

Supporting Information

A Route to Convert CO₂: Synthesis of 3,4,5-Trisubstituted Oxazolones

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1. Experimental Results

1.1 Table S1

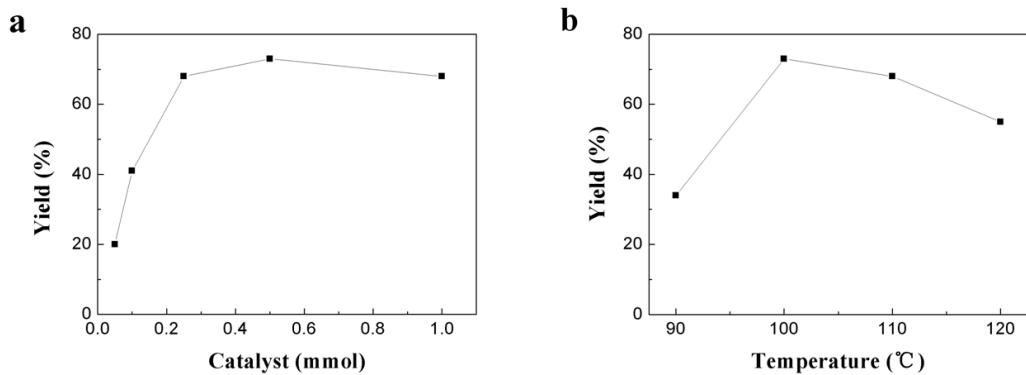
Table S1 Reaction of CO₂ with propargylic amine **1a** in various catalysts.^a

Entry	Catalyst	Conversion ^b /%	Yield ^b /%	Selectivity/%
1	--	0	0	--
2	Cs ₂ CO ₃	40	4	10
3	TBD ^c	>99	22	22
4	DBU ^d	>99	32	32
5	[Bmim][OAc]	>99	45	45

^aTypical reaction conditions were as follows until otherwise stated: 0.5 mmol of **1a**, 0.05 mmol of catalyst, 0.5 mL solvent of N,N-dimethylformamide, 0.1 MPa CO₂, 100 °C, 2 h. ^bConversion and yield were determined by ¹H NMR spectroscopy using 1,3,5-trioxane as an internal standard. ^cTBD =1,5,7-triazabicyclo[4.4.0]dec-5-ene. ^dDBU = 1,8-diazabicyclo[5.4.0]undec-7-ene.

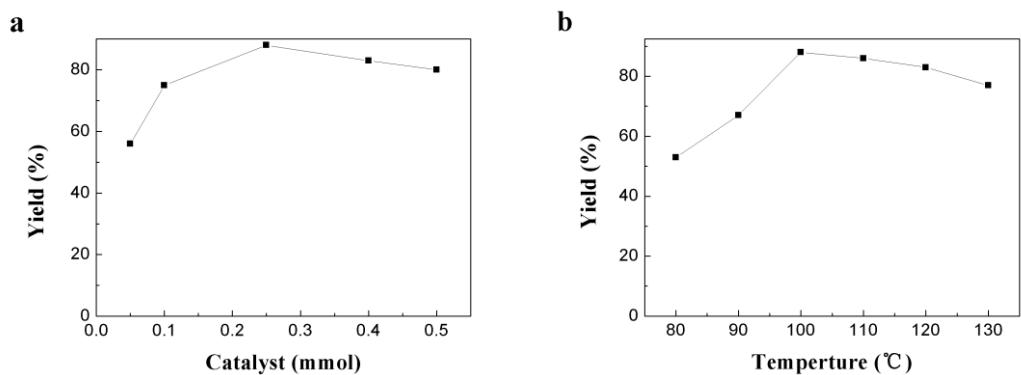
1.2. Figures S1-S4

Figure S1 Optimization of catalyst amount and temperature



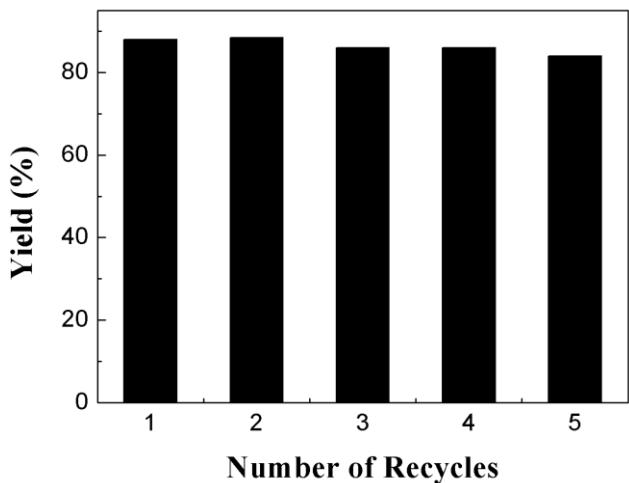
Typical reaction conditions were as follows until otherwise stated: **a)** 0.5 mmol of **1a**, 0.1 MPa CO₂ and different dosage of [Bmim][OAc], 100 °C, 2h. **b)** 0.5 mmol of **1a**, 0.5 mmol of [Bmim][OAc], 0.1 MPa CO₂, different temperatures, 2 h. Yields were determined by ¹H NMR spectroscopy using 1,3,5-trioxane as an internal standard.

Figure S2 Optimization of IL system amount and temperature



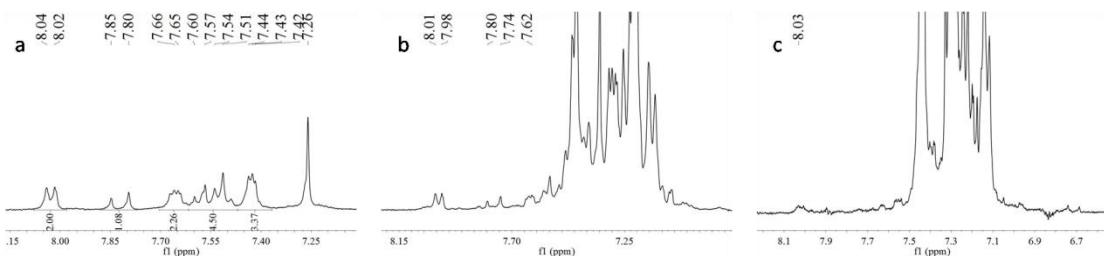
Typical reaction conditions were as follows until otherwise stated: **a)** 0.5 mmol of **1a**, 0.5 mmol of [Bmim][Tf₂N] and different dosage of [Bmim][OAc], 0.1 MPa CO₂, 100 °C, 12 h. **b)** 0.5 mmol of **1a**, 0.25 mmol of [Bmim][OAc] and 0.5 mmol of [Bmim][Tf₂N], 0.1 MPa CO₂, different temperatures, 12 h. Yields were determined by ¹H NMR spectroscopy using 1,3,5-trioxane as an internal standard.

Figure S3 Recyclability of the IL system



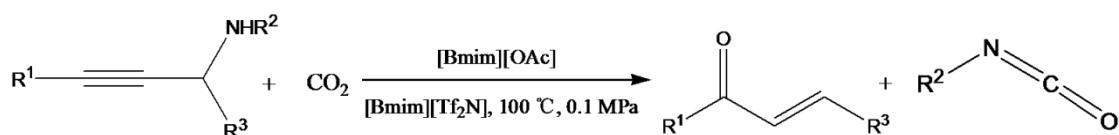
Typical reaction conditions were as follows until otherwise stated: 0.5 mmol of **1a**, 0.25 mmol of [Bmim][OAc] and 0.5 mmol of [Bmim][Tf₂N], 0.1 MPa CO₂, 100 °C, 12 h. After the reaction, the product was extracted by diethyl ether and yields were determined by ¹H NMR spectroscopy using 1,3,5-trioxane as an internal standard. The IL mixture was used directly for the next run after drying.

Figure S4 The characteristic peak of chalcone in different reaction



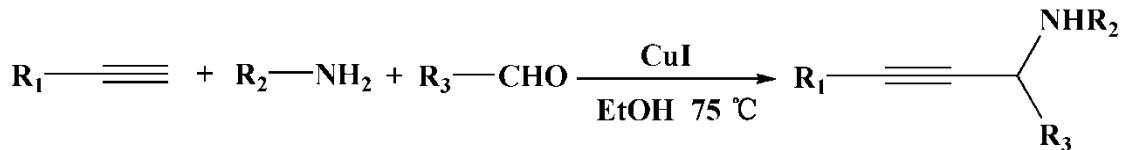
a: the pure chalcone isolated by silica gel column chromatography. b: reaction condition was 0.5 mmol of **1a**, 0.5 mmol of [Bmim][OAc], 0.1 MPa CO₂, 100 °C, 2 h. c: reaction condition was 0.5 mmol of **1a**, 0.25 mmol of [Bmim][OAc] and 0.5 mmol of [Bmim][Tf₂N], 0.1 MPa CO₂, 100 °C, 12 h.

Scheme S1 Possible side reaction.^{S2}

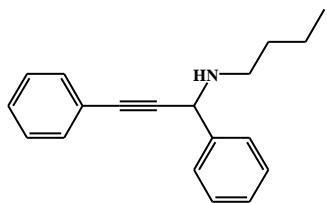


1.3. Substrates

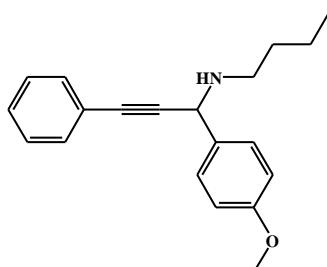
Preparation propargylic amines^{s3}



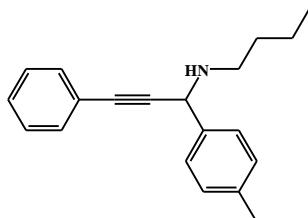
Data:



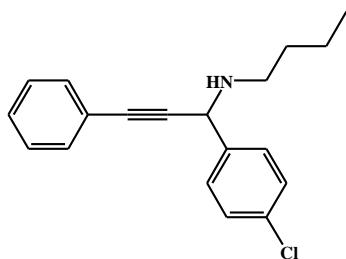
Butyl-(1,3-diphenyl-prop-2-ynyl)-amine (1a). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.61 (d, $J = 7.3$ Hz, 2H), 7.48 (dd, $J = 6.5, 2.9$ Hz, 2H), 7.39 (t, $J = 7.5$ Hz, 2H), 7.35 to 7.27 (m, 4H), 4.82 (s, 1H), 2.87 (dt, $J = 11.2, 7.3$ Hz, 1H), 2.74 (dt, $J = 11.2, 7.1$ Hz, 1H), 1.65 to 1.45 (m, 3H), 1.45 to 1.33 (m, 2H), 0.94 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 140.70, 131.74, 128.53, 128.27, 128.11, 127.70, 127.61, 123.27, 89.67, 85.27, 54.77, 47.11, 32.17, 20.53, 14.03.



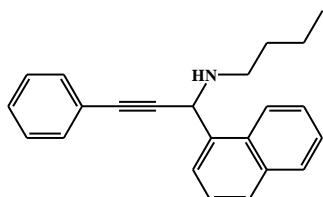
Butyl-[1-(4-methoxy-phenyl)-3-phenyl-prop-2-ynyl]-amine (1b). yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.63 to 7.42 (m, 4H), 7.40 to 7.27 (m, 3H), 6.91 (d, $J = 8.6$ Hz, 2H), 4.77 (s, 1H), 3.82 (s, 3H), 2.85 (dt, $J = 11.2, 7.3$ Hz, 1H), 2.72 (dt, $J = 11.2, 7.1$ Hz, 1H), 1.65 to 1.45 (m, 3H), 1.39 (dq, $J = 14.6, 7.3$ Hz, 2H), 0.93 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 159.16, 132.92, 131.72, 128.74, 128.26, 128.06, 123.31, 113.86, 89.91, 85.09, 55.32, 54.15, 47.02, 32.15, 20.55, 14.03.



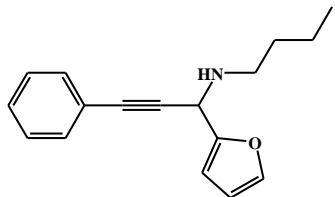
Butyl-(3-phenyl-1-p-tolyl-prop-2-ynyl)-amine (1c). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.51 (dd, $J = 8.6, 3.7$ Hz, 4H), 7.39 to 7.29 (m, 3H), 7.21 (d, $J = 7.8$ Hz, 2H), 4.81 (s, 1H), 2.89 (dt, $J = 11.3, 7.3$ Hz, 1H), 2.75 (dt, $J = 11.2, 7.1$ Hz, 1H), 2.39 (s, 3H), 1.56 (dt, $J = 13.8, 7.1$ Hz, 3H), 1.47 to 1.35 (m, 2H), 0.96 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 137.79, 137.37, 131.75, 129.22, 128.27, 128.07, 127.53, 123.37, 89.91, 85.10, 54.52, 47.10, 32.19, 21.17, 20.57, 14.07.



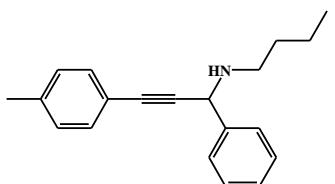
Butyl-[1-(4-chloro-phenyl)-3-phenyl-prop-2-ynyl]-amine (1d). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.57 (d, $J = 8.3$ Hz, 2H), 7.51 (dd, $J = 6.5, 3.0$ Hz, 2H), 7.42 to 7.28 (m, 5H), 4.81 (s, 1H), 2.86 (dt, $J = 11.2, 7.2$ Hz, 1H), 2.74 (dt, $J = 11.2, 7.0$ Hz, 1H), 1.56 (ddd, $J = 15.3, 10.1, 4.6$ Hz, 2H), 1.49 to 1.36 (m, 3H), 0.96 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 139.26, 133.45, 131.77, 129.06, 128.62, 128.36, 128.30, 123.07, 89.18, 85.67, 54.08, 46.99, 32.19, 20.55, 14.07.



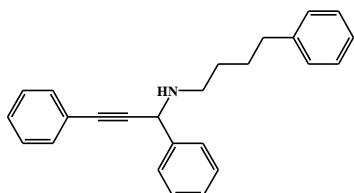
Butyl-(1-naphthalen-1-yl-3-phenyl-prop-2-ynyl)-amine (1e). yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 8.45 (d, $J = 8.4$ Hz, 1H), 8.04 (d, $J = 6.9$ Hz, 1H), 7.96 (d, $J = 8.1$ Hz, 1H), 7.90 (d, $J = 8.2$ Hz, 1H), 7.72 to 7.49 (m, 5H), 7.39 (d, $J = 5.0$ Hz, 3H), 5.60 (s, 1H), 3.20 to 3.04 (m, 1H), 3.04 to 2.87 (m, 1H), 1.89 to 1.57 (m, 3H), 1.57 to 1.43 (m, 2H), 1.04 (td, $J = 7.3, 1.7$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 136.04, 134.26, 131.86, 131.23, 128.92, 128.73, 128.40, 128.22, 126.31, 125.78, 125.45, 125.41, 123.97, 123.44, 89.86, 85.87, 52.47, 47.86, 32.28, 20.70, 14.16.



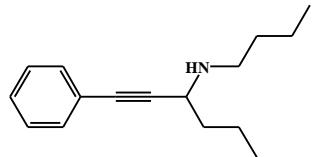
Butyl-(1-furan-2-yl-3-phenyl-prop-2-ynyl)-amine (1f). brown oil; ^1H NMR (400 MHz, CDCl_3) δ 7.47 (dd, $J = 6.5, 2.9$ Hz, 2H), 7.41 (s, 1H), 7.37 to 7.27 (m, 3H), 6.44 (d, $J = 3.1$ Hz, 1H), 6.35 (dd, $J = 2.9, 1.9$ Hz, 1H), 4.91 (s, 1H), 2.91 to 2.67 (m, 2H), 1.65 (s, 1H), 1.59 to 1.44 (m, 2H), 1.44 to 1.31 (m, 2H), 0.92 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 153.00, 142.42, 131.81, 128.32, 128.28, 122.86, 110.18, 107.36, 87.00, 84.29, 48.62, 46.59, 32.04, 20.49, 14.00.



Butyl-(1-phenyl-3-p-tolyl-prop-2-ynyl)-amine (1g). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.64 (d, $J = 7.6$ Hz, 2H), 7.41 (t, $J = 6.8$ Hz, 4H), 7.34 (t, $J = 7.3$ Hz, 1H), 7.15 (d, $J = 7.9$ Hz, 2H), 4.84 (s, 1H), 2.90 (dt, $J = 11.4, 7.3$ Hz, 1H), 2.77 (dt, $J = 11.3, 7.0$ Hz, 1H), 2.38 (s, 3H), 1.70 to 1.50 (m, 3H), 1.43 (dq, $J = 14.6, 7.3$ Hz, 2H), 0.98 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 140.85, 138.18, 131.67, 129.07, 128.54, 127.69, 127.67, 120.25, 88.95, 85.43, 54.82, 47.12, 32.21, 21.51, 20.59, 14.09.

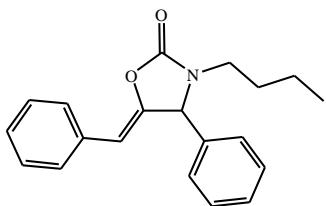


(1,3-Diphenyl-prop-2-ynyl)-(4-phenyl-butyl)-amine (1h). yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.56 (d, $J = 7.6$ Hz, 2H), 7.49 to 7.39 (m, 2H), 7.34 (t, $J = 7.4$ Hz, 2H), 7.30 to 7.18 (m, 6H), 7.13 (d, $J = 7.6$ Hz, 3H), 4.76 (s, 1H), 2.85 (dt, $J = 11.3, 7.1$ Hz, 1H), 2.77 to 2.65 (m, 1H), 2.59 (t, $J = 7.5$ Hz, 2H), 1.75 to 1.61 (m, 2H), 1.55 (dq, $J = 14.3, 7.0$ Hz, 2H), 1.40 (s, 1H). ^{13}C NMR (100 MHz, CDCl_3) δ 142.57, 140.72, 131.85, 128.64, 128.57, 128.40, 128.25, 127.83, 127.73, 125.83, 123.33, 89.73, 85.48, 54.85, 47.28, 35.92, 29.77, 29.34.

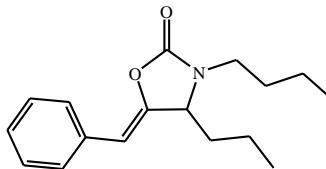


Butyl-(1-phenylethynyl-butyl)-amine (1i). yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.42 (dd, $J = 6.5, 3.0$ Hz, 2H), 7.36 to 7.26 (m, 3H), 3.59 (dd, $J = 7.7, 5.9$ Hz, 1H), 2.93 (ddd, $J = 11.1, 8.1, 6.6$ Hz, 1H), 2.66 (ddd, $J = 11.2, 8.2, 6.0$ Hz, 1H), 1.68 (ddd, $J = 13.5, 8.4, 6.3$ Hz, 2H), 1.63 to 1.46 (m, 4H), 1.38 (dd, $J = 15.3, 7.5$ Hz, 3H), 0.95 (dt, $J = 14.3, 7.3$ Hz, 6H). ^{13}C NMR (100 MHz, CDCl_3) δ 131.66, 128.21, 127.82, 123.52, 91.34, 83.55, 50.64, 47.33, 38.32, 32.25, 20.56, 19.44, 14.03, 13.94.

1.4. Intermediate data:

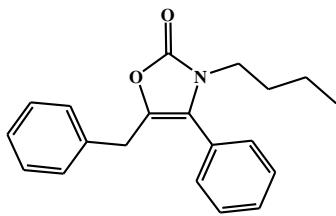


5-benzylidene-3-butyl-4-phenyloxazolidin-2-one (intermediate of 1a). white solid; ^1H NMR (400 MHz, CDCl_3) δ 7.52 (d, $J = 7.5$ Hz, 2H), 7.48 to 7.40 (m, 3H), 7.38 to 7.31 (m, 2H), 7.31 to 7.24 (m, 2H), 7.19 (t, $J = 7.4$ Hz, 1H), 5.39 (d, $J = 1.4$ Hz, 1H), 5.25 (d, $J = 1.7$ Hz, 1H), 1.58 (s, 1H), 1.53 to 1.38 (m, 2H), 1.38 to 1.18 (m, 2H), 0.88 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.05, 147.70, 137.33, 133.46, 129.37, 129.33, 128.45, 128.33, 127.81, 126.94, 104.55, 63.85, 41.61, 28.96, 19.81, 13.64.

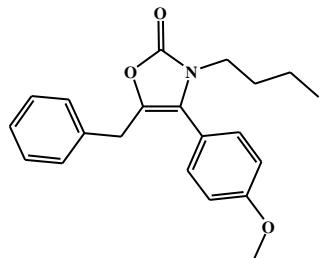


5-Benzylidene-3-butyl-4-propyl-oxazolidin-2-one (intermediate of 1i). colourless liquid; ^1H NMR (400 MHz, CDCl_3) δ 7.58 (d, $J = 7.5$ Hz, 2H), 7.32 (t, $J = 7.7$ Hz, 2H), 7.20 (t, $J = 7.4$ Hz, 1H), 5.47 (d, $J = 1.4$ Hz, 1H), 4.51 (s, 1H), 3.59 (dt, $J = 14.4, 8.0$ Hz, 1H), 3.02 (ddd, $J = 13.9, 8.3, 5.3$ Hz, 1H), 1.92 – 1.78 (m, 1H), 1.74 – 1.62 (m, 1H), 1.57 (ddd, $J = 21.5, 14.1, 7.9$ Hz, 2H), 1.44 – 1.25 (m, 4H), 0.95 (dd, $J = 7.8, 6.8$ Hz, 6H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.17, 147.04, 133.67, 128.46, 128.28, 126.72, 102.31, 58.43, 41.14, 34.41, 29.25, 19.91, 15.83, 13.90, 13.67.

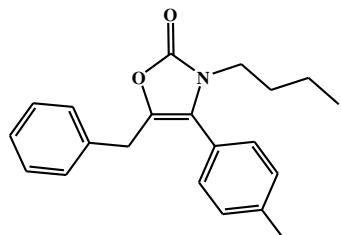
1.5. Products data:



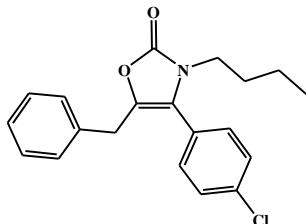
5-Benzyl-3-butyl-4-phenyl-3H-oxazol-2-one (2a). colorless oil; ^1H NMR (400 MHz, CDCl_3) δ 7.53 to 7.43 (m, 3H), 7.38 to 7.21 (m, 5H), 7.21 to 7.14 (m, 2H), 3.70 (s, 2H), 3.58 to 3.43 (m, 2H), 1.51 to 1.36 (m, 2H), 1.23 to 1.08 (m, 2H), 0.77 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.51, 136.93, 135.17, 129.68, 129.41, 129.04, 128.64, 128.42, 126.82, 126.79, 124.08, 42.01, 30.98, 30.67, 19.59, 13.45.



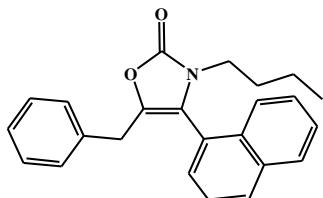
5-Benzyl-3-butyl-4-(4-methoxy-phenyl)-3H-oxazol-2-one (2b). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.35 to 7.11 (m, 7H), 6.99 (d, $J = 8.6$ Hz, 2H), 3.86 (s, 3H), 3.67 (s, 2H), 3.47 (t, $J = 7.4$ Hz, 2H), 1.54 to 1.37 (m, 2H), 1.24 to 1.08 (m, 2H), 0.78 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 160.39, 155.50, 137.11, 134.94, 131.16, 128.62, 128.42, 126.74, 123.80, 118.77, 114.49, 55.38, 41.91, 30.97, 30.72, 19.63, 13.52.



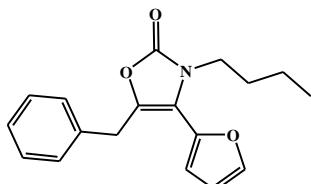
5-Benzyl-3-butyl-4-p-tolyl-3H-oxazol-2-one (2c). colorless oil; ^1H NMR (400 MHz, CDCl_3) δ 7.33 to 7.25 (m, 4H), 7.25 to 7.15 (m, 5H), 3.68 (s, 2H), 3.53 to 3.43 (m, 2H), 2.42 (s, 3H), 1.52 to 1.36 (m, 2H), 1.23 to 1.09 (m, 2H), 0.78 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.55, 139.49, 137.07, 134.99, 129.73, 129.59, 128.62, 128.43, 126.75, 124.06, 123.77, 41.97, 30.96, 30.70, 21.39, 19.62, 13.50.



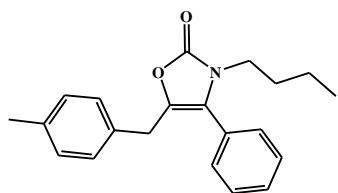
5-Benzyl-3-butyl-4-(4-chloro-phenyl)-3H-oxazol-2-one (2d). pale yellow oil; ^1H NMR (400 MHz, CDCl_3) δ 7.46 (d, $J = 8.3$ Hz, 2H), 7.33 to 7.19 (m, 5H), 7.16 (d, $J = 7.2$ Hz, 2H), 3.68 (s, 2H), 3.53 to 3.43 (m, 2H), 1.41 (dd, $J = 15.0, 7.7$ Hz, 2H), 1.24 to 1.08 (m, 2H), 0.79 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.34, 136.62, 135.66, 135.57, 130.94, 129.42, 128.71, 128.38, 126.91, 125.27, 122.94, 42.06, 30.99, 30.72, 19.60, 13.48.



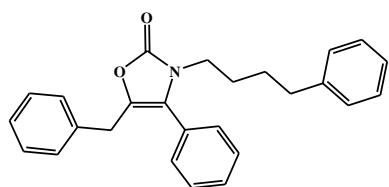
5-Benzyl-3-butyl-4-naphthalen-1-yl-3H-oxazol-2-one (2e). colorless oil; ^1H NMR (400 MHz, CDCl_3) δ 8.00 (d, $J = 8.2$ Hz, 1H), 7.97 to 7.90 (m, 1H), 7.74 (d, $J = 7.6$ Hz, 1H), 7.64 to 7.50 (m, 3H), 7.48 (d, $J = 6.8$ Hz, 1H), 7.25 to 7.13 (m, 3H), 7.09 (d, $J = 6.8$ Hz, 2H), 3.60 (q, $J = 15.8$ Hz, 2H), 3.47 (dt, $J = 14.3, 7.3$ Hz, 1H), 3.22 to 3.05 (m, 1H), 1.37 to 1.19 (m, 2H), 1.14 to 0.98 (m, 2H), 0.62 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.63, 136.61, 136.32, 133.66, 132.61, 130.54, 129.86, 128.67, 128.61, 128.51, 127.35, 126.74, 126.68, 125.28, 124.87, 123.80, 121.84, 42.31, 31.26, 30.80, 19.45, 13.33.



5-Benzyl-3-butyl-4-furan-2-yl-3H-oxazol-2-one (2f). croci oil; ^1H NMR (400 MHz, CDCl_3) δ 7.63 to 7.51 (m, 1H), 7.37 to 7.27 (m, 2H), 7.24 (dd, $J = 8.1, 4.7$ Hz, 3H), 6.50 (dt, $J = 11.8, 2.7$ Hz, 2H), 3.85 (s, 2H), 3.72 to 3.59 (m, 2H), 1.64 to 1.49 (m, 2H), 1.37 to 1.21 (m, 2H), 0.87 (t, $J = 7.3$ Hz, 3H). ^{13}C NMR (100 MHz, CDCl_3) δ 155.01, 143.75, 140.90, 137.16, 136.41, 128.73, 128.46, 126.95, 115.56, 111.31, 111.11, 42.75, 31.47, 30.83, 19.71, 13.60.



3-Butyl-5-(4-methyl-benzyl)-4-phenyl-3H-oxazol-2-one (2g). pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 7.57 to 7.40 (m, 3H), 7.32 (dd, *J* = 6.5, 2.9 Hz, 2H), 7.16 to 7.02 (m, 4H), 3.66 (s, 2H), 3.58 to 3.43 (m, 2H), 2.31 (s, 3H), 1.50 to 1.37 (m, 2H), 1.24 to 1.10 (m, 2H), 0.77 (t, *J* = 7.3 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 155.53, 136.38, 135.44, 133.88, 129.69, 129.36, 129.32, 129.02, 128.31, 126.89, 123.86, 41.99, 30.61, 21.03, 19.59, 13.46.

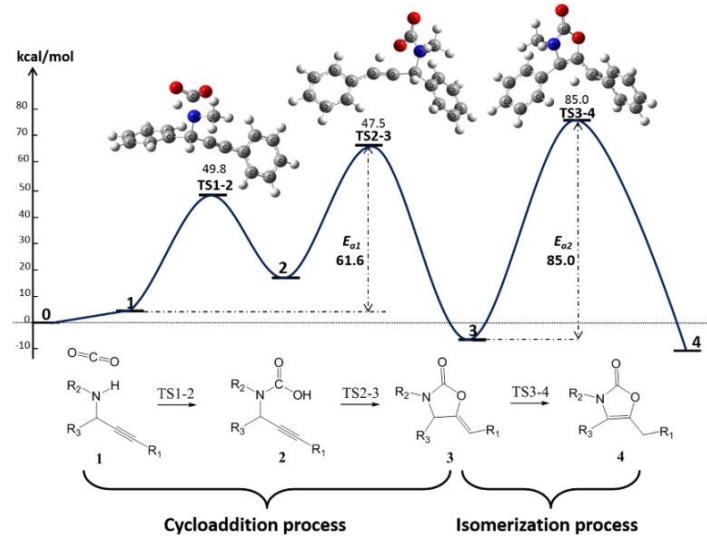


5-Benzyl-4-phenyl-3-(4-phenyl-butyl)-3H-oxazol-2-one (2h). pale yellow oil; ¹H NMR (400 MHz, CDCl₃) δ 7.51 to 7.38 (m, 3H), 7.33 to 7.19 (m, 7H), 7.15 (dd, *J* = 12.0, 7.0 Hz, 3H), 7.03 (d, *J* = 7.2 Hz, 2H), 3.68 (s, 2H), 3.52 (t, *J* = 6.7 Hz, 2H), 2.47 (t, *J* = 6.8 Hz, 2H), 1.59 to 1.37 (m, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 155.56, 141.77, 136.91, 135.27, 129.66, 129.45, 129.09, 128.68, 128.44, 128.35, 128.32, 126.84, 126.72, 125.82, 123.98, 42.04, 35.05, 30.97, 28.07, 27.97.

2. Theoretical Section

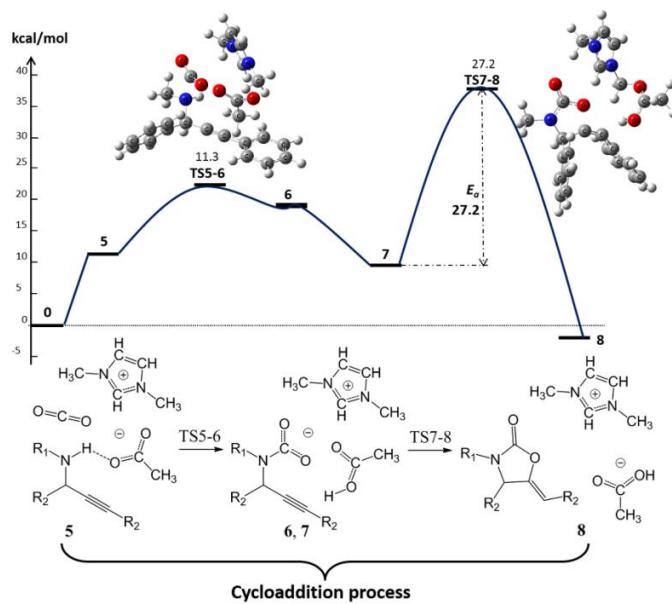
2.1. Figures S5-S7

Figure S5 Non-catalytic potential energy curve



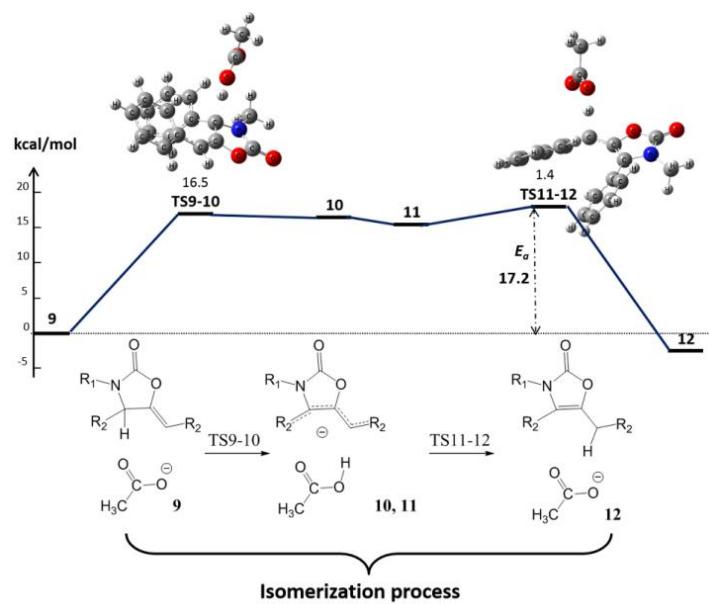
E_{a1} : the overall energy barrier (kcal/mol) of the cycloaddition process; E_{a2} : the overall energy barrier (kcal/mol) of the synthesis of 3,4,5-trisubstituted oxazolone from propargylic amines and CO₂.

Figure S6 [Bmim][OAc] promoted potential energy curve



E_a : the overall energy barrier (kcal/mol).

Figure S7 The potential energy curve of the double-bond isomerization promoted by $[\text{OAc}]^-$



E_a : the overall energy barrier (kcal/mol)

2.2. Optimized geometries of the transition states

TS1-2			
N	-0.61085976	0.79185519	0.00000000
H	0.62260424	1.03458119	0.34099900
C	-0.49071176	1.09231619	1.53987500
O	-1.39441676	1.09968519	2.32104500
O	0.77411724	1.31974819	1.56956400
C	-1.03764576	-0.60206781	-0.37632300
H	-0.90346376	-0.64409581	-1.46361500
C	-1.36192676	1.84439219	-0.71011700
H	-0.92923776	2.81416019	-0.46039700
H	-2.41452876	1.83129119	-0.41882800
C	-2.45253176	-0.83876581	-0.08735600
C	-3.62383976	-1.01626581	0.14020800
C	-0.09839376	-1.63406481	0.24061100
C	1.07334424	-1.97499081	-0.44137800
C	-0.36803576	-2.22750281	1.47539500
C	1.96786324	-2.89185181	0.10438300
H	1.28695824	-1.52446981	-1.40595000
C	0.52833324	-3.14308581	2.02264400
H	-1.27860276	-1.97437781	2.00505600
C	1.69625724	-3.47704581	1.33998800
H	2.87192624	-3.15084181	-0.43485800
H	0.31188724	-3.59597281	2.98338200
H	2.39053224	-4.19188081	1.76691700
C	-5.00492276	-1.20460881	0.44705900
C	-5.83555276	-1.95807381	-0.39964800
C	-5.55099276	-0.63063381	1.60868200
C	-7.18009376	-2.13152281	-0.08920400
H	-5.41748676	-2.40277381	-1.29480300
C	-6.89708276	-0.80797681	1.90959600
H	-4.91041076	-0.05060281	2.26214100
C	-7.71503276	-1.55739281	1.06393200
H	-7.81191976	-2.71567381	-0.74867900
H	-7.30836876	-0.36078181	2.80743400
H	-8.76362076	-1.69402181	1.30258400
H	-1.28028876	1.68463819	-1.78742700

Optimized geometries of transition states TS1-2. The imaginary vibrational frequency (cm^{-1}) is shown.

1724.1 *i*

TS2-3			
N	0.18099548	-0.04524887	0.00000000
H	2.62030348	-0.61176987	-1.59879700
C	0.64726748	0.76112413	-1.08638900
O	0.40111348	1.95444013	-1.10805000
O	1.29752648	0.03621813	-1.94817600
C	0.56552848	-1.44313887	0.05473300
H	0.69548948	-1.70682187	1.11843100
C	-0.65694152	0.49849713	1.05386900
H	-0.90791852	1.51804913	0.76734900
H	-0.13409452	0.52617813	2.01824700
C	1.92381048	-1.56231587	-0.48136400
C	3.11276948	-1.49163787	-0.88005700
C	-0.42391152	-2.44967287	-0.53823800
C	-0.79588752	-3.56815587	0.20985200
C	-0.96481652	-2.26176887	-1.81389600
C	-1.69942652	-4.49681787	-0.30682400
H	-0.38684452	-3.71467187	1.20544500
C	-1.86656252	-3.18951187	-2.32742800
H	-0.66942952	-1.39801387	-2.39643000
C	-2.23547052	-4.30775187	-1.57779200
H	-1.98707752	-5.35836587	0.28510900
H	-2.28277652	-3.03940187	-3.31718900
H	-2.94052052	-5.02514687	-1.98239500
C	4.49792248	-1.90663987	-0.81748000
C	4.90109748	-2.92999087	0.05627000
C	5.45074348	-1.28215087	-1.63461300
C	6.23475348	-3.31564287	0.10780500
H	4.16687448	-3.41908587	0.68559200
C	6.78503548	-1.67121887	-1.57407900
H	5.13993448	-0.49407487	-2.31023000
C	7.17944448	-2.68708287	-0.70517400
H	6.53921448	-4.10720587	0.78271400
H	7.51589148	-1.18171187	-2.20691700
H	8.21914648	-2.99017987	-0.66102500
H	-1.57856652	-0.08306787	1.16597200

Optimized geometries of transition states TS2-3. The imaginary vibrational frequency (cm^{-1}) is shown.

1354.5 *i*

TS3-4			
C	-0.13574661	0.38461538	0.00000000
C	0.66511739	0.78999638	-1.07397400
N	-0.62519461	1.64696138	0.51130900
C	-0.34535661	2.65027938	-0.38403300
O	-0.65228661	3.80903638	-0.36667000
O	0.44745639	2.09180838	-1.41557500
C	-1.44814561	1.86413438	1.69102200
H	-1.74490761	2.91176338	1.70589400
H	-0.88198061	1.63804138	2.59757100
C	1.68025539	-0.11617962	-1.46117000
H	0.93867739	-0.60615062	-0.10789000
H	1.82100739	-0.28342562	-2.53146800
C	-1.18072261	-0.71366562	-0.04249700
C	-1.04006061	-1.89147962	0.69368800
C	-2.34174761	-0.52037062	-0.80413200
C	-2.03168061	-2.87188862	0.65775800
H	-0.14913561	-2.04125062	1.29302800
C	-3.32794361	-1.50105362	-0.84457500
H	-2.46717961	0.39446338	-1.37314100
C	-3.17516261	-2.67970662	-0.11260100
H	-1.90815461	-3.78269162	1.23239600
H	-4.21369761	-1.34810262	-1.45076500
H	-3.94524361	-3.44202562	-0.14385800
C	2.93943239	-0.33630362	-0.69171000
C	3.78706739	-1.39176562	-1.06691900
C	3.34618039	0.49171738	0.36796300
C	4.98198939	-1.62541962	-0.39393600
H	3.49894739	-2.03652762	-1.89075600
C	4.53650139	0.24631038	1.04728900
H	2.73827839	1.34313938	0.65239400
C	5.36192439	-0.81347062	0.67453400
H	5.61780939	-2.44749562	-0.70441200
H	4.83037339	0.90325338	1.85886500
H	6.29336539	-0.99427562	1.19822900
H	-2.34007761	1.23464938	1.65891000

1671.9 *i*

Optimized geometries of transition states TS3-4. The imaginary vibrational frequency (cm^{-1}) is shown.

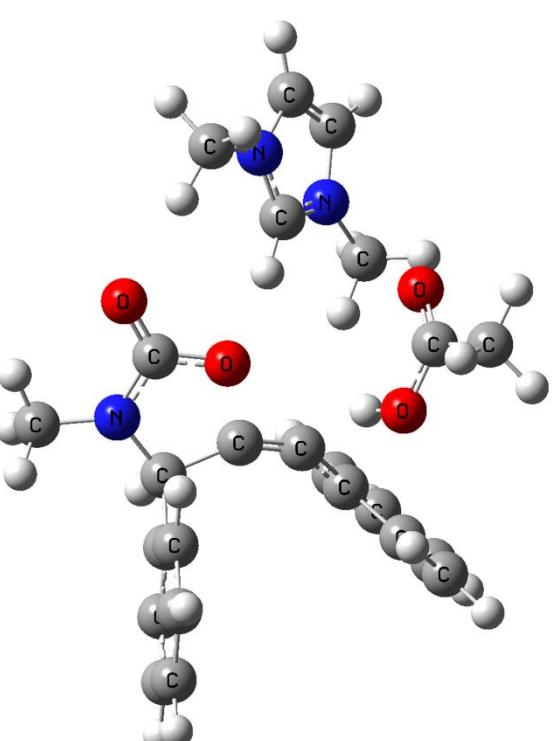
TS5-6			
O	-1.04865774	1.05704696	0.00000000
C	-1.38382474	0.10488196	0.64304500
O	-1.51468174	-1.07871404	0.75722400
C	2.05151926	0.35349296	1.07693200
N	2.17308826	0.81232096	-0.17692200
C	2.04702926	-0.24464304	-1.06121100
C	1.85590526	-1.36647104	-0.31155300
N	1.86029526	-0.97188804	1.01449600
C	2.26124926	2.22823596	-0.53890000
C	1.65784326	-1.84115504	2.18061000
H	2.51910226	2.80256296	0.35102900
H	3.03409526	2.36048596	-1.29992200
H	1.34217826	-1.21086204	3.01787700
H	2.58637926	-2.37022804	2.41400100
H	0.86544526	-2.55407804	1.94401800
H	1.69510726	-2.39294804	-0.60196500
H	2.08085726	-0.10477804	-2.13055000
H	2.08811426	0.96093896	2.00584700
H	1.29202326	2.56469996	-0.91494300
N	-1.93066974	0.81442096	2.47246100
C	-2.45160074	2.21519296	2.60599500
C	-2.78633874	-0.16898504	3.15694800
C	-1.43754874	3.19749196	2.20490300
H	-3.76760074	-0.21526804	2.67799400
H	-2.31154474	-1.14730604	3.08119800
H	-2.91038274	0.08632996	4.21846900
C	-0.60811274	4.05054196	1.96759900
H	-1.00630774	0.79200596	2.94434600
O	0.46907826	0.53053296	3.98503100
C	1.33026826	1.40800396	4.31537900
O	2.20533226	1.88925196	3.53694200
C	1.29536026	1.93638896	5.75093200
H	0.74523126	2.88678996	5.76540300
H	2.30810126	2.13842896	6.11372200
H	0.78537526	1.23497696	6.41739600
C	0.38314826	5.04314996	1.68392900
C	0.07128326	6.15731196	0.87847400
C	1.68430326	4.92004896	2.21657000
C	1.03833126	7.12668996	0.61229800
H	-0.93154974	6.25433896	0.47189400
C	2.64273726	5.89719796	1.94230200
H	1.92920326	4.05669196	2.83239500
C	2.32678026	7.00110096	1.14283000
H	0.78471126	7.98319896	-0.00764300
H	3.64103026	5.79613596	2.36157800
H	3.07749626	7.76025796	0.93719800
C	-3.79585474	2.42450096	1.90569900
C	-3.89191674	2.54379896	0.51162100
C	-4.96348074	2.51559696	2.67398800
C	-5.13431574	2.73432296	-0.09625100
H	-2.99157474	2.48998696	-0.09146600
C	-6.20817574	2.70946596	2.06726900
H	-4.90072674	2.43921996	3.75775200
C	-6.29645874	2.81701196	0.67787900
H	-5.19402774	2.82369696	-1.17826600
H	-7.10322974	2.77926796	2.68041900
H	-7.26160074	2.96951496	0.20120500
H	-2.62888074	2.37445896	3.68300000

Optimized geometries of transition states TS5-6. The imaginary vibrational frequency (cm^{-1}) is shown.

84.0*i*

TS7-8			
C	-3.25179500	-0.52165000	-0.42004900
N	-4.26686300	-1.38920100	-0.35036400
C	-5.42818200	-0.74748500	-0.74227200
C	-5.08710000	0.52965100	-1.06217600
N	-3.72485700	0.64671900	-0.86014200
C	-4.14737900	-2.78164200	0.10528500
C	-2.92496600	1.86580200	-1.02312000
H	-4.75495200	-3.41453300	-0.54151600
H	-3.09844300	-3.08284300	0.03118600
H	-3.09450200	2.27449900	-2.01966100
H	-1.87375900	1.61188500	-0.89393900
H	-3.20714400	2.58493900	-0.25584000
H	-5.68677100	1.35137800	-1.41319600
H	-6.37975700	-1.24933000	-0.76548200
H	-2.21379000	-0.73897600	-0.16098200
H	-4.49936700	-2.85907900	1.13514100
C	0.66722200	1.03517600	-0.34716500
C	0.85081900	-0.19018000	-0.53432500
O	-0.44849100	-1.31229000	0.09697700
C	1.74308800	-1.27603000	-1.06582300
H	2.06293200	-0.96635400	-2.07061000
C	-0.26848700	-2.45108900	-0.52212900
O	-1.08368800	-3.38022600	-0.55045600
N	0.94629100	-2.48668600	-1.17293000
C	1.40725500	-3.62971400	-1.93070000
H	2.35490900	-4.01014000	-1.53490400
H	0.65068900	-4.40882700	-1.85594700
H	1.55152900	-3.37112800	-2.98720700
C	-1.63824700	1.14784500	2.73386900
C	-1.83501300	1.26853300	4.22622600
H	-2.89183300	1.39281100	4.45366300
H	-1.45113600	0.37170800	4.71872700
H	-1.26511400	2.11791300	4.60925700
O	-0.35427900	1.05555900	2.41034900
O	-2.55210800	1.13604600	1.92901200
H	-0.21702000	0.95744800	1.42420100
C	3.00612400	-1.44908300	-0.21853700
C	4.21579200	-0.90556600	-0.65289800
C	2.96696000	-2.12939900	1.00213700
C	5.37131400	-1.03361100	0.11798300
H	4.25492600	-0.37287500	-1.59802300
C	4.11941200	-2.26106300	1.77193500
H	2.03160400	-2.55392900	1.34887300
C	5.32555700	-1.71310200	1.33278900
H	6.30408700	-0.60551600	-0.23236900
H	4.07719300	-2.79078300	2.71743900
H	6.22208300	-1.81708800	1.93384900
C	1.13898800	2.36104500	-0.62174800
C	0.93039500	2.96101800	-1.88416500
C	1.78790100	3.13432000	0.36580700
C	1.36171800	4.25806100	-2.14460100
H	0.43501700	2.38982200	-2.66223700
C	2.22378900	4.42571100	0.09154600
H	1.95048900	2.70449700	1.34731500
C	2.01349600	5.00345700	-1.16189800
H	1.19217200	4.68731100	-3.12710500
H	2.73028900	4.98967200	0.86822800
H	2.35023400	6.01294800	-1.36768000

Optimized geometries of transition states TS7-8. The imaginary vibrational frequency (cm^{-1}) is shown.



277.0 *i*

TS9-10			
C	-0.84470988	0.20477815	0.00000000
C	-2.17459688	-0.01148885	-0.24382600
O	-2.97510388	1.12958815	-0.13986200
C	-3.02710488	-1.14450285	-0.55096700
H	-3.11434988	-1.94555785	0.84651800
C	-4.29013688	0.81841215	-0.42489200
O	-4.90658988	-3.63814385	1.33491000
O	-5.17518888	1.65435815	-0.38722400
C	-4.03625688	-3.27082585	2.11373300
C	-3.94234488	-3.81851285	3.53027000
H	-4.03614888	-2.99886285	4.24729200
H	-4.72731188	-4.55447085	3.70165000
H	-2.96087988	-4.27368485	3.68505400
O	-3.09896988	-2.39202485	1.85730500
N	-4.34724888	-0.49823085	-0.69957500
C	-5.60713588	-1.17477685	-0.93308100
H	-5.74468088	-1.98579685	-0.21432300
H	-6.40031488	-0.43546685	-0.81954200
H	-5.64874388	-1.58809285	-1.94588000
H	-0.59951988	1.25135115	0.15259900
C	0.26114712	-0.71429685	0.14785600
C	1.56759912	-0.17859885	0.27714000
C	0.15183812	-2.12309785	0.21670900
C	2.68285612	-0.98625885	0.45263100
H	1.69079712	0.89991715	0.23860300
C	1.27200812	-2.92758785	0.39725300
H	-0.82026188	-2.58914685	0.14941800
C	2.54944512	-2.37653185	0.51215600
H	3.66429512	-0.52970885	0.54454500
H	1.14178512	-4.00441185	0.44903000
H	3.41812612	-3.01221585	0.64683000
C	-2.66914988	-2.10945485	-1.63825300
C	3.34827188	-3.33877085	-1.74582900
C	-1.65174588	-1.84908185	-2.57209600
C	-3.04439788	-4.24481985	-2.75870600
H	-4.10102288	-3.59184885	-1.00795900
C	-1.33777088	-2.76303785	-3.57584000
H	-1.09980288	-0.91912985	-2.50163700
C	-2.03663788	-3.96440985	-3.68309800
H	-3.586668788	-5.18391485	-2.81411000
H	-0.54231188	-2.53231085	-4.27803700
H	-1.79153788	-4.67672585	-4.46437600

Optimized geometries of transition states TS9-10. The imaginary vibrational frequency (cm^{-1}) is shown.
Charge = -1 Multiplicity = 1

367.6 *i*

TS11-12			
C	0.38395903	0.97269623	0.00000000
C	-0.68446697	0.02210523	0.21703900
O	-0.25832097	-1.31435677	0.10027300
C	-2.00124797	-0.00212477	0.59465300
C	-1.27921997	-2.16060177	0.40988200
O	-1.21452797	-3.37239877	0.32674200
N	-2.34133797	-1.38401577	0.78552400
C	-3.67010797	-1.95396177	0.87944400
H	-4.35266797	-1.45990377	0.18209500
H	-3.58481997	-3.01020477	0.62358400
H	-4.08460697	-1.87036077	1.88807500
H	0.93149303	0.73044623	-0.91347300
C	0.25648103	2.41946323	0.22818000
C	-0.33259897	2.97042623	1.38602800
C	0.85837503	3.32947323	-0.66693800
C	-0.33374797	4.34079223	1.62171200
H	-0.77697397	2.30715923	2.11877900
C	0.86100003	4.70057723	-0.42892200
H	1.34584503	2.93909023	-1.55455000
C	0.25818503	5.22386723	0.71619900
H	-0.79781897	4.72329923	2.52600400
H	1.33929303	5.36502423	-1.14287400
H	0.25641103	6.29270723	0.90298300
C	-2.97883197	1.06430923	0.78118800
C	-3.88785197	1.05678423	1.85882900
C	-3.05119297	2.14384023	-0.12032600
C	-4.83374697	2.06555223	2.01358800
H	-3.83297497	0.265557923	2.59699900
C	-3.98947097	3.15657623	0.04385500
H	-2.36190297	2.17725223	-0.95425100
C	-4.89462597	3.12295623	1.10531600
H	-5.51554097	2.03381023	2.85774700
H	4.01352697	3.97781623	-0.66467900
H	-5.62740697	3.91310223	1.22899700
C	3.46197803	0.18928723	1.24271900
O	2.25754703	0.27368923	1.72547900
O	3.80573203	0.45143823	0.09313100
C	4.48203603	-0.29790877	2.27437600
H	4.22887103	-1.31693177	2.58096200
H	5.48727803	-0.28150477	1.85177900
H	4.44022503	0.33136423	3.16742200
H	1.41055103	0.59020123	0.90979000

Optimized geometries of transition states TS9-10. The imaginary vibrational frequency (cm^{-1}) is shown.
Charge = -1 Multiplicity = 1

1135.9 *i*

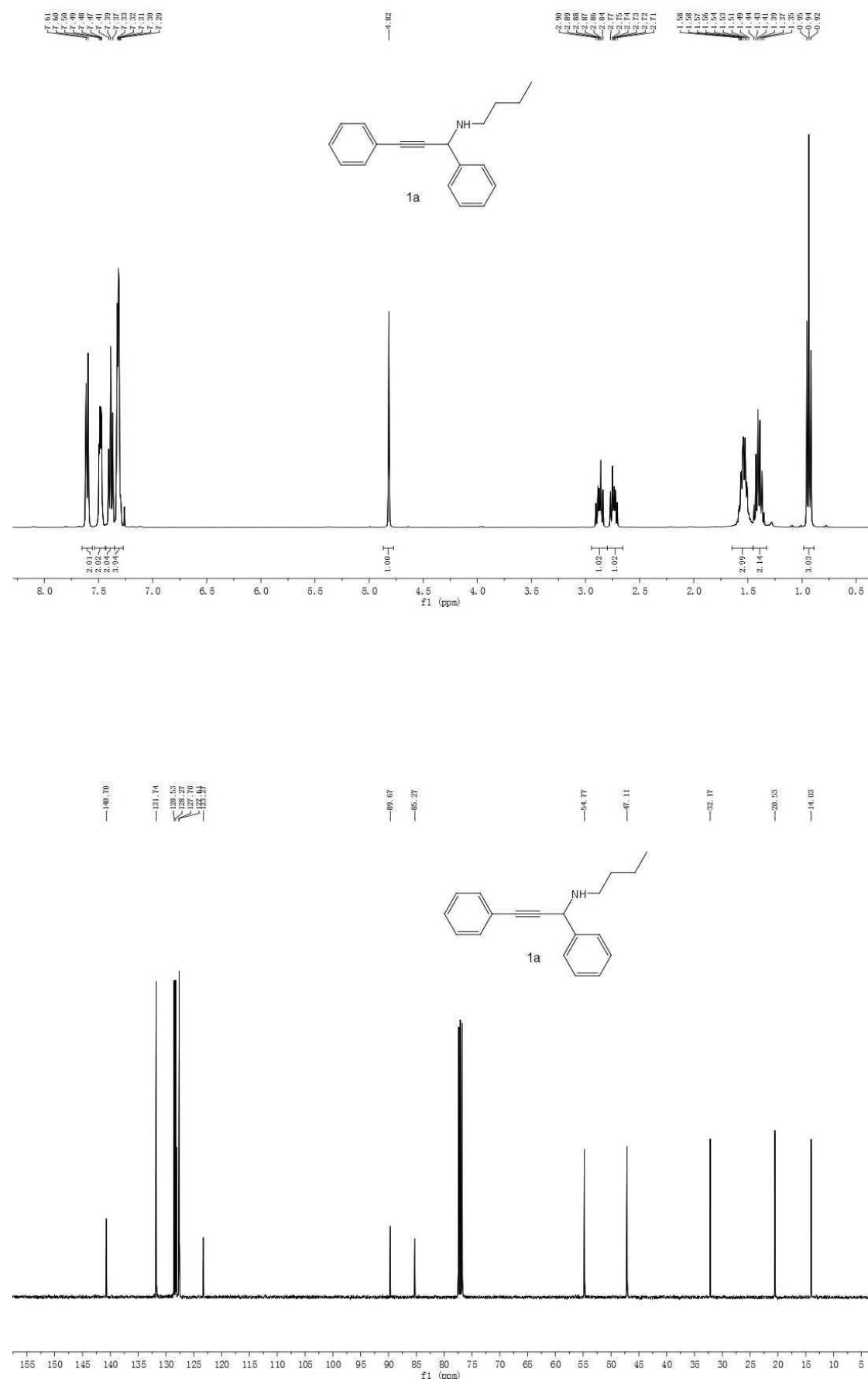
3. References

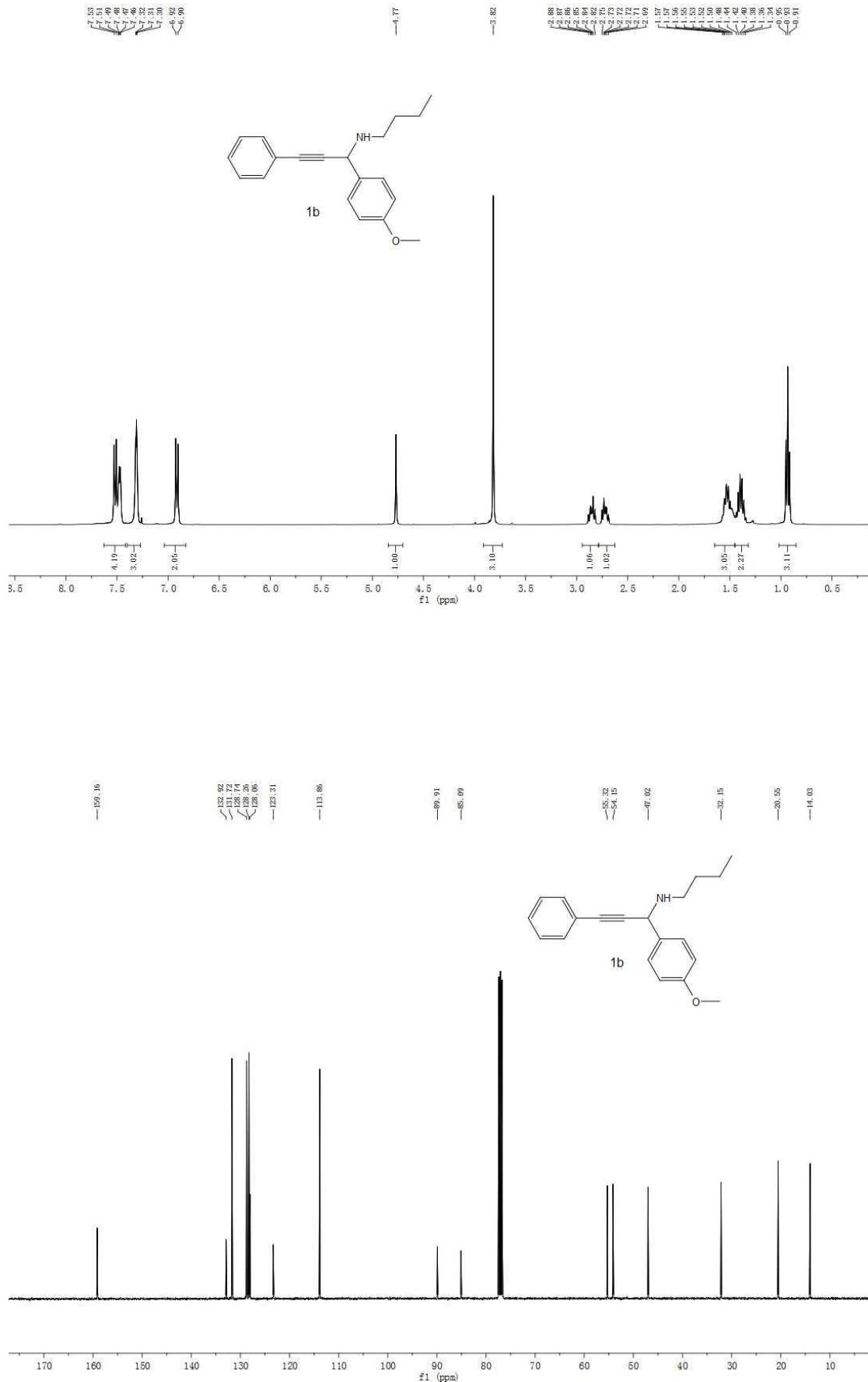
S1. Gaussian 09, Revision A.01, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, **2009**.

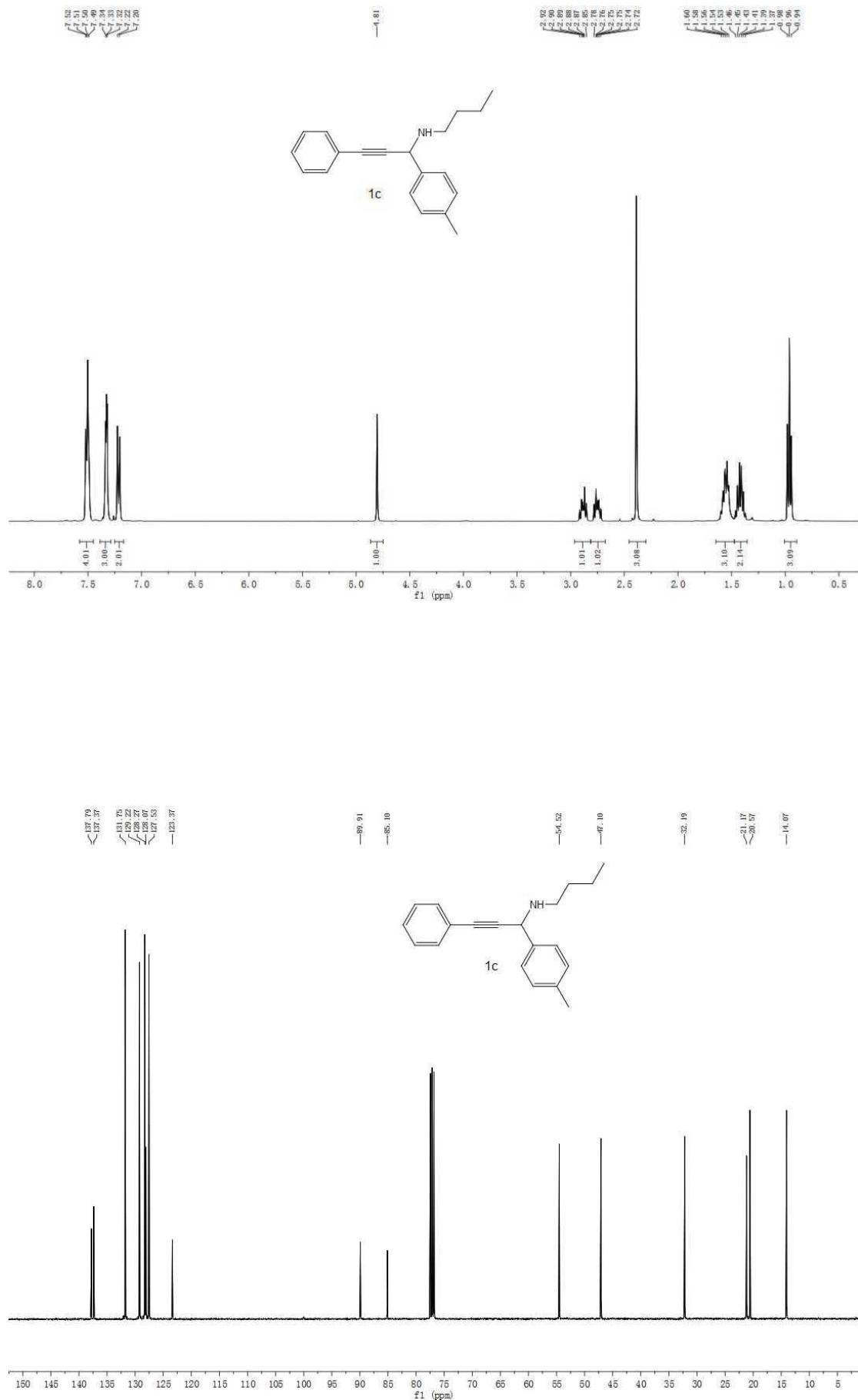
S2. Y. Sugawara, W. Yamada, S. Yoshida, T. Ikeno, T. Yamada, *J. Am. Chem. Soc.* 2007, **129**, 12902.

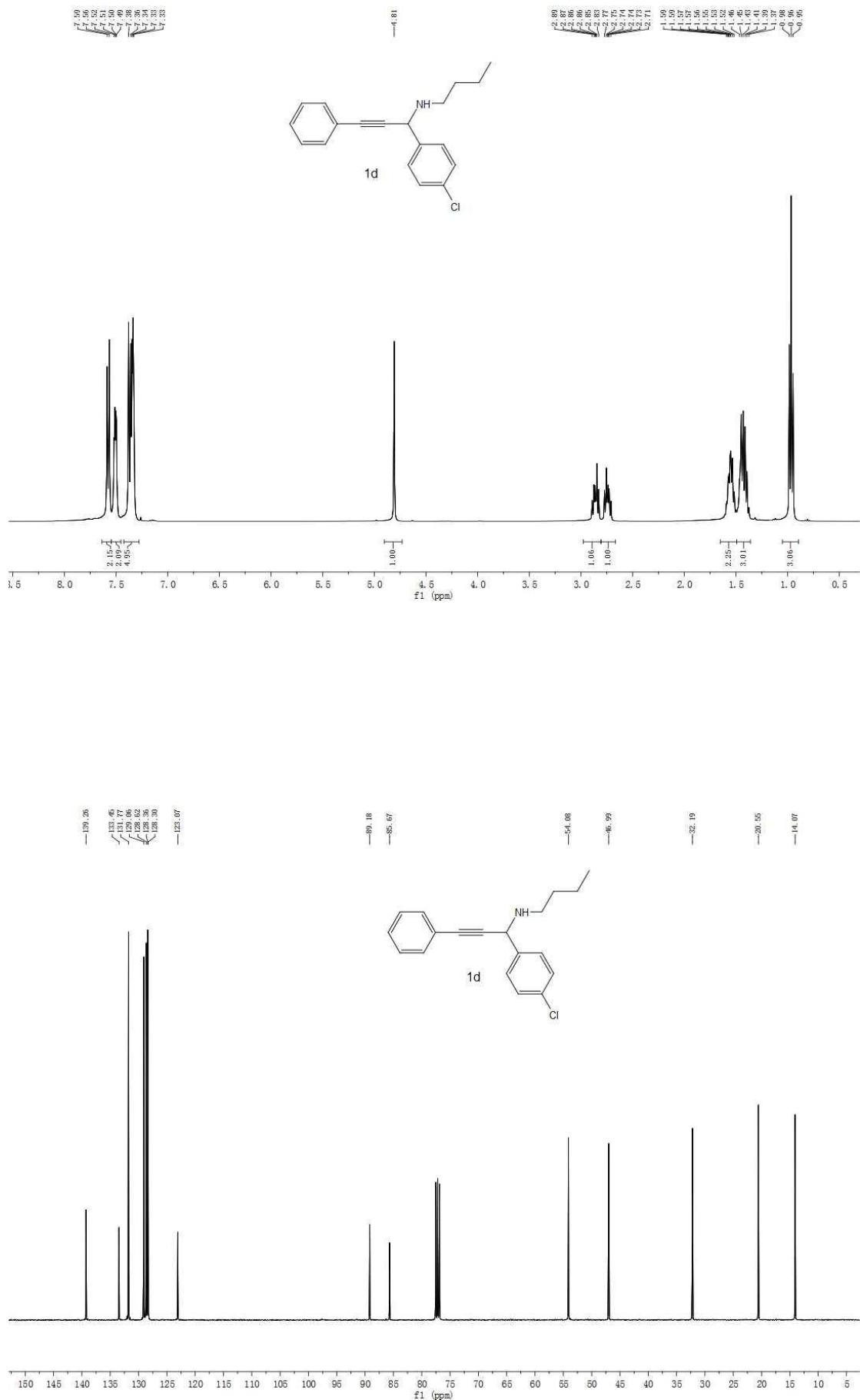
S3. W. J. Yoo and C. J. Li, *Adv. Synth. Catal.* 2008, **350**, 1503.

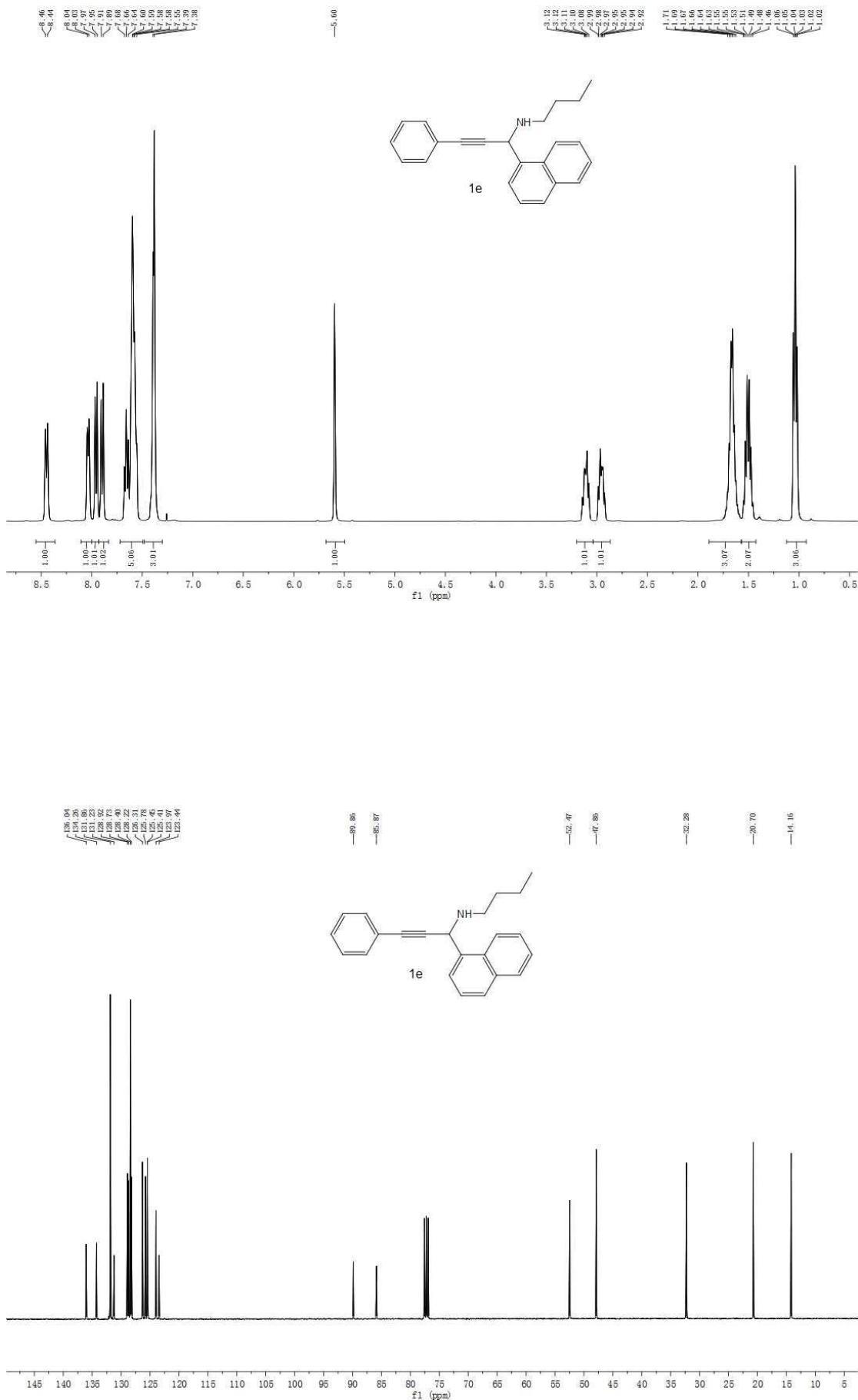
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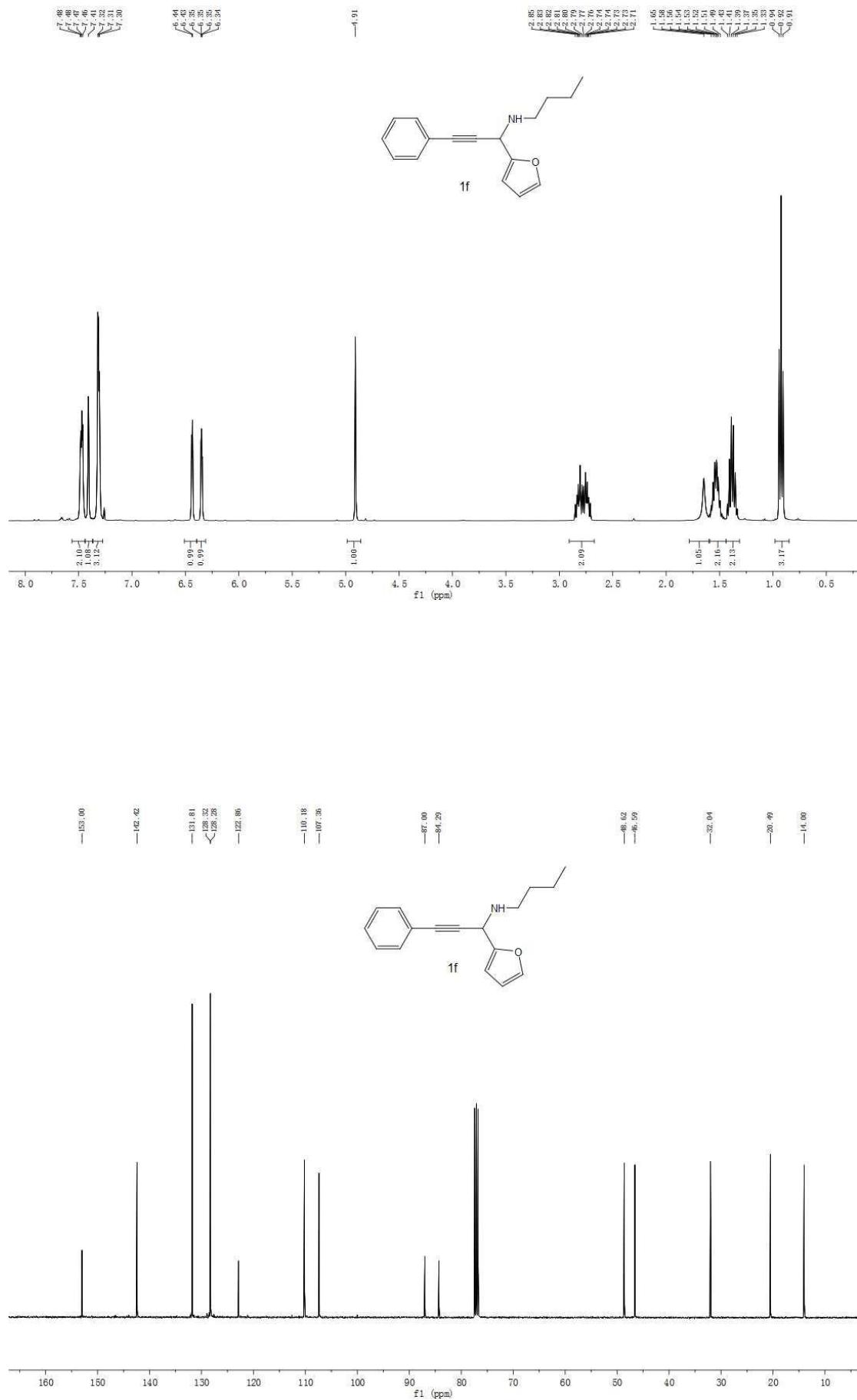


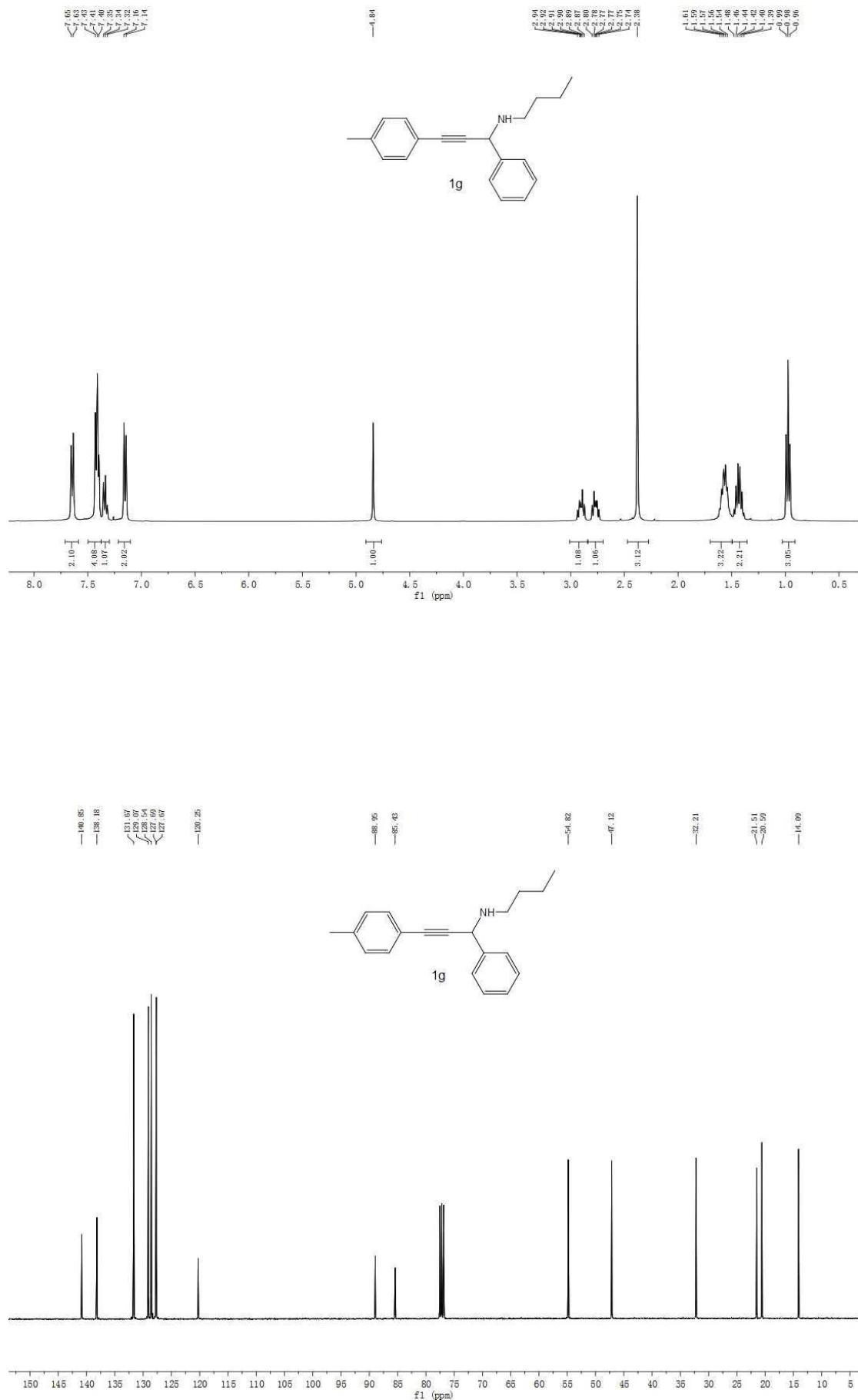


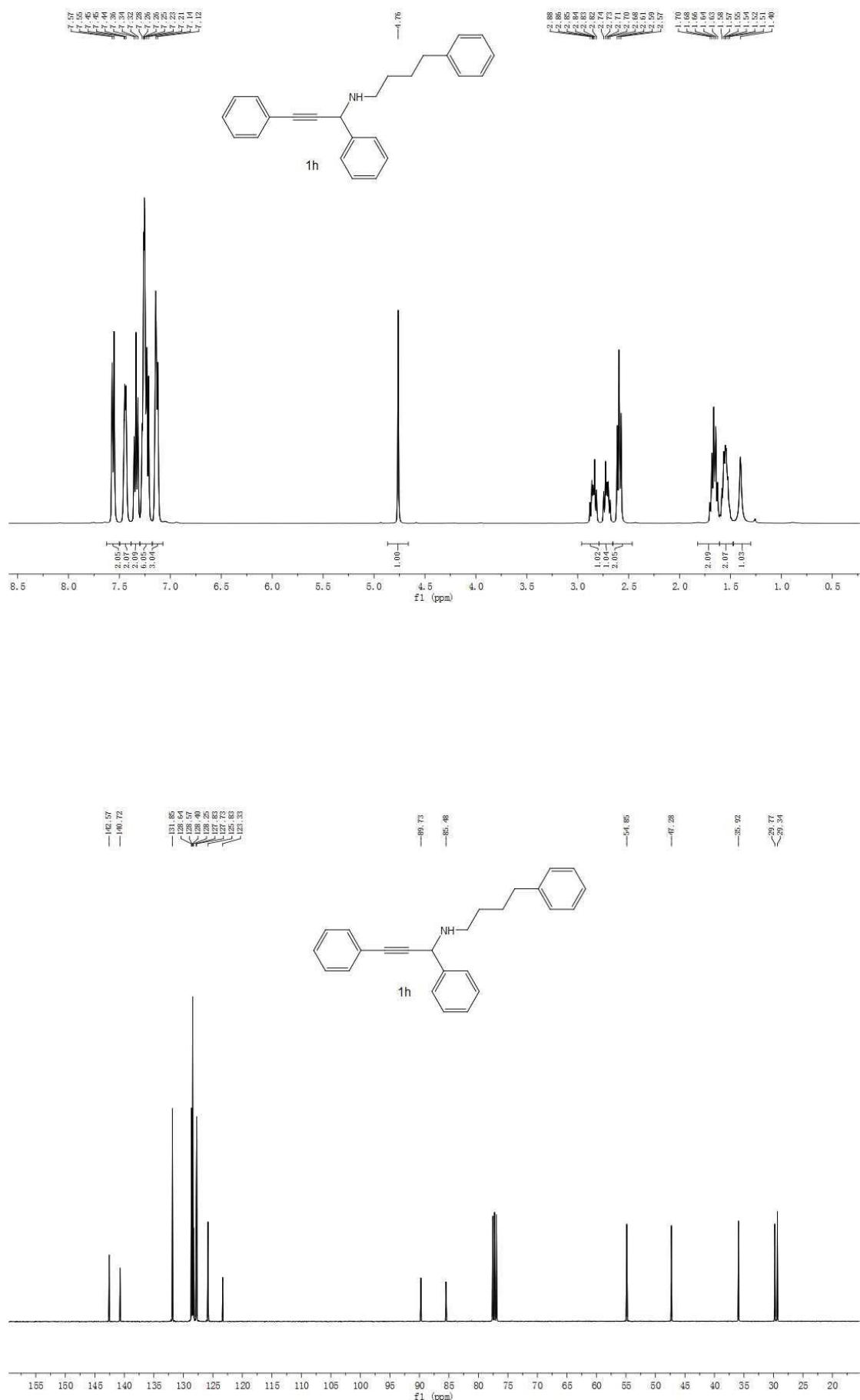


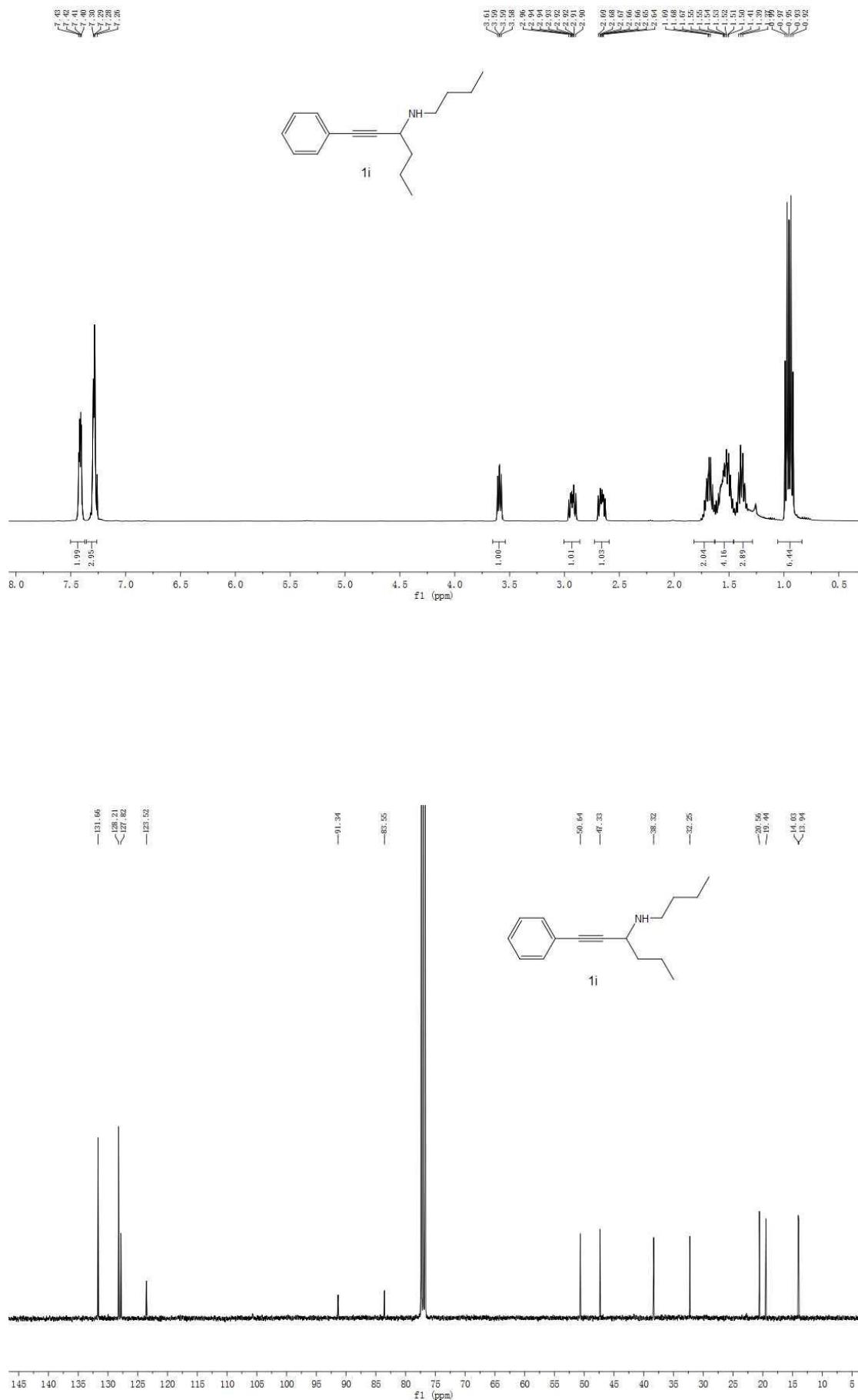


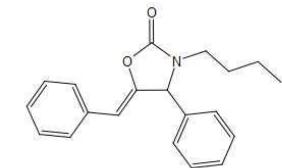




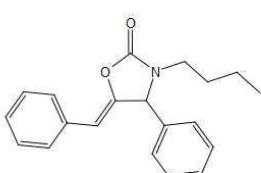
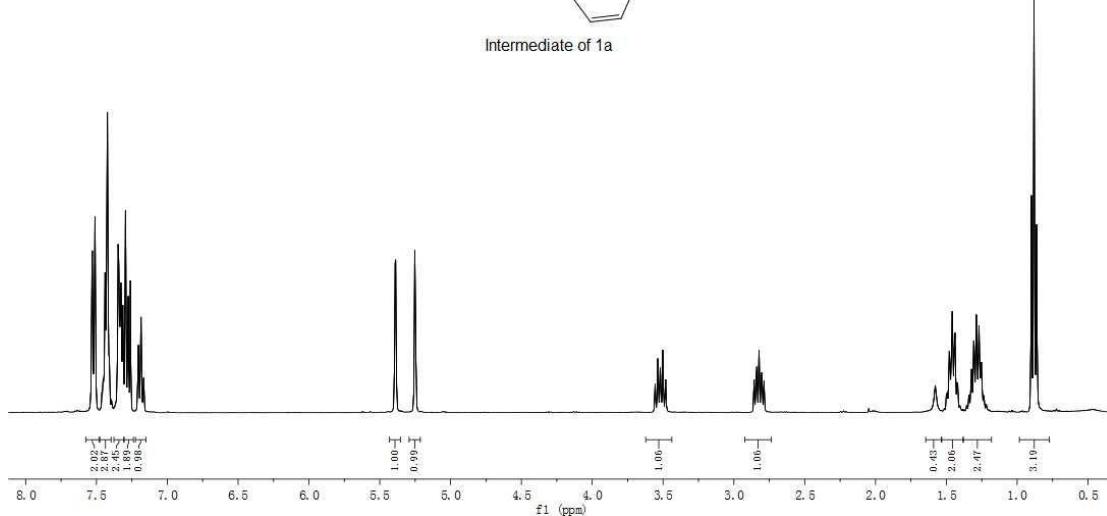




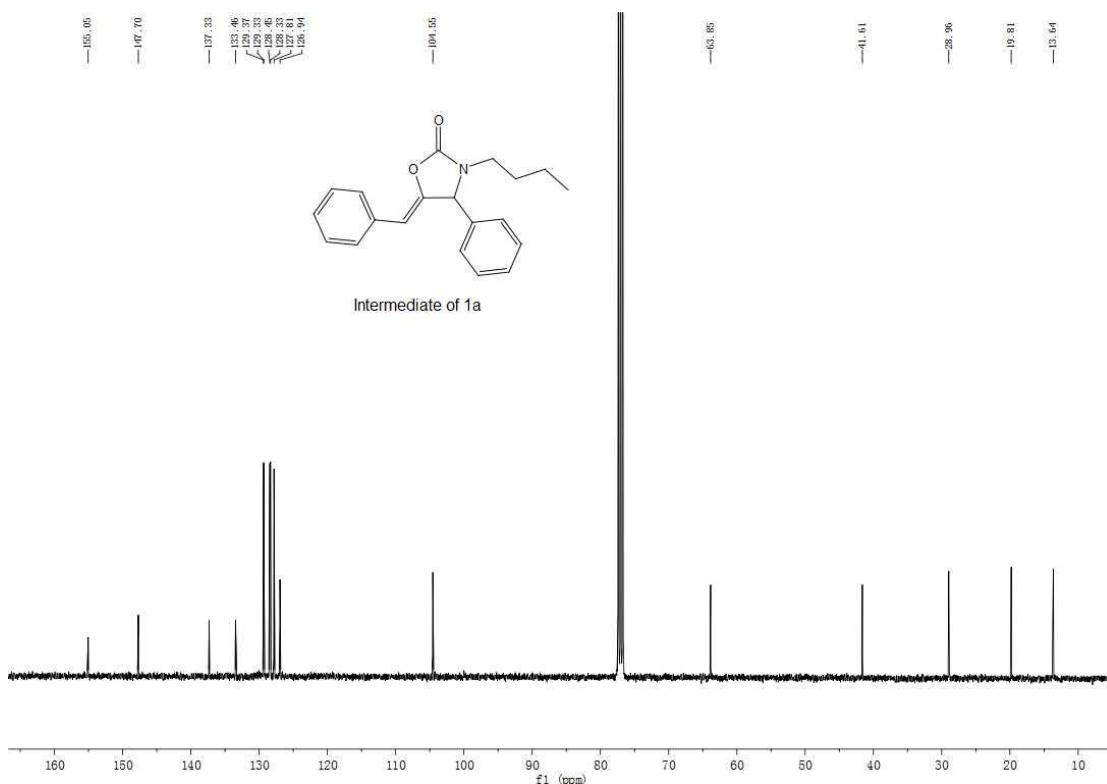


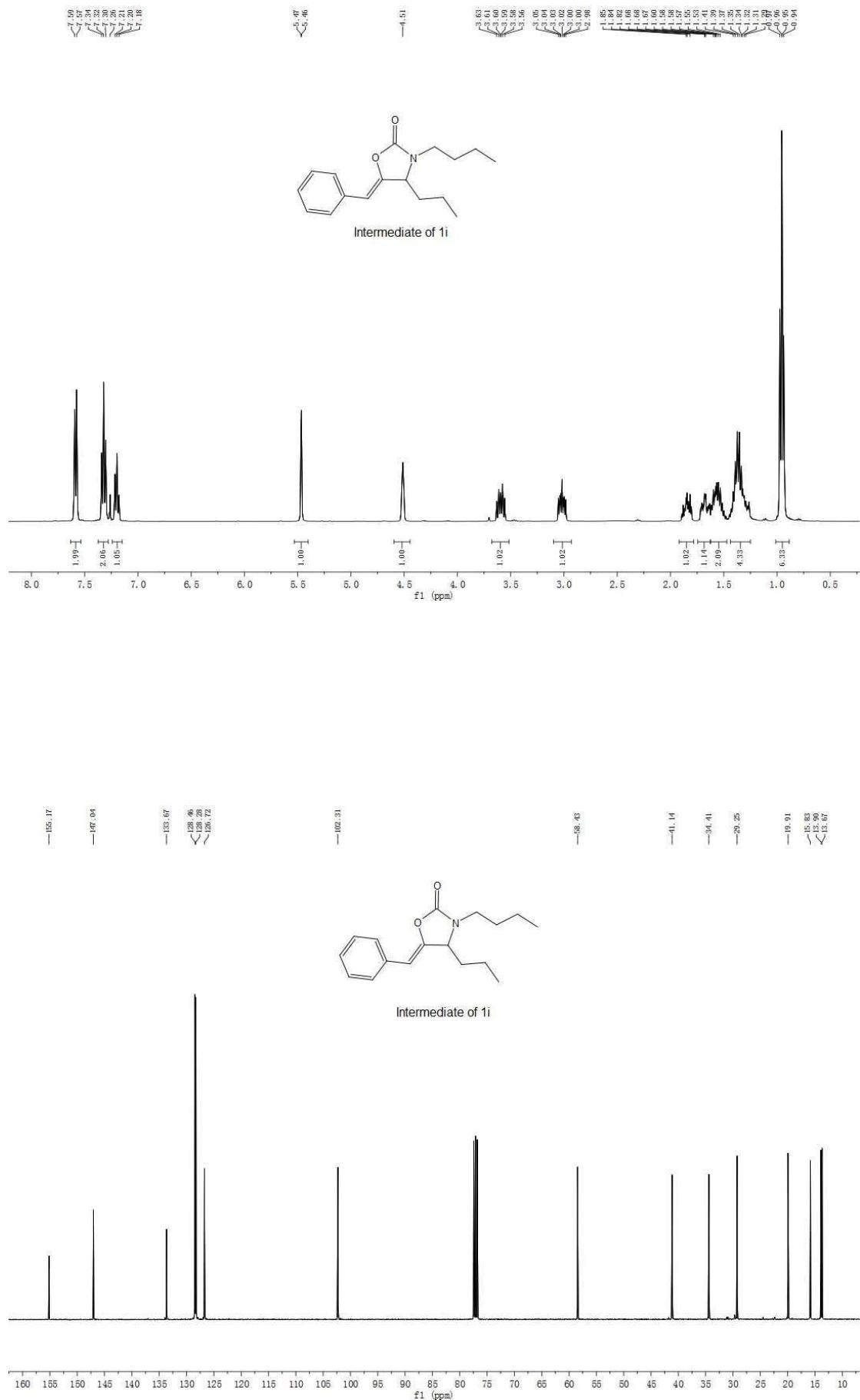


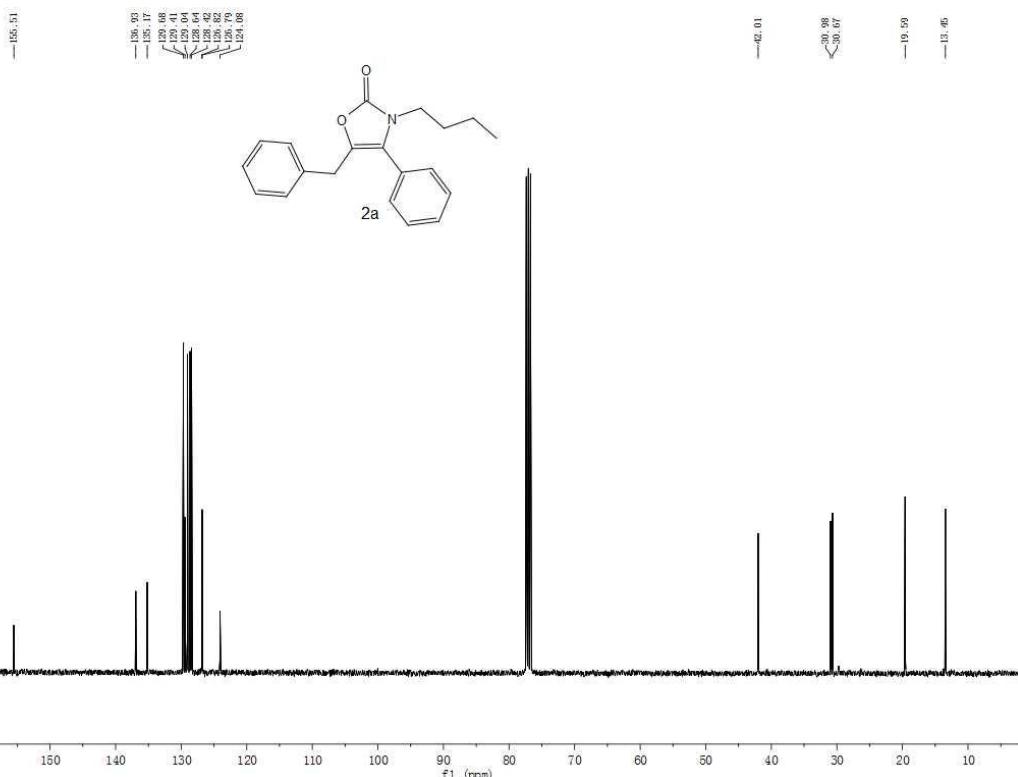
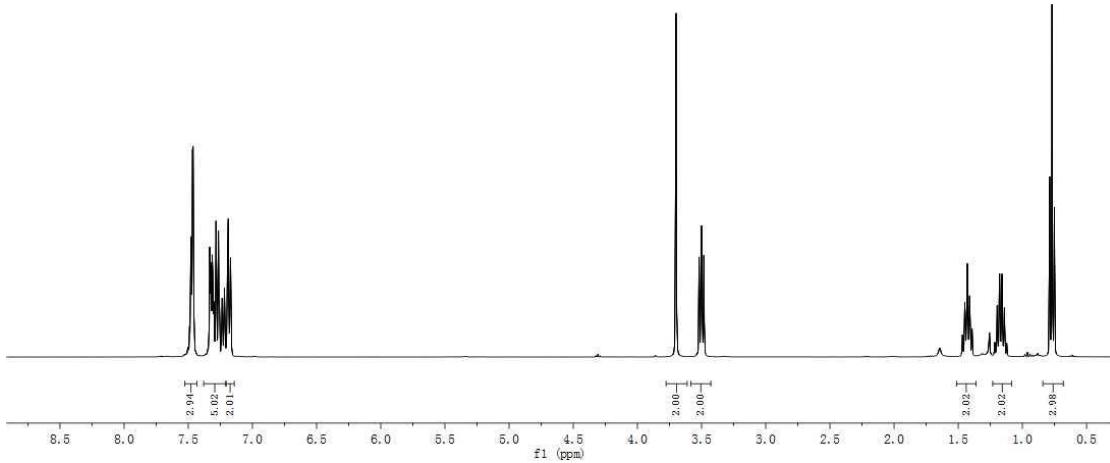
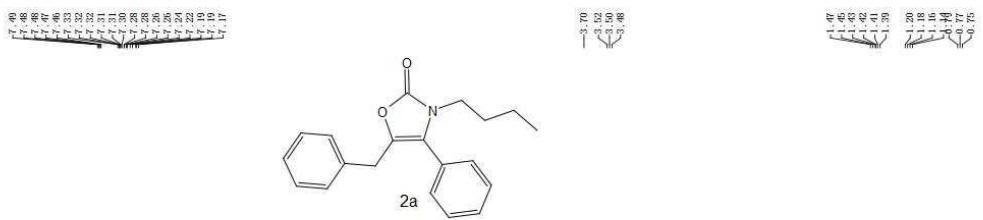
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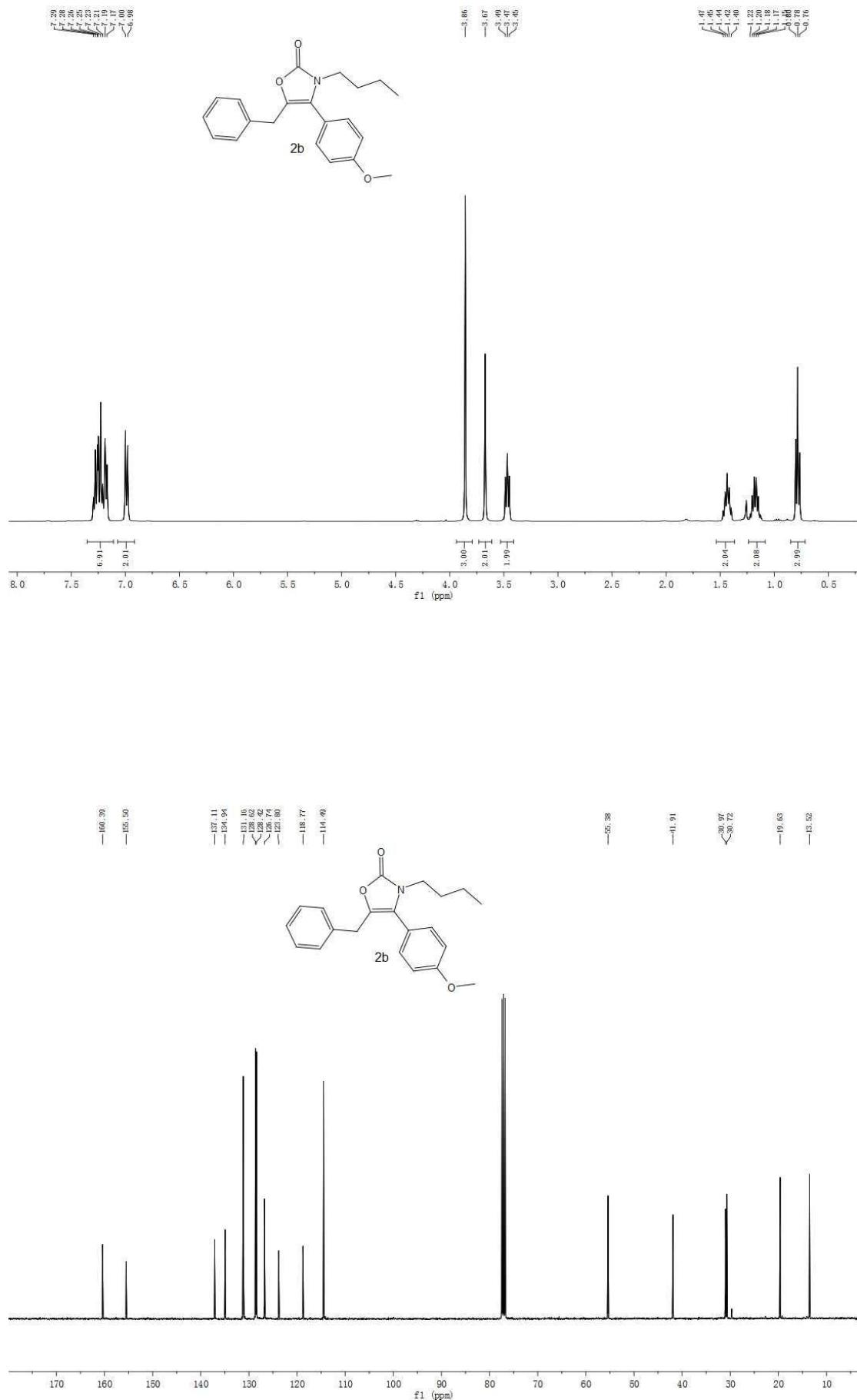


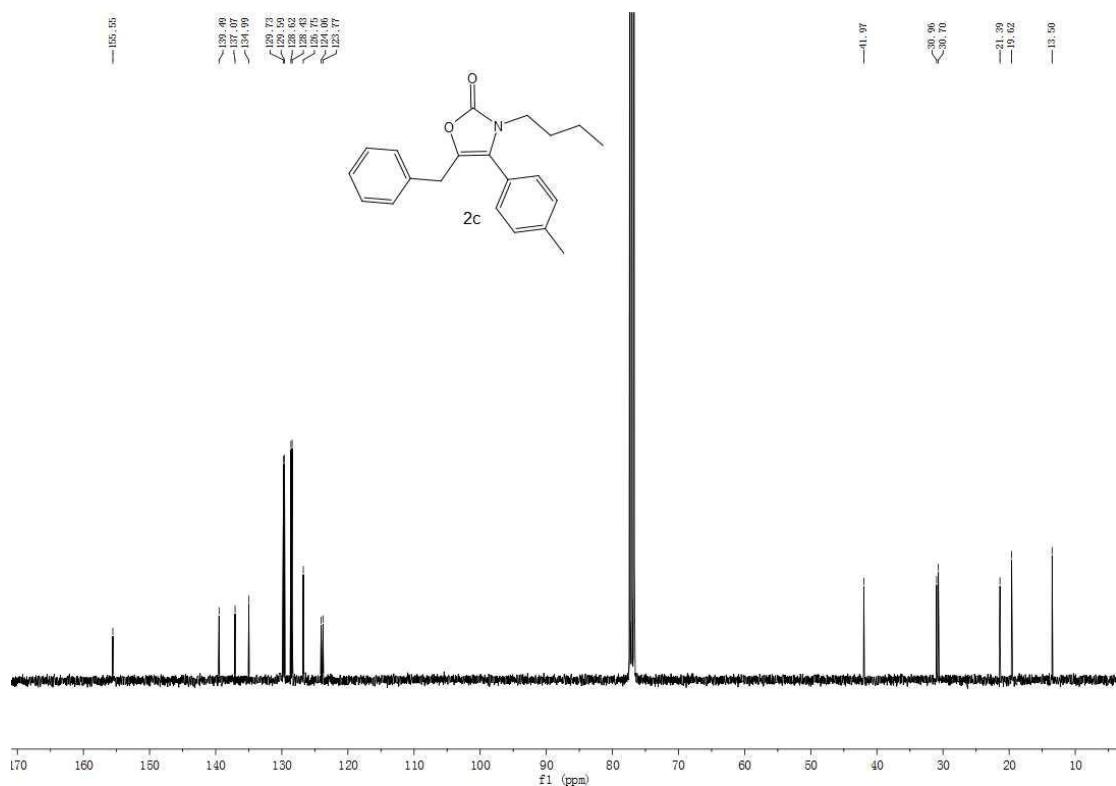
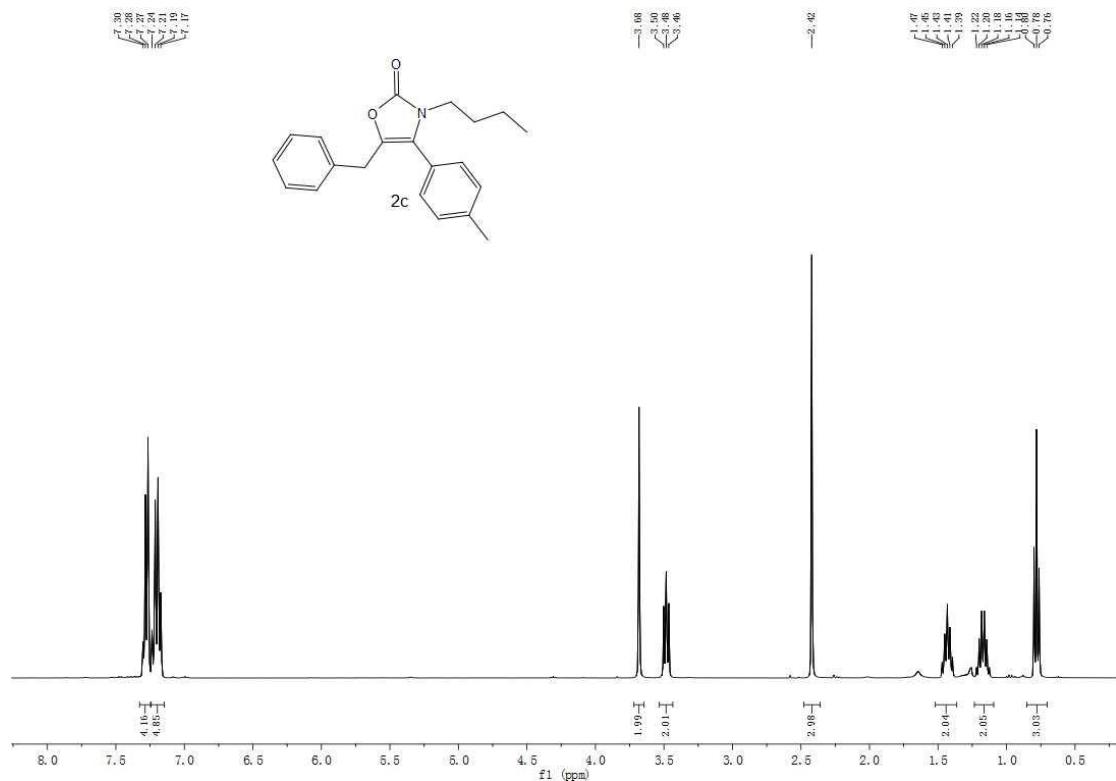
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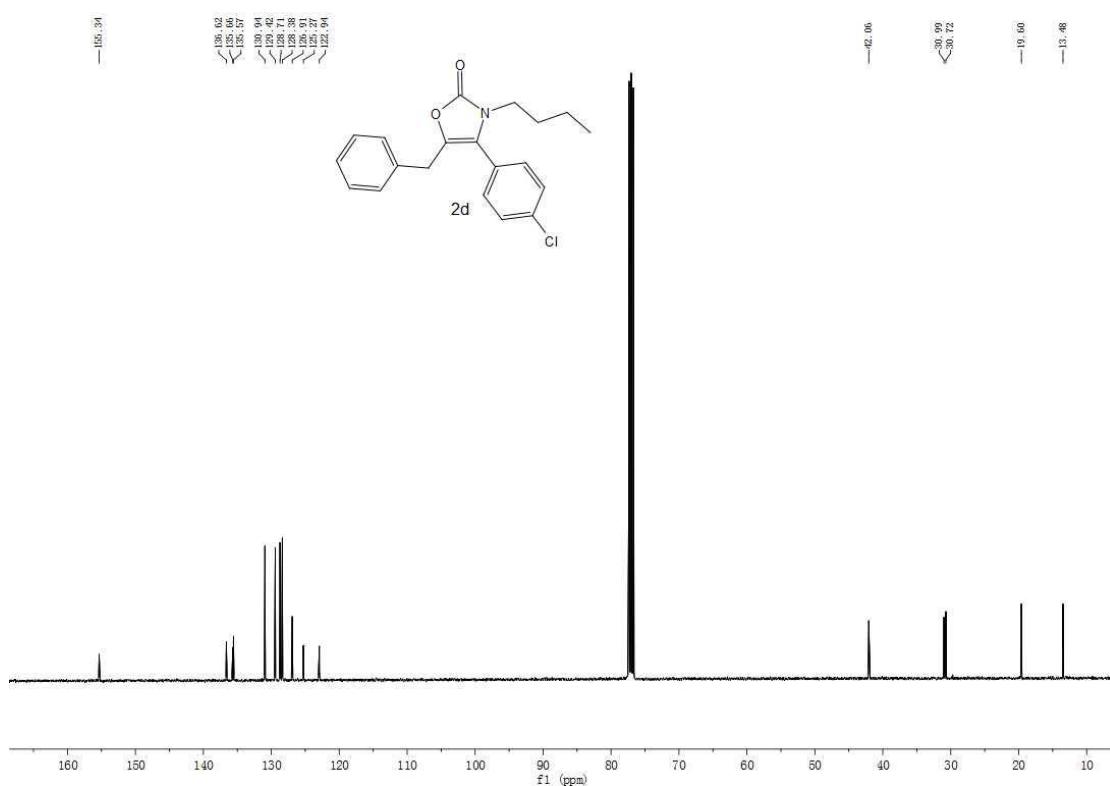
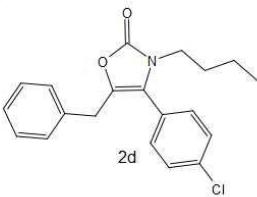
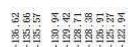
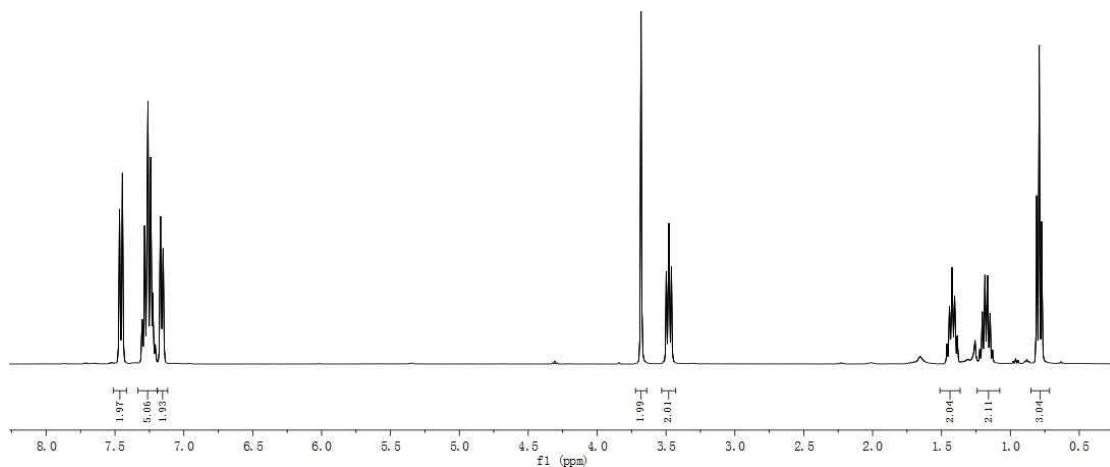
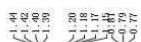
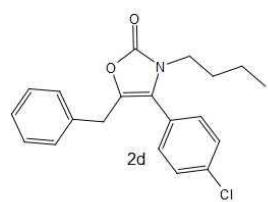


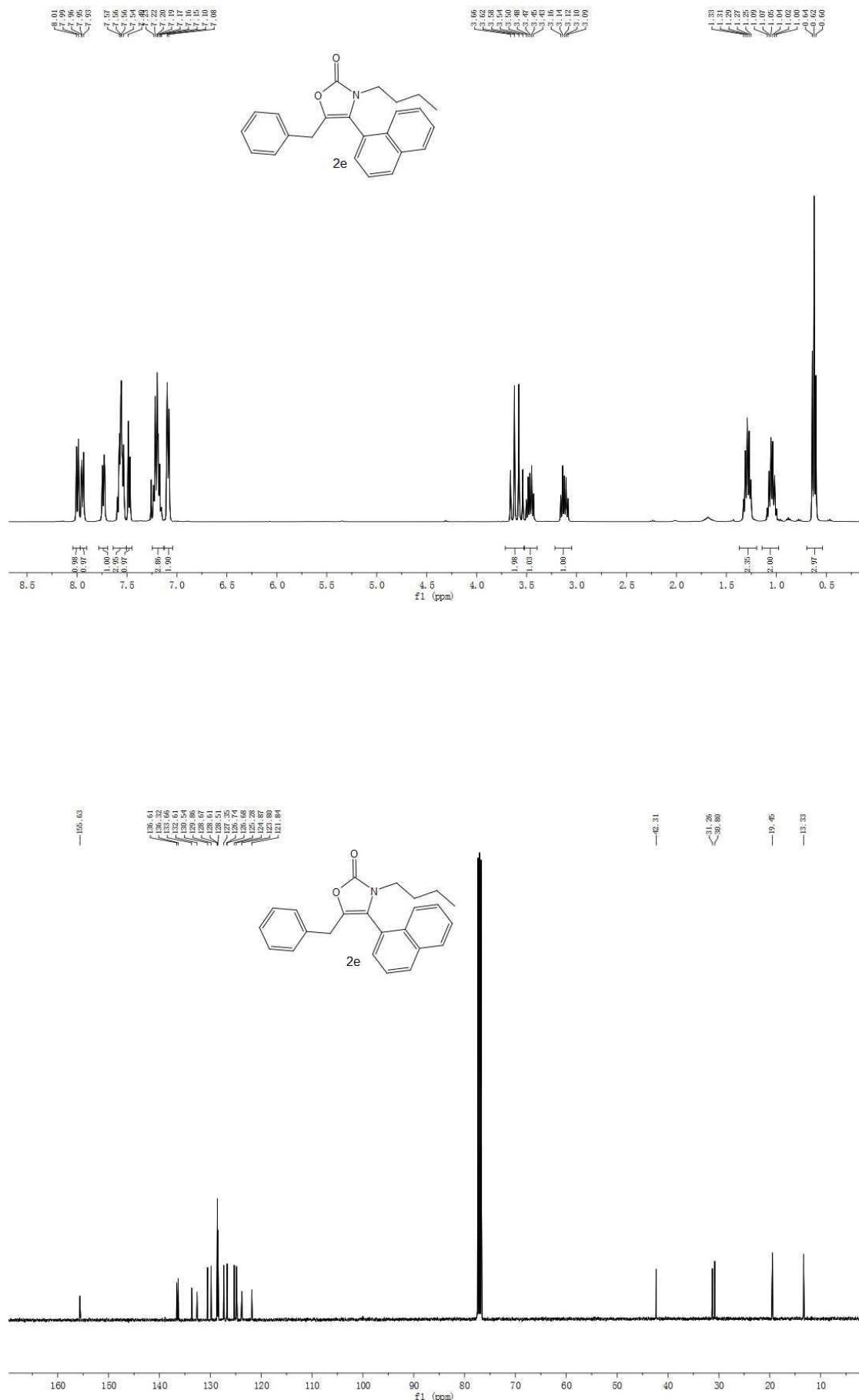


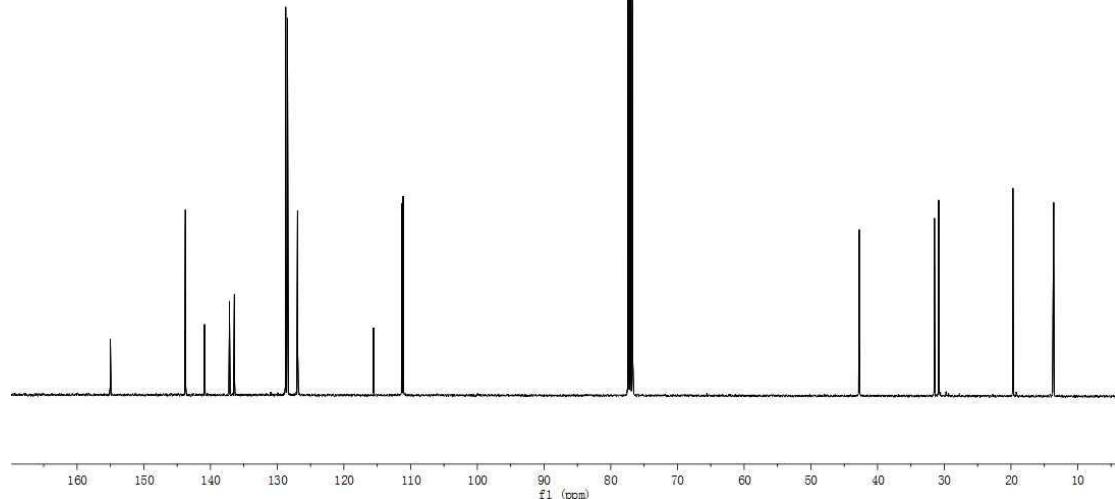
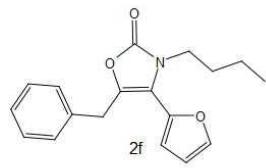
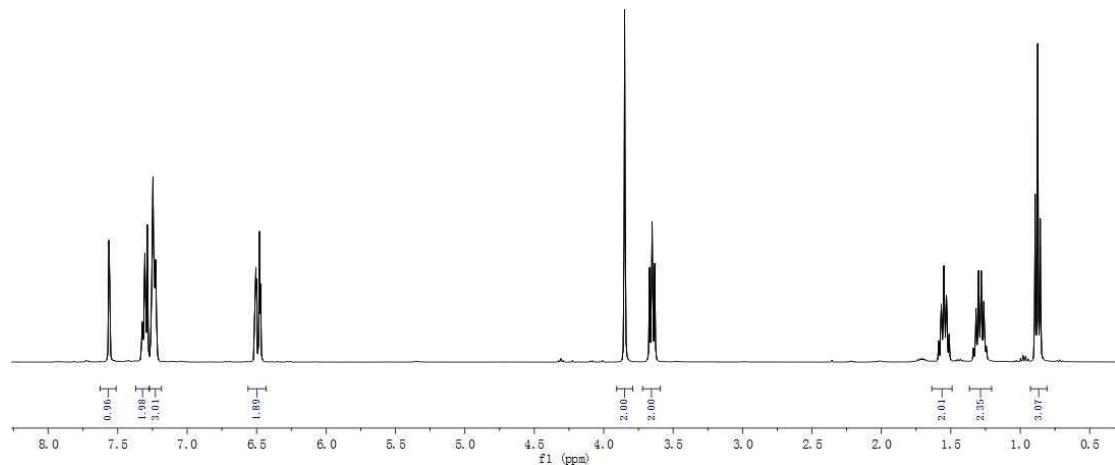
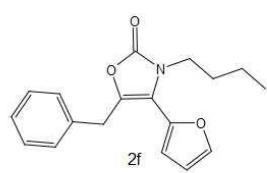


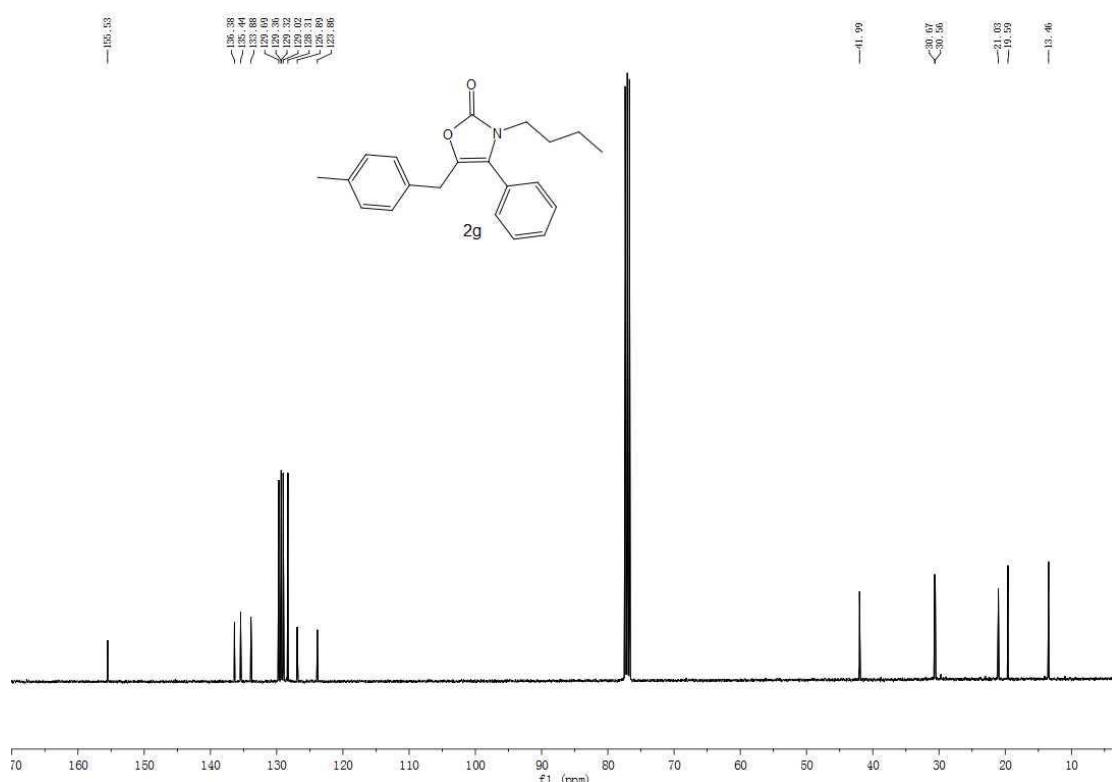
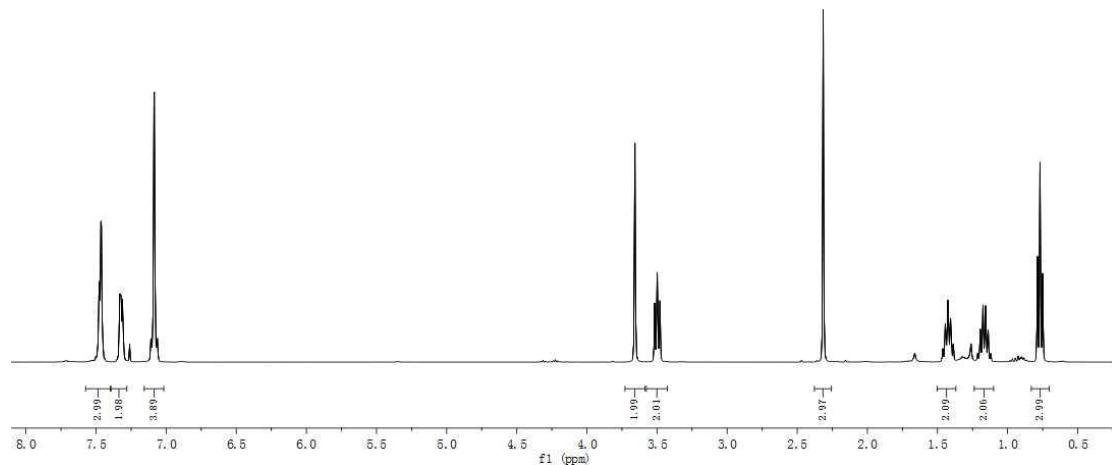


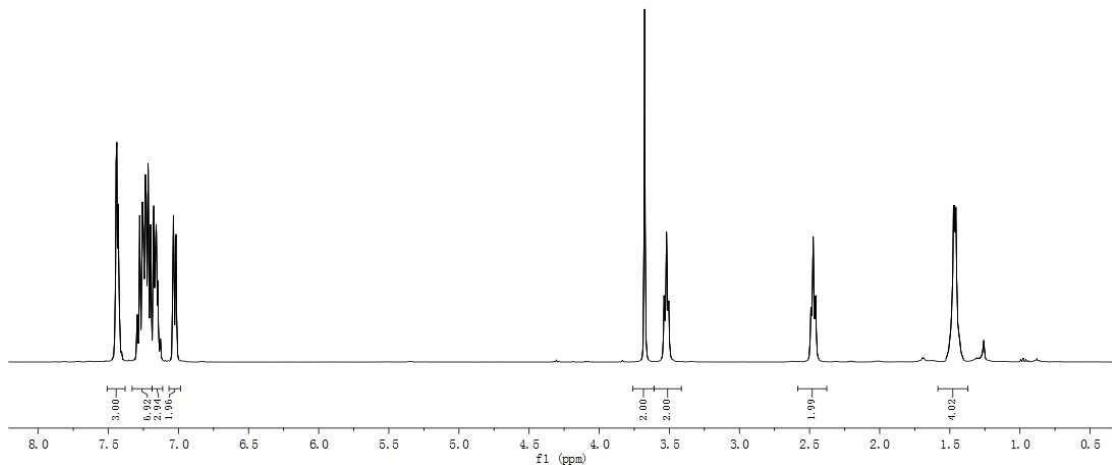
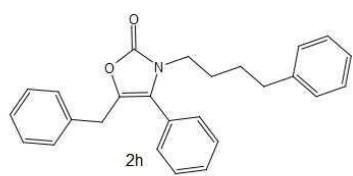








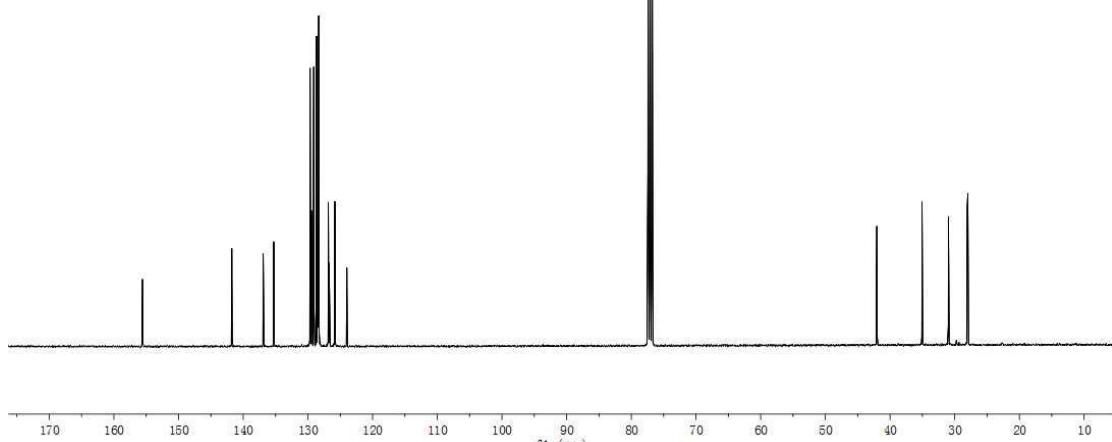
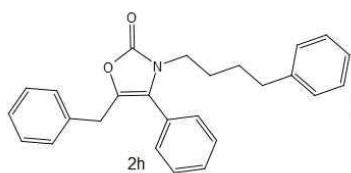


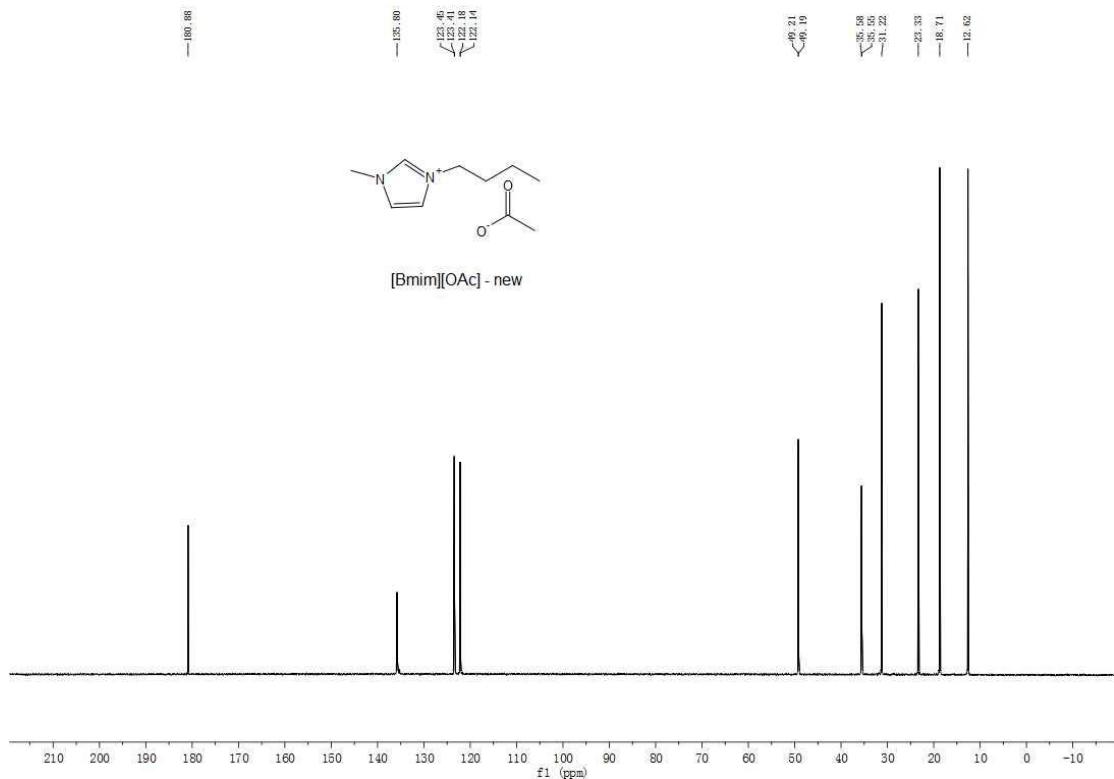
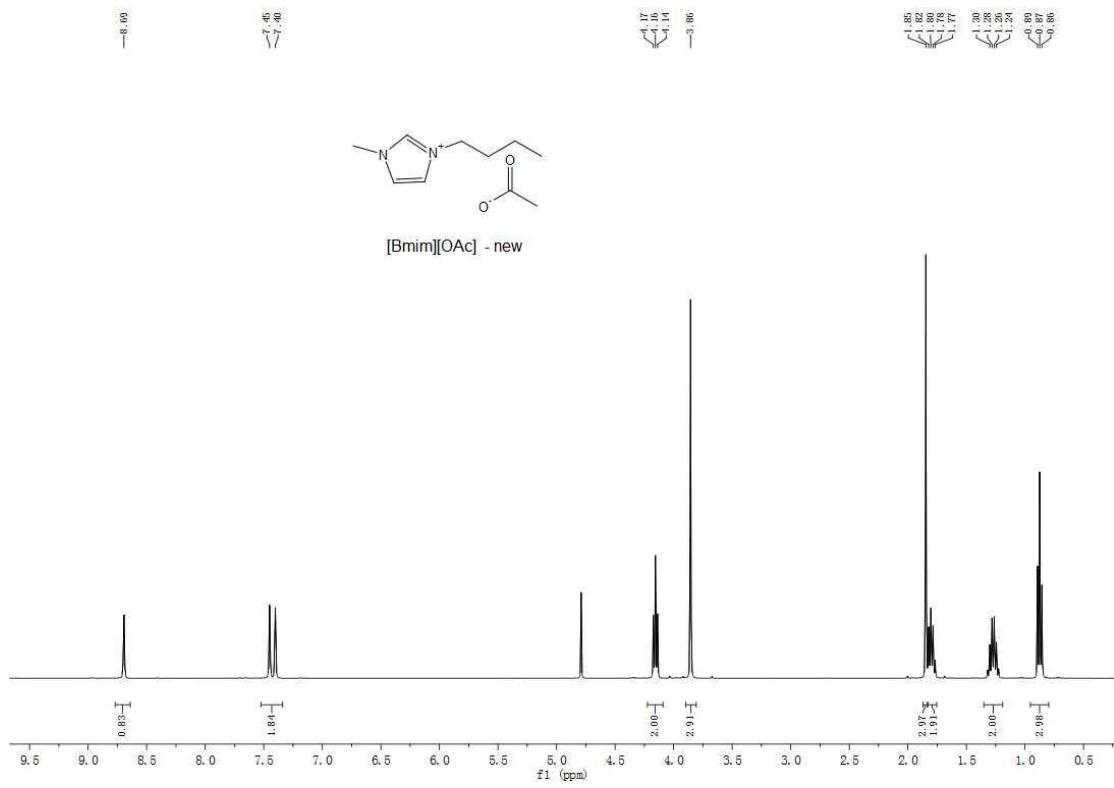


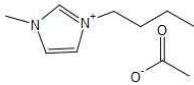
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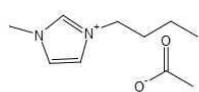
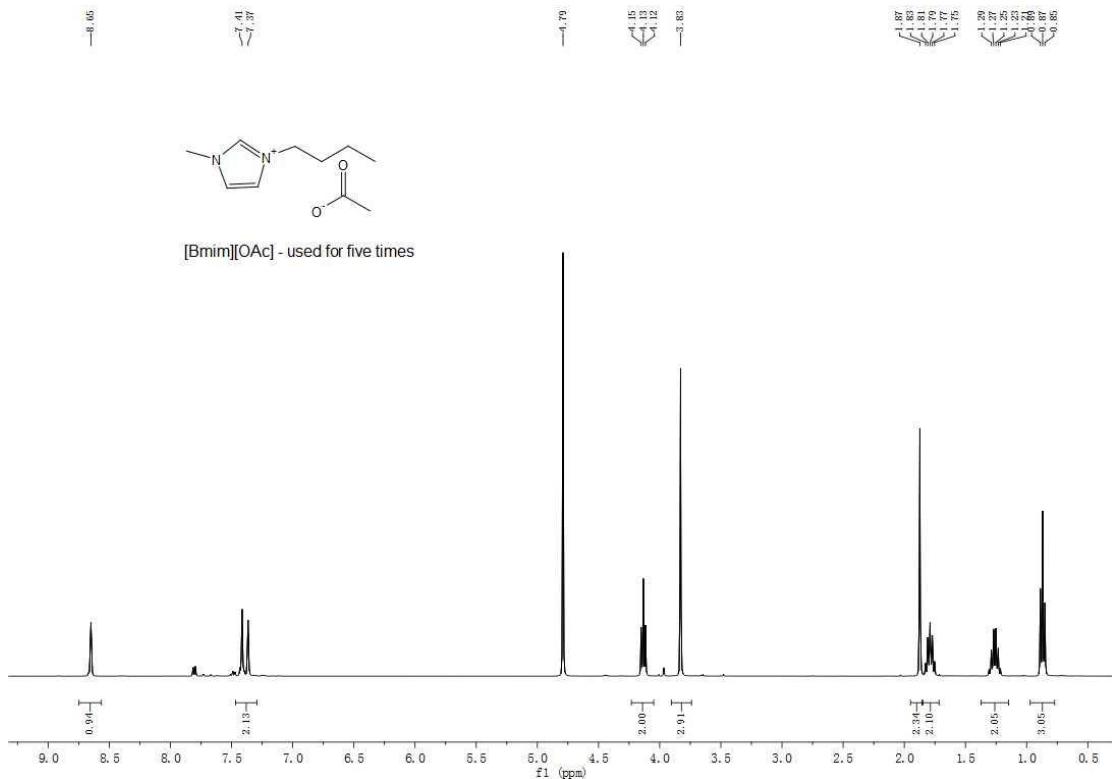
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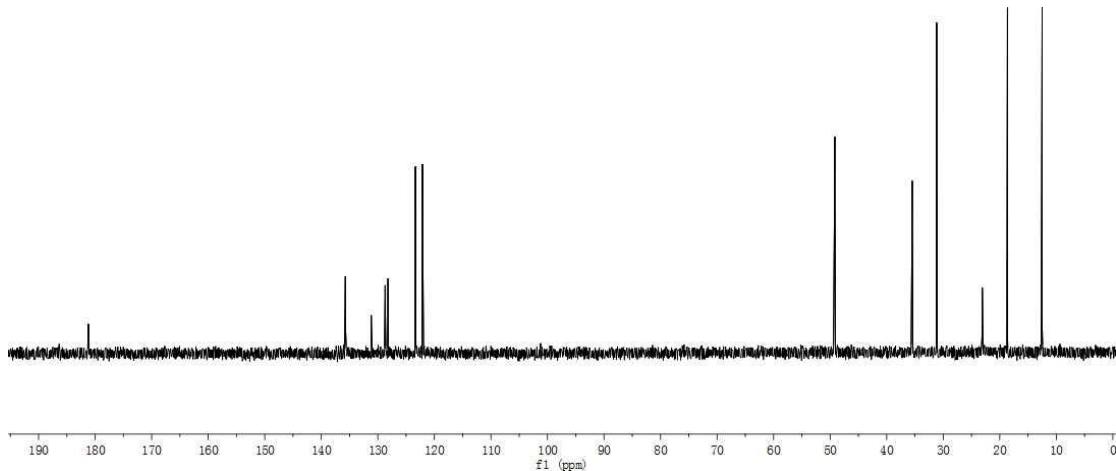


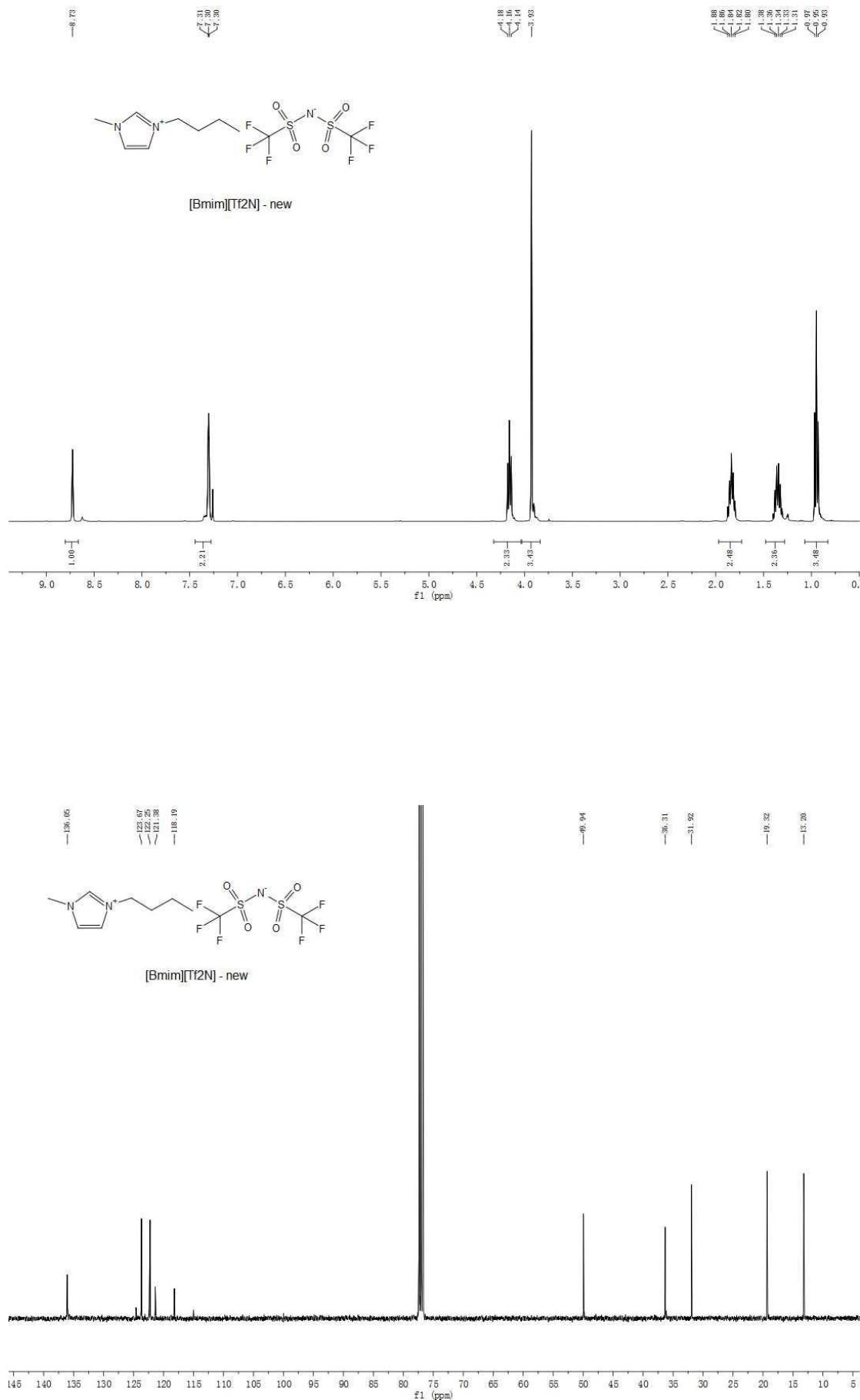


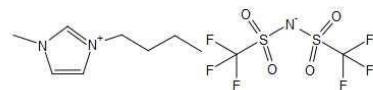
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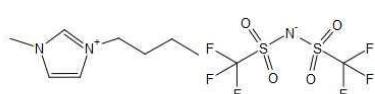
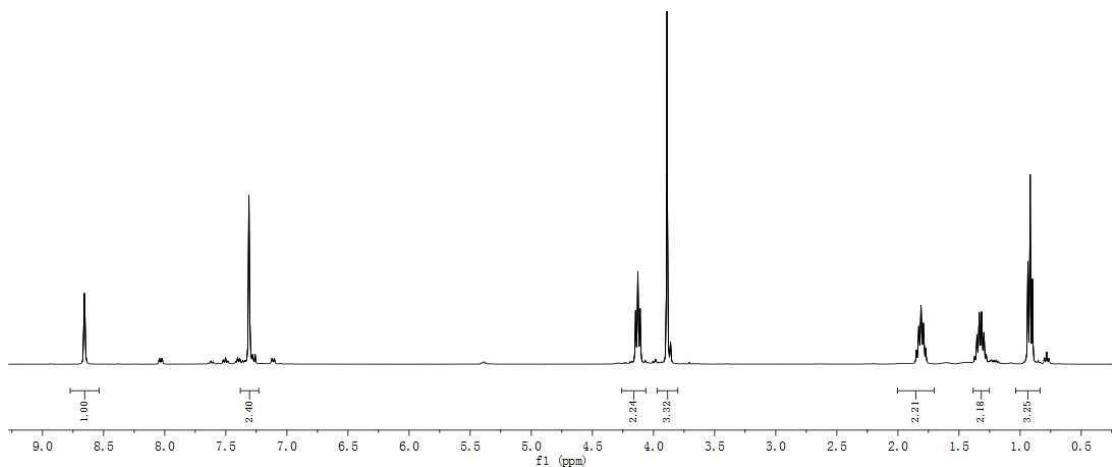
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