

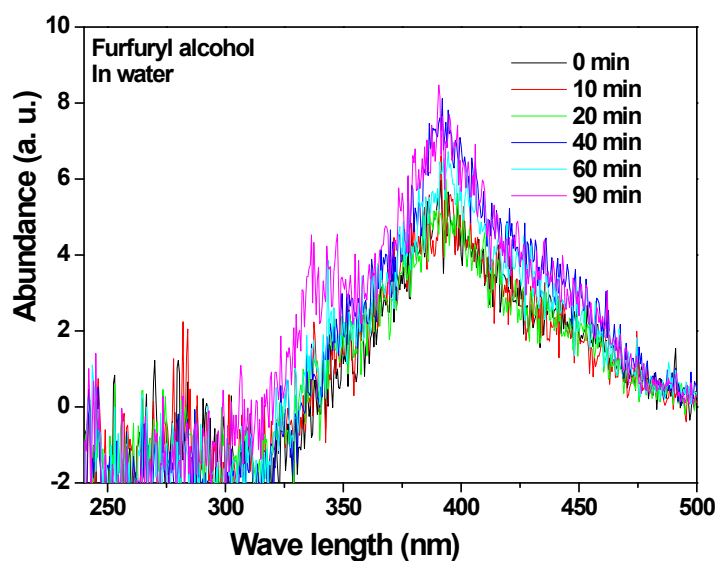
## Supporting Information

### Upgrading biomass-derived furans via acid-catalysis/hydrogenation: The remarkable difference between water and methanol as the solvent

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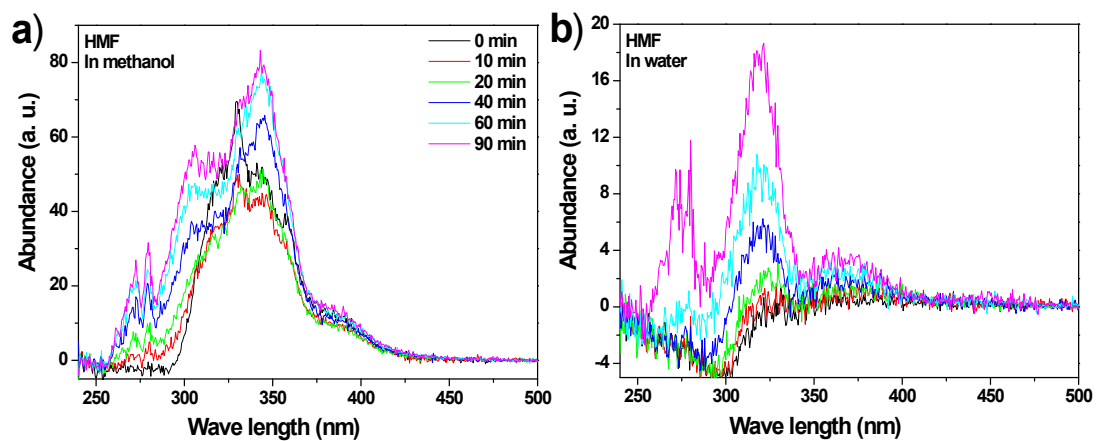
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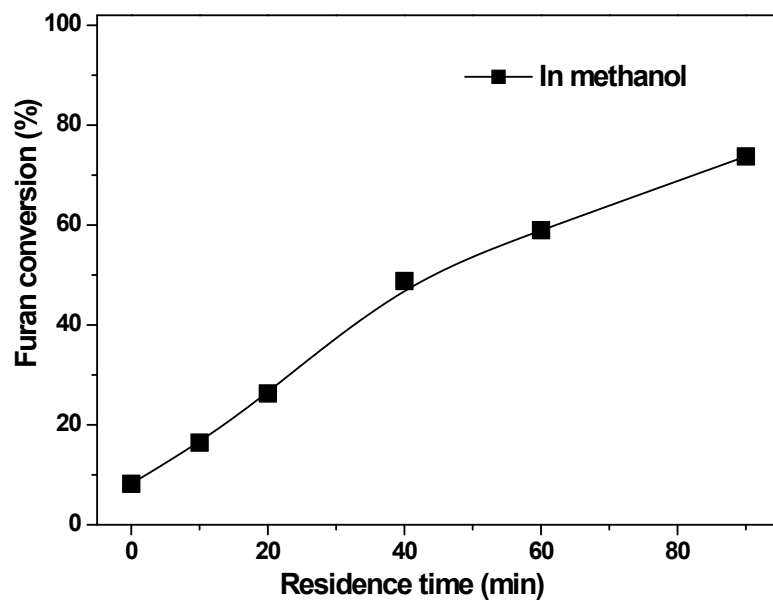


**Figure 1** UV-fluorescence spectra for determination of the soluble polymer formed in acid-treatment of furfuryl alcohol in water: “0 min” means the reaction temperature just reached 130°C.

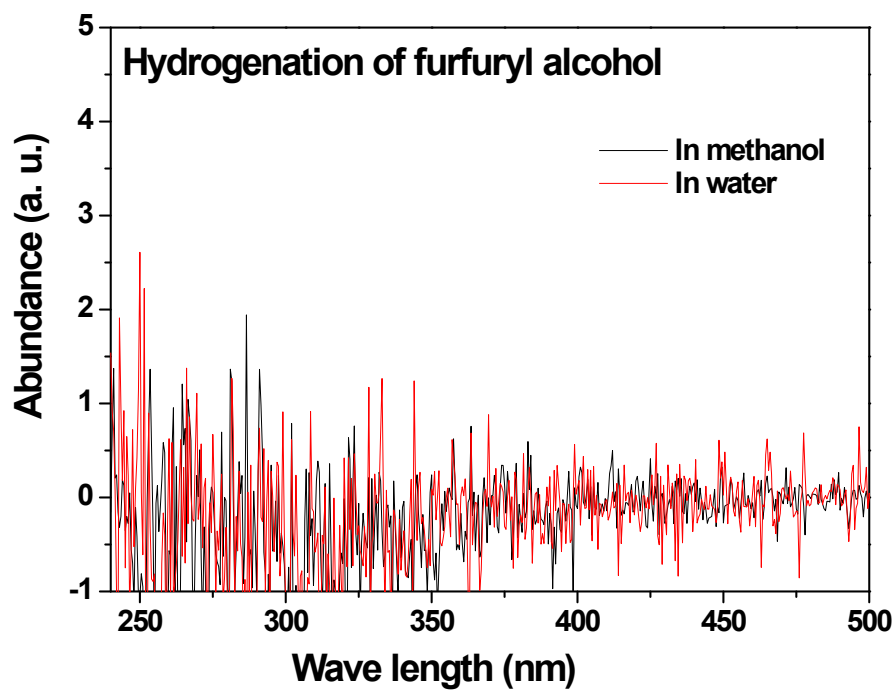
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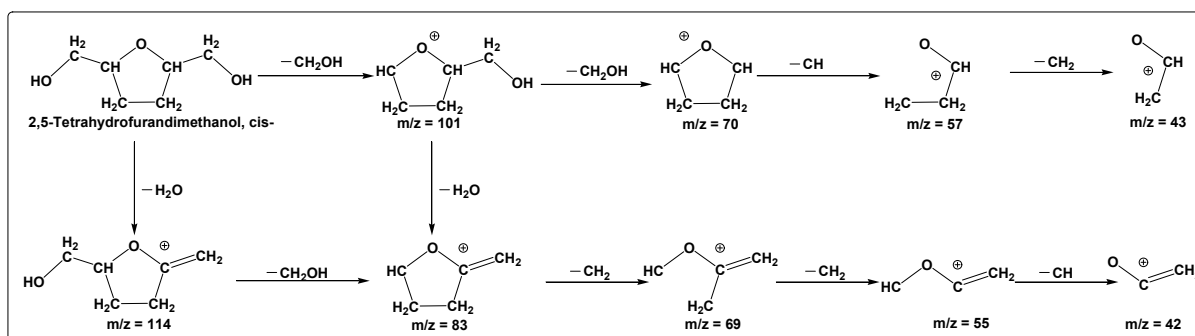
**Figure S2** UV-fluorescence spectra for determination of the soluble polymer formed in acid-treatment of HMF in methanol and in water. It needs to be noted that the higher abundance of the soluble polymer in methanol is due to the higher solubility of the soluble polymer in methanol.



**Figure S3** Conversion of furan in its acid-catalyzed conversion in methanol. Reaction conditions: furan loaded: 3 g, Amberlyst 70: 3 g, methanol: 100 ml,  $T = 170^{\circ}\text{C}$ , residence time: 90 min. The conversion of furan in water was not presented as the low solubility of furan in water makes it difficult to get the accurate conversion data.



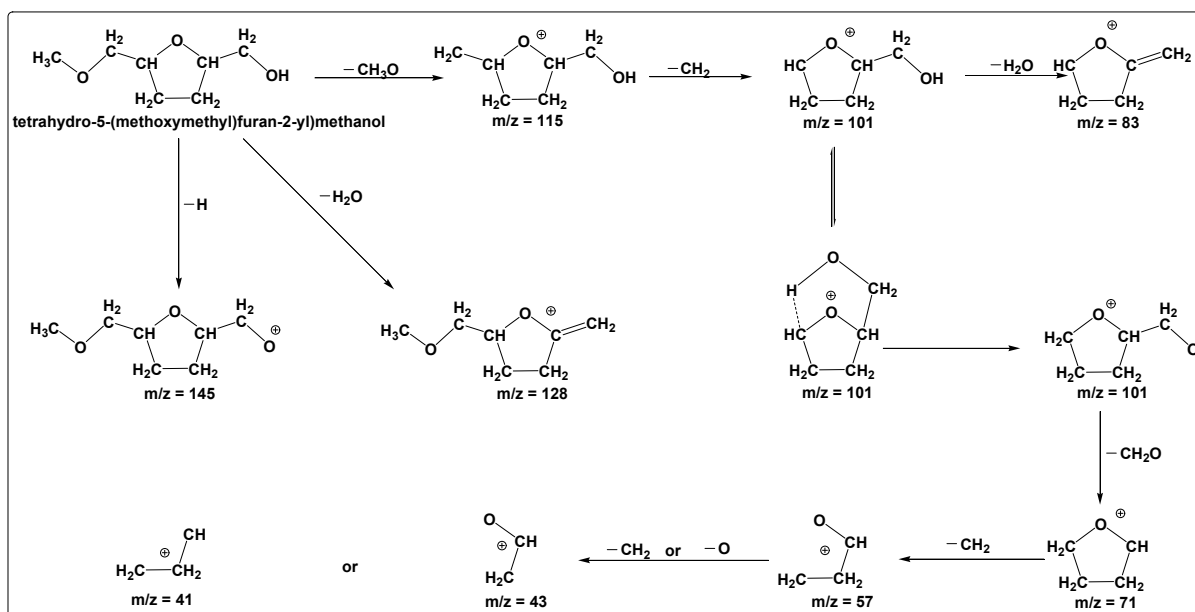
**Figure S4** UV-fluorescence spectra for determination of the soluble polymer formed in hydrogenation of furfuryl alcohol in methanol and in water.



**Figure S5** Proposed reaction pathways for the degradation of cis-2,5-tetrahydrofurandimethanol (TFM) in GC-MS.



Figure S6 MS spectrum for cis-2,5-tetrahydrofurandimethanol (TFM).

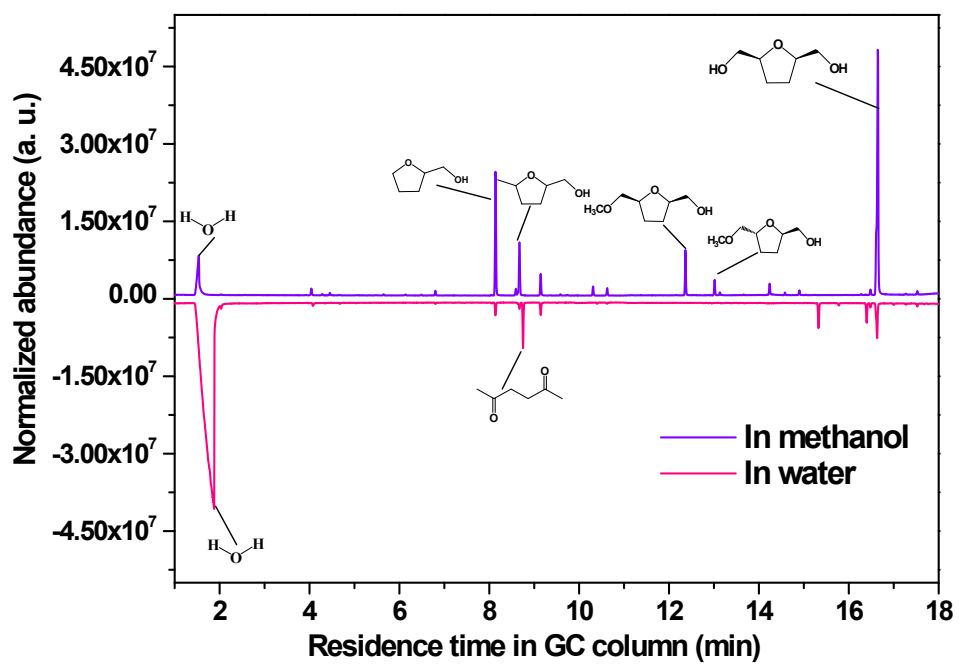


**Figure S7** Proposed reaction pathways for the degradation of tetrahydro-5-(methoxymethyl)furan-2-yl)methanol (THFM).

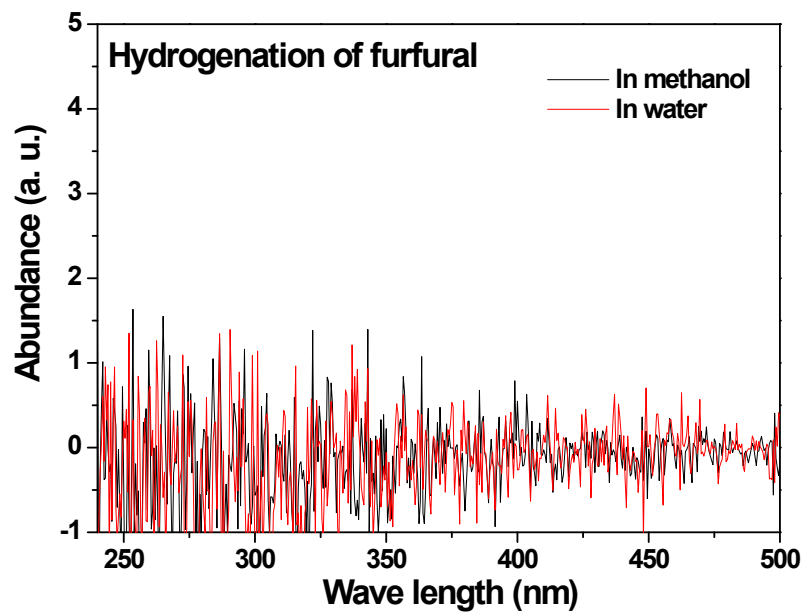


**Figure S8** MS spectrum for tetrahydro-5-(methoxymethyl)furan-2-yl)methanol (THFM).





**Figure S9** Normalized GC-MS spectra for the hydrogenation of HMF in methanol and water.



**Figure S10** UV-fluorescence spectra for determination of the soluble polymer formed in hydrogenation of furfuryl alcohol in methanol and in water.