

Supporting Information for:

## **Ruthenium Complexes of Tetradentate Bipyridine Ligands: Highly Efficient Catalysts for the Hydrogenation of Carboxylic Esters and Lactones**

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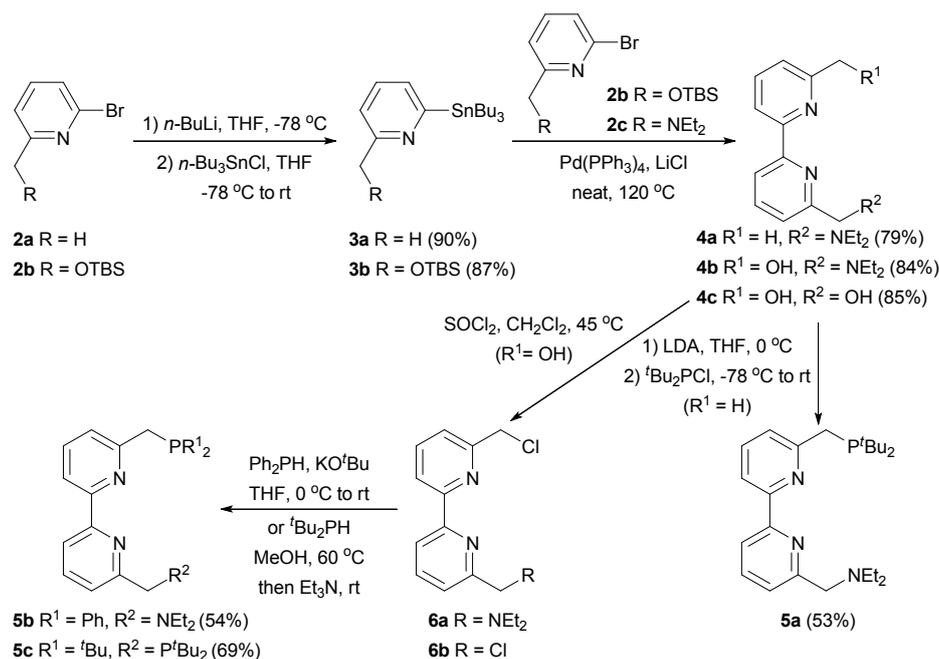
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**General:** Unless mentioned otherwise, all manipulations were performed in an argon-filled glove box MBRAUN LABstar or using standard Schlenk techniques. NMR spectra were recorded on a Bruker AV 400 spectrometer at 400 MHz (<sup>1</sup>H NMR), 101 MHz (<sup>13</sup>C NMR) and 162 MHz (<sup>31</sup>P NMR). Chemical shifts were reported in ppm relative to internal TMS for <sup>1</sup>H NMR data, deuterated solvent for <sup>13</sup>C NMR data, and external 85% H<sub>3</sub>PO<sub>4</sub> for <sup>31</sup>P NMR data, respectively. Data are presented in the following space: chemical shift, multiplicity, coupling constant in hertz (Hz), and signal area integration in natural numbers. High-resolution mass spectra were recorded on an IonSpec FT-ICR mass spectrometer with ESI resource. Elemental analysis was performed on Elementar Vario EL elemental analyzer. GC analysis were performed using a Agilent Technologies 7890A GC System.

Hydrogen gas (99.999%) was purchased from Boc Gas Inc., Tianjin. All the solvents used for reactions were distilled under argon after drying over an appropriate drying agent. [RuCl<sub>2</sub>(η<sup>6</sup>-p-cymene)]<sub>2</sub> was purchased from Strem Chemicals, Inc. Other commercially available reagents were purchased from Acros, Aldrich and Alfa Aesar Chemical Company. Pd(PPh<sub>3</sub>)<sub>4</sub> was prepared according to the literature procedure and kept in a refrigerator under argon.<sup>1</sup> HPPPh<sub>2</sub> and HP<sup>t</sup>Bu<sub>2</sub> were synthesized according to the literature procedure.<sup>2</sup> All of the liquid substrates for hydrogenation were distilled before use. All of the solid substrates except for long chain fatty esters and triglycerides for hydrogenation were recrystallized before use.

## (A) Preparation and Analytical Data of Ligands 5



### 1. Preparation and Analytical Data of Ligand 5a

#### 2-Bromo-6-diethylamminomethylpyridine (2c):

##### (1) 2-Bromo-6-(((4-methylphenyl)sulfonyl)oxy)methylpyridine

To a 500 mL round-bottom flask were added 2-bromo-6-hydroxymethylpyridine (11.2 g, 59.6 mmol) and THF (200 mL). A solution of NaOH (7.2 g, 180 mmol) in H<sub>2</sub>O (50 mL) was added in one portion with stirring and the mixture was cooled to 0 °C. To the mixture, a solution of TsCl (12.6 g, 66.1 mmol) in THF (50 mL) was added dropwise over 30 min. The mixture was allowed to slowly warm up to room temperature and stirred overnight. The solvent was removed under vacuum and the residue was diluted with CH<sub>2</sub>Cl<sub>2</sub> (300 mL) and washed with water and brine. The CH<sub>2</sub>Cl<sub>2</sub> solution was dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated, yielding crude title compound as a pale-yellow solid. It was pure enough for the next step. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.81 (d, *J* = 8.3 Hz, 2H), 7.56 (d, *J* = 8.0 Hz, 1H), 7.40 (d, *J* = 7.8 Hz, 2H), 7.34 (d, *J* = 8.1 Hz, 2H), 5.09 (s, 2H), 2.44 (s, 3H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 155.18 (s), 145.38 (s), 141.47 (s), 139.36 (s), 132.48 (s), 130.06 (s), 128.12 (s), 127.82 (s), 120.61 (s), 70.73 (s), 21.73 (s). HRMS (ESI) calcd for C<sub>13</sub>H<sub>13</sub>BrNO<sub>3</sub>S<sup>+</sup> ([M+H]<sup>+</sup>): 341.9794; Found: 341.9789.

##### (2) 2-Bromo-6-diethylamminomethylpyridine (2c)

To a 250 mL 2-neck round-bottom flask were added the above compound, diethylamine (43.9 g, 600 mmol) and THF (120 mL). The mixture was stirred at room temperature overnight. After removal of the solvent, the residue was diluted with CH<sub>2</sub>Cl<sub>2</sub> (300 mL), washed with saturated NaHCO<sub>3</sub>, water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo. The residue was distilled under vacuum, yielding 2-bromo-6-diethylamminomethylpyridine as a pale yellow oil (14.0 g, 96% for two steps). B.p.: 56 °C/0.1 mmHg. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.50 (d, *J* = 4.9 Hz, 2H), 7.32 (t, *J* = 4.8 Hz, 1H), 3.70 (s, 2H), 2.56 (q, *J* = 7.1 Hz, 4H), 1.03 (t, *J* = 7.1 Hz, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 163.09 (s), 141.04 (s), 138.73 (s), 125.85 (s), 121.37 (s),

59.09 (s), 47.51 (s), 12.08 (s). HRMS (ESI) calcd for  $C_{10}H_{16}BrN_2^+$  ( $[M+H]^+$ ): 243.0491; Found: 243.0487.

### **2-Tri-*n*-butylstannyl-6-methylpyridine (3a)<sup>3</sup>:**

To an oven-dried 500 mL 2-neck round-bottom flask under an argon atmosphere were added 2-bromo-6-methylpyridine (20.0 g, 116.3 mmol) and dry THF (200 mL). The solution was cooled to -78 °C and a solution of *n*-BuLi in hexane (2.4 M, 56 mL, 134.4 mmol) was added with a syringe during 30 min. The deep reddish brown solution was stirred at -78 °C for 2 hrs and tributyltin chloride (45.4 g, 139.5 mmol) was added dropwise with a syringe during 30 min. The mixture was stirred at -78 °C for 30 min then was slowly warm up to room temperature and stirred overnight. Thereafter, the mixture was evaporated to dryness and the residue was diluted with Et<sub>2</sub>O (300 mL), washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo. The residue was distilled under vacuum, yielding the title compound as a colorless oil (40.0 g, 90%). B.p.: 145 °C/0.4 mmHg. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.36 (t, *J* = 7.5 Hz, 1H), 7.18 (d, *J* = 7.4 Hz, 1H), 6.95 (d, *J* = 7.6 Hz, 1H), 2.54 (s, 3H), 1.69 – 0.76 (m, 27H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 173.11 (s), 158.70 (s), 133.40 (s), 129.48 (s), 121.62 (s), 29.22 (s), 27.47 (s), 24.98 (s), 13.82 (s), 9.95 (s). HRMS (ESI) calcd for  $C_{18}H_{34}NSn^+$  ( $[M+H]^+$ ): 384.1708; Found: 384.1712.

### **6-Diethylamminomethyl-6'-methyl-2,2'-bipyridine (4a):**

To an oven-dried 100 mL Schlenk flask were added Pd(PPh<sub>3</sub>)<sub>4</sub> (1.28 g, 1.1 mmol), anhydrous LiCl (4.1 g, 96.7 mmol), 2-tri-*n*-butylstannyl-6-methylpyridine (14.5 g, 37.9 mmol) and 2-bromo-6-diethylamminomethylpyridine (7.7 g, 31.7 mmol). The mixture was degassed, followed by the introduction of highly pure argon and then heated at 120 °C for 16 hrs with stirring, at which time a black precipitate was formed. GC analysis showed a full conversion of 2-bromo-6-diethylamminomethylpyridine. After cooling to room temperature, the mixture was diluted with ethyl acetate (100 mL) and filtered to remove unsolved material and then acidified with 6N HCl (aq). The organic layer was washed with water and the combined aqueous layer was made alkaline with 6N NaOH (aq) and extracted with ethyl ether (3 × 100 mL). The combined organic layer was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo. The residue was distilled under vacuum, yielding the title compound as a pale yellow oil (6.4 g, 79%). B.p.: 130 °C/0.1 mmHg. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.24 (d, *J* = 7.8 Hz, 1H), 8.19 (d, *J* = 7.8 Hz, 1H), 7.75 (t, *J* = 7.7 Hz, 1H), 7.68 (t, *J* = 7.7 Hz, 1H), 7.48 (d, *J* = 7.6 Hz, 1H), 7.14 (d, *J* = 7.6 Hz, 1H), 3.83 (s, 2H), 2.65 – 2.58 (m, 7H), 1.09 (t, *J* = 7.1 Hz, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 160.10 (s), 157.87 (s), 156.07 (s), 155.64 (s), 137.12 (s), 137.05 (s), 123.09 (s), 122.73 (s), 119.15 (s), 118.27 (s), 59.53 (s), 47.51 (s), 24.77 (s), 12.18 (s). HRMS (ESI) calcd for  $C_{16}H_{22}N_3^+$  ( $[M+H]^+$ ): 256.1808; Found: 256.1807.

### **6-Di-*tert*-butylphosphinomethyl-6'-diethylamminomethyl-2,2'-bipyridine (5a):**

To an oven-dried 100 mL Schlenk flask was added a degassed solution of 6-diethylamminomethyl-6'-methyl-2,2'-bipyridine (4.0 g, 15.7 mmol) in dry THF (10 mL) under an argon atmosphere. The solution was cooled to 0 °C and a solution of freshly prepared LDA (20 mL, 1.2 M) in THF was added with a syringe during 30 min. The mixture was stirred for 2 hrs at 0 °C and then cooled to -78 °C, and a degassed solution of di-*tert*-butylchlorophosphine (5.7 g, 31.5 mmol) in dry THF (10 mL) was added dropwise with a syringe during 30 min. The mixture was

allowed slowly warm up to room temperature and stirred overnight. The solvent was removed under vacuum. To the residue was added degassed water (30 mL) and degassed ethyl ether (30 mL). The ether phase was separated under an argon atmosphere. The aqueous phase was extracted with degassed ethyl ether ( $2 \times 10$  mL). The combined ether solutions were dried over  $\text{Na}_2\text{SO}_4$ , filtered, and the solvent was removed under vacuum. The residue was distilled under vacuum, yielding the title compound as a viscous pale-yellow oil (3.3 g, 53%), which was gradually solidified upon standing at room temperature in the glove box. B.p.: 180 °C /0.1 mmHg.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  37.12 (s).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.26 (d,  $J = 7.7$  Hz, 1H), 8.18 (d,  $J = 7.7$  Hz, 1H), 7.74 (t,  $J = 7.7$  Hz, 1H), 7.68 (t,  $J = 7.7$  Hz, 1H), 7.46 (d,  $J = 7.7$  Hz, 1H), 7.40 (d,  $J = 7.7$  Hz, 1H), 3.82 (s, 2H), 3.14 (d,  $J = 2.4$  Hz, 2H), 2.62 (q,  $J = 7.0$  Hz, 4H), 1.18 (d,  $J = 11.0$  Hz, 18H), 1.09 (t,  $J = 7.0$  Hz, 6H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  161.46 (d,  $J = 13.7$  Hz), 159.88 (s), 155.70 (s), 155.49 (s), 137.16 (s), 137.00 (s), 123.82 (d,  $J = 8.5$  Hz), 122.75 (s), 119.14 (s), 117.99 (s), 59.46 (s), 47.53 (s), 32.09 (d,  $J = 21.6$  Hz), 31.97 (d,  $J = 23.9$  Hz), 29.86 (d,  $J = 13.2$  Hz), 12.18 (s). HRMS (ESI) calcd for  $\text{C}_{24}\text{H}_{39}\text{N}_3\text{P}^+([\text{M}+\text{H}]^+)$ : 400.2876; Found: 400.2867.

## 2. Preparation and Analytical Data of Ligand 5b

### 2-((*tert*-Butyldimethylsilyloxy)methyl)-6-(tributylstannyl)pyridine (3b):

#### (1) 2-Bromo-6-((*tert*-butyldimethylsilyloxy)methyl)pyridine (2b)<sup>4</sup>

To an oven-dried 100 mL Schlenk flask were added TBSCl (10.2 g, 67.7 mmol), 2-bromo-6-hydroxymethylpyridine (10.6 g, 56.4 mmol), imidazole (15.4 g, 226.1 mmol) and anhydrous DMF (50 mL) under an argon atmosphere. The reaction mixture was stirred at room temperature until TLC showed complete conversion of the starting alcohol (1 hr). Water (50 mL) was added and then the mixture was extracted with  $\text{Et}_2\text{O}$  (100 mL  $\times 3$ ). The combined ethereal phases were washed with water and brine, dried over  $\text{Na}_2\text{SO}_4$ , filtered and concentrated. The resulting yellow oily residue was distilled under vacuum, yielding the title compound as a colorless oil (15.3 g, 90%). B.p.: 92 °C/0.1 mmHg.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.56 (t,  $J = 7.7$  Hz, 1H), 7.47 (dd,  $J = 7.6, 0.8$  Hz, 1H), 7.33 (dd,  $J = 7.8, 0.7$  Hz, 1H), 4.80 (s, 2H), 0.95 (s, 9H), 0.11 (s, 6H).

#### (2) 2-((*tert*-Butyldimethylsilyloxy)methyl)-6-(tributylstannyl)pyridine (3b)

To an oven-dried 250 mL 2-neck round-bottom flask were added the above silyl ether (15.5 g, 51.3 mmol) and dry THF (150 mL) under an argon atmosphere. The solution was cooled to -78 °C and a solution of *n*-BuLi in hexane (2.4 M, 25 mL, 60 mmol) was added with a syringe during 30 min. The deep reddish brown solution was stirred at -78 °C for 2 hrs and a solution of tributyltin chloride (20.0 g, 61.4 mmol) in dry THF (30 mL) was added dropwise with a syringe during 30 min. The mixture was stirred at -78 °C for 30 min then allowed slowly warm up to room temperature and stirred overnight. Thereafter, the mixture was evaporated to dryness and the residue was diluted with  $\text{Et}_2\text{O}$  (200 mL), washed with water and brine, dried over  $\text{Na}_2\text{SO}_4$ , filtered and evaporated in vacuo. The residue was distilled under vacuum, yielding the title compound as a colorless oil (23.0 g, 87%). B.p.: 170 °C/0.1 mmHg.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.49 (t,  $J = 7.6$  Hz, 1H), 7.31 (d,  $J = 7.4$  Hz, 1H), 7.24 (d,  $J = 7.8$  Hz, 1H), 4.83 (s, 2H), 1.60 – 1.49 (m, 6H), 1.36 – 1.28 (m, 6H), 1.12 – 1.05 (m, 6H), 0.95 (s, 9H), 0.87 (t,  $J = 7.3$  Hz, 9H), 0.12 (s, 6H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  172.63 (s), 161.56 (s), 133.78 (s), 130.57 (s), 118.49 (s), 66.77 (s), 29.22 (s), 27.46 (s), 26.10 (s), 18.52 (s), 13.84 (s), 9.98 (s), -5.16 (s). HRMS (ESI) calcd for  $\text{C}_{24}\text{H}_{48}\text{NOSiSn}^+$

([M+H]<sup>+</sup>): 514.2522; Found: 514.2530.

**6-Diethylamminomethyl-6'-Diphenylphosphinomethyl-2,2'-bipyridine (5b):**

(1) 6-Diethylamminomethyl-6'-Hydroxymethyl-2,2'-bipyridine (**4b**)

To an oven-dried 100 mL Schlenk flask were added Pd(PPh<sub>3</sub>)<sub>4</sub> (1.42 g, 1.2 mmol), anhydrous LiCl (4.5 g, 106 mmol), 2-bromo-6-diethylamminomethylpyridine (8.5 g, 35.0 mmol) and the tributyltin derivative **3b** (21.5 g, 42.0 mmol). The mixture was degassed, followed by the introduction of highly pure argon and then heated at 120 °C for 10 hrs with stirring, at which time a black precipitate was formed. GC analysis showed a full conversion of 2-bromo-6-diethylamminomethylpyridine. After cooling to room temperature, the mixture was diluted with ethyl acetate (100 mL) and filtered to remove unsolved material and then acidified with 6N HCl (aq). The organic layer was washed with water and the combined aqueous layer was made alkaline with 6N NaOH (aq) and extracted with ethyl ether (3 × 100 mL). The combined organic layer was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo. The residue was distilled under vacuum, yielding the title compound as a pale yellow oil (8.0 g, 84%). B.p.: 150 °C/0.1 mmHg. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.36 (d, *J* = 7.8 Hz, 1H), 8.26 (d, *J* = 7.8 Hz, 1H), 7.81, 7.79 (2t, *J* = 7.8 Hz, 2H), 7.52 (d, *J* = 7.8 Hz, 1H), 7.22 (d, *J* = 7.7 Hz, 1H), 4.83 (s, 2H), 4.06 (s, 1H), 3.84 (s, 2H), 2.63 (q, *J* = 7.1 Hz, 4H), 1.10 (t, *J* = 7.1 Hz, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 160.26 (s), 158.22 (s), 155.30 (s), 154.76 (s), 137.67 (s), 137.23 (s), 123.19 (s), 120.29 (s), 119.89 (s), 119.06 (s), 64.05 (s), 59.46 (s), 47.53 (s), 12.16 (s). HRMS (ESI) calcd for C<sub>16</sub>H<sub>22</sub>N<sub>3</sub>O<sup>+</sup> ([M+H]<sup>+</sup>): 272.1757; Found: 272.1762.

(2) 6-Chloromethyl-6'-diethylamminomethyl-2,2'-bipyridine (**6a**)

To a 250 mL 2-neck round-bottom flask were added the above compound (5.8 g, 21.4 mmol) and dry CH<sub>2</sub>Cl<sub>2</sub> (80 mL). The solution was cooled to 0 °C and a solution of SOCl<sub>2</sub> (13.0 g, 109 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) was added with a syringe during 30 min. The reaction solution was allowed to warm up to room temperature and stirred at 45 °C for 3 hrs. The reaction mixture was cooled and then poured into ice-water, made alkaline with 6N NaOH (aq) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 50 mL). The combined organic layer was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo, yielding crude title compound as a pale-brown oil. Due to the instability of the product, it was used without purification for the next step. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.36 (d, *J* = 7.9 Hz, 1H), 8.28 (d, *J* = 7.7 Hz, 1H), 7.83 (t, *J* = 7.7 Hz, 1H), 7.77 (t, *J* = 7.7 Hz, 1H), 7.50, 7.48 (2d, *J* = 8.8 Hz and 8.4 Hz, 2H), 4.75 (s, 2H), 3.83 (s, 2H), 2.62 (q, *J* = 7.0 Hz, 4H), 1.09 (t, *J* = 7.1 Hz, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 160.19 (s), 156.25 (s), 156.07 (s), 154.86 (s), 137.93 (s), 137.25 (s), 123.20 (s), 122.56 (s), 120.49 (s), 119.33 (s), 59.47 (s), 47.54 (s), 47.13 (s), 12.18 (s). HRMS (ESI) calcd for C<sub>16</sub>H<sub>21</sub>ClN<sub>3</sub><sup>+</sup> ([M+H]<sup>+</sup>): 290.1419; Found: 290.1427.

(3) 6-Diethylamminomethyl-6'-Diphenylphosphinomethyl-2,2'-bipyridine (**5b**)

To an oven-dried 250 mL Schlenk flask were added KO<sup>t</sup>Bu (3.4 g, 30.3 mmol) and degassed dry THF (50 mL) under an argon atmosphere. The solution was cooled to 0 °C and a solution of HPPH<sub>2</sub> (4.7 g, 25.2 mmol) in degassed dry THF (15 mL) was added with a syringe during 15 min. The reddish solution was stirred at 0 °C for 1 hr and a solution of the above chloride in degassed dry THF (15 mL) was added dropwise with a syringe during 30 min. The mixture was stirred at 0

°C for 30 min, then allowed slowly to warm up to room temperature and stirred overnight. Degassed water (30 mL) was added and the organic phase was separated, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and concentrated. The residue was distilled under vacuum, yielding the title compound as a pale yellow oil (5.0 g, 54%). B.p.: 230 °C/0.1 mmHg. <sup>31</sup>P {<sup>1</sup>H} NMR (162 MHz, CDCl<sub>3</sub>) δ -11.09 (s). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.18 (d, *J* = 7.8 Hz, 1H), 8.02 (d, *J* = 7.7 Hz, 1H), 7.69 (t, *J* = 7.7 Hz, 1H), 7.60 (t, *J* = 7.7 Hz, 1H), 7.53 – 7.47 (m, 4H), 7.45 (d, *J* = 7.7 Hz, 1H), 7.37 – 7.28 (m, 6H), 7.02 (d, *J* = 7.6 Hz, 1H), 3.81 (s, 2H), 3.71 (s, 2H), 2.62 (q, *J* = 7.1 Hz, 4H), 1.09 (t, *J* = 7.1 Hz, 6H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 159.93 (s), 157.46 (d, *J* = 7.5 Hz), 156.02 (s), 155.49 (s), 138.59 (d, *J* = 14.9 Hz), 137.06 (s, 2C), 133.15 (d, *J* = 19.0 Hz), 128.74 (s), 128.47 (d, *J* = 6.8 Hz), 123.47 (d, *J* = 5.5 Hz), 122.81 (s), 119.42 (s), 118.45 (d, *J* = 1.4 Hz), 59.53 (s), 47.55 (s), 38.72 (d, *J* = 16.3 Hz), 12.23 (s). HRMS (ESI) calcd for C<sub>28</sub>H<sub>31</sub>N<sub>3</sub>P<sup>+</sup> ([M+H]<sup>+</sup>): 440.2250; Found: 440.2258.

### 3. Preparation and Analytical Data of Ligand 5c

#### (1) 6,6'-Dihydroxymethyl-2,2'-bipyridine (4c)

To an oven-dried 100 mL Schlenk flask were added Pd(PPh<sub>3</sub>)<sub>4</sub> (1.14 g, 0.99 mmol), anhydrous LiCl (3.6 g, 84.9 mmol), the tributyltin derivative **3b** (16.9 g, 33.0 mmol) and 2-bromo-6-((*tert*-butyldimethylsilyloxy)methyl)pyridine (8.5 g, 28.1 mmol). The mixture was degassed, followed by the introduction of highly pure argon and then heated at 120 °C for 12 hrs with stirring, at which time a black precipitate was formed. GC analysis showed a full conversion of 2-bromo-6-((*tert*-butyldimethylsilyloxy)methyl)pyridine. After cooling to room temperature, the mixture was diluted with ethyl acetate (100 mL) and filtered to remove unsolved material and then acidified with 6N HCl (aq). The organic layer was washed with water and the combined aqueous layer was made alkaline with 6N NaOH (aq) and extracted with ethyl ether (3 × 100 mL). The combined organic layer was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo, yielding the title compound as a white solid (5.2 g, 85%). <sup>1</sup>H NMR (400 MHz, DMSO) δ 8.23 (d, *J* = 7.7 Hz, 2H), 7.93 (t, *J* = 7.8 Hz, 2H), 7.52 (d, *J* = 7.6 Hz, 2H), 5.48 (t, *J* = 5.8 Hz, 2H), 4.65 (d, *J* = 5.7 Hz, 4H).

#### (2) 6,6'-Dichloromethyl-2,2'-bipyridine (6b)

To an oven-dried 100 mL Schlenk flask were added 6,6'-dihydroxymethyl-2,2'-bipyridine (1.7 g, 7.9 mmol) and dry CH<sub>2</sub>Cl<sub>2</sub> (40 mL) under an argon atmosphere. The solution was cooled to 0 °C and a solution of SOCl<sub>2</sub> (4.8 g, 40.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added with a syringe during 15 min. The reaction solution was allowed to warm up to room temperature and stirred at 45 °C for 3 hrs. The reaction mixture was cooled and then poured into ice-water, made alkaline with 6N NaOH (aq) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 50 mL). The combined organic layer was washed with water and brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo, yielding crude title compound as a pale-yellow solid (1.8 g, 90%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.40 (d, *J* = 7.9 Hz, 2H), 7.85 (t, *J* = 7.8 Hz, 2H), 7.51 (d, *J* = 7.7 Hz, 2H), 4.75 (s, 4H).

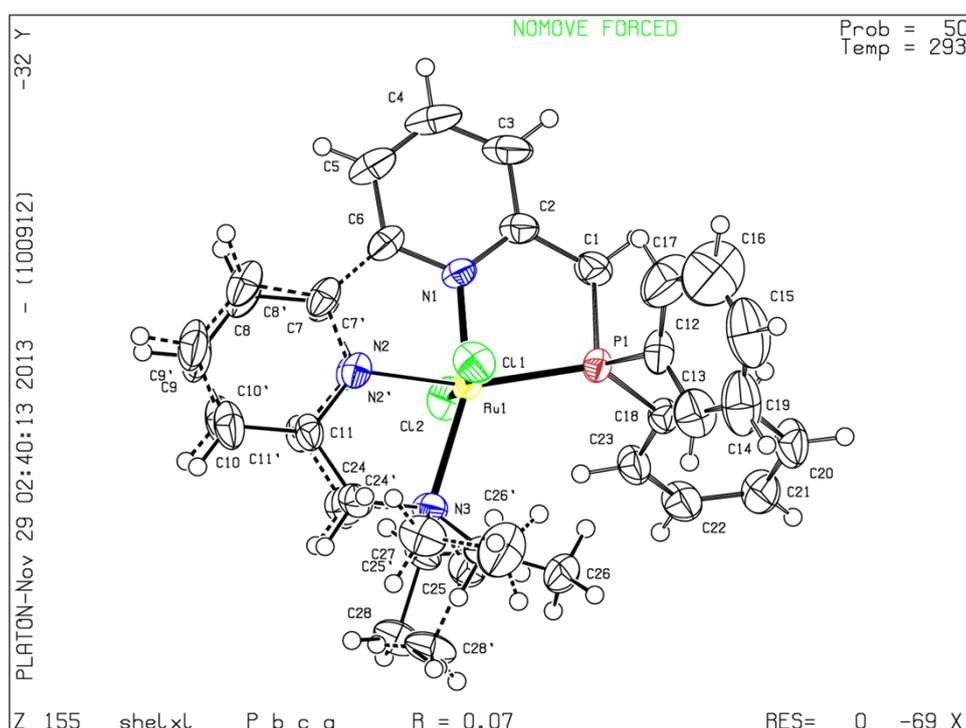
#### (3) 6,6'-Bis((di-*tert*-butylphosphino)methyl)-2,2'-bipyridine (5c)

To an oven-dried 50 mL Schlenk flask were added 6,6'-dichloromethyl-2,2'-bipyridine (1.0 g, 4.0 mmol), HP<sup>t</sup>Bu<sub>2</sub> (2.4 g, 16.4 mmol) and degassed dry MeOH (30 mL) under an argon atmosphere. The reaction solution was stirred at 60 °C for 48 hrs to give a white suspension. After



6.7 Hz, 2H), 3.20, 3.17 (2q,  $J = 6.4$  Hz and 6.6 Hz, 2H), 1.08 (t,  $J = 7.0$  Hz, 6H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  165.93 (d,  $J = 5.1$  Hz), 160.76 (d,  $J = 3.1$  Hz), 160.53 (s), 157.62 (s), 137.24 (d,  $J = 34.5$  Hz), 134.74 (s), 133.61 (d,  $J = 10.4$  Hz), 132.12 (s), 129.27 (s), 127.82 (d,  $J = 8.9$  Hz), 121.45 (d,  $J = 2.9$  Hz), 120.82 (s), 120.69 (s), 120.56 (d,  $J = 9.3$  Hz), 66.63 (s), 51.28 (s), 48.59 (d,  $J = 25.4$  Hz), 10.53 (s). HRMS (ESI) calcd for  $\text{C}_{28}\text{H}_{30}\text{ClN}_3\text{PRu}^+$  ( $[\text{M}-\text{Cl}]^+$ ): 576.0904; Found: 576.0897.

Single crystal of complex **1b** was grown by slow diffusion of *n*-hexane into its saturated solution in  $\text{CH}_2\text{Cl}_2$  at room temperature. Data of X-ray diffraction of single crystal of complex **1b** were collected at 293 K on a Rigaku 007 Saturn 70 CCD diffractometer. Positional disorder is observed in the atoms of one of the pyridine ring and the (diethylamino)methyl substituent (except for the N3 atom). These atoms are disordered over two orientations with a refined site-occupancy ratio of 0.629(5):0.371(5). The dihedral angles between the other pyridine ring and the major and minor components of the disordered pyridine ring are 12.2(5) and  $-15.2(9)^\circ$ , respectively.



**Figure S1** Single-crystal X-ray structure of complex **1b**, showing 50% probability displacement ellipsoids and the atom numbering scheme (H atoms are omitted for clarity). Open bonds indicate the minor disorder component.

**Table S1** Crystal data and structure of refinement for complex **1b**

Identification code	shelxl
Empirical formula	$\text{C}_{28}\text{H}_{30}\text{Cl}_2\text{N}_3\text{PRu}$
Formula weight	611.49
Temperature	293(2) K
Wavelength	0.71075 Å

Crystal system, space group	Orthorhombic, P b c a
Unit cell dimensions	a = 11.8935(5) Å $\alpha$ = 90 deg. b = 13.1976(6) Å $\beta$ = 90 deg. c = 33.9681(16) Å $\gamma$ = 90 deg.
Volume	5331.8(4) Å <sup>3</sup>
Z, Calculated density	8, 1.524 Mg/m <sup>3</sup>
Absorption coefficient	0.871 mm <sup>-1</sup>
F(000)	2496
Crystal size	0.420 × 0.320 × 0.120 mm
Theta range for data collection	2.090 to 29.499 deg.
Limiting indices	-16 ≤ h ≤ 16, -18 ≤ k ≤ 17, -47 ≤ l ≤ 46
Reflections collected/unique	42309/7351 [R(int) = 0.0397]
Completeness to theta = 25.242	99.4 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.840 and 0.691
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data/restraints/parameters	7351/49/361
Goodness-of-fit on F <sup>2</sup>	1.293
Final R indices [I > 2σ(I)]	R1 = 0.0659, wR2 = 0.1255
R indices (all data)	R1 = 0.0705, wR2 = 0.1273
Extinction coefficient	0.0042(2)
Largest diff. peak and hole	0.954 and -1.009 e. Å <sup>-3</sup>

**Table S2** Selected bond lengths [Å] and angles [deg] for complex **1b**

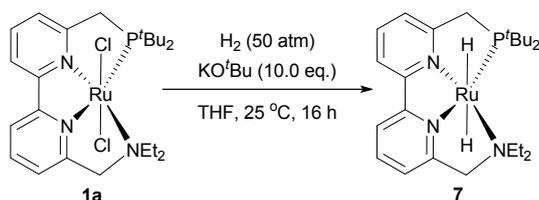
Ru(1)-N(1)	1.992(3)	Ru(1)-N(2)	1.946(5)
Ru(1)-N(2')	2.016(8)	Ru(1)-N(3)	2.262(3)
Ru(1)-P(1)	2.2821(11)	Ru(1)-Cl(1)	2.4196(12)
Ru(1)-Cl(2)	2.4088(12)		
N(1)-Ru(1)-N(2)	79.9(2)	N(2)-Ru(1)-N(3)	79.9(2)
N(1)-Ru(1)-N(3)	159.34(14)	N(1)-Ru(1)-P(1)	84.14(11)
N(2)-Ru(1)-P(1)	163.3(2)	N(3)-Ru(1)-P(1)	116.35(9)
N(1)-Ru(1)-Cl(1)	91.29(11)	N(1)-Ru(1)-Cl(2)	84.12(11)
N(2)-Ru(1)-Cl(1)	83.48(18)	N(2)-Ru(1)-Cl(2)	90.64(18)
N(3)-Ru(1)-Cl(1)	90.64(9)	N(3)-Ru(1)-Cl(2)	91.87(9)

P(1)-Ru(1)-Cl(1)	92.08(4)	P(1)-Ru(1)-Cl(2)	92.56(4)
Cl(1)-Ru(1)-Cl(2)	173.11(5)	N(2')-Ru(1)-N(1)	79.0(4)
N(2')-Ru(1)-N(3)	80.3(4)	N(2')-Ru(1)-P(1)	162.4(4)
N(2')-Ru(1)-Cl(1)	93.4(3)	N(2')-Ru(1)-Cl(2)	80.7(3)

### 3. Preparation and Analytical Data of Complex 1c

To an oven-dried 15 mL Schlenk tube was added PNNP ligand **5c** (113 mg, 0.24 mmol),  $[\text{RuCl}_2(\eta^6\text{-}p\text{-cymene})]_2$  (61 mg, 0.1 mmol), and degassed dry  $\text{CH}_2\text{Cl}_2$  (5 mL) under argon. The mixture was stirred at room temperature for 4 hrs to give a deep blue violet suspension. The solvent was removed under vacuum. The solid was washed with ethyl ether ( $3 \times 10$  mL), filtered and dried under vacuum to give pure complex **1c** as a deep blue violet solid (117 mg, 91%).  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  66.25 (s).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.87 (d,  $J = 7.8$  Hz, 2H), 7.63 (t,  $J = 7.8$  Hz, 2H), 7.57 (d,  $J = 7.7$  Hz, 2H), 3.87 (d,  $J = 8.3$  Hz, 4H), 1.45 (d,  $J = 11.3$  Hz, 36H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  166.50 (s), 159.51 (s), 134.04 (s), 122.30 (vt,  $J = 4.8$  Hz), 120.45 (s), 39.25 (dd,  $J = 11.8, 6.7$  Hz), 37.46 (vt,  $J = 3.8$  Hz), 30.70 (s). Anal. Calcd. for  $\text{C}_{28}\text{H}_{46}\text{Cl}_2\text{N}_2\text{P}_2\text{Ru}$ : C, 52.17; H, 7.19; N, 4.35. Found: C, 52.29; H, 6.91; N, 4.56.

### (C) Preliminary Study on the Reaction of 1a with Hydrogen in the Presence of KO<sup>t</sup>Bu



In a glove box, a hydrogenation vessel was charged with complex **1a** (11.5 mg, 20  $\mu\text{mol}$ ) and  $\text{KO}^t\text{Bu}$  (23 mg, 0.2 mmol). The vessel was placed into the autoclave and degassed dry THF (5.0 mL) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 50 atm. The reaction mixture was stirred at 25  $^\circ\text{C}$  for 16 hrs and then the pressure was released. The reaction mixture was transferred to a Schlenk tube under an argon atmosphere and the solvent was removed under vacuum. The remaining residue was taken back into the glove box, redissolved in degassed  $\text{C}_6\text{D}_6$  (1.0 mL), passed through a PTFE syringe filter, and transferred into a J-Young NMR tube. Then the sample was submitted to NMR analysis immediately. The preliminary  $^{31}\text{P}\{^1\text{H}\}$  NMR and hydride region  $^1\text{H}$  NMR study on the major product is shown in Figure S2 and S3, respectively.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  123.12 (s).  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  -11.08 (d,  $J = 14.4$  Hz, 2H). The chemical shift of -11.08 ppm in  $^1\text{H}$  NMR for the hydride ligands suggests the coordination of a ligand with a strong *trans* influence (such as a hydride) in the *trans* position. Based on these preliminary results, we tentatively assign the major product to be *trans*-dihydride complex **7**. Attempts to isolate and purify this complex, however, failed due to a lack of sufficient stability. Further studies are needed to fully characterize this complex.

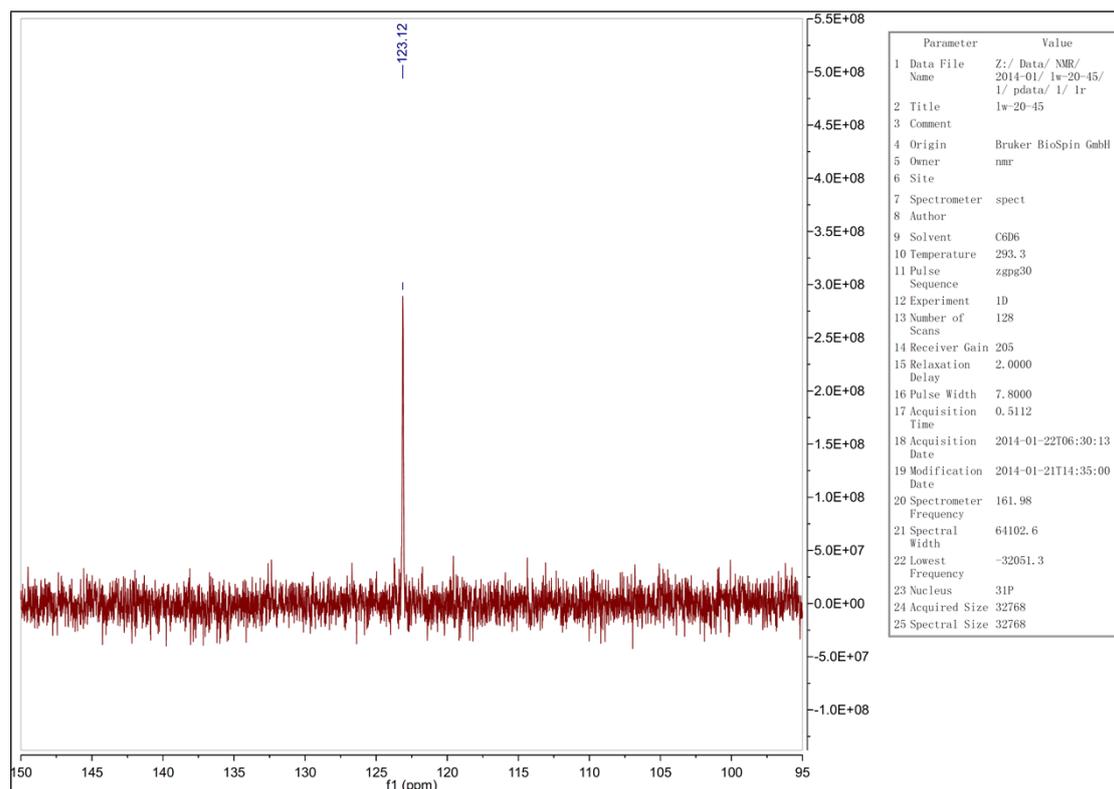


Figure S2 Preliminary  $^{31}\text{P}\{^1\text{H}\}$  NMR study of complex 7.

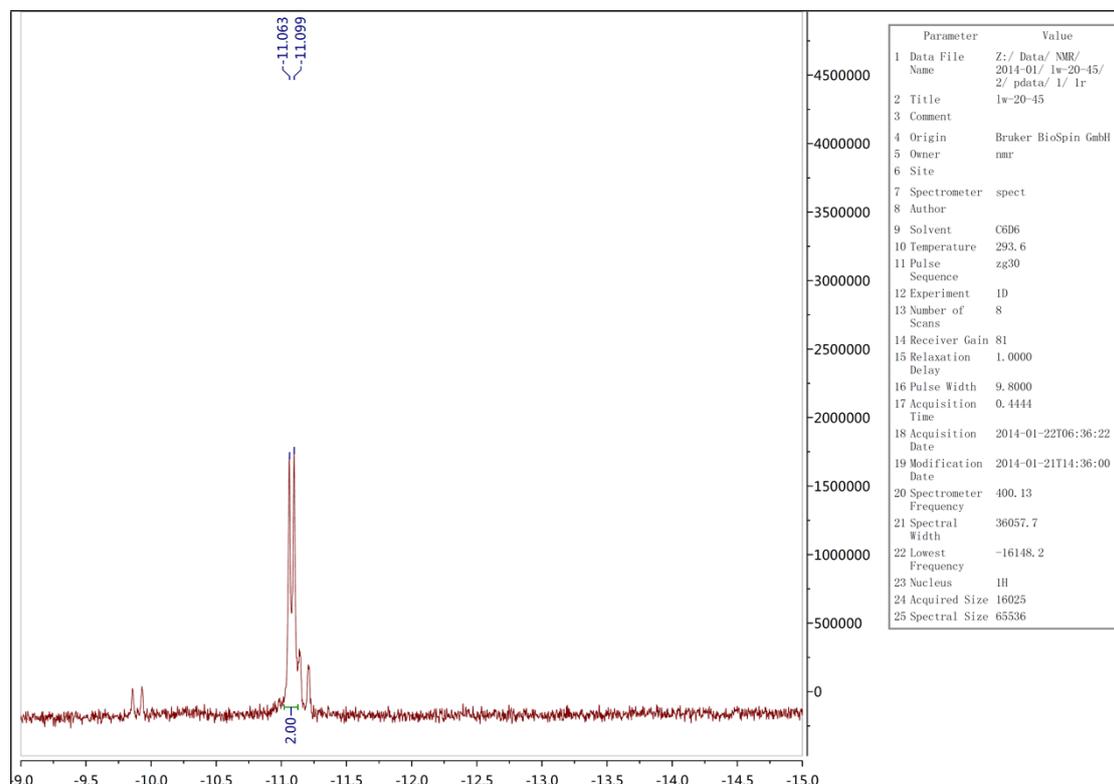


Figure S3 Preliminary hydride region  $^1\text{H}$  NMR study of complex 7.

## (D) Procedures for Hydrogenation

**Hydrogenation of GVL at S/C = 1000.** In a glove box, a hydrogenation vessel was charged

with complex **1a** (1.7 mg, 3.0  $\mu\text{mol}$ ) and NaOMe (17 mg, 0.3 mmol). The vessel was placed into the autoclave and a degassed solution of GVL (300.3 mg, 3.0 mmol) in dry *i*PrOH (5.0 mL) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 50 atm. The reaction mixture was stirred at 25 °C until no pressure drop was observed (4 hrs) and then the pressure was released. After adding 1,5-pentandiol (1,5-PDO) as an internal standard, an aliquot portion of the mixture was taken and filtered through a short silica column and submitted to analysis of yield of 1,4-PDO by GC (Agilent DB-35, 30 m  $\times$  0.32 mm  $\times$  0.25  $\mu\text{m}$ ) using a flame ionization detector (FID) operating at 250 °C. Injector temperature was set at 230 °C. The carrier gas was nitrogen with a flow rate of 1.0 mL/min. The following temperature program was used in the analysis: 50 °C – 10 °C/min – 250 °C (10 min). Using these conditions, the retention times for 1,4-PDO (7.918 min), GVL (8.295 min), and 1,5-PDO (9.035 min) were observed. The GC yield 1,4-PDO is 99%.

**Hydrogenation of GVL at S/C = 100,000.** A solution of complex **1a** ( $3.5 \times 10^{-4}$  mmol/mL) was made by dissolving complex **1a** (1.0 mg, 0.00175 mmol) in degassed dry *i*PrOH (5.0 mL). In a glove box, a hydrogenation vessel was charged with NaOMe (195 mg, 3.6 mmol) and then placed into the autoclave. A degassed solution of GVL (3.61 g, 36 mmol) in dry *i*PrOH (3.0 mL) and the freshly prepared complex **1a** solution (1.0 mL,  $3.5 \times 10^{-4}$  mmol) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 100 atm. The reaction mixture was stirred at 25 °C until no pressure drop was observed (48 hrs) and then the pressure was released. An aliquot portion of the mixture was taken and filtered through a short silica column and submitted to GC analysis. The GC yield is 91%. Silica gel was added to the rest of the mixture, the solvent was removed in vacuo and the residue was purified by flash column chromatograph (silica gel, eluents: petroleum ether/ethyl acetate (1:1), then ethyl acetate) to afford 1,4-PDO (3.40 g, 90%) as a colorless oil.

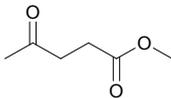
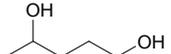
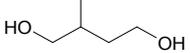
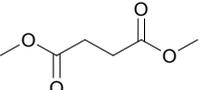
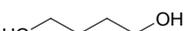
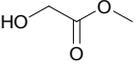
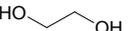
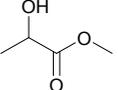
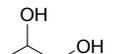
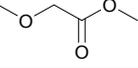
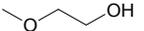
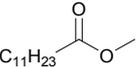
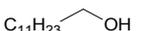
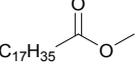
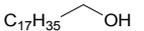
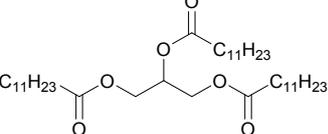
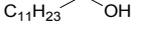
**Hydrogenation of Methyl Benzoate at S/C = 100,000.** In a glove box, a hydrogenation vessel was charged with complex **1a** (0.5 mg, 1.0  $\mu\text{mol}$ ) and NaOMe (540 mg, 10 mmol). The vessel was placed into the autoclave and a degassed solution of methyl benzoate (13.7 g, 100 mmol) in dry *i*PrOH (20 mL) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 100 atm. The reaction mixture was stirred at 25 °C until no pressure drop was observed (64 hrs) and then the pressure was released. An aliquot portion of the mixture was taken and filtered through a short silica column and submitted to GC analysis. The GC yield is 91%. Silica gel was added to the rest of the mixture, the solvent was removed in vacuo and the residue was purified by flash column chromatograph (silica gel, eluents: petroleum ether/ethyl acetate (3:1)) to afford benzyl alcohol (9.4 g, 87%) as a colorless oil.

**Hydrogenation of Ethyl Acetate at S/C = 100,000.** In a glove box, a hydrogenation vessel was charged with complex **1a** (1.1 mg, 2.0  $\mu\text{mol}$ ) and NaOEt (1.36 g, 20 mmol). The vessel was placed into the autoclave and degassed ethyl acetate (17.6 g, 200 mmol) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 50 atm. The reaction mixture was

stirred at 40 °C until no pressure drop was observed (48 hrs) and then the pressure was released. After adding toluene as an internal standard, an aliquot portion of the mixture was taken and filtered through a short silica column and submitted to GC analysis. The GC yield is 90%.

**General Procedure for the Hydrogenation of Other Substrates at S/C = 1000.** In a glove box, a hydrogenation vessel was charged with complex **1a** (1.7 mg, 3.0  $\mu$ mol) and NaOMe (17 mg, 0.3 mmol). The vessel was placed into the autoclave and a degassed solution of the substrate (3 mmol) in dry *i*PrOH (3.0 mL) was added through the injection port. The autoclave was tightened and purged three times with argon and three times with hydrogen before it was finally charged with hydrogen to 50 atm. The reaction mixture was stirred at 25 °C or 100 °C until no pressure drop was observed and then the pressure was released. An aliquot portion of the mixture was taken and filtered through a short silica column and submitted to GC analysis. For nonvolatile product, isolated yield was obtained by flash column chromatography.

**Table S3** Hydrogenation of other substrates with complex **1a**<sup>a</sup>

Entry	Substrate	Product	Time (h)	Yield (%) <sup>b</sup>
1 <sup>c</sup>			16	95
2			6	99(96)
3			6	99(96)
4			6	99(97)
5 <sup>d</sup>			2	99(97)
6 <sup>d</sup>			2	99(97)
7			4	99
8			16	99(95)
9 <sup>d</sup>			2	99(97)
10 <sup>e</sup>			2	99(97)

11 <sup>e</sup>			2	99(98)
12 <sup>f</sup>			24	97
13 <sup>f</sup>			24	96
14 <sup>g</sup>			16	99(99)
15 <sup>d</sup>			2	99(99)
16			16	99(96)
17 <sup>d</sup>			2	99(97)
18 <sup>g</sup>			16	99(95) 99 <sup>h</sup>

<sup>a</sup> Reaction conditions: 3.0 mmol substrate, 0.1 mol% **1a**, 3.0 mL <sup>i</sup>PrOH, 10 mol% NaOMe, P(H<sub>2</sub>) = 50 atm, 25 °C.

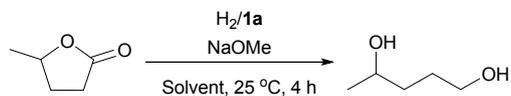
<sup>b</sup> GC yield, isolated yield in the parentheses. The balance of material present was unreacted starting material. <sup>c</sup> The

balance of material present was GVL. <sup>d</sup> 100 °C. <sup>e</sup> 100 °C, 0.3 mol% **1a**. <sup>f</sup> S/C = 10,000, 12 mmol substrate, 3.0 mL

<sup>i</sup>PrOH, 10 mol% NaOMe, P(H<sub>2</sub>) = 50 atm, 25 °C. <sup>g</sup> THF as solvent. <sup>h</sup> GC yield of methanol.

## (E) Complementary Reaction Optimization Data

**Table S4** Effect of solvent on the hydrogenation of GVL into 1,4-PDO with complex **1a**<sup>a</sup>



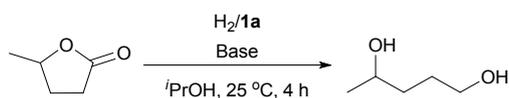
Entry	Solvent	Yield (%) <sup>b</sup>
1	THF	98
2	Et <sub>2</sub> O	94
3	2-MeTHF	98
4	CH <sub>2</sub> Cl <sub>2</sub>	0
5	DMF	0
6	DMSO	0
7	Toluene	78
8	MeOH	0

9	EtOH	20
10	<i>i</i> PrOH	99
11	Neat	95

<sup>a</sup> Reaction conditions: 3.0 mmol GVL, 0.1 mol% **1a**, 5.0 mL solvent, 10 mol% NaOMe, P(H<sub>2</sub>) = 50 atm, 25 °C, 4 h.

<sup>b</sup> GC yield. The balance of material present was unreacted GVL.

**Table S5** Effect of base on the hydrogenation of GVL into 1,4-PDO with complex **1a**<sup>a</sup>

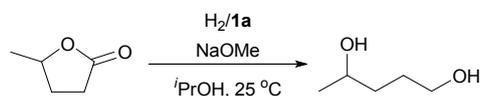


Entry	Base	Yield (%) <sup>b</sup>
1	NaOMe	99
2 <sup>c</sup>	NaOMe	62
3 <sup>c,d</sup>	NaOMe	95
4 <sup>e</sup>	NaOMe	0
5 <sup>e,f</sup>	NaOMe	0
6	NaOEt	96
7	NaO <sup><i>i</i></sup> Pr	79
8 <sup>f</sup>	NaO <sup><i>i</i></sup> Pr	99
9	NaO <sup><i>t</i></sup> Bu	72
10 <sup>f</sup>	NaO <sup><i>t</i></sup> Bu	99
11	KO <sup><i>t</i></sup> Bu	98
12	KOH	0
13	NaOH	0
14	Cs <sub>2</sub> CO <sub>3</sub>	0
15	K <sub>2</sub> CO <sub>3</sub>	0
16	DBU	0
17	Et <sub>3</sub> N	0

<sup>a</sup> Reaction conditions: 3.0 mmol GVL, 0.1 mol% **1a**, 5.0 mL *i*PrOH, 10 mol% Base, P(H<sub>2</sub>) = 50 atm, 25 °C, 4 h. <sup>b</sup>

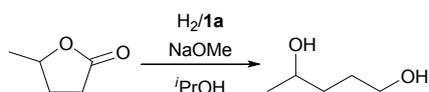
GC yield. The balance of material present was unreacted GVL. <sup>c</sup> 5 mol% NaOMe. <sup>d</sup> 10 h. <sup>e</sup> 1 mol% NaOMe. <sup>f</sup> 16 h.

**Table S6** Effect of H<sub>2</sub> pressure on the hydrogenation of GVL into 1,4-PDO with complex **1a**<sup>a</sup>



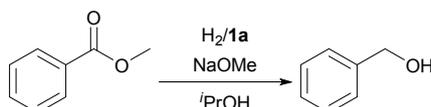
Entry	P(H <sub>2</sub> ) (atm)	Time (h)	Yield (%) <sup>b</sup>
1	50	4	99
2	30	4	92
3	30	6	98
4	10	4	51
5	10	16	95

<sup>a</sup> Reaction conditions: 3.0 mmol GVL, 0.1 mol% **1a**, 5.0 mL *i*PrOH, 10 mol% NaOMe, 25 °C. <sup>b</sup> GC yield. The balance of material present was unreacted GVL.

**Table S7** Hydrogenation of GVL at low catalyst loading with complex **1a**<sup>a</sup>

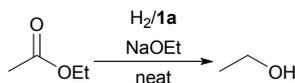
Entry	P(H <sub>2</sub> ) (atm)	Temp. (°C)	Time (h)	Yield (%) <sup>b</sup>
1 <sup>c</sup>	50	25	10	98
2	50	25	64	50
3	50	40	48	76
4	100	25	48	91(90)
5 <sup>d</sup>	100	25	48	67
6 <sup>e</sup>	100	25	48	30

<sup>a</sup> Reaction conditions: 36 mmol GVL, 0.001 mol% **1a**, 4.0 mL *i*PrOH, 10 mol% NaOMe. <sup>b</sup> GC yield, isolated yield in the parentheses. The balance of material present was unreacted GVL. <sup>c</sup> 12 mmol GVL, 0.01 mol% **1a**, 1.0 mL *i*PrOH, 10 mol% NaOMe. <sup>d</sup> 5 mol% NaOMe. <sup>e</sup> 1 mol% NaOMe.

**Table S8** Hydrogenation of methyl benzoate at low catalyst loading with complex **1a**<sup>a</sup>

Entry	P(H <sub>2</sub> ) (atm)	Temp. (°C)	Time (h)	Yield (%) <sup>b</sup>
1 <sup>c</sup>	50	25	16	99(97)
2	50	25	64	47
3	50	40	64	60
4	100	25	48	58
5	100	25	64	91(87)

<sup>a</sup> Reaction conditions: 100 mmol methyl benzoate, 0.001 mol% **1a**, 20 mL *i*PrOH, 10 mol% NaOMe. <sup>b</sup> GC yield, isolated yield in the parentheses. The balance of material present was unreacted methyl benzoate. <sup>c</sup> 12 mmol methyl benzoate, 0.01 mol% **1a**, 3.0 mL *i*PrOH, 10 mol% NaOMe.

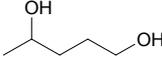
**Table S9** Hydrogenation of ethyl acetate at low catalyst loading with complex **1a**<sup>a</sup>

Entry	P(H <sub>2</sub> ) (atm)	Temp. (°C)	Time (h)	Yield (%) <sup>b</sup>
1	50	25	64	79
2	50	40	21	55
3	50	40	48	90
4	100	25	48	88

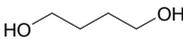
<sup>a</sup> Reaction conditions: 200 mmol ethyl acetate, 0.001 mol% **1a**, 10 mol% NaOEt. <sup>b</sup> GC yield. The balance of material present was unreacted ethyl acetate.

## (F) Analytical Data of the Hydrogenation Products

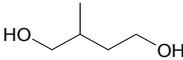
### Pentane-1,4-diol<sup>5</sup>

 Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.92 – 3.81 (m, 1H), 3.76 – 3.62 (m, 2H), 2.06 (br s, 2H), 1.78 – 1.45 (m, 4H), 1.22 (d, *J* = 6.2 Hz, 3H). GC-MS (EI, 70 eV): *m/z* (%) = 89(5), 71(100), 58(16), 45(82), 31(16).

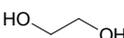
### Butane-1,4-diol<sup>6</sup>

 Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.92 – 3.58 (m, 4H), 2.57 (br s, 2H), 1.93 – 1.59 (m, 4H). GC-MS (EI, 70 eV): *m/z* (%) = 71(42), 57(31), 42(100), 31(71).

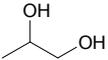
### 2-Methylbutane-1,4-diol<sup>7</sup>

 Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.89 – 3.35 (m, 4H), 2.86 (br s, 2H), 1.94 – 1.73 (m, 1H), 1.75 – 1.49 (m, 2H), 0.93 (d, *J* = 6.9 Hz, 3H). GC-MS (EI, 70 eV): *m/z* (%) = 86(3), 71(30), 58(34), 56(100), 55(43), 43(30), 41(46), 31(36).

### Ethane-1,2-diol<sup>8</sup>

 Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.93 – 3.53 (m, 4H), 2.23 – 1.85 (m, 2H). GC-MS (EI, 70 eV): *m/z* (%) = 62([M<sup>+</sup>], 7), 43(18), 31(100).

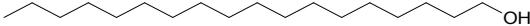
### Propane-1,2-diol<sup>9</sup>

 Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.99 – 3.85 (m, 1H), 3.70 – 3.59 (m, 1H), 3.47 – 3.33 (m, 1H), 2.00 (d, *J* = 3.9 Hz, 1H), 1.83 (t, *J* = 5.8 Hz, 1H), 1.17 (d, *J* = 6.4 Hz, 3H). GC-MS (EI, 70 eV): *m/z* (%) = 76([M<sup>+</sup>], 1), 61(8), 45(100), 43(20), 31(12), 15(2).

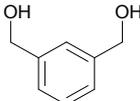
### Dodecan-1-ol<sup>10</sup>

 White solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.63 (t, *J* = 6.6 Hz, 2H), 1.61 – 1.50 (m, 2H), 1.40 – 1.18 (m, 19H), 0.87 (t, *J* = 6.8 Hz, 3H). GC-MS (EI, 70 eV): *m/z* (%) = 168(3), 140(18), 125(13), 111(42), 97(74), 83(94), 69(98), 55(100), 43(64), 31(38).

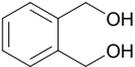
### Octadecan-1-ol<sup>10</sup>

 White solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 3.63 (t, *J* = 6.6 Hz, 2H), 1.76 (br s, 1H), 1.63 – 1.50 (m, 2H), 1.39 – 1.16 (m, 30H), 0.87 (t, *J* = 6.8 Hz, 3H). GC-MS (EI, 70 eV): *m/z* (%) = 252(3), 224(6), 196(3), 182(3), 168(4), 153(6), 139(11), 125(29), 111(54), 97(92), 83(100), 69(89), 55(93), 43(75), 31(22).

### 1,3-Phenylenedimethanol<sup>11</sup>

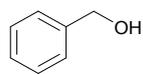
 White solid. <sup>1</sup>H NMR (400 MHz, DMSO) δ 7.33 – 7.15 (m, 4H), 5.21 (t, *J* = 5.7 Hz, 2H), 4.51 (d, *J* = 5.7 Hz, 4H). GC-MS (EI, 70 eV): *m/z* (%) = 138([M<sup>+</sup>], 67), 120(22), 107(63), 91(85), 79(100), 77(64), 65(13), 51(15), 39(9), 31(7).

### 1,2-Phenylenedimethanol<sup>12</sup>

 White solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.40 – 7.29 (m, 4H), 4.75 (s, 4H), 2.79 (s, 2H). GC-MS (EI, 70 eV): *m/z* (%) = 120(100), 91(83), 77(29), 65(16), 51(9),

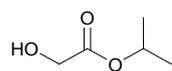
39(7), 29(3).

### Phenylmethanol<sup>13</sup>



Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.40 – 7.28 (m, 5H), 4.71 (d, *J* = 6.0 Hz, 2H), 1.62 (t, *J* = 6.0 Hz, 1H). GC-MS (EI, 70 eV): *m/z* (%) = 108([M<sup>+</sup>], 79), 91(16), 79(100), 77(62), 65(7), 51(23), 39(9), 29(3).

### Isopropyl glycolate<sup>14</sup>



Colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 5.12 (hept, *J* = 6.3 Hz, 1H), 4.10 (d, *J* = 5.2 Hz, 2H), 2.50 (t, *J* = 5.3 Hz, 1H), 1.27 (d, *J* = 6.3 Hz, 6H). GC-MS (EI, 70 eV): *m/z* (%) = 75(20), 59(9), 45(64), 43(100), 41(44), 39(11), 31(21), 27(14).

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