

Supporting Information

The D-glucosamine-derived pyridyl-triazole@palladium recoverable catalyst for Mizoroki-Heck reactions under solvent-free conditions

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1. General Information

a. Materials

All reagents were commercially available and used without purification, unless otherwise noted. Aryl halides and sulfinic acid salts were purchased from Alfa Aesar. Other chemicals were obtained commercially and used without any prior purification. ^1H NMR spectra were recorded on a Bruker AvanceII 400 spectrometer using TMS as the internal standard. All products were isolated by short chromatography on a silica gel (200–300 mesh) column using petroleum ether (60–90 °C), unless otherwise noted. The pivaloylated sugar substrates were prepared according to our previous reports.¹ All compounds were characterized by ^1H NMR, ^{13}C NMR and mass spectroscopy, which are consistent with those reported in the literature.²⁻⁵

b. Methods

Melting points were determined on an X-5 Data microscopic melting point apparatus. ^1H NMR and ^{13}C NMR spectra were recorded on a Bruker Advance 400 spectrometer at ambient temperature with CDCl_3 or $\text{DMSO-}d_6$ as solvent unless otherwise noted and tetramethylsilane (TMS) as the internal standard. Mass spectra (GC-MS) were acquired on an Agilent 5975 spectrometer. IR spectra were recorded on a Nicolet 380 FT-IR spectrophotometer using KBr discs. Scanning electron microscope (SEM) images were collected on FEI XL40 instrument. Transmission electron microscopy (TEM) images were taken on FEI T20 microscope. The small-angle X-ray diffraction (SAXRD) data were taken on a German Bruker D4 X-ray diffractometer with Niltered Cu Ka radiation (40 kV, 40 mA). Thermogravimetric analyses were performed with a SII Nano Technology EXTAR TG/DTA7220 thermal analyzer at 10 $^\circ\text{C}/\text{min}$ in nitrogen atmosphere (10 ml/min). 5 mg of each sample in an alumina pan was analyzed in the 40-900 $^\circ\text{C}$ temperature range. Analytical thin layer chromatography (TLC) was performed on Merk precoated TLC (silica gel 60 F254) plates.

2. Experimental Section

General procedure for the Mizoroki-Heck reaction in the presence of the catalyst (5)

To a flask, a mixture of D-glucosamine-derived triazole@palladium catalyst **5** (0.05 g of the catalyst, 0.1 mol%), aryl halide (1 mmol), olefin (2 mmol) and Et_3N (3 mmol) were added and heated at 80 $^\circ\text{C}$ under solvent-free conditions. After completion of the reaction, ethylacetate (10 mL) was added to the flask. The catalyst was separated by simple filtration. Water (3×15 mL) was added to the ethylacetate phase and decanted. The organic layer was dried over anhydrous Na_2SO_4 . After evaporation of the solvent, the resulted crude products was purified by column chromatography (hexane/ethylacetate) giving the pure products in excellent yields.

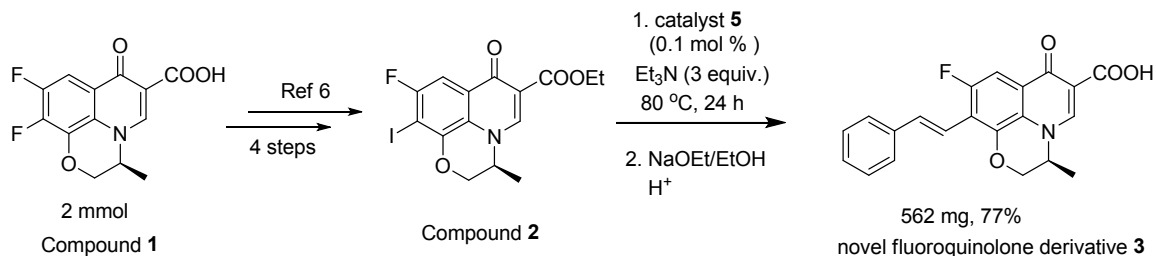
Recycling of the catalyst in Mizoroki-Heck reaction

After completion of the reaction at the first run, the reaction mixture was cooled down to room temperature and ethylacetate (5 mL) was added to the reaction mixture to extract organics. The ethylacetate phase was sucked from the vial by a syringe and the catalyst was dried under vacuum. After complete drying, the catalyst was reused for the similar reaction. This process was repeated for five runs.

General procedure for synthesis of Axitini under solvent-free condition.

To a flask, a mixture of intermediate **11** (1 mmol), 2-vinylpyridine **7e** (2 mmol), Et₃N (3 mmol) and D-glucosaminederived triazole@palladium catalyst **5** (0.1 mol%) were added and heated at 100 °C under solvent-free conditions. The reaction progress was monitored by TLC. After completion of the reaction, the mixture was allowed to cool to room temperature, ethylacetate (10 mL) was added to the flask. The catalyst was separated by simple filtration and the aqueous phase was extracted with CH₂Cl₂ for 3 times (3 × 2 mL). Then the combined organic layers were dried over anhydrous Na₂SO₄, concentrated under vacuum and purified by column chromatography (hexane/ethyl acetate 10:1) to afford the desired product. Residual Pd content in solvent was determined to be not more than 20 ppm by atomic absorption spectroscopy.

General procedure for synthesis of novel fluoroquinolone derivative under solvent-free condition.



Scheme S1

To a flask, a mixture of compound **2** (2 mmol) (which was prepared from commercially available compound **1** by 4 steps⁶), styrene (4 mmol), Et₃N (6 mmol) and D-glucosamine

derived triazole@palladium catalyst **5** (0.2 mol%) were added and heated at 100 °C under solvent-free conditions. The reaction progress was monitored by TLC. After completion of the reaction, the mixture was allowed to cool to room temperature and the catalyst was separated by simple filtration. Then 10 mL NaOEt/EtOH solution was added and the mixture was stirred for 3 h. Upon completion, 1 M HCl was added until the solution was at neutral pH, and was filtered. The solution was concentrated in vacuo and purified by flash chromatography 7:1 ethyl acetate/MeOH to give the novel fluoroquinolone derivative **3** (562 mg, 77%)

3. Characterization of the Catalysts and Prouducts

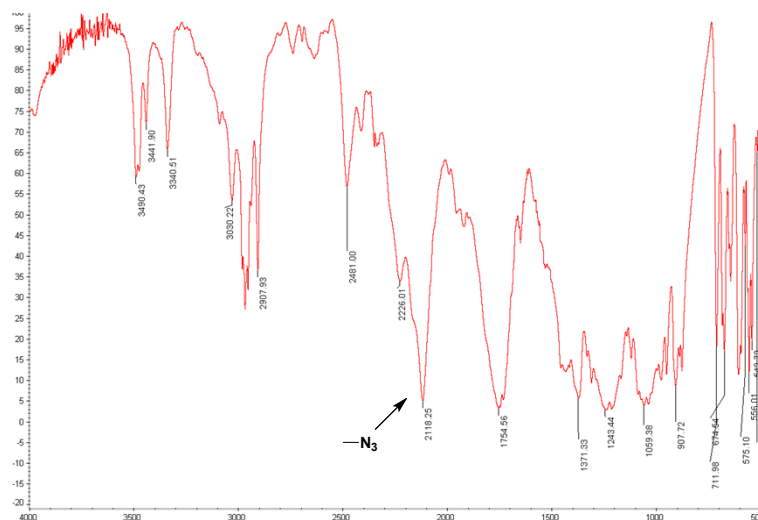


Figure S1. FT-IR spectrum of D-glucosamine-based azide (3).

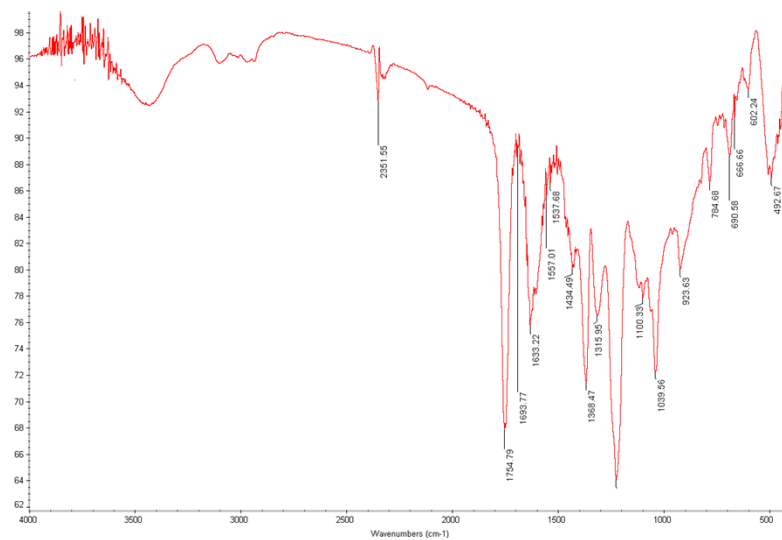


Figure S2. FT-IR spectrum of D-glucosamine-derived triazole(4) .



Figure S3. FT-IR spectrum of catalyst(5) .

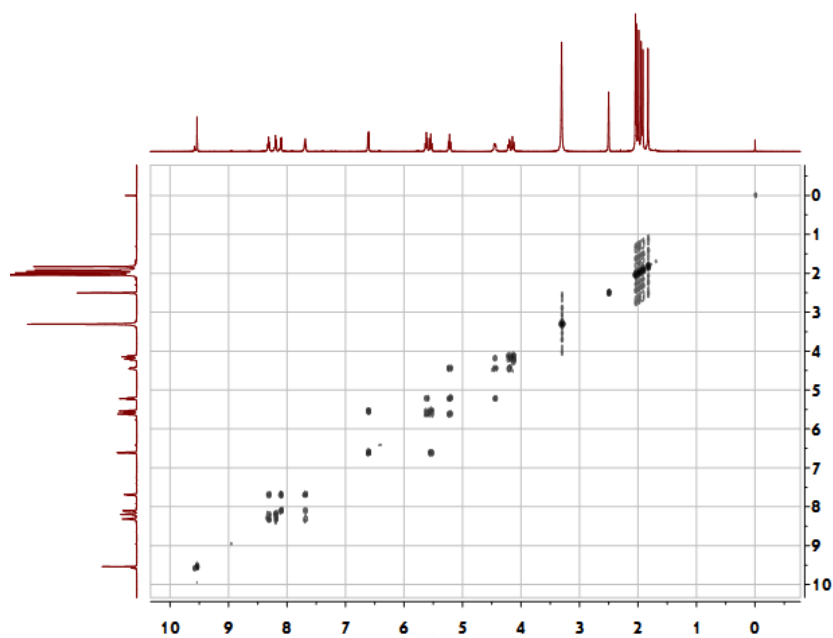
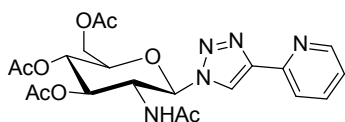


Figure S4. 2D ROESY NMR spectrum of Pd catalyst 5.

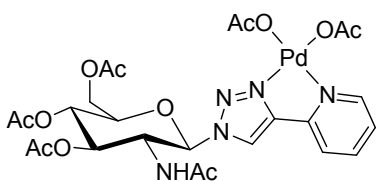
D-glucosamine-derived triazole(4):



Obtained as a white solid in 92% yield; M.p. 214-215 °C. ¹H

NMR (500 MHz, CDCl₃): δ 9.54 (s, 1H), 8.32 (t, *J* = 6.0 Hz, 1H), 8.20 (t, *J* = 6.0 Hz, 1H), 8.11 (d, *J* = 4.0 Hz, 1H), 7.69 (t, *J* = 5.2 Hz, 1H), 6.62 (d, *J* = 7.2 Hz, 1H, G₁H), 5.63 (t, *J* = 7.2 Hz, 1H, G₂H), 5.55 (t, *J* = 7.2 Hz, 1H, G₄H), 5.23 (t, *J* = 7.6 Hz, 1H, G₃H), 4.46-4.44 (m, 1H, G₅H), 4.23-4.13 (m, 2H, G₆H), 2.05-1.84 (m, 19H). ¹³C NMR (100 MHz, CDCl₃): δ 170.5, 170.0, 169.8, 150.2, 137.7, 122.7, 84.6, 73.8, 72.5, 70.7, 68.1, 20.9, 20.8, 20.7, 20.4. Anal. calcd. for C₂₁H₂₅N₅O₈: C, 53.05; H, 5.30; N, 14.73; found: C, 53.12; H, 5.38; N, 14.77.

D-glucosamine-derived triazoles@ Pd catalyst 5

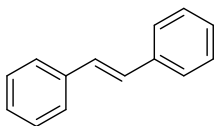


Obtained as a white solid in 90% yield; M.p. 289-291 °C.

¹H NMR (500 MHz, CDCl₃): δ 9.54 (s, 1H), 8.32 (t, *J* = 6.0 Hz, 1H), 8.20 (t, *J* = 6.0 Hz, 1H), 8.11 (d, *J* = 4.0 Hz, 1H), 7.69 (t, *J* = 5.2 Hz, 1H), 6.62 (d, *J* = 7.2 Hz, 1H, G₁H), 5.63 (t, *J* = 7.2 Hz, 1H, G₂H), 5.55 (t, *J* = 7.2 Hz, 1H, G₄H), 5.23 (t, *J* = 7.6 Hz, 1H, G₃H), 4.46-4.44 (m, 1H, G₅H), 4.23-4.13 (m, 2H, G₆H), 2.05-1.84 (m, 19H). ¹³C NMR (100 MHz, CDCl₃): δ 170.5, 170.0, 169.8, 150.2, 137.7, 122.7, 84.6, 73.8, 72.5, 70.7, 68.1, 20.9,

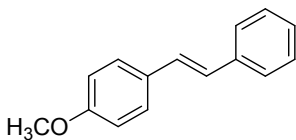
20.8, 20.7, 20.4. Anal. calcd. for C₂₅H₃₁N₅O₁₂Pd: C, 42.90; H, 4.46; N, 10.01; found: C, 42.93; H, 4.42; N, 10.05.

(E)-1,2-diphenylethene 8a²:



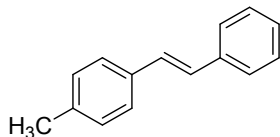
¹H NMR (500 MHz, CDCl₃): δ 7.58 (d, *J* = 7.5 Hz, 4H), 7.43 (dd, *J* = 9.8, 5.5 Hz, 4H), 7.36 – 7.29 (m, 2H), 7.18 (d, *J* = 1.3 Hz, 2H). GC-MS (EI) [M]⁺: *m/z* calcd. for C₁₄H₁₂: 180.0, found: 180.

(E)-1-methoxyl-4-styrylbenzene 8b³:



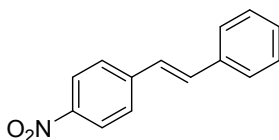
¹H NMR (500 MHz, CDCl₃): δ 7.50 – 7.40 (m, 4H), 7.33 (dd, *J* = 10.6, 4.7 Hz, 2H), 7.21 (t, *J* = 3.7 Hz, 1H), 7.05 (d, *J* = 16.3 Hz, 1H), 6.96 (d, *J* = 16.3 Hz, 1H), 6.90 – 6.85 (m, 2H), 3.80 (s, 3H). GC-MS (EI) [M]⁺: *m/z* calcd. for C₁₅H₁₄O: 210.0, found: 210.

(E)-1-methyl-4-styrylbenzene 8c²:



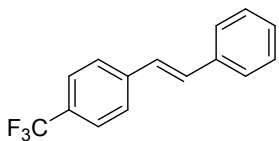
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.60 – 7.54 (m, 2H), 7.51 – 7.45 (m, 2H), 7.45 – 7.38 (m, 2H), 7.32 (dd, $J = 8.3, 5.4$ Hz, 1H), 7.26 – 7.20 (m, 2H), 7.17 – 7.11 (m, 2H), 2.49 – 2.37 (m, 3H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{15}\text{H}_{14}$: 194.0, found: 194.

(E)-1-nitro-4-styrylbenzene 8d²:



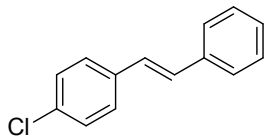
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.24 (d, $J = 8.8$ Hz, 2H), 7.65 (d, $J = 8.7$ Hz, 2H), 7.58 (d, $J = 7.3$ Hz, 2H), 7.43 (t, $J = 7.5$ Hz, 2H), 7.39 – 7.34 (m, 1H), 7.29 (t, $J = 4.4$ Hz, 1H), 7.17 (d, $J = 16.3$ Hz, 1H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{14}\text{H}_{11}\text{NO}_2$: 225.0, found: 225.

(E)-1-styryl-4-(trifluoromethyl)benzene 8e³:



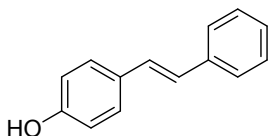
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.65 (q, $J = 8.6$ Hz, 4H), 7.58 (d, $J = 7.3$ Hz, 2H), 7.44 (t, $J = 7.6$ Hz, 2H), 7.37 (dd, $J = 8.2, 6.5$ Hz, 1H), 7.24 (d, $J = 16.3$ Hz, 1H), 7.16 (d, $J = 16.3$ Hz, 1H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{15}\text{H}_{11}\text{F}_3$: 248.0, found: 248.

(E)-1-chloro-4-styrylbenzene 8f⁴:



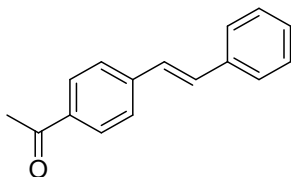
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.55 (d, $J = 7.4$ Hz, 2H), 7.46 (t, $J = 9.4$ Hz, 2H), 7.41 (t, $J = 7.6$ Hz, 2H), 7.37 (d, $J = 8.5$ Hz, 2H), 7.33 (t, $J = 7.3$ Hz, 1H), 7.15 – 7.06 (m, 2H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{14}\text{H}_{11}\text{Cl}$: 214.0, found: 214.

(E)-4-styrylphenol 8g²:



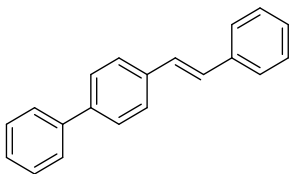
¹H NMR (500 MHz, DMSO): δ 9.55 (d, J = 12.0 Hz, 1H), 7.54 (d, J = 7.3 Hz, 2H), 7.46 – 7.38 (m, 2H), 7.35 (t, J = 7.7 Hz, 2H), 7.26 – 7.18 (m, 1H), 7.18 – 7.09 (m, 1H), 7.02 (d, J = 16.4 Hz, 1H), 6.83 – 6.72 (m, 2H). GC-MS (EI) [M]⁺: m/z calcd. for C₁₄H₁₂O: 196.0, found: 196.

(E)-4-acetyl-4-styrylbenzene 8h³:



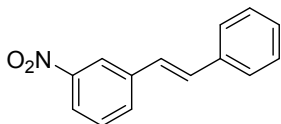
¹H NMR (500 MHz, CDCl₃): δ 7.97 (d, J = 8.4 Hz, 2H), 7.60 (d, J = 8.3 Hz, 2H), 7.56 (d, J = 7.5 Hz, 2H), 7.41 (t, J = 7.6 Hz, 2H), 7.33 (t, J = 7.3 Hz, 1H), 7.25 (d, J = 16.3 Hz, 1H), 7.15 (d, J = 16.3 Hz, 1H), 2.62 (s, 3H). GC-MS (EI) [M]⁺: m/z calcd. for C₁₆H₁₄O: 222.0, found: 222.

(E)-4-phenyl-4-styrylbenzene 8i⁴:



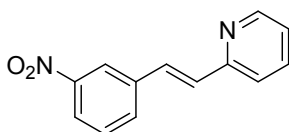
¹H NMR (500 MHz, CDCl₃): δ 7.67 – 7.62 (m, 3H), 7.58 – 7.55 (m, 1H), 7.48 (dd, J = 10.5, 4.9 Hz, 1H), 7.42 – 7.36 (m, 2H), 7.30 (d, J = 7.4 Hz, 1H), 7.18 (s, 1H). GC-MS (EI) [M]⁺: m/z calcd. for C₂₀H₁₆: 256.0, found: 256.

(E)-1-nitro-3-styrylbenzene 8j²:



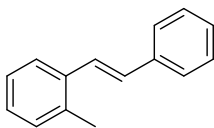
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.39 (s, 1H), 8.12 (dd, $J = 8.2, 1.3$ Hz, 1H), 7.82 (d, $J = 7.7$ Hz, 1H), 7.56 (dd, $J = 15.7, 7.7$ Hz, 3H), 7.47 – 7.39 (m, 2H), 7.35 (t, $J = 7.3$ Hz, 1H), 7.29 (s, 1H), 7.16 (d, $J = 16.3$ Hz, 1H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{14}\text{H}_{11}\text{NO}_2$: 225.0, found: 225.

(E)-1-nitro-3-styrylpyridine 8k:



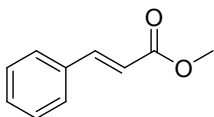
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.66 (d, $J = 4.2$ Hz, 1H), 8.46 (s, 1H), 8.16 (dd, $J = 8.2, 1.2$ Hz, 1H), 7.88 (d, $J = 7.7$ Hz, 1H), 7.74 (dd, $J = 8.8, 7.1$ Hz, 2H), 7.56 (t, $J = 7.9$ Hz, 1H), 7.44 (d, $J = 7.8$ Hz, 1H), 7.29 (d, $J = 3.6$ Hz, 1H), 7.24 (dd, $J = 6.7, 4.9$ Hz, 1H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{13}\text{H}_{10}\text{N}_2\text{O}_2$: 226.0, found: 226.

(E)-1-methyl-2-styrylbenzene 8l2:



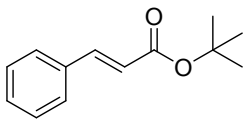
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.65 (d, $J = 7.3$ Hz, 1H), 7.58 (d, $J = 7.3$ Hz, 2H), 7.44 – 7.37 (m, 3H), 7.34 – 7.30 (m, 1H), 7.29 – 7.26 (m, 1H), 7.24 (t, $J = 3.7$ Hz, 2H), 7.06 (d, $J = 16.2$ Hz, 1H), 2.49 (s, 3H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{15}\text{H}_{14}$: 194.0, found: 194.

Methyl cinnamate 8m⁴:



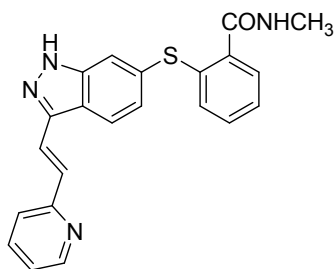
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 7.72 (d, $J = 16.0$ Hz, 1H), 7.57 – 7.49 (m, 2H), 7.44 – 7.35 (m, 3H), 6.47 (d, $J = 16.0$ Hz, 1H), 3.83 (s, 3H). GC-MS (EI) $[\text{M}]^+$: m/z calcd. for $\text{C}_{10}\text{H}_{10}\text{O}_2$: 162.0, found: 162.

tert-butyl cinnamate 8n⁴:



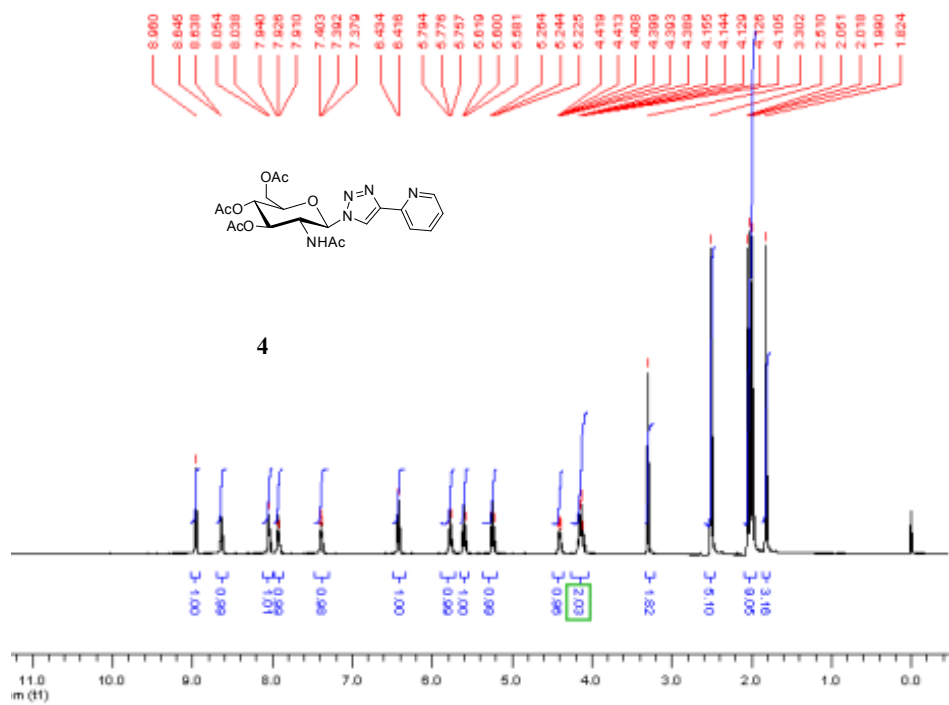
¹H NMR (500 MHz, CDCl₃): δ 7.58 (d, *J* = 16.0 Hz, 1H), 7.51 – 7.48 (m, 2H), 7.37 – 7.33 (m, 3H), 6.36 (d, *J* = 16.0 Hz, 1H), 1.53 (s, 9H). GC-MS (EI) [M]⁺: *m/z* calcd. for C₁₃H₁₆O₂: 204.0, found: 204.

Axitinib⁵

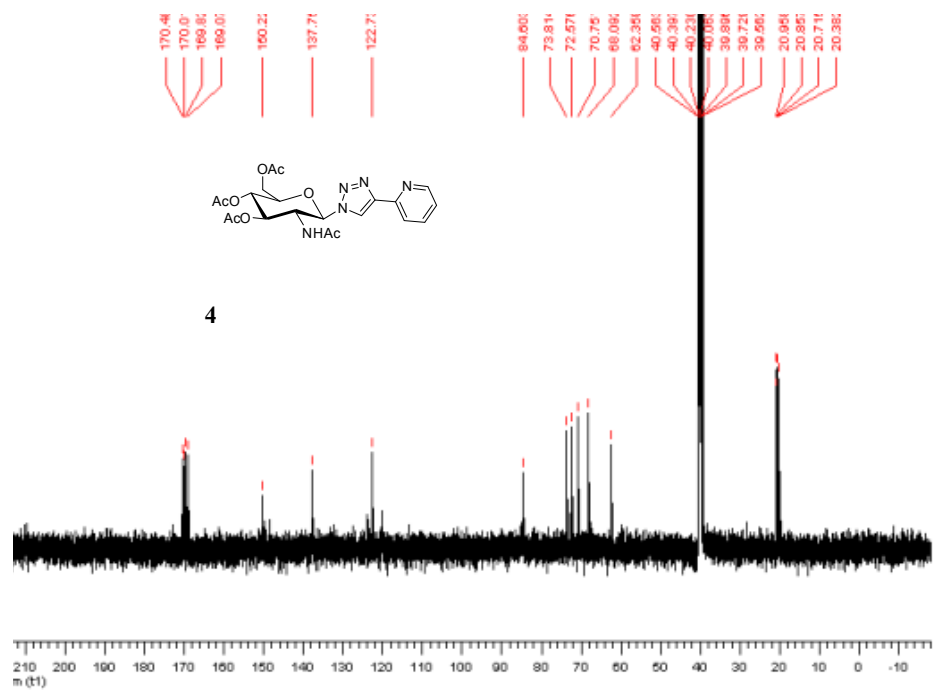


White solid, Mp 225-226 °C. ¹H NMR (500 MHz, DMSO - *d*₆): δ 13.34 (1H, s), 8.61 (d, *J* = 2.0 Hz, 1H), 8.37 (d, *J* = 3.5 Hz, 1H), 8.21 (d, *J* = 8.5 Hz, 1H), 7.97 (d, *J* = 16.5 Hz, 1H), 7.82-7.79 (m, 1H), 7.67 (d, *J* = 7.5 Hz, 1H), 7.61-7.57 (m, 2H), 7.32-7.28 (m, 3H), 7.20 (d, *J* = 9.0 Hz, 1H,), 7.07 (d, *J* = 7.5 Hz, 1H), 2.78 (d, *J* = 3.5 Hz, 3H). ¹³C NMR (75MHz, DMSO-*d*₆): δ 168.33, 155.38, 150.01, 142.49, 142.34, 137.61, 137.30, 135.98, 133.07, 130.73, 130.59, 129.76, 128.25, 126.66, 125.95, 124.11, 123.06, 122.93, 122.18, 120.76, 115.09, 26.54.

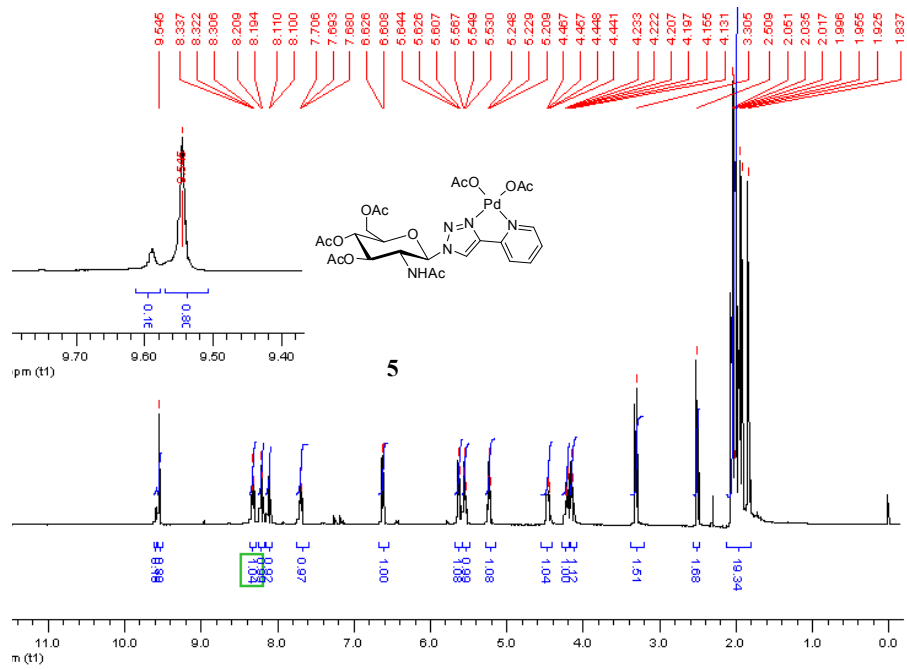
¹H NMR



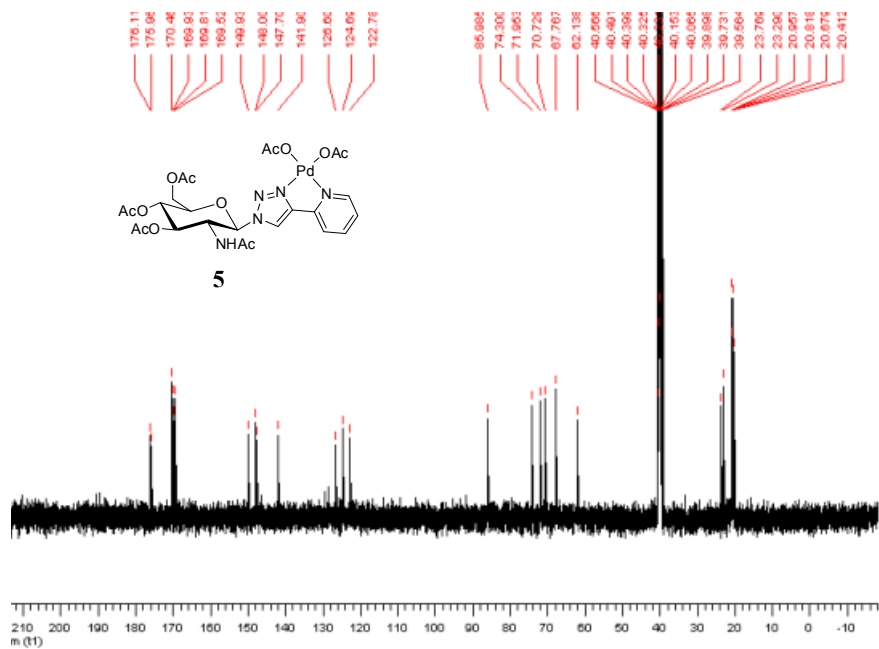
¹³C NMR



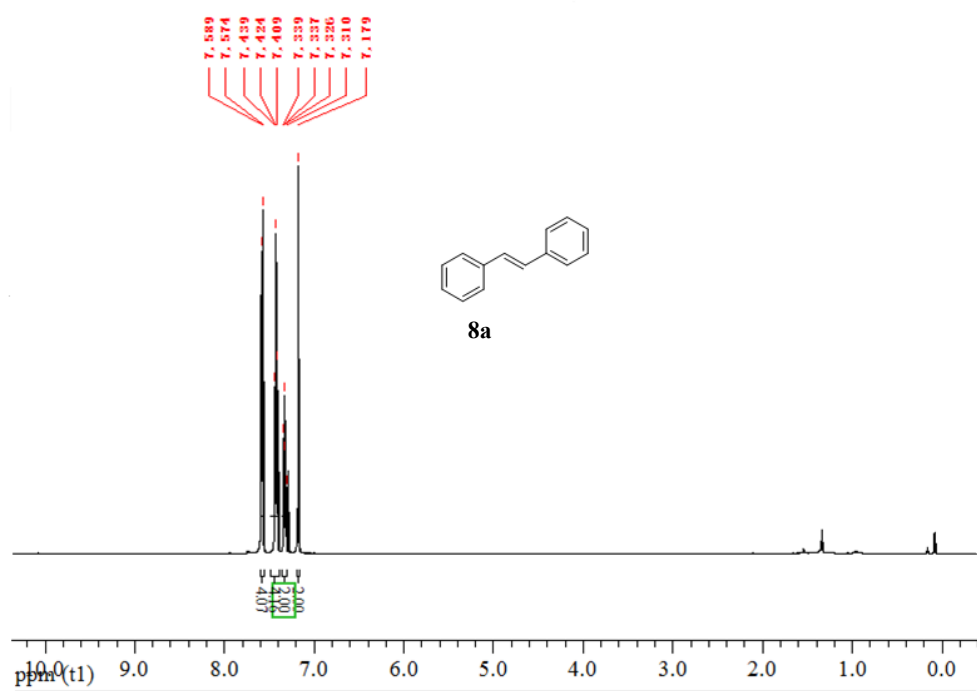
¹H NMR



¹³C NMR

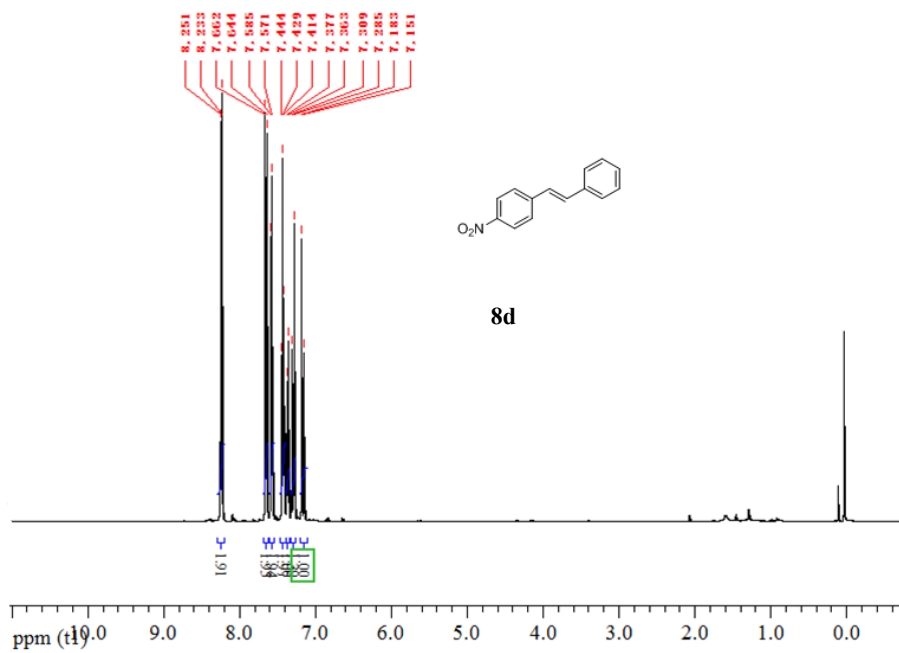


$^1\text{H NMR}$

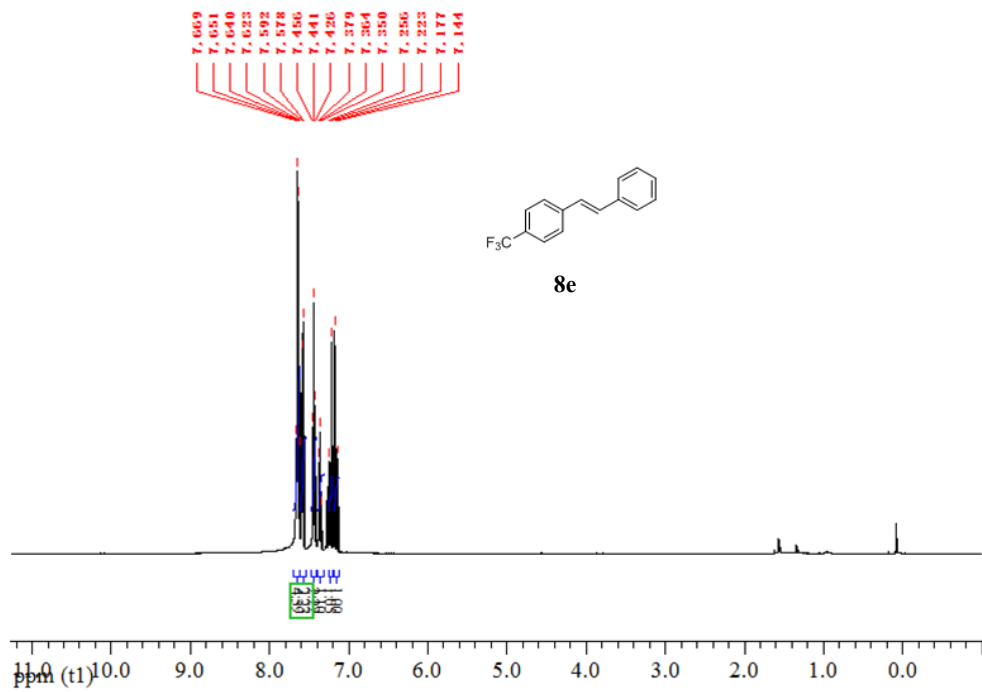


$^1\text{H NMR}$

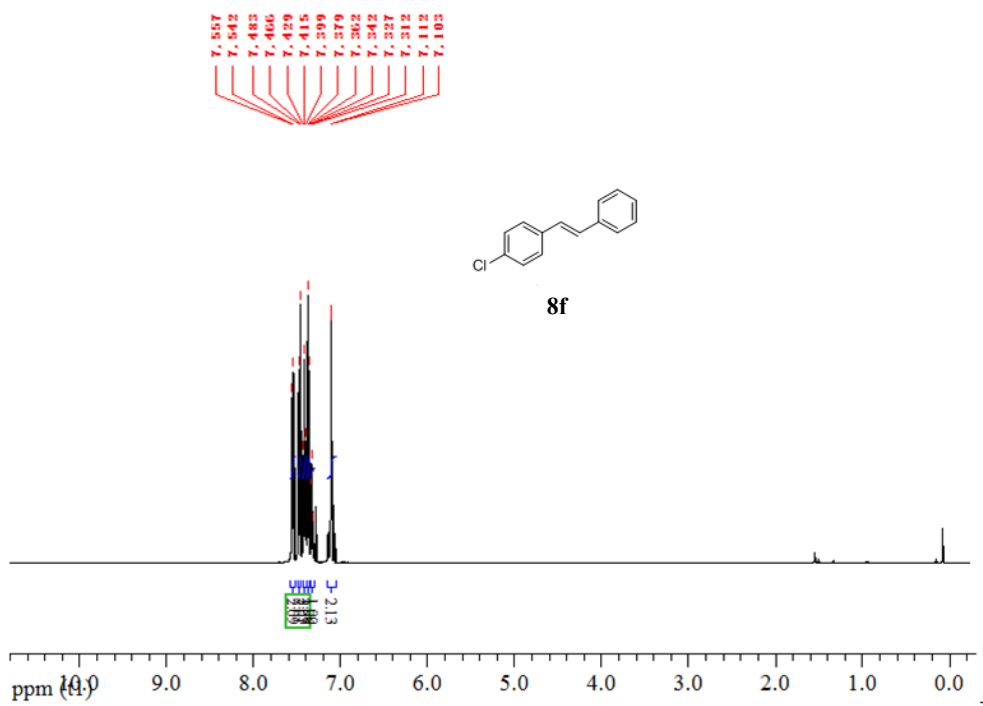
^1H NMR



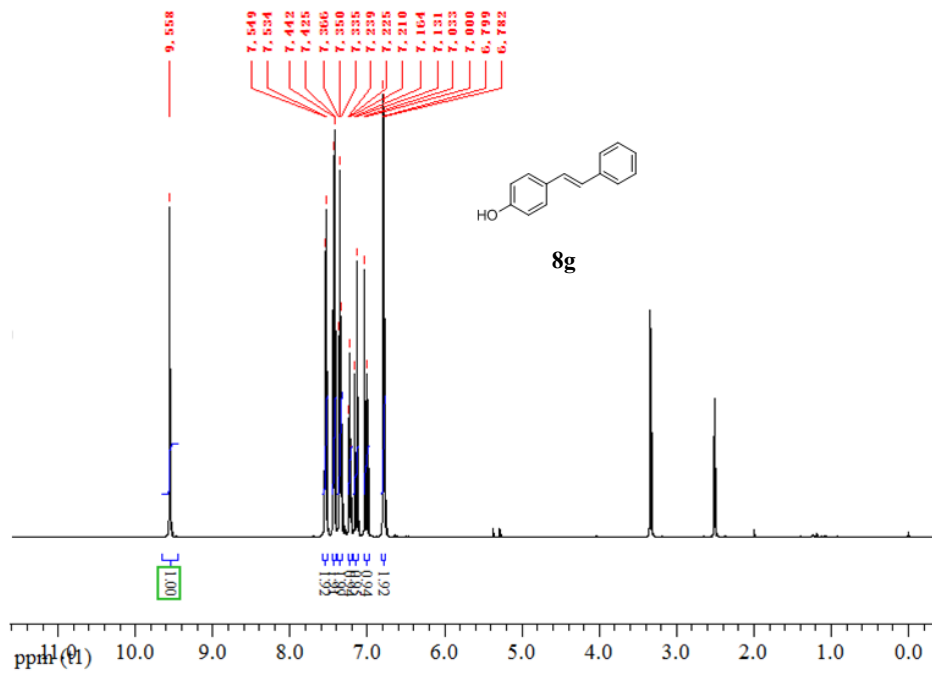
^1H NMR



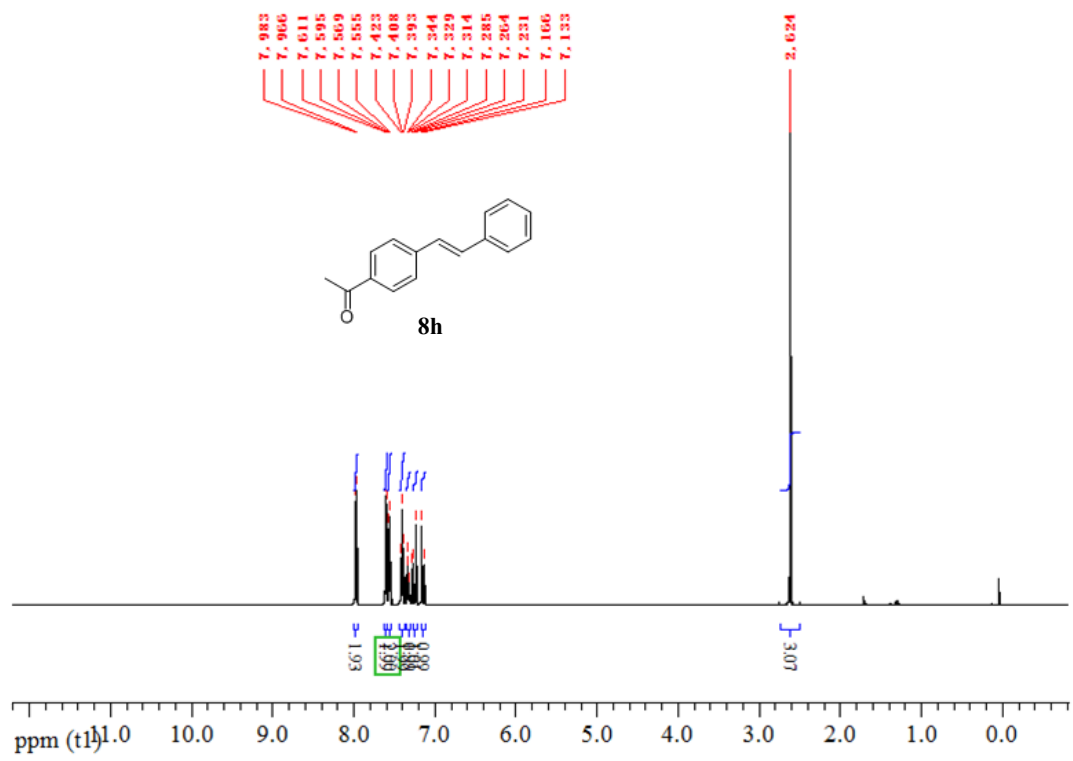
¹H NMR



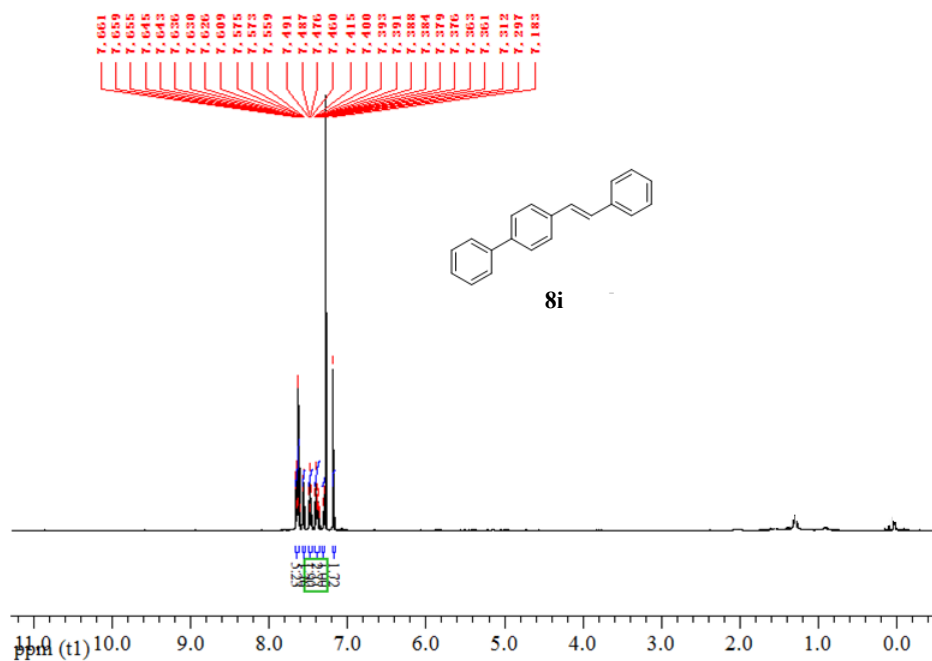
^1H NMR



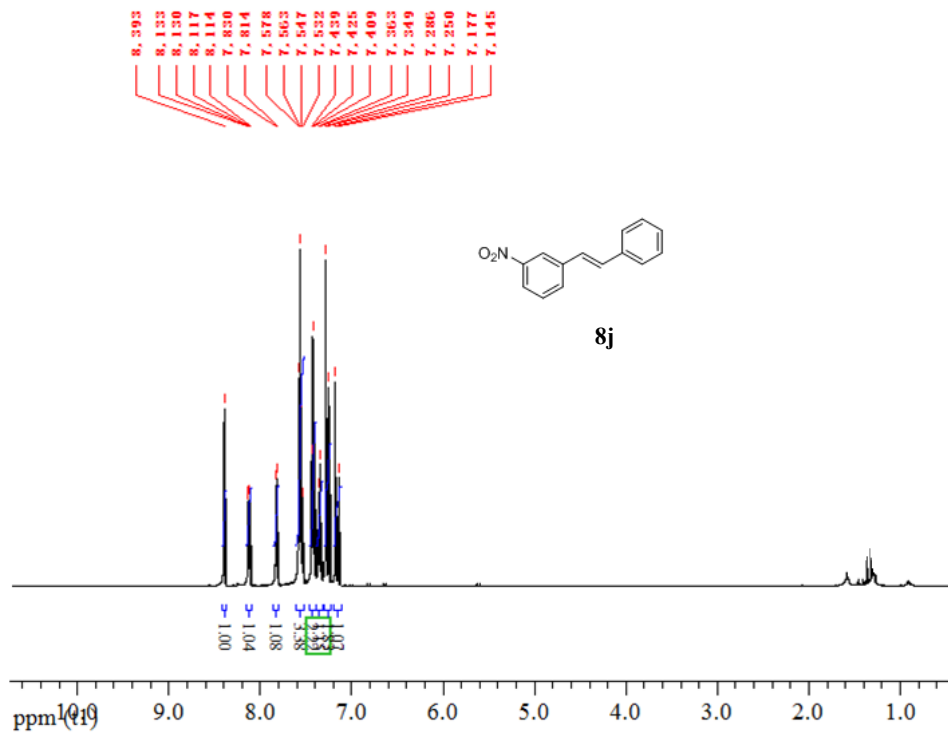
^1H NMR



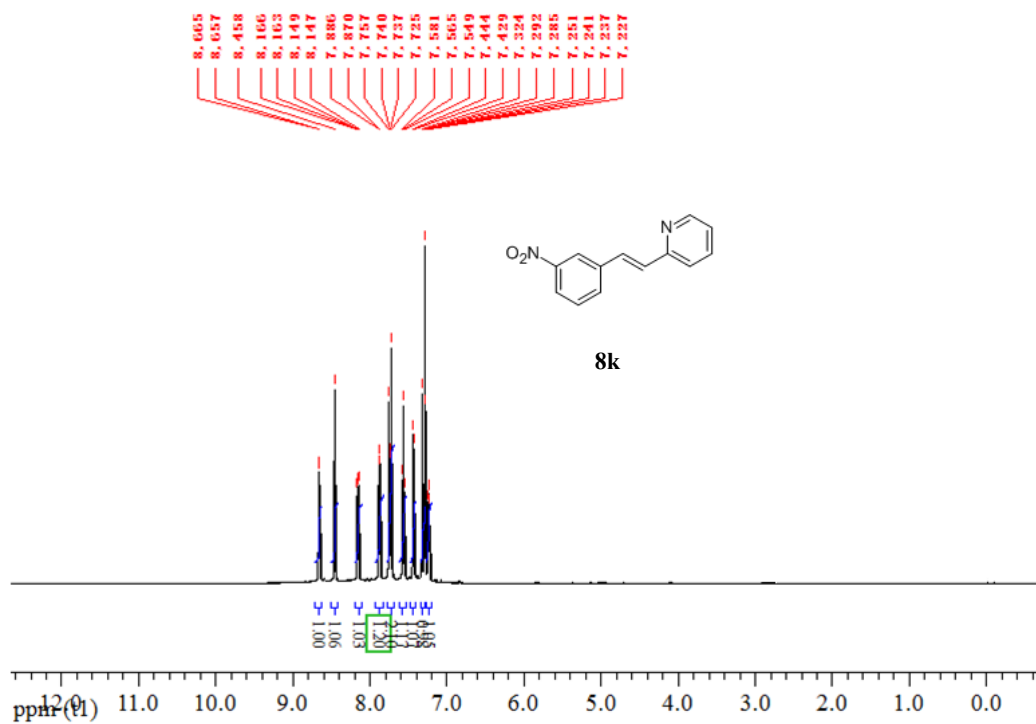
¹H NMR



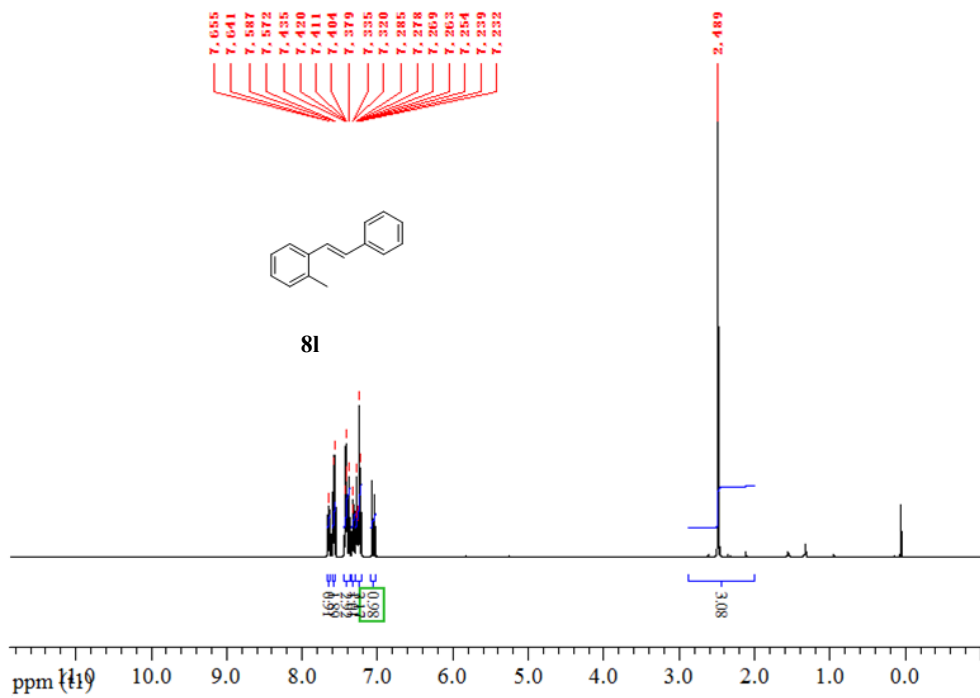
¹H NMR



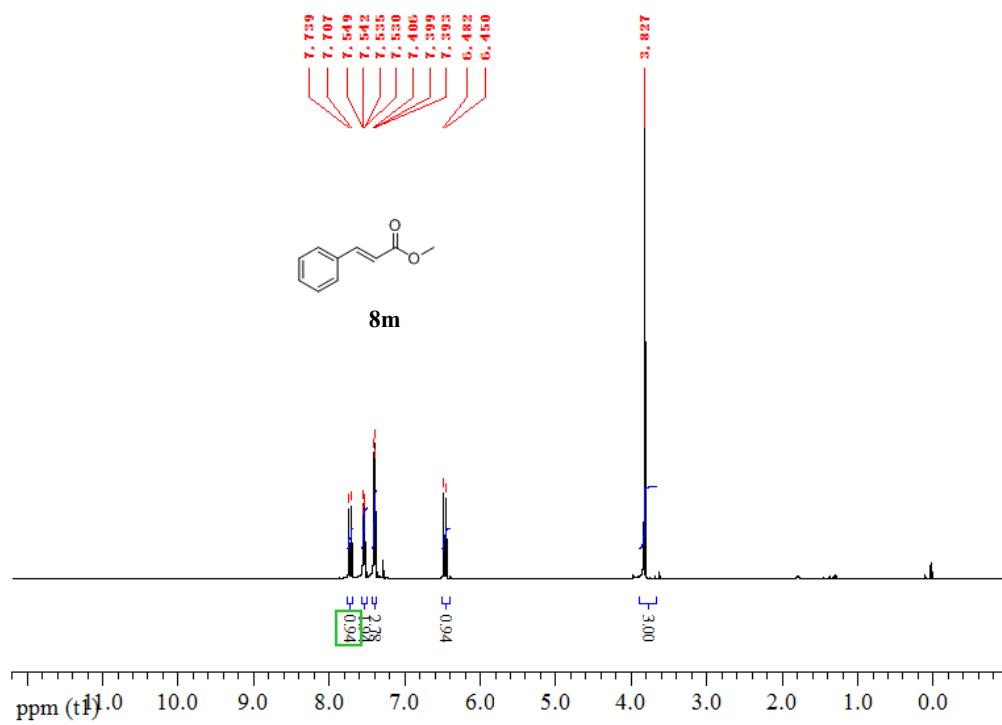
^1H NMR



^1H NMR

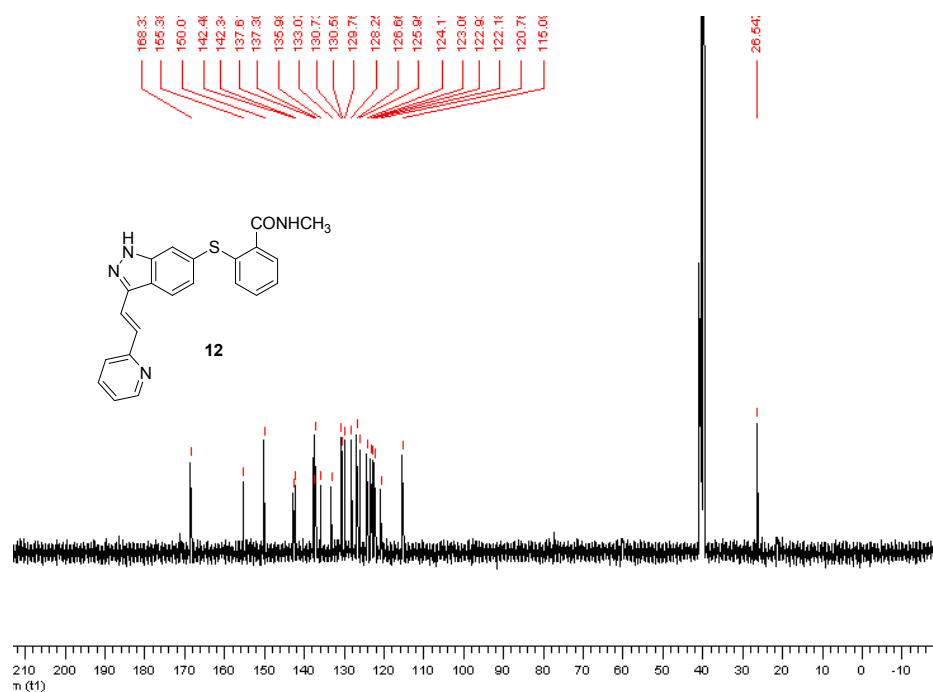


¹H NMR



¹H NMR

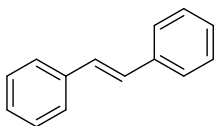
^{13}C NMR

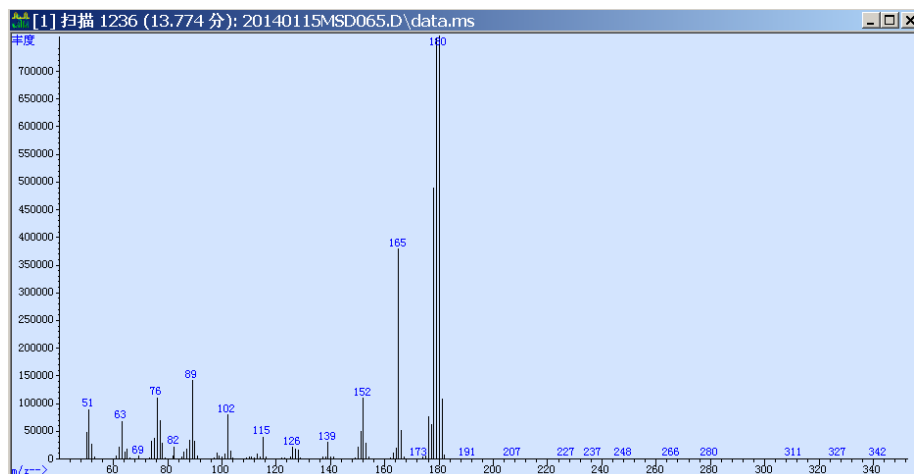


The selected GC-MS chromatogram of products:

(E)-1,2-diphenylethene 8a:

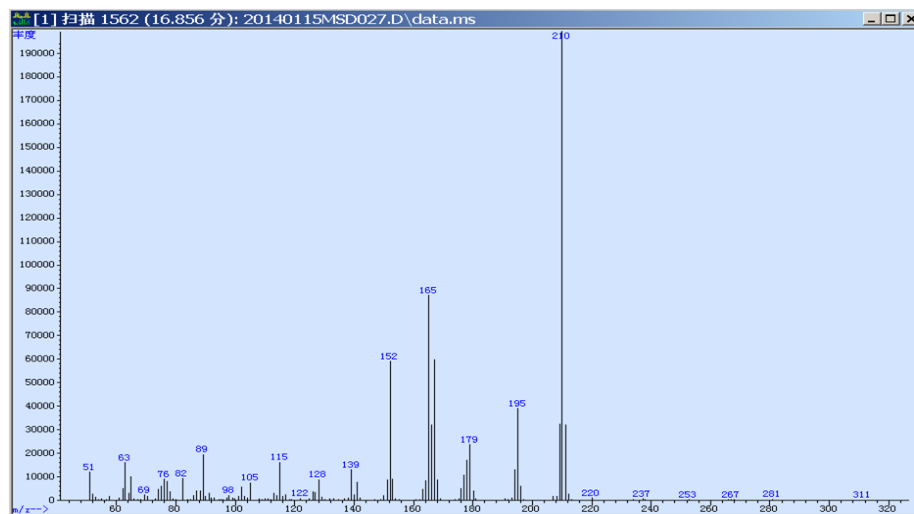
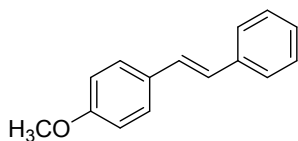
GC-MS (EI) $[M]^+$: m/z calcd. for $\text{C}_{14}\text{H}_{12}$: 180.0, found: 180.





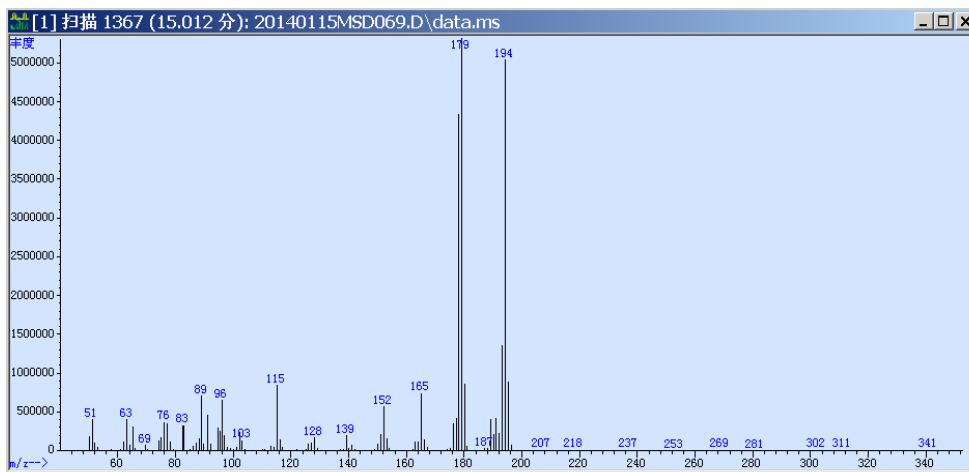
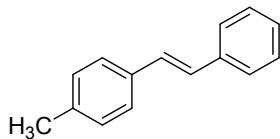
(E)-1-methoxy-4-styrylbenzene 8b:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₅H₁₄O: 210.0, found: 210.



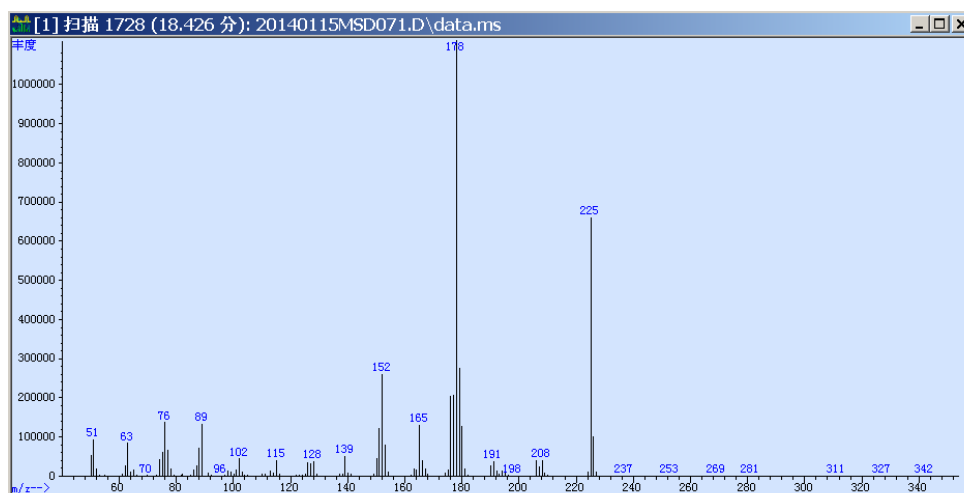
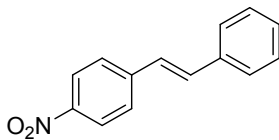
(E)-1-methyl-4-styrylbenzene 8c:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₅H₁₄: 194.0, found: 194.



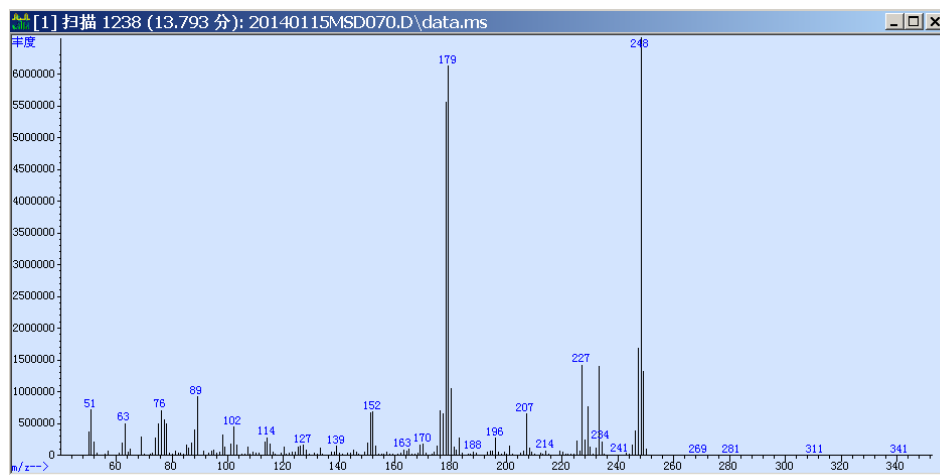
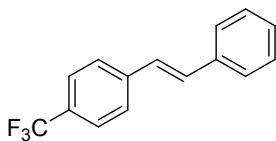
(E)-1-nitro-4-styrylbenzene 8d:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₄H₁₁NO₂: 225.0, found: 225.



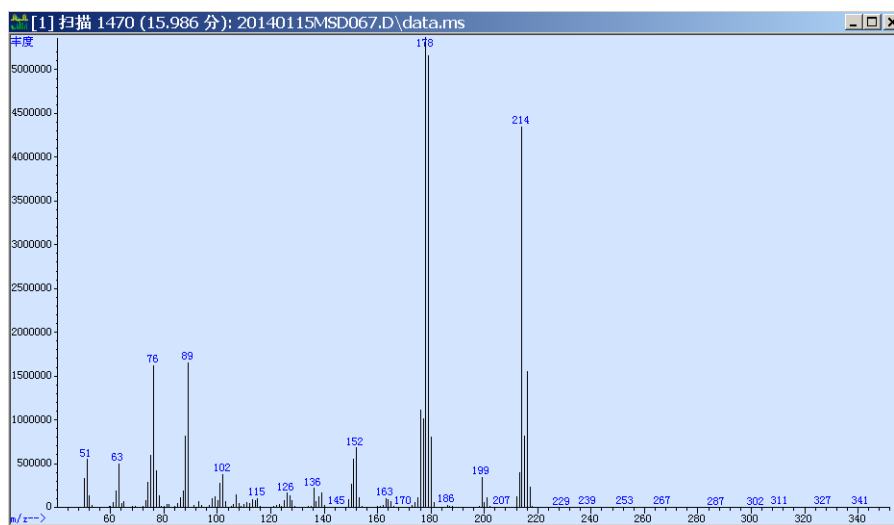
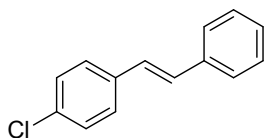
(E)-1-styryl-4-(trifluoromethyl)benzene 8e:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₅H₁₁F₃: 248.0, found: 248.



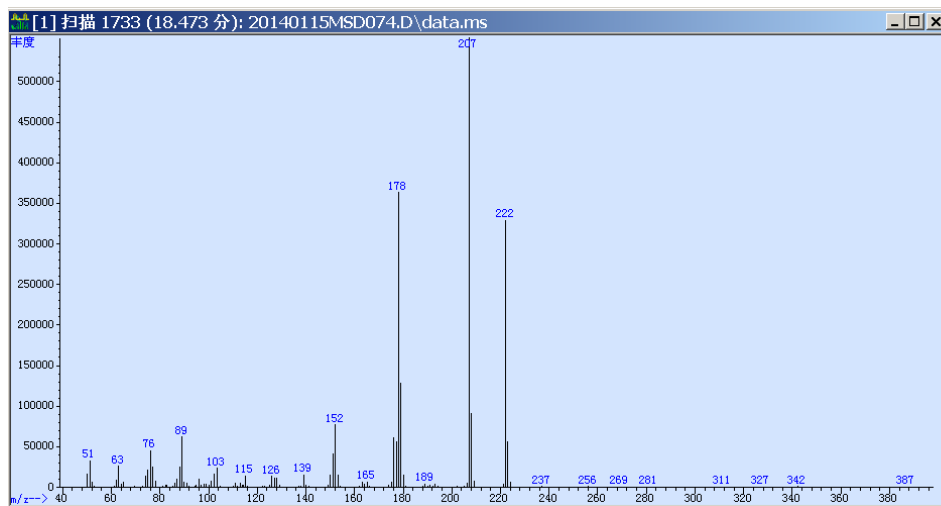
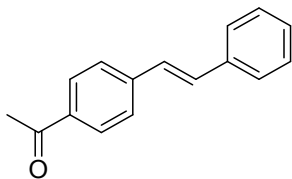
(E)-1-chloro-4-styrylbenzene 8f:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₄H₁₁Cl: 214.0, found: 214.



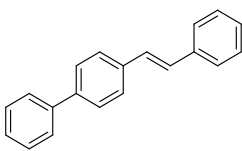
(E)-4-acetyl-4-styrylbenzene 8h:

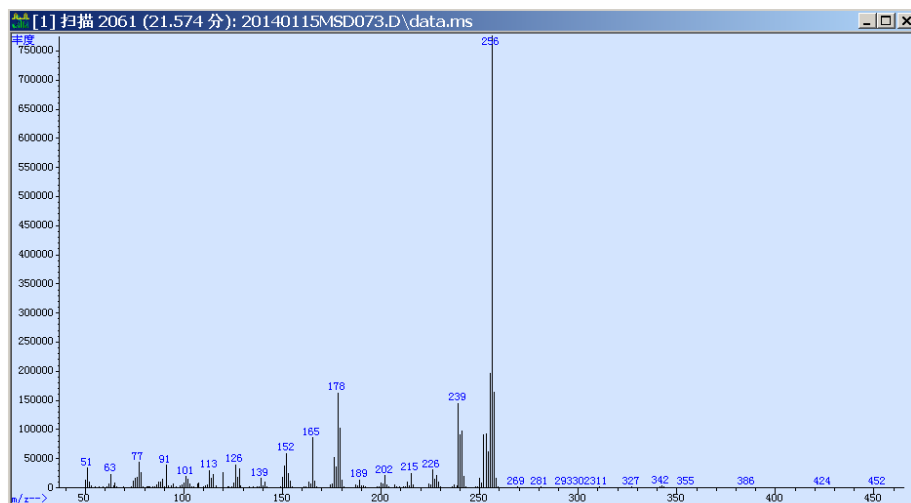
GC-MS (EI) [M]⁺: m/z calcd. for C₁₆H₁₄O: 222.0, found: 222.



(E)-4-phenyl-4-styrylbenzene 8i:

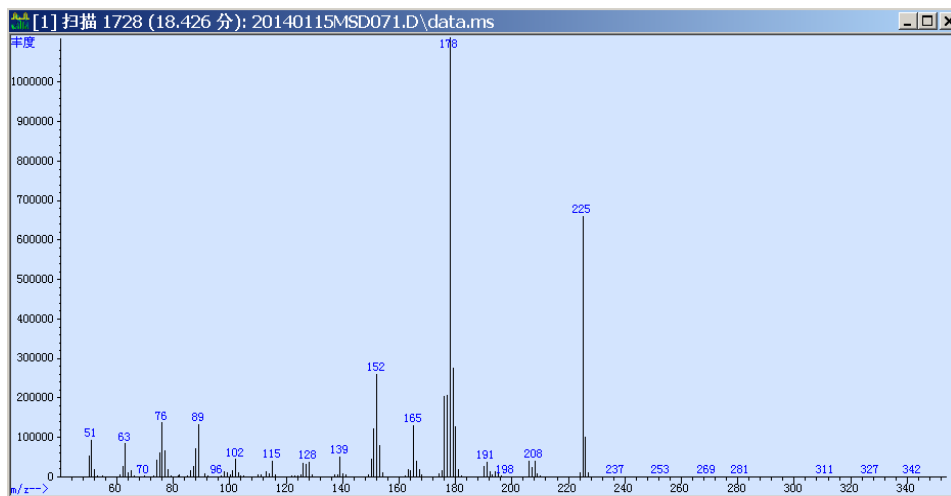
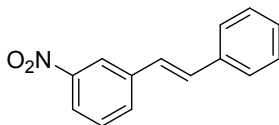
GC-MS (EI) [M]⁺: m/z calcd. for C₂₀H₁₆: 256.0, found: 256.





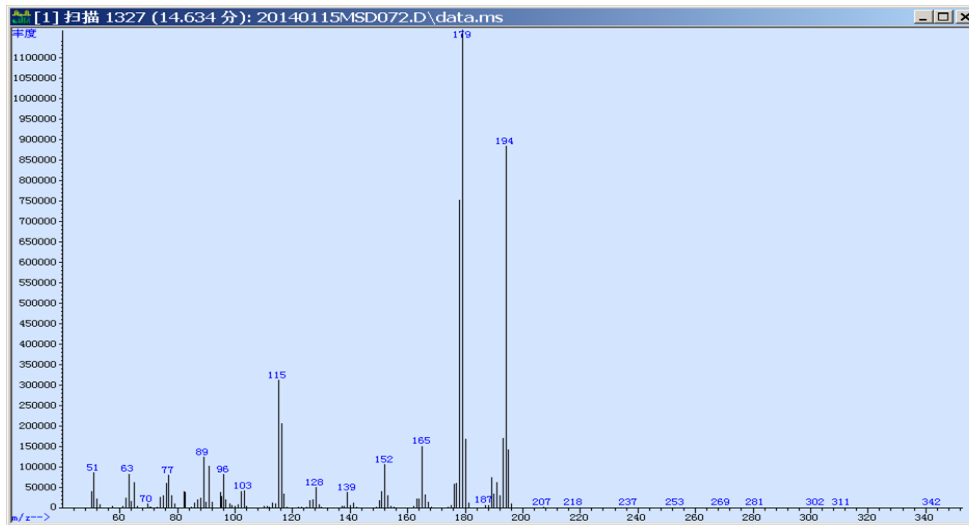
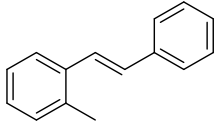
(E)-1-nitro-3-styrylbenzene 8j:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₄H₁₁NO₂: 225.0, found: 225.



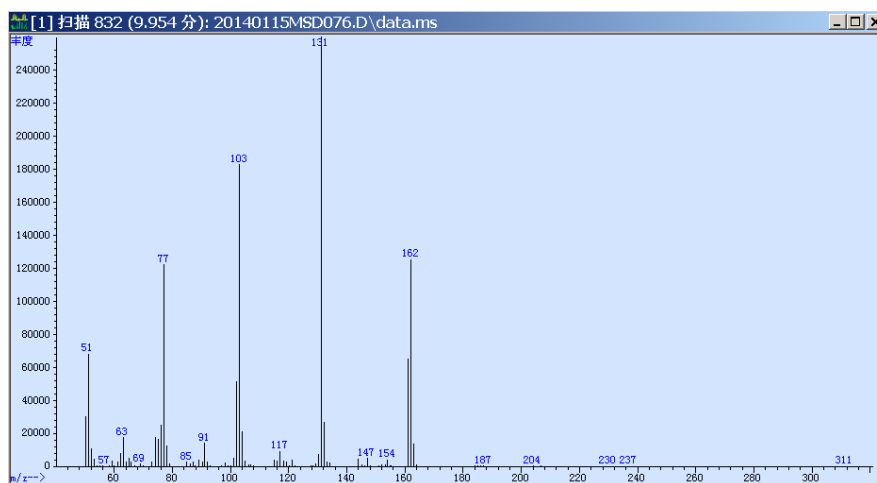
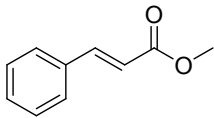
(E)-1-methyl-2-styrylbenzene 8l:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₅H₁₄: 194.0, found: 194.



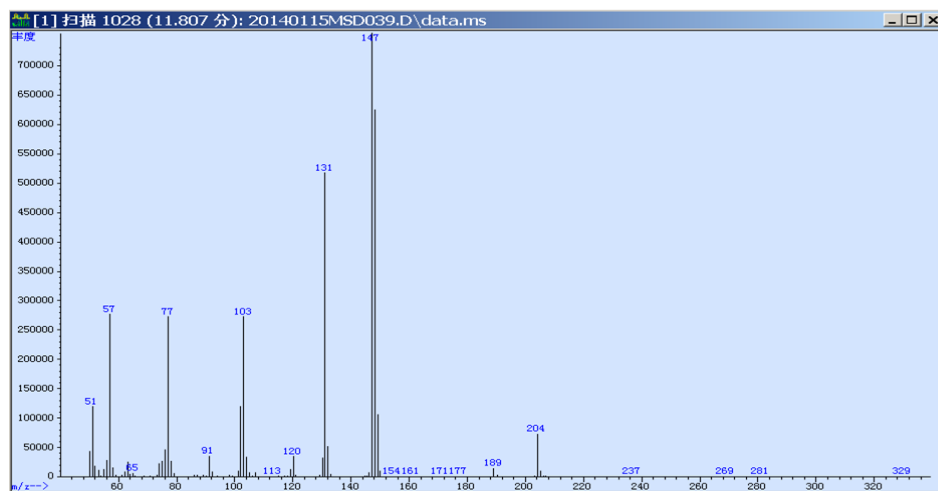
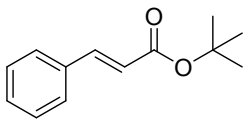
Methyl cinnamate 8m:

GC-MS (EI) [M]⁺: m/z calcd. for C₁₀H₁₀O₂: 162.0, found: 162.



***tert*-butyl cinnamate 8n:**

GC-MS (EI) [M]⁺: m/z calcd. for C₁₃H₁₆O₂: 204.0, found: 204.



4. References

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