

Synthesis of Periodic Mesoporous Organosilicas with Incorporated Aluminium

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Electronic Supplementary Information

Figure S1. Powder X-ray diffraction patterns of as-synthesised Al-CH₂-CH₂-PMO with different Si/Al ratios.

Figure S2. ²⁷Al MAS NMR spectra of Al-CH₂-CH₂-PMOs (as-synthesised) with different Si/Al ratios.

Figure S3. ¹H-²⁹Si MAS NMR spectra of as-synthesised Al-CH₂-CH₂-PMOs with different Si/Al ratios.

Identification of acidic sites by FTIR spectroscopy of adsorbed pyridine.

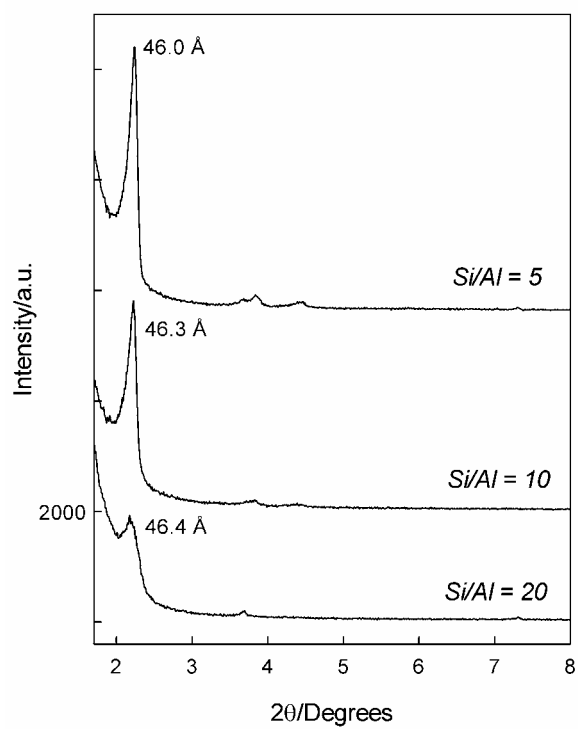


Figure S1. Powder X-ray diffraction patterns of as-synthesised Al-CH₂-CH₂-PMO with different Si/Al ratios.

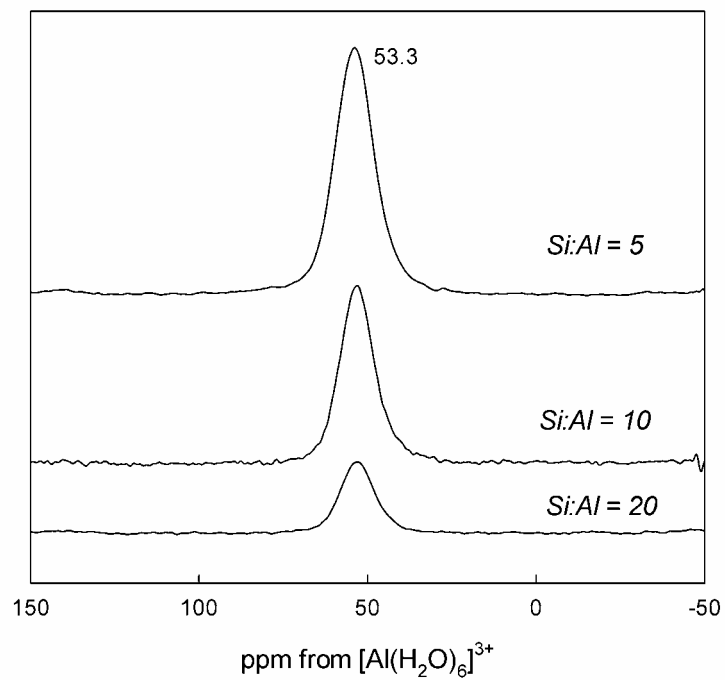


Figure S2. ^{27}Al MAS NMR spectra of Al-CH₂-CH₂-PMOs (as-synthesised) with different Si/Al ratios.

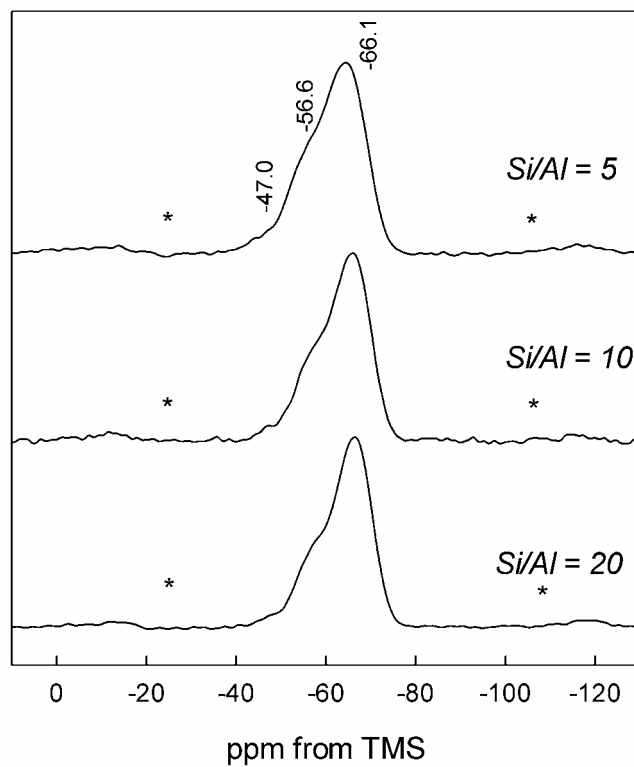


Figure S3. ^1H - ^{29}Si MAS NMR spectra of as-synthesised Al-CH₂-CH₂-PMOs with different Si/Al ratios.

Identification of acidic sites by FTIR spectroscopy of adsorbed pyridine

The adsorption of pyridine was carried on the Al-CH₂-CH₂-PMO with different contents of Al. The solids were thermally treated at 150°C for 24 hours before the adsorption of pyridine. The solids were pressed with KBr into the pellets (KBr was calcined at 550°C for 24 hours before use). The spectra were recorded on a PerkinElmer FTIR spectrometer.

The spectra of pyridine adsorbed on a Al-CH₂-CH₂-PMOs show the presence of several bands. The lines at 1540_(m.) and 1640_(m.) cm⁻¹ registered for the samples with different Al-contents are indicative of pyridine bound to Bronsted acidic sites, while the line at *ca.* 1490_(s.) cm⁻¹ can be ascribed to the pyridine bound to both Bronsted and Lewis acidic sites. The intense peak at 1444_(s.) cm⁻¹ is attributable to the H-bonded pyridine typical of the silica based molecular sieves.