

Structural-functional evaluation of ionic liquid libraries for the design of co-solvents in lipase-catalysed reactions

ELECTRONIC SUPPLEMENTARY INFORMATION

5

10 Materials and methods

Lipolytic activity assay method - 2,3-dimercapto-1-propanol tributyrate (Sigma Aldrich), DMPTB, was dissolved in Triton X-100 (Sigma Aldrich) and in 50 mM trizma base buffer (Sigma Aldrich), pH 7.2. The final stock solution is 10 mM DMPTB and 2% Triton X-100 and was stored at -20 °C. A 40 mM 15 solution of 5,5'-dithiobis(2-nitro benzoic acid) (Sigma Aldrich), DTNB, in isobutanol was prepared daily. Thermomyces lanuginosus lipase (Sigma Aldrich, L0777), TIL, was dissolved in 50 mM trizma base buffer pH 7.2 with 0.0025% of triton X-100. The standard reaction mixture contained 4.44 mM DMPTB, 0.88 mM DTNB, 0.0013% Triton X-100, and 26.70 mM Trizma base, pH 7.2. The concentration of enzyme used was 70 nM. Each well of the 96-well microplate was filled with 100 µl 20 of a 50% IL (prepared in water), 50 µl of the standard reaction mixture and 50 µl of the enzyme sample. Controls without the enzyme and/or the IL were also performed.

The time-dependent absorbance change was monitored at 405 nm for 10 min of reaction in a Biotek Synergy 2 multi-mode microplate reader. The lipolytic activity was calculated based on the initial velocity of the reaction (Abs/s). Lipolytic activity is calculated as a percentage of the activity of TIL in 25 buffer in the presence of the selected ionic liquids and represents the average of at least 2 replicates.

Differential Scanning Fluorimetry (DSF)- Differential scanning fluorimetry, runs were performed by monitoring the fluorescence of the exogenous probe Sypro Orange (Life Technologies, 5000x concentrate in DMSO) in a Bio Rad IQ5 Multicolor Real-Time PCR detection system equipped with a 30 charge-coupled device (CCD) camera and a Cy3 filter with excitation and emission wavelengths of 490 and 575 nm, respectively. 75 µL of 6.67 µM enzyme solution in 20x DMSO concentrated sypro orange probe was left in contact with 25 µL of 25% (w/w) IL solution for 15 minutes. Each well of the 96-Well PCR plate was filled with 20 µL of this mixture. The assays were performed in a Bio Rad IQ5 Multicolor Real-Time PCR detection system with heating rates of 1.0 °C min⁻¹ from 20 to 90 °C. 35 Controls without the enzyme and/or the IL were also performed.

Table S1 – List of ionic liquids used in this work and respective activity and stability data for T_{IL}

Compound	Activity (%)	Normalized intensity	Apparent T_m by DSC (°C)	ΔT_m (°C)	Transition steepness (%)	logP (Cation)	logP (Anion)	LogP (IL)
Control (Water)	100	0	70	0	100	-	-	-
(CH ₃) ₂ N ₂ CNH ₂ N(CN) ₂	19	#N/D	#N/D	#N/D	#N/D	-2.043	0.282	-1.761
N _{1111(2OH)} Bicarbonate	10	3	59	-11	68.3	-1.574	-0.516	-2.09
N _{1111(2OH)} NO ₃	12	8	54	-16	55.1	-1.574	1.31	-0.264
N _{1111(2OH)} C ₃ COO	39	5	64	-6	67.2	-1.574	0.215	-1.359
N _{1111(2OH)} (C ₁) ₂ PO ₄	42	6	69	-1	93.6	-1.574	-1.094	-2.668
N _{1111(2OH)} C ₂ COO	44	6	67	-3	80.2	-1.574	-0.241	-1.815
N _{1111(2OH)} C ₄ COO	46	12	60	-10	36.5	-1.574	0.672	-0.902
N _{1111(2OH)} C ₂ SO ₃	54	12	63	-7	74.0	-1.574	-0.768	-2.342
N _{1111(2OH)} Pivalate	54	19	60	-10	38.9	-1.574	0.639	-0.935
N _{1111(2OH)} Salicylate	57	22	35	-35	38.6	-1.574	0.489	-1.085
N _{1111(2OH)} Cl	61	7	40	-30	63.1	-1.574	0.77	-0.804
N _{1111(2OH)} OTf	71	9	55	-15	59.9	-1.574	1.019	-0.555
N _{1111(2OH)} C ₁ COO	87	4	70	0	86.1	-1.574	-0.908	-2.482
N _{1111(2OH)} (C ₁) ₂ CCOO	88	7	60	-10	59.9	-1.574	0.222	-1.352
N _{1111(2OH)} C ₁ SO ₃	100	9	66	-4	89.7	-1.574	-1.117	-2.691
N _{1111(2OH)} C ₅ COO	#N/D	92	54	-16	8.0	-1.574	1.128	-0.446
N _{1111(2OH)} H ₂ PO ₄	#N/D	2	37	-33	27.0	-1.574	-1.911	-3.485
N _{1112(2OH)} Br	33	29	63	-7	60.1	-1.225	0.89	-0.335
N _{1111(2OH)} Cl	70	6	57	-13	65.0	-1.189	0.77	-0.419
N _{1113(2OH)} Br	23	35	50	-20	53.8	-0.702	0.89	0.188
N _{1114(2OH)} Cl	97	15	60	-10	71.2	-0.316	0.77	0.454
N _{1114(2OH)} Br	6	30	59	-11	16.2	-0.072	0.89	0.818
C ₁ C ₁ im C ₁ SO ₃	46	42	40	-30	40.7	0.025	-1.117	-1.092
N _{1115(2OH)} Br	7	75	53	-17	2.8	0.211	0.89	1.101
C ₂ C ₁ im N(CN) ₂	0	85	No transition	No transition	No transition	0.374	0.282	0.656
C ₂ C ₁ im HCOO	11	42	59	-11	25.7	0.374	-0.516	-0.142
C ₂ C ₁ im C ₂ SO ₃	12	21	65	-5	19.1	0.374	-0.768	-0.394
C ₂ C ₁ im Cl	17	38	57	-13	10.5	0.374	0.77	1.144
C ₂ C ₁ im Pivalate	19	29	55	-15	21.2	0.374	0.639	1.013
C ₂ C ₁ im C ₂ COO	19	12	62	-8	29.4	0.374	-0.241	0.133
C ₂ C ₁ im C ₁ COO	24	35	64	-6	40.6	0.374	-0.908	-0.534
C ₂ C ₁ im C ₁ SO ₃	24	88	67	-3	35.8	0.374	-1.117	-0.743
C ₂ C ₁ im C ₆ SO ₃	27	76	No transition	No transition	No transition	0.374	1.124	1.498
C ₂ C ₁ im C ₄ SO ₃	35	57	No transition	No transition	No transition	0.374	0.212	0.586
C ₂ C ₁ im Gala	71	1	66	-4	41.5	0.374	-3.153	-2.779
C ₂ C ₁ im C ₄ SO ₄	#N/D	50	No transition	No transition	No transition	0.374	0.058	0.432
C ₂ C ₁ im C ₈ SO ₄	#N/D	2	No transition	No transition	No transition	0.374	0.97	1.344
N _{111Bz(2OH)} Cl	1	60	38	-32	2.9	0.469	0.77	1.239
C ₄ C ₁ pyrr Lactate	19	31	58	-12	24.6	0.755	-0.989	-0.234
C ₄ C ₁ pyrr Cl	#N/D	23	49	-21	12.4	0.755	0.77	1.525
C ₃ C ₁ im C ₃ SO ₃	12	44	54	-16	27.8	0.898	-0.245	0.653
C ₃ C ₁ im C ₂ SO ₃	13	36	61	-9	15.0	0.898	-0.768	0.13
C ₃ C ₁ im Cl	20	38	43	-27	12.0	0.898	0.77	1.668
C ₃ C ₁ im C ₄ SO ₃	35	55	45	-25	3.5	0.898	0.212	1.11
C ₄ C ₁ pip Cl	36	45	No transition	No transition	No transition	1.211	0.77	1.981
C ₄ C ₁ im N(CN) ₂	2	78	50	-20	1.0	1.354	0.069	1.423
C ₄ C ₁ im C ₂ SO ₃	6	44	53	-17	3.4	1.354	-0.768	0.586
C ₄ C ₁ im Cl	10	42	No transition	No transition	No transition	1.354	0.77	2.124
C ₄ C ₁ im C ₁ COO	18	29	58	-12	16.6	1.354	-0.908	0.446
C ₆ C ₁ pyrr Cl	#N/D	34	No transition	No transition	No transition	1.667	0.77	2.437
C ₅ C ₁ im Cl	20	78	No transition	No transition	No transition	1.81	0.77	2.58
C ₆ C ₁ im C ₁ SO ₃	2	69	No transition	No transition	No transition	2.267	-1.117	1.15
C ₆ C ₁ im Cl	6	96	No transition	No transition	No transition	2.267	0.77	3.037
C ₆ C ₁ im C ₂ SO ₃	19	59	No transition	No transition	No transition	2.267	-0.768	1.499
C ₈ C ₁ im Cl	2	96	45	-25	5.9	3.179	0.77	3.949
N ₁₁₁₄₈ Cl	3	66	No transition	No transition	No transition	3.448	0.77	4.218
N _{888H} C ₇ COO	12	78	48	-22	13.7	8.22	2.04	10.26
N _{888H} C ₁ COO	56	15	36	-34	70.4	8.22	-0.908	7.312
N ₈₈₈₂₁ C ₂ SO ₄	3	100	41	-29	4.8	5.272	-0.922	4.35
N _{888H} (2-C ₄)C ₇ COO	39	28	72	2	55.7	8.22	3.871	12.091
C ₄ C ₁ im OTf	2	85	No transition	No transition	No transition	1.354	1.019	2.373

#N/D = not determined

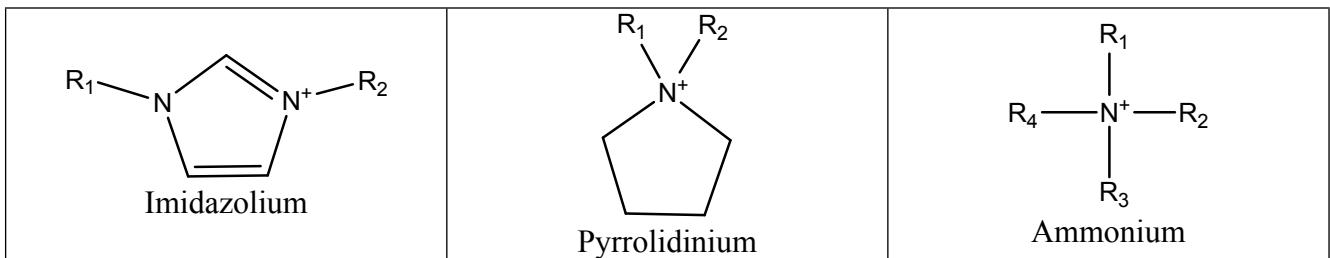
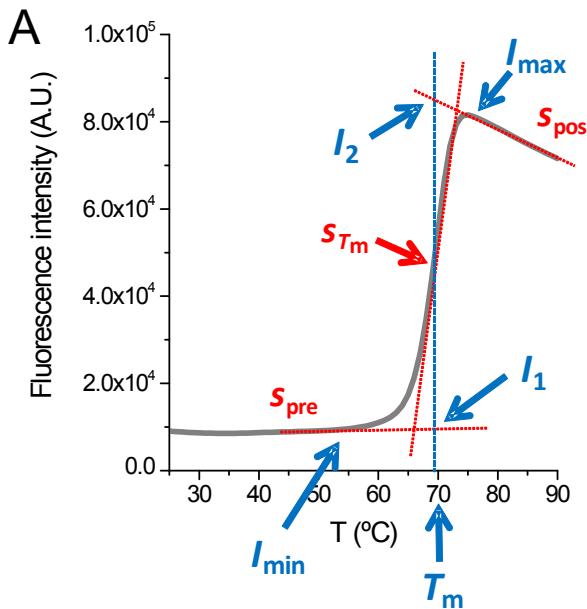


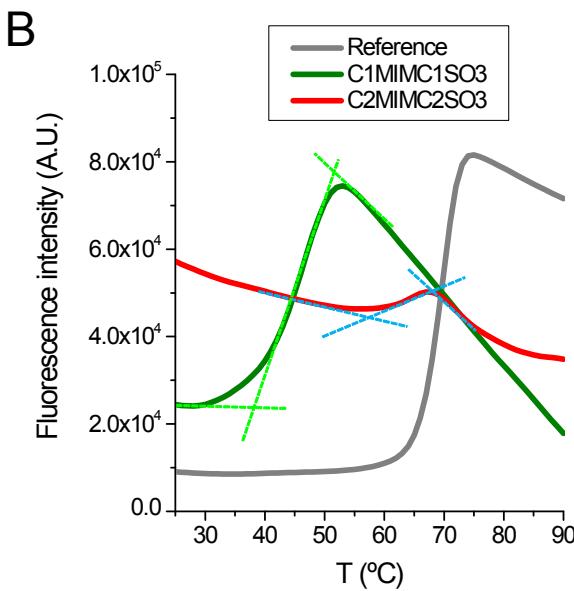
Figure S1 –Main cation structure of ILs



$$\text{Transition steepness} = \frac{s_{T_m}}{(I_2 - I_1)}$$

$$I_2 = I_{\max} - s_{\text{pos}} \times (T_{I_{\max}} - T_m)$$

$$I_1 = I_{\min} - s_{\text{pre}} \times (T_{I_{\min}} - T_m)$$



C

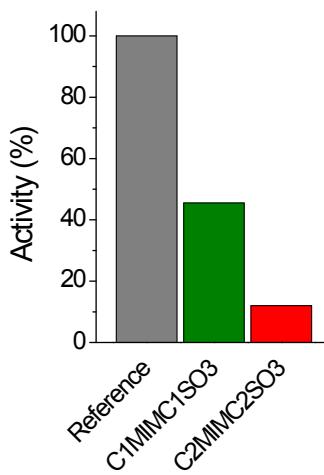


Figure S2 –Examples of DSF curves and parameter determination. A) The melting temperature (T_m) is the temperature at midpoint transition, and the transition steepness is the slope of the intensity change at T_m , normalized by the amplitude of the transition, which is calculated by the difference in maximum and minimum intensities extrapolated through T_m using slopes for the post- and pre-transitions, respectively. B) Different types of curves obtained by DSF using distinct solvents and C) activity data for the corresponding examples.

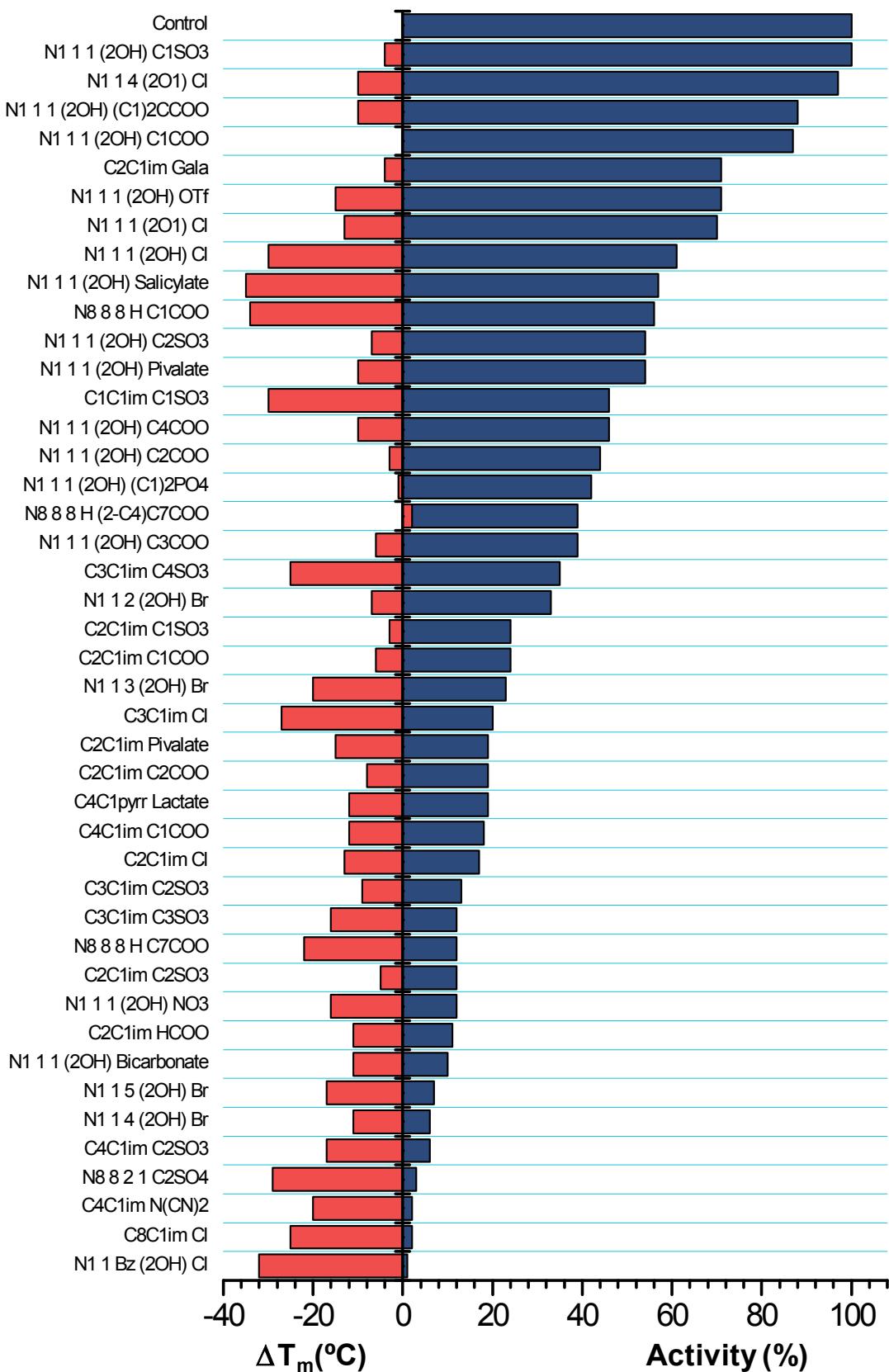


Figure S3 –Comparison between the activity of *TIL* and the T_m determined by DSF

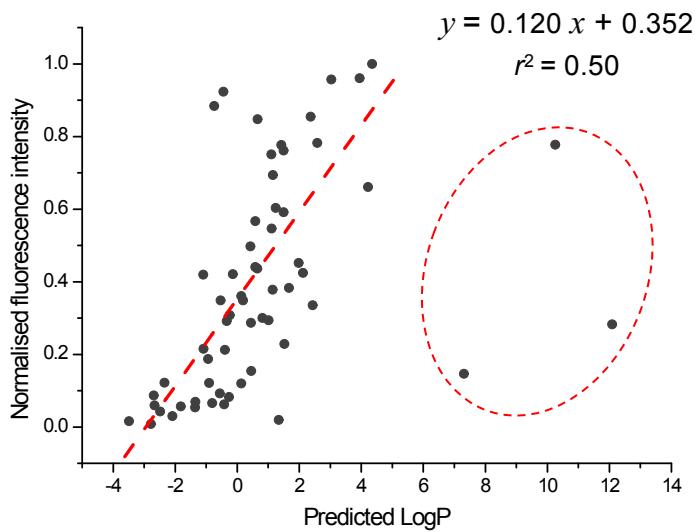


Figure S4 – Effect of the hydrophobic character of IL on the fluorescence intensity of Sypro Orange in the corresponding solvent condition.

5

10

15

20

25

30

35

40

Table S2 – List of ionic liquids used in this work and source

Ionic liquids	Source
(CH ₃) ₂ N ₂ CNH ₂ N(CN) ₂	GVSM Carrera, RFM Frade, J Aires de Sousa, CAM Afonso, LC Branco, <i>Tetrahedron</i> 2010, 12 , 643
N _{111(2OH)} Bitartrate	Sigma Aldrich
N _{111(2OH)} H ₂ Citrate	Sigma Aldrich
N _{111(2OH)} Bicarbonate	Sigma Aldrich
N _{111(2OH)} NO ₃	Iolitec
N _{111(2OH)} C ₃ COO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} (C ₁) ₂ PO ₄	Iolitec
N _{111(2OH)} C ₂ COO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} C ₄ COO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} C ₂ SO ₃	A.J.L. Costa, M.R.C. Soromenho, K. Shimizu, J.M.S.S. Esperança, J.N. Canongia Lopes, L.P.N. Rebelo, <i>EUCHEM</i> 2012, Wales, UK, August 5-10, 2012
N _{111(2OH)} Pivalate	S Shahriari, LC Tomé, JMM Araújo, LPN Rebelo, JAP Coutinho, IM Marrucho, MG Freire, <i>RSC Adv</i> 2013, 3 , 183
N _{111(2OH)} Salicylate	Sigma Aldrich
N _{111(2OH)} Cl	Sigma Aldrich
N _{111(2OH)} OTf	Iolitec
N _{111(2OH)} C ₁ COO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} (C ₁) ₂ CCOO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} C ₁ SO ₃	A.J.L. Costa, M.R.C. Soromenho, K. Shimizu, J.M.S.S. Esperança, J.N. Canongia Lopes, L.P.N. Rebelo, <i>EUCHEM</i> 2012, Wales, UK, August 5-10, 2012
N _{111(2OH)} C ₅ COO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
N _{111(2OH)} H ₂ PO ₄	Iolitec
N _{112(2OH)} Br	Sigma Aldrich
N ₁₁₁₍₂₀₁₎ Cl	AJL Costa, P Papis, PM Reis, K Shimizu, J Szydłowski, JN Canongia Lopes, JMSS Esperança, LPN Rebelo, <i>COIL-5</i> , Vilamoura, Portugal, April 21-25, 2013
N _{113(2OH)} Br	AJL Costa, MRC Soromenho, K Shimizu, IM Marrucho, JMSS Esperança, JN Canongia Lopes, LPN Rebelo, <i>ChemPhysChem</i> 2012, 13 , 1902
C ₂ OH ₂ C ₁ im Cl	Iolitec
N ₁₁₃₍₂₀₁₎ Cl	AJL Costa, P Papis, PM Reis, K Shimizu, J Szydłowski, JN Canongia Lopes, JMSS Esperança, LPN Rebelo, <i>COIL-5</i> , Vilamoura, Portugal, April 21-25, 2013
N _{114(2OH)} Br	AJL Costa, MRC Soromenho, K Shimizu, IM Marrucho, JMSS Esperança, JN Canongia Lopes, LPN Rebelo, <i>ChemPhysChem</i> 2012, 13 , 1902
C ₁ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
N _{115(2OH)} Br	AJL Costa, MRC Soromenho, K Shimizu, IM Marrucho, JMSS Esperança, JN Canongia Lopes, LPN Rebelo, <i>ChemPhysChem</i> 2012, 13 , 1902
C ₂ C ₁ im N(CN) ₂	Iolitec
C ₂ C ₁ im OTf	Iolitec
C ₂ C ₁ im SCN	Iolitec
C ₂ C ₁ im HCOO	M Petkovic, JL Ferguson, HQ Nimal Gunaratne, R Ferreira, MC Leitão, KR Seddon, LPN Rebelo, C Silva Pereira, <i>Green Chem</i> 2010, 12 , 643
C ₂ C ₁ im Br	Iolitec
C ₂ C ₁ im C ₂ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₂ C ₁ im Cl	Iolitec
C ₂ C ₁ im Pivalate	It was prepared by several of the authors in accordance with the protocol cited in J Blath, N Deubler, T Hirth, T Schiestel, <i>Chem Eng J</i> 2012, 181 , 152
C ₂ C ₁ im C ₂ COO	B Zhao, L Greiner, W Leitne, <i>RSC advances</i> 2012, 2 , 2476
C ₂ C ₁ im C ₁ COO	Sigma Aldrich
C ₂ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₂ C ₁ im C ₆ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₂ C ₁ im C ₄ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₂ C ₁ im Galactate	They were prepared by some of the authors in QUILL centre (Queen's University Ionic Liquids Laboratory, Belfast, UK).
C ₂ C ₁ im C ₂ SO ₄	Merck
C ₂ C ₁ im C ₄ SO ₄	Merck
C ₂ C ₁ im C ₈ SO ₄	Merck
N _{11Bz(2OH)} Cl	Sigma Aldrich
C ₄ C ₁ pyrr Lactate	They were prepared by some of the authors in QUILL centre (Queen's University Ionic

	Liquids Laboratory, Belfast, UK).
C ₄ C ₁ pyrr OTf	Iolitec
C ₆ C ₁ pyrr Cl	Iolitec
C ₃ C ₁ im C ₃ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₃ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₃ C ₁ im C ₂ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₃ C ₁ im Cl	Iolitec
C ₃ C ₁ im C ₄ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₂ py Cl	Iolitec
C ₄ C ₁ pip Cl	Iolitec
C ₄ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₄ C ₁ im N(CN) ₂	Iolitec
C ₄ C ₁ im C ₂ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₄ C ₁ im BF ₄	Iolitec
C ₄ C ₁ im Cl	Iolitec
C ₄ C ₁ im Alalinate	Ohno H, Fukumoto K, <i>Acc Chem Res</i> 2007, 40 , 1122
C ₄ C ₁ im C ₁ COO	Sigma Aldrich
C ₆ C ₁ pyrr Cl	Iolitec
C ₅ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₅ C ₁ im Cl	They were prepared by some of the authors in QUILL centre (Queen's University Ionic Liquids Laboratory, Belfast, UK).
C ₄ py Cl	Iolitec
C ₆ C ₁ im C ₁ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₆ C ₁ im Cl	Iolitec
C ₆ C ₁ im C ₂ SO ₃	Blesic et al. <i>Phys Chem Chem Phys</i> 2009, 11 , 8939
C ₈ C ₁ pyrr Cl	They were prepared by some of the authors in QUILL centre (Queen's University Ionic Liquids Laboratory, Belfast, UK).
C ₄ C ₁ py Cl	Iolitec
C ₈ C ₁ im Cl	They were prepared by some of the authors in QUILL centre (Queen's University Ionic Liquids Laboratory, Belfast, UK).
N ₁₁₄₈ Cl	Cheng Chen, Guang Xian Zhang, Feng Xiu Zhang, Hui Zheng, <i>Adv Materials Res</i> 2012, 549 , 278
C ₆ C ₁ py Cl	Iolitec
C ₆ C ₁ py C ₁ SO ₃	AB Pereiro, A Rodríguez, M Blesic, K Shimizu, JNC Lopes, LPN Rebelo, <i>J Chem Eng Data</i> 2011, 56 , 4356
C ₈ C ₁ py Cl	AB Pereiro, A Rodríguez, M Blesic, K Shimizu, JNC Lopes, LPN Rebelo, <i>J Chem Eng Data</i> 2011, 56 , 4356
N _{888H} C ₇ COO	Bioniqs
N _{888H} C ₁ COO	Bioniqs
P ₄₄₄₂ (C ₂) ₂ PO ₄	Cytec
N ₈₈₂₁ C ₂ SO ₄	Bioniqs
N _{888H} (2-C ₄)C ₇ COO	Bioniqs
C ₄ C ₁ im OTf	Iolitec