

Comparison of microsublimation and ion exchange chromatography for boron isolation preceding its isotopic analysis via multi-collector ICP-

MS

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Supporting Information

1. Influence of remaining natural B on final $\delta^{11}\text{B}$ of isotopic reference materials

1.1 Boron isotope ratio of mixture of two sources

The final $^{11}\text{B}/^{10}\text{B}$ isotope ratio in the samples $(^{11}\text{B}/^{10}\text{B})_{\text{mix}}$ can be considered as a mixture of two sources, the remaining natural B $(^{11}\text{B}/^{10}\text{B})_{\text{bl}}$ and the spiked reference material AE120, AE121 or AE122 $(^{11}\text{B}/^{10}\text{B})_{\text{sa}}$. The final isotope ratio of the mixture can be calculated as:

$$\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{\text{mix}} = \frac{^{11}\text{B}_{\text{sa}} + ^{11}\text{B}_{\text{bl}}}{^{10}\text{B}_{\text{sa}} + ^{10}\text{B}_{\text{bl}}} \quad (\text{Eq. S1})$$

In this case, both $(^{11}\text{B}/^{10}\text{B})_{\text{bl}}$ and $(^{11}\text{B}/^{10}\text{B})_{\text{sa}}$ are known, as well as the total mass of B present in the blank and the sample. Hence, each of the four terms in the equation can be calculated. For example:

$$^{11}\text{B}_{\text{sa}} = n(\text{B}_{\text{sa}}) \times f^{11}\text{B}_{\text{sa}} \quad (\text{Eq. S2})$$

where $n(\text{B}_{\text{sa}})$ is the number of moles of B and $f^{11}\text{B}_{\text{sa}}$ is the fraction of ^{11}B in the standard spike. Both terms in this equation can be calculated following:

$$f^{11}\text{B}_{\text{sa}} = \frac{100 \times \left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{\text{sa}}}{1 + \left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{\text{sa}}} \quad (\text{Eq. S3}) \quad \text{and} \quad n(\text{B}_{\text{sa}}) = \frac{m(\text{B}_{\text{sa}})}{MM_{\text{sa}}}$$

(Eq. S4)

In the latter equation, $m(B_{sa})$ is the mass of B and MM_{sa} is the molar mass of B in the spiked standard. The molar mass can be calculated using the following equation:

$$MM_{sa} = \frac{f^{11}B_{sa}}{100} \times 11.00931 + \frac{100 - f^{11}B_{sa}}{100} \times 10.01924$$

(Eq. S5)

The three other terms in equation S1 can be determined in a similar way.

1.2 Calculation of $\delta^{11}B_{mix}$ in seawater, spinach and silicate glass

Using the equations outlined above, the $\delta^{11}B_{mix}$ values were calculated in order to evaluate if the low concentrations of remaining natural B in the sample matrices could have affected the final result of the spiked standard. In literature reported $\delta^{11}B$ values for seawater, plant material and NIST 610 silicate glass were used as input parameters for the calculation of $\delta^{11}B_{mix}$. All data input and final results of $\delta^{11}B_{mix}$ in seawater samples are given in Table S1. It is clear that final $\delta^{11}B$ values equal those of the reference materials.

Table S1 - Input parameters and final $\delta^{11}B$ values in seawater matrices spiked with certified boron reference materials

Seawater						
	Microsublimation			Chromatography		
$m(B_{bl})$ (ng)	0.3			0.6		
$\delta^{11}B_{bl}$ (‰)	39.6 [S1]			39.6		
$m(B_{sa})$ (ng)	800			4000		
$\delta^{11}B_{sa}$ (‰)	-20.2	19.9	39.7	-20.2	19.9	39.7
$\delta^{11}B_{mix}$ (‰)	-20.2	19.9	39.7	-20.2	19.9	39.7

To our knowledge, no $\delta^{11}B$ data on NIST 1570 spinach powder have been published. Therefore, two outer values of the range in $\delta^{11}B$ observed in plant material were used as input parameters, *i.e.* -30 and +40 ‰ [S2]. For microsublimated samples, a maximum deviation of 0.1 ‰ can be expected due to the 26 µg/l of natural B remaining in the samples.

Table S2 - Input parameters and final $\delta^{11}B$ values in spinach matrices spiked with certified boron reference materials

Spinach powder												
	Microsublimation						Chromatography					
$m(B_{bl})$ (ng)	1.3			1.3			2.6		2.6			
$\delta^{11}B_{bl}$ (‰)	-30			40			-30		40			
$m(B_{sa})$ (ng)	800			800			4000		4000			
$\delta^{11}B_{sa}$ (‰)	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7
$\delta^{11}B_{mix}$ (‰)	-20.2	19.8	39.6	-20.1	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7

Finally, the lowest and highest $\delta^{11}B$ values of NIST 610 silicate glass published in literature are -8.4 and 0.0 ‰, respectively [S3, S4]. When these values, together with a mass of remaining B

calculated from the limit of detection, are used as input parameters, no significant deviation from the certified values is expected.

Table S3 - Input parameters and final $\delta^{11}\text{B}$ values in silicate glass matrices spiked with certified boron reference materials

NIST 610 silicate glass												
	Microsublimation						Chromatography					
$m(\text{B}_{\text{bl}})$ (ng)	0.1			0.1			0.2			0.2		
$\delta^{11}\text{B}_{\text{bl}}$ (‰)	-8.4			0.0			-8.4			0.0		
$m(\text{B}_{\text{sa}})$ (ng)	800			800			4000			4000		
$\delta^{11}\text{B}_{\text{sa}}$ (‰)	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7
$\delta^{11}\text{B}_{\text{mix}}$ (‰)	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7	-20.2	19.9	39.7

2. Mass discrimination correction

In this work, a combination of internal correction using the common analyte internal standardization (CAIS) method and external correction relying on sample-standard bracketing (SSB) was used. The isotopic reference material Li_2CO_3 IRMM-016 was spiked to all standards and samples as an internal standard. Pure NIST SRM 951 boric acid bracketing standards and 2 M HCl - doped standards were included in the MC-ICP-MS measurement sequence.

The first step in the application of CAIS internal correction, is establishing a linear relationship between the measured (uncorrected) $^{11}\text{B}/^{10}\text{B}$ and the $^7\text{Li}/^6\text{Li}$ isotope ratios in the pure and 2 M HCl doped standards, $(^{11}\text{B}/^{10}\text{B})_{\text{ref,meas}}$ and $(^7\text{Li}/^6\text{Li})_{\text{ref,meas}}$, respectively. An example of such a linear relationship is given in Figure S1. Hence, for each sample, the expected measured $(^{11}\text{B}/^{10}\text{B})_{\text{ref,meas}}$ can be calculated from the observed $(^7\text{Li}/^6\text{Li})_{\text{ref,meas}}$. The ratio of the true $(^{11}\text{B}/^{10}\text{B})_{\text{ref,true}}$ to the calculated $(^{11}\text{B}/^{10}\text{B})_{\text{ref,meas}}$ now gives the extent of mass discrimination and this factor is subsequently used to correct the measured $(^{11}\text{B}/^{10}\text{B})_{\text{sample,meas}}$ into $(^{11}\text{B}/^{10}\text{B})_{\text{sample,corr}}$. This procedure can be summarized via the following equations:

$$\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{\text{sample,corr}} = \left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{\text{sample,meas}} \times \frac{(^{11}\text{B}/^{10}\text{B})_{\text{ref,true}}}{(^{11}\text{B}/^{10}\text{B})_{\text{ref,meas}}} \quad (\text{Eq.S6})$$

Where:

$$(^{11}\text{B}/^{10}\text{B})_{\text{ref,meas}} = a \times \left(\frac{^7\text{Li}}{^6\text{Li}}\right)_{\text{ref,meas}} + b \quad (\text{Eq.S7})$$

and $\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{ref,true}$ is 4.0436 ± 0.0027 [S5]

Finally, in equation S7, the parameters a and b are deduced from the linear regression between $\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{ref,meas}$ and $\left(\frac{^7\text{Li}}{^6\text{Li}}\right)_{ref,meas}$ in the pure and doped standards.

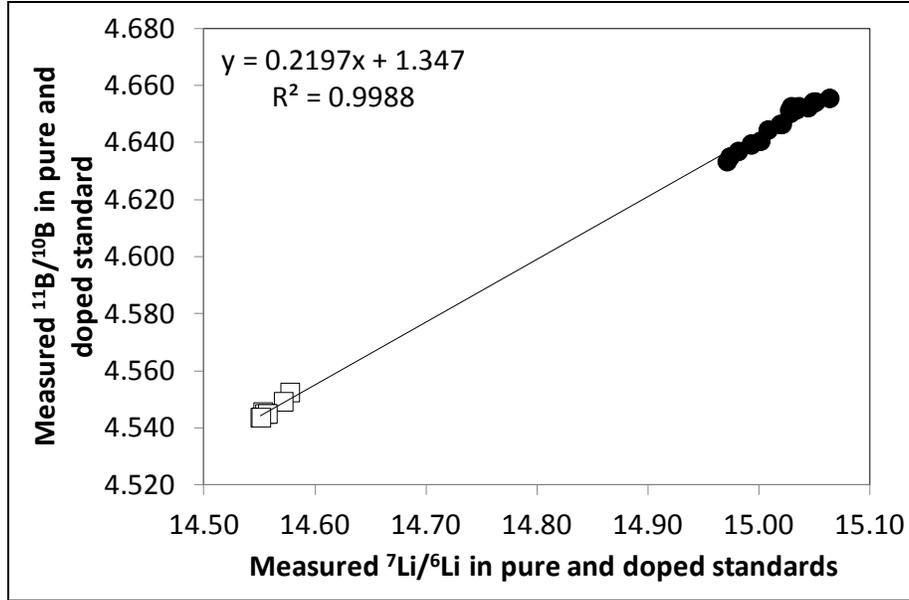


Figure S1 - Linear correlation between $\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{ref,meas}$ and $\left(\frac{^7\text{Li}}{^6\text{Li}}\right)_{ref,meas}$ observed in pure (●) and HCl doped (□) standards.

After correction using the CAIS method, external correction, relying on SSB, is applied according to equation S8.

$$\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{corr,final} = \left(\frac{\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{sample,corr}}{\left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{std-1,corr} + \left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{std+1,corr}} \right) \times \left(\frac{^{11}\text{B}}{^{10}\text{B}}\right)_{ref,true}$$

(Eq.S8)

3. Boron concentration in procedural blanks after microsublimation and ion exchange chromatography

Boron concentrations in procedural blanks were determined after four-fold dilution in 0.29 M HF. Determined concentration in each replicate, although typically < LOD, is shown in Table S4.

Table S4 – B concentration in procedural blank samples after microsublimation and ion exchange chromatography

	Microsublimation			Chromatography		
	A	B	C	A	B	C
Ca ²⁺ solution	< 0.08	< 0.08	< 0.08	< 0.08	< 0.08	< 0.08
Seawater	< 0.08	< 0.08	< 0.08	< 0.08	< 0.08	< 0.08
Spinach	< 0.08	< 0.08	0.44	0.50	0.40	0.60
Silicate glass	< 0.08	< 0.08	< 0.08	0.60	0.40	< 0.08

4. Investigation of correlation between $\delta^{11}\text{B}$ offset and B recovery in individual samples

Figure S2 plots $\delta^{11}\text{B}$ offset against the B recovery for each individual sample. The offset is defined as the difference between the experimentally determined $\delta^{11}\text{B}$ value and the corresponding reference value. The horizontal lines represent the 0.6 ‰ expanded uncertainty as indicated on the certificate of ERM AE 120, 121 and 122 $\delta^{11}\text{B}$ reference materials. From the figure, it is clear that within each sample set, no correlation between offset and recovery can be found.

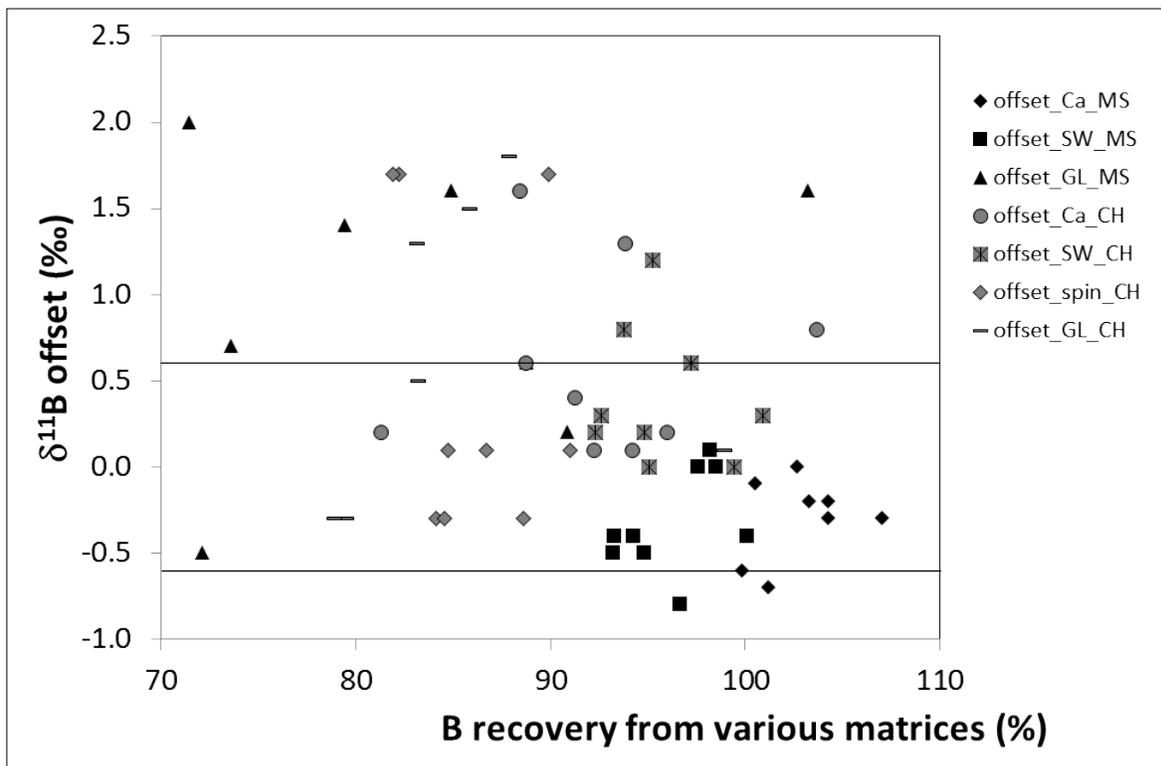


Figure S2 - Scatterplot of $\delta^{11}\text{B}$ offset vs. B recovery in individual samples

4. Reference

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[S4] Jochum, K.P., Weis, U., Stoll, B., Kuzmin, D., Yang, Q., Raczek, I., Jacob, D.E., Stracke, A., Birbaum, K., Frick, D.A., Günther, D., Enzweiler, J., *Geostand Geoanal Res*, 2011, **35**(4), 397-429.

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