

## Supplementary data

# A non-nitric acid method of adipic acid synthesis: organic solvent- and promotor-free oxidation of cyclohexanone with oxygen over hollow-structured Mn/TS-1 catalysts

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**Table S1** Comparison between the novel catalyst and the others reported previously.

Catalyst	Conv.	AA Sele.	Oxidant	Temp.	Solvent	Reference
Mn-HTS	68.4	93.1	O <sub>2</sub>	90	No	This work
Pt/Carbon/monolith	100	22	O <sub>2</sub>	140	Water	[1]
Pt/Carbon	100	38.8	O <sub>2</sub>	140	Acetic acid/water	[2]
Mn(OAc) <sub>2</sub>	99.8	75	O <sub>2</sub>	65	Acetic acid/CF <sub>3</sub> COOH	[3]
Mn(NO <sub>3</sub> ) <sub>2</sub> /Co(NO <sub>3</sub> ) <sub>2</sub>	97.5	93.4	O <sub>2</sub>	40	Acetic acid/HNO <sub>3</sub>	[4]
Co/Mn cluster	97.6	86.6	O <sub>2</sub>	100	Acetic acid/water	[5]
H <sub>5</sub> PMo <sub>10</sub> V <sub>2</sub> O <sub>40</sub>	99	51	O <sub>2</sub>	70	Acetic acid/water	[6]
H <sub>7</sub> PMo <sub>8</sub> V <sub>4</sub> O <sub>40</sub>	98	55	O <sub>2</sub>	60	Acetonitrile/methanol	[7]

[1] E. Crezee, F. Kapteijn and J. A. Moulijn, *Catal. Today*, 2001, 69, 283–290.

[2] M. Besson, F. Gauthard, B. Horvath and P. Gallezot, *J. Phys. Chem. B*, 2005, 109, 2461–2467

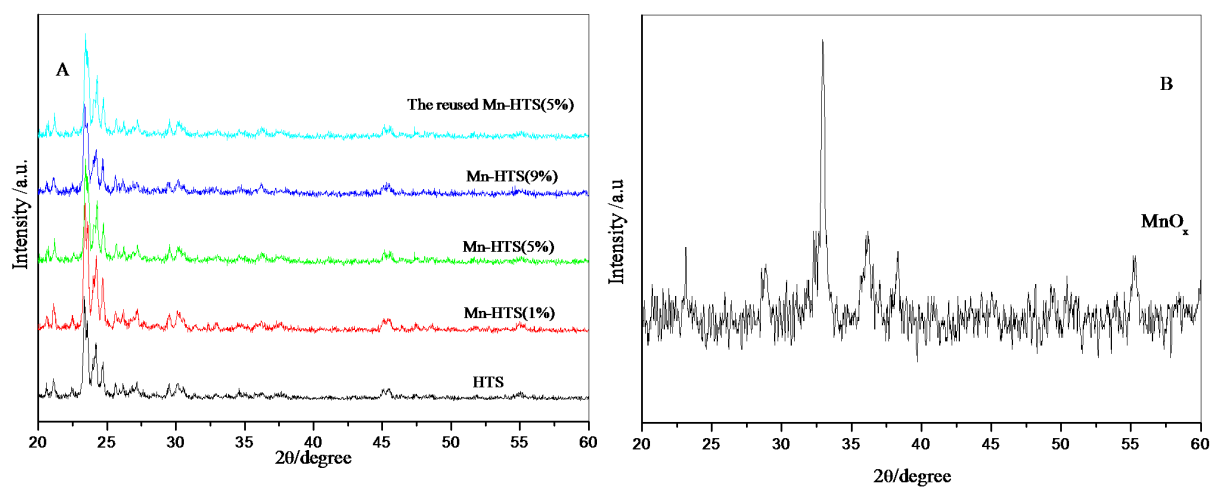
[3] M. Constantini and L. Krumenacker, *FR Patent 2541993*, 1983 (patent held by Rhone Poulenc).

[4] C. Fumagalli, F. Minisci and R. Pirola, *WO Patent 01/87815*, 2001 (patent held by Lonza SpA).

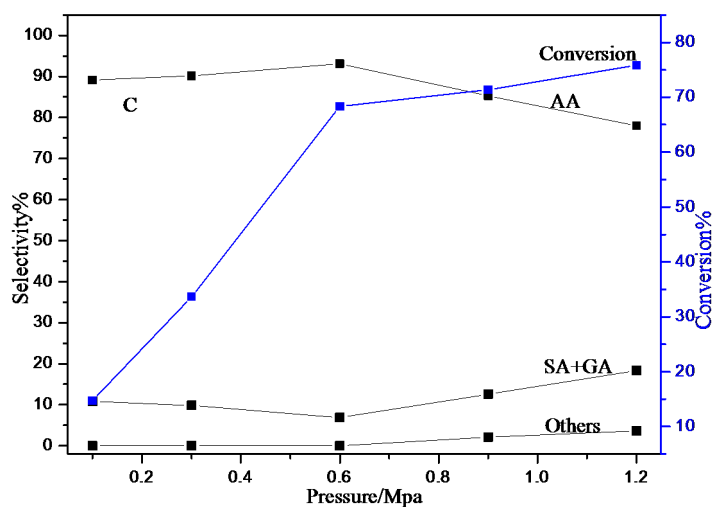
[5] S. A. Chavan, D. Srinivas and P. Ratnasamy, *J. Catal.* 2002, **212**, 39–45.

[6] A. Atlamsani, J.-M. Brégeault and M. Ziyad, *J. Org. Chem.*, 1993, **58**, 5663–5665.

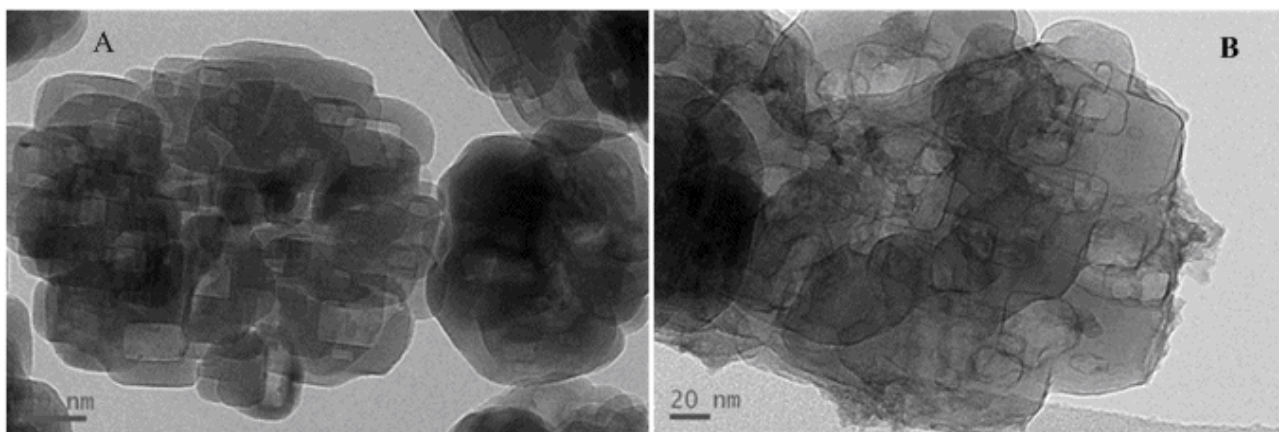
[7] J.M. Brégeault, E.A. Bassam and J. Martin, *US Patent 4983767*, 1991 (patent held by Rhone Poulenc Chimie).



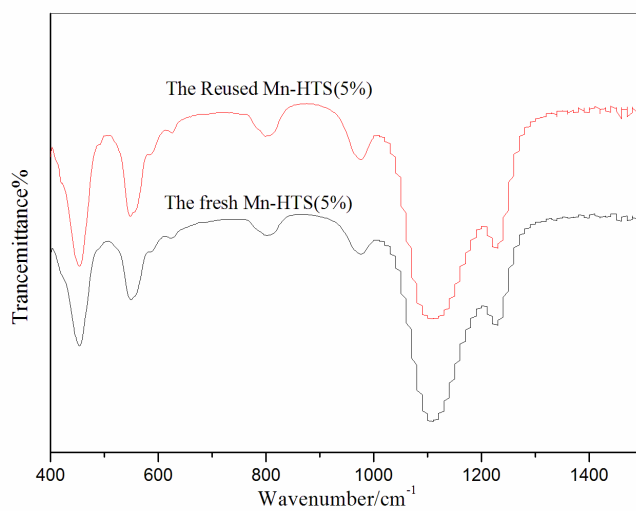
**Fig. S1** XRD of samples parent HTS, the xMn-HTS, the reused Mn-HTS(5%) (A) and MnO<sub>x</sub> (B)



**Fig. S2** Effect of the reaction pressure on the conversion of cyclohexanone and AA selectivity over the synthesized 5% Mn-HTS. (reaction conditions: 40 g cyclohexanone; GA+SA: glutarate acid and succinate acid; Others include CO<sub>2</sub> and CO).



**Fig. S3** TEM images of the reused Mn-HTS(5%)



**Fig. S4** FT-IR spectra of the fresh Mn-HTS (5%) and the reused Mn-HTS(5%)