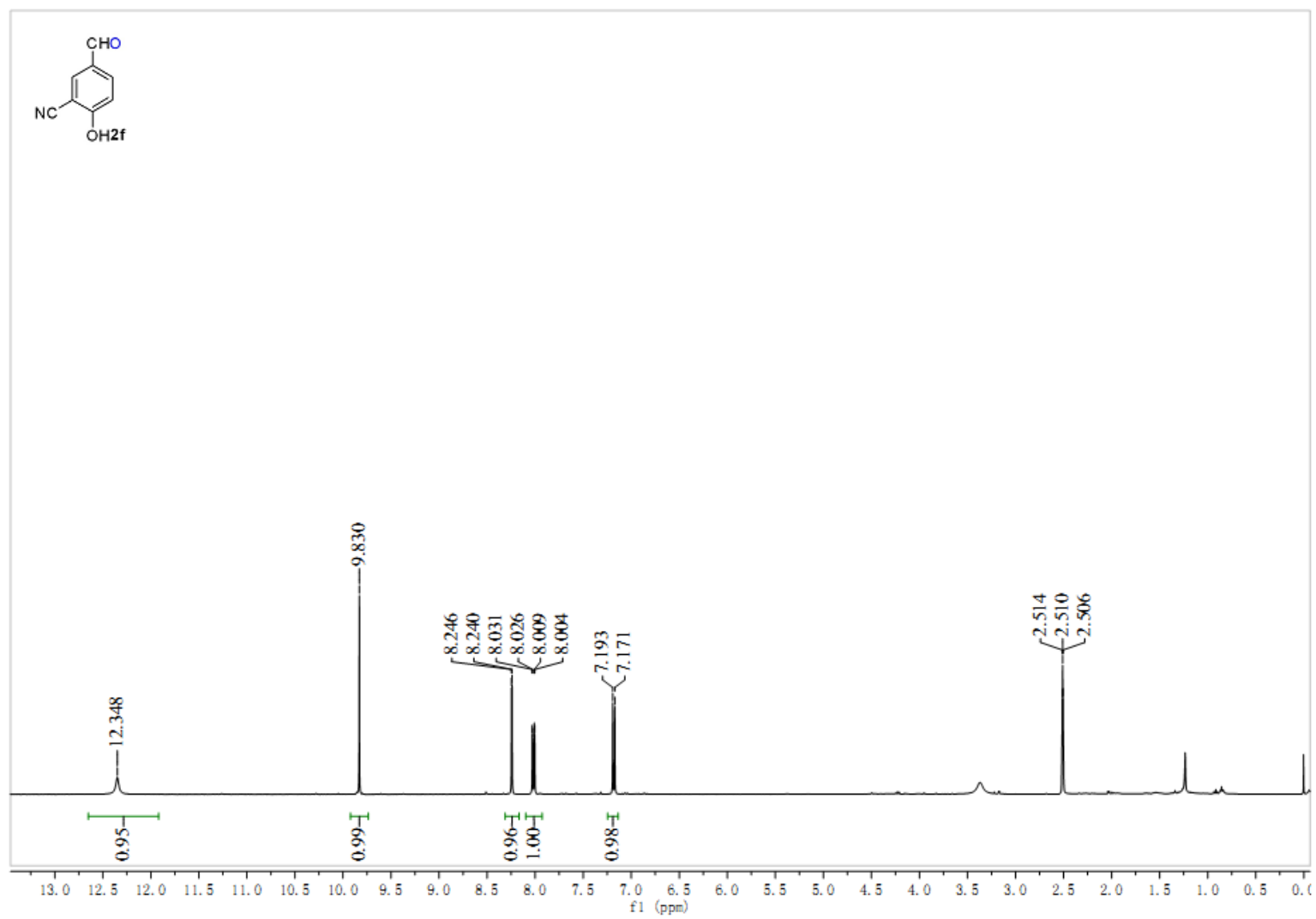
Maximum: 100.00

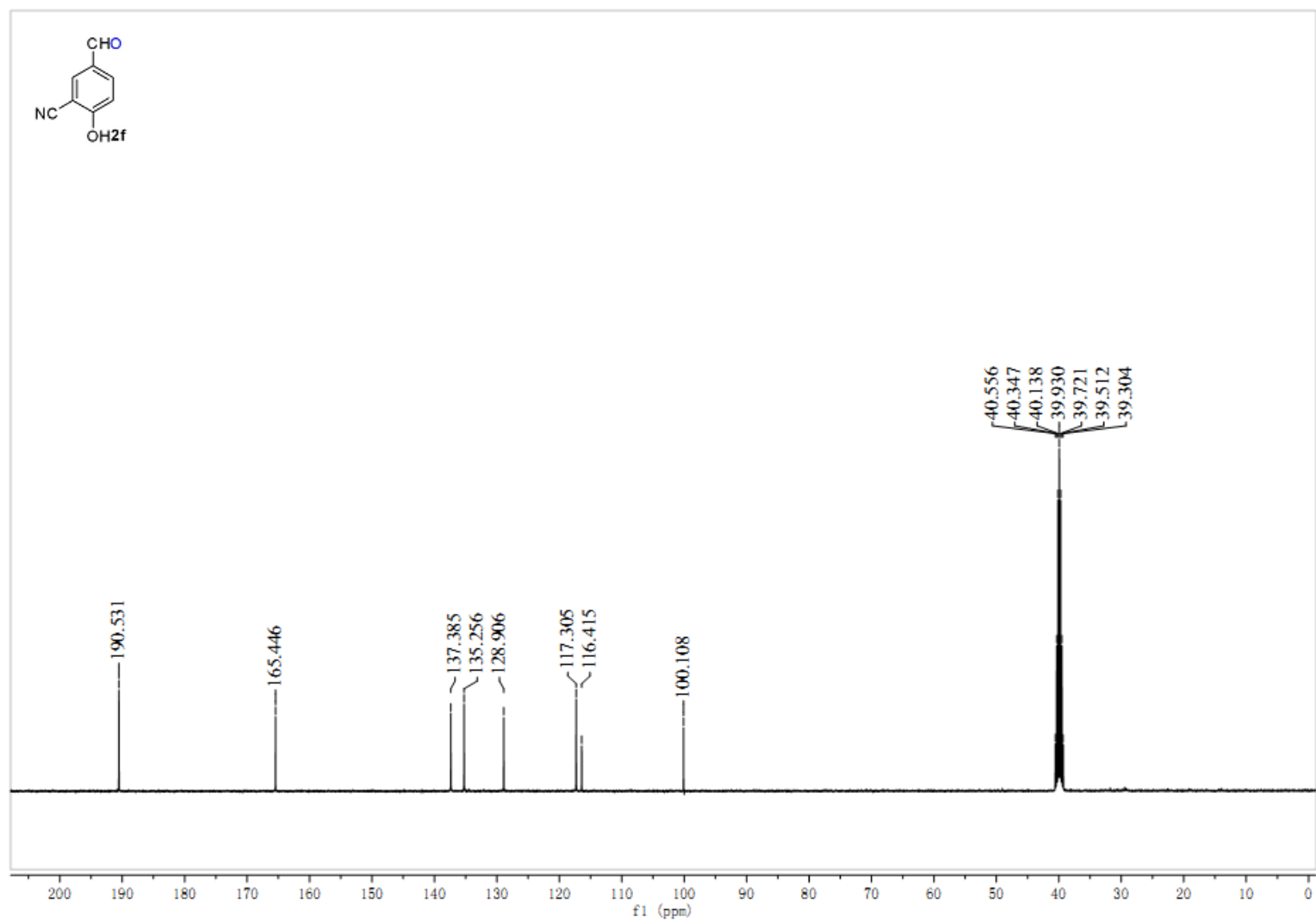
5.0

10.0

50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
140.0270	79.65	140.0274	-0.4	-2.6	5.0	1	C7 H5 O2 F





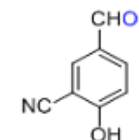
Elemental Composition Report

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Page 1

Monoisotopic Mass, Even Electron Ions

34 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 N: 0-3 O: 0-3

YF-JI

ECUST institute of Fine Chem

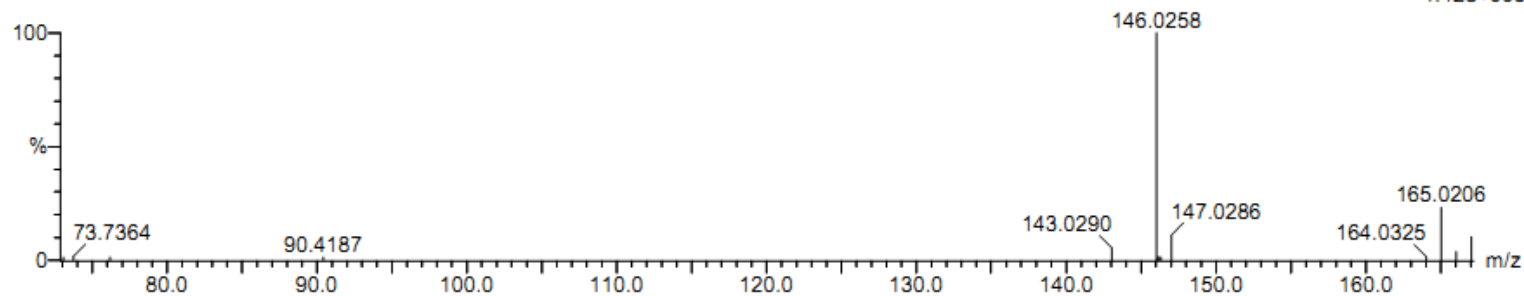
16-Dec-2013

21:44:39

2: TOF MS ES-

1.12e+003

JYF-JA-6 27 (0.934) Cm (27:28)



Minimum:

-1.5

Maximum:

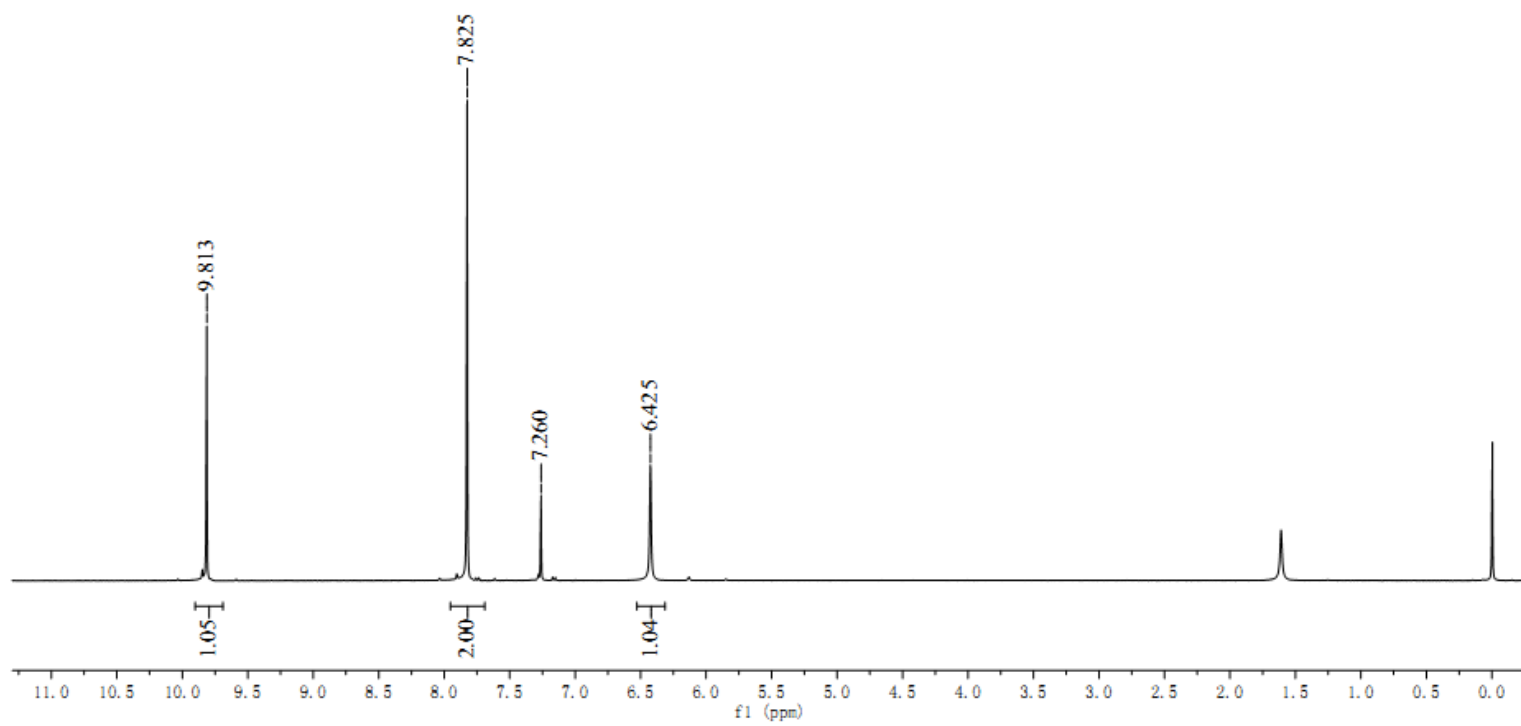
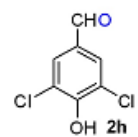
30.0

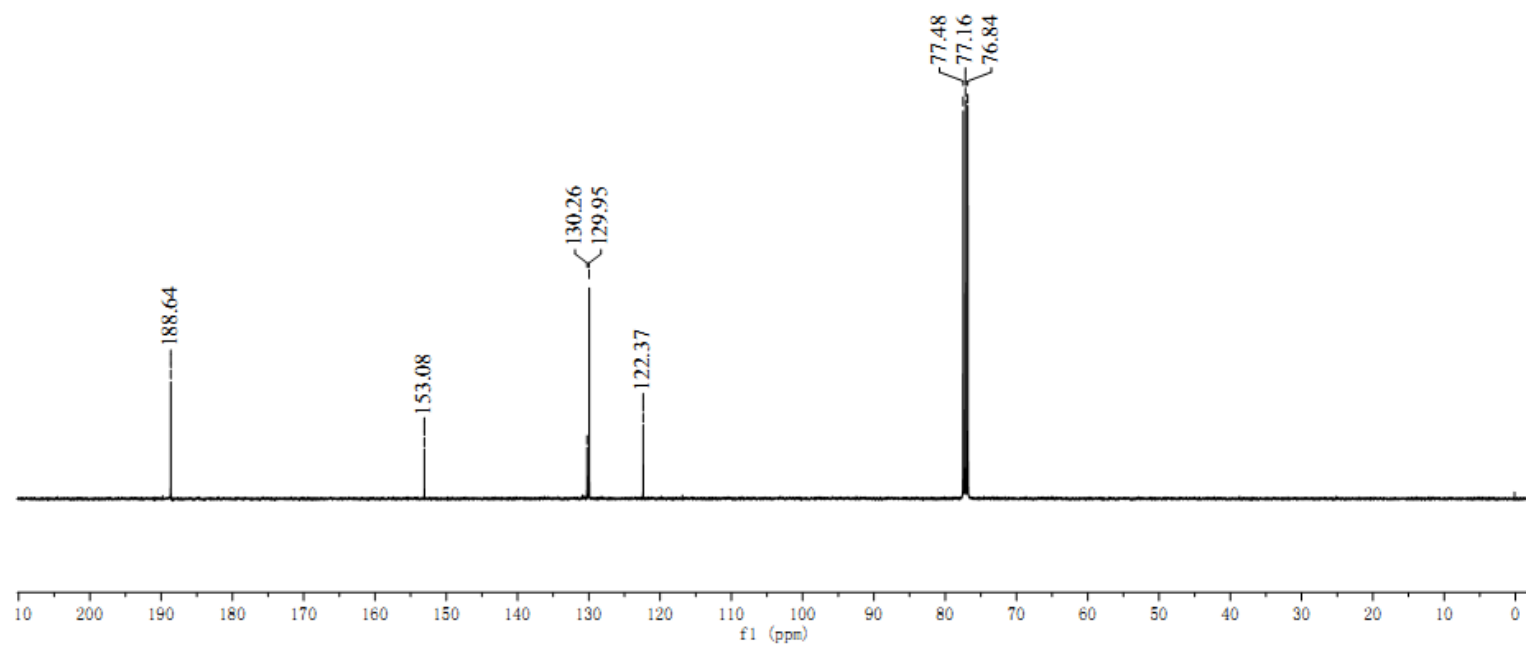
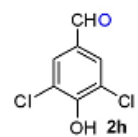
50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
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146.0258	146.0242	1.6	11.0	7.5	22.0	0.0	C8 H4 N O2
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Elemental Composition Report

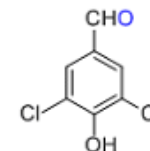
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

23 formula(e) evaluated with 2 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 O: 0-2 Cl: 0-2

YF-JI

ECUST institute of Fine Chem

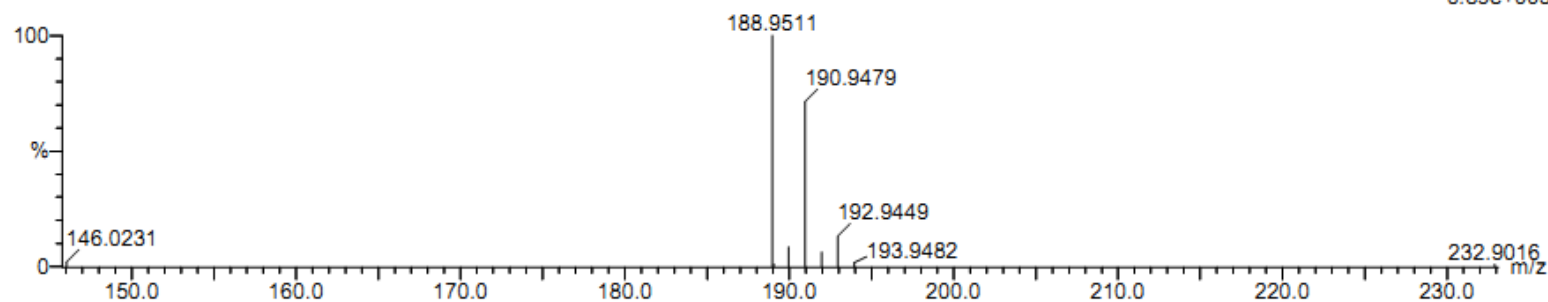
16-Dec-2013

21:34:38

JYF-JA-2 5 (0.271) Cm (4:8)

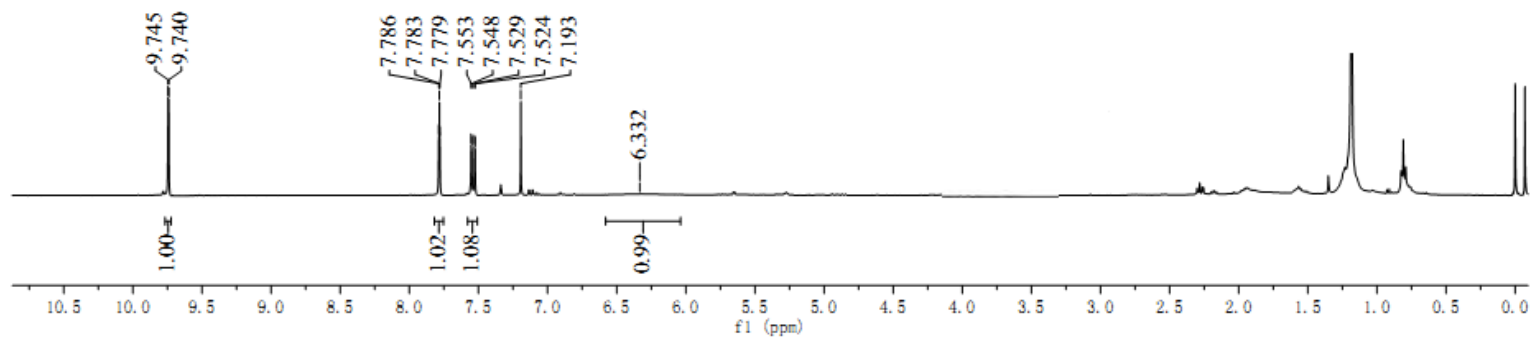
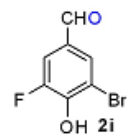
2: TOF MS ES-

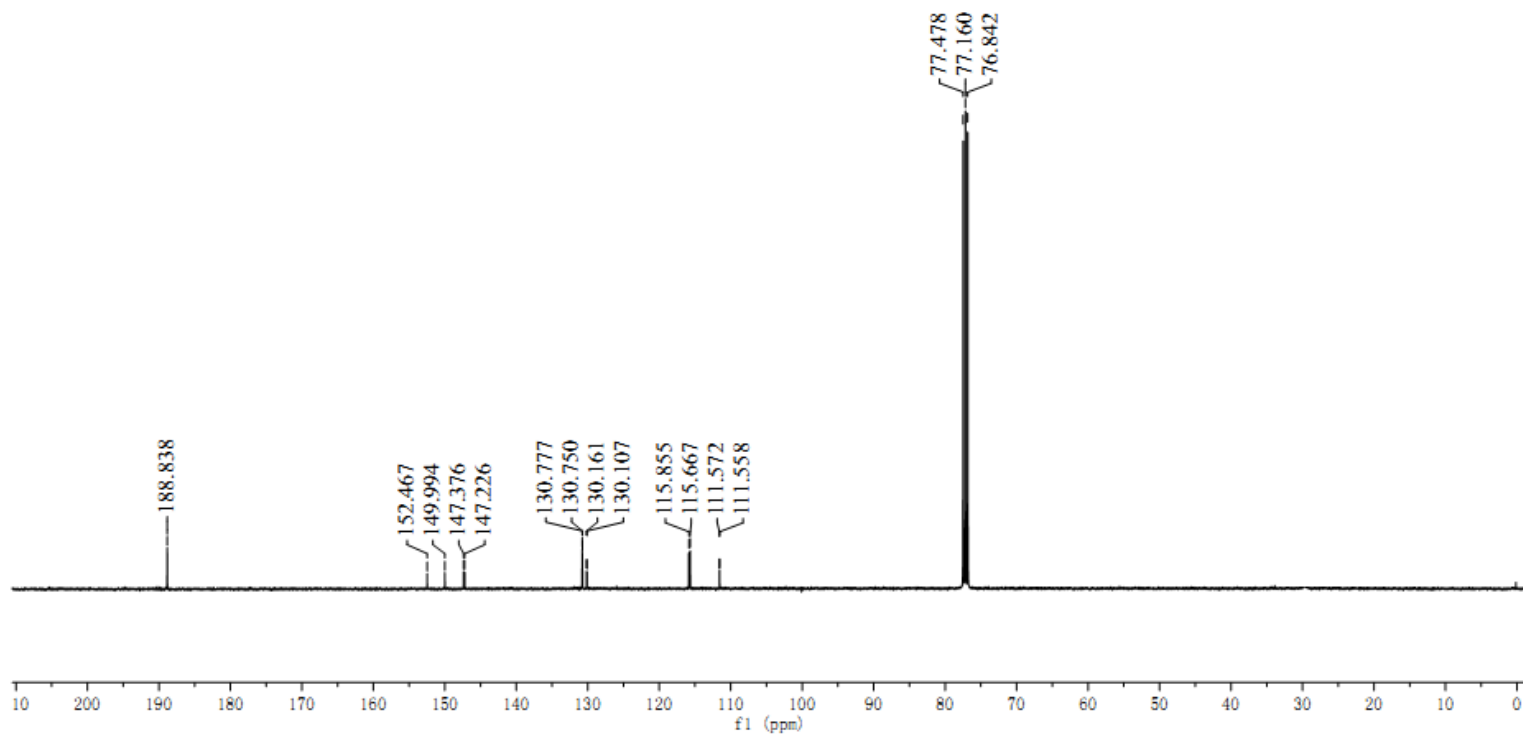
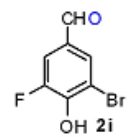
6.89e+003



Minimum: -1.5
Maximum: 30.0 50.0 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
188.9511	188.9510	0.1	0.5	5.5	14.8	0.0	C7 H3 O2 Cl2





Elemental Composition Report

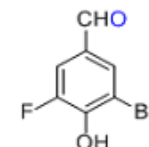
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

68 formula(e) evaluated with 7 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 O: 0-2 Br: 0-2 F: 0-3

YF-JI

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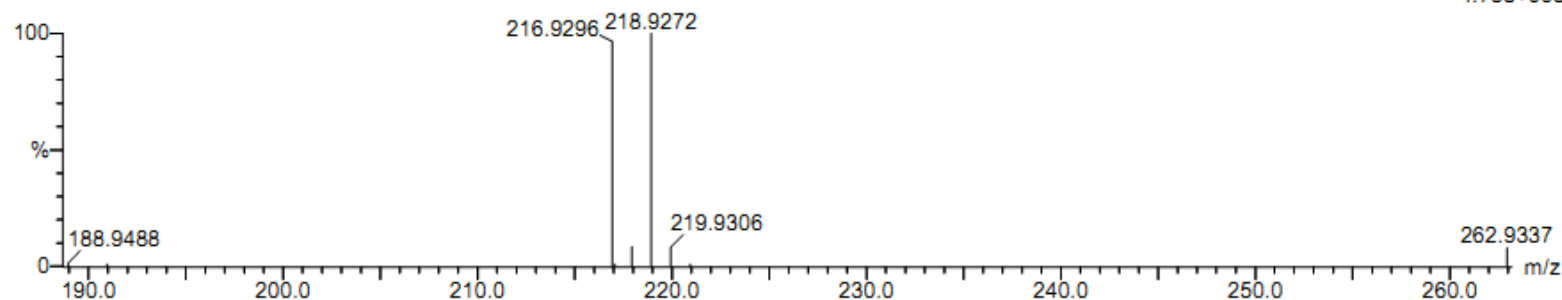
16-Dec-2013

21:36:55

JYF-JA-3 10 (0.419) Cm (9:11)

2: TOF MS ES-

4.75e+003



Minimum:

-1.5

Maximum:

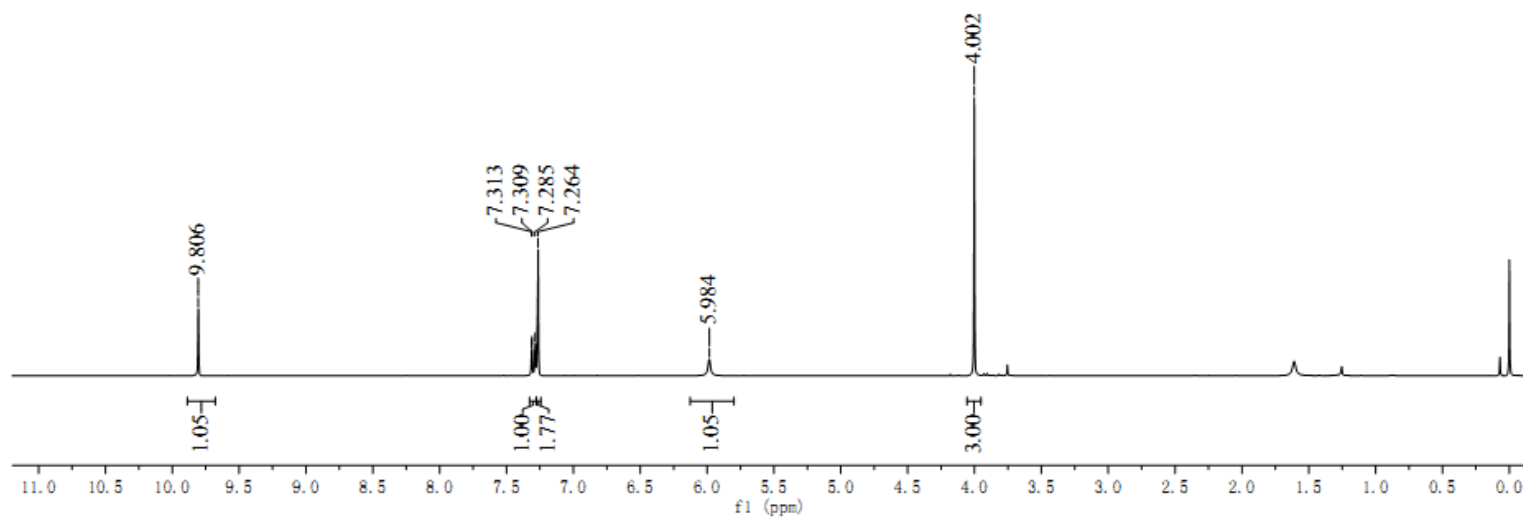
30.0

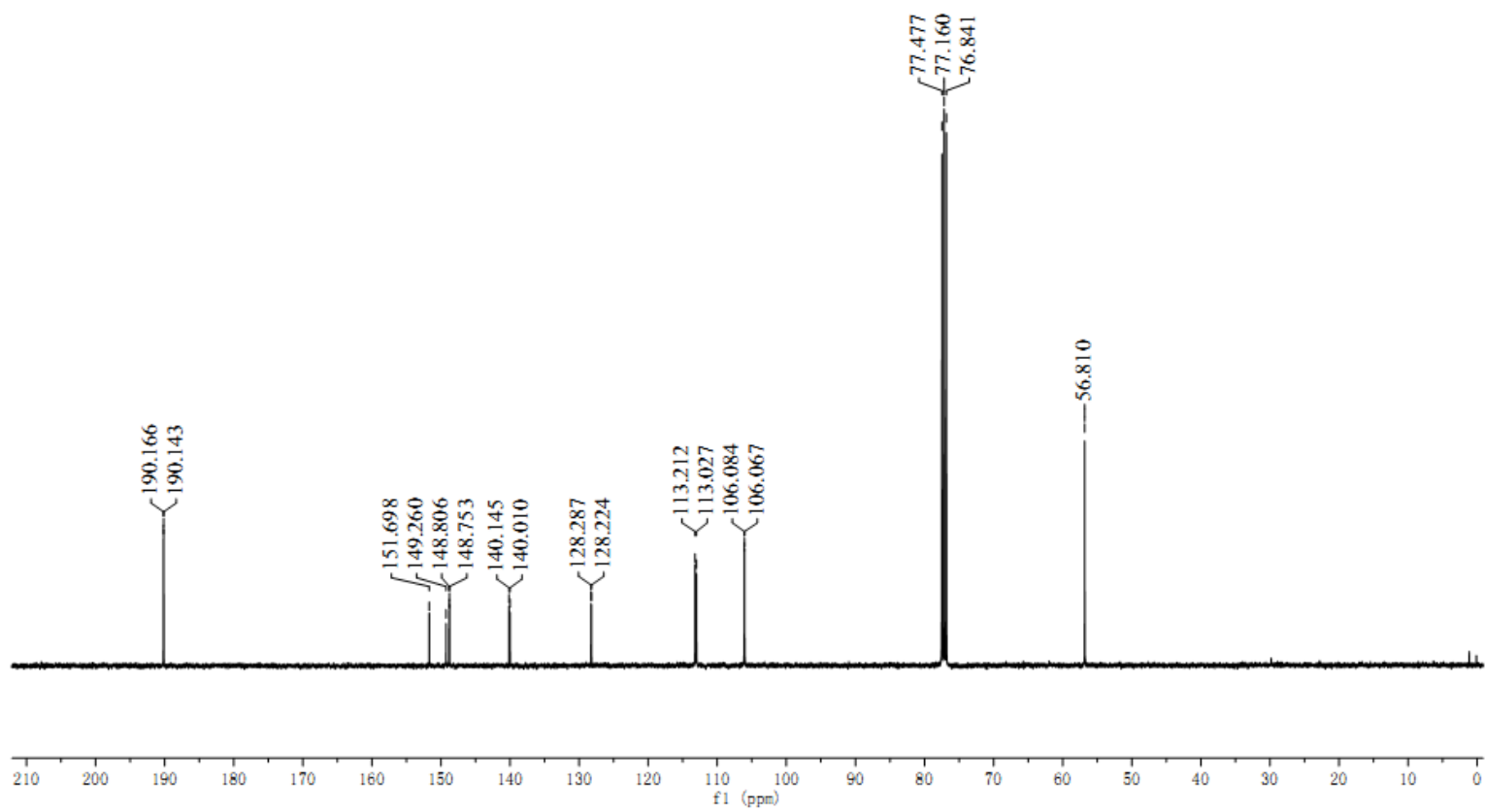
50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
------	------------	-----	-----	-----	-------	--------------	---------

216.9296	216.9300	-0.4	-1.8	5.5	13.8	0.0	C7 H3 O2 Br F
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Elemental Composition Report

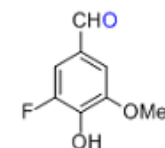
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

37 formula(e) evaluated with 10 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 O: 0-3 F: 0-3

YF-JI

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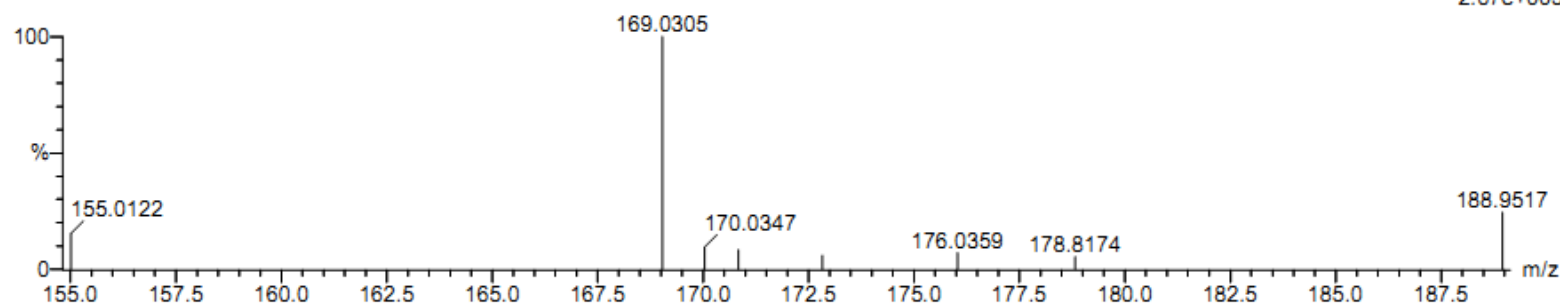
16-Dec-2013

21:39:51

2: TOF MS ES-

2.67e+003

JYF-JA-4 12 (0.466) Cm (7:14)



Minimum:

-1.5

Maximum:

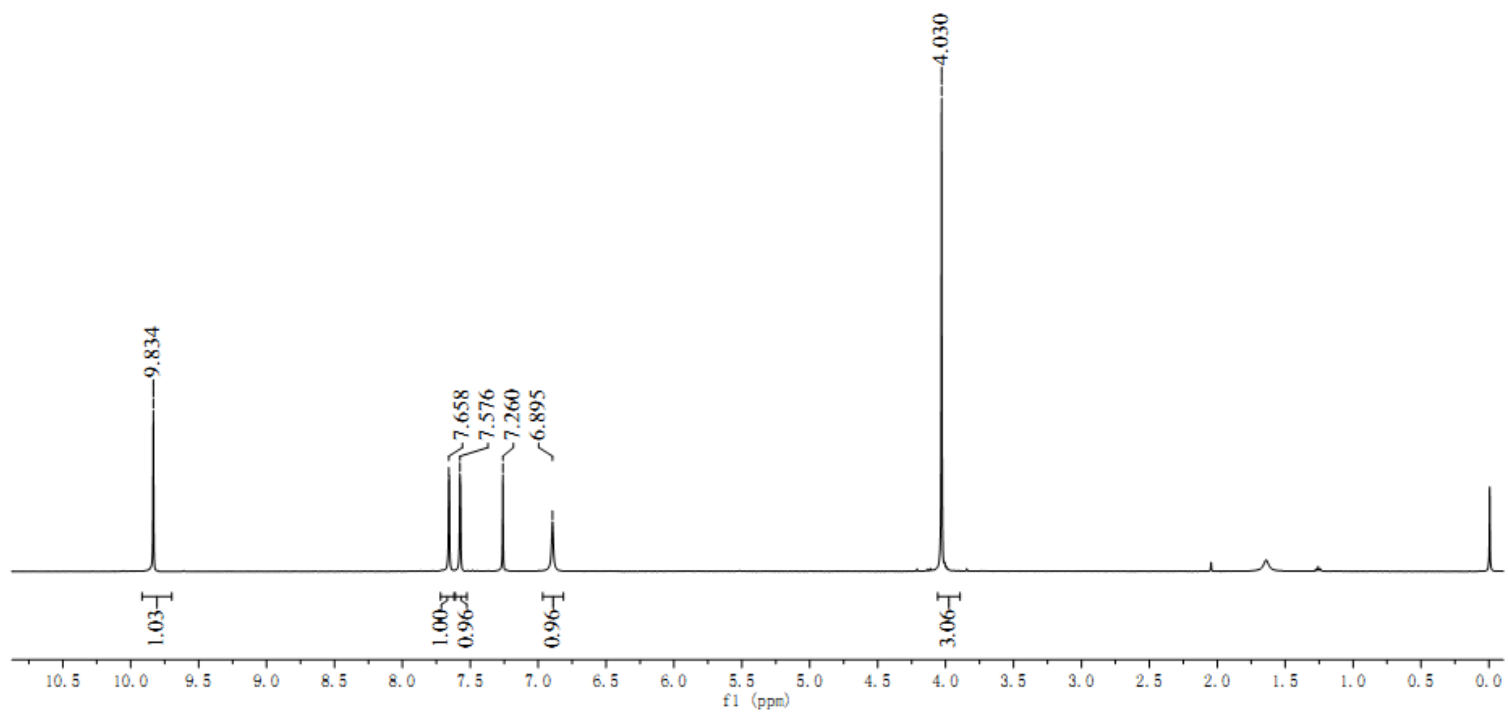
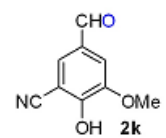
30.0

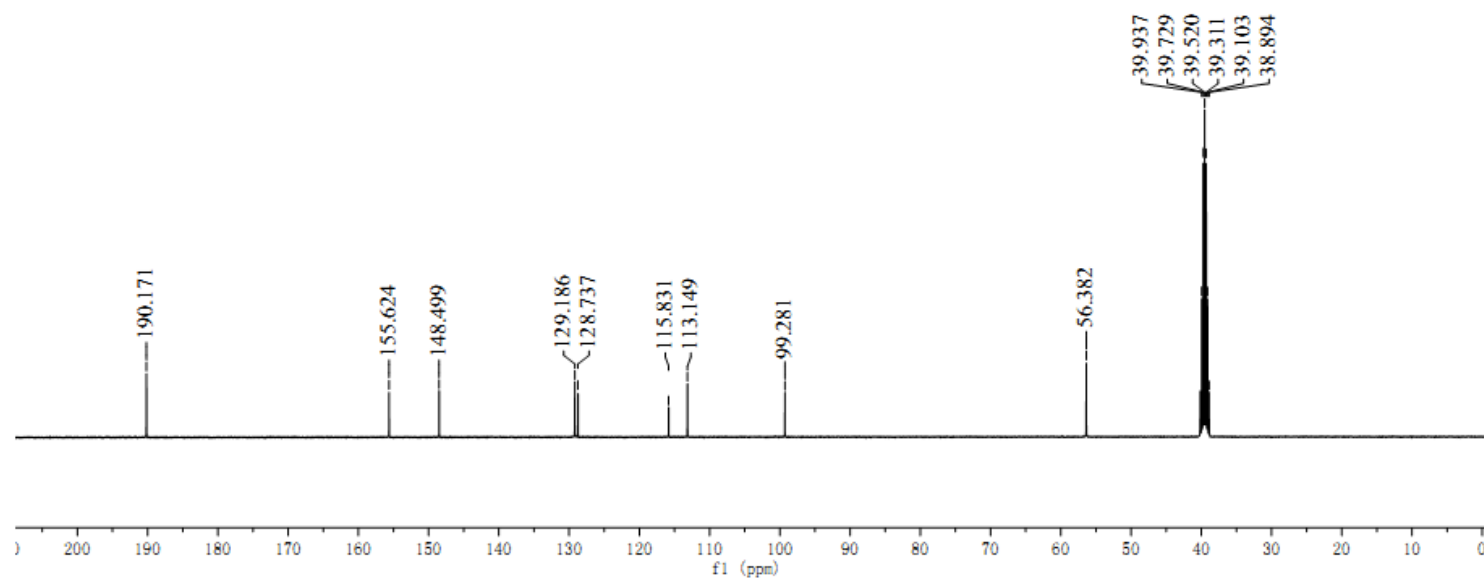
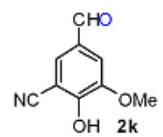
50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
------	------------	-----	-----	-----	-------	--------------	---------

169.0305	169.0301	0.4	2.4	5.5	5.4	0.0	C8 H6 O3 F
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Elemental Composition Report

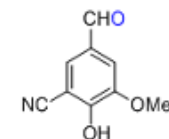
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

40 formula(e) evaluated with 6 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-45 H: 0-70 N: 0-3 O: 0-3

YF-JI

ECUST institute of Fine Chem

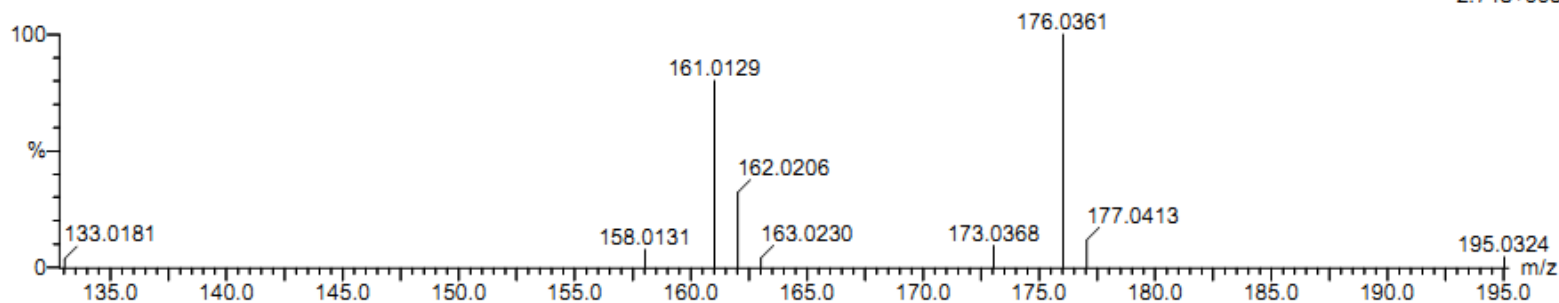
16-Dec-2013

21:41:57

2: TOF MS ES-

2.74e+003

JYF-JA-5 17 (0.639) Cm (17:20)



Minimum:

-1.5

Maximum:

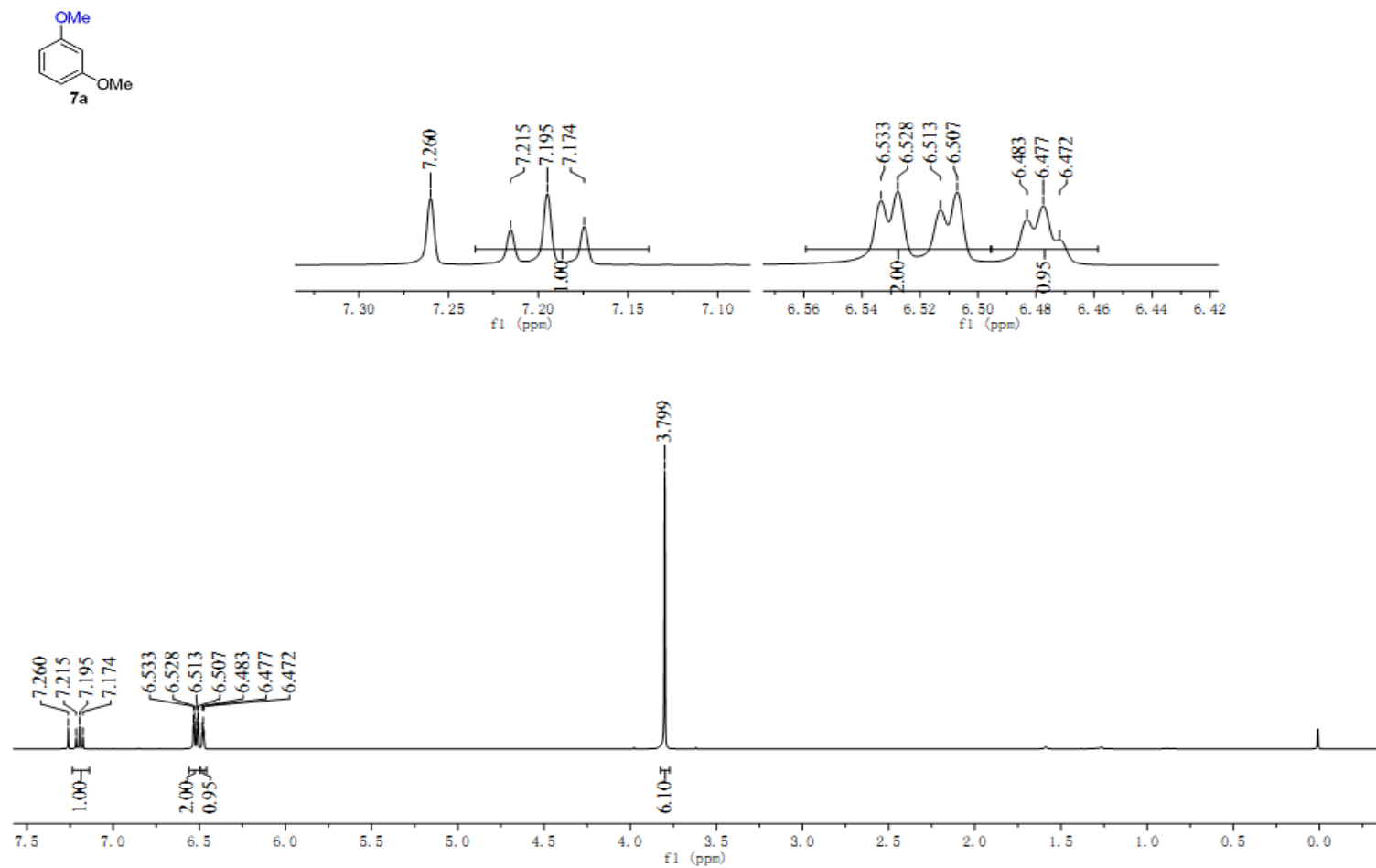
30.0

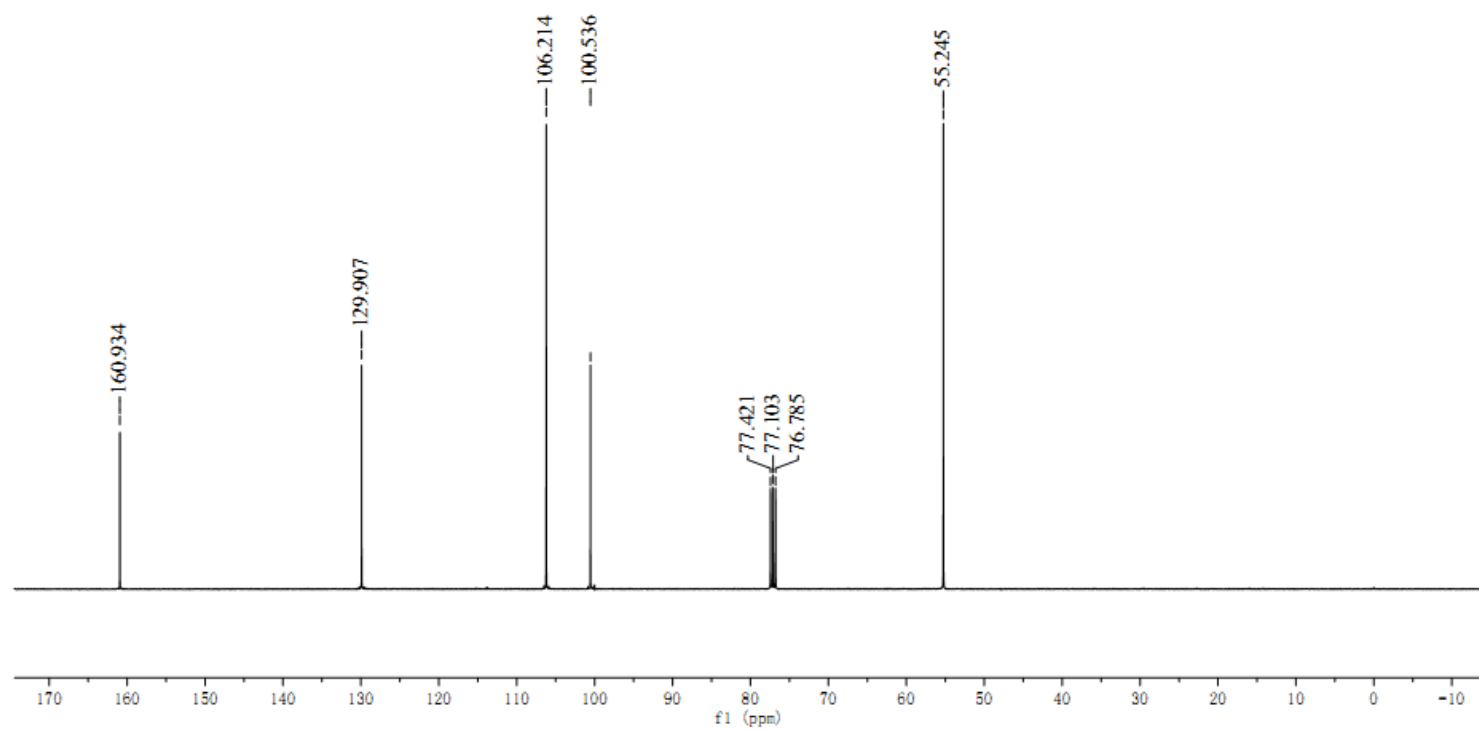
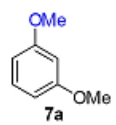
50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
176.0361	176.0348	1.3	7.4	7.5	8.3	0.0	C9 H6 N O3

4.8 Copies of spectra for CuCl/HCOOMe-catalyzed methoxylation.





Elemental Composition Report

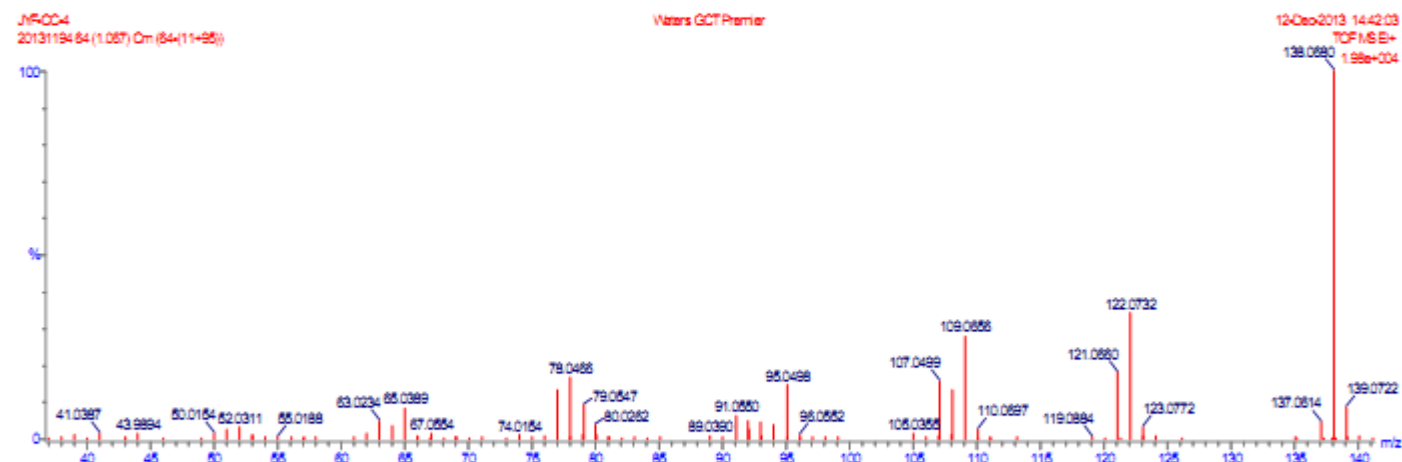
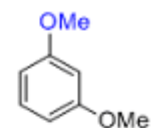
Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50

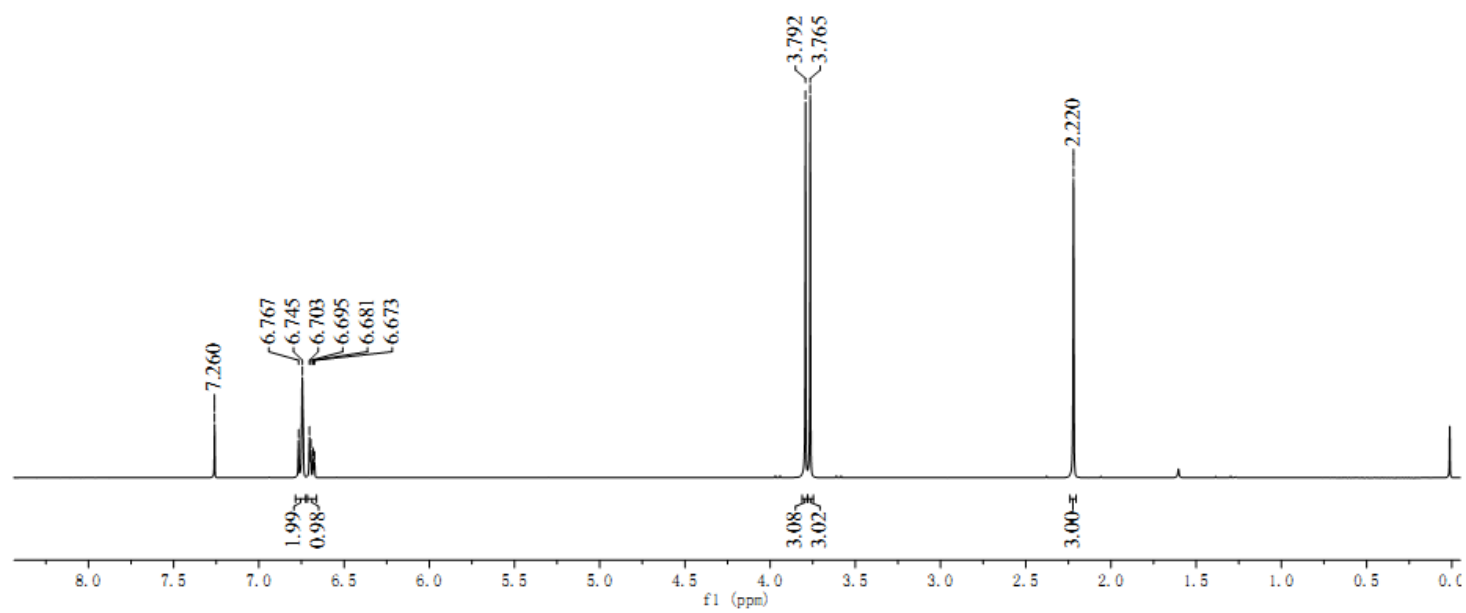
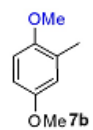
Monoisotopic Mass, Odd and Even Electron Ions

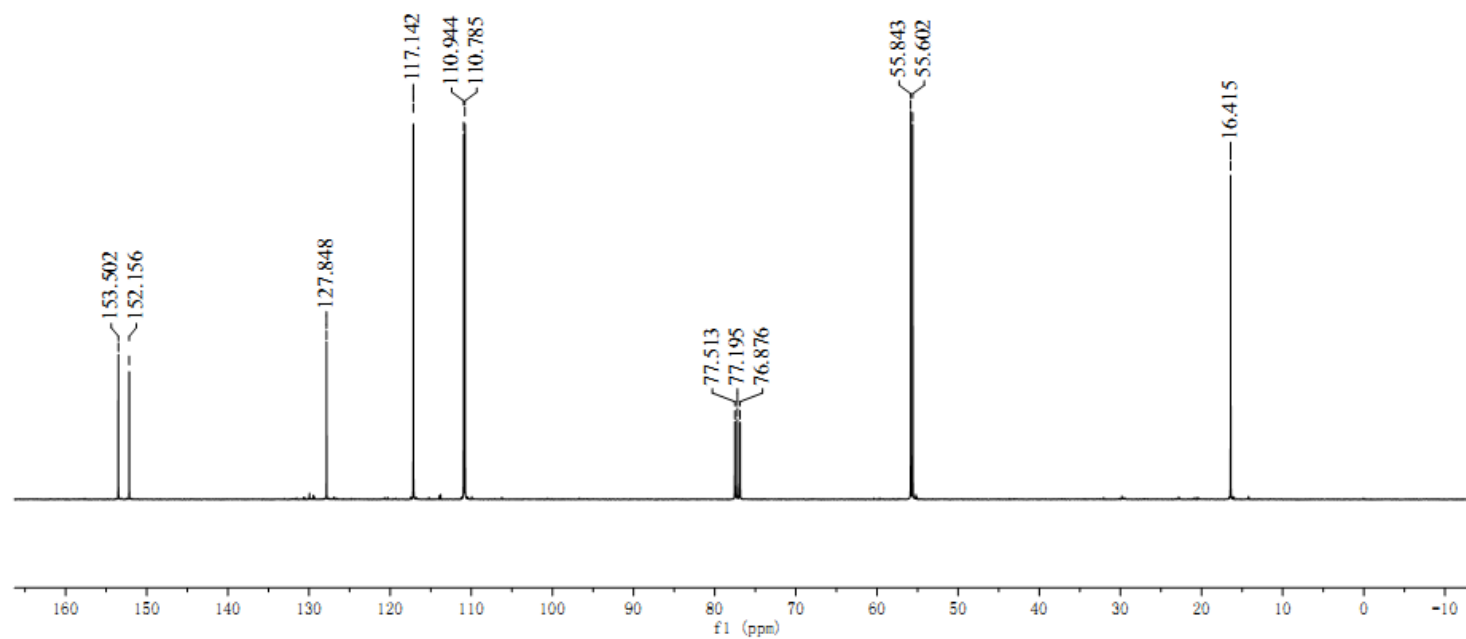
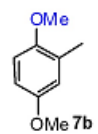
63 formula(e) evaluated with 23 results within limits (up to 50 closest results for each mass)

Elements Used: C: 0-8 H: 0-10 O: 0-2



Minimum:	3.00							-1.5
Maximum:	100.00		5.0	10.0	50.0			
Mass	RA	Calc. Mass	mDa	PPM	DBE	I-FIT	Formula	
138.0680	100.00	138.0681	-0.1	-0.7	4.0	4.9	C8 H10 O2	





Elemental Composition Report

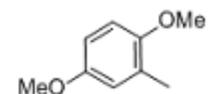
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Odd and Even Electron Ions

19 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-35 H: 0-100 O: 0-10

YF-JI

ECUST institute of Fine Chem

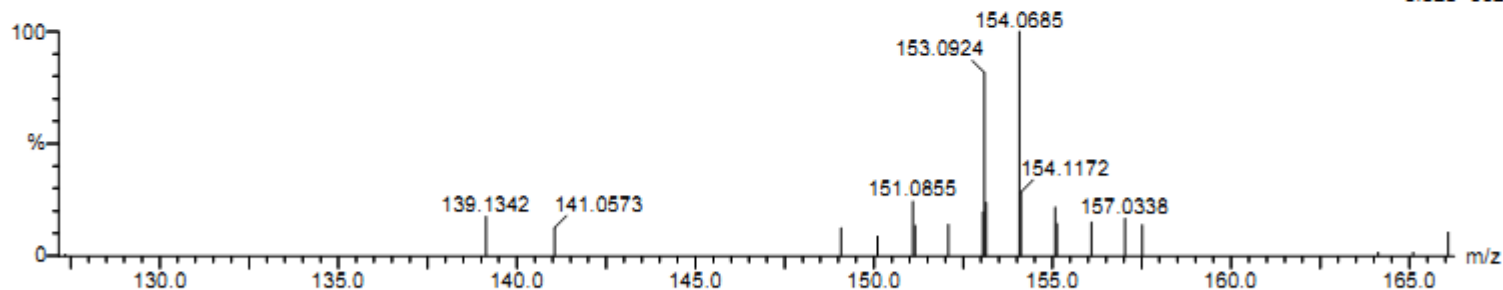
13-Nov-2013

16:19:07

1: TOF MS ES+

5.82e+002

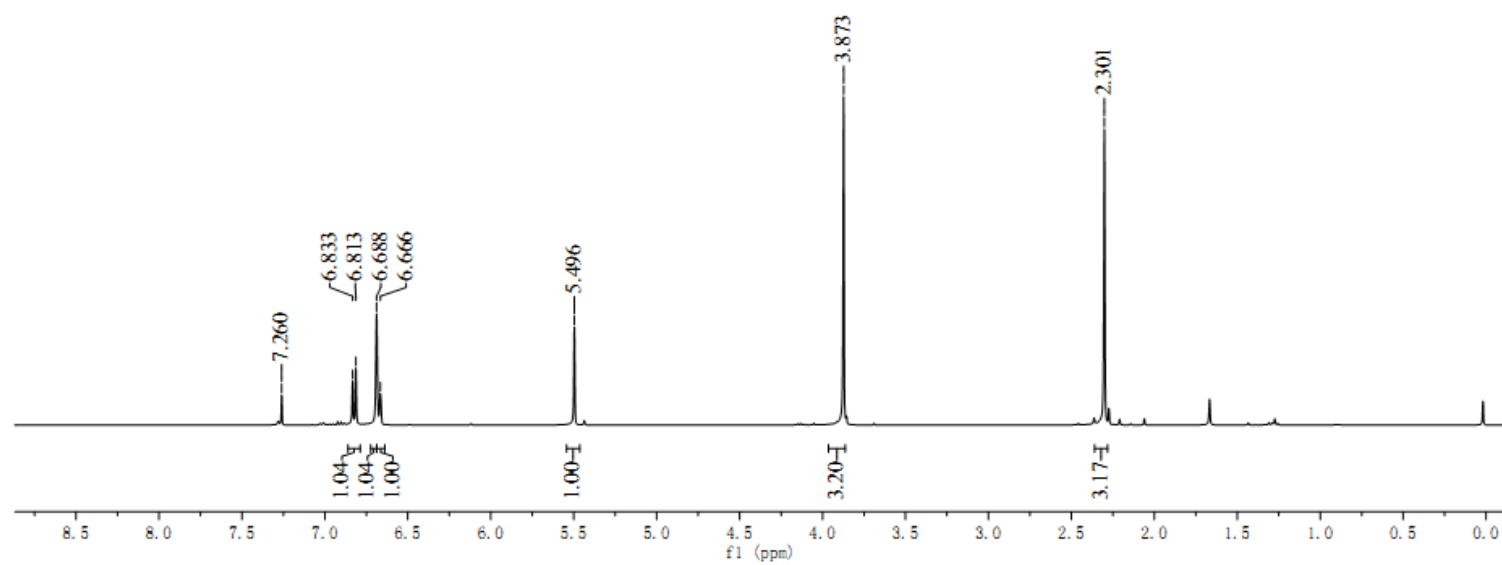
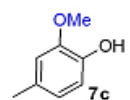
JYF-JA-1 110 (0.775) Cm (109:122)

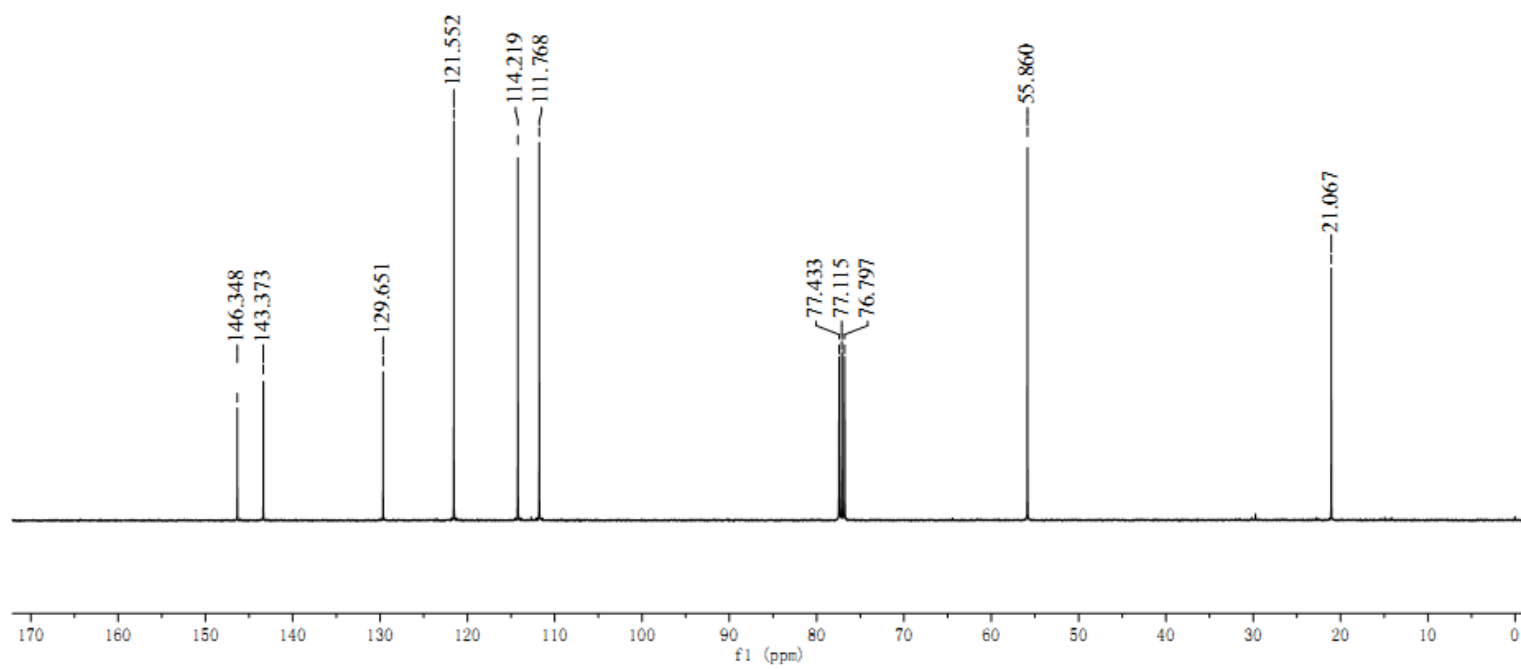
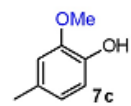


Minimum:

Maximum: 30.0 50.0 -1.5 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
153.0924	153.0916	0.8	5.2	3.5	42.0	0.0	C9 H13 O2





Elemental Composition Report

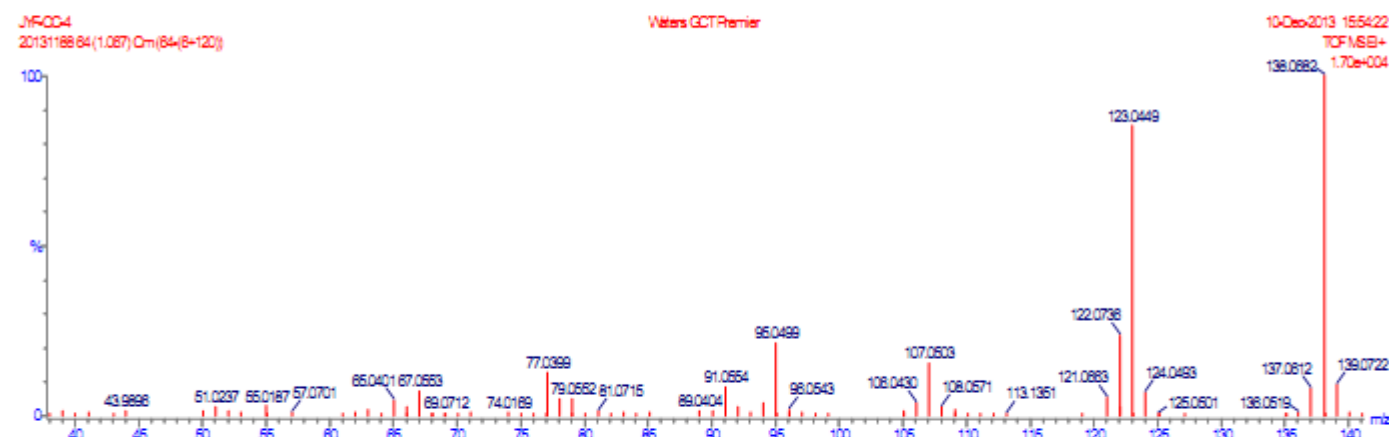
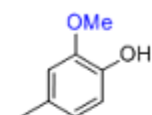
Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50

Monoisotopic Mass, Odd and Even Electron Ions

40 formula(e) evaluated with 16 results within limits (up to 50 closest results for each mass)

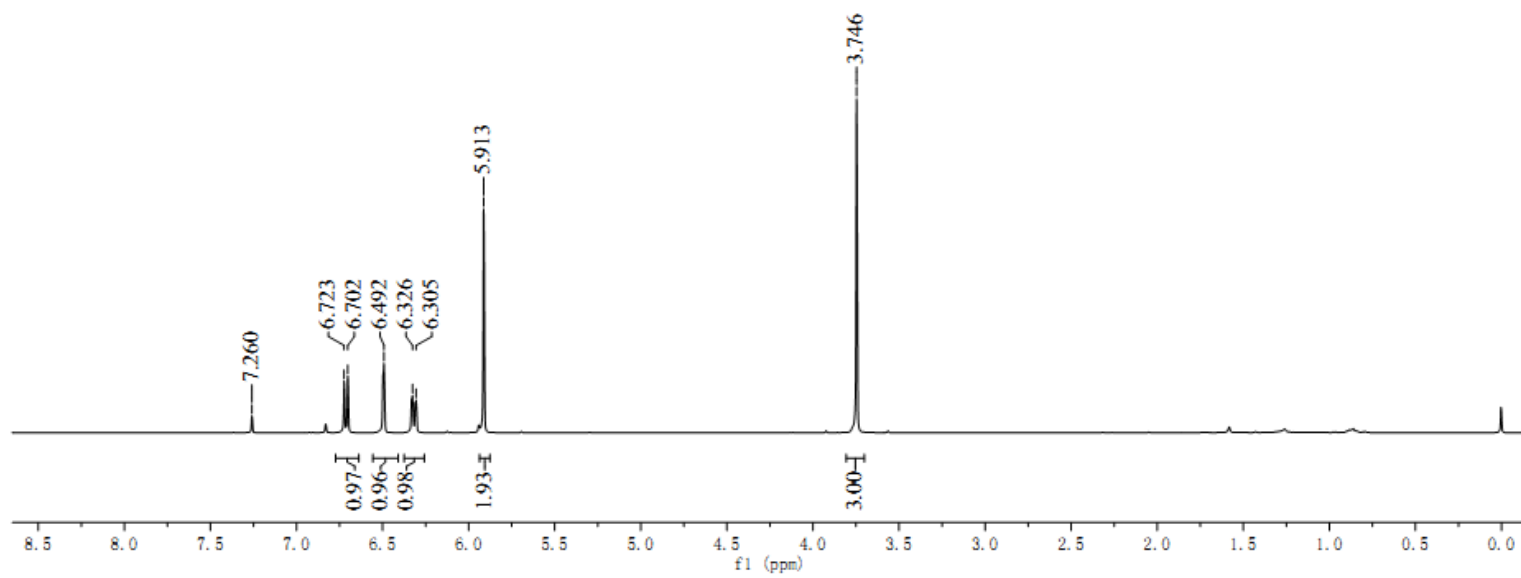
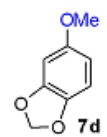
Elements Used: C: 0-8 H: 0-10 O: 0-2

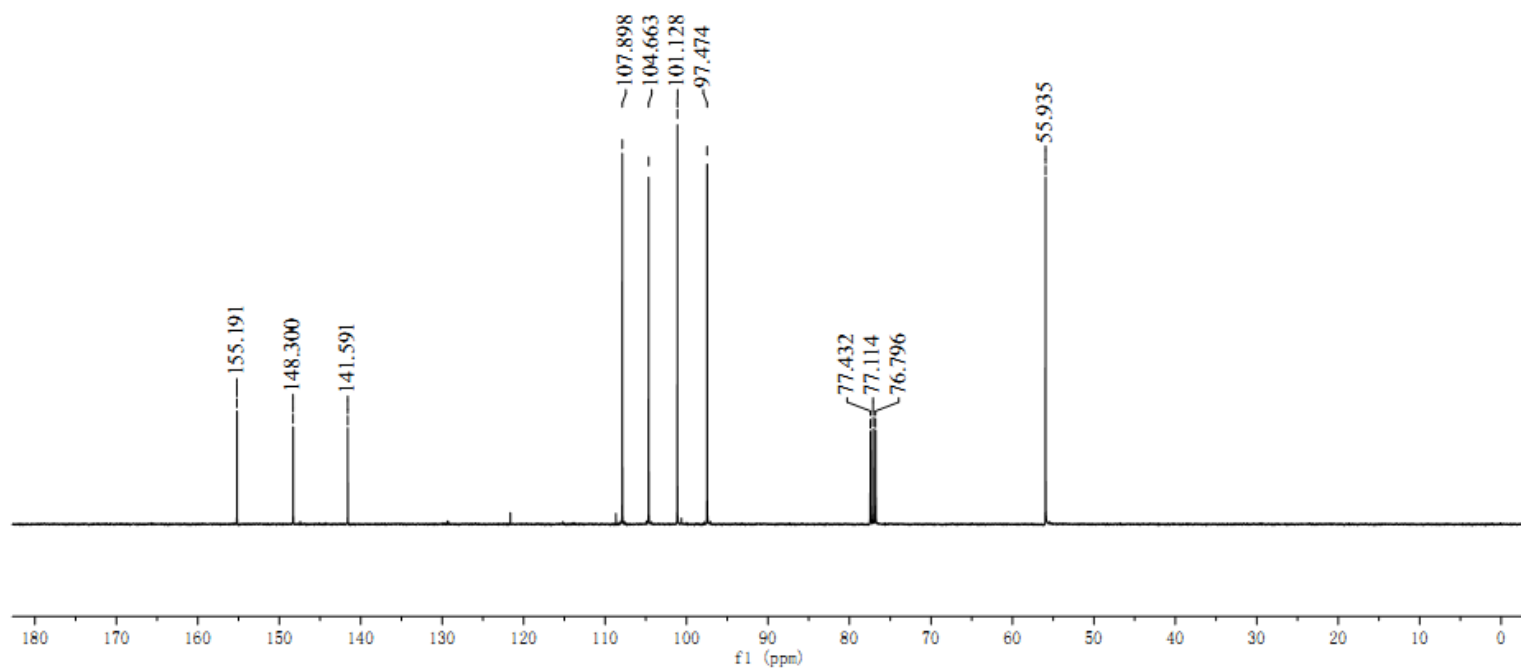
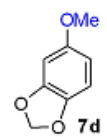


Minimum: 3.00 -1.5

Maximum: 100.00 5.0 10.0 50.0

Mass	RA	Calc. Mass	mDa	PPM	DBE	I-FIT	Formula
138.0682	100.00	138.0681	0.1	0.7	4.0	1.8	C8 H10 O2





Elemental Composition Report

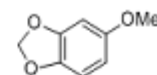
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

10 formula(e) evaluated with 3 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-35 H: 0-50 O: 0-3

YF-JI

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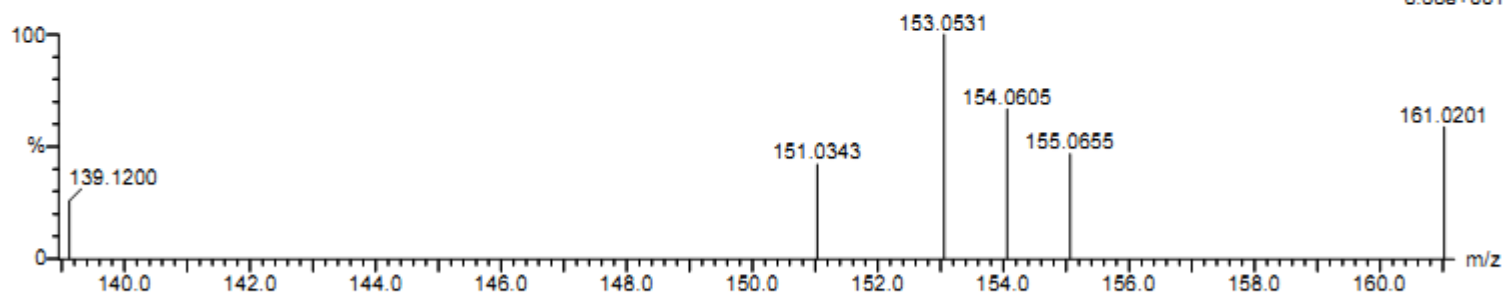
15-Nov-2013

12:51:21

1: TOF MS ES+

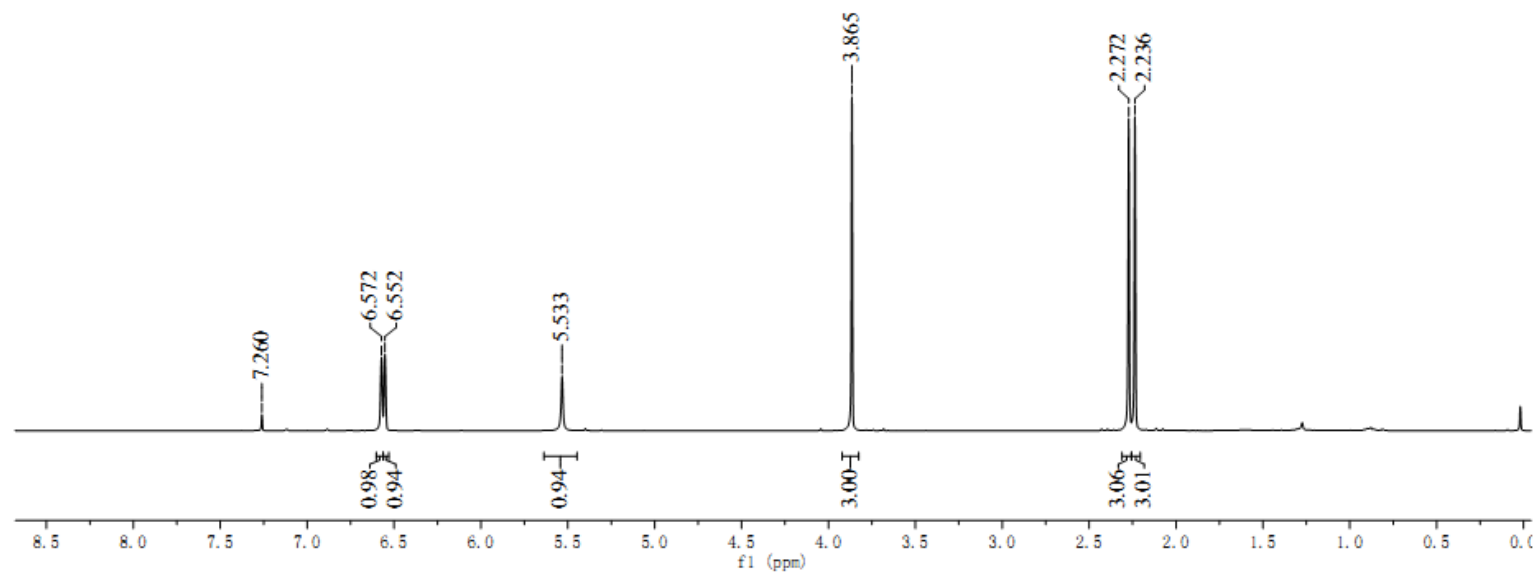
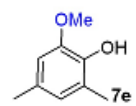
8.68e+001

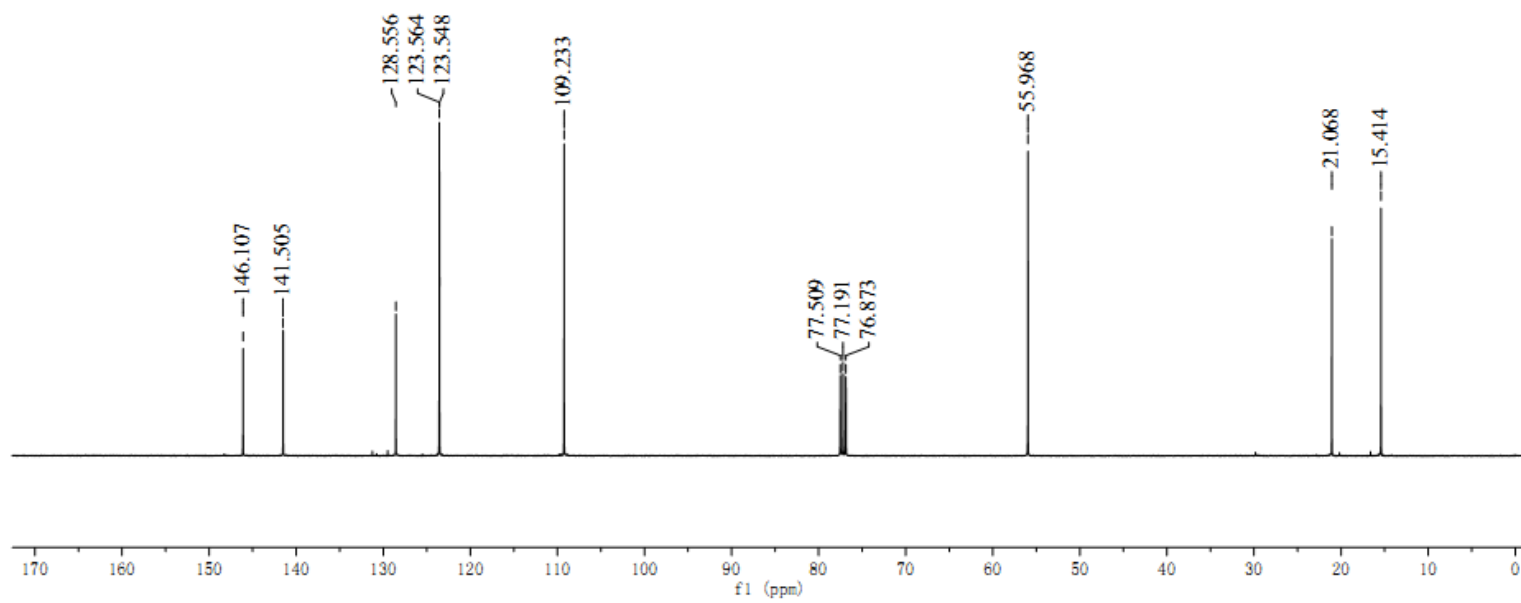
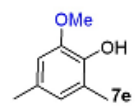
JYF-JA-4 2 (0.160) Cm (1:5)



Minimum: -1.5
Maximum: 30.0 50.0 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
153.0531	153.0552	-2.1	-13.7	4.5	11.8	0.0	C8 H9 O3





Elemental Composition Report

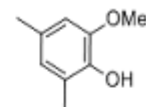
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

10 formula(e) evaluated with 2 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-35 H: 0-50 O: 0-3

YF-JI

ECUST institute of Fine Chem

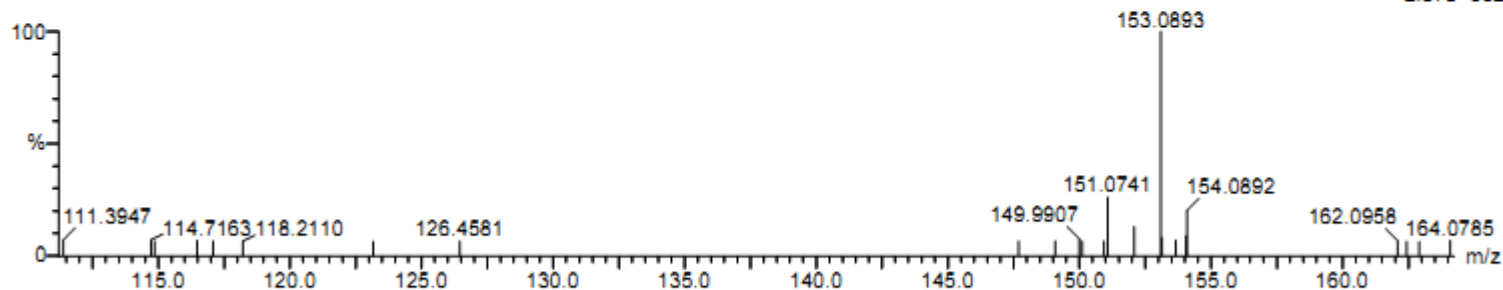
15-Nov-2013

13:06:17

1: TOF MS ES+

2.37e+002

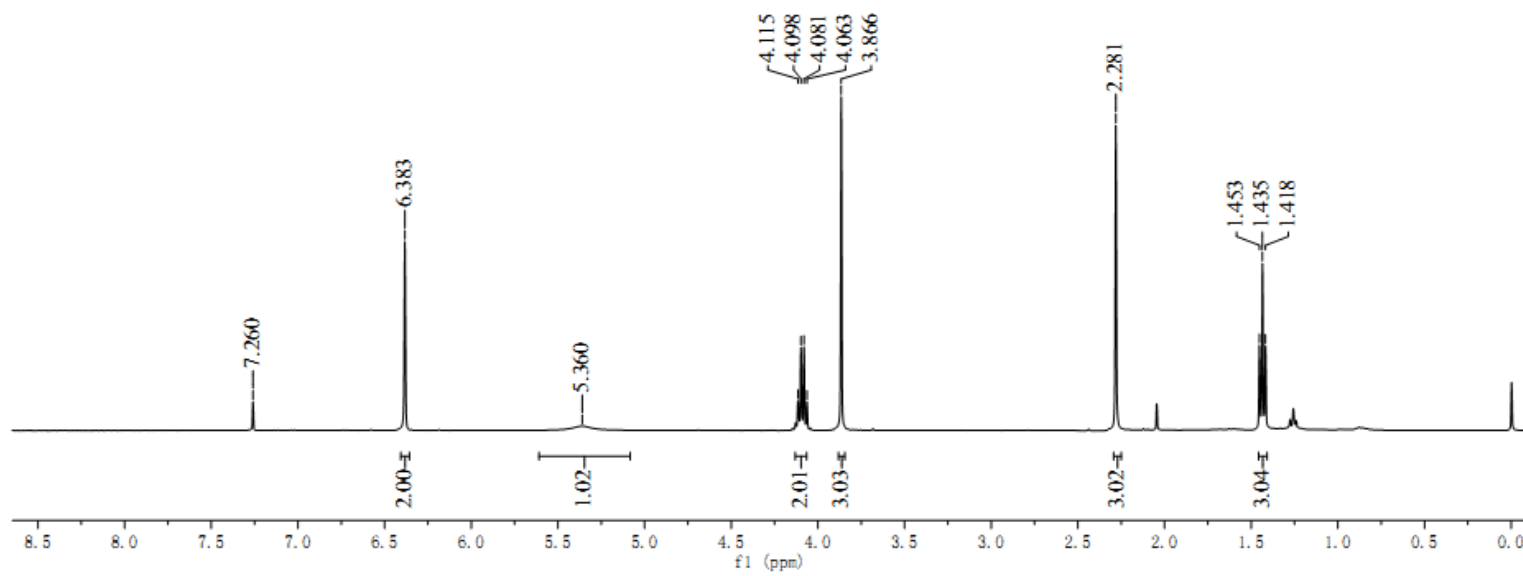
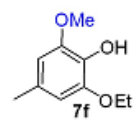
JYF-JA-5 52 (1.681) Cm (49:52)

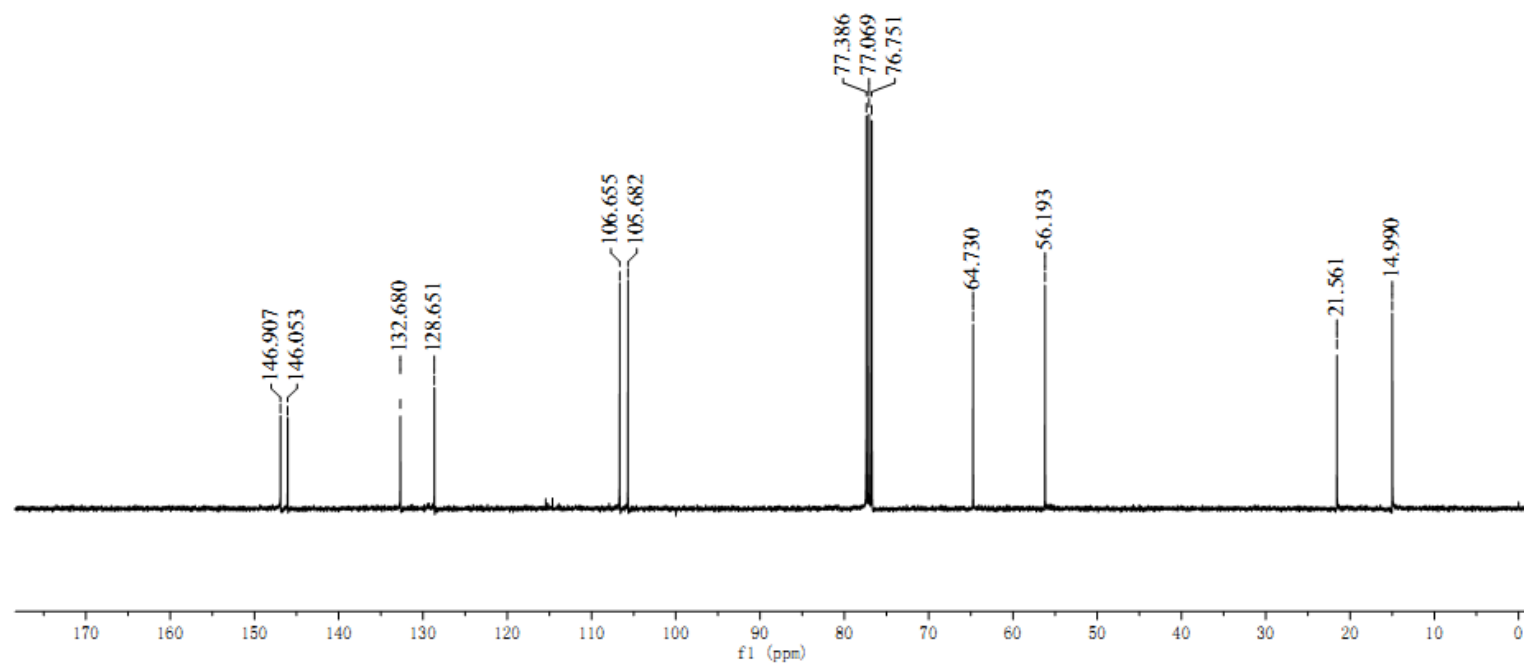
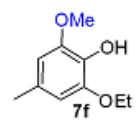


Minimum:

Maximum: 30.0 50.0 -1.5 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
153.0893	153.0916	-2.3	-15.0	3.5	27.2	0.0	C9 H13 O2





Elemental Composition Report

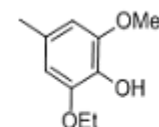
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

24 formula(e) evaluated with 5 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-37 H: 0-60 O: 0-9

YF-JI

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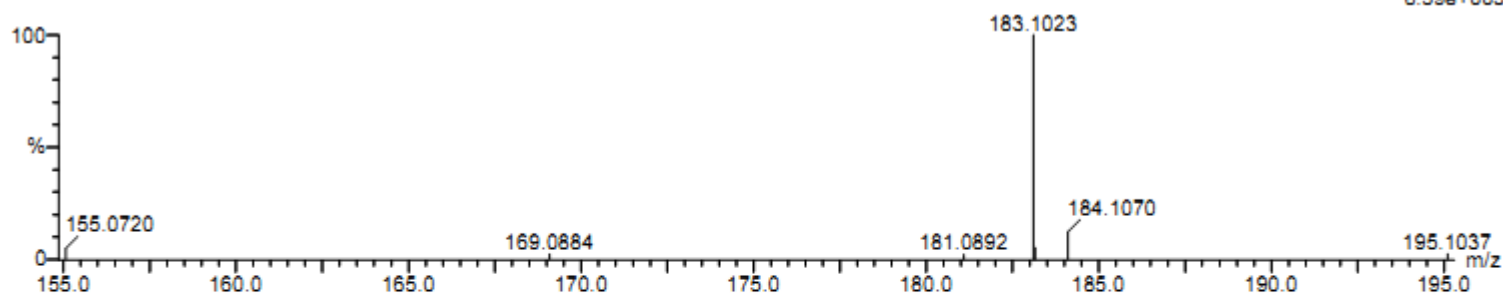
15-Nov-2013

13:47:19

1: TOF MS ES+

8.59e+003

JYF-JA-6 24 (0.815) Cm (14:25)



Minimum:

Maximum:

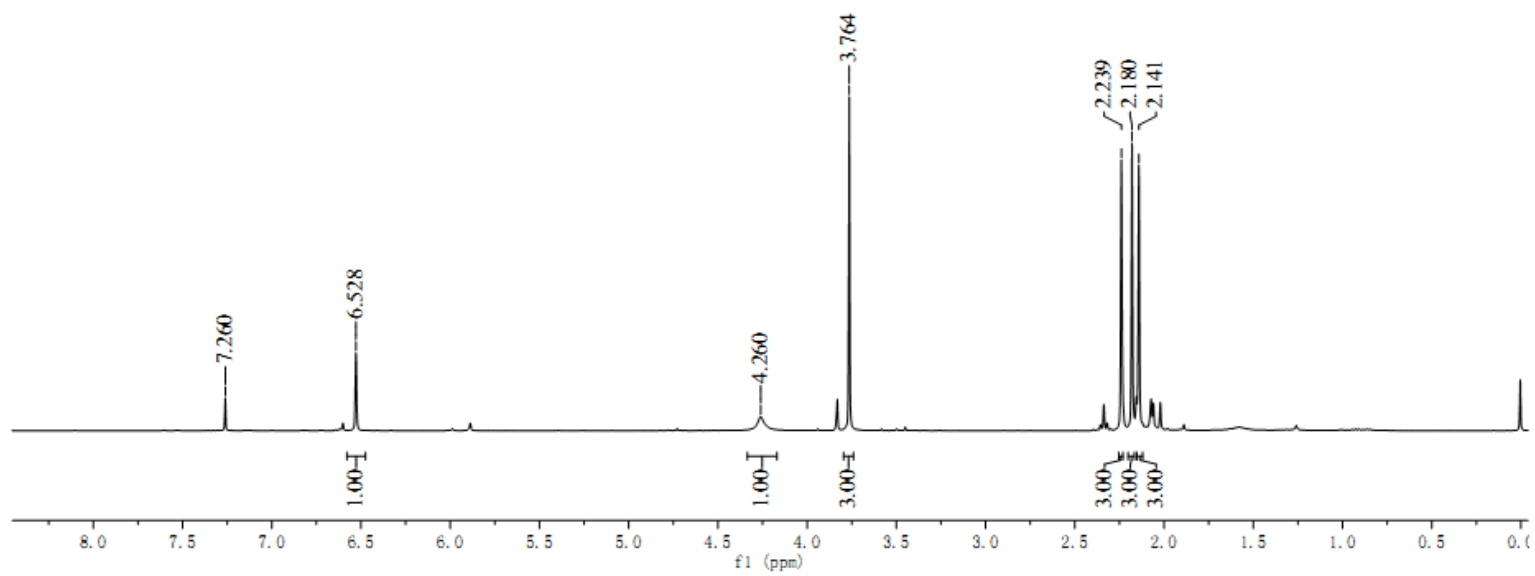
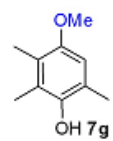
-1.5

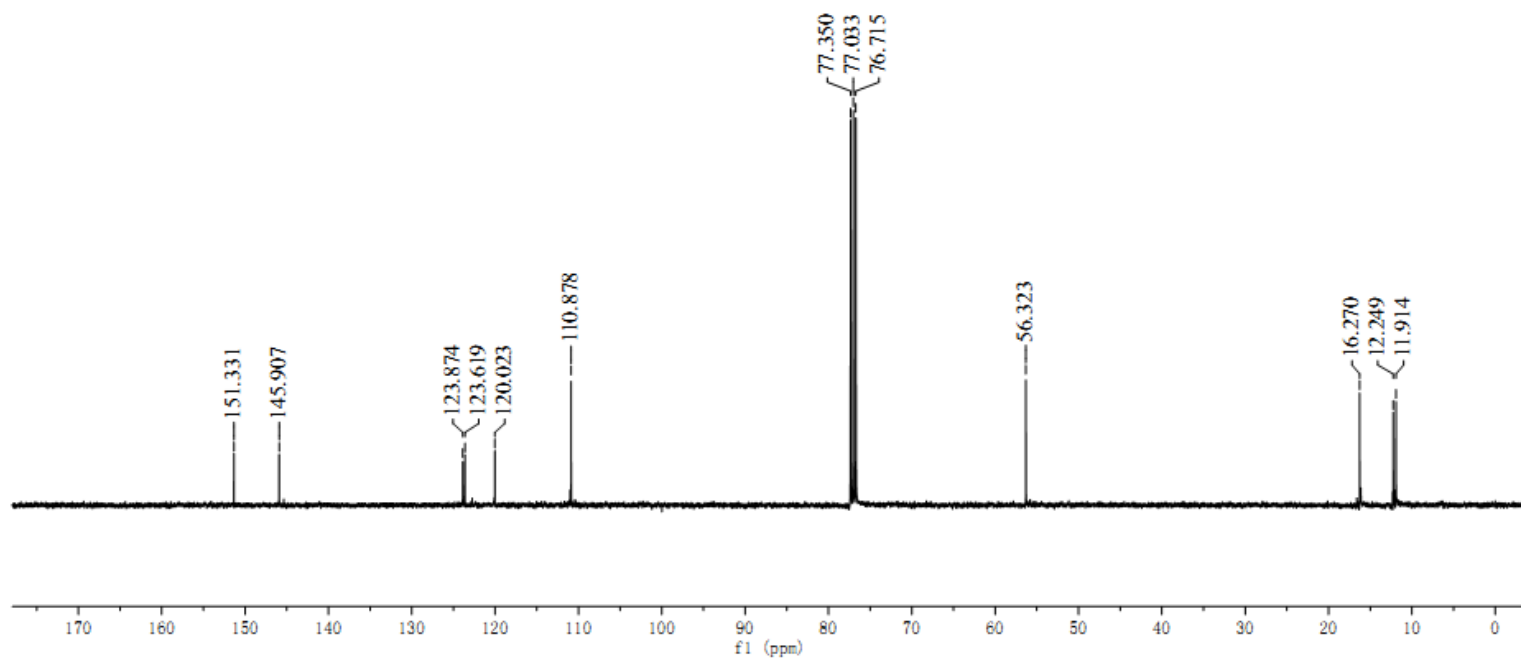
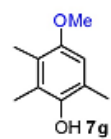
30.0

50.0

100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
183.1023	183.1021	0.2	1.1	3.5	21.1	0.0	C10 H15 O3





Elemental Composition Report

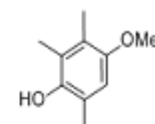
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

23 formula(e) evaluated with 4 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-37 H: 0-60 O: 0-9

YF-JI

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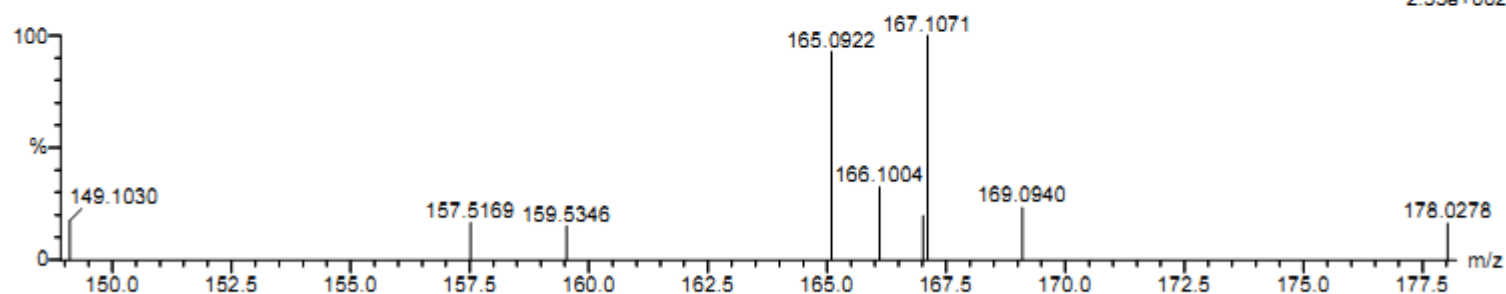
15-Nov-2013

13:58:39

1: TOF MS ES+

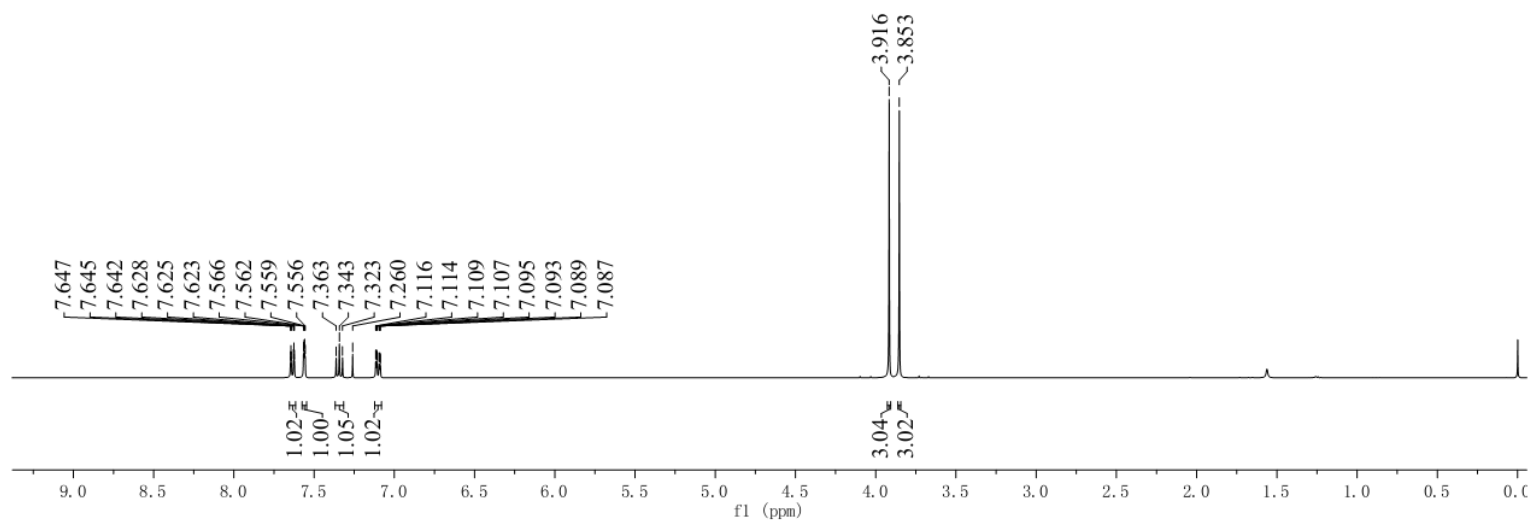
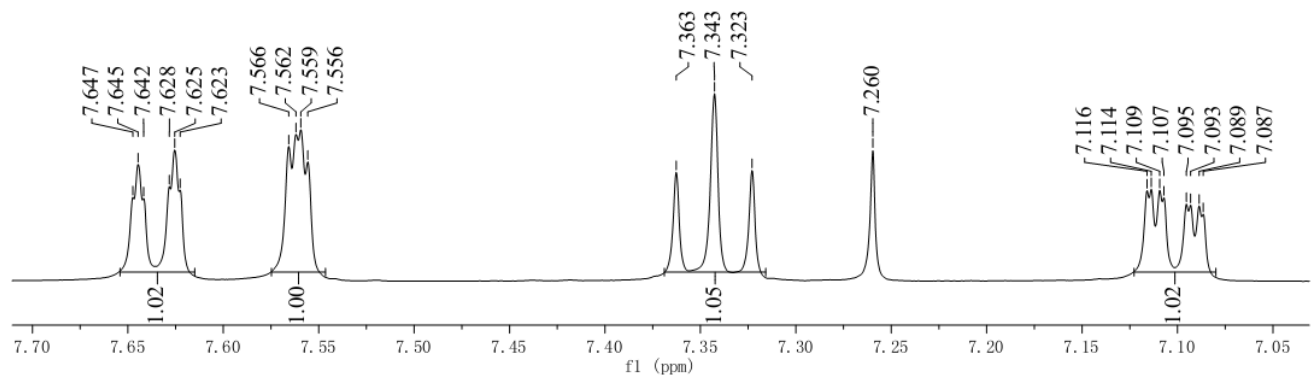
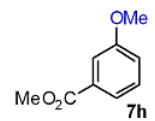
2.55e+002

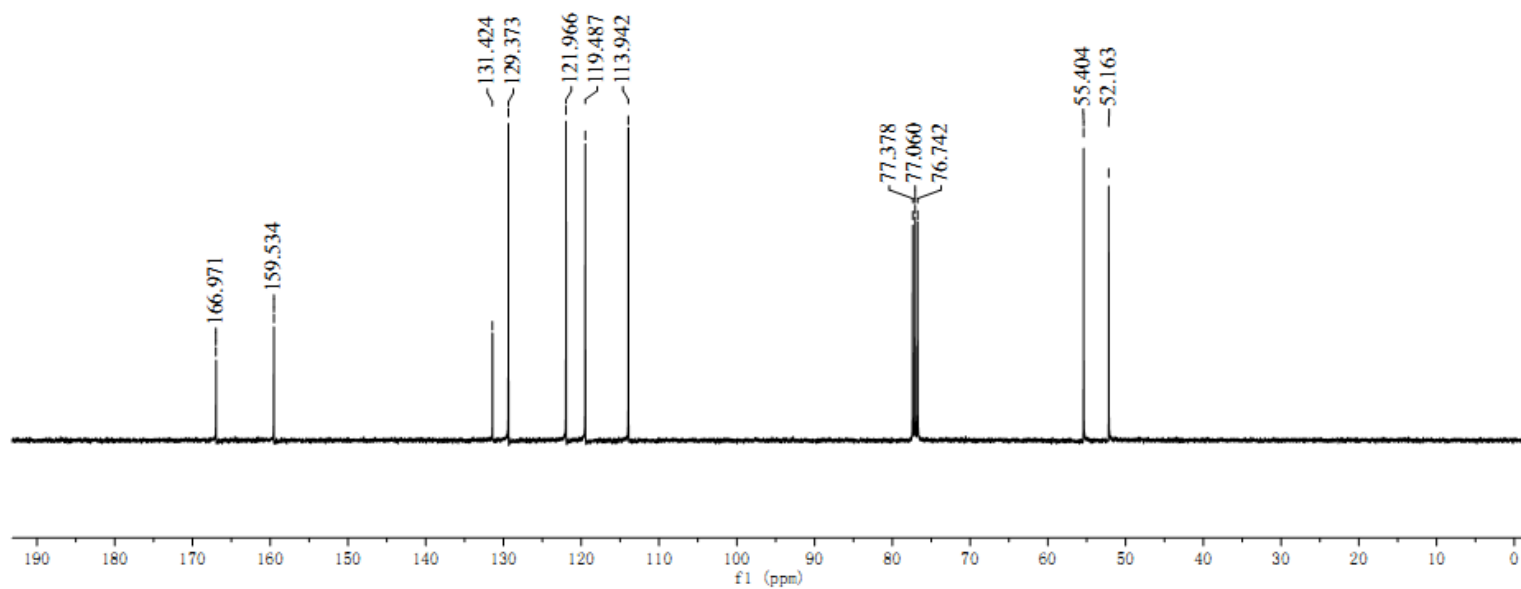
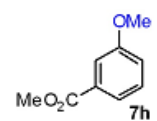
JYF-JA-8 27 (0.913) Cm (22:28)



Minimum: -1.5
Maximum: 30.0 50.0 100.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
167.1071	167.1072	-0.1	-0.6	3.5	15.4	0.0	C10 H15 O2





Elemental Composition Report

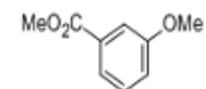
Page 1

Single Mass Analysis

Tolerance = 30.0 mDa / DBE: min = -1.5, max = 100.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 2



Monoisotopic Mass, Even Electron Ions

23 formula(e) evaluated with 5 results within limits (up to 1 closest results for each mass)

Elements Used:

C: 0-37 H: 0-60 O: 0-9

YF-JI

ECUST institute of Fine Chem

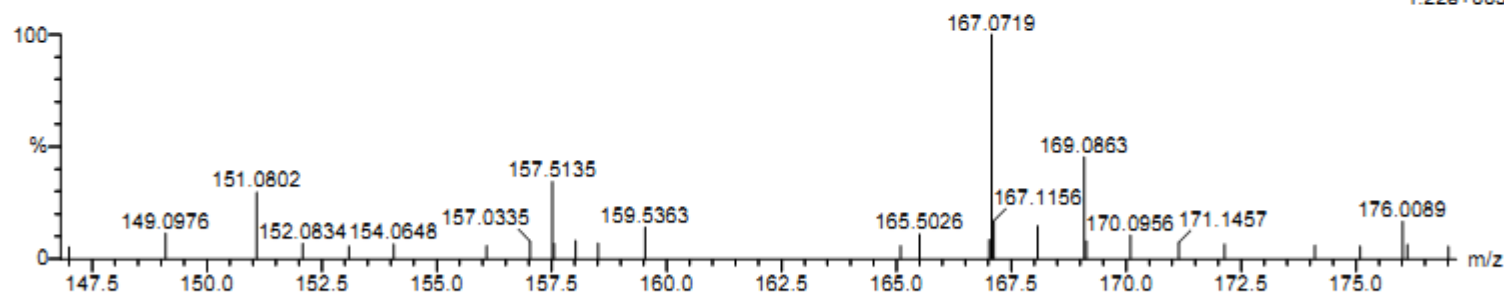
15-Nov-2013

14:10:21

1: TOF MS ES+

1.22e+003

JYF-JA-11 2 (0.152) Cm (1:22)



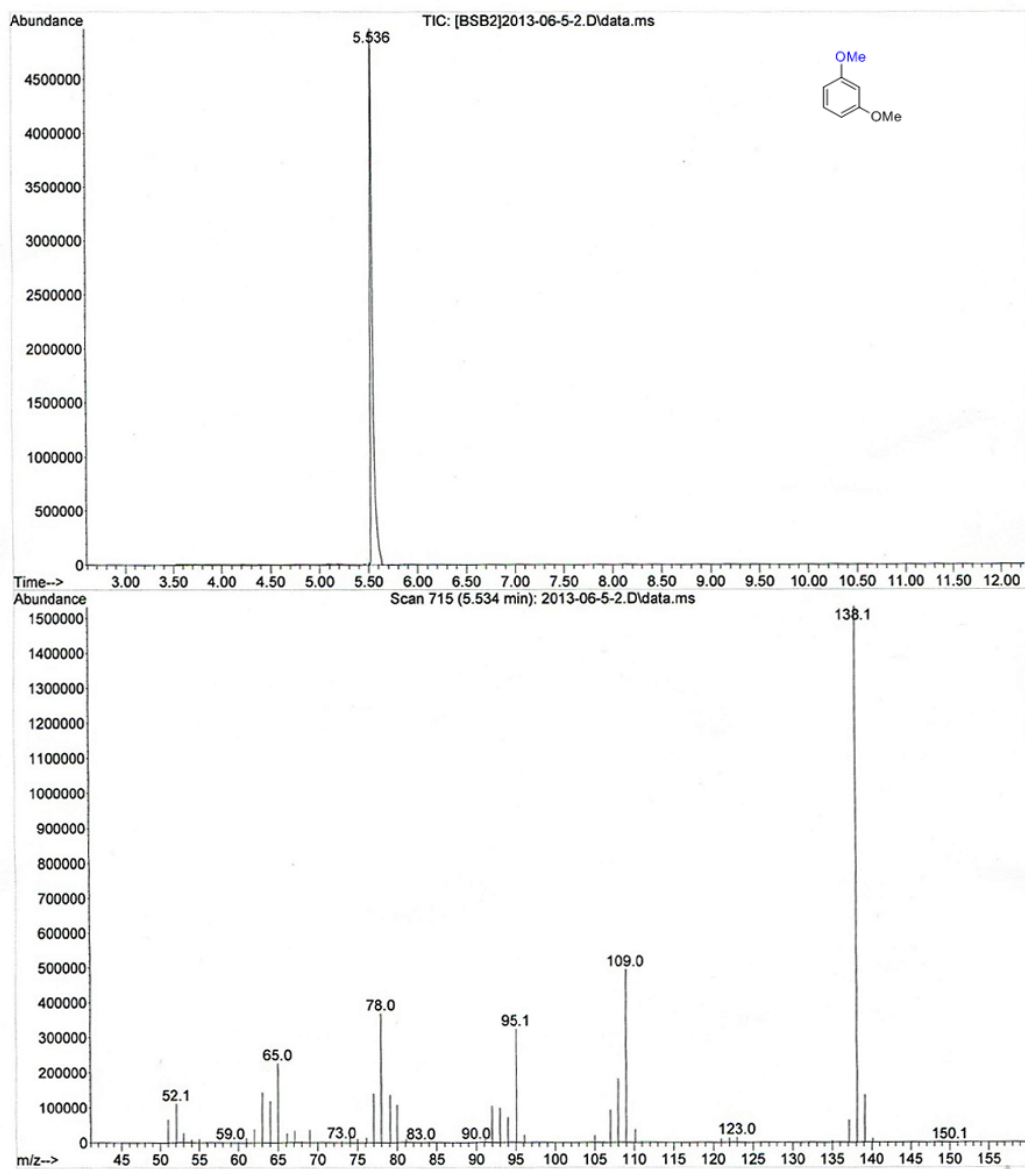
Minimum:

Maximum: 30.0 50.0 -1.5 100.0

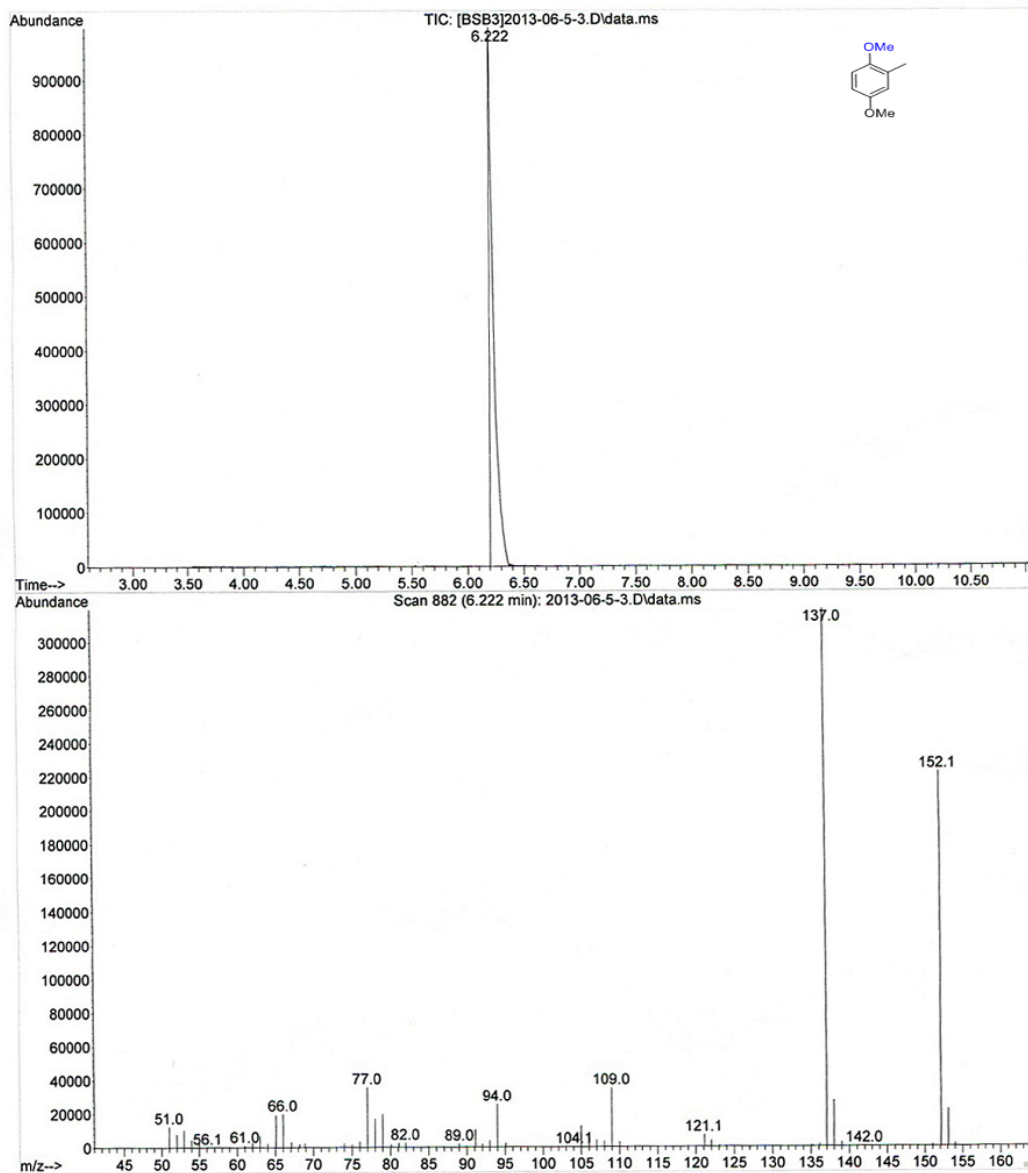
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	i-FIT (Norm)	Formula
167.0719	167.0708	1.1	6.6	4.5	19.2	0.0	C9 H11 O3

4.9 Copies of GC-MS for CuCl/HCOOMe-catalyzed methoxylation of aryl bromides.

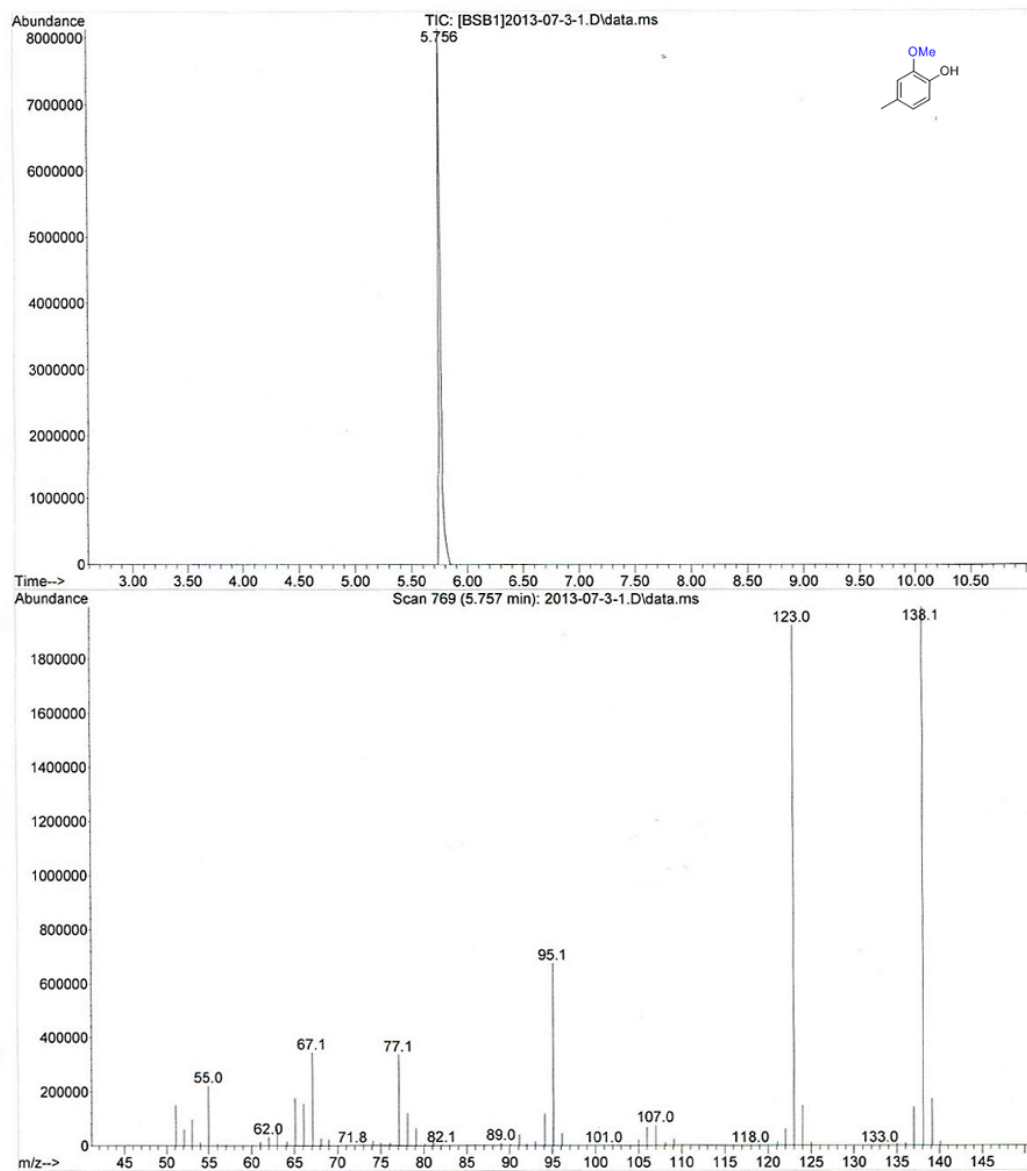
file :D:\data\2013-06-5\BSB\2013-06-5-2.D
operator : [BSB2]
acquired : 5 Jun 2013 13:53 using AcqMethod METHOD-01.M
instrument : GC-MSD
sample Name: 2013-06-5-8
disc Info :
trial Number: 18



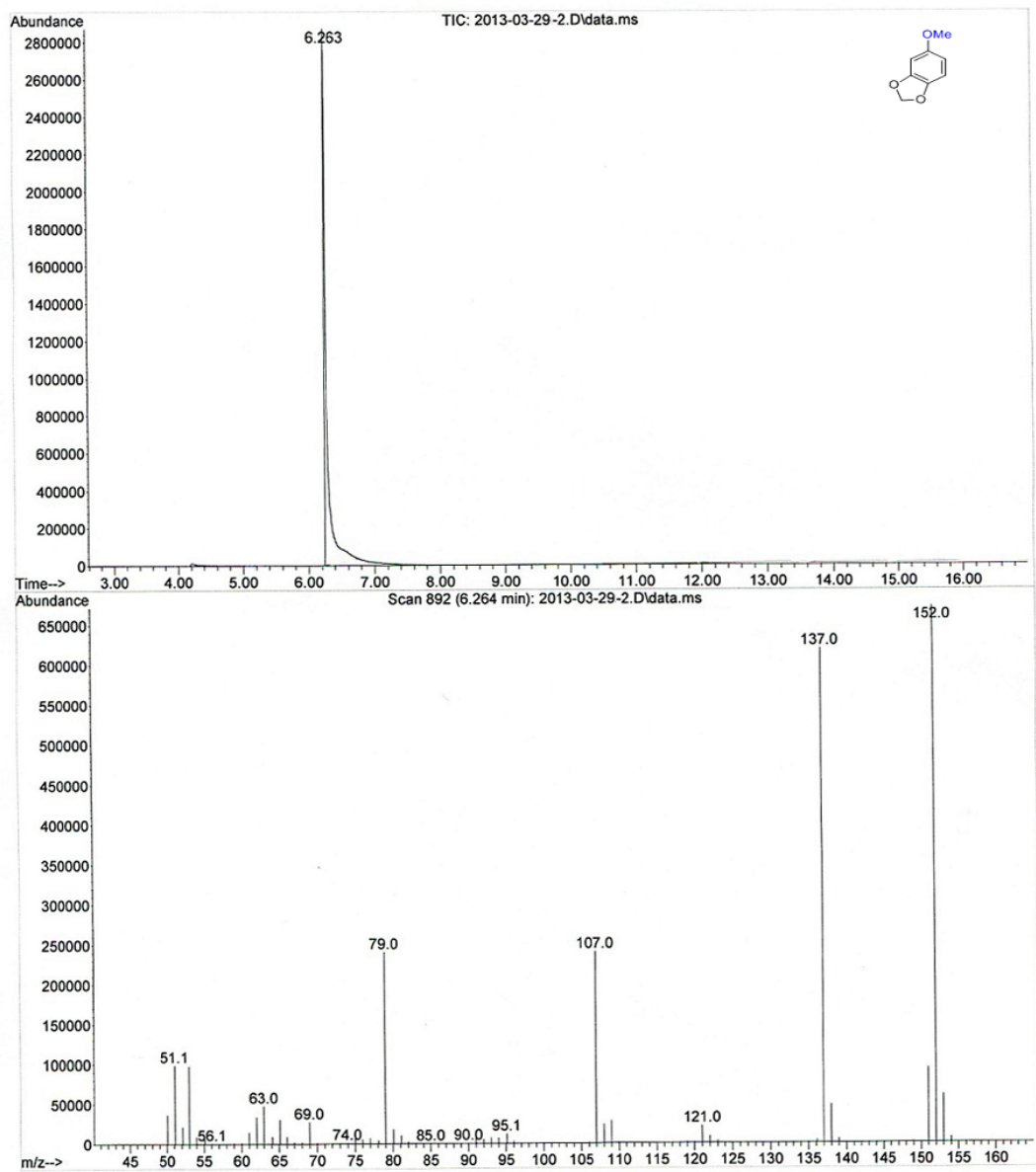
File :D:\data\2013-06-5\BSB\2013-06-5-3.D
Operator : [BSB3]
Acquired : 5 Jun 2013 14:09 using AcqMethod METHOD-01.M
Instrument : GC-MSD
Sample Name: 2013-06-5-9
Disc Info :
Serial Number: 19



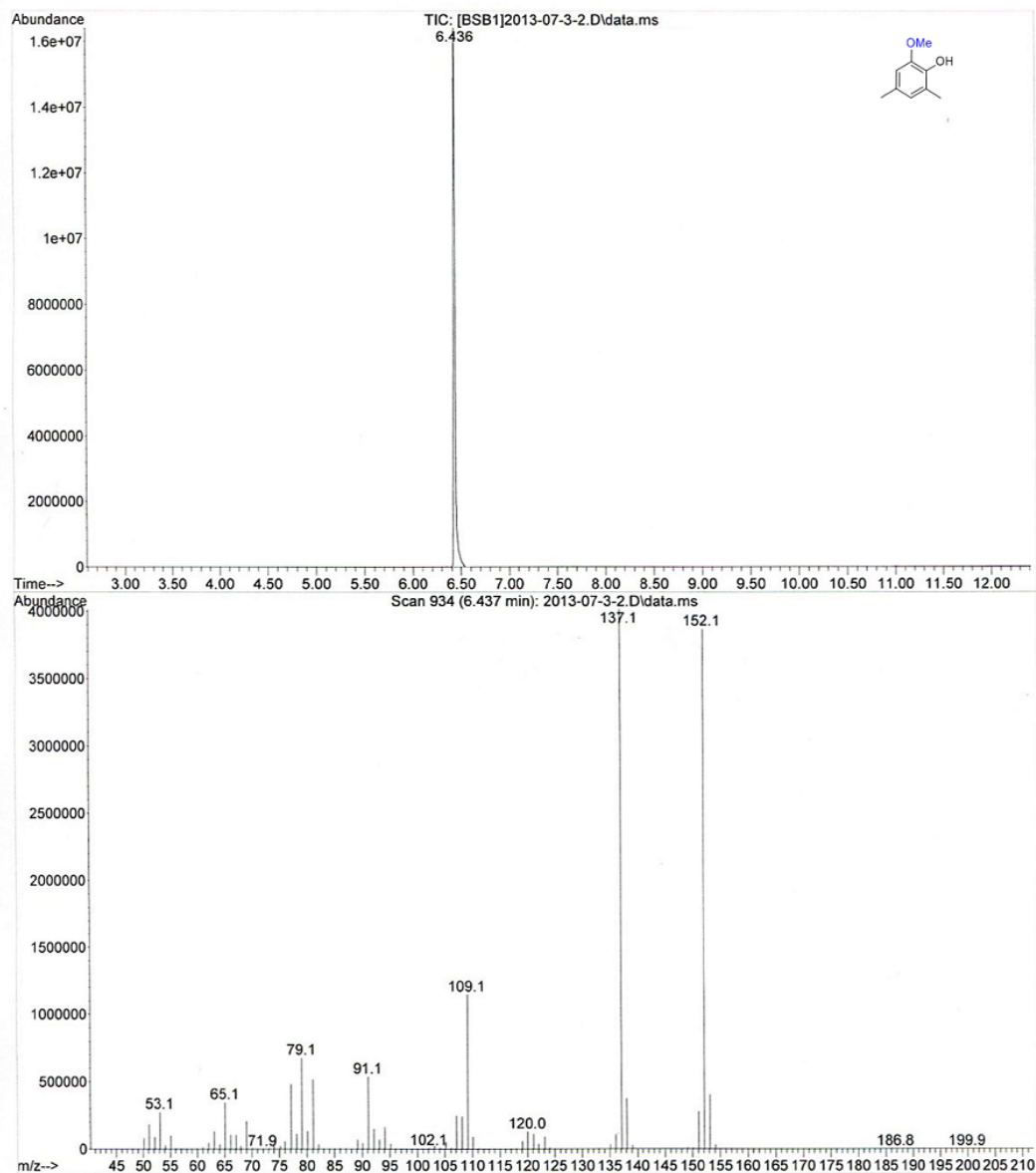
File :D:\data\2013-07-3\BSB\2013-07-3-1.D
Operator : [BSB1]
Acquired : 3 Jul 2013 14:10 using AcqMethod METHOD-01.M
Instrument : GC-MSD
Sample Name: 2013-07-3-8
Misc Info :
Vial Number: 15



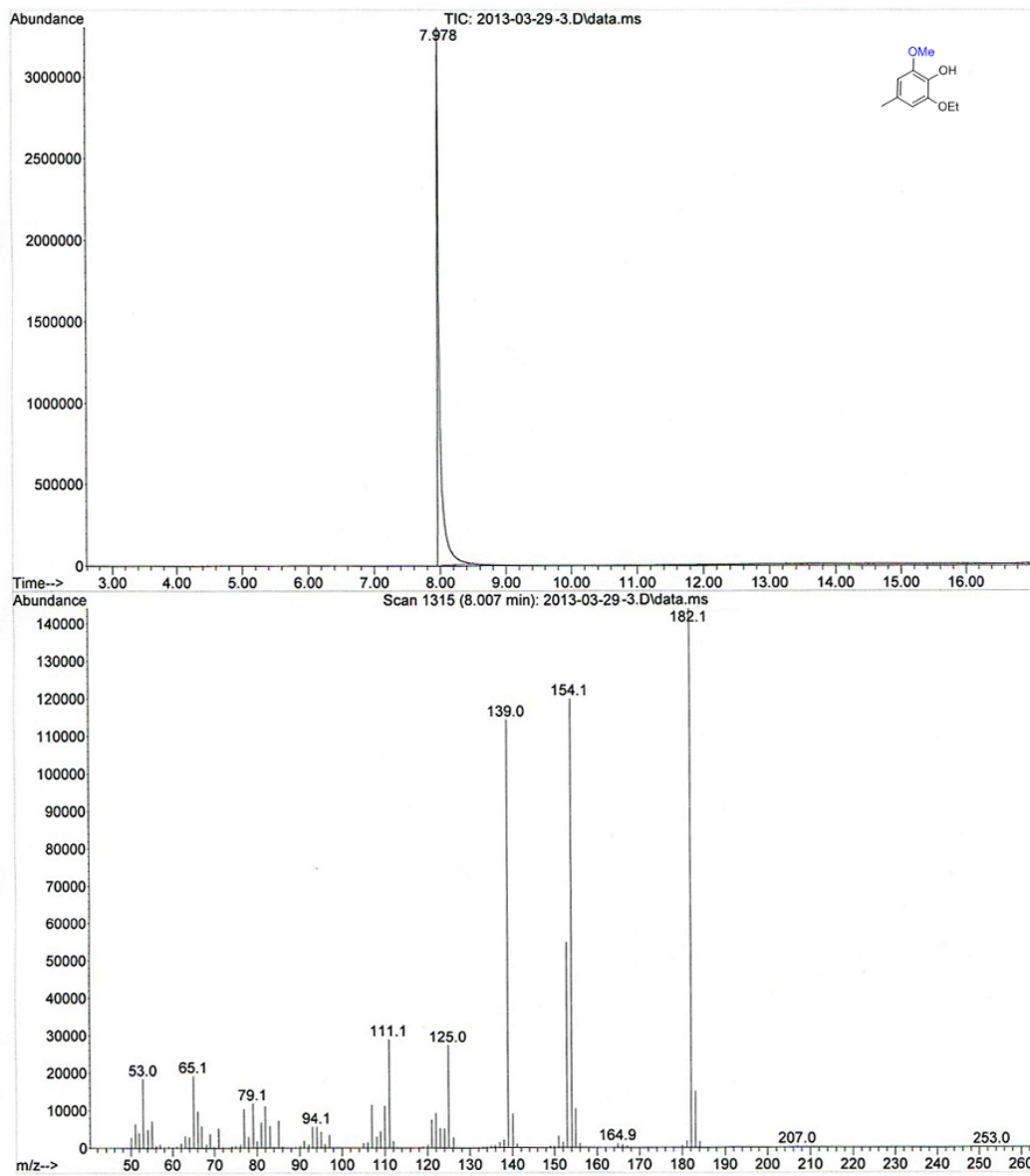
File :D:\data\2013-03-29\2013-03-29-2.D
Operator :
Acquired : 29 Mar 2013 13:46 using AcqMethod METHOD-01.M
Instrument : GC-MSD
Sample Name: 2013-03-29-5
Misc Info :
Vial Number: 15



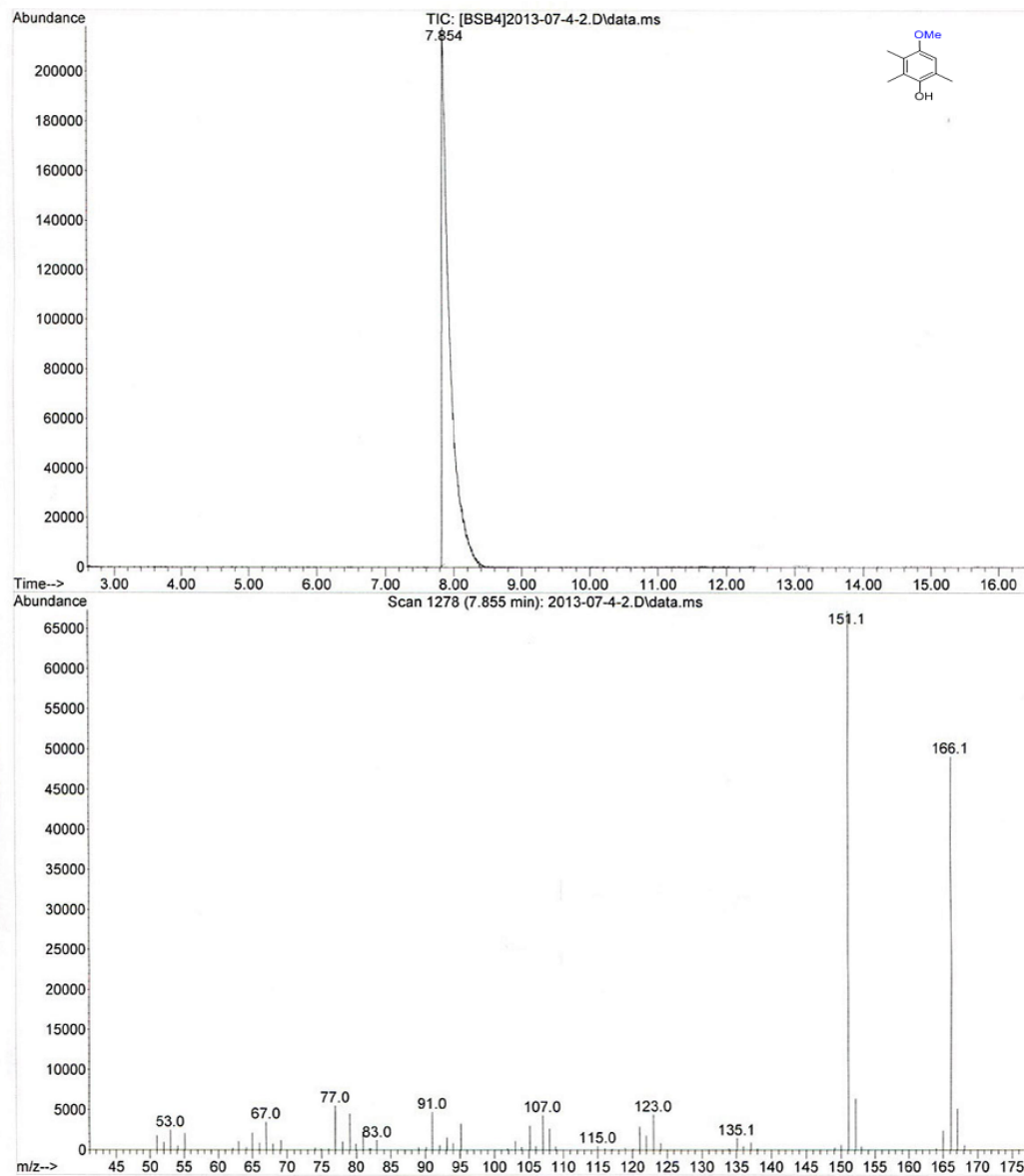
.le :D:\data\2013-07-3\BSB\2013-07-3-2.D
perator : [BSB1]
quired : 3 Jul 2013 14:24 using AcqMethod METHOD-01.M
strument : GC-MSD
mple Name: 2013-07-3-9
sc Info :
ial Number: 16



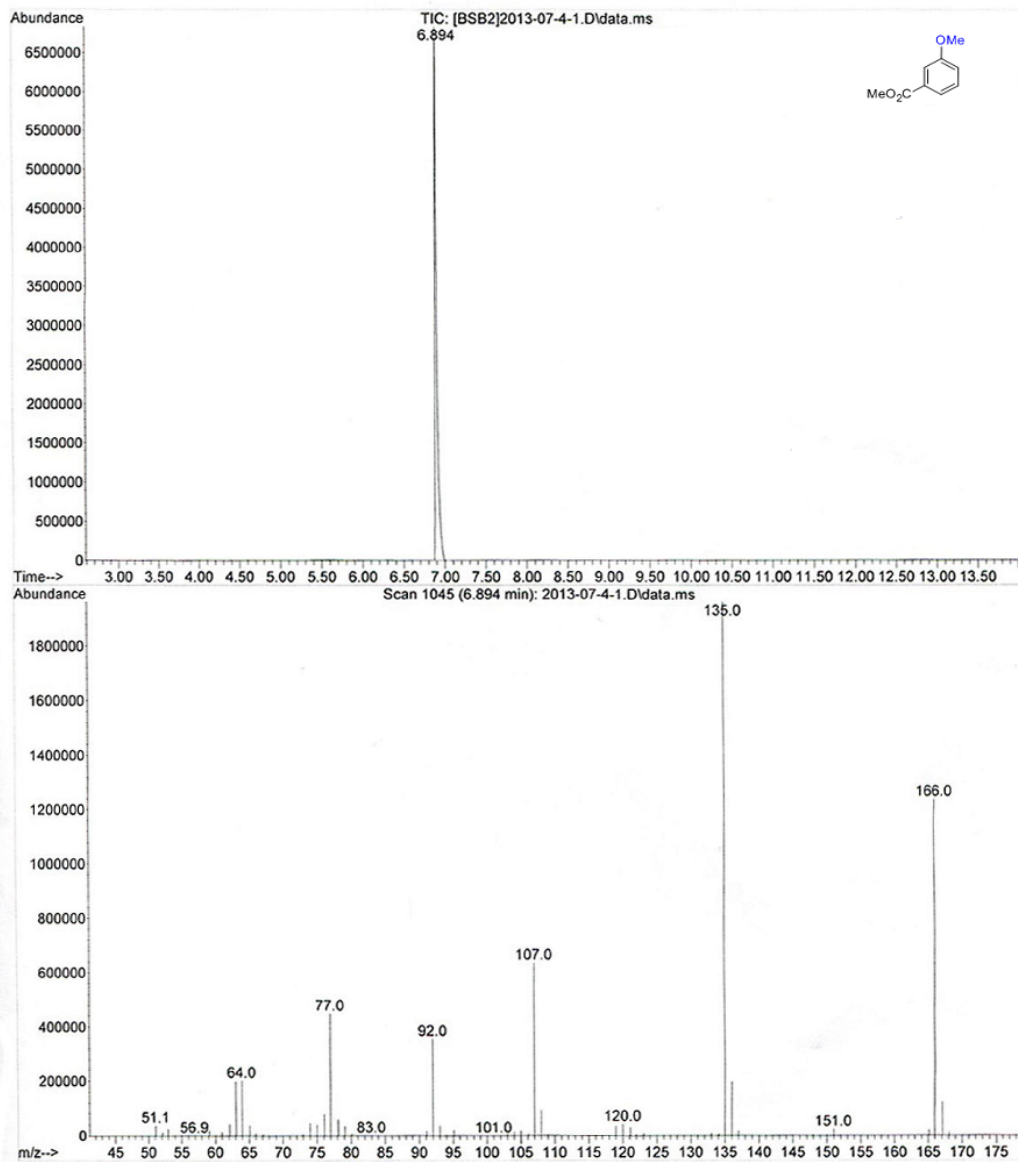
le :D:\data\2013-03-29\2013-03-29-3.D
erator :
quired : 29 Mar 2013 14:06 using AcqMethod METHOD-01.M
strument : GC-MSD
mple Name: 2013-03-29-6
sc Info :
al Number: 16



file :D:\data\2013-07-4\BSB\2013-07-4-2.D
operator : [BSB4]
acquired : 4 Jul 2013 12:00 using AcqMethod METHOD-01.M
instrument : GC-MSD
sample Name: 2013-07-4-2
disc Info :
serial Number: 11

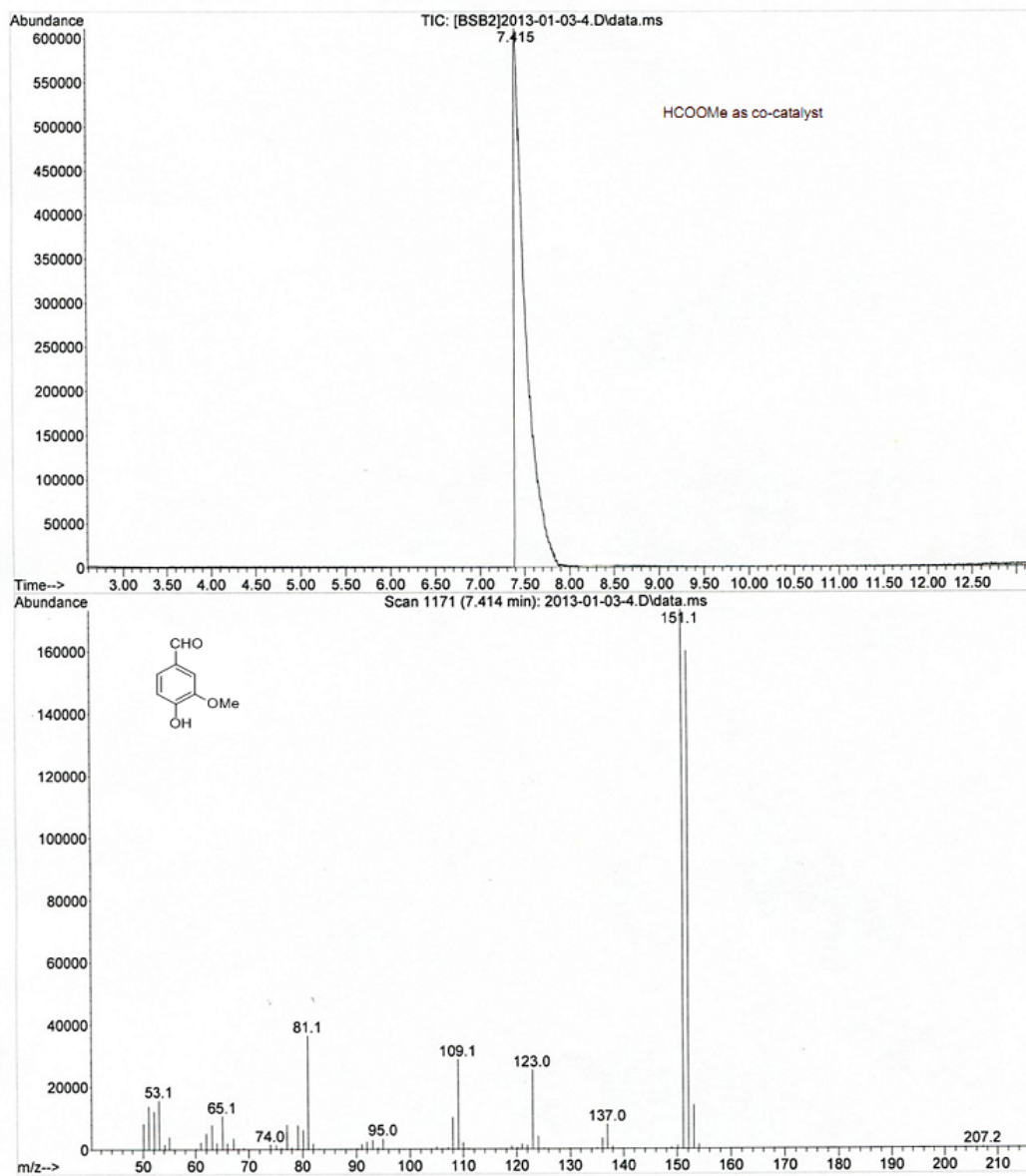


file :D:\data\2013-07-4\BSB\2013-07-4-1.D
operator : [BSB2]
acquired : 4 Jul 2013 11:42 using AcqMethod METHOD-01.M
instrument : GC-MSD
sample Name: 2013-07-4-1
disc Info :
vial Number: 10



4.10 Copies of GC-MS for CuCl-catalyzed methoxylation of 2a with different co-catalysts.

File :D:\data\2013-01-03\BSB\2013-01-03-4.D
Operator : [BSB2]
Acquired : 3 Jan 2013 12:25 using AcqMethod METHOD-01.M
Instrument : GC-MSD
Sample Name: 2013-01-03-2
Misc Info :
Vial Number: 12

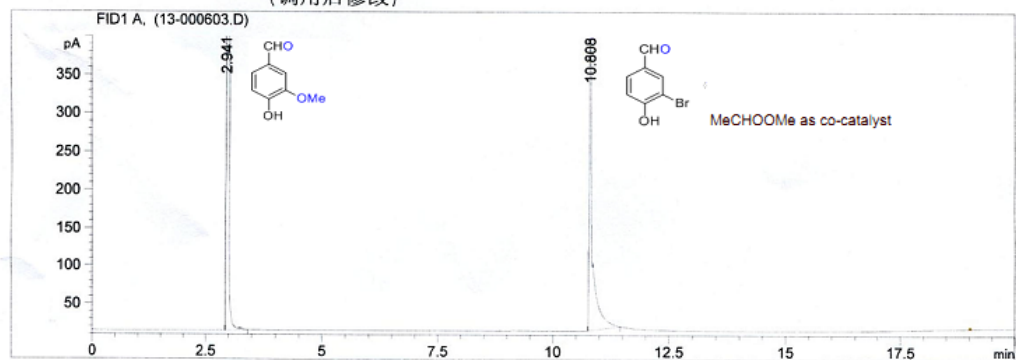


数据文件: C:\CHEM32\1\DATA\13-000603.D
样品名称: JYF-CC-2

=====

操作者	:		位置	:	样品瓶 101
仪器	:	仪器 1	进样次数	:	1
进样日期	:	05-Dec-13, 14:51:52	进样量	:	手动
采集方法	:	C:\CHEM32\1\METHODS\JYF.M			
最后修改	:	2013-12-5 14:48:00			
分析方法	:	C:\CHEM32\1\METHODS\气相条件.M			
最后修改	:	2013-12-5 15:14:51			

(调用后修改)



=====
面积百分比报告
=====

排序 : 信号
乘积因子: : 1.0000
稀释因子: : 1.0000
内标使用乘积因子和稀释因子

信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	2.941	BB S	0.0286	4.14442e4	1.96233e4	95.15157
2	10.808	BB S	0.0449	2111.78394	649.31158	4.84843

总量 : 4.35560e4 2.02726e4

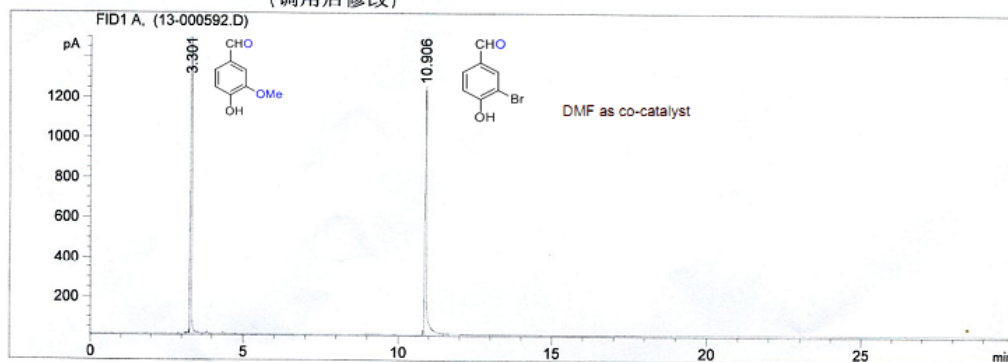
=====
*** 报告结束 ***

数据文件: C:\CHEM32\1\DATA\13-000592.D
样品名称: JFY-CC-1

=====

操作者	:		位置	: 样品瓶 101
仪器	:	仪器 1	进样次数	: 1
进样日期	:	02-Dec-13, 12:45:53	进样量	: 手动
采集方法	:	C:\CHEM32\1\METHODS\气相条件.M		
最后修改	:	2013-11-20 17:09:12		
分析方法	:	C:\CHEM32\1\METHODS\气相条件.M		
最后修改	:	2013-12-5 10:29:52		

(调用后修改)



=====
面积百分比报告
=====

排序 : 信号
乘积因子: : 1.0000
稀释因子: : 1.0000
内标使用乘积因子和稀释因子

信号 1: FID1 A,

峰 #	保留时间 [min]	类型	峰宽 [min]	峰面积 [pA*s]	峰高 [pA]	峰面积 %
1	3.301	VB S	0.0279	6.82191e4	4.10916e4	93.97011
2	10.906	BB	0.0512	4377.49268	1240.88257	6.02989

总量 : 7.25966e4 4.23324e4

=====
*** 报告结束 ***