Supporting Information

Hierarchical micro-/mesoporous N- and O-enriched carbon derived from disposable cashmere: A competitive cost-effective material for high performance electrochemical capacitors

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Fig. S1. Contact angle of the hierarchical CDMMC sample



Fig. S2. Relationship between electrochemically cathodic and anodic redox peak currents and the sweep rate of the hierarchical CDMMC electrode in $1 \text{ M H}_2\text{SO}_4$ electrolyte



Fig. S3. EIS spectra of the CDMMC electrode in three-electrode configurations with (a) 6 M KOH and (b) 1 M H_2SO_4 solutions. The insets in (a, b) for the corresponding high-frequency regions of the EIS spectra, respectively

In general, the intersection of the plots at the X-axis represents solution resistance (R_s), which is associated with the following three items: the resistance of the aqueous solution, the intrinsic resistance of the electroactive material itself, and the contact resistance at the interface between electroactive material and current collector. As observed in **Fig. S3a**, **b**, the values for the R_s in the 6 M KOH and 1 M H₂SO₄ are observed as ~0.56 Ohm and 0.41 Ohm, respectively.

SCs (F g ⁻¹)/		electrochemical	Ref.
discharge current	electrolyte	window	
density (A g ⁻¹)		(V)	
363/0.5	КОН	1.0	This work
460/0.5	H_2SO_4	1.0	This work
220/0.5	KOH	1.0	[1]
319/0.5	KOH	1.0	[2]
340/2.0	КОН	1.0	[3]
280/<1.0	КОН	0.8	[4]
400/0.5	КОН	1.0	[5]
347/0.2	H_2SO_4	1.0	[6]
326	H_2SO_4	1.0	[7]
327/1.0	H_2SO_4	1.0	[8]
300/0.2	H_2SO_4	1.0	[9]
340/0.1	H_2SO_4	1.2	[10]
239/0.5	H_2SO_4	1.0	[11]
243/0.05	КОН	1.0	[12]
	SCs (F g ⁻¹)/ discharge current density (A g ⁻¹) 363/0.5 460/0.5 220/0.5 319/0.5 340/2.0 280/<1.0 400/0.5 347/0.2 326 327/1.0 300/0.2 340/0.1 239/0.5 243/0.05	SCs (F g-1)/ discharge current density (A g-1)electrolyte density (A g-1) $363/0.5$ KOH $460/0.5$ H ₂ SO ₄ $220/0.5$ KOH $319/0.5$ KOH $319/0.5$ KOH $340/2.0$ KOH $280/<1.0$ KOH $400/0.5$ KOH $347/0.2$ H ₂ SO ₄ 326 H ₂ SO ₄ $327/1.0$ H ₂ SO ₄ $300/0.2$ H ₂ SO ₄ $340/0.1$ H ₂ SO ₄ $340/0.5$ H ₂ SO ₄ $340/0.5$ H ₂ SO ₄ $340/0.1$ H ₂ SO ₄ $340/0.5$ H ₂ SO ₄ $239/0.5$ H ₂ SO ₄ $243/0.05$ KOH	SCs (F g ⁻¹)/electrochemicaldischarge currentelectrolytewindowdensity (A g ⁻¹)(V) $363/0.5$ KOH 1.0 $460/0.5$ H ₂ SO ₄ 1.0 $220/0.5$ KOH 1.0 $220/0.5$ KOH 1.0 $319/0.5$ KOH 1.0 $340/2.0$ KOH 1.0 $280/<1.0$ KOH 0.8 $400/0.5$ KOH 1.0 $347/0.2$ H ₂ SO ₄ 1.0 326 H ₂ SO ₄ 1.0 $327/1.0$ H ₂ SO ₄ 1.0 $300/0.2$ H ₂ SO ₄ 1.0 $340/0.1$ H ₂ SO ₄ 1.0 $340/0.1$ H ₂ SO ₄ 1.0 $340/0.5$ KOH 1.0 $340/0.5$ H ₂ SO ₄ 1.0

Table S1. SCs of the CDMMC *vs.* recently published heteroatom-rich carbons all tested in threeelectrode configurations with different electrolytes

[1] G. Y. Xu, B. Ding, P. Nie, L. F. Shen, J. Wang, X. G. Zhang, Chem. Eur. J. 2013, 19, 12306.

[2] L. Qie, W. Chen, H. Xu, X. Xiong, Y. Jiang, F. Zou, X. Hu, Y. Xin, Z. Zhang, Y. Huang, *Energy Environ. Sci.* 2013, 6, 2497

[3] W. J. Qian, F. X. Sun, Y. H. Xun, L. H. Qiu, C. H. Liu, S. D. Wang, F. Yan, *Energy Environ. Sci.* 2014, 7, 379.

[4] H. M. Jeong, J. W. Lee, W. H. Shin, Y. J. Choi, H. J. Shin, J. K. Kang, J. W. Choi, *Nano Lett.***2011**, 11, 2472.

- [5] M. Biswal, A. Banerjee, M. Deo, S. Ogale, Energy Environ. Sci. 2013, 6, 1249.
- [6] Y. Chen, X. O. Zhang, D. C. Zhang, P. Yu, Y. W. Ma, Carbon 2011, 49, 573.
- [7] S. Y. Yang, K. H. Chang, H. W. Tien, Y. F. Lee, S. M. Li, Y. S. Wang, J. Y. Wang, C. C. M.
- Ma, C. C. Hu, J. Mater. Chem. 2011, 21, 2374.
- [8] H. Sun, L. Cao, L. Lu, Energy Environ. Sci. 2012, 5, 6206.
- [9] L. Zhao, L. Z. Fan, M. Q. Zhou, H. Guan, S. Y. Qiao, M. Antonietti, M. M. Titirici, *Adv. Mater.*2010, 22, 5202.
- [10] C. O. Ania, V. Khomenko, E. Raymundo-Pinero, J. B. Parra, F. Beguin, *Adv. Funct. Mater.*2007, 17, 1828.
- [11] T. Bordjiba, M. Mohamedi, L. H. Dao, Adv. Mater. 2008, 20, 815.
- [12] X. J. He, P. H. Ling, M. X. Yu, X. T. Wang, X. Y. Zhang, M. D. Zheng, *Electrochim. Acta* 2013, 105, 635.



Fig. S4. EIS pattern of the CDMMC-based symmetric EC with 1 M TEABF₄/PC electrolyte. The inset for the corresponding high-frequency region of the EIS spectrum